

Functional Structure/Activity Relationships

Design and synthesis of novel 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one derivatives for use as herbicides and evaluation of their mode of action

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1 **Design and synthesis of novel 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one**
2 **derivatives for use as herbicides and evaluation of their mode of action**

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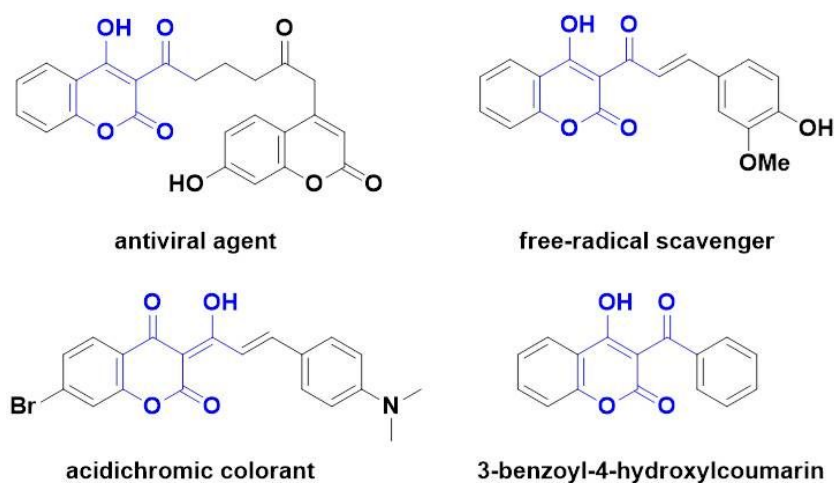
13 **ABSTRACT**

14 In order to develop a novel herbicide containing the β -triketone motif, a series of
15 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one derivatives were designed and
16 synthesized. Bioassay results showed that compound **III5** had good pre-emergent
17 herbicidal activity even at a dosage of 187.5 g ha⁻¹. Moreover, compound **III5**
18 showed a broader spectrum of weed control when compared with a commercial
19 herbicide 2,4-dichlorophenoxyacetic acid (2,4-D), and displayed good crop safety to
20 *Triticum aestivum L.* and *Zea mays Linn.* when applied at 375 g ha⁻¹ under
21 pre-emergence conditions, which indicated its great potential as a herbicide. More
22 importantly, studying the molecular mode of action of compound **III5** revealed that
23 the novel triketone structure is a proherbicide of its corresponding phenoxyacetic acid
24 auxin herbicide, which has a herbicidal mechanism similar to that of 2,4-D. The
25 present work indicates that the 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one motif
26 may be a potential lead structure for further development of novel auxin-type
27 herbicides.

28 **Keywords:** 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one, β -triketone, synthesis,
29 herbicidal activity, auxin herbicide

31 INTRODUCTION

32 β -Triketones are known as potential agrochemicals because they exhibit several
33 biological activities, including herbicidal ¹⁻⁶, antibacterial ⁷, antifungal ^{8, 9} and
34 insecticidal activity ¹⁰. Among the various β -triketones reported to date, pyran-based
35 diketone lactones have been extensively investigated, and many pyran-based diketone
36 lactones (**Fig. 1**) are reported owing to their effective role as antiviral agents ¹¹,
37 free-radical scavengers ¹², and acidichromic colorants ^{13, 14}. However, relatively little
38 is known about the application of pyran-based diketone lactones in pesticide
39 formulation ¹⁵. In our previous work, it was shown that
40 3-benzoyl-4-hydroxycoumarin derivatives (**Fig. 1**), which belong to the family of
41 pyran-based diketone lactones, display herbicidal activity and fungicidal activity ^{16, 17}.
42 Therefore, this encouraged us to continue to explore the biological activity of
43 pyran-based diketone lactones in pesticide.



44 **Figure 1.** Structures of various pyran-based diketone lactones.

46 Phenoxyacetic acid and its derivatives are key intermediates in the synthesis of
47 biologically active compounds. Over the last few decades, many

48 phenoxyacetyl-containing derivatives have been successfully synthesized, and these
49 derivatives display a wide range of biological activity in drug chemistry¹⁸⁻²².
50 Moreover, phenoxyacetic acid and its derivatives have also been reported to exhibit
51 herbicidal²³, insecticidal²⁴, and fungicidal activity²⁵. Based on the above-mentioned
52 facts, we envisaged that introducing phenoxyacetyl at the 3rd-position of
53 4-hydroxycoumarin and 4-hydroxy-6-methyl-2*H*-pyran-2-one can help construct a
54 variety of 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one derivatives (**Fig. 2**,
55 compounds **I** and **II**), which should possess some interesting biological activity.
56 Therefore, as a continuation of our work on synthesizing and evaluating the biological
57 activity of pyran-based diketone lactones in pesticide formulation, 21
58 3-(2-phenoxyacetyl)-4-hydroxycoumarin derivatives (**Fig. 2**, compound **I**) and 21
59 4-hydroxy-6-methyl-3-(2-phenoxyacetyl)-2*H*-pyran-2-one derivatives (**Fig. 2**,
60 compound **II**) were synthesized, and their herbicidal activities were evaluated to
61 develop a novel herbicide containing β -triketone motif. To the best of our knowledge,
62 this is the first report on the herbicidal activity of
63 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one derivatives.

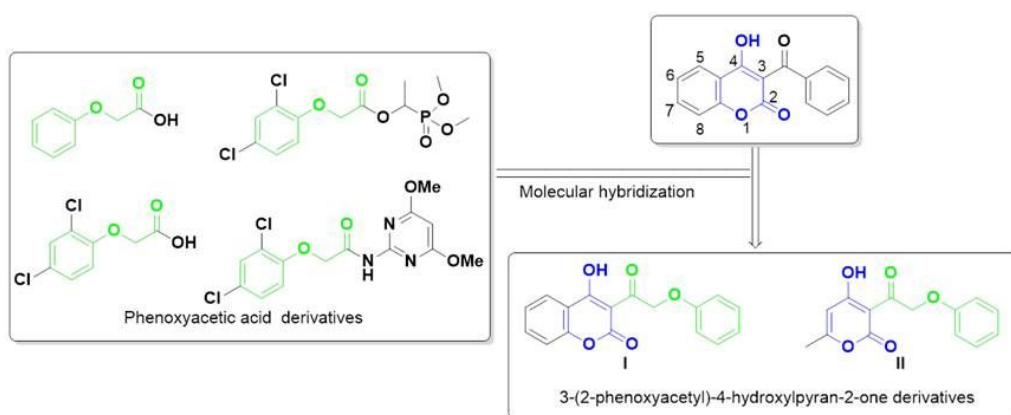
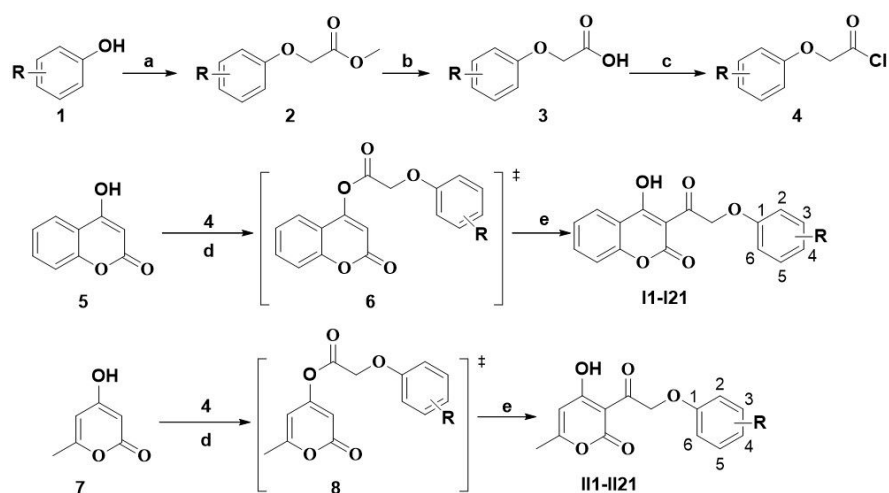


Figure 2. Design strategy used to prepare the target compounds.

66 **MATERIALS AND METHODS**67 **Chemical synthesis procedures**68 The synthetic pathway used to prepare the target compounds is outlined in **Scheme**69 **1**. The yields were not optimized.

Reagent and conditions: (a) methyl chloroacetate, K₂CO₃, DMF, 50 °C, 12 h; (b) LiOH, MeOH/H₂O, 50 °C, 12 h; (c) oxalyl chloride, DMF, DCM, 25 °C, 12 h; (d) Et₃N, 4-dimethylaminopyridine, DCM, 25 °C, 12 h; (e) KCN, 18-crown-6, Et₃N, DCM, 25 °C, 72 h.

I1/II1 R=H; I2/II2 R=2-Cl; I3/II3 R=3-Cl; I4/II4 R=4-Cl; I5/II5 R=2-Me; I6/II6 R=3-Me; I7/II7 R=4-Me
 I8/II8 R=2-OMe; I9/II9 R=3-OMe; I10/II10 R=4-OMe; I11/II11 R=2,3-2Cl; I12/II12 R=2,4-2Cl; I13/II13 R=2,5-2Cl
 I14/II14 R=3,5-2Cl; I15/II15 R=2-Cl-4-F; I16/II16 R=2-Cl-4-Br; I17/II17 R=2-Me-4-F; I18/II18 R=2-Me-4-Cl;
 I19/II19 R=2-Me-4-Br; I20/II20 R=2-Br-4-F; I21/II21 R=2-Br-4-Cl

70

71 **Scheme 1**. Synthetic route of preparing target compounds **I1-I21** and **II1-II21**.72 *General procedure for the synthesis of aryloxyacetyl chloride derivatives 4*

73 The key intermediate aryloxyacetyl chloride derivatives **4** were synthesized
 74 following a reported procedure⁵. The appropriate phenol **1** (20 mmol) and potassium
 75 carbonate (40 mmol) were added successively to *N,N*-dimethylformamide (100 mL)
 76 in a 250 mL flask with stirring, and the mixture was heated up to 50 °C for 1 h.
 77 Methyl chloroacetate (20 mmol) was added to the mixture and the suspension was
 78 stirred at 50 °C for 12 h. Then, the mixture was cooled to room temperature, poured
 79 into water, and stirred for another 0.5 h. To the solution was added 200 mL of ethyl
 80 acetate, the ethyl acetate layer was separated and concentrated by rotary evaporation.

81 Methanol (50 mL) and water (5 mL) were added to the resulting residue, and then
82 lithium hydroxide (100 mmol) was added to the solution with stirring. The solution
83 was heated up to 50 °C for 12 h. After completion of the reaction, 30 mL of methanol
84 was removed by rotary evaporation. The resulting solution was cooled to room
85 temperature and acidified by hydrochloric acid solution (6 M) to pH = 1. The
86 resulting solid was collected by filtration, washed with water, and dried in a vacuum
87 to afford aryloxyacetic acids **3**. Subsequently, aryloxyacetic acids **3** (10 mmol) were
88 dissolved in dichloromethane (20 mL) in a 50 mL flask, oxalyl chloride (20 mmol)
89 and *N,N*-dimethylformamide (one drop) were added. The mixture was stirred at room
90 temperature for 12 h. After completion of the reaction, dichloromethane was removed
91 by rotary evaporation to provide aryloxyacetyl chloride derivatives **4**, which used in
92 the next step without purification.

93 *General procedure for the synthesis of 3-(2-phenoxyacetyl)-4-hydroxypyran-2-one*
94 *derivatives (compound II-II21 and compound III-II21)*

95 3-(2-phenoxyacetyl)-4-hydroxypyran-2-one derivatives were prepared following a
96 reported method ^{16, 17, 26}. 4-Hydroxycoumarin **5** (5 mmol) and 20 mL of
97 dichloromethane were added to a 100 mL flask. Aryloxyacetyl chloride derivatives **4**
98 was added to the solution at 0 °C, the solution was stirred for another 30 min, then,
99 triethylamine (10 mmol) and 4-dimethylaminopyridine (0.5 mmol) were added
100 successively to the mixture, and the solution was stirred for 12 h. After completion of
101 the reaction, aqueous hydrochloric acid solution (1 M, 20 mL) was added. The
102 dichloromethane layer was separated and washed with H₂O, saturated sodium chloride

103 solution, dried by anhydrous sodium sulfate, concentrated by rotary evaporation, and
104 the residue was scratched from ethyl acetate and petroleum ether (1/1 by volume) to
105 give desired product **6**, which used in the next step without further purification.

106 A solution of compound **6** (2 mmol) in dichloromethane (20 mL) was added to
107 triethylamine (3 mmol), 18-crown-6 (0.2 mmol) and potassium cyanide (1 mmol).
108 The mixture was stirred at room temperature for 72 h. The mixture was poured into
109 water and extracted with dichloromethane. The combined organic phase was dried
110 with anhydrous sodium sulfate, filtered and removed by rotary evaporation. The
111 residue was scratched from ethyl acetate and ethanol (1/1 by volume) to give target
112 compounds **I1-I21**.

113 Compounds **III-II21** were synthesized by the similar procedure to compounds
114 **I1-I21**. The data of ^1H NMR, ^{13}C NMR, and HRMS of all target compounds is given
115 in the Supporting Information.

116 **X-ray diffraction of compounds I9 and III14**

117 Compound **I9** was recrystallized from a mixture of dichloromethane and methanol
118 (1/1 by volume) to afford a suitable single crystal. Compound **III14** was recrystallized
119 from dichloromethane to afford a suitable single crystal. Crystallographic data for
120 compounds **I9** and **III14** had been deposited with the Cambridge Crystallographic Data
121 Centre as supplementary publications with the deposition numbers 1834640 and
122 1874641, respectively. The detail data can be obtained free of charge from
123 <http://www.ccdc.cam.ac.uk/>.

124 **Herbicidal activity evaluation**

125 Based on the reported procedure²⁶⁻²⁹, herbicidal activity was evaluated with three
126 replicates per treatment. Mesotrione and 2,4-D were selected as positive control. The
127 preliminary herbicidal activity of target compounds was determined with *Brassica*
128 *campestris* root test and *Echinochloa crusgalli* cup test. Further herbicidal activity
129 study of target compounds against four species representative of monocotyledonous
130 and dicotyledonous plants (*Brassica campestris*, *Amaranthus retroflexus*,
131 *Echinochloa crusgalli*, *Digitaria sanguinalis*) was performed in the greenhouse.
132 Compound **III5** was selected to study the herbicidal spectra. The procedure is given
133 in the Supporting Information.

134 **Crop selectivity**

135 Based on the reported procedure⁵, the crop selectivity of compound **III5** was
136 evaluated with three replicates per treatment. Three representative crops, namely,
137 *Triticum aestivum* L., *Zea mays* Linn. and *Gossypium spp*, were selected for crop
138 selectivity studies in the greenhouse. The procedure is given in the Supporting
139 Information.

140 **Phenotypic study of *Arabidopsis thaliana***

141 *Arabidopsis thaliana* employed in this study is in the Columbia (Col-0) background.
142 Surface-sterilized seeds sown onto Murashige and Skoog (MS) plates at 4°C for 3
143 days under the darkness to favor vernalization and then grown at 22 °C for 2 days
144 under LD (16h light/8h darkness). Subsequently, plants were transferred to 1/2MS
145 plates supplemented with compound **III5** (from 1 μM to 1.0 nM), 2,4-D (from 1 μM
146 to 1.0 nM), 2-(2-chloro-4-fluorophenoxy) acetic acid (from 1 μM to 0.1 μM),

147 4-hydroxy-6-methyl-pyran-2-one (1 μ M) and mesotrione (1 μ M), respectively. The
148 solution of compound **III5**, 2,4-D, 2-(2-chloro-4-fluorophenoxy) acetic acid,
149 4-hydroxy-6-methyl-pyran-2-one and mesotrione was prepared in 0.1% MeOH and
150 diluted in the 1/2MS to obtain the tested concentrations, respectively. Solvent (0.1%
151 MeOH) was added also to control plates. Plants were grown on vertically oriented
152 plates at 22 °C for 7 days under LD (16h light/8h darkness) for phenotypic
153 investigation and root length measurement.

154 **Quantitative real-time-PCR**

155 Surface-sterilized seeds were sown onto MS plates. The plant was grown on plates
156 for two weeks from day of sowing under LD (16h light/8h darkness) at 22 °C.
157 Subsequently, the plant was carefully placed in a petri dish with filter paper covered.
158 The solution of compound **III5** (10 μ M), 2,4-D (10 μ M), and mesotrione (10 μ M) was
159 added to the petri dish, respectively. Control plants were treated with distilled water.
160 The leaves of seedlings were collected for RNA extraction after 0 h, 12 h, and 24 h,
161 respectively. For quantitative real-time PCR (qRT-PCR), total RNAs were extracted
162 from plant tissues using Trizol reagent (TaKaRa, Japan) according to the
163 manufacturer's instructions. Reverse transcription reactions were performed with the
164 PrimeScript RT reagent kit with gDNA Eraser (Takara, Japan). qRT-PCR reactions
165 were performed on real-time thermal cycling system (Bio-Rad, USA), and the
166 SYBR-Green Ex Taq II kit (TaKaRa, Japan) was used for detecting gene expression
167 abundances. The gene expression levels were normalized with the reference gene
168 *ACTIN-2*. Primers used for PCR are given in the Supporting Information (**Table S1**).

169 The degradation study of compound III15 in *Arabidopsis thaliana*

170 Surface-sterilized seeds were sown onto MS plates. Plants were grown on plates at
171 22 °C for 14 days under LD (16 h light/8 h dark). Subsequently, plants were carefully
172 placed in a petri dish with filter paper covered. The solution of compound **III15** (50
173 μM) and 2-(2-chloro-4-fluorophenoxy) acetic acid (50 μM) was added to the petri dish,
174 respectively. After 15 min of incubation, the treated plants were transferred to the MS
175 plates. The leaves of seedlings were collected for extraction after 0 h, 1 h, 1.5 h, 2 h,
176 and 2.5 h, respectively. Methanol was used as extraction fluid. The ingredients of
177 extraction were analyzed by HPLC on a Shimadzu HPLC system using a 5 μm C18
178 column (150 \times 4.6 mm; Agilent Zorbax) and 3-indolebutyric acid (IBA) was used as
179 the internal standard. A linear gradient with increasing methanol (solvent A) against
180 double-distilled H₂O (solvent B) at a flow rate of 1 mL min⁻¹ over 35 min was used.
181 Both solutions contained 0.1 % phosphoric acid. HPLC conditions were as follows: 0
182 min, 30 % A; 20 min, 70 % A; 22 min, 30 % A; 35 min, stop. The results are
183 presented at 230 nm.

184 Statistical analysis

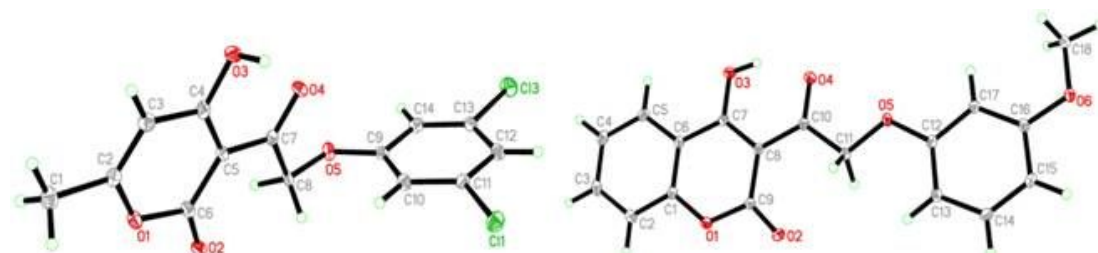
185 The values shown in each table are mean values \pm SD of at least three repeated
186 experiments. SPSS 22.0 (SPSS, Chicago, IL) was used as the statistical software
187 program.

188 RESULTS AND DISCUSSION**189 Chemistry**

190 As depicted in **Scheme 1**, the target compounds can be prepared via a five-step

191 synthetic route using several phenols as the starting material. In the presence of
192 potassium carbonate as base, phenols **1** reacted with methyl chloroacetate, and the
193 corresponding methyl aryloxyacetates **2** were produced. Hydrolysis of methyl
194 aryloxyacetates **2** with lithium hydroxide as a base gave aryloxyacetyl acids **3**.
195 Subsequently, aryloxyacetyl acids **3** were transformed into their corresponding
196 aryloxyacetyl chloride derivatives **4** using oxalyl chloride. Reacting compound **5** with
197 the aryloxyacetyl chloride derivatives **4** in the presence of 4-dimethylaminopyridine
198 and trimethylamine gave intermediate **6**, which was treated with potassium cyanide
199 and 18-crown-6 at room temperature to give target compounds **I1-I21** in 41– 61%
200 yield.

201 Compounds **I1-I21** were synthesized using a procedure similar to that used for
202 compounds **I1-I21**. The structures of all the target compounds were identified using
203 ^1H and ^{13}C NMR spectroscopy, and HRMS. Furthermore, the structures of compound
204 **I9** and compound **I14** were confirmed using X-ray diffraction analysis (CCDC
205 1834640 and 1874641; **Fig. 3**).



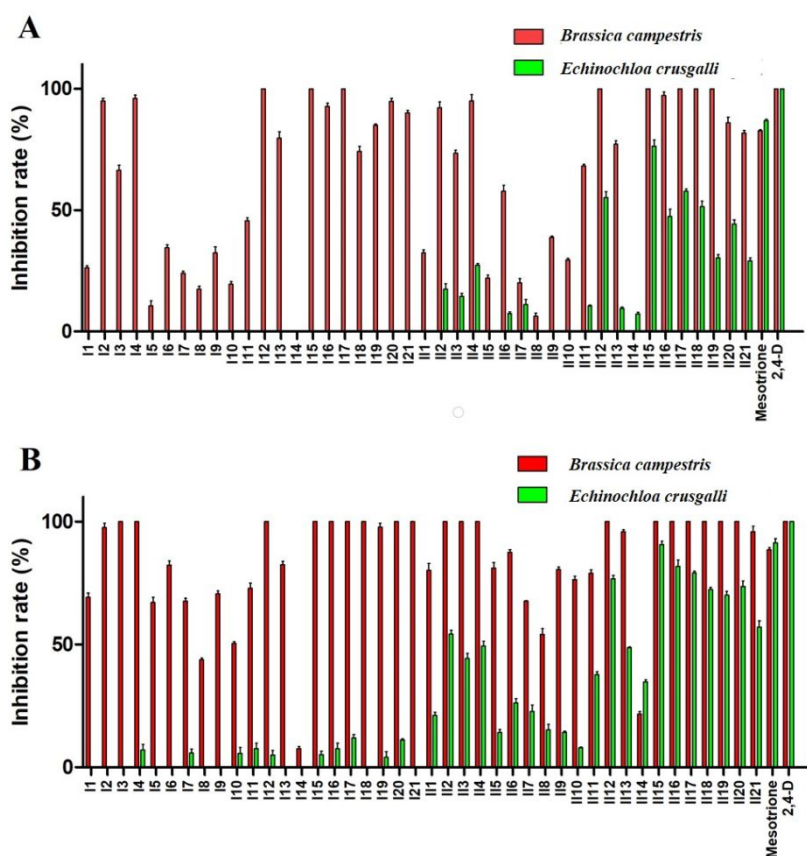
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207 **Figure 3.** X-ray crystal structures of compounds **I9** (right) and **I14** (left).

208 *In vitro* herbicidal activity

209 The herbicidal activity of the target compounds was preliminarily determined using
210 a *Brassica campestris* root test and *Echinochloa crusgalli* cup test. Mesotrione and

211 2,4-D were selected as positive control samples. As shown in **Fig. 4A**, the target
 212 compounds, such as **I2**, **I4**, **II2**, **II5-II17**, **I20**, **I21**, **II2**, **II4**, **II12**, and **II15-II19**,
 213 exhibited excellent herbicidal activity against the dicotyledonous plant *Brassica*
 214 *campestris* with inhibition >90% even at a dosage of 10 $\mu\text{g mL}^{-1}$. However, only
 215 some of the compound **II**, such as **II12** and **II15-II20**, exhibited >70% control against
 216 *Echinochloa crusgalli* at a dosage of 100 $\mu\text{g mL}^{-1}$ (**Fig. 4B**). These preliminary results
 217 indicate that the target compounds have good herbicidal activity and exhibit better
 218 herbicidal activity against dicotyledonous plants.



219
 220 **Figure 4.** Effects (% inhibition) of compounds **I1-I21** and **II1-II21** on mustard root
 221 and barnyardgrass seedling growth; **(A)** dosage = 10 $\mu\text{g mL}^{-1}$; **(B)** dosage = 100 μg
 222 mL^{-1} ; The error bars represent the standard deviation (s. d.) from three biological
 223 replicates.

224 **Molecular mode of action of the target compounds**

225 In order to explore the molecular mode of action of target compounds, compound
226 **II15** was selected to study the herbicidal mechanism with *Arabidopsis thaliana* as a
227 model plant. As shown in **Fig. 5A**, *Arabidopsis thaliana* treated with mesotrione at a
228 dosage of 1.0 μM developed bleaching symptoms, whereas *Arabidopsis thaliana*
229 treated with 2,4-D, 2-(2-chloro-4-fluorophenoxy) acetic acid,
230 4-hydroxy-6-methyl-pyran-2-one or compound **II15** did not develop any bleaching
231 symptoms, indicating these compounds have a herbicidal mechanism different to that
232 of mesotrione. Subsequently, the phenotypes of *Arabidopsis thaliana* in response to a
233 range of compound **II15** concentrations were investigated. It was found that
234 compound **II15** displays significant inhibitory activity against *Arabidopsis thaliana*
235 root growth with IC_{50} value of 26.42 nM, which comparable with commercial
236 herbicide 2,4-D (IC_{50} =13.55 nM) (Inhibition curve is given in Supporting Information,
237 **Fig. S1**). Interestingly, *Arabidopsis thaliana* treated with compound **II15** at a lower
238 concentration (such as 0.1 and 0.3 μM) exhibited obviously inhibited root elongation
239 and induced in lateral root hairs formation, which consistent with the observation that
240 plants exposed to 2,4-D or 2-(2-chloro-4-fluorophenoxy) acetic acid (**Fig. 5B**). These
241 preliminary phenotypes indicate that compound **II15** may be an auxin-type compound
242 and has a herbicidal mechanism similar to that of 2,4-D.

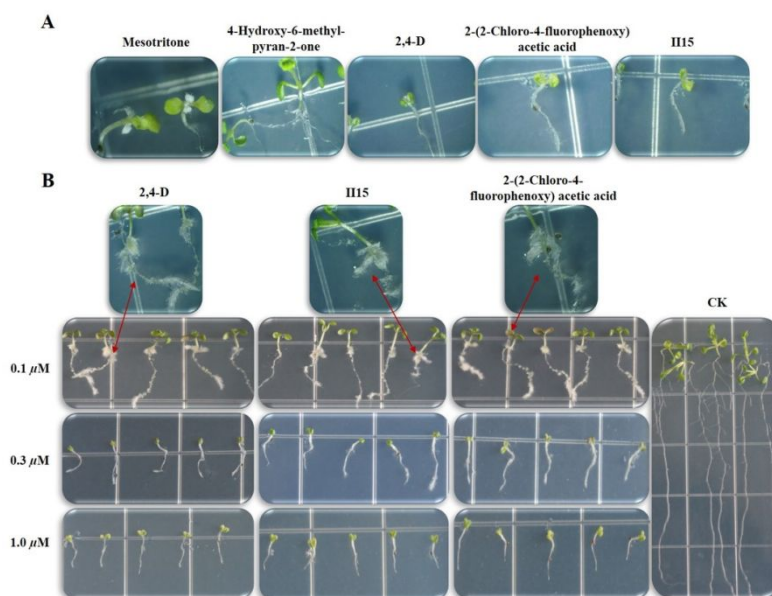
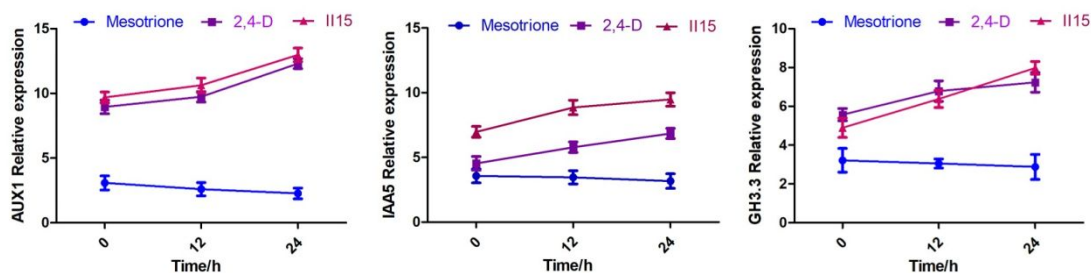


Figure 5. Photographs illustrating seedling and root phenotype of *Arabidopsis*

thaliana

To further verify whether compound **III5** is an auxin-type compound, the effect of compound **III5** on the expression levels of auxin related genes was investigated. The auxin induced gene (*IAA5*), auxin regulated gene (*GH3.3*) and auxin transport gene (*AUX1*) were selected for test, since these genes play an important role in auxin early response and transport³⁰⁻³⁷. As shown in **Fig. 6**, the trend in the change observed for the tested genes induced by mesotrione was different to that of 2,4-D. The expression levels of *IAA5*, *GH3.3* and *AUX1* were down-regulated after *Arabidopsis thaliana* treated with mesotrione. Since mesotrione is not an auxin analogue, the downregulation of these genes may be induced by its damage to the plant. To our delight, upon increasing the treatment time with compound **III5**, the expression levels of *IAA5*, *GH3.3* and *AUX1* was up-regulation, and the change trend was consistent with those observed for 2,4-D, which confirmed the auxin-type property of compound **III5**.



259

260 **Figure 6.** Effect of compound **III15**, mesotrione, and 2,4-D on the expression levels of261 *IAA5*, *GH3.3* and *AUX1*; The error bars represent the standard deviation (s. d.) from

262

three biological replicates.

263

To explore whether the auxin-type property of compound **III15** are due to its264 degradation products, the degradation of compound **III15** in *Arabidopsis thaliana* was

265 investigated. 2-(2-Chloro-4-fluorophenoxy) acetic acid was selected as a control. It

266 was found that 2-(2-chloro-4-fluorophenoxy) acetic acid and compound **III15** were267 gradually degraded in the plant (**Fig. 7A and 7B**). Interestingly, with the degradation268 of compound **III15**, a new peak was observed at 19 min, which is same as the peak269 time observed for 2-(2-chloro-4-fluorophenoxy) acetic acid (**Fig. 7B**). LC–MS270 analysis showed that there are two dominant ions m/z 311.17 ($M-H^+$) and m/z 203.14271 ($M-H^+$) in negative ionization mode, which corresponds to the molecular weight of272 compound **III15** ($M = 312.02$) and 2-(2-chloro-4-fluorophenoxy) acetic acid ($M =$ 273 204.00) (**Fig. 7C**). This result indicates that compound **III15** is degraded to

274 2-(2-chloro-4-fluorophenoxy) acetic acid in the plant. It is worth noting that, after 2.5

275 h, 2-(2-chloro-4-fluorophenoxy) acetic acid was completely degraded, while

276 compound **III15** undergone continuous degradation to release277 2-(2-chloro-4-fluorophenoxy) acetic acid, indicating compound **III15** will prolong the

278 lifetime of phenoxycarboxylic acid in the plant and exert a sustained herbicidal effect.

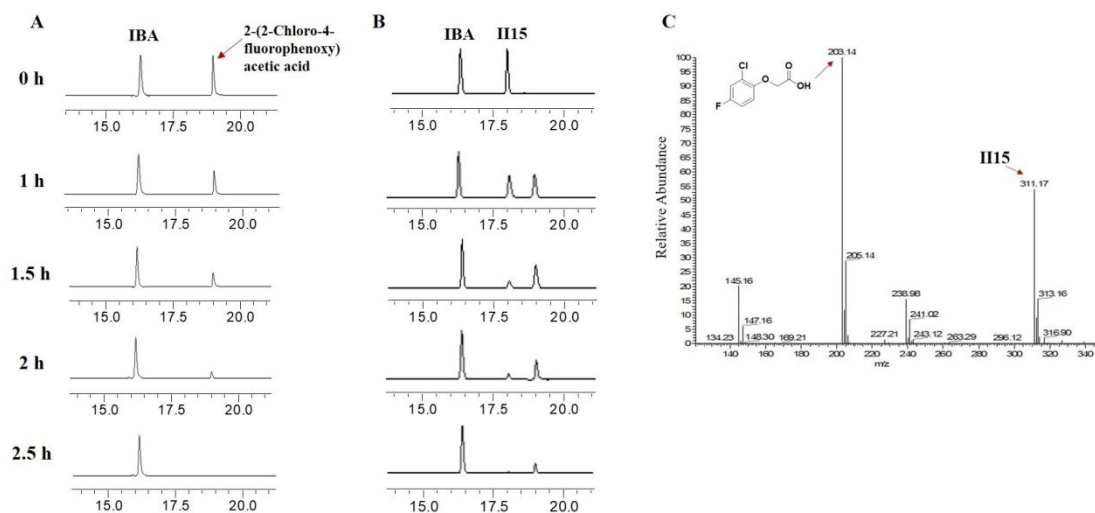


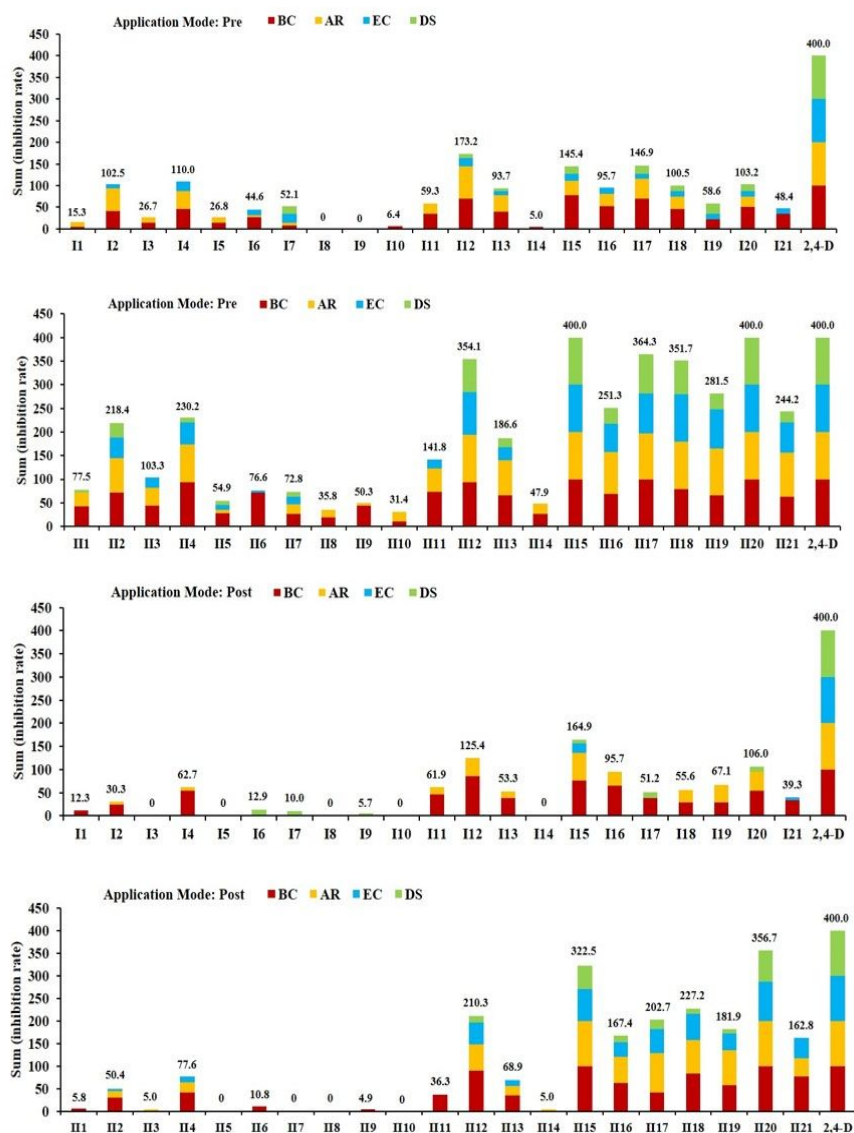
Figure 7. Degradation analysis of 2-(2-chloro-4-fluorophenoxy) acetic acid and

compound **III5** in *Arabidopsis thaliana*

Herbicidal activity in greenhouse tests and SARs study

Based on the above preliminary bioassay results, the herbicidal activity of all target compounds was further tested on four species representative of monocotyledonous and dicotyledonous plants at a dosage of 1500 g ha⁻¹ located in a greenhouse. 2,4-D was selected as a positive control. As shown in **Figure 8**, most of the target compounds displayed stronger herbicidal activity against dicotyledonous plants when compared to monocotyledonous plants, which was consistent with the preliminary *in vitro* results. Moreover, in most cases, the target compounds had stronger pre-emergent herbicidal activity than post-emergent herbicidal activity against all the weeds tested. This may be attributed to the target compounds being easily absorbed or degraded in the soil when compared to on the surface of the plant. Furthermore, we were encouraged to observe that some of target compound **II**, such as **III5** and **II20**, exhibited 400% sum inhibition against all weeds tested under pre-emergence conditions, which equal to the herbicidal activity of 2,4-D. These bioassay results

296 indicate that compounds **II15** and **II20** are worthy of further study.



297

298 **Figure 8.** Effects (% inhibition) of compounds **I1-I21** and **II1-II21** on loss of plant

299 weight at a dosage of 1500 g ha⁻¹ in greenhouse testing; Pre: pre-emergence; Post:

300 post-emergence; BC: *Brassica campestris*; AR: *Amaranthus retroflexus*; EC:

301 *Echinochloa crusgalli*; DS: *Digitaria sanguinalis*.

302 Subsequently, compounds **II15** and **II20** were tested against all weeds to further

303 evaluated their activity using a dose reduction with serial two-fold dilutions. As

304 shown in **Table 1**, the herbicidal activity of target compounds **II15** and **II20** became

305 progressively lower against the weeds tested at a dosage of 750, 375, and 187.5 g ha⁻¹.
 306 In addition, upon decreasing the dosage, the herbicidal activity of compounds **II15**
 307 and **II20** under post-emergence conditions decreased faster than that observed under
 308 pre-emergence conditions, implying these compounds exhibit enhanced herbicidal
 309 activity under pre-emergence conditions than under post-emergence conditions.
 310 Compounds **II15** and **II20** exhibit good herbicidal activity against the weeds tested at
 311 a dosage of 750 g ha⁻¹ under pre-emergence conditions, which is comparable with the
 312 activity of 2,4-D. However, compound **II20** displayed lower herbicidal activity than
 313 did 2,4-D, when the dosage was reduced to 187.5 g ha⁻¹. To our delight, compound
 314 **II15** still displayed good pre-emergent herbicidal activity against *Brassica campestris*
 315 and *Digitaria sanguinalis* with 82.7 and 61.8% inhibition at a dosage of 187.5 g ha⁻¹,
 316 respectively, which is higher than the herbicidal activity observed for 2,4-D.
 317 Meanwhile, compound **II15** displays good pre-emergent herbicidal activity against
 318 *Amaranthus retroflexus* with 83.5% inhibition at a dosage of 187.5 g ha⁻¹, which is
 319 comparable with the activity observed for 2,4-D. These promising results indicate that
 320 compound **II15** has good herbicidal activity under pre-emergence conditions and may
 321 be serve as a lead compound for further optimization.

322 **Table 1.** Effects (% inhibition) of compounds **II15** and **II20** on loss of plant weight at
 323 a dosage of 1500 g ha⁻¹ in greenhouse testing ^a

| Comp. | <i>Brassica</i> | | <i>Amaranthus</i> | | <i>Echinochloa</i> | | <i>Digitaria</i> | |
|----------------------------|-------------------|------|--------------------|------|--------------------|------|--------------------|------|
| | <i>campestris</i> | | <i>retroflexus</i> | | <i>crusgalli</i> | | <i>sanguinalis</i> | |
| | Pre | Post | Pre | Post | Pre | Post | Pre | Post |
| Rate (g ha ⁻¹) | | | | | | | | |

| | | | | | | | | | |
|-------------|-------|----------|----------|----------|----------|----------|----------|----------|----------|
| | 750 | 100 | 100 | 100 | 83.3±0.9 | 77.0±1.2 | 34.7±1.8 | 98.3±0.9 | 31.2±0.9 |
| II15 | 375 | 95.5±1.4 | 41.5±1.2 | 95.4±0.9 | 75.9±1.6 | 62.4±2.0 | 8.9±1.0 | 94.7±0.7 | 15.9±2.1 |
| | 187.5 | 82.7±1.8 | 27.9±2.3 | 83.5±2.2 | 51.9±0.8 | 23.8±2.5 | 0 | 61.8±1.2 | 0 |
| | 750 | 100 | 75.0±1.1 | 100 | 78.2±1.8 | 91.8±1.8 | 29.7±1.3 | 72.3±2.1 | 40.4±1.7 |
| II20 | 375 | 91.4±0.6 | 47.8±1.2 | 95.5±0.7 | 59.1±1.6 | 72.4±3.7 | 9.8±3.1 | 60.3±1.2 | 34.7±0.7 |
| | 187.5 | 52.0±1.5 | 19.1±2.7 | 56.1±3.2 | 30.5±1.7 | 0 | 0 | 30.2±1.3 | 0 |
| | 750 | 100 | 100 | 100 | 100 | 99.3±0.7 | 51.1±1.8 | 100 | 52.8±2.5 |
| 2,4-D | 375 | 94.3±1.2 | 100 | 98.5±1.5 | 100 | 77.1±2.1 | 28.3±2.3 | 61.0±2.5 | 15.9±1.7 |
| | 187.5 | 47.7±4.6 | 100 | 94.4±1.7 | 100 | 54.6±1.4 | 13.8±2.1 | 20.0±1.7 | 0 |

324 ^a Each value represents the mean ±SD of three experiments

325 Based on the bioassay data obtained from the greenhouse tests at a dosage of 1500
 326 g ha⁻¹ under pre-emergence conditions, the SARs were preliminarily investigated
 327 performing parallel activity contrast studies between compounds **II2** and **II5-II21**, and
 328 compounds **II12**, and **II15-II21** against *Brassica campestris*, *Amaranthus retroflexus*,
 329 *Echinochloa crusgalli*, and *Digitaria sanguinalis*. The results shown in **Fig. 9** indicate
 330 that series **II** always displayed higher herbicidal activity against the weeds tested
 331 when compared to series **I**. The possible reason for this observation is that the plants
 332 have different absorption and/or metabolic capacity toward series **I** and **II**. The
 333 preliminary SARs indicate that removing the benzene ring from the
 334 4-hydroxylcoumarin fragment in series **I** was beneficial toward improving the
 335 herbicidal activity.

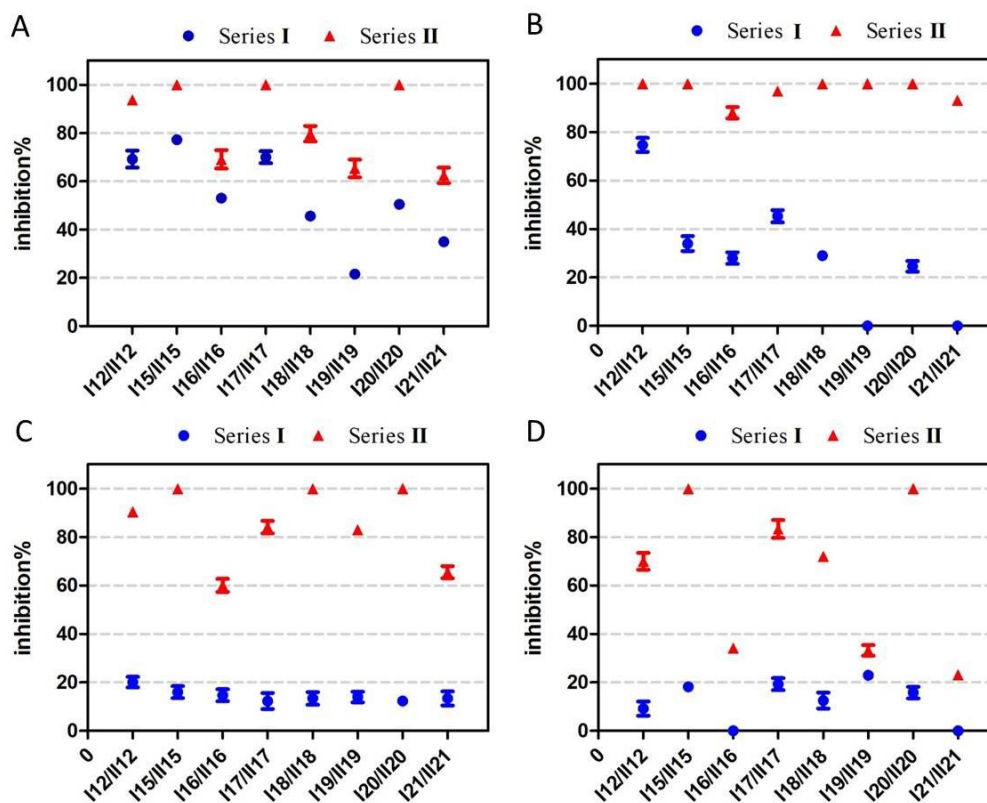


Figure 9. Parallel activity contrast studies between compounds **I12** and **I15-I21** and compounds **II12** and **II15-II21** against *Brassica campestris* (A), *Amaranthus retroflexus* (B), *Echinochloa crusgalli* (C), and *Digitaria sanguinalis* (D). The inhibition activity was tested at 1500 g ha⁻¹ and expressed in ordinate.

Analyzing the pre-emergent herbicidal activity against *Brassica campestris* among compounds **II1-I10** reveals that the compounds without a substituent or with an electron-donating group (compounds **II1** and **II5-II10**) generally had lower activities than did the compounds bearing an electron-withdrawing group (compounds **II2** and **II3**). For compounds **II11-II14** with two chlorine substituents on the benzene ring, the herbicidal activity against *Brassica campestris* could be placed in the following order: **II12** (2-Cl-4-Cl) > **II11** (2-Cl-3-Cl) > **II13** (2-Cl-5-Cl) > **II14** (3-Cl-5-Cl), which demonstrates that the 2,4-disubstitution pattern is the most active and the

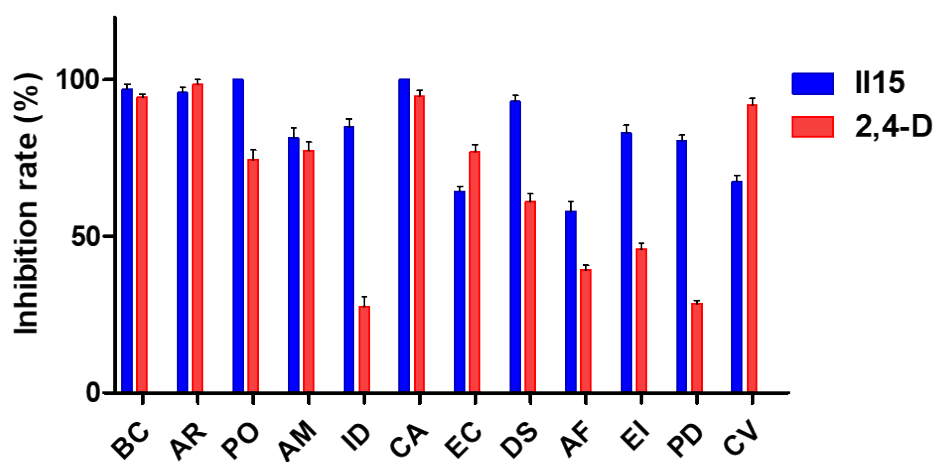
349 3,5-disubstitution pattern is the least active. These findings indicate that the
350 electron-withdrawing group introduced into benzene ring in compound **II** is beneficial
351 toward improving the herbicidal activity, and the position of the substituent on the
352 benzene ring is very important for herbicidal activity, with the 2,4-disubstitution
353 pattern confirmed as the optimal orientation.

354 Among the 2,4-disubstituted analogues (compounds **III2** and **III5-III21**),
355 compounds bearing 2-Cl-4-F, 2-Me-4-F, and 2-Br-4-F substituents on the benzene
356 ring display higher herbicidal activity against *Brassica campestris* than do those with
357 other types of substituents at the same positions. For example, the results of obtained
358 for the herbicidal activity of compounds **III2** and **III5-III19** show that their herbicidal
359 activity can be placed in the following order: **III5** (2-Cl-4-F) > **III2** (2-Cl-4-Cl) >
360 **III6** (2-Cl-4-Br); **III7** (2-Me-4-F) > **III8** (2-Me-4-Cl) > **III9** (2-Me-4-Br). These
361 findings indicate that a F atom introduced at the 4-position on the benzene ring of
362 compound **II** is beneficial to improving the herbicidal activity against *Brassica*
363 *campestris*.

364 **Herbicidal spectrum and crop safety of compound III5**

365 To evaluate whether compound **III5** has the potential to be developed as a
366 herbicide, we tested its herbicidal spectra and crop safety at a dosage of 375 g ha⁻¹.
367 The herbicidal spectrum experiments show that compound **III5** displays >80%
368 inhibition against 9 of the 12 tested weeds at 375 g ha⁻¹ (**Fig. 10**). It was worth noting
369 that compound **III5** displays >80% inhibition against *Ixeris denticulate*, *Eleusine*
370 *indica*, and *Puccinellia distans*, whereas 2,4-D only displays <50% inhibition to these

371 weeds. This finding indicates that compound **II15** has a broader spectrum of weed
 372 control than does the commercial herbicide 2,4-D at a dosage of 375 g ha⁻¹ under
 373 pre-emergence conditions. Subsequently, three representative crops, *Triticum*
 374 *aestivum* L., *Zea mays* Linn. and *Gossypium* spp, were selected for further crop
 375 selectivity studies (**Table 2**). The results showed that *Triticum aestivum* L. and *Zea*
 376 *mays* Linn. displayed a high tolerance toward compound **II15** at a dosage of 375 g
 377 ha⁻¹, but compound **II15** was not selective for *Gossypium* spp (51.3% injury). These
 378 promising results indicate that compound **II15** has the potential to be developed as a
 379 pre-emergence herbicide for weed control in *Triticum aestivum* L. and *Zea mays* Linn.
 380 fields.



Abbreviations: BC, *Brassica campestris*; AR, *Amaranthus retroflexus*; PO, *Portulaca oleracea*; AM, *Abutilon theophrasti* Medicus; ID, *Ixeris denticulate*; CA, *Chenopodium album*; EC, *Echinochloa crusgalli*; DS, *Digitaria sanguinalis*; AF, *Avena fatua*; EI, *Eleusine indica*; PD, *Puccinellia distans*; CV, *Chloris virgate*.

381

382 **Figure 10.** Herbicidal spectrum testing of compounds **II15** under pre-emergence
 383 condition at the dosage of 375 g ha⁻¹.

384 **Table 2.** Pre-emergence crop selectivity of compound **II15** at the dosage of 375 g ha⁻¹
 385 (injury inhibition)^a

| Comp. | % injury | | |
|--------|-----------------------------|-----------------------|----------------------|
| | <i>Triticum aestivum L.</i> | <i>Zea mays Linn.</i> | <i>Gossypium spp</i> |
| III5 | 0 | 0 | 51.3±2.6 |
| 2, 4-D | 0 | 0 | 0 |

^a Each value represents the mean ±SD of three experiments

386 In summary, a series of 4-hydroxyl-3-(2-phenoxyacetyl)-pyran-2-one derivatives
 387 have been designed using molecular hybridization between pyran-based diketone
 388 lactones and aryloxyacetic acid moieties. Forty-two 4-hydroxyl-3-(2-phenoxyacetyl)-
 389 pyran-2-one derivatives were prepared in moderate to good yield. Our bioassay results
 390 showed that compound **III5** displayed good pre-emergent herbicidal activity, even at
 391 a dosage of 187.5 g ha⁻¹. Our herbicidal spectrum study revealed that compound **III5**
 392 had a broader spectrum of weed control than did the commercial herbicide 2,4-D, and
 393 displayed good crop safety against *Triticum aestivum L.* and *Zea mays Linn.* at a
 394 dosage of 375 g ha⁻¹, which indicated its great potential as a herbicide for weed
 395 control in *Triticum aestivum L.* and *Zea mays Linn.* fields. By investigating the
 396 phenotypes of *Arabidopsis thaliana*, detecting the effect on auxin response genes and
 397 studying of degradation of compound **III5**, it was found that compound **III5** is
 398 metabolized to form an corresponding aryloxy acetic acid in the plant and has a
 399 herbicidal mechanism similar to that of 2,4-D, which indicates compound **III5** may
 400 be a potential lead structure for the further development of novel auxin-type
 401 herbicides. It is worth mentioning that, in our current work, the compounds obtained
 402 by molecular hybridization prolong the lifetime of herbicides in the plant, which may

403 exert a sustained herbicidal effect. This is a step forward as less herbicide need be
404 used, which would have a very important meaning in practical applications. Further
405 studies on the structural optimization of compound **III5** are ongoing in our laboratory.

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411 **SUPPORTING INFORMATION**

412 Supporting information may be found in the online version of this article.

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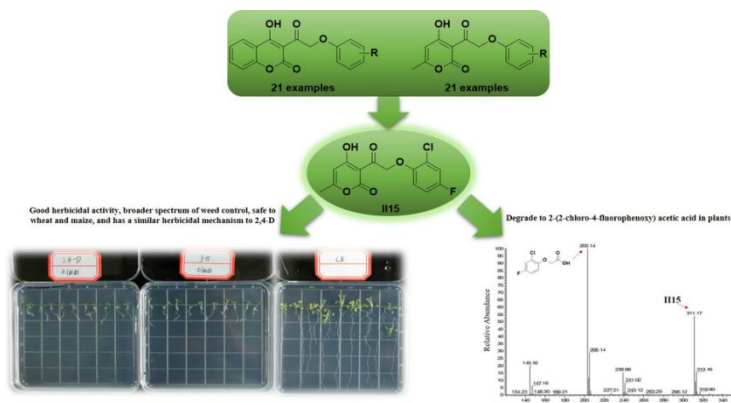
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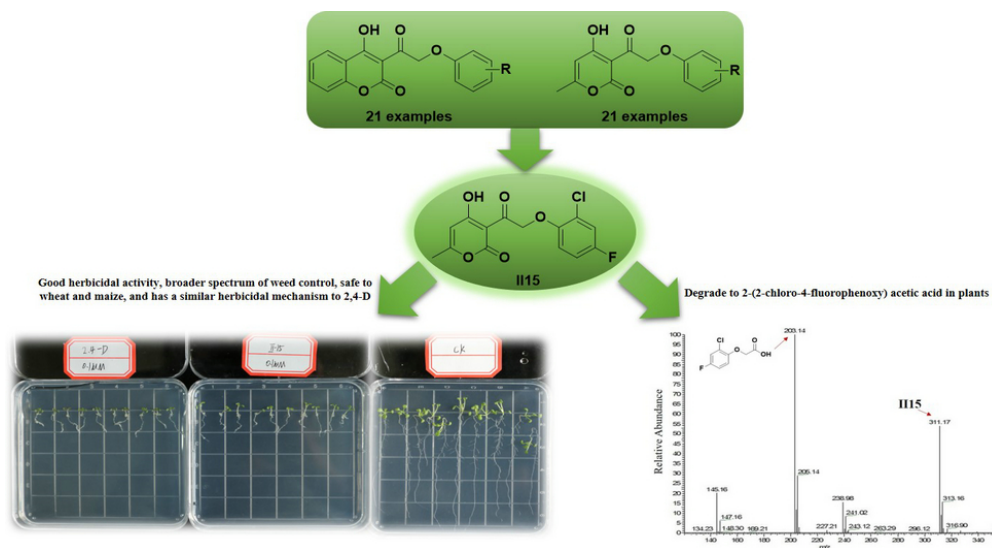
Design and synthesis of novel 4-hydroxy-3-(2-phenoxyacetyl)-pyran-2-one

531

derivatives for use as herbicides and evaluation of their mode of action



532



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