

Article

Subscriber access provided by Northern Illinois University

Catalyst-free Rearrangement of Allenyl Aryldiazoacetates into 1,5-Dihydro-4H-pyrazol-4-ones

Kostiantyn O. Marichev, Huang Qiu, Austin C. Offield, Hadi Arman, and Michael P Doyle J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.6b01833 • Publication Date (Web): 08 Sep 2016 Downloaded from http://pubs.acs.org on September 9, 2016

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



The Journal of Organic Chemistry is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Catalyst-free Rearrangement of Allenyl Aryldiazoacetates into 1,5-Dihydro-4*H*-pyrazol-4-ones

Kostiantyn O. Marichev, Huang Qiu, Austin C. Offield, Hadi Arman, and Michael P. Doyle*

Department of Chemistry, The University of Texas at San Antonio, San Antonio, Texas 78249, United

States

E-mail: michael.doyle@utsa.edu

ABSTRACT: Phenylpropargyl diazoacetates exist in equilibrium with 1-phenyl-1,2-dien-1-yl diazoacetate - allenes that are rapidly formed at room temperature through 1,3-acyloxy migration catalyzed by gold(I) or gold(III) compounds, and these catalysts react solely with the π -donor rather than with the diazo group. The product allene of the aryldiazoacetates undergoes rearrangement that is not catalyzed by gold in which the terminal nitrogen of the diazo functional group adds to the central carbon of the allene initiating a sequence of bond forming reactions resulting in the production of 1,5-dihydro-4*H*-pyrazol-4-ones in good yields. These 1,5-dihydro-4*H*-pyrazol-4-ones undergo intramolecular 1,3-acyl migration to form an equilibrium mixture and can quantitatively transfer the acyl group to an external nucleophile with formation of 4-hydroxypyrazoles. Reactions of phenylpropargyl phenyldiazoacetates catalyzed by cationic gold complexes are initiated at the diazo functional group to form a gold carbene whose subsequent cascade process (intramolecular addition then aromatic substitution) results in the formation of a product that is uniquely characteristic of this pathway.



INTRODUCTION

 Gold catalysis provides powerful methodologies for unique metallo-organic transformations.¹ Activation of organic functional groups by gold(I) and gold(III) complexes initiates diverse pathways for often unique reactions.² One of the most useful of these transformations has been that of propargyl esters,³ which form allenes by 1,3-acyloxy migration, that undergo varied catalytic reactions and in the presence of weak bases are converted to conjugated dienes.⁴ We have been intrigued with the reactions of propargyl diazoacetates because these substrates contain two reactive centers for gold catalysts, and both are known to initiate reaction processes that are characteristic of their requisite functional groups.^{2,5} In particular, the diazo functionality of a propargyl phenyldiazoacetate is expected to react with the gold catalyst that functions as a metallo- σ -bond acceptor to displace dinitrogen through back-donation and form a metal carbene. This metal carbene is proposed to undergo intramolecular addition to the proximal carbon-carbon triple bond to form a derivative metal carbene that completes the domino transformation by electrophilic substitution into the aromatic nucleus of the original aryldiazoacetate (Scheme 1).^{6,7} π -Bond-forming reactions by gold catalysts that occur initially at the carbon-carbon triple bond proceed by nucleophilic addition, then 1,3-acyloxy migration, to a gold-coordinated allene⁸ from which an acylium ion intermediate⁹ may be the key to undetermined product(s). Both pathways are well documented, although the product outcome of the π -bond initiated pathway is revealed here for the first time.

Scheme 1. Catalyst-Dependent Divergence of Reaction Pathway Based on σ -Bond Association with a Diazo Functional Group versus π -Bond Association with an Alkyne.



RESULTS AND DISCUSSION

Selectivity. Initial assessment of gold catalysis was evaluated with propargyl aryldiazoacetates using cationic gold(I) catalysts, including those formed from ligated gold(I) chloride and silver salts. Cationic gold catalysts are well-established σ -bond acceptors with aryldiazoacetates,⁵ and they have been widely used for dinitrogen extrusion reactions of diazo compounds that are reported to produce the corresponding gold carbenes. Catalytic reactions of phenylpropargyl phenyldiazoacetate **1** with representative Au(I)⁺ catalysts formed the product from

The Journal of Organic Chemistry

the carbene-initiated domino transformation **2** in good yield (Scheme 2). Complete conversion of the reactant was observed at room temperature, and no other discernable product was identified. Each of the gold catalysts was soluble in the reaction solution.

Scheme 2. Domino Reaction of Propargyl Phenyldiazoacetate 1 Catalyzed by Cationic Gold(I).



Neutral gold(I) and gold(III) compounds are reported to have varying influences on reactions with diazo and alkyne functional groups.^{1,2,5} A survey of representative compounds for reactions with **1** at 20 °C revealed that 5 mol % of chloro(tetrahydrothiophene)gold(I) [AuCl(C_4H_8S)] catalyzed its complete conversion to three discernable products in the highest yield (Scheme 3). The surprising conversion of **1** to **3** and **4** retained all of the atoms of the reactant but gave extensive rearrangement, including the conversion of the reactant ester to product 1,5-dihydro-4*H*-pyrazol-4-ones.

Scheme 3. Products from Gold(I)-Catalyzed Rearrangement of Phenylpropargyl Phenyldiazoacetate 1.



Compounds **3** and **4** were observed spectroscopically to be related as structural isomers, and the crystal structure of **4** was obtained (Figure 1a) after its isolation from the reaction mixture. Recognition of **3** and **4** as potential benzoyl transfer agents prompted us to convert **3** and **4**, individually and as a mixture, quantitatively to the same 4-hydroxypyrazole **6** by treatment with nucleophilic bases that included hydrazines and amines (X-Ray structure of analog **6'**, Figure 1b). Compound **5**, although present in very low yield, was confirmed by spectral and chromatographic comparison with the previously characterized compound.¹⁰

 Examination of this reaction as a function of solvent did not provide any improvement in the product yield, but did show significant variation in the **3:4** product ratios (see Supporting Information). To determine if this product variation could be due to an equilibrium between **3** and **4** under the reaction conditions, the formation of **3** and **4** in DCE was followed as a function of time over ten days at room temperature with high product accountability, and the results from this investigation (Figure 2) clearly show an equilibrium between **3** and **4**.



Figure 1. X-Ray structures of 1,5-dihydro-4*H*-pyrazol-4-one **4** and deacylation product **6'** with 50% thermal ellipsoid probability.

Slow deacylation resulting in 6 complicates the overall picture, but conclusions can be drawn that compound 3 is the product of kinetic control at room temperature, and 4 is the product of thermodynamic control.



Figure 2. Time course for the formation of **3**, **4**, and **6** in DCE at 20 °C. Yields were determined by ¹H NMR and HPLC analyses of the reaction mixture after selected periods of time using 1,3,5-trimethoxybenzene as the internal standard; reported yields are the averages of two runs.

Experiments performed at higher temperatures (50 and 84 °C) are consistent with this interpretation. Compound **4** is the major isomer even at lower conversions of diazo compound **1**. At 50 °C the maximum total yield of 3+4 is 81% at a 12 h reaction time, and the ratio 3:4 is 2:3; whereas at 84 °C the reaction is complete

Page 5 of 24

The Journal of Organic Chemistry

(total yield of 3+4 is 80%) after 6 h, and 3:4 is 1:1. Continuing the reaction further at these higher temperatures leads to lower product yield as the result of deacylation, from which 4-hydroxypyrazole 6 is formed.

Cationic gold(I) and the neutral chloro(tetrahydrothiophene)gold(I) catalysts are mutually exclusive for product formation from reactions with **1**. Both catalysts are active at room temperature, and their application provides products from reactions at the diazo functional group [**2** from cationic gold(I)] and the alkyne functional group [**3**, **4**, and **5** from AuCl(C₄H₈S)] in good yields. Neutral gold(I) chloride complexes having strongly coordinating ligands (Table 1) show very low or negligible conversion of phenylpropargyl phenyldiazoacetate **1** under the same conditions, even to reaction times of three days.

Entry ^a	Gold Catalyst	Temp (°C)	Conversion (%)	Yield (%)	Yield (%)	Ratio	Yield (%)
			1	2^{b}	$3+4^{b}$	3 : 4 ^{<i>b</i>}	5^{b}
1	AuCl(IPr)	84 ^d (20)	68 (<5)	46	_	_	_
2	AuCl(CO)	84 (20)	>95 (14)	16 (9)	14	47:53	9
3	AuCl[P(OMe) ₃]	84 ^c (20)	40 (<5)	25	_	_	3
4	AuCl(PPh ₃)	84 ^c (20)	35 (<5)	23	_	_	_
5	AuCl(PMe ₃)	84 ^c (20)	45 (<5)	20	8	40:60	6
6		84 ^c (20)	70 (<5)	57	_	-	_
7	$\operatorname{AuCl}(C_4H_8S)^i$	20	>95	-	76	75:25	2
8	$AuCl(C_4H_8S)$	84	>95	-	80	45:55	4
9	$AuCl(Me_2S)$	20	>95	-	42	62:38	3
10	AuCl(Me ₂ S)	84	>95	-	46	45:55	5
11^e	AuPicCl ₂ ^j	20	>95	_	21	50:50	14
12 ^f	AuPicCl ₂ ^j	84	>95	_	14	40:60	18
13 ^g	AuCl ₃ (Pyridine) ^{<i>i</i>}	20	>95	_	25	60:40	35
14^h	AuCl ₃ (Pyridine) ^{<i>i</i>}	84	>95	_	18	45:55	39

 Table 1. Screening of Gold(I) and Gold(III) Catalysts in Reactions with Phenylpropargyl Phenyldiazoacetate 1.

^{*a*}Reactions were performed on 0.10 mmol scale: a solution of **1** (0.10 mmol) in DCE (1.0 mL) was added to a solution of Au-catalyst (5 mol %) in DCE (1.0 mL) under a nitrogen atmosphere at 20 °C (% conversion in parenthesis for entries 1-6) or in refluxing DCE, and the resulting reaction mixture was stirred for defined periods of time. ^{*b*}Conversion of **1**, yields of **2**, (**3**+**4**), **5**, and ratios **3**:**4** were determined by ¹H NMR analysis of characteristic peaks using 2,4,6-trimethoxybenzene as the internal standard. Reaction times were 72 h for the reactions carried out at 20 °C, and 6 h at 84 °C. Reaction times: ^{*c*}12 h; ^{*d*}48 h. Yield of product **6**: ^{*e*}24%; ^{*f*}27%; ^{*g*}32%; ^{*h*}38%. ^{*i*}At 50 °C after 12 h: 81% yield of **3**+4 with **3**:**4** = 40:60. ^{*j*}Reduction of the catalyst to elemental gold is significant.

However, heating these complexes to the temperature of refluxing 1,2-dichloroethane for up to 48 h gives 2 exclusively or as the major product (entries 1-6), albeit usually with low product yield and, generally, low % conversion.¹¹ The dimethylsulfide coordinated AuCl (entries 9-10) is not as effective a catalyst as its tetrahydrothiophene analog (entries 7-8) with product yields only about half those found with AuCl(C_4H_8S). Gold(III) catalysts (entries 11–14) are also less efficient than the AuCl(R_2S) catalysts in the formation of compounds **3** and **4**. However, diene **5** is formed in the presence of these catalysts in yields up to 39%, and there is no evidence for the formation of **2**. In separate experiments we have established that the dimethylsulfide ligand of AuCl remains on Au(I) throughout the reaction; also, analysis by ¹H NMR spectroscopy shows no apparent association of AuCl(Me₂S) with the diazo functional group of methyl phenyldiazoacetate at room temperature, and <5% diazo decomposition occurred over 24 h.

Role of Allene. Close examination of the reaction process by NMR spectroscopy revealed that the key transformation in the formation of **3** and **4** was the generation of allene (Scheme 1), which occurred within 5 min at 20 °C. For these investigations the dimethyl analog (7) of cyclopentylpropargyl phenyldiazoacetate **1** was used (Eq 1) to assess the course of the reaction. Allene **8** formation, which occurs by 1,3-acyloxy transfer, begins to take place immediately upon addition of the catalyst [5 mol % AuCl(Me₂S) or AuCl(C₄H₈S)] as the only reaction product, and reaches an apparent equilibrium (**7:8** = 45:55 with either catalyst).



With added gold catalyst up to 2 equiv. based on 7 no further change in the 7:8 ratio occurred. Confirmation of this process was made with phenyldimethylpropargyl benzoate under the same conditions (allene:alkyne = 3:2) and with 1 (allene:alkyne = 1:2). The validity of the reversibility of Au(I)-catalyzed [3,3]-rearrangements of propargylic esters has been reported,¹² and with propargyl diazoacetates additional stabilization by the diazo group of the cyclic carbocation intermediate on the pathway to allene formation could be expected. Furthermore, when the allene was isolated from the reaction mixture free of gold catalyst, the allene was observed to undergo slow conversion to the rearranged products corresponding to 3 and 4, but without any obvious formation of 5. This conversion was first order in 7 with a rate constant of 1.05×10^{-5} sec⁻¹ at 20 °C, and addition of AuCl(Me₂S) did not influence the reaction rate. The activation energy measured over 20–60 °C was 18.8 kcal/mol (R² = 0.999) with $\Delta H^{\ddagger} = 18.2$ kcal/mol and $\Delta S^{\ddagger} = -19.4$ cal/deg-mol at 298 K.

Reaction Mechanism. A stepwise mechanism for the neutral gold(I)-catalyzed rearrangement of arylpropargyl aryldiazoacetate **9** is given in Scheme 4. Consistent with the reported allene formation in related propargyl systems,¹³ the AuCl(R₂S) catalyzed 1,3-acyloxy rearrangement of **9** produces carboxyallene **C** that undergoes a unique rearrangement between allene **C** and the diazo functional group to form the observed **3** directly in a concerted fashion or stepwise through intermediate **D**. The rearrangement of **C** to **10** is a rare example of a diazo compound reacting as an *N*-electrophile,¹⁴ and the only one that involves an allene. Diene **12** is formed from

The Journal of Organic Chemistry

gold-activated carboxyallene **A**, presumably through gold-coordinated acylium ion intermediate **B**. Compound **10** is formed initially from **A** and undergoes acyl migration to reach equilibrium with **11**. This mechanism explains the formation of **10** from carboxyallene **C**, the equilibrium that exists between **10** and **11**, and the formation of only (*Z*)-diene **12** from **A** through **B** without evidence for the formation of the isomeric (*E*)-diene that was reported when the same reaction is performed in the presence of 4-chloropyridine-*N*-oxide.^{10,15}

Scheme 4. Proposed Mechanism for the Formation of 10, 11, and 12 from 9.



The formation of **12** is consistent with the results reported for the rearrangement of an acetate analog of $1,^{9c}$ and the conversion of **B** to **12** is consistent with the formation of diene **12** as the sole geometrical isomer. The difference between the two pathways for diene formation (in the presence and absence of pyridine-*N*-oxide) is that the pyridine-*N*-oxide intercepts activated allene **A** as a base to form both (*E*)- and (*Z*)-**12**, whereas, without the added base, **B** is formed and undergoes intramolecular proton transfer to form the observed (*Z*)-**12**.

Although diazo compounds are normally regarded as nucleophiles, with both the diazo carbon and the terminal nitrogen as nucleophilic centers,¹⁶ there are examples in which the terminal nitrogen atom of diazo compounds exhibits electrophilic reactivity.^{14b,17} These reactions are intermolecular and involve obvious nucleophilic attack on the terminal nitrogen of diazocarbonyl compounds and result in the formation of hydrazone derivatives. The formation of **10** can be considered to be initiated by intramolecular electrophilic addition of the *N*-terminus of the aryldiazoacetates onto the central carbon of the allene that is continued with hydrogen transfer and C4-C2 bond formation, which occurs with the release of the Ar₂CO acyl group. To confirm that the allene forms both (*Z*)- and (*E*)-dienes with a pyridine-*N*-oxide, the latter undergoing [4+2]-cycloaddition,¹⁰ propargyl phenyldiazoacetate **1** was combined with 1.1 equiv of 4-chloropyridine-*N*-oxide and heated at 60 °C in the presence of 5 mol % AuCl(C₄H₈S) for 6 h. Allene formation was observed prior to addition of 4-chloropyridine-*N*-oxide, and both the

(*Z*)-diene **5** and the cycloaddition product from the (*E*)-diene were the sole reaction products (Scheme 5). Diene formation is not observed in the absence of gold(I) catalyst, but with only 20 mol % of 4-chloropyridine-*N*-oxide all four products (**3**, **4**, (*Z*)-**5** and (*E*)-**5**) were obtained from the reaction performed with 5 mol % AuCl(C_4H_8S) at 60 °C for 6 h.¹⁸

Scheme 5. Role of the Pyridine-N-oxide in Diene Formation from Phenylpropargyl Phenyldiazoacetate 1.



Substrate Scope. The substrate scope of phenylpropargyl aryldiazoacetates 9a-k (Table 2) demonstrates that the AuCl(C₄H₈S)-catalyzed reaction is broadly appropriate for the synthesis of the 1,5-dihydro-4*H*-pyrazol-4-ones 10 and 11, and these reactions are tolerant of both electron-donating and electron-withdrawing substituents in the aryl group of the aryldiazoacetate. High total yields of the 10+11 composites are observed, and their 10:11 ratio is near 3:1 (Table 2). Substrate 9j with a 12-membered ring gives a high yield of the 10j+11j composite, but its 10:11 ratio is 3:2. Tetrahydropyranyl derivatives 10k+11k are obtained in only 46% yield and a 78:22 10:11 ratio.

Table 2. Substrate Scope of Gold(I)-Catalyzed Reaction of Phenylpropargyl Aryldiazoacetates 9.^a

	$ \begin{array}{c} $	$\frac{O_{\text{Ph}}}{N_{\text{N}}} + \frac{V_{\text{Ph}}}{V_{\text{H}}}$	O Ph N-N Ar H
0.11	\mathbf{y}	IU X:-11 (0/) ^{b. C}	D-4:- 10.11 ^b
9-11	$Ar(Y = CH_2)$	Y 1eld (%)	Katio 10:11
a	C_6H_5	76	75:25
b	p-MeO-C ₆ H ₄	75	72:28
с	p-Me-C ₆ H ₄	85	72:28
d	p-Ph-C ₆ H ₄	78	88:12
e	p-Br-C ₆ H ₄	87	85:15
f	p-Cl-C ₆ H ₄	83	83:17
g	<i>p</i> -CF ₃ -C ₆ H ₄	73	70:30

The Journal of Organic Chemistry

h	$2-C_4H_3S^d$	67	88:12 Ratio 10:11 ^b	
9–11	$Y (Ar = C_6 H_5)$	Yield (%) ^{<i>b</i>, <i>c</i>}		
i	(CH ₂) ₂	73	78:22	
j	$(CH_2)_8$	86	59:41	
k	CH ₂ O	46	78:22	

^aReactions were carried out at 20 °C on a 0.20 mmol scale: a solution of 1 in 2.0 mL DCE was added to a 5 mol % solution of AuCl(C₄H₈S) in 2.0 mL DCE under a nitrogen atmosphere, and the resulting reaction mixture was stirred for 72 h. ^bRatios and yields were determined by ¹H NMR spectroscopy using 2,4,6trimethoxybenzaldehyde as the internal standard. ^cIndividual compounds 10 and 11 were isolated and fully characterized (see Experimental Section and SI). ${}^{d}2$ -C₄H₃S = 2-Thiophene.

With substrates having electron-withdrawing or electron-donating groups (EWG or EDG) in both arene nuclei, the product outcome is quite similar (Scheme 6) to those presented in Table 2. However, having an EWG as Z and an EDG as X (compound 90) results in a lower yield of 100+110 (47%) but a higher 10:11 ratio of 89:11. High yield (79%) and almost exclusive formation of compound 10q (10:11 = 91:9) are observed from the dimethyl analog 9q, whereas aryl unsubstituted analog 9p results in lower yield of 10p+11p (68%) with a 88:12 10:11 ratio.

Scheme 6. Substrate Scope of Gold(I)-Catalyzed Reaction of Arylpropargyl Aryldiazoacetates 9.

NL-							
		10,11	Х	Υ	Ζ	Yield (%)	10:11
	5 mol % AuCl(C ₄ H ₈ S)	1	CI	CH ₂	н	80	77:23
	DCE, 20 °C,	m	OMe	CH_2	Н	77	62:38
	72 h	n	OMe	CH_2	OMe	79	70:30
		0	OMe	CH_2	CF_3	47	89:11
X		р	Н	Н	Н	68	88:12
9		q	Н	Н	OMe	79	91:9

The cyclobutyl analog 9r (Scheme 7) forms neither 10 nor 11. Instead, treatment of 9r with 5 mol % AuCl(C₄H₈S) affords 13 in good yield (70%), but only at 80 °C over 12 h. This product is consistent with initial formation of a gold-carbene intermediate through the diazo functional group. The alternate pathway through a strained allene intermediate is apparently too restrictive and, although the initial step in the 1,3-acyl rearrangement that produces the stabilized intermediate E is favorable, release of the C-O bond to form the allene does not occur.

Scheme 7. Outcome of Gold(I)-Catalyzed Reaction of Cyclobutyl Propargyl Reactant 9r.



Diazoacetate Influence. The conversion of **A** to **10** in Scheme 4 was expected to be dependent on the relative basicity of the ester carbonyl for 1,3-acyl transfer and the relative electrophilicity of the diazo terminal nitrogen for rearrangement. To evaluate these dependencies we prepared acceptor and acceptor-acceptor analogues (**14** and **15**) of the donor-acceptor phenylpropargyl aryldiazoacetate **9**, as well as another donor-acceptor diazo compound **16**, and performed gold-catalyzed reactions with them (Schemes 8a and 8b) under the same conditions as those performed with **1** (see Table 1).

Scheme 8a. Product Dependence on Diazo Compound Catalyzed by AuCl(C₄H₈S).



Scheme 8b. Product Dependence on Diazo Compound Catalyzed by AuPicCl₂.



The Journal of Organic Chemistry

In all cases the allene intermediate was formed and reached equilibrium within 5 min, but the rearrangement product 22 was not observed (Scheme 9). In contrast to reactions with propargyl aryldiazoacetates 1 or 9, however, both 14 and 15 were unreactive with $AuCl(C_4H_8S)$ at room temperature but in refluxing 1,2-dichloroethane underwent cleavage to form enone 17 (Scheme 8a), whose formation is consistent with the pathway through intermediate B in Scheme 4 (equivalent to F in Scheme 9), and with 18, the acid-catalyzed elimination product for which the diazoacetic acid is the leaving group. The absence of rearranged product 22 in reactions with 14 or 15 can be attributed to the diminished electrophilicity of their diazo terminal nitrogens, but the facility with which donor-acceptor diazo compound 16 forms cleavage product 17 instead of 22 is possibly related to other factors.

With the gold(III) catalyst AuPicCl₂ the same reactions occurred with diazo compounds 14–16 (Scheme 8b) but at lower temperatures and, except for 14, gave the same products as with AuCl(C₄H₈S). Diazoacetoacetate 14 gave diene 21 as the major product whose formation through intermediate F occurs in competition with the production of cleavage product 17. Once again, the rearranged product 22 or its acyl transfer product is absent in the product mixture, suggesting the uniqueness of the overall process that forms 22 and its analogues.

Scheme 9. Diazo Compound Alternate Pathways to Product Formation.



CONCLUSION

We have discovered a unique gold(I)-catalyzed rearrangement of propargyl aryldiazoacetates and a platform on which catalytic activity of gold complexes can be assessed. Arylpropargyl aryldiazoacetates have two reactive sites for reactions with gold complexes. If initial interaction occurs at the diazo carbon, a gold carbene is the outcome, and a subsequent cascade process results in the formation of a product that is uniquely characteristic of this pathway. If gold coordination occurs with the carbon-carbon triple bond, 1,3-acyloxy migration of the propargylic ester promoted by the diazoester takes place. The corresponding allene formed rapidly is a stable

intermediate, and it determines the course of subsequent processes, one of which is to undergo gold-catalyzed formation of a gold-acylacylium ion intermediate to form stable products by cleavage or intramolecular hydrogen migration. Alternatively, the allene intermediate undergoes a complex transformation, not catalyzed by gold, in which the terminal nitrogen of the diazo functional group adds at the central carbon of the allene to initiate a sequence of bond forming reactions resulting in the production of 1,5-dihydro-4*H*-pyrazol-4-ones in good yields. As acyl transfer agents these unique products undergo intramolecular 1,3-acyl migration to form an equilibrium mixture of two isomeric 1,5-dihydro-4*H*-pyrazol-4-ones under the reaction conditions or quantitatively transfer the acyl group to an external nucleophile with formation of 4-hydroxypyrazoles. Cationic gold(I) complexes initiate their reactions at room temperature exclusively with the diazo functional group of propargyl phenyldiazoacetates, whereas both AuCl(R₂S) and gold(III) catalysts undergo initial reaction exclusively at the carbon-carbon triple bond. Neutral gold(I) chloride complexes having strongly coordinating ligands show very low or negligible conversion of propargyl phenyldiazoacetates at room temperatures to give mainly the product from reaction at the diazo functional group, which may have occurred after chloride dissociation,^{2e} sometimes in competition with products from reaction at the carbon-carbon triple bond.

EXPERIMENTAL SECTION

General Information. All reactions were carried out under an atmosphere of dry nitrogen in oven dried glassware using freshly distilled solvents. All solvents were purified and dried using standard techniques. Thin Layer Chromatography (TLC) analyses were performed on precoated analytical plates Silica Gel 60 F_{254} , and visualized with the use of UV light or iodine stain (I₂ and Silica Gel 60). High-resolution mass spectra (HRMS) were performed on a microTOF-ESI mass spectrometer. Exact masses were reported for the molecular ions [M+Na]⁺, [M+H]⁺ or [M-H]⁻. Column chromatography was performed on a CombiFlash purification system using normal phase disposable columns. IR spectra were recorded using a FT-IR spectrometer. NMR spectra were recorded on spectrometers at 300, 400 or 500 MHz (¹H NMR) and 76, 100 or 126 MHz (¹³C NMR). Chemical shifts are reported in ppm using residue CHCl₃ (δ 7.26 ppm)/H₂O (δ 1.56 ppm), CH₃OH (δ 3.31 ppm)/H₂O (δ 4.87 ppm), DMSO (δ 2.50 ppm)/H₂O (δ 3.33 ppm) for ¹H NMR reference and the central resonance of CDCl₃ (δ 77.16 ppm), CD₃OD (δ 49.00 ppm), DMSO-d6 (δ 39.52 ppm) for ¹³C NMR reference. ¹H NMR spectra are reported as follows (s = singlet, br = broad singlet, d = doublet, t = triplet, q = quartet, p = pentet, m = unresolved multiplet, comp = composite of magnetically nonequivalent protons); coupling constants (*J*) are given in Hertz (Hz).

Materials. AuCl(PPh₃), AuCl(PMe₃), AuCl[P(OMe)₃], Chloro[2-dicyclohexyl(2',6'-dimethoxybiphenyl)phosphine]-gold(I), and AuPicCl₂ were purchased from Sigma-Aldrich; AuCl(C₄H₈S), AuCl(Me₂S), AuCl(CO), (IPr)AuCl, and AuCl₃(Py) were purchased from Strem Chemicals. All other chemicals were obtained from commercial sources and used as received without further purification.

Characterization of Propargyl Diazoacetates. Detailed procedures for the preparation of diazo compounds **9** and characterization data of compounds **1** (**9a**), **9c–g**, **9l**, **m** were previously reported.¹⁰ The reactions were carried out on a 5 mmol scale; reported are total yields of two steps: DCC coupling and diazo transfer.

The Journal of Organic Chemistry

1-(Phenylethynyl)cyclopentyl 2-Diazo-2-(4-methoxyphenyl)acetate (9b). 1.24 g, 69% yield. Orange oil. ¹H NMR (500 MHz, CDCl₃) δ 7.49 – 7.44 (comp, 2H), 7.42 (d, *J* = 8.9 Hz, 2H), 7.33 – 7.28 (comp, 3H), 6.95 (d, *J* = 9.0 Hz, 2H), 3.82 (s, 3H), 2.42 (ddd, *J* = 12.3, 7.8, 3.7 Hz, 2H), 2.35 – 2.26 (comp, 2H), 1.90 – 1.78 (comp, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 164.1, 158.0, 131.8, 128.3, 128.1, 125.9, 122.7, 117.1, 114.6, 89.5, 85.0, 82.0, 55.4, 40.8, 23.5. IR (neat) 2954, 2075, 1702, 1512, 1255, 1145, 997 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₂H₂₀N₂O₃Na [M+Na]⁺ 383.1366, found: 383.1361.

*I-(Phenylethynyl)cyclopentyl 2-Diazo-2-(thiophen-2-yl)acetate (***9***h).* 874 mg, 52% yield. Dark-red oil. ¹H NMR (500 MHz, CDCl₃) δ 7.46 (dd, *J* = 6.4, 2.9 Hz, 2H), 7.34 – 7.28 (m, 4H), 7.07 – 7.03 (m, 1H), 6.84 (d, *J* = 3.6 Hz, 1H), 2.49 – 2.41 (m, 2H), 2.31 (dt, *J* = 15.3, 7.7 Hz, 2H), 1.87 – 1.81 (m, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 163.4, 131.9, 128.3, 128.1, 126.9, 125.9, 125.4, 122.6, 120.9, 89.0, 85.3, 82.9, 40.8, 23.4. IR (neat) 2928, 2076, 1701, 1443, 1285, 1231, 1127, 974, 691 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₉H₁₆N₂O₂SNa [M+Na]⁺ 359.0825, found: 359.0822.

1-(Phenylethynyl)cyclohexyl 2-Diazo-2-phenylacetate (9i). 1.14 g, 66% yield. Yellow solid, m.p. 50 – 51 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.53 (dd, J = 8.5, 1.1 Hz, 2H), 7.50 – 7.45 (comp, 2H), 7.42 – 7.36 (comp, 2H), 7.34 – 7.28 (comp, 3H), 7.21 – 7.16 (m, 1H), 2.36 – 2.21 (comp, 2H), 2.09 (comp, 2H), 1.81 – 1.65 (comp, 4H), 1.62 – 1.53 (m, 1H), 1.45 (ddd, J = 13.1, 8.5, 4.8 Hz, 1H). ¹³C NMR (126 MHz, CDCl₃) δ 163.1, 131.9, 128.9, 128.3, 128.1, 125.8, 125.7, 124.0, 122.7, 89.1, 86.5, 77.1, 37.5, 25.2, 22.7. IR (neat) 2936, 2083, 1707, 1241, 754, 691 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₂H₂₀N₂O₂Na [M+Na]⁺ 367.1417, found: 367.1410.

I-(Phenylethynyl)cyclododecyl 2-Diazo-2-phenylacetate (9j). 1.33 g, 62% yield. Yellow solid, m.p. 92.5 – 93.5 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.53 (dd, *J* = 7.6, 0.9 Hz, 2H), 7.50 – 7.44 (comp, 2H), 7.42 – 7.35 (comp, 2H), 7.34 – 7.27 (comp, 3H), 7.19 (ddd, *J* = 8.6, 2.2, 1.1 Hz, 1H), 2.37 – 2.23 (comp, 2H), 2.12 – 1.98 (comp, 2H), 1.68 (d, *J* = 8.8 Hz, 2H), 1.43 (comp, 16H). ¹³C NMR (126 MHz, CDCl₃) δ 163.1, 131.9, 128.9, 128.3, 128.1, 125.8, 125.7, 124.0, 122.7, 89.4, 86.2, 79.6, 33.4, 26.1, 25.9, 22.3, 22.1, 19.4. IR (neat) 2928, 2079, 1705, 1470, 1241, 1146, 1006, 754 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₈H₃₂N₂O₂Na [M+Na]⁺ 451.2356, found: 451.2343.

4-(*Phenylethynyl*)*tetrahydro-2H-pyran-4-yl* 2-*Diazo-2-phenylacetate* (**9***k*). 1.12 g, 65% yield. Yellow solid, m.p. 85 – 86 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.57 – 7.45 (comp, 4H), 7.44 – 7.36 (comp, 2H), 7.36 – 7.28 (comp, 3H), 7.23 – 7.17 (m, 1H), 3.93 (dt, J = 9.2, 4.4 Hz, 2H), 3.85 (ddd, J = 11.9, 9.0, 2.9 Hz, 2H), 2.45 – 2.38 (comp, 2H), 2.23 (ddd, J = 13.1, 9.0, 4.0 Hz, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 163.0, 131.9, 128.9, 128.7, 128.2, 125.9, 125.4, 124.1, 122.2, 87.5, 87.5, 74.1, 64.5, 38.0. IR (neat) 2960, 2086, 1707, 1498, 1241, 1140, 755 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₈N₂O₃Na [M+Na]⁺ 369.1214, found: 369.1210.

l-((4-Methoxyphenyl)ethynyl)cyclopentyl 2-Diazo-2-(4-methoxyphenyl)acetate (**9n**). 1.13 g, 58% yield. Orange oil. ¹H NMR (500 MHz, CDCl₃) δ 7.41 (dd, J = 10.1, 9.0 Hz, 4H), 6.94 (d, J = 9.0 Hz, 2H), 6.82 (d, J = 8.9 Hz, 2H), 3.81 (s, 3H), 3.80 (s, 3H), 2.46 – 2.37 (m, 2H), 2.34 – 2.24 (m, 2H), 1.87 – 1.78 (m, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 164.1, 159.6, 158.0, 133.3, 125.9, 117.1, 114.8, 114.5, 113.8, 88.1, 84.9, 82.3, 55.34, 55.25, 40.8, 23.5. IR (neat) 2956, 2080, 1703, 1511, 1248, 1147, 830, 734 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₃H₂₂N₂O₄Na [M+Na]⁺ 413.1472, found: 413.1483.

l-((4-Methoxyphenyl)ethynyl)cyclopentyl 2-Diazo-2-[(4-(trifluorometh-yl)phenyl]acetate (90). 1.18 g, 55% yield. Orange oil. ¹H NMR (500 MHz, CDCl₃) δ 7.72 – 7.52 (comp, 4H), 7.40 (d, J = 8.6 Hz, 2H), 6.83 (d, J = 8.7 Hz, 2H), 3.81 (s, 3H), 2.50 – 2.37 (comp, 2H), 2.37 – 2.22 (comp, 2H), 1.85 (comp, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 162.8, 159.7, 133.3, 130.3, 127.5 (q, J = 32.5 Hz), 125.7 (q, J = 3.8 Hz), 123.5, 114.6, 113.8, 87.6, 85.3, 82.9, 77.3, 77.0, 76.8, 55.3, 40.8, 23.4. IR (neat) 2957, 2087, 1706, 1509, 1323, 1244, 1070, 832 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₃H₁₉F₃N₂O₃Na [M+Na]⁺ 451.1240, found: 451.1225.

2-*Methyl-4-phenylbut-3-yn-2-yl 2-Diazo-2-phenylacetate* (**9***p*). 958 mg, 63% yield. Orange oil. ¹H NMR (300 MHz, CDCl₃) δ 7.53 – 7.43 (comp, 4H), 7.41 – 7.35 (comp, 2H), 7.32 – 7.27 (comp, 3H), 7.21 – 7.14 (m, 1H), 1.85 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 163.4, 131.9, 128.9, 128.4, 128.2, 125.7, 125.7, 124.0, 122.5, 90.0, 84.5, 73.8, 29.4. IR (neat) 2936, 2079, 1705, 1498, 1349, 1247, 1120, 754, 691 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₉H₁₆N₂O₂Na [M+Na]⁺ 327.1104, found: 327.1102.

2-Methyl-4-phenylbut-3-yn-2-yl 2-Diazo-2-(4-methoxyphenyl)acetate (7 or 9q). 1.14 g, 68% yield. Orange oil. ¹H NMR (500 MHz, CDCl₃) δ 7.49 – 7.44 (comp, 2H), 7.42 (d, J = 9.0 Hz, 2H), 7.34 – 7.28 (comp, 3H), 6.95 (d, J = 9.0 Hz, 2H), 3.82 (s, 3H), 1.85 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 163.9, 158.0, 131.9, 128.4, 128.1, 126.0, 122.6, 117.2, 114.6, 90.1, 84.4, 73.7, 55.4, 29.5. IR (neat) 2988, 2079, 1705, 1514, 1258, 1122, 998, 827 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₀H₁₈N₂O₃Na [M+Na]⁺ 357.1210, found: 357.1209.

1-(Phenylethynyl)cyclobutyl 2-Diazo-2-phenylacetate (9r). 822 mg, 52% yield. Yellow solid, m.p. 76 – 77 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.52 (dd, *J* = 8.5, 1.1 Hz, 2H), 7.50 – 7.45 (comp, 2H), 7.42 – 7.36 (comp, 2H), 7.34 – 7.29 (comp, 3H), 7.23 – 7.16 (m, 1H), 2.81 – 2.71 (comp, 2H), 2.62 (ddd, *J* = 12.7, 9.8, 2.6 Hz, 2H), 2.15 – 2.05 (m, 1H), 2.00 (ddd, *J* = 9.8, 8.6, 4.9 Hz, 1H). ¹³C NMR (126 MHz, CDCl₃) δ 163.2, 131.9, 128.9, 128.4, 128.2, 125.8, 125.5, 124.0, 122.5, 89.2, 84.8, 73.3, 37.2, 14.7. IR (neat) 2952, 2081, 1704, 1497, 1351, 1242, 1144, 1088, 754, 691 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₀H₁₆N₂O₂Na [M+Na]⁺ 339.1104, found: 339.1107.

2-Methyl-4-phenylbut-3-yn-2-yl 2-Diazo-3-oxobutanoate (14). Prepared according to general procedure¹ from acetoacetic acid; 797 mg, 59% yield. Pale yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.49 – 7.40 (comp, 2H), 7.35 – 7.27 (comp, 3H), 2.49 (s, 3H), 1.83 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 190.3, 159.6, 131.8, 128.6, 128.2, 122.2, 89.2, 85.0, 74.7, 29.4, 28.3. IR (neat) 2988, 2137, 1719, 1655, 1315, 1121, 1058, 693 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₅H₁₄N₂O₃Na [M+Na]⁺ 293.0897, found: 293.0891.

2-*Methyl-4-phenylbut-3-yn-2-yl* 2-*Diazoacetate* (15). Prepared according to literature procedure^{7e} for the related diazo compounds; 524 mg, 46% total yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.45 (dt, *J* = 8.2, 3.7 Hz, 2H), 7.30 (dd, *J* = 4.7, 2.3 Hz, 3H), 4.72 (s, 1H), 1.80 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 131.8, 128.4, 128.1, 122.5, 90.1, 84.2, 73.5, 46.7, 29.4. IR (neat) 2987, 2111, 1701, 1370, 1186, 1124, 983, 693 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₃H₁₂N₂O₂Na [M+Na]⁺ 251.0791, found: 251.0798.

2-*Methyl-4-phenylbut-3-yn-2-yl* 3-(*tert-Butyldimethylsilyl*)*oxy-2-diazo-but-3-enoate* (16). Synthesized according to the published procedure¹⁹ from compound 14; 1.67 g, 88% yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.48 – 7.43 (comp, 2H), 7.34 – 7.28 (comp, 3H), 5.03 (d, *J* = 2.1 Hz, 1H), 4.26 (d, *J* = 2.1 Hz, 1H), 1.81 (s, 6H), 0.93 (s, 9H), 0.24 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 162.5, 140.9, 131.8, 128.4, 128.1, 122.5, 90.4, 89.9, 84.4, 73.6, 29.4, 25.6, 18.1, -4.8. IR (neat) 2956, 2088, 1715, 1342, 1255, 1057, 842 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₂₈N₂O₃SiNa [M+Na]⁺ 407.1761, found: 407.1745.

General Procedure for Gold(I)-Catalyzed Domino Cascade Transformation of Phenylpropargyl Phenyldiazoacetates. To a flame-dried 10 mL Schlenk flask charged with a magnetic stirring bar cationic gold(I) catalyst (0.010 mmol) and 2.0 mL DCM were added under nitrogen atmosphere. 1-(Phenylethynyl)cyclopentyl 2-diazo-2-phenylacetate 1 (0.20 mmol) dissolved in 2.0 mL of DCM was added in one portion into the solution under the flow of nitrogen. The resulting mixture was stirred for 12 h at room temperature (20 °C); the reaction mixture was purified by column chromatography (100:1 to 10:1 gradient of hexanes: ethyl acetate as eluents) to afford pure *8'-Phenylspiro[cyclopentane-1,1'-indeno[1,2-c]furan]-3'(8'H)-one* (2) as a white solid (52 mg, 86% using catalyst [Au(JohnPhos)(MeCN)]SbF₆); m.p. 128 – 129 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.74 (d, *J* = 8.0 Hz, 1H), 7.40 – 7.36 (m, 1H), 7.33 – 7.30 (comp, 3H), 7.27 – 7.24 (comp, 2H), 7.08 – 7.06 (comp, 2H), 4.80 (s, 1H), 2.11 – 2.08 (comp, 2H), 2.07 – 2.02, (m, 1H), 2.01 – 1.98 (m, 1H), 1.98 – 1.82 (m, 1H), 1.81 – 1.69 (m, 1H), 1.67 – 1.55 (m, 1H), 1.33 – 1.25 (m, 1H); ¹³C NMR (400 MHz, CDCl₃) δ 179.0, 166.4, 152.0, 136.1, 135.9, 134.3, 129.1, 128.1, 127.9, 127.7, 125.0, 120.9, 95.5, 52.4, 38.5, 36.7, 24.4. IR (neat) 2961, 1751, 1453, 1090, 953, 771 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₈O₂Na [M+Na]⁺ 325.1199, found: 325.1200.

8'-Phenylspiro[cyclobutane-1,1'-indeno[1,2-c]furan]-3'(8'H)-one (13). Obtained using 5 mol % of Au(C₄H₈S)Cl at 80 °C in DCE for 12 h: 40.3 mg, 70% yield. White solid, m.p. 150–151 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.72 (d, J= 8.2 Hz, 1H), 7.38 – 7.32 (comp, 4H), 7.27 – 7.21 (comp, 2H), 7.15 – 7.13 (comp, 2H), 4.89 (s, 1H), 2.82 – 2.74 (m, 1H), 2.51 – 2.47 (comp, 2H), 1.84 – 1.73 (comp, 2H), 1.60 – 1.50 (m, 1H); ¹³C NMR (400 MHz, CDCl₃) δ 178.2, 166.2, 152.2, 135.8, 135.5, 133.8, 129.0, 128.1, 127.9, 127.0, 125.0, 120.9, 85.9, 52.4, 33.3, 31.7, 12.2. IR (neat) 1756, 1454, 1154, 1066, 991, 722, 700 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₀H₁₆O₂Na [M+Na]⁺ 311.1043, found: 311.1050.

General Procedure for the Synthesis of 1,5-Dihydro-4*H*-pyrazol-4-ones 10 and 11. To a flame-dried 10 mL Schlenk flask charged with a magnetic stirring bar, $AuCl(C_4H_8S)$ catalyst (0.010 mmol) and 2.0 mL DCE were added under nitrogen atmosphere. Propargyl aryldiazoacetate 1 or 9 (0.20 mmol) dissolved in 2.0 mL of DCE was added in one portion into the solution under the flow of nitrogen. The resulting mixture was stirred for 72 h at room temperature (20 °C). Solvent was evaporated, and products were purified by column chromatography (100:1 to 10:1 gradient of hexanes: ethyl acetate as eluents) to afford pure compounds 10 and 11.

5-Benzoyl-5-(cyclopent-1-en-1-yl)-3-phenyl-1,5-dihydro-4H-pyrazol-4-one (3 or 10a). 37.6 mg, 57% yield. Yellow solid, m.p. 115–116 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.18 (dd, J = 7.9, 1.2 Hz, 2H), 8.08 (dd, J = 7.9, 1.2 Hz, 2H), 7.90 (br, 1H), 7.62 – 7.57 (m, 1H), 7.47 (t, J = 7.7 Hz, 2H), 7.42 – 7.36 (comp, 3H), 5.88 – 5.84 (m, 1H), 2.50 – 2.41 (comp, 3H), 2.35 – 2.29 (m, 1H), 1.98 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.1, 189.7, 143.8, 138.5, 134.0, 133.4, 131.6, 130.7, 129.2, 129.1, 128.5, 128.4, 126.0, 81.5, 32.4, 32.4, 23.0. IR (neat) 3358, 1721, 1674, 1246, 903 cm⁻¹; HRMS (ESI) *m*/*z* calculated for C₂₁H₁₈N₂O₂Na [M+Na]⁺ 353.1260, found: 353.1247.

5-Benzoyl-3-(cyclopent-1-en-1-yl)-5-phenyl-1,5-dihydro-4H-pyrazol-4-one (4 or 11a). 12.5 mg, 19% yield. Yellow solid, m.p. 130–131 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.84 (d, J = 7.5 Hz, 2H), 7.66 (br, 1H), 7.50 (t, J = 7.5 Hz, 1H), 7.42 – 7.33 (comp, 7H), 6.84 – 6.80 (m, 1H), 2.75 – 2.70 (comp, 2H), 2.56 – 2.49 (comp, 2H), 1.96 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.7, 189.5, 143.2, 135.7, 134.1, 133.7, 133.1, 132.0, 130.9, 129.5, 128.8, 128.5, 126.5, 81.8, 34.1, 32.7, 22.2. IR (neat) 3343, 1732, 1675, 1447, 1231, 738 cm⁻¹; HRMS (ESI) m/z calculated for C₂₁H₁₈N₂O₂Na [M+Na]⁺ 353.1260, found: 353.1249.

5-Benzoyl-5-(cyclopent-1-en-1-yl)-3-(4-methoxyphenyl)-1,5-dihydro-4H-pyrazol-4-one (10b). 36.3 mg, 54% yield. Yellow solid, m.p. 44–46 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.18 (d, J = 7.3 Hz, 2H), 8.03 (d, J = 9.0 Hz, 2H), 7.73 (s, 1H), 7.62 – 7.56 (m, 1H), 7.50 – 7.44 (comp, 2H), 6.92 (d, J = 9.0 Hz, 2H), 5.87 – 5.82 (m, 1H), 3.84 (s, 3H), 2.52 – 2.39 (comp, 3H), 2.31 (dddd, J = 17.5, 9.3, 5.8, 2.1 Hz, 1H), 2.02 – 1.84 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.5, 189.9, 160.5, 144.2, 138.7, 133.9, 133.5, 131.4, 130.7, 128.3, 127.6, 121.7, 114.0, 81.3, 55.3, 32.5, 32.4, 23.0. IR (neat) 3338, 2931, 1712, 1674, 1608, 1510, 1251, 1176, 835 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₂H₁₉N₂O₃ [M–H]⁻ 359.1401, found: 359.1396.

5-Benzoyl-3-(cyclopent-1-en-1-yl)-5-(4-methoxyphenyl)-1,5-dihydro-4H-pyrazol-4-one (11b). 14.1 mg, 21% yield. Yellow solid, m.p. 60.5–61.5 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.86 (d, J = 8.5 Hz, 2H), 7.73 (s, 1H), 7.53 – 7.47 (m, 1H), 7.36 (dd, J = 11.7, 4.0 Hz, 2H), 7.26 (d, J = 7.8 Hz, 2H), 6.92 (d, J = 7.8 Hz, 2H), 6.82 – 6.81 (m, 1H), 3.80 (s, 3H), 2.76 – 2.68 (m, 2H), 2.52 (d, J = 5.6 Hz, 2H), 1.96 – 1.88 (m, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 193.1, 189.7, 159.8, 143.2, 133.9, 133.6, 133.1, 132.1, 131.0, 128.4, 127.9, 127.8, 114.9, 81.6, 55.3, 34.1, 32.7, 22.2. IR (neat) 3344, 2954, 1736, 1673, 1608, 1510, 1252, 692 cm⁻¹; HRMS (EI) *m/z* calculated for $C_{22}H_{19}N_2O_3$ [M–H]⁻ 359.1401, found: 359.1406.

5-Benzoyl-5-(cyclopent-1-en-1-yl)-3-(p-tolyl)-1,5-dihydro-4H-pyrazol-4-one (**10c**). 42.0 mg, 61% yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 8.19 – 8.14 (comp, 2H), 7.97 – 7.92 (comp, 2H), 7.78 (s, 1H), 7.60 – 7.56 (m, 1H), 7.47 – 7.43 (comp, 2H), 7.21 – 7.18 (comp, 2H), 5.85 – 5.82 (m, 1H), 2.46 – 2.38 (comp, 3H), 2.36 (s, 3H), 2.31 – 2.26 (m, 1H), 1.96 – 1.88 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.3, 189.8, 144.2, 139.3, 138.6, 134.0, 133.5, 131.5, 130.7, 129.2, 128.4, 128.3, 126.0, 81.4, 32.5, 32.4, 23.0, 21.4. IR (neat) 3344, 1722, 1676, 1231, 951 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₂H₂₀N₂O₂Na [M+Na]⁺ 367.1417, found: 367.1404.

5-Benzoyl-3-(cyclopent-1-en-1-yl)-5-(p-tolyl)-1,5-dihydro-4H-pyrazol-4-one (11c). 16.5 mg, 24% yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.88 – 7.80 (comp, 2H), 7.67 (s, 1H), 7.51 – 7.47 (m, 1H), 7.37 – 7.32 (comp, 2H), 7.24 – 7.16 (comp, 4H), 6.83 – 6.78 (m, 1H), 2.74 – 2.68 (comp, 2H), 2.55 – 2.48 (comp, 2H), 2.34 (s, 3H), 1.96 – 1.88 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.9, 189.6, 143.2, 138.8, 133.9, 133.7, 133.1, 132.9, 132.0, 131.0, 130.2, 128.4, 126.4, 81.8, 34.1, 32.7, 22.2, 21.1. IR (neat) 3348, 1737, 1675, 1230, 809 cm⁻¹; HRMS (ESI) *m/z* calculated for $C_{22}H_{20}N_2O_2Na$ [M+Na]⁺ 367.1417, found: 367.1407.

3-([1,1'-Biphenyl]-4-yl)-5-benzoyl-5-(cyclopent-1-en-1-yl)-1,5-dihydro-4H-pyrazol-4-one (10d). 56.0 mg, 69% yield. Yellow solid, m.p. 49–51 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.23 – 8.15 (comp, 2H), 7.98 (s, 1H), 7.68 – 7.58 (comp, 6H), 7.47 (dt, J = 12.5, 7.9 Hz, 5H), 7.37 (t, J = 7.4 Hz, 1H), 5.89 (m, 1H), 2.59 – 2.40 (comp, 3H), 2.35 (ddd, J = 15.2, 11.0, 4.0 Hz, 1H), 2.04 – 1.86 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.1, 189.8, 143.5, 141.8, 140.5, 138.5, 134.0, 133.5, 131.7, 130.7, 128.8, 128.4, 128.1, 127.6, 127.2, 127.0, 126.4, 81.6, 32.5, 23.0. IR (neat) 3346, 2957, 1725, 1674, 1233, 907, 733 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₇H₂₁N₂O₂ [M–H]⁻ 405.1609, found: 405.1610.

5 - ([1, 1'-Biphenyl] - 4-yl) - 5-benzoyl - 3-(cyclopent - 1-en - 1-yl) - 1, 5-dihydro - 4H-pyrazol - 4-one (11d). 7.3 mg, 9% yield. Yellow solid, m.p. 80–81 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.92 – 7.88 (comp, 2H), 7.78 (s, 1H), 7.67 – 7.61 (comp, 3H), 7.60 – 7.55 (comp, 2H), 7.52 (dd, J = 11.1, 3.7 Hz, 1H), 7.48 – 7.42 (comp, 3H), 7.38 (dd, J = 11.1, 3.7 Hz, 1H), 7.48 – 7.42 (comp, 3H), 7.48 – 7.42 (comp, 3H)

The Journal of Organic Chemistry

15.9, 7.7 Hz, 3H), 6.88 – 6.83 (m, 1H), 2.79 – 2.72 (comp, 2H), 2.58 – 2.51 (comp, 2H), 1.98 – 1.90 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.8, 189.5, 143.2, 141.7, 140.0, 134.6, 134.2, 133.8, 133.1, 132.0, 131.0, 128.9, 128.6, 128.2, 127.7, 127.1, 126.9, 81.6, 34.1, 32.7, 22.2. IR (neat) 3342, 2953, 1728, 1674, 1487, 1229, 1008, 735, 696 cm⁻¹; HRMS (EI) *m/z* calculated for $C_{27}H_{21}N_2O_2$ [M–H]⁻ 405.1609, found: 405.1604.

5-Benzoyl-3-(4-bromophenyl)-5-(cyclopent-1-en-1-yl)-1,5-dihydro-4H-pyrazol-4-one (10e). 60.4 mg, 74% yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 8.19 – 8.14 (comp, 2H), 7.97 – 7.92 (comp, 3H), 7.61 – 7.57 (m, 1H), 7.52 – 7.49 (comp, 2H), 7.47 – 7.43 (comp, 2H), 5.86 – 5.83 (m, 1H), 2.47 – 2.39 (comp, 3H), 2.32 – 2.26 (m, 1H), 1.96 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 191.7, 189.5, 142.7, 138.3, 134.1, 133.3, 131.8, 131.7, 130.8, 128.4, 128.1, 127.4, 123.4, 81.7, 32.5, 32.4, 22.9. IR (neat) 3336, 1724, 1673, 1231, 1009 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₇N₂O₂NaBr [M+Na]⁺ 431.0366, found: 431.0350.

5-Benzoyl-5-(4-bromophenyl)-3-(cyclopent-1-en-1-yl)-1,5-dihydro-4H-pyrazol-4-one (11e). 10.6 mg, 13% yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.83 – 7.79 (comp, 2H), 7.62 (s, 1H), 7.54 – 7.50 (comp, 3H), 7.39 – 7.35 (comp, 2H), 7.26 – 7.23 (comp, 2H), 6.83 – 6.80 (m, 1H), 2.74 – 2.68 (comp, 2H), 2.54 – 2.50 (comp, 2H), 1.95 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.4, 189.1, 143.3, 134.7, 134.6, 133.9, 132.9, 132.6, 131.9, 130.8, 128.7, 128.2, 123.2, 81.0, 77.3, 77.0, 76.8, 34.1, 32.7, 22.2. IR (neat) 3338, 1719, 1674, 1230, 908 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₇N₂O₂NaBr [M+Na]⁺ 431.0366, found: 431.0345.

5-Benzoyl-3-(4-chlorophenyl)-5-(cyclopent-1-en-1-yl)-1,5-dihydro-4H-pyrazol-4-one (10f). 50.2 mg, 69% yield. Yellow solid, m.p. 77.5–78.5 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.17 (d, *J* = 8.5 Hz, 2H), 8.02 (d, *J* = 8.5 Hz, 2H), 7.91 (s, 1H), 7.62 – 7.57 (m, 1H), 7.49 – 7.45 (comp, 2H), 7.38 – 7.34 (comp, 2H), 5.87 – 5.83 (m, 1H), 2.49 – 2.39 (comp, 3H), 2.33 – 2.26 (m, 1H), 1.97 – 1.88 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 191.8, 189.5, 142.7, 138.3, 135.0, 134.1, 133.3, 131.8, 130.7, 128.7, 128.4, 127.6, 127.2, 81.7, 32.5, 32.4, 22.9. IR (neat) 3345, 1716, 1673, 1231, 1116 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₇N₂O₂NaCl [M+Na]⁺ 387.0871, found: 387.0857.

5-Benzoyl-5-(4-chlorophenyl)-3-(cyclopent-1-en-1-yl)-1,5-dihydro-4H-pyrazol-4-one (11f). 10.2 mg, 14% yield. Yellow solid, m.p. 94–95 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.84 – 7.80 (comp, 2H), 7.67 (s, 1H), 7.55 – 7.51 (m, 1H), 7.40 – 7.36 (comp, 4H), 7.34 – 7.30 (comp, 2H), 6.85 – 6.81 (m, 1H), 2.75 – 2.69 (comp, 2H), 2.56 – 2.51 (copm, 2H), 1.96 – 1.90 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.5, 189.1, 143.3, 135.0, 134.5, 134.1, 133.9, 132.9, 131.9, 130.8, 129.7, 128.6, 127.9, 80.9, 34.1, 32.7, 22.2. IR (neat) 3344, 1717, 1671, 1228, 933 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₇N₂O₂NaCl [M+Na]⁺ 387.0871, found: 387.0858.

5-Benzoyl-5-(cyclopent-1-en-1-yl)-3-[4-(trifluoromethyl)phenyl]-1,5-dihydro-4H-pyrazol-4-one (10g). 40.6 mg, 51% yield. Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 8.22 – 8.16 (comp, 4H), 8.07 (br, 1H), 7.63 (d, *J* = 8.5 Hz, 2H), 7.60 – 7.56 (m, 1H), 7.50 – 7.47 (comp, 2H), 5.88 – 5.86 (m, 1H), 2.47 – 2.40 (comp, 3H), 2.33 – 2.27 (m, 1H), 1.98 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 191.4, 189.3, 141.9, 138.1, 134.2, 133.6, 133.3, 132.8, 132.6, 131.9(q, *J* = 30.0 Hz), 129.3, 128.4, 127.9, 125.4 (q, *J* = 3.8 Hz) 124.5 (q, *J* = 278.8 Hz), 81.9, 32.4, 32.4, 23.0. IR (neat) 3334, 1709, 1673, 1321, 1120 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₂H₁₇N₂O₂F₃Na [M+Na]⁺ 421.1134, found: 421.1116.

5-Benzoyl-3-(cyclopent-1-en-1-yl)-5-[4-(trifluoromethyl)phenyl]-1,5-dihydro-4H-pyrazol-4-one (11g). 17.5 mg, 22% yield. Yellow solid, m.p. 45.5–46.5 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.82 – 7.76 (comp, 2H), 7.68 –

7.63 (comp, 3H), 7.55 – 7.50 (comp, 3H), 7.38 (t, J = 7.8 Hz, 2H), 6.85 – 6.83 (m, 1H), 2.75 – 2.69 (comp, 2H), 2.56 – 2.51 (comp, 2H), 1.97 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.1, 188.9, 143.3, 139.3, 134.8, 134.1, 132.9, 131.8, 131.0 (q, J = 32.5 Hz) 130.6 128.8, 126.9, 126.4 (q, J = 3.8 Hz), 121.5 (q, J = 270 Hz) 80.8, 34.1, 32.7, 22.2. IR (neat) 3335, 1729, 1675, 1119, 907 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₂H₁₇N₂O₂F₃Na [M+Na]⁺ 421.1134, found: 421.1119.

 5-Benzoyl-5-(cyclopent-1-en-1-yl)-3-(thiophen-2-yl)-1,5-dihydro-4H-pyrazol-4-one (10h). 39.7 mg, 59% yield.²⁰ Yellow oil. ¹H NMR (500 MHz, CDCl₃) δ 8.18 (d, J = 7.3 Hz, 2H), 7.82 (s, 1H), 7.80 (dd, J = 3.7, 0.8 Hz, 1H), 7.60 (t, J = 7.4 Hz, 1H), 7.47 (t, J = 7.8 Hz, 2H), 7.35 (d, J = 5.1 Hz, 1H), 7.08 (dd, J = 4.9, 3.8 Hz, 1H), 5.88 – 5.84 (m, 1H), 2.48 – 2.37 (comp, 3H), 2.37 – 2.26 (m, 1H), 1.91 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 190.7, 189.5, 141.6, 138.3, 134.1, 130.7, 128.4, 127.7, 126.7, 81.2, 32.4, 23.0. IR (neat) 3337, 2926, 1722, 1674, 1228, 731, 692 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₉H₁₆N₂O₂SNa [M+Na]⁺ 359.0825, found: 359.0823.

5-Benzoyl-5-(cyclohex-1-en-1-yl)-3-phenyl-1,5-dihydro-4H-pyrazol-4-one (10i). 39.2 mg, 57% yield. Yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 8.18 (dd, J = 5.3, 3.3 Hz, 2H), 8.13 – 7.95 (m, 2H), 7.84 (s, 1H), 7.59 (ddd, J = 6.7, 4.0, 1.3 Hz, 1H), 7.52 – 7.43 (comp, 2H), 7.42 – 7.32 (comp, 3H), 5.84 (t, J = 2.7 Hz, 1H), 2.11 (comp, 2H), 1.63 (comp, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 192.3, 190.4, 143.7, 134.1, 133.9, 133.5, 130.8, 129.2, 129.1, 128.5, 128.5, 128.3, 128.2, 127.7, 126.0, 84.9, 25.9, 25.5, 22.6, 21.6. IR (neat) 3343, 2927, 1729, 1674, 1447, 1231, 695 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₂H₁₉N₂O₂ [M–H]⁻ 343.1452, found: 343.1449.

5-Benzoyl-3-(cyclohex-1-en-1-yl)-5-phenyl-1,5-dihydro-4H-pyrazol-4-one (11i). 11.0 mg, 16% yield. Yellow solid, m.p. 63–64 °C. ¹H NMR (300 MHz, CDCl₃) δ 7.83 (dd, J = 8.4, 1.2 Hz, 2H), 7.56 (br, 1H), 7.49 (dd, J = 10.5, 4.4 Hz, 1H), 7.41 – 7.31 (comp, 7H), 7.02 (m, 1H), 2.40 (comp, 2H), 2.18 (comp, 2H), 1.76 – 1.55 (comp, 4H). ¹³C NMR (126 MHz, CDCl₃) δ 193.1, 189.6, 145.0, 135.9, 133.7, 133.1, 131.5, 131.0, 129.5, 128.7, 128.5, 127.7, 126.5, 82.1, 25.7, 25.2, 22.3, 21.9. IR (neat) 3339, 2927, 1732, 1675, 1447, 1231, 698 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₂H₁₉N₂O₂ [M–H]⁻ 343.1452, found: 343.1448.

5-Benzoyl-5-(cyclododec-1-en-1-yl)-3-phenyl-1,5-dihydro-4H-pyrazol-4-one (**10***j*). 43.7 mg, 51% yield. Yellow solid, m.p. 69–70 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.15 – 8.05 (comp, 2H), 7.89 (s, 1H), 7.64 – 7.31 (comp, 8H), 5.48 – 5.42 (m, 1H), 2.69 – 2.44 (comp, 2H), 2.32 – 2.18 (m, 1H), 2.09 – 1.96 (m, 1H), 1.55 – 1.19 (comp, 16H). ¹³C NMR (126 MHz, CDCl₃) δ 192.2, 190.3, 143.5, 143.5, 135.7, 135.0, 133.7, 131.6, 131.4, 129.0, 128.5, 128.2, 127.8, 127.1, 125.9, 85.7, 26.4, 25.53, 25.48, 25.09, 25.05, 24.74, 24.72, 24.1, 22.6, 22.4. IR (neat) 3351, 2928, 1699, 1677, 1448, 1344, 693 cm⁻¹; HRMS (ESI) *m*/*z* calculated for C₂₈H₃₂N₂O₂Na [M+Na]⁺ 451.2356, found: 451.2347.

5-Benzoyl-3-(cyclododec-1-en-1-yl)-5-phenyl-1,5-dihydro-4H-pyrazol-4-one (11j). 30.0 mg, 35% yield. Yellow solid, m.p. 71–72 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.85 (dt, *J* = 8.5, 1.5 Hz, 2H), 7.66 (s, 1H), 7.52 – 7.47 (m, 1H), 7.44 – 7.31 (comp, 7H), 6.79 (t, *J* = 8.1 Hz, 1H), 2.57 (t, *J* = 6.8 Hz, 2H), 2.30 – 2.17 (comp, 2H), 1.67 – 1.59 (comp, 2H), 1.54 (dt, *J* = 9.6, 7.2 Hz, 2H), 1.49 – 1.41 (comp, 4H), 1.40 – 1.31 (comp, 6H), 1.30 – 1.23 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 193.4, 189.6, 145.0, 136.0, 135.4, 133.6, 133.2, 130.9, 129.9, 129.5, 128.7, 128.5, 126.5, 81.9, 26.6, 26.1, 25.6, 25.2, 25.1, 24.9, 24.6, 23.7, 22.9, 22.2. IR (neat) 3347, 2924, 2850,

The Journal of Organic Chemistry

1729, 1672, 1447, 1229, 737, 696 cm⁻¹; HRMS (ESI) m/z calculated for C₂₈H₃₂N₂O₂Na [M+Na]⁺ 451.2356, found: 451.2349.

5-Benzoyl-5-(3,6-dihydro-2H-pyran-4-yl)-3-phenyl-1,5-dihydro-4H-pyrazol-4-one (10k). 24.9 mg, 36% yield. Yellow solid, m.p. 67–68 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.23 – 8.12 (comp, 2H), 8.10 – 8.03 (m, 1H), 7.84 (dd, J = 6.3, 3.1 Hz, 1H), 7.56 – 7.30 (comp, 5H), 7.27 (comp, 2H), 5.91 – 5.88 (m, 1H), 4.22 (dd, J = 5.3, 2.6 Hz, 2H), 3.89 (ddt, J = 16.8, 11.9, 6.2 Hz, 1H), 3.78 – 3.64 (m, 1H), 2.70 – 2.41 (m, 1H), 2.37 – 2.05 (m, 1H). ¹³C NMR (126 MHz, CDCl₃) δ 191.8, 189.6, 144.1, 134.2, 131.7, 131.6, 131.0, 130.7, 129.4, 128.6, 128.5, 126.7, 126.0, 83.5, 65.3, 64.0, 25.8. IR (neat) 3332, 2925, 2085, 1710, 1677, 1447, 1232, 700 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₁H₁₇N₂O₃ [M–H]⁻ 345.1245, found: 345.1243.

5-Benzoyl-3-(3,6-dihydro-2H-pyran-4-yl)-5-phenyl-1,5-dihydro-4H-pyrazol-4-one (11k). 6.9 mg, 10% yield. Yellow solid, m.p. 70–71 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.84 (d, J = 8.3 Hz, 2H), 7.73 (s, 1H), 7.52 (td, J = 7.6, 1.0 Hz, 1H), 7.44 – 7.33 (comp, 7H), 6.99 (m, 1H), 4.29 (comp, 2H), 3.89 (dd, J = 10.5, 5.2 Hz, 2H), 2.54 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.6, 189.3, 143.2, 135.6, 133.8, 133.0, 130.9, 129.6, 128.9, 128.5, 128.5, 126.4, 125.1, 82.4, 65.4, 64.1, 25.1. IR (neat) 3322, 2923, 2085, 1708, 1675, 1447, 1232, 1131, 694 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₁H₁₇N₂O₃ [M–H]⁻ 345.1245, found: 345.1239.

5-(4-Chlorobenzoyl)-5-(cyclopent-1-en-1-yl)-3-phenyl-1,5-dihydro-4H-pyrazol-4-one (10l). 45.1 mg, 62% yield. Yellow solid, m.p. 85–86 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.21 – 8.15 (comp, 2H), 8.08 – 8.03 (comp, 2H), 7.92 (s, 1H), 7.46 – 7.43 (comp, 2H), 7.41 – 7.37 (comp, 3H), 5.85 – 5.82 (m, 1H), 2.47 – 2.40 (comp, 3H), 2.29 – 2.24 (m, 1H), 1.97 – 1.88 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.0, 188.7, 143.9, 140.7, 138.4, 132.4, 132.1, 131.5, 129.3, 129.0, 128.7, 128.5, 126.0, 81.6, 77.3, 77.0, 76.8, 32.5, 32.4, 22.9. IR (neat) 3347, 1724, 1674, 1114, 1010 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₁H₁₇N₂O₂NaCl [M+Na]⁺ 387.0871, found: 387.0855.

5-(4-Chlorobenzoyl)-3-(cyclopent-1-en-1-yl)-5-phenyl-1,5-dihydro-4H-pyrazol-4-one (111). 13.1 mg, 18% yield. Yellow solid, m.p. 92–93 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.84 – 7.80 (comp, 2H), 7.77 (s, 1H), 7.40 – 7.36 (comp, 3H), 7.33 – 7.30 (comp, 2H), 7.28 – 7.26 (comp, 2H), 6.82 – 6.79 (m, 1H), 2.75 – 2.71 (comp, 2H), 2.55 – 2.51 (comp, 2H), 1.95 – 1.89 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.7, 188.4, 143.3, 140.5, 135.7, 134.1, 132.7, 132.0, 131.0, 129.6, 129.0, 128.8, 126.7, 82.2, 77.3, 77.0, 76.8, 34.1, 32.7, 22.2. IR (neat) 3342, 1741, 1676, 1091, 907 cm⁻¹; HRMS (ESI) *m/z* calculated for $C_{21}H_{17}N_2O_2NaCl [M+Na]^+$ 387.0871, found: 387.0856.

5-(Cyclopent-1-en-1-yl)-5-(4-methoxybenzoyl)-3-phenyl-1,5-dihydro-4H-pyrazol-4-one (**10m**). 34.6 mg, 48% yield. Yellow solid, m.p. 53–54 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.29 – 8.18 (comp, 2H), 8.11 – 8.02 (comp, 2H), 7.98 (s, 1H), 7.43 – 7.33 (comp, 3H), 6.98 – 6.90 (comp, 2H), 5.88 – 5.79 (m, 1H), 3.89 (s, 3H), 2.52 – 2.37 (comp, 3H), 2.32 – 2.24 (m, 1H), 2.00 – 1.84 (comp, 2H). ¹³C NMR (126 MHz, cdcl₃) δ 192.6, 187.8, 164.2, 144.0, 139.0, 133.6, 131.4, 129.2, 129.1, 128.5, 126.1, 126.0, 113.6, 81.6, 55.5, 32.6, 32.4, 23.0. IR (neat) 3336, 2932, 2842, 1714, 1663, 1596, 1244, 1171, 1028, 842, 732 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₂H₁₉N₂O₃ [M–H]⁻ 359.1401, found: 359.1397.

3-(Cyclopent-1-en-1-yl)-5-(4-methoxybenzoyl)-5-phenyl-1,5-dihydro-4H-pyrazol-4-one (11m). 20.9 mg, 29% yield. Yellow solid, m.p. 72–73 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.94 – 7.84 (comp, 2H), 7.78 (s, 1H), 7.41 –

7.34 (comp, 3H), 7.33 – 7.29 (comp, 2H), 6.86 – 6.77 (comp, 3H), 3.81 (s, 3H), 2.77 – 2.69 (comp, 2H), 2.53 (ddd, J = 9.6, 4.8, 2.2 Hz, 2H), 1.93 (dt, J = 15.0, 7.4 Hz, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 193.4, 187.8, 163.9, 143.4, 136.4, 133.9, 133.8, 132.1, 129.4, 128.9, 128.7, 126.7, 125.5, 113.7, 82.2, 55.5, 34.1, 32.7, 22.2. IR (neat) 3310, 2933, 2841, 1736, 1667, 1599, 1510, 1247, 1172, 733 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₂H₁₉N₂O₃ [M–H]⁻ 359.1401, found: 359.1394.

 5-(Cyclopent-1-en-1-yl)-5-(4-methoxybenzoyl)-3-(4-methoxyphenyl)-1,5-dihydro-4H-pyrazol-4-one (10n). 42.9 mg, 55% yield. Yellow solid, m.p. 52–53 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.29 – 8.17 (comp, 2H), 8.07 – 7.98 (comp, 2H), 7.90 (s, 1H), 6.97 – 6.87 (comp, 4H), 5.86 – 5.78 (m, 1H), 3.87 (s, 3H), 3.82 (s, 3H), 2.51 – 2.36 (comp, 3H), 2.28 (dddd, *J* = 15.3, 9.3, 5.8, 2.1 Hz, 1H), 1.99 – 1.84 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 193.0, 187.9, 164.2, 160.4, 144.2, 139.2, 133.5, 131.1, 127.6, 126.2, 121.8, 113.9, 113.6, 81.3, 55.5, 55.3, 32.5, 32.4, 23.0. IR (neat) 3340, 2956, 2840, 1717, 1665, 1598, 1508, 1248, 1172, 1029, 837 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₃H₂₁N₂O₄ [M–H]⁻ 389.1507, found: 389.1501.

3-(Cyclopent-1-en-1-yl)-5-(4-methoxybenzoyl)-5-(4-methoxyphenyl)-1,5-dihydro-4H-pyrazol-4-one (**11n**). 18.7 mg, 24% yield. Yellow solid, m.p. 67–68 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.94 – 7.86 (comp, 2H), 7.83 (s, 1H), 7.24 – 7.17 (comp, 2H), 6.92 – 6.85 (comp, 2H), 6.86 – 6.77 (comp, 3H), 3.81 (s, 3H), 3.80 – 3.77 (s, 3H), 2.77 – 2.67 (comp, 2H), 2.56 – 2.46 (comp, 2H), 1.96 – 1.87 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 193.7, 188.0, 163.8, 159.7, 143.3, 133.9, 133.6, 132.2, 128.5, 128.1, 125.6, 114.8, 113.6, 82.0, 55.5, 55.3, 34.1, 32.7, 22.2. IR (neat) 3337, 2933, 2839, 1721, 1663, 1596, 1508, 1241, 1169, 1027, 839, 734 cm⁻¹; HRMS (EI) *m/z* calculated for $C_{23}H_{21}N_2O_4$ [M–H]⁻ 389.1507, found: 389.1497.

5-(Cyclopent-1-en-1-yl)-5-(4-methoxybenzoyl)-3-[4-(trifluoromethyl)phenyl]-1,5-dihydro-4H-pyrazol-4-one (100). 36.0 mg, 42% yield. Yellow solid, m.p. 35–36 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.23 (comp, 5H), 7.63 (d, J = 8.3 Hz, 2H), 6.99 – 6.91 (comp, 2H), 5.89 – 5.83 (m, 1H), 3.89 (s, 3H), 2.51 – 2.39 (comp, 3H), 2.33 – 2.24 (m, 1H), 2.00 – 1.87 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 191.9, 187.4, 164.4, 141.9, 138.6, 133.6, 131.7, 130.5 (q, J = 32.8 Hz), 125.9, 125.4 (q, J = 3.8 Hz), 113.6, 82.0, 55.5, 32.5, 32.4, 23.0. IR (neat) 3326, 2930, 2845, 1663, 1596, 1321, 1243, 1164, 1065, 842 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₃H₁₈F₃N₂O₃ [M–H]⁻ 427.1275, found: 427.1274.

3-(Cyclopent-1-en-1-yl)-5-(4-methoxybenzoyl)-5-[4-(trifluoromethyl)phenyl]-1,5-dihydro-4H-pyrazol-4-one (110). 4.3 mg, 5% yield. Yellow solid, m.p. 67 – 68 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.85 (dd, J = 9.3, 2.6 Hz, 2H), 7.71 (s, 1H), 7.66 (dd, J = 8.5, 2.5 Hz, 2H), 7.48 (dd, J = 8.5, 2.5 Hz, 2H), 6.92 – 6.76 (comp, 3H), 3.84 (s, 3H), 2.82 – 2.67 (comp, 2H), 2.61 – 2.42 (comp, 2H), 2.00 – 1.84 (comp, 2H). ¹³C NMR (126 MHz, CDCl₃) δ 192.8, 187.1, 164.2, 143.6, 140.1, 134.6, 133.6, 131.9, 130.8 (q, J = 34.0 Hz), 127.7, 127.1, 126.3 (q, J = 3.8 Hz), 125.2, 125.1, 114.0, 81.2, 55.6, 34.1, 32.7, 22.2. IR (neat) 3337, 2930, 1725, 1667, 1598, 1323, 1243, 1170, 1120, 1070, 841 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₃H₁₈F₃N₂O₃ [M–H]⁻ 427.1275, found: 427.1277.

5-Benzoyl-3-phenyl-5-(prop-1-en-2-yl)-1,5-dihydro-4H-pyrazol-4-one (**10p**). 36.5 mg, 60% yield.²⁰ Yellow solid, m.p. 39 – 40 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.19 (d, *J* = 7.4 Hz, 2H), 8.08 (d, *J* = 6.7 Hz, 2H), 7.93 (s, 1H), 7.61 (t, *J* = 7.4 Hz, 1H), 7.48 (t, *J* = 7.6 Hz, 2H), 7.43 – 7.35 (comp, 3H), 5.27 (s, 1H), 5.20 (s, 1H), 1.91 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 192.0, 189.9, 143.7, 140.3, 134.1, 133.4, 130.7, 129.3, 129.0, 128.5, 128.5,

The Journal of Organic Chemistry

126.0, 116.7, 84.1, 20.0. IR (neat) 3339, 1727, 1671, 1447, 1229, 692 cm⁻¹; HRMS (ESI) *m/z* calculated for $C_{19}H_{16}N_2O_2Na [M+Na]^+$ 327.1104, found: 327.1098.

5-Benzoyl-3-(4-methoxyphenyl)-5-(prop-1-en-2-yl)-1,5-dihydro-4H-pyrazol-4-one (**10q**). 48.1 mg, 72% yield.²⁰ Yellow solid, m.p. 44–45 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.18 (d, *J* = 7.4 Hz, 2H), 8.03 (d, *J* = 9.0 Hz, 2H), 7.74 (s, 1H), 7.60 (t, *J* = 7.4 Hz, 1H), 7.48 (t, *J* = 7.7 Hz, 2H), 6.92 (d, *J* = 9.0 Hz, 2H), 5.25 (s, 1H), 5.18 (s, 1H), 3.83 (s, 3H), 1.90 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 192.4, 190.0, 160.5, 144.1, 140.5, 134.1, 133.4, 130.7, 128.4, 127.6, 121.6, 116.5, 114.0, 83.9, 55.3, 20.0. IR (neat) 3342, 2934, 1730, 1674, 1251, 1231, 1173, 835, 690 cm⁻¹; HRMS (EI) *m/z* calculated for C₂₀H₁₇N₂O₃ [M–H]⁻ 333.1245, found: 333.1251.

Synthesis of 4-Hydroxypyrazoles. The experiments were carried out with both: (a) reaction mixtures containing 10d + 11d and (b) individual compounds 10d or 11d.

(a) Procedure for the synthesis of 6d from a reaction mixture containing 10d and 11d.

To the reaction mixture after the gold(I)-catalyzed reaction of **9d** that contained 1,5-dihydro-4*H*-pyrazol-4-ones **10d** and **11d** (156 mg, 0.384 mmol) of in DCE (8.0 mL) benzylamine (49.4 mg, 0.461 mmol) dissolved in DCE (0.5 mL) was added in one portion under a nitrogen atmosphere. The resulting mixture was stirred at 20 °C for 2 h, and the white precipitate of **6'** was formed. The precipitate of **6'** was dissolved by adding methanol (5.0 mL); the solution was filtered through Celite to remove gold particles. Solvents were evaporated, and the obtained mixture of products **6'** and *N*-benzylbenzamide was separated by column chromatography using hexanes/ethyl acetate (v/v 2:1) as the eluent to afford *3-([1,1'-Biphenyl]-4-yl)-5-(cyclopent-1-en-1-yl)-1H-pyrazol-4-ol* (**6')** (100 mg, 87% yield) as a white solid; m. p. > 300 °C (decomp.). ¹H NMR (300 MHz, DMSO-d6) δ 12.54 (br, 1H), 8.04 – 8.01 (comp, 3H), 7.69 (d, *J* = 7.3 Hz, 4H), 7.45 (t, *J* = 7.5 Hz, 2H), 7.34 (t, *J* = 7.2 Hz, 1H), 6.29 (s, 1H), 2.70 – 2.68 (comp, 2H), 2.49 – 2.47 (comp, 2H), 1.99 – 1.78 (comp, 2H). ¹³C NMR (126 MHz, DMSO-d6) δ 140.3, 138.7, 136.0, 129.4, 127.8, 127.1, 126.9, 126.3, 126.0, 33.4, 33.3, 22.8. IR (neat) 3246 (br), 1634, 1487, 1269, 1016, 844, 767 cm⁻¹; HRMS (ESI) *m/z* calculated for C₂₀H₁₉N₂O [M+H]⁺ 303.1492, found: 303.1496.

5-(*Cyclopent-1-en-1-yl*)-3-phenyl-1H-pyrazol-4-ol (6). 81.3 mg, 90% yield. White solid, m. p. 198 – 200 °C (decomp.). ¹H NMR (500 MHz, CD₃OD) δ 7.85 (dd, *J* = 8.1, 0.9 Hz, 2H), 7.45 – 7.42 (m, 2H), 7.35 – 7.32 (m, 1H), 6.48 (dt, *J* = 4.2, 2.1 Hz, 1H), 2.78 (ddd, *J* = 9.8, 4.4, 2.1 Hz, 2H), 2.57 (tdd, *J* = 7.5, 4.8, 2.4 Hz, 2H), 2.07 – 1.93 (comp, 2H). ¹³C NMR (126 MHz, CD₃OD) δ 136.7, 135.1, 134.3, 131.6, 130.4, 128.7, 128.2, 127.5, 126.1, 32.9, 32.6, 22.4. IR (neat) 3249 (br), 1635, 1482, 1274, 1026, 845 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₄H₁₅N₂O [M+H]⁺ 227.1179, found: 227.1181.

(b) Procedure for the synthesis of 6' from pure 10d.

To a solution of **10d** (81.0 mg, 0.200 mmol) in DCE (4.0 mL) benzylamine (25.7 mg, 0.240 mmol) dissolved in DCE (0.3 mL) was added in one portion under a nitrogen atmosphere. The resulting mixture was stirred at 20 °C for 2 h. Afterward, DCE was evaporated, and the obtained mixture of products **6'** and *N*-benzylbenzamide was separated by column chromatography using hexanes/ethyl acetate (v/v 2:1) as the eluent to afford compound **6'** (54.0 mg, 89% yield). The same procedure was used for the synthesis of **6'** from **11d**.

Gold-Catalyzed Reactions of Propargyl Diazoacetates 14-16.

(a) General Procedure for the Catalysis by AuCl(C₄H₈S). To a stirred solution of propargyl diazoacetate (0.30 mmol) in DCE (2.0 mL) AuCl(C₄H₈S) (4.8 mg, 0.015 mmol) dissolved in DCE (1.0 mL) was added rapidly under a nitrogen atmosphere, and the reaction mixture was stirred at room temperature for 12 h (16) or at 84 °C for 48 h (14 and 15). Solvent was evaporated and the reaction mixture was purified via flash-chromatography (using gradient hexanes to 9:1 hexanes/ethyl acetate as the eluent) to afford compounds 17 and 18. Reported in Scheme 8 are isolated yields.

*3-Methyl-1-phenylbut-2-en-1-one (17).*²¹ Pale yellow liquid. ¹H NMR (300 MHz, CDCl₃) δ 7.93 (d, *J* = 8.1 Hz, 2H), 7.58 – 7.37 (comp, 3H), 6.75 (s, 1H), 2.21 (s, 3H), 2.02 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 191.5, 156.6, 139.2, 132.3, 128.4, 128.2, 121.2, 28.0, 21.2.

(*3-Methylbut-3-en-1-yn-1-yl)benzene* (*18*).²² Colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ 7.52 – 7.42 (comp, 2H), 7.39 – 7.29 (comp, 3H), 5.43 (s, 1H), 5.33 (s, 1H), 2.02 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 131.7, 131.6, 128.3, 126.9, 123.3, 122.0, 90.6, 88.5, 23.5.

(b) General Procedure for the Catalysis by AuPicCl₂. To a stirred solution of propargyl diazoacetate 14 (81 mg, 0.30 mmol) in DCE (2.0 mL) AuPicCl₂ (5.9 mg, 0.015 mmol) dissolved in DCE (1.0 mL) was added rapidly under a nitrogen atmosphere, and the reaction mixture was stirred at 40 °C for 12 h. Solvent was evaporated and the reaction mixture was purified via flash-chromatography (using gradient 19:1 hexanes/ethyl acetate to 9:1 hexanes/ethyl acetate as the eluent) to afford 17 (6.7 mg, 14%) and 21 (66.4 mg, 82%).

(*Z*)-3-*Methyl-1-phenylbuta-1,3-dien-1-yl* 2-*Diazo-3-oxobutanoate* (21). Yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.49 – 7.40 (comp, 2H), 7.35 – 7.27 (comp, 3H), 2.49 (s, 3H), 1.83 (s, 6H). ¹³C NMR (126 MHz, CDCl₃) δ 190.3, 159.6, 131.8, 128.6, 128.2, 122.2, 89.2, 85.0, 74.7, 29.4, 28.3. IR (neat) 2925, 2142, 1726, 1659, 1246, 1139, 1058, 697 cm⁻¹; HRMS (ESI) *m/z* calculated for C₁₅H₁₄N₂O₃Na [M+Na]⁺ 293.0897, found: 293.0888.

ASSOCIATED CONTENT

The Supporting Information is available free of charge on the ACS Publications website at DOI: Experimental tables, kinetic experiments details, copies of ¹H and ¹³C NMR spectra of all compounds, crystallographic reports (PDF).

Crystallographic data for **4** (CIF) Crystallographic data for **6**' (CIF)

AUTHOR INFORMATION

Corresponding Author

*michael.doyle@utsa.edu

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

Support for this research from the National Science Foundation (CHE-1533833 and CHE-1464690), The Welch Foundation (AX-1871), and UTSA is gratefully acknowledged. HQ thanks the China Scholarship

Council for a fellowship. The HRMS used in this research is supported by a grant from the National Institutes of Health (G12MD007591). We thank W. G. Griffith for extensive mass spectral analyses.

REFERENCES

(1) (a) Hashmi, A. S. K.; Toste, F. D. (Eds.) Modern Gold Catalyzed Synthesis; Wiley-VCH: Weinheim, Germany, 2012. (b) Toste, F. D.; Michelet, V. (Eds.) Gold Catalysis: An Homogeneous Approach (Catalytic Science Series – Vol. 13); Imperial College Press, 2014.

(2) Recent Reviews: (a) Pflästerer, D.; Hashmi A. S. K. Chem. Soc. Rev. 2016, 45, 1331. (b) Dorel, R.;
Echavarren, A. M. Chem. Rev. 2015, 115, 9028. (c) Ranieri, B.; Escofet, I.; Echavarren, A. M. Org. Biomol. Chem. 2015, 13, 7103.

(3) Propargylic systems and gold: (a) Wang, S.; Zhang, G.; Zhang, L. *Synlett* **2010**, 692. (b) Garayalde, D.; Nevado, C. *ACS Catal.* **2012**, *2*, 1462.

(4) (a) Shiroodi, R. K.; Dudnik, A. S.; Gevorgyan, V. J. Am. Chem. Soc. 2012, 134, 6928. (b) Briones, J. F.;
Davies, H. M. L. J. Am. Chem. Soc. 2012, 134, 11916. (c) Shiroodi, R. K.; Gevorgyan, V. Chem. Soc. Rev. 2013, 42, 4991. (d) Fensterbank, L.; Malacria, M. Acc. Chem. Res. 2014, 47, 953. (e) Yu, Z.; Ma, B.; Chen, M.; Wu, H.-H.; Liu, L.; Zhang, J. J. Am. Chem. Soc. 2014, 136, 6904.

(5) Diazo compounds and gold catalysis: (a) Liu, L.; Zhang, J. *Chem. Soc. Rev.* 2016, 45, 506. (b) Fructos, M. R.; Díaz-Requejo, M. M.; Pérez, P. J. *Chem. Commun.* 2016, 52, 7326. (c) Wei, F.; Song, C.; Ma, Y.; Zhou, L.; Tung, C.-H.; Xu, Z. *Science Bull.* 2015, 60 (17), 1479. (d) Wang, J.; Yao,X.; Wang, T.; Han, J.; Zhang, J.; Wang, P.; Zhang, Z. *Org. Lett.* 2015, 17, 5124.

(6) Xu, X.; Trong, P.; Doyle, M. P. in *Science of Synthesis: Applications of Domino Transformations in Organic Synthesis 1*, Snyder, S. A., Ed., Georg Thieme Verlag KG, New York, 2016; Chapter 1.6.2, 511.

(7) (a) Padwa, A.; Chiacchio, U.; Garreau, Y.; Kassir, J. M.; Krumpe, K. E.; Schoffstall, A. M. J. Org. Chem.
1990, 55, 414. (b) Kinder, F. R.; Padwa A. Tetrahedron Lett. 1990, 31, 6835. (c) Padwa, A.; Dean, D. C.; Fairfax, D. J.; Xu, S. L. J. Org. Chem. 1993, 58, 4646. (d) Padwa, A.; Weingarten, M. D. J. Org. Chem. 2000, 65, 3722. (e) Jansone-Popova, S.; May, J. A. J. Am. Chem. Soc. 2012, 134, 17877. (f) Jansone-Popova, S.; Le, P. Q.; May, J. A. Tetrahedron 2014, 70, 4118. (g) Zheng, Y.; Mao, J.; Weng, Y.; Zhang, X.; Xu, X. Org. Lett. 2015, 17, 5638. (h) Yao, R.; Rong, G.; Yan, B.; Qiu, L.; Xu. X. ACS Catal. 2016, 6, 1024.

(8) Reviews on gold catalysis with allenes: (a) Krause, N.; Winter, C. *Chem. Rev.* 2011, *111*, 1994. (b)
Alcaide, B.; Almendros, P. *Acc. Chem. Res.* 2014, *47*, 939. (c) Yang, W.; Hashmi, A. S. K. *Chem. Soc. Rev.* 2014, *43*, 2941. (d) Soriano, E.; Fernandez, I. *Chem. Soc. Rev.* 2014, *43*, 3041.

(9) (a) Zhang, L. J. Am. Chem. Soc. 2005, 127, 16804. (b) Zhang, L.; Wang, S. J. Am. Chem. Soc. 2006, 128, 1442. (c) Wang, S.; Zhang, L. J. Am. Chem. Soc. 2006, 128, 8414. (d) Luo, T.; Schreiber, S. L. J. Am. Chem. Soc. 2009, 131, 5667. (e) Hashmi, A. S. K.; Wang, T.; Shi, S.; Rudolph, M. J. Org. Chem. 2012, 77, 7761.

(10) Qiu, H.; Srinivas, H. D.; Zavalij, P. Y.; Doyle, M. P. J. Am. Chem. Soc. 2016, 138, 1808.

(11) *o*-Acetylenyl-substituted phenyldiazoacetates have been reported to undergo a sequential reaction with water resulting in 1*H*-isochromene derivatives using (IPr)AuCl as the catalyst (80 °C, 24 h): Zhou, L.; Liu, Y.; Zhang, Y.; Wang, J. *Beilstein J. Org. Chem.* **2011**, *7*, 631.

(12) Manuleón, P.; Krinsky, J. L.; Toste, F. D. J. Am. Chem. Soc. 2009, 131, 4513.

(13) (a) Tejedor, D.; Mendez-Abt, G.; Cotos, L.; Garcia,-Tellado, F. Chem. Soc. Rev. 2013, 42, 458. (b) Xing,

Y.; Wei, Y.; Zhou, H. Curr. Org. Chem. 2012, 16, 1594.

(14) (a) Li, L.; Chen, J.-J.; Li, Y.-J.; Bu, X.-B.; Liu, Q.; Zhao, Y.-L. Angew. Chem. Int. Ed. 2015, 54, 12107.
(b) Kuznetsov, A. V.; Gulevich, A. V., Wink, D. J.; Gevorgyan V. Angew. Chem. Int. Ed. 2014, 53, 9021. (c) António, J. P. M.; Frade, R. F. M.; Santos, F. M. F.; Coelho, J. A. S.; Alfonso, C. A. M.; Gois, P. M. P.; Trindade, A. F. RSC Adv. 2014, 4, 29352. (d) Santos, F. M. F.; Rosa, J. N.; Andr, V.; Duarte, M. T.; Veiros, L. F.; Gois, P. M. P. Org. Lett. 2013, 15, 1760.

(15) The same outcome occurred when phenylpropargyl benzoate (PhCCCMe₂OOCPh) was treated with AuCl(Me₂S): the corresponding allene product from 1,3-acyl transfer was first formed, then slow rearrangement at 20 °C over 20 h formed only the (*Z*)-diene product [PhCOOC(Ph)=CHC(Me)=CH₂] quantitatively.

(16) (a) Doyle, M. P.; McKervey, M. A.; Ye, T. Modern Catalytic Methods for Organic Synthesis with Diazo Compounds: From Cyclopropanes to Ylides; Wiley-Interscience: New York, 1998. (b) Davies, H. M. L.; Parr, B. T. In: Contemporary Carbene Chemistry; Moss, R. A.; Doyle, M. P. Eds.; Wiley: Hoboken, N. Y., 2014; Chapter 15, 363. (c) Ford, A.; Miel, H.; Ring, A.; Slattery, C. N.; Maguire, A. R.; McKervey, M. A. Chem. Rev. 2015, 115, 9981.

(17) Li, W.; Liu, X.; Hao, X.; Hu, X.; Chu, Y.; Cao, W.; Qin, S.; Hu, C.; Lin, L.; Feng, X. J. Am. Chem. Soc. **2011**, *133*, 15268.

(18) Isolated yields: **3** (22%), **4** (28%), (Z)-diene **5** (18%), [4+2]-cycloaddition (9%).

(19) Deng, Y.; Jing, C.; Doyle, M. P. Chem. Commun. 2015, 51, 12924.

(20) Pure compound 11 was not isolated.

(21) Reported compound: Egi, M.; Yamaguchi, Y.; Fujiwara, N.; Akai, S. Org. Lett. 2008, 10, 1867.

(22) Reported compound: Hatakeyama, T.; Yoshimoto, Y.; Gabriel, T.; Nakamura, M. Org. Lett. 2008, 10, 5341.