

Synthesis and transformations of metallacycles

28.* Reactions of allenes with EtAlCl₂ and Et₂AlCl catalyzed by Ti and Zr complexes

A. G. Ibragimov,* L. O. Khafizova, G. N. Gil'fanova, and U. M. Dzhemilev

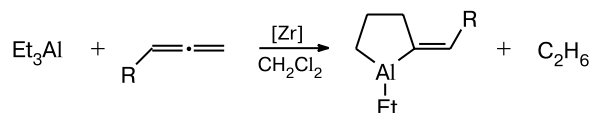
Institute of Petrochemistry and Catalysis,
Bashkortostan Republic Academy of Sciences and Ufa Research Center of the Russian Academy of Sciences,
141 prosp. Oktyabrya, 450075 Ufa, Russian Federation.
Fax: +7 (347 2) 31 2750. E-mail: ink@anrb.ru

Catalytic cycloaluminum of allenes with EtAlCl₂ in the presence of Ti or Zr complexes afforded methylidene- and alkyl(benzyl)idenealuminacyclopropanes and the corresponding aluminacyclopentanes, which were identified by analyzing the hydrolysis products. The reactions with the use of Et₂AlCl instead of EtAlCl₂ produced 1,2- and 1,4-dialuminum compounds.

Key words: organoaluminum compounds, catalysis, cycloaluminum, aluminacyclopropanes, aluminacyclopentanes, allenes.

Cycloaluminum of allenes with Et₃Al catalyzed by Cp₂ZrCl₂ afforded 2-alkylidene-1-ethylaluminacyclopentanes (Scheme 1).²

Scheme 1



With the aim of extending the scope of this reaction, applying it to other organoaluminum compounds (OAC), and synthesizing new classes of three- and five-membered unsaturated OAC, we studied cycloaluminum of alkyl- and phenylallenes with EtAlCl₂ in the presence of metallic Mg as the acceptor of chloride ions and Cp₂TiCl₂ as the catalyst, which exhibits^{3–6} high activity and selectivity in the formation of three-membered OAC from EtAlCl₂ and α -olefins or acetylenes.

Results and Discussion

We studied the reactions of allenes which differ in the nature of the substituent (Ph (**1a**), CH₂Ph (**1b**), Alk) and the length of the alkyl chain (C₆H₁₃, C₈H₁₇, C₉H₁₉ (**1c–e**, respectively)) with EtAlCl₂. Preliminary experiments demonstrated that the reactions in the presence of Mg and the catalyst Cp₂TiCl₂ (molar ratio RCH=C=CH₂ : [Al] : Mg : [Ti] = 20 : 26 : 24 : 1) afforded in 8–12 h unsaturated cyclic OAC, viz., 3-alkyl(ben-

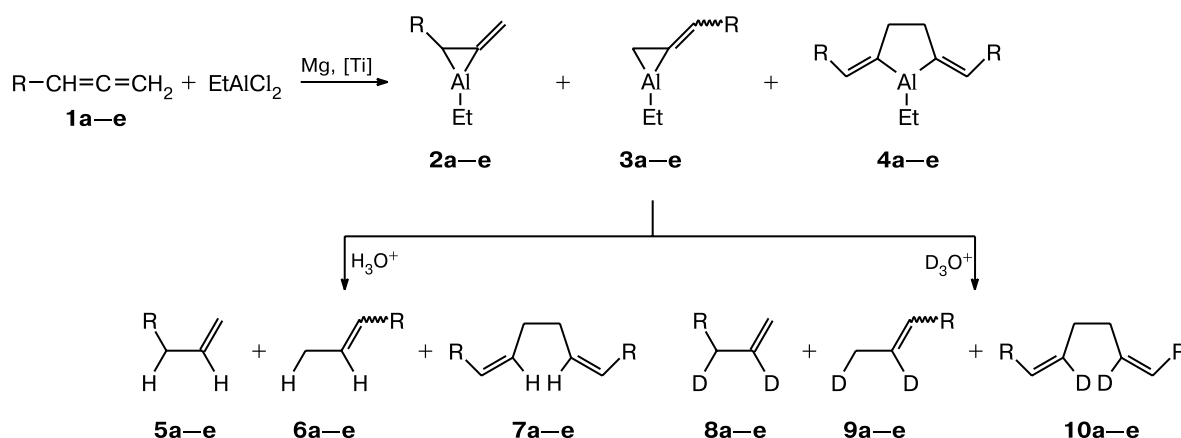
zyl,phenyl)-1-ethyl-2-methylidenealuminacyclopropanes (**2**), 2-alkyl(benzyl)idene-1-ethylaluminacyclopropanes (**3**), and 2,5-di(alkyl(benzyl)idene)-1-ethylaluminacyclopentanes (**4**), which were identified by analyzing the hydrolysis products.

The effect of the nature of the solvent on the yields and compositions of the reaction products was studied. In THF, cycloaluminum of nona-1,2-diene (**1c**) with EtAlCl₂ in the presence of 5 mol.% of Cp₂TiCl₂ afforded organoaluminum compounds **2c**, **3c**, and **4c** in the ratio 6 : 1 : 2 in a total yield of ~80% (Scheme 2). In aliphatic (hexane, heptane, cyclohexane) or aromatic (benzene, toluene, xylene) solvents, cycloaluminum of allenes did not virtually take place. In diethyl ether, cyclic organoaluminum compounds **2c**, **3c**, and **4c** were generated in the ratio 9 : 1 : 10 in a total yield of ~60%.

The structures of unsaturated cyclic organoaluminum compounds **2–4** were established by analyzing the hydrolysis (**5–7**) or deuterolysis products (**8–10**). According to the ¹³C NMR spectroscopic data, alkenes **6a,b** (**9a,b**) and **7a–e** (**10a–e**) contain *Z*-disubstituted double bonds, whereas the double bonds in molecules **6c–e** (**9c–e**) have the *Z/E* configurations in a ratio of ~2 : 1. According to the ¹H NMR spectra, hydrolysis products **7a–e** contain the *cis*-double bonds (spin-spin coupling constant of the alkene protons ³*J* is 11.7 Hz for **7a**; cf. lit. data⁷: ³*J*_{trans} = 18 Hz). The *cis* configuration of the double bond in compounds **7c–e** is confirmed by the fact that their ¹³C NMR spectra have characteristic⁸ high-field signals for the allylic C atoms at δ 26–28. Mass spectrometric analysis of deuterolysis products showed that the mixtures contained dideuterioalkenes **8–10** along with mono-deuterated alkenes as the minor products (10–12%),

* For Part 27, see Ref. 1.

Scheme 2



R = Ph (**a**), CH_2Ph (**b**), $n\text{-C}_6\text{H}_{13}$ (**c**), $n\text{-C}_8\text{H}_{17}$ (**d**), $n\text{-C}_9\text{H}_{19}$ (**e**);
[Ti] = Cp_2TiCl_2

which indicates that cyclometallation was accompanied by competitive hydrometallation of the starting allenes.

The formation of unsaturated cyclic organoaluminum compounds **2–4** is attributable to transmetalation of the *in situ* generated unsaturated titanacyclopentane and titanacyclopentane intermediates with the starting EtAlCl_2 according to a scheme analogous to that proposed in our previous study⁹ for cyclometallation of arylenes. In these processes, the configurations of the *in situ* generated alkylidenetitanacyclopentane and -titanacyclopentane intermediates dictate the stereoconfigurations of the alkylidene-substituted organoaluminum compounds **3** and **4**. Apparently, hydride titanium complexes are responsible for the formation of minor hydrometallation products of allenes. Under the reaction conditions, these complexes are generated through dehydrogenolysis of THF with low-valent titanium complexes,¹⁰ as we have demonstrated previously¹¹ in the investigation of hydrometallation of α -olefins.

When studying cycloaluminum of nona-1,2-diene (**1c**) with EtAlCl_2 , we found that the reactions with the use of titanium alkoxides or halides ($\text{Ti}(\text{OPr}^i)_4$ or TiCl_4) instead of Cp_2TiCl_2 proceeded less selectively to give 1-ethyl-3-(*n*-hexyl)-2-methylidenealuminacyclopentane **2c** (15–20%), 1-ethyl-2-heptylidenealuminacyclopentane **3c** (8–10%), and a mixture of unidentified unsaturated aluminacyclopentanes and aluminacycloheptanes (55–60%). This is attributable to the insertion of an additional molecule of the starting allene at the Ti–C bond of intermediate titanacyclopentanes followed by transmetalation of *in situ* generated unsaturated titanacycloheptanes with the starting EtAlCl_2 . Cycloaluminum of nona-1,2-diene (**1c**) with EtAlCl_2 in the presence of 5 mol.% of Cp_2ZrCl_2 or ZrCl_4 afforded organoaluminum compounds **2c** and **3c** in a ratio of ~5 : 1 in ~30% yield

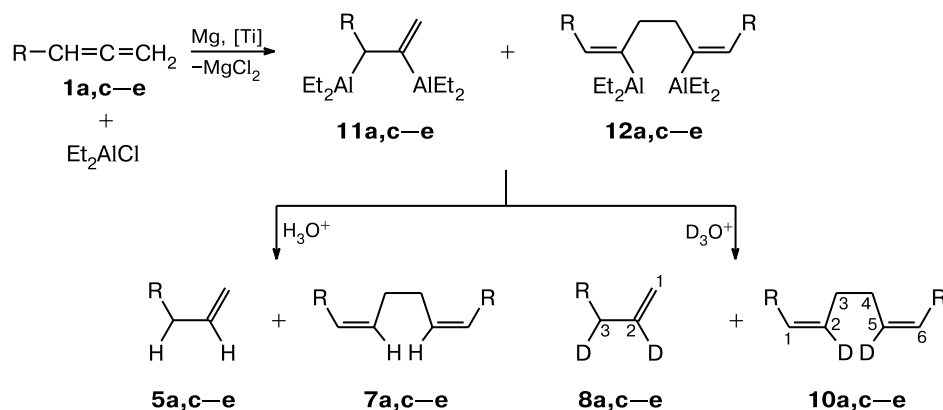
and a complex mixture of unsaturated aluminacyclopentanes in ~50% yield.

The reactions with Et_2AlCl instead of EtAlCl_2 follow another pathway. Under the optimum conditions, these reactions afford predominantly unsaturated 1,2- and 1,4-dialuminum compounds (Scheme 3). Actually, the reactions of allenes **1a,c–e** with Et_2AlCl in THF in the presence of Cp_2TiCl_2 as the catalyst taken in the [Al] : $\text{RCH}=\text{C}=\text{CH}_2$: Mg : [Ti] ratio of ~44 : 20 : 24 : 1 gave rise to 2-alkyl(phenyl)-1,2-bis(diethylaluminumio)-1-methylidenoethanes (**11a,c–e**) and 1,4-dialkyl(benzylidene)-1,4-bis(diethylaluminumio)butanes (**12a,c–e**) in a ratio of ~5 : 4 in a total yield of 75–80%. Mass-spectrometric analysis demonstrated that deuterolysis afforded not only 1,2- and 1,4-dideuterated compounds **8** and **10** but also monodeuterated alkenes in 10–15% yields, which indicates that dialumination was accompanied by competitive hydroalumination of allenes according to a scheme proposed previously.¹⁰

It should be noted that dialumination of allenes in the presence of Cp_2ZrCl_2 or ZrCl_4 (5 mol.%) as the catalyst facilitates the formation of unsaturated acyclic organoaluminum compounds **11** and a complex mixture of unsaturated regioisomeric 1,4-dialuminum compounds (according to the results of GLC and mass spectrometry of deuterolysis products) in a ratio of 1 : 2 in a total yield of ~80%.

The experimental results obtained in the present study and the published data^{1–6} on cycloaluminum of unsaturated compounds allowed us to suggest the most probable scheme of 1,2- and 1,4-dialumination of allenes in the presence of Cp_2TiCl_2 as the catalyst (Scheme 4). According to this scheme, Mg used in this reaction reduces Cp_2TiCl_2 to "Cp₂Ti". The formally divalent "titanocene" coordinates the allene molecule to form titanacyclo-

Scheme 3



R = Ph (**a**), $n-C_6H_{13}$ (**c**), $n-C_8H_{17}$ (**d**), $n-C_9H_{19}$ (**e**);
 [Ti] = Cp_2TiCl_2

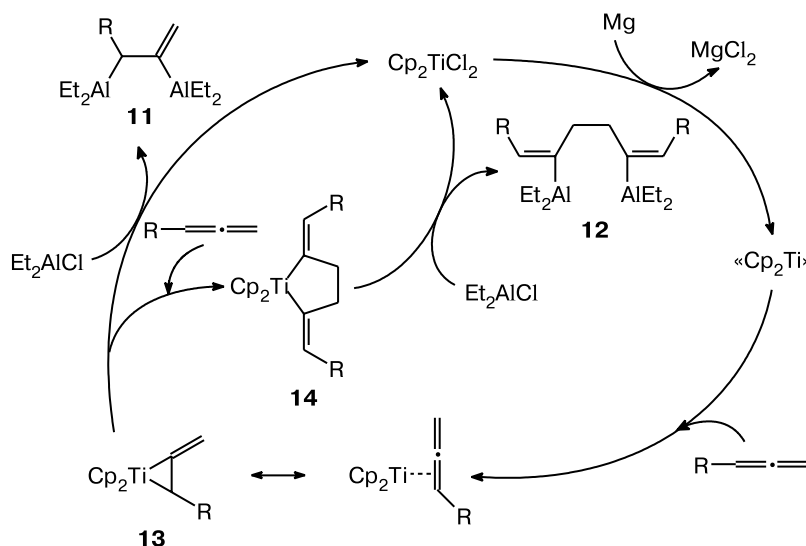
propane intermediate **13** whose transmetalation with Et_2AlCl affords 2-alkyl(phenyl)-1,2-bis(diethylaluminum)-1-methylideneethane **11**. The insertion of the second allene molecule at the Ti—C bond of titanacyclopentane intermediate **13** facilitates the highly regioselective formation of 2,5-dialkyl(benzyl)idenetitanocyclopentane **14** whose transmetalation with the starting Et_2AlCl affords 1,4-dialkyl(benzyl)idene-1,4-bis(diethylaluminum)butane **12**.

Thus, the reactions of allenes with $EtAlCl_2$ or Et_2AlCl in the presence of Ti- or Zr-containing complex catalysts afford unsaturated aluminacyclopropanes and aluminacyclopentanes or 1,2- and 1,4-dialuminum compounds depending on the structures of the starting OAC.

Experimental

The reactions with organometallic compounds were carried out under dry argon. The starting allenes, Cp_2TiCl_2 , and Cp_2ZrCl_2 were prepared according to known procedures.^{12–15} Commercially available 86% $EtAlCl_2$ and 86% Et_2AlCl were purchased from the Redkinskii pilot-production plant (Russia). Tetrahydrofuran was distilled from $LiAlH_4$ immediately before use. The GLC analysis of the hydrolysis and deuteration products was carried out on a Chrom-5 chromatograph (Ar as the carrier gas, 1200×3-mm column with 5% SE-30 or 15% PEG-6000 on Chromaton N-AW). The IR spectra were measured on a UR-20 spectrometer (films). The mass spectra were recorded on an MKh-1306 mass spectrometer (70 eV, 200 °C). The 1H NMR spectra were measured on a Tesla BS-567

Scheme 4



spectrometer (100 MHz, Me₄Si as the internal standard) in CDCl₃. The ¹³C NMR spectra were recorded on JEOL FX-90Q (22.5 MHz) and Bruker AM-300 spectrometers (75.46 MHz) with the JMOD mode broad-band and off-resonance proton decoupling. Elemental analysis was performed on a Carlo Erba instrument (model 1106). The yields and ratios of unsaturated OAC were determined by GLC of hydrolysis or deutero-lysis products. The spectroscopic parameters of unsaturated OAC **2–4**, **11**, and **12** are not reported because of the complexity of their interpretation (¹³C NMR spectra of these compounds have high-field signals for the C_α atoms characteristic of OAC, which are broadened due to relaxation on the ²⁷Al nuclei).

Reactions of allenes with EtAlCl₂ catalyzed by Cp₂TiCl₂ (ZrCl₄). A mixture of Cp₂TiCl₂ (ZrCl₄) (0.5 mmol), Mg (powder, 12 mg-at.), THF (10 mL), 1,2-diene (10 mmol), and EtAlCl₂ (13 mmol) was placed with stirring in a glass vessel under dry argon at 0 °C. The mixture was warmed to ~20 °C and stirred for 8–12 h. The reaction mixture was treated with 7–10% HCl or DCl in D₂O. The products were extracted with hexane, dried with MgSO₄, and isolated by distillation.

The reactions of allenes with Et₂AlCl catalyzed by Cp₂TiCl₂ (ZrCl₄) were carried out under the same conditions with the use of 2.2 mol. equiv. of Et₂AlCl (with respect to 1,2-diene).

Allylbenzene (5a). B.p. 74–75 °C (20 Torr). IR, ν/cm⁻¹: 3020, 2950, 2850, 1600, 1490, 1450, 940, 850. ¹H NMR, δ: 3.36 (d, 2 H, PhCH₂, *J* = 7 Hz); 4.98–5.64 (m, 3 H, –CH=CH₂); 6.94 (m, 5 H, Ph). ¹³C NMR, δ: 40.16 (t, C(1')); 115.63 (t, C(3')); 126.03 (d, C(4)); 128.24 (d, C(3), C(5)); 128.57 (d, C(2), C(6)); 137.45 (d, C(2')); 143.90 (s, C(1)). MS, *m/z*: 118 [M]⁺.

But-3-enylbenzene (5b). B.p. 62–63 °C (10 Torr). ¹H NMR, δ: 2.25–2.42 (m, 2 H, CH₂–C=); 3.37 (t, 2 H, PhCH₂, *J* = 7 Hz); 4.85–6.08 (m, 3 H, –CH=CH₂); 7.00–7.29 (m, 5 H, Ph). ¹³C NMR, δ: 35.35 (t, C(2')); 39.00 (t, C(1')); 114.60 (t, C(4')); 125.72 (d, C(4)); 128.11 (d, C(3), C(5)); 129.15 (d, C(2), C(6)); 137.61 (d, C(3')); 138.03 (s, C(1)). MS, *m/z*: 132 [M]⁺.

Non-1-ene (5c). B.p. 58–59 °C (30 Torr). IR, ν/cm⁻¹: 2990, 2950, 2850, 1630, 1470, 910. ¹H NMR, δ: 0.81–0.93 (m, 3 H, Me); 1.07–1.46 (m, 10 H, CH₂); 1.81–2.17 (m, 2 H, CH₂–CH=CH₂); 4.86–5.11 (m, 2 H, CH₂=); 5.57–6.04 (m, 1 H, CH=CH₂). ¹³C NMR, δ: 14.22 (q, C(9)); 22.87 (t, C(8)); 29.17, 29.37, 29.63 (t, C(4), C(5), C(6)); 32.10 (t, C(7)); 34.05 (t, C(3)); 114.17 (t, C(1)); 139.82 (d, C(2)). MS, *m/z*: 126 [M]⁺.

Undec-1-ene (5d). B.p. 77–78 °C (15 Torr). ¹H NMR, δ: 0.77–0.89 (m, 3 H, Me); 1.07–1.46 (m, 14 H, CH₂); 1.80–2.17 (m, 2 H, CH₂–CH=CH₂); 4.83–5.11 (m, 2 H, CH₂=); 5.35–5.58 (m, 1 H, –CH=CH₂). ¹³C NMR, δ: 14.09 (q, C(11)); 22.74 (t, C(10)); 28.98; 29.17; 29.37 (2 C); 29.56 (t, C(4), C(5), C(6), C(7), C(8)); 31.97 (t, C(9)); 33.26 (t, C(3)); 114.04 (t, C(1)); 139.27 (d, C(2)). MS, *m/z*: 154 [M]⁺.

Dodec-1-ene (5e). B.p. 90–91 °C (10 Torr). ¹H NMR, δ: 0.83–0.96 (m, 3 H, Me); 1.09–1.46 (m, 16 H, CH₂); 1.81–2.17 (m, 2 H, CH₂–CH=CH₂); 4.86–5.07 (m, 2 H, CH₂=); 5.33–5.58 (m, 1 H, –CH=CH₂). ¹³C NMR, δ: 14.22 (q, C(12)); 22.87 (t, C(11)); 29.17; 29.37; 29.56; 29.89 (t, 3 C, C(4), C(5), C(6), C(7), C(8), C(9)); 32.17 (t, C(10)); 34.05 (t, C(3)); 114.17 (t, C(1)); 139.22 (t, C(2)). MS, *m/z*: 168 [M]⁺.

((Z)-Prop-1-enyl)benzene (6a). B.p. 67–68 °C (15 Torr). ¹H NMR, δ: 1.58 (d, 3 H, Me, *J* = 7 Hz); 5.33–5.73 (m, 1 H, MeCH=); 6.40 (d, 1 H, =CHPh, ³*J*_{H,H} = 12.0 Hz); 6.98–7.30

(m, 5 H, Ph). ¹³C NMR, δ: 14.00 (q, C(3')); 125.31 (d, C(2')); 126.03 (d, C(4)); 128.57 (d, C(2), C(6)); 129.37 (d, C(3), C(5)); 131.04 (t, C(1')); 139.95 (s, C(1)). MS, *m/z*: 118 [M]⁺.

((Z)-But-2-enyl)benzene (6b). B.p. 64–65 °C (10 Torr). ¹H NMR, δ: 1.58 (d, 3 H, =CMe, *J* = 7 Hz); 3.36 (d, 2 H, CH₂Ph, *J* = 7 Hz); 5.27–5.98 (m, 2 H, CH=CH); 7.02–7.30 (m, 5 H, Ph). ¹³C NMR, δ: 13.99 (q, C(4')); 39.20 (t, C(1')); 122.39 (d, C(3')); 125.90 (d, C(4)); 128.38 (d, C(3), C(5)); 128.51 (d, C(2), C(6)); 129.74 (t, C(2')); 141.81 (s, C(1)). MS, *m/z*: 132 [M]⁺.

Non-2-ene (Z(E)) (6c). B.p. 66–67 °C (40 Torr). ¹H NMR, δ: 0.82–0.94 (m, 3 H, Me); 1.20–1.52 (m, 8 H, CH₂); 1.56–1.68 (m, 3 H, =CMe); 1.77–2.13 (m, 2 H, –CH₂–C=); 5.21–5.58 (m, 2 H, –HC=CH–). ¹³C NMR, δ: 12.66 (17.86) (q, C(1)); 14.15 (q, C(9)); 22.80 (t, C(8)); 27.03 (32.82) (t, C(4)); 29.17, 29.37 (t, C(5), C(6)); 31.90 (t, C(7)); 123.60 (124.57) (d, C(2)); 130.95 (131.70) (d, C(3)). MS, *m/z*: 126 [M]⁺.

Undec-2-ene (Z(E)) (6d). B.p. 83–84 °C (20 Torr). ¹H NMR, δ: 0.82–0.96 (m, 3 H, Me); 1.04–1.28 (m, 12 H, CH₂); 1.56–1.68 (m, 3 H, =CMe); 1.77–2.13 (m, 2 H, –CH₂–CH=); 5.27–5.55 (m, 2 H, –HC=CH–). ¹³C NMR, δ: 12.86 (17.96) (q, C(1)); 22.87 (t, C(10)); 27.03 (33.82) (t, C(4)); 28.46, 29.17, 29.56, 29.82 (t, C(5), C(6), C(7), C(8)); 32.17 (t, C(9)); 123.53 (124.51) (d, C(2)); 130.88 (131.73) (d, C(3)). MS, *m/z*: 154 [M]⁺.

Dodec-2-ene (Z(E)) (6e). B.p. 94–95 °C (5 Torr). ¹H NMR, δ: 0.82–0.96 (m, 3 H, Me); 1.09–1.28 (m, 14 H, CH₂); 1.56–1.68 (m, 3 H, =CMe); 1.74–2.13 (m, 2 H, –CH₂–CH=); 5.27–5.58 (m, 2 H, –HC=CH–). ¹³C NMR, δ: 12.82 (18.02) (q, C(1)); 14.22 (t, C(12)); 22.87 (t, C(11)); 29.17, 29.37, 29.56, 29.89 (2 s, C(5), C(6), C(7), C(8), C(9)); 32.82 (t, C(10)); 123.96 (124.87) (d, C(2)); 130.95 (131.79) (d, C(3)). MS, *m/z*: 168 [M]⁺.

(1Z,5Z)-1,6-Diphenylhexa-1,5-diene (7a). B.p. 180–182 °C (5 Torr). Found (%): C, 92.03; H, 7.20. C₁₈H₁₈. Calculated (%): C, 92.30; H, 7.70. IR, ν/cm⁻¹: 3050, 3020, 2920, 2250, 1600, 1500, 1490, 1450, 1080, 1030, 1000, 910, 720, 700. ¹H NMR, δ: 2.20–2.45 (m, 4 H, CH₂–C=C); 5.45–5.73 (m, 2 H, CH₂–CH=); 6.39 (d, 2 H, =CH–Ph, ³*J*_{H,H} = 11.7 Hz); 7.04–7.43 (m, 10 H, Ph). ¹³C NMR, δ: 28.72 (t, C(3), C(4)); 126.59 (d, C_o Ph); 128.15 (d, C_o Ph); 128.73 (d, C_m Ph); 129.45 (d, C(1), C(6)); 131.47 (d, C(2), C(5)); 137.51 (s, C_{ipso} Ph). MS, *m/z*: 234 [M]⁺.

(2Z,6Z)-1,8-Diphenylocta-2,6-diene (7b). B.p. 182–183 °C (2 Torr). Found (%): C, 91.06; H, 8.25. C₂₀H₂₂. Calculated (%): C, 91.60; H, 8.40. IR, ν/cm⁻¹: 3050, 3020, 2900, 2800, 1600, 1450, 900, 720, 700. ¹H NMR, δ: 2.20–2.55 (m, 4 H, CH₂–C=C); 3.36 (d, 4 H, Ph–CH₂–C=, *J* = 7 Hz); 5.45–5.80 (m, 2 H, CH₂–CH=); 7.00–7.40 (m, 10 H, Ph). ¹³C NMR, δ: 28.93 (t, C(4), C(5)); 34.49 (t, C(1), C(8)); 126.62 (d, C_p Ph); 128.51 (d, C_o Ph); 129.10 (d, C_m Ph); 129.42 (d, C(2), C(7)); 130.75 (d, C(3), C(6)); 140.33 (s, C_{ipso} Ph). MS, *m/z*: 262 [M]⁺.

(7Z,11Z)-Octadeca-7,11-diene (7c). B.p. 137–139 °C (2 Torr). Found (%): C, 85.90; H, 13.42. C₁₈H₃₄. Calculated (%): C, 86.40; H, 13.60. IR, ν/cm⁻¹: 3000, 2950, 2900, 2820, 1640, 1450, 900, 730. ¹H NMR, δ: 0.76–0.90 (m, 6 H, Me); 1.06–1.43 (m, 16 H, CH₂); 1.90–2.33 (m, 8 H, CH₂–C=C); 5.28–5.38 (m, 4 H, CH=CH). ¹³C NMR, δ: 14.22 (q, C(1), C(18)); 22.80 (t, C(2), C(17)); 27.35 (t, C(6), C(13)); 27.55 (t, C(9), C(10)); 29.43, 29.63 (t, C(4), C(5), C(14), C(15)); 32.04 (t, C(3),

C(16)); 129.25 (d, C(8), C(11)); 130.49 (d, C(7), C(12)). MS, m/z : 250 $[M]^+$.

(9Z,13Z)-Docosa-9,13-diene (7d). B.p. 189–191 °C (2 Torr). Found (%): C, 86.12; H, 13.50. $C_{22}H_{42}$. Calculated (%): C, 86.27; H, 13.73. IR, ν/cm^{-1} : 3000, 2950, 2850, 1640, 1450, 1380, 970, 810, 790, 720. 1H NMR, δ : 0.76–0.89 (m, 6 H, Me); 1.10–1.45 (m, 24 H, CH_2); 1.95–2.12 (m, 8 H, $CH_2-C=C$); 5.22–5.35 (m, 4 H, $CH=CH$). ^{13}C NMR, δ : 14.22 (q, C(1), C(22)); 22.80 (t, C(2), C(21)); 27.35 (t, C(8), C(15)); 27.55 (t, C(11), C(12)); 29.43 (2 C); 29.63, 29.89 (t, C(4), C(5), C(6), C(7), C(16), C(17), C(18), C(19)); 32.04 (t, C(3), C(20)); 129.25 (d, C(10), C(13)); 130.49 (d, C(9), C(14)). MS, m/z : 306 $[M]^+$.

(10Z,14Z)-Tetracos-10,14-diene (7e). B.p. 217–218 °C (2 Torr). Found (%): C, 85.96; H, 13.60. $C_{24}H_{46}$. Calculated (%): C, 86.23; H, 13.77. IR, ν/cm^{-1} : 3000, 2950, 2920, 2850, 1640, 1480, 1380, 970, 810, 790, 720. 1H NMR, δ : 0.81–0.94 (m, 6 H, Me); 1.19–1.28 (m, 28 H, CH_2); 1.99–2.11 (m, 8 H, $CH_2-C=C$); 5.32–5.44 (m, 4 H, $CH=CH$). ^{13}C NMR, δ : 14.22 (q, C(1), C(24)); 22.80 (t, C(2), C(23)); 27.35 (t, C(9), C(16)); 27.55 (t, C(12), C(13)); 29.50 (2 C); 29.70 (3 C) (t, C(4), C(5), C(6), C(7), C(8), C(17), C(18), C(19), C(20), C(21)); 32.04 (t, C(3), C(22)); 129.25 (d, C(11), C(14)); 130.49 (d, C(10), C(15)). MS, m/z : 334 $[M]^+$.

(1,2-Dideuterioallyl)benzene (8a). B.p. 81–82 °C (30 Torr). Found (%): C, 89.62; H, D, 9.87. $C_9H_8D_2$. Calculated (%): C, 90.00; H, 6.67, D, 3.33. IR, ν/cm^{-1} : 3070, 3050, 3010, 2910, 2170 (C–D), 1600, 1490, 1450, 1020, 920, 700. 1H NMR, δ : 3.35 (s, H, Ph–CHD); 5.04 (s, 2 H, $CH_2=CD$); 6.38–7.27 (m, 5 H, Ph). ^{13}C NMR, δ : 39.80 (d, C(1'), $J_{C,D}$ = 19.8 Hz); 115.63 (t, C(3')); 126.03 (d, C(4)); 128.24 (d, C(2), C(6)); 128.37 (d, C(3), C(5)); 137.12 (s, C(2'), $J_{C,D}$ = 23.4 Hz); 143.91 (s, C(1)). MS, m/z : 120 $[M]^+$.

(2,3-Dideuteriobut-3-enyl)benzene (8b). B.p. 70–71 °C (15 Torr). Found (%): C, 89.16; H, D, 9.90. $C_{10}H_{10}D_2$. Calculated (%): C, 89.55; H, 7.46; D, 2.99. IR, ν/cm^{-1} : 3075, 3020, 2920, 2840, 2170 (C–D), 1600, 1490, 1450, 1020, 910, 740, 700. 1H NMR, δ : 2.25–2.46 (m, 1 H, CHD–C=C); 3.23–3.44 (m, 2 H, CH_2 –Ph); 5.03 (s, 2 H, $CH_2=CD$); 6.94–7.25 (m, 5 H, Ph). ^{13}C NMR, δ : 35.03 (d, C(2'), $J_{C,D}$ = 19.8 Hz); 38.99 (t, C(1')); 114.62 (t, C(4')); C(3')*; 125.68 (d, C(4)); 128.06 (d, C(3), C(5)); 128.21 (d, C(2), C(6)); 138.03 (s, C(1)). MS, m/z : 134 $[M]^+$.

2,3-Dideuterionon-1-ene (8c). B.p. 52–54 °C (20 Torr). Found (%): C, 83.92; H, D, 15.20. $C_9H_{16}D_2$. Calculated (%): C, 84.37; H, 12.50; D, 3.13. IR, ν/cm^{-1} : 3070, 2920, 2830, 2160 (C–D), 1460, 910, 800, 740. 1H NMR, δ : 0.76–0.96 (m, 3 H, Me); 1.29–1.76 (m, 10 H, CH_2); 1.96–2.05 (m, 1 H, CHD–C=C); 4.95 (s, 2 H, $CH_2=CD$). ^{13}C NMR, δ : 14.15 (q, C(9)); 22.87 (t, C(8)); 29.34; 29.59; 29.76 (t, C(4), C(5), C(6)); 32.10 (t, C(7)); 34.05 (d, C(3), $J_{C,D}$ = 20.0 Hz); 115.00 (t, C(1)), 139.20 (s, C(2), $J_{C,D}$ = 23.0 Hz). MS, m/z : 128 $[M]^+$.

2,3-Dideuterioundec-1-ene (8d). B.p. 72–73 °C (10 Torr). Found (%): C, 84.15; H, D, 15.12. $C_{11}H_{20}D_2$. Calculated (%): C, 84.62; H, 12.82; D, 2.56. IR, ν/cm^{-1} : 3070, 2950, 2910, 2850, 2170 (C–D), 1620, 1450, 900, 810, 750. 1H NMR, δ : 0.76–0.96 (m, 3 H, Me); 1.24–1.59 (m, 14 H, CH_2); 1.94–2.16 (m, 1 H, CHD–C=C); 4.93 (s, 2 H, $CH_2=CD$). ^{13}C NMR, δ : 14.11 (q, C(11)); 22.76 (t, C(10)); 28.94; 29.26; 29.46; 29.65 (2 C) (t, C(4), C(5), C(6), C(7), C(8)); 32.06 (t, C(9)); 32.91

(d, C(3), $J_{C,D}$ = 20.5 Hz); 114.00 (t, C(1)); 138.84 (s, C(2), $J_{C,D}$ = 23.4 Hz). MS, m/z : 156 $[M]^+$.

2,3-Dideuteriododec-1-ene (8e). B.p. 72–73 °C (2 Torr). Found (%): C, 84.22; H, D, 15.06. $C_{12}H_{22}D_2$. Calculated (%): C, 84.70; H, 12.94; D, 2.36. IR, ν/cm^{-1} : 3065, 2950, 2900, 2800, 2170 (C–D), 1480, 900, 820, 750. 1H NMR, δ : 0.76–0.96 (m, 3 H, Me); 1.20–1.44 (m, 16 H, CH_2); 1.94–2.14 (m, 1 H, CHD–C=C); 4.89 (s, 2 H, $CH_2=CD$). ^{13}C NMR, δ : 14.22 (q, C(12)); 22.80 (t, C(11)); 28.98; 29.30; 29.50; 29.69; 29.76 (2 C) (t, C(4), C(5), C(6), C(7), C(8), C(9)); 32.04 (t, C(10)); 33.47 (t, C(3), $J_{C,D}$ = 20.0 Hz); 114.04 (t, C(1)); 138.96 (s, C(2), $J_{C,D}$ = 23.0 Hz). MS, m/z : 170 $[M]^+$.

(2,3-Dideuterio-(Z)-prop-1-enyl)benzene (9a). B.p. 73–74 °C (20 Torr). Found (%): C, 89.32; H, D, 9.66. $C_9H_8D_2$. Calculated (%): C, 90.00; H, 6.67; D, 3.33. IR, ν/cm^{-1} : 3020, 2950, 2850, 2160 (C–D), 1650, 1450, 940, 850. 1H NMR, δ : 1.58 (br.s, 2 H, CH_2D); 6.39 (br.s, 1 H, –CH=CD); 7.00–7.27 (m, 5 H, Ph). ^{13}C NMR, δ : 13.89 (t, C(3'), $J_{C,D}$ = 19.0 Hz); C(2')*; 126.00 (d, C(4)); 128.40 (d, C(2), C(6)); 129.76 (d, C(3), C(5)); 130.20 (d, C(1')); 139.95 (s, C(1)). MS, m/z : 120 $[M]^+$.

(3,4-Dideuterio-(Z)-but-2-enyl)benzene (9b). B.p. 72–73 °C (15 Torr). Found (%): C, 89.14; H, D, 10.27. $C_{10}H_{10}D_2$. Calculated (%): C, 89.55; H, 7.46; D, 2.99. IR, ν/cm^{-1} : 3080, 2920, 2160 (C–D), 1650, 1460, 900, 740, 700. 1H NMR, δ : 1.60 (br.s, 2 H, CH_2D); 3.36 (d, 2 H, –CH₂–Ph, J = 7 Hz); 5.42 (t, 1 H, CH=CD, J = 7 Hz); 7.00–7.25 (m, 5 H, Ph). ^{13}C NMR, δ : 13.89 (t, C(4'), $J_{C,D}$ = 20.4 Hz); C(3')*; 39.25 (t, C(1')); 125.90 (d, C(4)); 128.35 (d, C(3), C(5)); 128.52 (d, C(2), C(6)); 130.00 (d, C(2')); 141.81 (s, C(1)). MS, m/z : 134 $[M]^+$.

1,2-Dideuterionon-2-ene (Z(E)-9c). B.p. 67–68 °C (40 Torr). Found (%): C, 83.88; H, D, 15.36. $C_9H_{16}D_2$. Calculated (%): C, 84.37; H, 12.50; D, 3.13. IR, ν/cm^{-1} : 3030, 2950, 2850, 2170 (C–D), 1640, 1450, 900, 850, 720. 1H NMR, δ : 0.82–0.94 (m, 3 H, Me); 1.04–1.38 (m, 8 H, CH_2); 1.56 and 1.68 (both s, 2 H, $CH_2D-C=C$); 1.86–2.34 (m, 2 H, $CH_2-C=C$); 5.21–5.52 (m, 1 H, CH=CD). ^{13}C NMR, δ : 12.48 (17.99) (t, C(1), $J_{C,D}$ = 19.5 Hz); 14.15 (q, C(9)); 23.06 (t, C(8)); 26.96 (33.99) (t, C(4)); 29.63, 29.80 (t, C(5), C(6)); 32.40 (t, C(7)); 123.53 (124.51) (s, C(2), $J_{C,D}$ = 22.4 Hz); 130.88 (131.73) (t, C(3)). MS, m/z : 128 $[M]^+$.

1,2-Dideuterioundec-2-ene (Z(E)-9d). B.p. 83–84 °C (20 Torr). Found (%): C, 84.17; H, D, 15.14. $C_{11}H_{20}D_2$. Calculated (%): C, 84.62; H, 12.82; D, 2.56. IR, ν/cm^{-1} : 3020, 2900, 2850, 2170 (C–D), 1620, 1450, 900, 720. 1H NMR, δ : 0.82 (m, 3 H, Me); 1.04–1.36 (m, 12 H, CH_2); 1.56 and 1.61 (both s, 2 H, $CH_2D-C=C$); 1.92–2.12 (m, 2 H, $CH_2-C=C$); 5.25–5.56 (m, 1 H, CH=CD). ^{13}C NMR, δ : 12.76 (17.87) (t, C(1), $J_{C,D}$ = 19.0 Hz); 14.22 (q, C(11)); 22.80 (t, C(10)); 26.96 (33.83) (t, C(4)); 29.17, 29.55, 29.75, 29.86 (t, C(5), C(6), C(7), C(8)); 32.04 (t, C(9)); 123.66 (124.57) (s, C(2), $J_{C,D}$ = 22.4 Hz); 131.01 (131.79) (d, C(3)). MS, m/z : 156 $[M]^+$.

1,2-Dideuteriododec-2-ene (Z(E)-9e). B.p. 94–95 °C (5 Torr). Found (%): C, 84.37; H, D, 15.06. $C_{12}H_{22}D_2$. Calculated (%): C, 84.70; H, 12.94; D, 2.36. IR, ν/cm^{-1} : 3025, 2950, 2900, 2850, 2170 (C–D), 1620, 1450, 920, 710. 1H NMR, δ : 0.82–0.94 (m, 3 H, Me); 1.05–1.41 (m, 14 H, CH_2); 1.56 and 1.61 (both s, 2 H, $CH_2D-C=C$); 1.92–2.16 (m, 2 H, $CH_2-C=C$); 5.32–5.65 (m, 1 H, CD=CH). ^{13}C NMR, δ : 12.75 (17.99) (t, C(1), $J_{C,D}$ = 20.5 Hz); 14.15 (q, C(12)); 22.80

* The signal is not observed.

* The signal is not observed.

(t, C(11)); 27.45 (t, C(9)); 27.03 (34.05) (t, C(4)); 29.40, 29.68, 29.76, 29.80 (2 C) (t, C(5), C(6), C(7), C(8)); 32.04 (t, C(10)); 123.66 (124.57) (s, C(2), $J_{C,D} = 22.4$ Hz); 130.95 (131.79) (d, C(3)). MS, m/z : 170 $[M]^+$.

2,5-Dideuterio-(1Z,5Z)-1,6-diphenylhexa-1,5-diene (10a). B.p. 180–182 °C (5 Torr). Found (%): C, 91.16; H, D, 8.26. $C_{18}H_{16}D_2$. Calculated (%): C, 91.53; H, 6.77; D, 1.70. IR, ν/cm^{-1} : 3030, 2900, 2800, 2165 (C–D), 1600, 1500, 1450, 1080, 900, 720, 700. 1H NMR, δ : 2.42 (br.s, 4 H, $CH_2=C=C$); 6.40 (br.s, 2 H, $C=CH-Ph$); 7.04–7.55 (m, 10 H, Ph). ^{13}C NMR, δ : 28.72 (t, C(3), C(4)); 126.59 (d, C_p Ph); 128.15 (d, C_o Ph); 128.73 (d, C_m Ph); 129.45 (d, C(1), C(6)); 131.47 (s, C(2), C(5), $J = 23.4$ Hz); 137.51 (s, C_{ipso} Ph). MS, m/z : 236 $[M]^+$.

3,6-Dideuterio-(2Z,6Z)-1,8-diphenylocta-2,6-diene (10b). B.p. 182–183 °C (2 Torr). Found (%): C, 90.56; H, D, 8.94. $C_{20}H_{20}D_2$. Calculated (%): C, 90.90; H, 7.58; D, 1.52. IR, ν/cm^{-1} : 3050, 3020, 2900, 2800, 2160 (C–D), 1600, 1500, 1450, 900, 720, 700. 1H NMR, δ : 2.45 (br.s, 4 H, $CH_2=C=C$); 3.36 (d, 4 H, $Ph-CH_2-$, $J = 7$ Hz); 5.43 (t, 2 H, $CD=CH$, $J = 7$ Hz); 7.00–7.35 (m, 10 H, Ph). ^{13}C NMR, δ : 28.90 (t, C(4), C(5)); 34.42 (t, C(1), C(8)); 126.60 (d, C_p Ph); 128.52 (d, C_o Ph); 129.06 (d, C_m Ph); 129.40 (d, C(2), C(8)); 130.40 (s, C(3), C(6), $J_{C,D} = 22.5$ Hz); 140.38 (s, C_{ipso} Ph). MS, m/z : 264 $[M]^+$.

8,11-Dideuterio-(7Z,11Z)-octadeca-7,11-diene (10c). B.p. 137–138 °C (2 Torr). Found (%): C, 85.33; H, D, 14.08. $C_{18}H_{32}D_2$. Calculated (%): C, 85.71; H, 12.70; D, 1.59. IR, ν/cm^{-1} : 3040, 3020, 2900, 2850, 2170 (C–D), 1640, 1450, 900, 720, 700. 1H NMR, δ : 0.82–0.95 (m, 6 H, Me); 1.07–1.46 (m, 16 H, CH_2); 1.81–2.07 (m, 8 H, $CH_2=C=C$); 5.38 (t, 2 H, $CD=CH$, $J = 7$ Hz). ^{13}C NMR, δ : 14.22 (q, C(1), C(18)); 22.80 (t, C(2), C(17)); 27.42 (t, C(6), C(9), C(10), C(13)); 29.43, 29.63 (t, C(4), C(5), C(14), C(15)); 31.90 (t, C(3), C(16)); C(8)*; C(11)*; 130.30 (d, C(7), C(12)). MS, m/z : 252 $[M]^+$.

10,13-Dideuterio-(9Z,13Z)-docosa-9,13-diene (10d). B.p. 190–192 °C (2 Torr). Found (%): C, 85.41; H, D, 13.96. $C_{22}H_{40}D_2$. Calculated (%): C, 85.71; H, 12.99; D, 1.30. IR, ν/cm^{-1} : 3040, 3025, 2900, 2850, 2170 (C–D), 1640, 1450, 900, 720, 700. 1H NMR, δ : 0.82–0.95 (m, 6 H, Me); 1.08–1.51 (m, 24 H, CH_2); 1.89–2.07 (m, 8 H, $CH_2=C=C$); 5.36 (t, 2 H, $CD=CH$, $J = 7$ Hz). ^{13}C NMR, δ : 14.15 (q, C(1), C(22)); 22.87 (t, C(2), C(21)); 27.41 (t, C(8), C(11), C(12), C(15)); 29.56 (4 C); 29.76 (4 C) (t, C(4), C(5), C(6), C(7), C(16), C(17), C(18), C(19)); 32.17 (t, C(3), C(20)); C(10)*; C(13)*; 130.17 (d, C(9), C(14)). MS, m/z : 308 $[M]^+$.

11,14-Dideuterio-(10Z,14Z)-tetracos-10,14-diene (10e). B.p. 218–219 °C (2 Torr). Found (%): C, 85.29; H, D, 14.06. $C_{24}H_{44}D_2$. Calculated (%): C, 85.71; H, 13.09; D, 1.20. IR, ν/cm^{-1} : 3035, 2900, 2850, 2170 (C–D), 1620, 1450, 970, 720, 700. 1H NMR, δ : 0.82–0.94 (m, 6 H, Me); 1.08–1.52 (m, 28 H, CH_2); 1.90–2.07 (m, 8 H, $CH_2=C=C$); 5.36 (t, 2 H, $CD=CH$, $J = 7$ Hz). ^{13}C NMR, δ : 14.15 (q, C(1), C(24)); 22.80

(t, C(2), C(23)); 27.42 (t, C(9), C(12), C(13), C(16)); 29.56 (2 C); 29.76 (3 C) (t, C(4), C(5), C(6), C(7), C(8), C(17), C(18), C(19), C(20), C(21)); 32.04 (t, C(3), C(22)); 130.40 (d, C(10), C(15)); C(11)*; C(14)*. MS, m/z : 236 $[M]^+$.

References

1. S. V. Rusakov, L. M. Khalilov, L. V. Parfenova, A. G. Ibragimov, O. A. Ponomarev, and U. M. Dzhemilev, *Izv. Akad. Nauk, Ser. Khim.*, 2001, 2229 [*Russ. Chem. Bull., Int. Ed.*, 2001, **50**, 2336].
2. L. O. Khafizova, A. G. Ibragimov, G. N. Gil'fanova, L. M. Khalilov, and U. M. Dzhemilev, *Izv. Akad. Nauk, Ser. Khim.*, 2001, 2089 [*Russ. Chem. Bull., Int. Ed.*, 2001, **50**, 2188].
3. U. M. Dzhemilev, A. G. Ibragimov, L. O. Khafizova, S. V. Rusakov, and L. M. Khalilov, *Mendeleev Commun.*, 1997, 198.
4. U. M. Dzhemilev and A. G. Ibragimov, *J. Organometal. Chem.*, 1994, **466**, 1.
5. U. M. Dzhemilev and A. G. Ibragimov, *Izv. Akad. Nauk, Ser. Khim.*, 1998, 816 [*Russ. Chem. Bull.*, 1998, **47**, 786 (Engl. Transl.)].
6. U. M. Dzhemilev and A. G. Ibragimov, *Usp. Khim.*, 2000, **69**, 134 [*Russ. Chem. Rev.*, 2000, **69**, 121 (Engl. Transl.)].
7. A. J. Gordon and R. A. Ford, *The Chemist's Companion*, J. Wiley and Sons, New York—London—Sydney—Toronto, 1972.
8. E. Breit-Maier and W. Voelter, *^{13}C NMR Spectroscopy*, Verlag Chemie GmbH, Weinheim, 1974, 276.
9. A. G. Ibragimov, L. O. Khafizova, K. G. Satenov, L. M. Khalilov, L. G. Yakovleva, S. V. Rusakov, and U. M. Dzhemilev, *Izv. Akad. Nauk, Ser. Khim.*, 1999, 1594 [*Russ. Chem. Bull.*, 1999, **48**, 1574 (Engl. Transl.)].
10. P. Sobota, T. Pluzinski, B. Jezowska-Trzebiatowska, and S. Rummel, *J. Organometal. Chem.*, 1980, **185**, 69.
11. A. G. Ibragimov, L. O. Khafizova, I. V. Zagrebel'naya, L. V. Parfenova, R. M. Sultanov, L. M. Khalilov, and U. M. Dzhemilev, *Izv. Akad. Nauk, Ser. Khim.*, 2001, 280 [*Russ. Chem. Bull., Int. Ed.*, 2001, **50**, 292].
12. F. Sato, K. Oguro, and M. Sato, *Chem. Lett.*, 1978, 805.
13. A. N. Nesmeyanov, T. V. Nikitina, and O. V. Nogina, *Metody elementoorganicheskoi khimii. Podgruppy aktinoidy* [Methods of Organometallic Chemistry. Actinide Subgroup], Nauka, Moscow, 1974, 335 pp. (in Russian).
14. G. Wilkinson and J. M. Birmingham, *J. Am. Chem. Soc.*, 1954, **76**, 4281.
15. R. Kh. Freidlina, E. M. Brainina, and A. N. Nesmeyanov, *Dokl. Akad. Nauk SSSR*, 1961, **138**, 1369 [*Dokl. Chem.*, 1969 (Engl. Transl.)].

* The signal is not observed.