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Asymmetric Cyclization of 3,4-Dihydro-2-Vinyl-2H-1,4-Benzoxazine Catalyzed by Palladium-BHMP Catalyst¹

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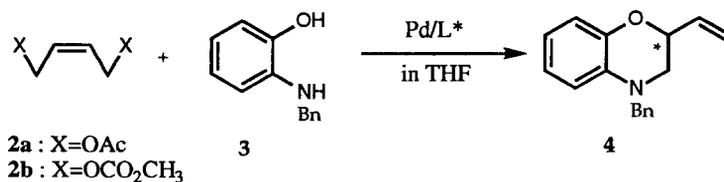
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Abstract: The reaction of 1,4-diacetoxy-*cis*-2-butene **2a** with 2-(benzylamino)phenol **3** in THF in the presence of Et₃N and a catalytic amount of Pd(0)-BHMP-β-Ala **1c** gave optically active 4-benzyl-2-vinylbenzoxazine of up to 56.2%ee. The reaction of (*Z*)-2-butene-1,4-diylbis(methylcarbonate) **2b** instead of **2a** with 2-(benzylamino)phenol **3**, **4** was obtained with e.e. up to 71.4%. We could improve the enantioselectivity of (*R*)-**4** by introducing a carboxyl group at the terminal position of the pendant side chain on the bisphosphine ligand and by using a methyl carbonate ester **2b** instead of diacetate **2a**.

There are many therapeutically and biologically active compounds in the 3,4-dihydro-2H-1,4-benzoxazine series.² Catalytic asymmetric construction of heterocycles is difficult, so only a few methods have been reported to date.³ Saegusa and co-workers reported the construction of morpholine and piperazine skeletons using a palladium catalyst bearing triisopropyl phosphite ligand.⁴ Hayashi et.al. reported that the reaction of 1,4-diacetoxy-*cis*-2-butene with 2-(benzylamino)ethanol was catalyzed by a palladium complex coordinated with (*R*)-BINAP to give optically active (*R*)-4-benzyl-2-vinylmorpholine in up to 65%ee.⁵ Similarly Sinou et.al. reported the asymmetric synthesis of 2-vinyl-1,4-benzodioxane in the presence of a catalytic amount of a palladium(0) with BINAP.⁶

In a previous paper, we showed the effectiveness of the catalytic cyclization of 2-vinylmorpholine using a palladium-BHMP catalyst, and improved the enantioselectivity of 2-vinylmorpholine up to 83.2%ee.⁷ In this paper we examine the extension of this reaction to other heterocycles containing an aromatic ring by use of a chiral bisphosphine ligand bearing a heterofunctional group on the side chain, expecting that the heterofunctional group on the BHMP ligand would interact with the incoming nucleophile.

Scheme 1



Reaction of (*Z*)-2-butene-1,4-diylbis(methylcarbonate) **2b** with 2-(benzylamino)phenol **3** was carried out in the presence of a palladium complex generated in situ by mixing a chiral ligand with Pd₂(dba)₃•CHCl₃ (1/Pd=1) as catalyst. Solutions of the chiral ligand (BHMP-β-Ala **1c**) (0.013mmol) and Pd (0.013mmol) in 2.0ml of THF was stirred at 20°C for 90 min. To the solution was added **2b** (0.25 mmol) and **3** (0.25 mmol), and the mixture was stirred at 20°C for 37hr. The solvent was removed in vacuo, the product (*R*)-4-benzyl-2-vinyl-benzoxazine **4** (31mg 48%) was isolated by silica gel column chromatography. The enantiomeric excess was determined by HPLC analysis (CHIRALCEL OB-H, n-hexane/2-propanol=300/1) to be 71.4%ee: [α]_D²² -11.6 (c 0.6 CHCl₃). The results are summarized in Table 1.

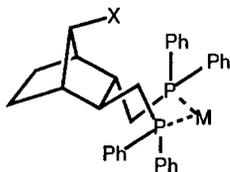
Table 1. Asymmetric Cyclization of 3,4-Dihydro-2-Vinyl-2H-1,4-Benzoxazine Catalyzed by Palladium-BHMP Catalyst.^a

entry	chiral ligand	substrate	°C, h	yield (%) ^b	ee% ^c (confign) ^d
1	(<i>S</i>)-BINAP	2b , -	20, 36	31	8.6 (<i>2R</i>)
2	(<i>2S,3S</i>)-NORPHOS	2b , -	20, 36	0	-
3	(<i>R</i>)-(<i>S</i>)-BPPFA	2b , -	20, 36	31	22.8 (<i>2S</i>)
4	BHMP (1a)	2a , Et ₃ N	45, 22	71	0
5	BHMP-Gly (1b)	2a , Et ₃ N	45, 44	32	27.5 (<i>2R</i>)
6	BHMP-Gly (1b)	2b , -	23, 18	71	50.4 (<i>2R</i>)
7	BHMP-Gly (1b)	2b , -	-20, 72	30	53.6 (<i>2R</i>)
8	BHMP-β-Ala (1c)	2a , Et ₃ N	40, 15	79	56.2 (<i>2R</i>) ^e
9	BHMP-β-Ala (1c)	2b , -	20, 37	48	71.4 (<i>2R</i>) ^e
10	BHMP-Ad-PrOH (1d)	2a , Et ₃ N	45, 40	99	3.3 (<i>2R</i>)

^a All entries were carried out under Ar in the presence of palladium complex prepared in situ by mixing a chiral ligand with Pd₂(dba)₃•CHCl₃ (1/Pd=1) as catalyst. ^b Isolated yield after silica gel column chromatography.

^c Determined by HPLC analysis with a chiral stationary phase column (DAICEL CHIRALCEL-OB-H).

^d Determined by the sign of the specific rotation. ^e [α]_D²² -11.6 (c 0.61 CHCl₃).



BHMP 7X (**1**)

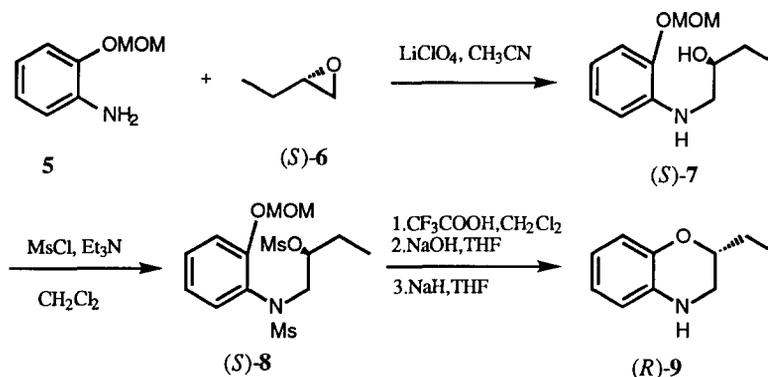
BHMP (1a): X=H
 BHMP-Gly (1b): X=OCH₂CONHCH₂COOH
 BHMP-β-Ala (1c): X=OCH₂CONH(CH₂)₂COOH
 BHMP-Ad-PrOH (1d): X=OCH₂CONH(CH₂)₃OH

The most stereoselective phosphine ligand was BHMP-β-Ala **1c**. The use of methylcarbonate ester **2b**⁸ instead of **2a**, was found to increase the enantioselectivity to 71.4% (entry 9), this trend was applied to the use of palladium-BHMP-Gly **1b** catalyst (entry 5, 6). Other heterofunctional groups on the bisphosphine ligand was examined. The use of BHMP-Ad-PrOH **1d** containing an alcohol unit at the terminal position of the pendant side chain that is about the same length of **1c**, gave **4** with low enantioselectivity (entry 10). Palladium complexes of other phosphine ligands including ((*S*)-(1,1'-binaphthalene)-2,2'-diylbis(diphenylphosphine)) (*S*)-BINAP⁹, ((*2S,3S*)bicyclo[2,2,1]hept-5-ene-2,3-diylbis(diphenylphosphine)) (*2S,3S*)-NORPHOS¹⁰, ((*R*)-1-[(*S*)-1',2-bis(diphenylphosphino)ferrocenyl]ethylamine) (*R*)-(*S*)-BPPFA¹¹ were

less stereoselective and/or showed no reaction (entry 1-3). As shown in Table 1, the carboxylic group on the ligand have an important influence on the enantioselectivity of the cyclization.

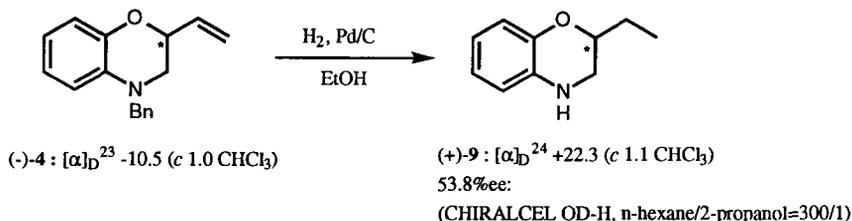
The absolute configuration of **4** was determined by correlation with (*R*)-**9** which was prepared from (*S*)-(-)-1,2-epoxybutane (**6**) by the following reactions: 2-methoxymethoxy aniline **5** reacted with (*S*)-1,2-epoxybutane **6**¹² in the presence of LiClO₄ to give aminoalcohol (*S*)-**7**, the treatment of which with methanesulfonyl chloride in the presence of Et₃N gave (*S*)-**8**. Deprotection of the hydroxy group (*S*)-**8** with trifluoroacetic acid and followed by reaction with NaOH in THF solution and finally treatment with sodium hydride gave the cyclized product (*R*)-**9** as a pure material (Scheme 2).

Scheme 2



Comparison of the HPLC analysis (CHIRALCEL OD-H ; n-hexane/2-propanol=300/1) of (+)-**9** ($[\alpha]_{\text{D}}^{24} +22.3$ (c 1.1 CHCl₃)) which was obtained from (-)-**4** ($[\alpha]_{\text{D}}^{23} -10.5$ (c 1.0 CHCl₃)) by the palladium-catalyzed hydrogenation with that of authentic (*R*)-**9** revealed the absolute configuration of (+)-**9** to be (*R*)-(+)-**9** (Scheme 3).

Scheme 3



In conclusion, 3,4-dihydro-2-vinyl-2H-1,4-benzoxazine **4** was efficiently synthesized by a palladium catalyzed asymmetric cyclization. This functionalized benzoxazine is a potentially versatile intermediate for the synthesis of various biologically active compounds.

Acknowledgment

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References

1. Asymmetric Reaction Catalyzed by Chiral Metal Complexes LXXI.
2. (a) Bartsch, H.; Schwarz, O. *J. Heterocyclic. Chem.* **1982**, *19*, 1189. (b) Bulter, C. M. R.; B. Chapleo, B. C.; Myers, L. P and Welbourn, P. A. *J. Heterocyclic. Chem.* **1985**, *22*, 177. (c) Atarashi, S.; Tsurumi, H.; Fujiwara, T.; Hayakawa, I. *J. Heterocyclic. Chem.* **1991**, *28*, 329. (d) Atarashi, S.; Yokohama, S.; Yamazaki, K.; Sakano, K.; Imamura, M.; Hayakawa, I. *Chem. Pharm. Bull.* **1987**, *35* (5), 1896. (e) D'Ambr, E. T.; Estep, G. K.; Bell, R. M.; Eissenstat, A. M.; Josef, A. K.; Ward, J. S.; Haycock, A. D.; Baizman, R. E.; Casiano, M. F.; Beblin, C. N.; Chippari, M. S.; Greo, D. J.; Kullnig, K. R.; Daley, T. G. *J. Med. Chem.* **1992**, *35*, 124. (f) Combs, D. W.; Rampulla, M. S.; Bell, S. C.; Klaubert, D. H.; Tobia, A. J.; Falotico, R.; Haertlein, B.; Lakas-Weiss, C.; Moor, J. B. *J. Med. Chem.* **1990**, *33*, 380. (g) Chapleo, C. B.; Butler, R. C. M.; England, D. C.; Myers, P. L.; Roach, A. G.; Smith, C. F. C.; Stillings, M. R.; Tulloch, I. F. *J. Med. Chem.* **1989**, *32*, 1627. (h) Heine, H. W.; Schairer, W. C.; Suriano, J. A. *Tetrahedron.* **1988**, *44*, 3181. (i) LARGERON, M.; Dupuy, H.; Fleury, M-B. *Tetrahedron*, **1995**, *51*, 4953.
3. (a) Hayashi, T.; Yamamoto, A.; Ito, Y. *Tetrahedron Lett.* **1988**, *29*, 669. (b) Mori, M.; Nukui, S. Shibasaki, M. *Chem. Lett.* **1991**, 1797. (c) Trost, B.M.; Van Vranken, D.L.; Bingel, C. *J. Am. Chem. Soc.* **1992**, *114*, 9327. (d) Nukui, S.; Sodeoka, M.; Shibasaki, M. *Tetrahedron Lett.* **1993**, *34*, 4965.
4. Tsuda, T.; Kiyoi, T.; Saegusa, T. *J. Org. Chem.* **1990**, *55*, 3388.
5. Uozumi, Y.; Tanahashi, A.; Hayashi, T. *J. Org. Chem.* **1993**, *58*, 6826.
6. Massacret, M.; Goux, C.; Lhoste, P.; Sinou, D. *Tetrahedron Lett.* **1994**, *35*, 6093.
7. Yamazaki, A.; Achiwa, K. *Tetrahedron: Asymmetry.* **1995**, *5*, 1021.
8. (a) Tsuji, J.; Shimizu, I.; Minami, I.; Ohashi, Y.; *Tetrahedron Lett.* **1982**, *23*, 4809. (b) Tsuji, J.; Shimizu, I.; Minami, I.; Ohashi, Y.; Sugiura, T.; Takahashi, K. *J. Org. Chem.* **1985**, *50*, 1523.
9. Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; and Noyori, R. *J. Am. Chem. Soc.* **1980**, *102*, 7932.
10. Brunner, H.; Pieronczyk, W.; Schonhammer, B.; Steng, K.; Bernal, I.; and Korp, J. *Chem. Ber.* **1981**, *114*, 1137.
11. Hayashi, T.; Mise, T.; Fukushima, M.; Kagotani, M.; Nagashima, N.; Hamada, Y.; Matsumoto, A.; Kawakami, S.; Konishi, M.; Yamamoto, K.; Kumada, M. *Bull. Chem. Soc. Jpn.* **1980**, *53*, 1138.
12. Mori, K.; Sasaki, M.; Tamada, S.; Suguro, T.; and Masuda, S. *Tetrahedron.* **1979**, *35*, 1601.

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