

## Polymer-supported Chain Homologation

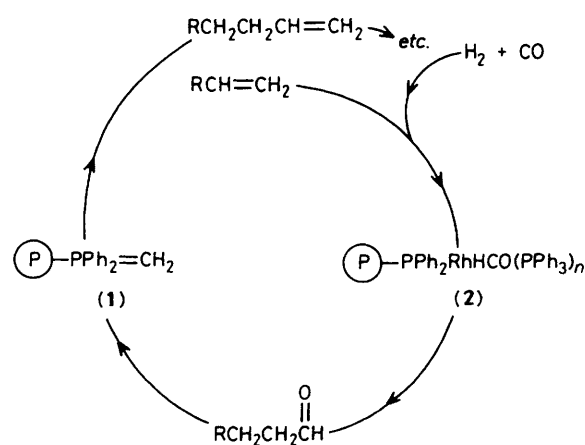
Steven L. Regen\*† and Mitsuo Kodomari

Department of Chemistry, Marquette University, Milwaukee, Wisconsin 53233, U.S.A.

Addition of hept-1-ene to a suspension of polystyrene-bound forms of hydridocarbonyltris(triphenylphosphine)rhodium(I) and methylenetriphenylphosphorane in tetrahydrofuran, under a hydrogen/carbon monoxide atmosphere (60 °C; 120 lb in<sup>-2</sup>), produces 2-methylheptanal (10%), n-octanal (45%), n-decanal (12%), n-dodecanal (2%), and n-tetradecanal (0.3%) as major products.

The use of synthesis gas (H<sub>2</sub> + CO) in combination with Fischer–Tropsch catalysts provides a practical synthetic route to a homologous series of n-aliphatic alcohols.<sup>1</sup> Major limitations of this method are that product mixtures are complex (generally obeying a Schultz–Flory distribution<sup>2</sup>) and that high reaction temperatures (>200 °C) are required. Here we report the coupling of a polymeric hydroformylation process with a polymeric Wittig reaction, resulting in a non-Schultz–Flory distribution of aldehydes, under mild conditions (60 °C; 120 lb in<sup>-2</sup>).

The method involves co-suspension of a resinous Wittig reagent (1) with an immobilized hydroformylation catalyst (2) in an organic solvent. A suitable alkene is introduced into the system, and the mixture is heated under an atmosphere of H<sub>2</sub> + CO. Diffusion of alkene into and aldehyde product out of



Scheme 1

† Present address: Department of Chemistry, Lehigh University, Bethlehem, Pennsylvania 18015, U.S.A.

the catalyst (**2**), followed by alkene regeneration *via* methylene transfer from the Wittig reagent (**1**), completes one cycle and extends the chain by two carbon atoms.

A stainless steel reaction vessel (75 ml) equipped with a Teflon-coated magnetic stirring bar, was charged with the Wittig reagent (**1**) (0.81 g, 1.0 mmol; generated by treatment of a 2% cross-linked, 53% ring-substituted polymeric phosphonium salt with *n*-butyl-lithium in hexane<sup>3</sup>), the catalyst (**2**) (0.840 g, 0.30 mmol Rh; produced from a 2% cross-linked, 53% ring-substituted phosphine polymer, 0.36 mmol Rh per g),<sup>4</sup> tetrahydrofuran (10 ml; distilled from sodium benzophenone ketyl), and hept-1-ene (0.049 g, 0.5 mmol) under nitrogen. A 1 : 1 mixture of H<sub>2</sub> and CO was introduced (120 lb in<sup>-2</sup>) at room temperature. The system was heated to 60 °C, stirred for 16 h, cooled to *ca.* 5 °C, vented, and analysed by g.l.c. Major products identified by g.l.c.-mass spectrometric comparisons with authentic samples were 2-methylheptanal (10%), *n*-octanal (45%), *n*-decanal (12%), *n*-dodecanal (2%), and *n*-tetradecanal (0.3%). Control experiments carried out with the Wittig reagent (**1**) plus soluble hydridocarbonyltris-(triphenylphosphine)rhodium(I) yielded only unchanged hept-1-ene.

Although the required use of stoichiometric quantities of Wittig reagent (**1**) precludes any practical development of this system, these results demonstrate that mild, non-Schulz-Flory chain homologation based on synthesis gas is possible. Efforts are being directed toward the development of a fully catalytic chain homologation employing synthesis gas.

We are grateful to the Division of Basic Energy Sciences of the U.S. Department of Energy for financial support of this work.

*Received, 22nd April 1987; Com. 539*

## References

- 1 R. B. Anderson, 'The Fischer-Tropsch Synthesis,' Academic Press, New York, 1984.
- 2 G. Odian, 'Principles of Polymerization,' 2nd edn., Wiley, New York, 1981, p. 89.
- 3 M. Bernard and W. T. Ford, *J. Org. Chem.*, 1983, **48**, 326.
- 4 C. U. Pittman, Jr., L. R. Smith and R. M. Hanes, *J. Am. Chem. Soc.*, 1975, **97**, 1742.