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Design, synthesis and algicides activities of thiourea derivatives as the Novel Scaffold Aldolase Inhibitors

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ABSTRACT

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Keywords: Aldolases Fragment-based Virtual Screen thiourea algicides activities De novo Discovery By using a new Fragment-Based Virtual Screen strategy, two series of novel FBA-II inhibitors (thiourea derivatives) were *de novo* discovered based on the active site of fructose-1, 6-bisphosphate aldolase from Cyanobacterial (CyFBA). In comparison, most of the *N*-(2-benzoylhydrazine-1-carbonothioyl) benzamide derivatives (**L14-L22**) exhibit higher CyFBA-II inhibitory activities compared to *N*-(phenylcarbamothioyl) benzamide derivatives (**L14-L13**). Especially, compound **L14** not only shows higher CyFBA-II activity ($K_i = 0.65 \mu$ M), but also exhibits most potent *in vivo* activity against Synechocystis sp. PCC 6803 (EC₅₀=0.09 ppm), higher (7-fold) than that of our previous inhibitor (EC₅₀=0.6 ppm). The binding modes of compound **L14** and CyFBA-II were further elucidated by jointly using DOX computational protocol, MM-PBSA and site-directed mutagenesis assays. The positive results suggest that strategy adopted in this study was promising to rapidly discovery the potent inhibitors with novel scaffolds. The satisfactory algicide activities suggest that the thiourea derivatives is very likely to be a promising lead for the development of novel specific algicides to solve Cyanobacterial harmful algal blooms (CHABs).

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1. Introduction

Cyanobacterial harmful algal blooms (CHABs) have been increasing in frequency all over the world.[1] Toxin-producing CHABs have directly threatened to the human health, living resources, and water economies.[2-5] To control these CHABs, many approaches included mechanical, physical/chemical, and biological controls have been used so far.[6] In comparison, chemical control was the most effective method. Chemical control involves the use of compounds (copper compounds, chemical oxidants, and herbicides).[1, 7] Especially, the compounds designed for the specific target of cyanobacteria provide us a useful pathway to the development of potential algaecides. Previously, using the systematically structure-centric approach, several novel inhibitors of class II fructose-1, 6-bisphosphate aldolase from Cyanobacteria (CyFBA-II) were designed and synthesized in our group.[8, 9] In contrast, the N-(2,5-Dioxo-2,5-dihydro-1H-pyrrol-1-yl) benzamide derivatives exhibit higher inhibitory activities against Synechocystis sp. PCC 6803.[9] However, the instability of those compounds limits their further applications. To address this concern, two series of novel scaffold CyFBA-II inhibitors (thiourea derivatives) were *de novo* designed by using Fragment-Based Virtual Screen (FBVS) strategy developed in our group based on the substrate site of CyFBA-II. The proposed binding model of novel inhibitors targeting into CyFBA-II were systematically identified by integrating Docking-ONIOM-XO (DOX)[10] computational protocol, Molecular Mechanics combined with the Poisson–Boltzmann and Surface Area continuum solvation (MM-PBSA),[11] chemical synthesis, and site-directed mutant assays.

2. Results and Discussion

2.1. Design of thioureas derivatives as novel CyFBA-II inhibitors

In previous work, we found that *N*-(2,5-Dioxo-2,5-dihydro-1*H*-pyrrol-1-yl) benzamide derivatives shown potent inhibition against CyFBA-II.[8] However, these compounds appears to be unstable due to the hydrazone moiety. Recently, *de novo* design (i.e. FBVS) are well-developed and mature technique for lead discovery as well as lead optimization.[12, 13] Some *de novo* design computational methods have already been implemented in commercial or academic software, such as LUDI,[14] EA-inventor,[15] SPROUT,[16] BUILDER,[17] PROLIGANDS,[18] LigBuilder[19] and LEA3D,[20] especially, many protocols have successfully applied to the discovery of highly potent inhibitors.[21-23] However, most of the software mentioned-above were performed based on their specific docking algorithm and scoring function. Our previous studies[8, 9, 24] suggested that Surflex-Dock,[25] evaluated by using an empirical function ChemScore, was suitable for the discovery of novel FBA inhibitors, thus Surflex-Dock were used in the present FBVS protocol to generate and score the conformations of fragments and resulting compounds based on the CyFBA-II active pocket. To design the algaecides with desirable druglike properties, several rings and frameworks in FDA-approved drugs[26] were used as the fragments dataset (**Figure S1**) for this FBVS protocol (**Figure 1**).



Figure 1. Fragment-Based Virtual Screen (FBVS) protocol

The 3D-Conformation of CyFBA-II were constructed as our previous work.[8, 9] The Zinc ion (Zn^{2+}) were derived from the template structure (PDB ID: 3GAY). The Molecular Dynamics (MD) simulation of CyFBA-II and substrate (FBP) were performed prior to the docking, as illustrated in **Figure S2** and **Figure S3** in supporting information. In MD procedure, the charge of Zn ion is set as 2, the parameters of Zn ion is optimized by Pang group[27, 28] in Amber package. For the FBVS protocol, the fragments were docked firstly into the active pocket of CyFBA-II by using Surflex-Dock. Twenty poses were generated for each fragment. Then, the cluster analysis for each fragment were performed (RMSD < 1Å), and six conformations with high score (surflex-dock score/number of atoms > 0.15) will be selected for the connection between fragments. Subsequently, about 3512 molecules were generated by using Surflex-Dock, and the Xscore developed by Wang et al.,[29] were performed for all docking conformations. Then, top 500 compounds with high average score (Ave_score) of protein-ligand given by X-score were selected for the further Lead-like evaluation by using Percepta@ACDLabs software. At this step, about 174 reasonable compounds were remained. Finally, top 10 compounds with smaller Molecular Weight (MW) and solubility larger than 0.1 mg/mL were selected for our further analysis. Top 10 potent hit compounds generated by FBVS routing were listed in **Table 1**.



According to the total number of HDonors&HAcceptors and synthesis step number, compound 435, which was re-named L1, was finally selected to further investigation; The synthetic route were illustrated in Scheme 1, which has been documented by Rauf et al.[30]





Scheme 1. General synthetic route for compound L1~L13.

Fortunately, the inhibitory constant (K_i) of L1 against CyFBA-II was of 16.7 μ M (Table 2), this result indicates the FBVS route adopted in the present study seems to be reliable.

2.2. Structure-activity-relationship (SAR) analysis

To discover more potent CyFBA-II inhibitors, it is necessary to analyze the structure-activity-relationship (SAR). Firstly, we fixed R_2 , R_3 and R_4 to Hydrogen, and introduced the Cl, NO₂, and OH group into R_1 position (Table 2). However, no remarkable CyFBA-II inhibition (Inhibition Rate < 40%) were observed, such as compounds L2 ~ L4. In comparison, compound L4 exhibit higher inhibition rate (37%) than those of L2 and L3. This indicate that hydroxyl group seems to be favorable in R_1 position. Furthermore, R_1 , R_3 and R_4 were fixed to hydrogen, and R_2 was replaced by OH (L5), Br (L6) and NO₂ (L7), the results show that compound L7 have relatively high inhibition rate (60%) compared with those of compounds L5 (22%) and L6 (42%). Taken together, we believe that OH at R_1 and

NO₂ at R_2 position appears to be favorable. To explain this experiment observation, L11 was chosen as the representative compound, the binding model of L11 and CyFBA-II was predicted by using the DOX strategy, which has been documented previously.[10] The possible binding conformation of L11 and CyFBA-II was illustrated in Figure 2.

Table 2. Substitution patterns of compounds L1 - L13 and their inhibition activity against CyFBA-II

		$\begin{array}{c} R_1 \\ R_2 \\ R_3 \end{array} \xrightarrow{R_1} H \\ S \\ O \\ R_4 $					
	Compds	R ₁	R_2	<i>R</i> ₃	R ₄	<i>K</i> _i (μM)/ Inhibition Rate ^a	
	L1	ОН	NO ₂	Н	Н	16.7±0.4	
	L2	Cl	Н	Н	Н	30%ª	
	L3	NO_2	Н	Н	Н	15%ª	
	L4	ОН	Н	Н	Н	37%ª	
	L5	Н	ОН	Н	Н	22% ª	
	L6	Н	Br	Н	Н	42% ^a	
	L7	Н	NO ₂	Н	Н	60% ^a	
	L8	ОН	NO ₂	Н	CF ₃	5.5±0.10	
	L9	ОН	NO ₂	Н	CH ₃	5.5±0.20	
	L10	ОН	NO ₂	Н	Ph	2.5±0.10	
	L11	ОН	NO_2	Н	NO ₂	1.9±0.02	
C	L12	ОН	Н	NO ₂	Н	9.1±0.20	
1	L13	ОН	Н	NO_2	Cl	9.1±0.10	

As shown in Figure 2, the hydroxyl group and sulfur of L11 can coordinate with the Zn^{2+} . Moreover, the hydroxyl can also form hydrogen bond with residue N275. To verify the importance of residue N275, the site-directed mutagenesis was performed. As listed in **Table 3**, the inhibitory activity of L11 against N275A (37.9µM) decreased about 20-fold than parental CyFBA-II (1.9 µM), the possible reason for this case is that N275A results in the disappear of the hydrogen-bond between N275 and the hydroxyl moiety in benzene ring (Figure 2). Additionally, Figure 2 illustrate that the nitro group in R_2 position can form hydrogen bonds with residues K202 and T278. As expected, the inhibitions of L11 against T278A decreased about 46.3-fold than parental CyFBA-II (**Table 3**). Unfortunately, K202A mutant lacked detectable enzymatically activity, thus the interaction of compound and K202 could not be determined using mutagenesis experiment. Combing the DOX results (**Figure 2**), we believe that the interaction between nitro group in R_2 position and T278 is probably existed. Therefore, the compounds with hydroxyl group in R_1 and NO₂ group in R_2 position could exhibit high inhibitory activities against CyFBA-II.

Subsequently, we fixed R_1 and R_2 to OH and NO₂, and tried to vary the R_4 to CF₃ (L8), CH₃ (L9), Ph (L10) and NO₂ (L11). Compared with L1 (16.7 µM), L8 ~ L11 show higher inhibitions ($K_i = 1.9 \sim 5.5$ µM) against CyFBA-II. Especially, the L11 exhibit highest CyFBA-II inhibitions with K_i value of 1.9 µM. These findings reveal that the introduction of substituents in R_4 position, hydrophilic or hydrophobic group, might improve the CyFBA-II inhibitions of hit compounds. However, our DOX prediction (Figure 2) show that the nitro group in R_4 position locate outside of the active pocket, no remarkable interactions of nitro group in R_4 position could be observed. Therefore, the reason that compounds L8~L11 have higher CyFBA-II inhibitions is not clear so far.

In addition, one can notice from **Table 2** that **L12** exhibit higher inhibitory activity (9.1 μ M) compared with compound **L4** (37%), this indicate that the NO₂ in *R*₃ position seems to be favorable than H. The reason is likely that NO₂ in *R*₃ position could also form hydrogen-bond with T278.

One can notice from **Figure 2** that, the NH moiety in the middle linkage of L11 might form a hydrogen-bond with D277, and the carboxyl group of L11 may form hydrogen-bond with residues R281 and R302. However, the D277A mutant couldn't affect the inhibitory activity of L11 (2.4-fold), this is not consistent with the predicted interaction in **Figure 2**, maybe due to the flexibility of residue D277. In contrast, the mutants of R281A and R302A could result in 12.9 and 76-fold decrease of the L11 inhibitions (**Table 3**). In addition, the Van der Waals interaction between thiourea linkage and R281 could also be observed in **Figure 2**. These may be the partial reason that the inhibitions of L11 against R281A significantly decreased about 76-fold compared with parental CyFBA-II.



Figure 2. The predicted binding modes of compound L11 and CyFBA-II

2.3. Optimization of hit compounds.

Based upon the mutant data (**Table 3**), it is concluded that R281 is significantly important for the binding of L11 against CyFBA-II. However, due to the effect of the coordination between sulfur and Zn^{2+} , the hydrogen-bond distance of L11 and R281 is larger than 3 Å (**Figure 2**). Thus, we propose that if one amide group were interposed between hydroxyl and sulfur, the hit compounds will be closer to R281, which in turn improve the inhibitory activities of hit compounds against CyFBA-II. For this purpose, L14~L22 were further synthesized (**Table 4**), the synthetic route was illustrated in **Scheme 2**.

	L11(μM) K _i K _i (mutant)/K _i (WT) 1.9±0.02 1 3.9±0.20 2.1		L14	(μΜ)
	$K_{ m i}$	$K_{i}(mutant)/K_{i}(WT)$	$K_{ m i}$	$K_{\rm i}({\rm mutant})/K_{\rm i}({\rm WT})$
WT	1.9±0.02	1	0.7±0.02	1
S50A	3.9±0.20	2.1	15.2±0.30	23.4
R51A	3.3±0.20	1.7	3.9±0.50	6.0
855A	3.4±0.10	2.1	1.0±0.10	1.5
N275A	37.9±1.4	20	15.9±1.0	24.5
D277A	4.6±0.60	2.4	7.2±0.50	11
T278A	87.5±0.50	46.3	4.3±0.50	6.6
R281A	143.7±1.5	76.0	52.3±0.40	80.5
R302A	24.4±1.1	12.9	20.4±0.10	31

Table 3. The K_i values of the compound L11 and L14 against wildtype (WT) and mutants of CyFBA-II

As listed in **Table 4**, most of these compounds have high inhibitory activities against CyFBA-II ($K_i \leq 5.2 \,\mu$ M). When R_1 was replaced by hydroxyl group, the compounds (**L14~L16**) exhibit higher inhibitory activities against CyFBA-II ($K_i = 0.65 \sim 0.88 \mu$ M). Especially, compound **L14** exhibits highest inhibition with K_i of 0.65 μ M. These results indicate that the hydroxyl group in R_1 position are significantly important for the inhibitory activities of hit compounds. The possible binding modes (**Figure 3**) of compound **L14** and CyFBA-II were also predicted by using DOX strategy and MD simulation, the plots of MD were illustrated in **Figure S4**.



R₁=H, OH; R₂=H, Br, NO₂; R₃=H, OH, F, Br, NO₂; R₄=H, F, Cl

Scheme 2. General synthetic route for L14~L22

Similar to L11, the hydroxyl group in R_1 of L14 could form hydrogen bond with N275 (Figure 3). When N275 was mutated to Ala, the inhibitions of L14 against N275A decreased about 25-fold compared to WT. Therefore, we believe that the hydrogen-bond between N275 and L14 is essential for binding of L14 and CyFBA-II (Figure 3).

 Table 4. Substitution patterns of compounds L14 – L22 and their inhibition activity against CyFBA-II and Synechocystis sp. PCC 6803

R₂

		R ₃ H H H H H H H H H H H R ₄						
_	Compds	<i>R</i> ₁	<i>R</i> ₂	<i>R</i> ₃	<i>R</i> ₄	<i>K</i> _i (μM)	ЕС ₅₀ (ррт)	
-	L14	ОН	Н	Н	Н	0.65 ± 0.05	0.09 ± 0.02	
	L15	ОН	Н	Н	Cl	0.63±0.03	0.98 ± 0.34	
	L16	ОН	Н	Н	F	0.88±0.03	5.36±0.73	
	L17	ОН	Н	ОН	Н	1.25±0.16	6.4±0.80	
	L18	Н	NO ₂	Н	Н	1.87±0.09	0.39 ± 0.13	
	L19	Н	Br	Н	Н	3.1±0.17	5.9±1.1	
	L20	Н	Н	F	Н	3.5±0.71	8.2±0.60	
	L21	Н	Н	Br	Н	5.2±0.10	1.58±0.20	
	L22	Н	Н	NO ₂	Н	2.8±0.12	16.3±6.2	

Additionally, one carbonyl group in the left of the thiourea linkage could coordinate with Zn^{2+} , while the other carbonyl group could form coordination with Zn^{2+} via water molecule (**Figure 3**). The remarkable hydrogen bonds between thiourea linkage and surrounding residues (S50, R281 and R302) could be

observed. As expected, when the S50, R281 and R302 were mutated to Ala, the corresponding inhibitory activities of L14 decrease about 23.4, 80.5, 31-fold (Table 3). This is satisfactory consistent with our theoretical predictions that these three residues could form hydrogen bond with compound L14. Noticeably, the R281A mutant could result in the 80-fold decrease of L14 inhibitory activities (Figure 3), this is owning to the Van der Waals interaction between thiourea linkage and R281 and R302, which is similar to the binding-modes of L11 and CyFBA-II (Figure 2).

It should be noticed that the NO₂ moiety of L11 can form a hydrogen-bond with T278 (Figure 2), the K_i of L11 against T278A is significantly increased 46.3-fold (Table 3) compared to parental CyFBA-II. In contrast, no hydrogen-bond between L14 and D278 could be formed (Figure 3), thus the effect of D278A on the inhibitory activity of L14 is slightly, with the decrease of 6.6-fold (Table 3). In addition, our DOX&MD results show no interaction between L14 and residues R51 and S55. As expected, when S55, R51 were mutated, the corresponding K_i value decreased less than 10-fold. The well agreement between experimental and theoretical results show that our predicted interactions between L14 and CyFBA-II is reasonable.



Figure 3. The possible binding modes of L14 and CyFBA-II obtained by DOX strategy and molecular dynamic (MD) simulations

The R_4 substituent of hit compound located out of the active cavity, therefore, our substituent in R_4 position actually couldn't improve the inhibition activities of hit compounds, such as L14~L16.

Furthermore, the binding free energies of L14 and the mutants (N275A, R281A, S50A and R302A) were also calculated by using MMGBSA method based on the MD results. Our MMGBSA results (**Table 5**) show that the R281A and R302A could result in the remarkable decrease (6.29 and 6.21 kcal/mol) of $\Delta\Delta G_{bind}$, while R275A and S50A mutants could lead to relatively small decrease (5.99 and 3.44 kcal/mol) of $\Delta\Delta G_{bind}$. These are qualitatively agreement with the experimental K_i value. Taken together, we could believe that the binding modes of L14 and CyFBA-II predicted by DOX and MD were reasonable, and R281 maybe a critical pharmacophore to the binding of hit compounds.

2.4. Synechocystis Activity

The inhibition activities (EC₅₀) of compounds L14~L22 against Synechocystis sp. PCC 6803 were further determined (**Table 4**). As listed in **Table 4**, most compounds with high CyFBA-II inhibition activities generally exhibit potent Synechocystis activities. For example, the inhibition constant (K_i) of compounds L14 ~ L22 were less than 6 µM, correspondingly, the EC₅₀ of most of these compounds is less than 10 ppm. Especially, compound L14 shows highest CyFBA-II inhibition activity (0.65µM), and also exhibits most potent algicide activity (EC₅₀ = 0.09 ppm) which is 7-fold potent than our previous inhibitor (Y10) with EC₅₀ of 0.6 ppm.[24] However, it is hard to make direct correlations between CyFBA-II inhibitions and algicide activities from the currently available experimental data, because the absorption, distribution, and metabolism performance of individual compound obtained in the present study. For instance, the CyFBA-II inhibitory activities of compounds L16 (0.88µM), L17 (1.25µM) and L22 (2.7µM) is higher, but these compounds show relatively weak algicide activities, with the EC₅₀ of 5.36 ppm, 6.4 ppm, and 16.3 ppm, respectively.

5	, 0	1			
	$\Delta Gbing(kcal/mol)$	$\Delta\Delta$ Gbind(kcal/mol)	$K_{\rm i}(\mu{ m M})$	$K_{i}(mutant)/K_{i}(WT)$	
WT	-30.56	0	0.65±0.02	1	_
N275A	-24.57	5.99	15.93±1.0	24.5	
R281A	-24.27	6.29	52.34±0.41	80.5	0
S50A	-27.12	3.44	15.2±0.29	23.4	
R302A	-24.35	6.21	20.44±0.14	31.0	
					_

Table 5. Binding free energy of compound L14 and mutant and wild-type (WT) enzyme of CyFBA-II calculated by MMGBSA. together with experimental K_i values

To further confirm that the algicide activities of these hit compounds is achieved by inhibiting CyFBA-II enzyme, the inhibition of the representative compound L14 against overexpressed Synechocystis sp. PCC 6803 (PSBII), whose FBA-II gene was overexpressed (Figure 4B). Due to the increase of FBA-II gene in Synechocystis sp. PCC 6803, the algicide activity of L14 against PSBII ($EC_{50}=4.98\mu$ M, Figure 5B) was decreased about 18-fold compared to the activity of L14 against WT Synechocystis sp. PCC 6803 ($EC_{50}=0.28\mu$ M). The result demonstrated that L14 could actually inhibit CyFBA-II in Synechocystis sp. PCC 6803, in turn inhibit the growth of the Cyanobacterial.



Figure 4. The inhibition curve of compound L14 against Synechocystis sp. PCC 6803. (A) Synechocystis sp. PCC 6803 from wild-type (WT) strains; (B) Synechocystis sp. PCC 6803 from CyFBA-II overexpression strains (PSBII).

3. Conclusions

Herein, using a new Fragment-Based Virtual Screen (FBVS) strategy developed in our group, two series of novel FBA-II inhibitors (thiourea derivatives) were de novo discovered based on the active site of fructose-1, 6-bisphosphate aldolase from Cyanobacterial (CyFBA). In comparison, most of the N-(2benzoylhydrazine-1-carbonothioyl) benzamide derivatives (L14~L22) exhibit higher CyFBA-II inhibitory activities compared to N-(phenylcarbamothioyl) benzamide derivatives (L1~L13). When the OH was introduced on the benzovlhydrazine moiety of the benzamide derivatives, the corresponding compounds, such as L14(0.65µM), L15(0.63µM) and L16(0.88 µM), exhibit high CyFBA-II inhibitory activities. Furthermore, the algicide activities were determined for Synechocystis sp. PCC 6803. The results suggest that most of the compounds with high CvFBA-II inhibition might exhibit potential in vivo activities. Especially, compound L14 not only shows high CyFBA-II activities ($K_i = 0.65 \mu$ M), but also exhibits highest algicide activities (EC₅₀ = 0.09 ppm), which is 7-fold potent than our previous inhibitor (EC₅₀=0.6 ppm). The possible interactions of representative compound L14 and surrounding key residues in the active site of CyFBA-II were elucidated by jointly using DOX, MD simulations, MM-PBSA and Site-directed mutagenesis assays. The well agreement between theoretical and experimental results suggest that the DOX strategy is a reliable approach for the prediction of the binding-model of inhibitor and target, and the present design strategies were successfully implemented to rapidly screen and optimization of novel inhibitors. The satisfactory algicide activities suggest that the thiourea derivatives are very likely to be a promising lead compound for the development of novel specific algicides to solve CHABs.

4. Experimental

4.1. Molecular dynamics (MD) simulations and Binding Free Energies

To verify the binding mode of hit compounds and CyFBA-II, the MD simulations were further performed by using pmemd.cuda module of AMBER16. The hit compounds were optimized by Gaussian 09 at the ω b97xd/6-31g(d) level, and the whole protein was optimized using Amber force field. The AMBER10 force field and gaff force field were used for CyFBA-II and inhibitors, respectively. The whole system was immersed with TIP3P water molecules[31] in a truncated octahedron box of 10 Å from any solute atoms. The system was neutralized with the counterions of Na⁺. SHAKE[32] was used to constrain bonds involving hydrogen atoms, and the time step was 2.0 fs. the long-range electrostatics were calculated by the particle mesh Ewald (PME) algorithm.[33] The nonbonded cutoff was set to 10 Å. Based on the MD trajectories, the binding free energies **L14** were also calculated using MM/GBSA. The electrostatic desolvation free energy was calculated by the Generalized Born (GB) model.[34] The binding interface of CyFBA-II (or it's mutant) and inhibitor is relatively hydrophobic. The solvent and the solute dielectric constants of GB were set to 80 and 1, respectively.[35] The nonpolar desolvation term was estimated from the solvent accessible surface area (SASA): 0.0072 × SASA + 0.00.[36]

4.2. Chemicals and Reagents.

The proton (¹H) and carbon (¹³C) NMR spectra were recorded on a Varian Mercury-Plus 400 or 600 MHz spectrometer in CDCl₃, DMSO- d_6 or acetone- d_6 . Chemical shifts were given in parts per million (ppm) with TMS as the internal reference. Flash chromatography purifications were performed on Merck silica gel 60 (230–400 mesh) as the stationary phase, and MeOH and CH₃COCH₃ were used as eluents. High-resolution mass spectrometry (HRMS) analysis was performed using a hydride IT-TOF mass spectrometer with ESI interface (Shimadzu, Kyoto, Japan) and Agilent 6224 Accurate-Mass time-of-flight mass spectrometer with ESI interface (Agilent Technologies, Waldbronn, Germany). The synthetic procedure, ¹H and ¹³C nuclear magnetic resonance (NMR) and high-resolution mass spectrometry (HRMS) are listed in the Supporting Information.

4.3. Enzyme inhibition activities and Site-Directed Mutagenesis.

Triosephosphate isomerase (TIM), glycerol 3-phosphate dehydrogenase (GPDH), and FBA-I were from rabbit muscle and purchased from Sigma Corporation. The inhibition constant (K_i) of the hit compounds were determined at the CyFBA-II (or it's mutant) recombinant protein level as described previously.[37] The recombinant protein was expressed in E. coli BL21(DE3) cells, it was purified as previous document.[9, 37] Commercial preparations from Sigma of glycerol 3-phosphate dehydrogenase (GPDH) from rabbit muscle and triosephosphate isomerase (TIM) from rabbit muscle were used. The inhibitory activity of hit compounds against CyFBA-II were determined by using an NADH linked enzymatic assay.[38, 39] The activity of cleavage reaction was examined in Tris-HCl buffer (50mM, pH 7.5), NADH (0.6 mM), glycerol 3-phosphate dehydrogenase (GPDH, 0.3 U, Sigma), triosephosphate isomerase (TIM, 1 U, Sigma), aldolase (1.4 µg/mL), and various concentrations of fructose-1, 6-bisphosphate (FBP, 0.04–5mM) in a cuvette to give the final volume of 0.4 mL. The reaction progress was measured by monitoring the change in absorbance of NADH at 340 nm on the microplate reader (SpectraMax M5, Molecular Devices) over 5 min at 30 °C. The dose-response curves of compounds L10, L11, L14, L15 and L16 against wild-type CyFBA-II were illustrated in **Figure S5**.

 K_i values were determined by the inhibition of the reaction at more than two concentrations of the substrate (FBP). In present study, The L11 and L14 were selected for the validation of the competitive inhibitors, as shown in **Figure S6**. For the competitive inhibitors, K_i may be calculated by the equation[40]: $K_i = IC_{50}/(1+S/K_m)$, where IC_{50} is the half-maximal inhibitory concentration, S is the concentration of FBP and K_m is the Michaelis-Menten constant. The K_m value of CyFBA-II is 68 μ M. The inhibition rate (I%) was calculated using the following formula: $I\% = [(V_0 - V)/V_0] \times 100\%$, where V_0 represent the maximum velocity ([I] = 0 μ M), and V indicate the velocity when an inhibitor was added. All data were analyzed by GraphPad Prism 6.0 software (Prism Software).

Mutations were performed by introducing specific base changes into a double-stranded DNA plastid. All highly effective site-directed mutagenesis were performed using KODPlus Mutagenesis kit (Toyobo, Osaka, Japan) in term of the manufacturer's protocol as our previous document.[41] The wild-type CyFBA-II plasmid and mutagenic primers were denatured, annealed, and polymerized in a final volume of 50 μ L for 25 cycles (95 °C for 30s, 60 °C for 1 min, and 72 °C for 3 min). Additionally, *Nde I* and *BamH I* restriction enzyme (Fermentas, China) was used to digest the parental methylated and hemi-methylated DNA. Then, the plasmids of mutations were transformed into *Escherichia coli* (*E. coli*) strain BL21 (DE3) cells, and the sequence analysis were performed for validations. By using SDS-PAGE, the identical motilities were exhibited for all purified enzymes (wild and mutants) to ensure the purities more than 90%.

4.4. Inhibitory Assays on Synechocystis sp. PCC 6803.

Synechocystis sp. PCC 6803 were cultured photoautotrophically in BG11 medium[42] at 28 °C, it is in light for 12 h and in dark for 12 h alternatively, and 50–55% relative humidity for seven days in 96-well

microtiter plates as described previously.[9] This is similar to the antibacterial bioassay.[43] The inhibition rate (I%) of test compounds on the cyanobacteria growth was determined according to previous work.[8] The half maximal effective concentration (EC₅₀) values were obtained from the inhibition curves for the inhibition rate against various concentrations of hit compounds. The curves were fitted by nonlinear regression using logistic equation in Origin 7.7 software.

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Supplementary Material

Experimental procedures, analytical data, enzyme assays, and biological evaluation. This material can be found online

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Design, synthesis and algicides activities of thiourea derivatives as the Novel Scaffold Aldolase Inhibitors

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