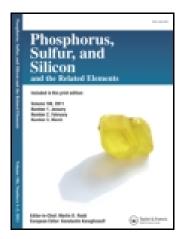
This article was downloaded by: [University of Victoria] On: 29 January 2015, At: 19:19 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/gpss20

Reactions with Hydrazonoyl Halides 63: Synthesis and Anticancer Activity of Some New 1,3,4-Thiadiazoles, 1,3,4-Selenadiazoles, and 1,2,4-Triazolo[4,3a]pyrimidines

Eman K. A. Abdelall^a, Mahmoud A. Mohamed^b & Abdou O. Abdelhamid^c

^a Department of Pharmaceutical Organic Chemistry , Faculty of Pharmacy, Beni-Suef University , Beni-Suef, Egypt

^b Department of Textile, Faculty of Industrial Education, Beni-Suef University, Beni-Suef, Egypt

 $^{\rm c}$ Department of Chemistry , Faculty of Science, Cairo University , Giza, Egypt

Published online: 25 Aug 2010.

To cite this article: Eman K. A. Abdelall , Mahmoud A. Mohamed & Abdou O. Abdelhamid (2010) Reactions with Hydrazonoyl Halides 63: Synthesis and Anticancer Activity of Some New 1,3,4-Thiadiazoles, 1,3,4-Selenadiazoles, and 1,2,4-Triazolo[4,3-a]pyrimidines, Phosphorus, Sulfur, and Silicon and the Related Elements, 185:9, 1862-1874, DOI: <u>10.1080/10426500903348013</u>

To link to this article: <u>http://dx.doi.org/10.1080/10426500903348013</u>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions



Phosphorus, Sulfur, and Silicon, 185:1862–1874, 2010 Copyright © Taylor & Francis Group, LLC ISSN: 1042-6507 print / 1563-5325 online DOI: 10.1080/10426500903348013

REACTIONS WITH HYDRAZONOYL HALIDES 63: SYNTHESIS AND ANTICANCER ACTIVITY OF SOME NEW 1,3,4-THIADIAZOLES, 1,3,4-SELENADIAZOLES, AND 1,2,4-TRIAZOLO[4,3-*a*]PYRIMIDINES

Eman K. A. Abdelall,¹ Mahmoud A. Mohamed,² and Abdou O. Abdelhamid³

 ¹Department of Pharmaceutical Organic Chemistry, Faculty of Pharmacy, Beni-Suef University, Beni-Suef, Egypt
 ²Department of Textile, Faculty of Industrial Education, Beni-Suef University, Beni-Suef, Egypt
 ³Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt

2,3-Dihydro-1,3,4-thiadiazoles, 2,3-dihydro-1,3,4-selenadiazoles, and triazolino[4,3a]pyrimidines containing benzoxazole or benzothiazole moieties were prepared from the reaction of each of ethyl 3-aza-3-(benzoxazol-2-ylamino)-2-chloroprop-2-enoate and ethyl 3-aza-3-(benzothiazolo-2-ylamino)-2-chloroprop-2-enoate with each of potassium thiocyanate, potassium selenocyanate, alkyl carbodithioate, and pyrmidine-2-thione derivatives. All the newly synthesized compounds were confirmed by elemental analysis, spectral data, and alternative route synthesis whenever possible. Some of the newly synthesized compounds were screened toward certain cancer tumors.

Supplemental materials are available for this article. Go to the publisher's online edition of Phosphorus, Sulfur, and Silicon and the Related Elements to view the free supplemental file.

Keywords Anticancer activity; hydrazonoyl bromide; 1,3,4-selenadiaziles; 1,3,4-thiadiazoles; triazolo[4,3-a]pyrimidines

INTRODUCTION

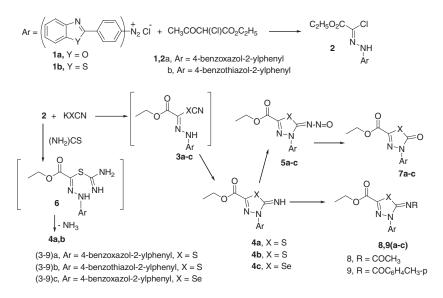
Hydrazonoyl halides have been widely employed for the synthesis of heterocyclic compounds.¹⁻⁴ 1,3,4-Thiadiazole derivatives have become useful in medicine, agriculture, and in many fields of technology.⁵ In addition, 2-(4-aminophenyl)benzothiazole and oxazole and their analogues are a novel class of potent and selective antitumor agents.^{6–10} As an extension of our study^{11–16} and of our synthesis of 1,3,4-thiadiazoles, we report here the synthesis of hydrazonoyl halides containing 4-phenylbenzothiazole and 4-phenylbenzoxazole moieties and their reactions toward some potassium thiocyanate, potassium selenocyanate, alkyl carbodithioate, and pyrimidine-2-thione derivatives.

Received 1 June 2009; accepted 18 September 2009.

Address correspondence to Abdou O. Abdelhamid, Department of Chemistry, Faculty of Science, Cairo University, Giza 12613, Egypt. E-mail: aabdou_abdelhamid@yahoo.com

RESULTS AND DISCUSSION

Treatment of the appropriate diazotized [4-(1,3-benzoxazol-2-yl)phenyl]amine⁶ (1a) and diazotized [4-(1,3-benzothiazol-2-yl)phenyl]amine¹⁰ (1b) with ethyl 2chloro-3-oxobutanoate in ethanolic sodium acetate gave ethyl 3-aza-3-[(benzoxazol-2-ylphenyl)amino]-2-chloroprop-2-enoate (2a) and ethyl 3-aza-3-(benzothiazolo-2ylphenyl)amino]-2-chloroprop-2-enoate (2b), respectively. Structure 2 was elucidated by elemental analysis and spectral data. The ¹H NMR spectrum of 2a showed signals at $\delta = 1.37$ (t, 3H, J = 7.5 Hz, CH₂CH₃), 4.35 (q, 2H, J = 7.5 Hz, CH₂CH₃), 7.40 (d, 2H, J = 8Hz, ArH's), 7.50 (d, 2H, J = 8Hz, ArH's), 8.00 (d, 2H, J = 8Hz, ArH's), 8.07 (d, 2H, J = 8Hz, ArH's), and 10.89 (s, br., 1H, NH). Treatment of the appropriate 2a with each of potassium thiocyanate and potassium selenocyanate gave ethyl 3-(4-(benzoxazol-2-ylphenyl)-2-imino-1,3,4-thiadiazoline-5-carboxylate (4a) and ethyl 3-(4-(benzoxazol-2-ylphenyl)-2-imino-1,3,4-selenadiazoline-5-carboxylate (4c), respectively (Scheme 1). The structure of 4 was elucidated on the basis of elemental analyses, spectral data, alternative synthetic route, and its nitrosation and acylation reactions.





In addition, treatment of the appropriate **2a** with thiourea in boiling ethanol gave a product identical in all respects (mp, mixed mp, and spectra) with **4a**. These results indicate that hydrazone **3a** and amidiazone **6a** are not the final products, and that **3a** and **6a** readily gave **4a** either by cyclization or by elimination of one molecule of ammonia (Scheme 1). Acylation of **4a** with acetic anhydride or with 4-methylbenzoyl chloride in pyridine afforded ethyl 2-(1-aza-2-oxopropylidene)-3-(4-benzoxazol-2-ylphenyl)-1,3,4thiadiazoline-5-carboxylate (**8a**) and ethyl 2-[1-aza-2-(4-methylphenyl)-2-oxoethylidene]-3-(4-benzoxazol-2-ylphenyl)-1,3,4-thiadiazoline-5-carboxylate (**9a**), respectively. Spectral data and elemental analyses confirmed their structures. ¹H NMR spectrum of **8a** showed signals at $\delta = 1.45$ (t, 3H, J = 7.5 Hz, CH₂CH₃), 2.41 (s, 3H, CH₃CON=), 4.47(q, 2H, J = 7.5 Hz, CH₂CH₃), 7.40 (d, 2H, J = 8Hz, ArH's), 7.50 (d, 2H, J = 8Hz, ArH's), and 8.44 (d, 2H, J = 8Hz, ArH's).

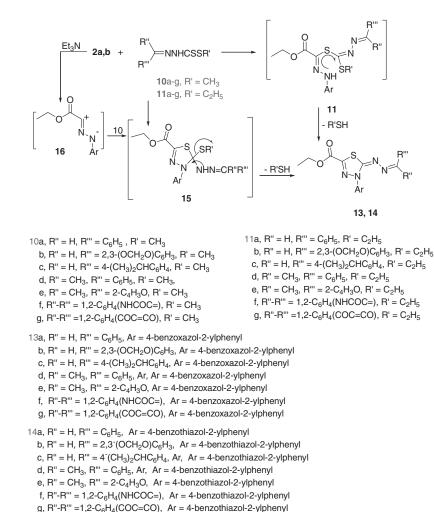
Analogously, acylation of each of **4b** and **4c** gave **8b,c** and **9b,c**, respectively. Nitrosation of each **4a** and **4c** with saturated sodium nitrite in acetic acid at $0-5^{\circ}$ C gave ethyl 2-(azanitrosomethylene)-3-(benzoxazol-2-ylphenyl)-1,3,4-thiadiazoline-5-carboxylate (**5a**) and ethyl 2-(azanitrosomethylene)-3-(benzoxazol-2-ylphenyl)-1,3, 4-selenadiazoline-5-arboxylate (**5c**), respectively. Ethyl 3-(4-benzoxazol-2-ylphenyl)-2oxo-1,3,4-thia/selenadiazoline-5-carboxylate **7a,c** and 3-(4-benzothiazol-2-ylphenyl)-2oxo-1,3,4-thiadiazoline-5-carboxylate **7b** were prepared by thermolysis of **5a,c** and **5b** in boiling xylene. The IR spectrum of **7** revealed a band near $\nu = 1685$ cm⁻¹ (CO).

Compound **2a** reacted with the methyl carbodithioates^{18,19} **10a** to give the 1,3,4thiadiazoline derivative **13a**. The structure of **13a** was confirmed by elemental analysis, spectral data, and alternative synthetic route. ¹H NMR of **13a** showed signals at $\delta =$ 1.36(t, 3H, **CH**₃CH₂), 4.39 (q, 2H, **CH**₂CH₃), 7.54–8.33 (m, 13H, ArH's), and 8.57 CH (vinyl)). Also, treatment of **2a** with ethyl carbodithioates **11a** in ethanolic triethylamine gave a product identical in all respects (mp, mixed mp, and spectra) with **13a**. Product **13a** was assumed to be formed via elimination of alkanethiol (R¹SH) from the corresponding cycloadduct **15**, which formed from 1,3-dipolar cycloaddition (or 1,3-addition) of nitrile imide **16** (generated in situ from hydrazonoyl chlorides **2** and triethylamine) to C=S **14** (Scheme 2). Analogously, treatment of the appropriate **2a,b** with the appropriate **10a–g** (or **11a–g**) in ethanolic triethylamie gave the thiadiazolines **13a–g** and **14 a–g**, respectively (Scheme 2).

Next, treatment of ethyl 4-methyl-6-phenyl-2-thioxo-1,3,6-trihydro-pyrimidine-5carboxylate (**17a**) with **2a** in chloroform and triethylamine gave the 1,2,4-triazolo[4,3*a*]pyrimidine-5-carboxylates **20a**. The structure of **20a** was elucidated by elemental analysis, spectra, and alternative synthesis. The ¹H NMR spectrum of **20a** showed signals at δ = 1.23 (t, 3H, **CH**₃CH₂), 1.12 (t, 3H, **CH**₃CH₂), 2.88 (s, 3H, CH₃), 4.02 (q, 2H, **CH**₂CH₃), 4.11 (q, 2H, **CH**₂CH₃), 6.67 (s, 1H, CH), and 6.8–8.43 (m, 13H, ArH). Its IR spectrum revealed bands at ν = 1735 cm⁻¹ (CO). Ethyl 6-methyl-4-[4-phenyl-2-methylthio-3,4dihydropyrimidine-5-carboxylate (**22a**) reacted with **2a** in boiling ethanolic sodium ethoxide solution gave products identical in all aspects (mp, mixed mp, and spectra) with the corresponding **20a**. Analogously, treatment of each **2a** with the appropriate **17b,c** and **2b** with the appropriate **17a–c** gave triazolo[4,3-*a*]pyrimidines **20b,c** and **21a–c**, respectively (Scheme 3).

The formation of 20 can be explained via 1,3-dipolar cycloaddition or 1,3-addition of nitrile imide 16 (prepared in situ from hydrazonoyl chlorides 2 with triethylamine or sodium ethoxide) to C=S of 16 (or NH of 22) to give intermediate 18 (or 23), with ring opening and ring closure to afford the final products 20 by elimination of hydrogen sulfide from 19 (or methyl mercaptan from 23) (Scheme 3).

Treatment of **2a** with methyl benzoylhydrazinecarbodithioate (**24**) in ethanolic triethylamine gave 2,3-dihydro-1,3,4-thiadiazole **26** (Scheme 4). Its structure was elucidated by elemental analysis, spectral data, and alternative synthetic route. The ¹H NMR spectrum showed signals at $\delta = 1.39$ (t, 3H, **CH**₃CH₂), 4.42 (q, 2H, CH₃**CH**₂), 7.42–8.35 (m, 13H, ArH), and 11.42 (s, br, 1H, NH). Thus, treatment of **2a** with 5-phenyl-1,3,4-oxadiazole-2-thiol (**27**) gave products identical in all respects (mp, mixed mp, and spectra) with **26a** (Scheme 4). Similarly, treatment of the appropriate **2a,b** with **28** gave thiadiazolines **29a** and **29b**, respectively.



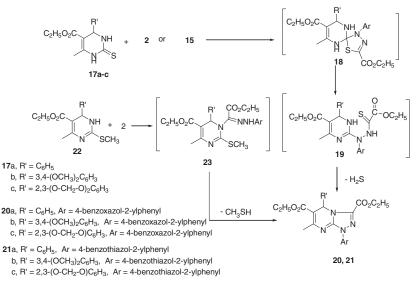
Scheme 2

ANTICANCER AND CYTOTOXIC ACTIVITY

Twelve of the synthesized compounds were subjected to anticancer screening, and some of these compounds showed a cytotoxic effect. Female Swiss albino mice from the Animal House of the Egyptian Cancer Institute were maintained on standard pellet diet and water, and in vitro test for cytotoxic effect was operated using Ehrlich ascites carcinoma (EAC)¹⁷ induced through interperitonial transplantation in female Swiss albino mice. (See the Supplemental Materials, available online, for full details.)

EXPERIMENTAL

All melting points were determined on an electrothermal apparatus and are uncorrected. IR spectra were recorded (KBr discs) on a Shimadzu FT-IR 8201 PC spectrophotometer. ¹H NMR spectra were recorded in CDCl₃ and (CD₃)₂SO solutions on a Varian

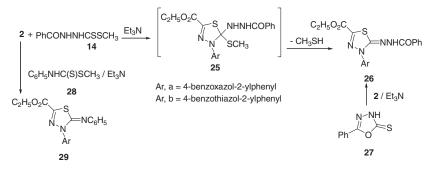




Gemini 300 MHz spectrometer, and chemical shifts were expressed in δ units using TMS as internal reference. Elemental analyses were carried out at the Microanalytical Center of the Cairo University. Alkyl carbodithioates^{18,19} were prepared as previously reported.

Synthesis of Ethyl 3-Aza-3-(benzoxazol-2-ylamino)-2-chloroprop-2enoate (2a) and Ethyl 3-Aza-3-(benzothiazolo-2-ylamino)-2chloroprop-2-enoate (2b)

The appropriate amount of diazotized [4-(1,3-benzoxazol-2-yl)phenyl]amine (1a) and diazotized [4-(1,3-benzothiazol-2-yl)phenyl]amine (1b) (0.005 mol) were added dropwise to cold solution (0°C) of ethyl 2-chlorobutanoate (1.65 g, 0.005 mol) and sodium acetate trihydrate (0.65 g, 0.005 mol) while stirring. The yellow precipitate was collected and recrystallized from ethanol to give 2a and 2b, respectively (Tables I and II).



Scheme 4

REACTIONS WITH HYDRAZONOYL HALIDES

% Analyses, Calcd./Found Mol. Formula Yield ^a % Mp, °C Comp					Compound		
S	Ν	Н	С	Mol. Formula Mol. Wt.	Color	Solvent	no.
	12.22	4.10	59.40	C ₁₇ H ₁₄ ClN ₃ O ₃	82	199–201	2a
	12.02	3.99	59.23	343.76	Yellow	EtOH	
8.91	11.68	3.92	56.74	$\mathrm{C_{17}H_{14}ClN_{3}O_{2}S}$	80	172-174	2b
9.01	11.79	3.99	56.59	359.83	Orange	EtOH	
8.75	15.29	3.85	59.01	$C_{18}H_{14}N_4O_3S$	60	170-174	4a
8.59	15.32	3.77	59.12	366.39	Brown	EtOH	
16.77	14.65	3.69	56.53	$C_{18}H_{14}N_4O_2S_2$	60	165-168	4b
16.70	14.65	3.80	56.66	382.46	Brown	EtOH	
	13.56	3.41	52.31	$C_{18}H_{14}N_4O_3Se$	65	169-170	4c
	13.50	3.32	52.32	413.29	Brown	EtOH	
3.11	17.71	3.31	54.68	$C_{18}H_{13}N_5O_4S$	85	134-137	5a
3.01	17.65	3.22	54.60	395.39	Red	Acetone	
15.59	17.02	3.18	52.54	$C_{18}H_{13}N_5O_3S_2$	80	128-131	5b
15.48	16.99	3.10	52.44	411.46	Pale rose	Acetone	
	15.83	2.96	48.88	C ₁₈ H ₁₃ N ₅ O ₄ Se	80	134-137	5c
	15.75	3.01	48.75	442.29	Pale rose	Acetone	
8.73	11.44	3.57	58.85	C ₁₈ H ₁₃ N ₃ O ₄ S	60	155-157	7a
3.70	11.41	3.77	58.77	367.38	Shiny yellow	EtOH	
16.72	10.96	3.42	56.38	$C_{18}H_{13}N_3O_3S_2$	65	151-155	7b
16.66	11.01	3.32	56.30	383.44	Shiny yellow	EtOH	
_	10.14	3.16	52.19	C ₁₈ H ₁₃ N ₃ O ₄ Se	60	124-125	7c
	9.86	3.11	52.09	414.27	Shiny yellow	DMF/EtOH	
7.85	13.72	3.95	58.81	C ₂₀ H ₁₆ N ₄ O ₄ S	85	164-166	8a
7.69	13.69	3.79	58.79	408.43	Buff	EtOH	
15.11	13.20	3.80	56.59	C ₂₀ H ₁₆ N ₄ O ₃ S ₂	80	155-156	8b
1501	13.30	3.87	56.55	424.5	Buff	DMF/EtOH	
	12.30	3.54	52.76	C ₂₀ H ₁₆ N ₄ O ₄ Se	80	152-154	8c
	12.36	3.44	52.59	455.33	Buff	Aq. EtOH	
5.62	11.56	4.16	64.45	C26H20N4O4S	90	202-204	9a
5.55	11.39	4.17	64.39	484.53	Buff	DMF/EtOH	
2.81	11.19	4.03	62.38	$C_{26}H_{20}N_4O_3S_2$	85	207-209	9b
2.69	11.22	4.11	62.40	500.59	Buff	DMF/EtOH	- 14
_	10.54	3.79	58.76	$C_{26}H_{20}N_4O_4Se$	85	175–177	9c
	10.39	3.63	58.59	531.42	Buff	DMF/EtOH	
5.83	14.92	4.08	63.95	$C_{25}H_{19}N_5O_3S$	75	213–214	13a
5.72	14.79	4.01	63.75	469.52	Yellow	DMF/EtOH	104
5.24	13.64	3.73	60.81	C ₂₆ H ₁₉ N ₅ O ₅ S	76	236–238	13b
5.20	13.59	3.66	60.77	513.52	Yellow	DMSO	155
5.20	13.69	4.93	65.74	C ₂₈ H ₂₅ N ₅ O ₃ S	75	187–189	13c
5.20	13.66	4.90	65.59	511.59	Yellow	DMF	150
5.63	14.48	4.38	64.58	C ₂₆ H ₂₁ N ₅ O ₃ S	65	170–173	13d
5.55 5.55	14.45	4.33	64.52	483.54	Yellow	DMF/EtOH	150
5.77	14.49	4.04	60.88	$C_{24}H_{19}N_5O_4S$	70	220-222	13e
5.77 5.71	14.79	4.04	60.88 60.95	473.5	Yellow	220–222 DMF	150
5.28	14.00	4.00 3.55	61.17	$C_{26}H_{18}N_6O_4S$	72	289–290	13f
5.28 5.21	16.46		61.09			289–290 DMF/EtOH	131
5.21 5.12	10.33	3.49		510.52 CoeHeeN-O-S	Orange 78		12.
		3.27	61.94	C ₂₇ H ₁₇ N ₅ O ₅ S		234–236	13g
5.00	13.31	3.11	61.70	523.52 C H N O S	Red	DMF/EtOH	14-
							14a
3.21 3.09	14.42 14.31	3.94 3.79	61.84 61.77	$\begin{array}{c} C_{25}H_{19}N_5O_2S_2\\ 485.58\end{array}$	72 Orange	245–247 DMF/EtOH	14a

Table I Characterization data of the newly synthesized compounds

(Continued on next page)

% Analyses, Calcd./Found			und	Mol. Formula	Yield ^a %	Mp, °C	Compound
S	Ν	Н	С	Mol. Wt.	Color	Solvent	no.
12.11	13.22	3.62	58.97	$C_{26}H_{19}N_5O_4S_2$	70	247–249	14b
12.00	13.09	3.69	58.89	529.59	Pale orange	DMF	
12.15	13.27	4.78	63.73	$C_{28}H_{25}N_5O_2S_2$	70	181-184	14c
12.01	13.09	4.71	63.66	527.66	Pale orange	DMF	
12.84	14.02	4.24	62.50	$C_{26}H_{21}N_5O_2S_2$	74	267-269	14d
12.90	14.21	4.05	62.41	499.61	Orange	DMF	
13.10	14.31	3.91	58.88	$C_{24}H_{19}N_5O_3S_2$	76	251-253	14e
13.01	14.19	3.82	58.69	489.57	Yellow	DMF	
12.18	15.96	3.45	59.30	$C_{26}H_{18}N_6O_3S_2$	78	> 300	14f
12.09	16.02	3.33	59.49	526.59	Pale orange	DMF/EtOH	
11.89	12.98	3.18	60.10	C ₂₇ H ₁₇ N ₅ O ₄ S ₂	80	241-243	14g
11.79	12.78	3.01	60.00	539.58	Red	DMF/EtOH	
	12.74	4.95	67.75	C ₃₁ H ₂₇ N ₅ O ₅	66	267-269	20a
	12.69	4.79	67.69	549.58	Yellow	CHCl ₃ /MeOH	
	11.49	5.13	65.02	C33H31N5O7	64	244-246	20b
	11.33	5.23	64.92	609.63	Orange	CHCl ₃ /MeOH	
	11.80	4.58	64.75	C ₃₂ H ₂₇ N ₅ O ₇	60	252-254	20c
	11.77	4.66	64.96	593.59	Orange	CHCl ₃ /MeOH	
5.67	12.38	4.81	65.82	C31H27N5O4S	61	260-263	21a
5.61	12.49	4.78	65.69	565.64	Orange	CHCl ₃ /MeOH	
5.12	11.19	4.99	63.35	C33H31N5O6S	62	181-182	21b
5.00	11.30	4.89	63.46	625.69	Orange	CHCl ₃ /MeOH	
5.26	11.49	4.46	63.04	C32H27N5O6S	66	267-269	21c
5.16	11.36	4.33	63.22	609.65	Orange	CHCl ₃ /MeOH	
6.60	14.42	3.94	61.85	C25H19N5O4S	78	251-253	26a
6.67	14.53	3.81	61.79	485.51	Yellow	EtOH	
7.25	12.66	4.10	65.14	C24H18N4O3S	68	228-231	29a
7.16	12.73	4.00	65.01	442.49	Yellow	Acetic acid	
13.99	12.22	3.96	62.86	$C_{24}H_{18}N_4O_2S_2$	65	221-223	29b
13.82	12.00	3.82	62.89	458.56	Yellow	Acetic acid	

 Table I Characterization data of the newly synthesized compounds (Continued)

Synthesis of Ethyl 3-(4-(Benzoxazol-2-ylphenyl)-2-imino-1,3,4thiadiazoline-5-carboxylate (4a), Ethyl 3-(4-(Benzothiazol-2-ylphenyl)-2-imino-1,3,4-thiadiazoline-5-carboxylate (4b), and Ethyl 3-(4-(Benzoxazol-2-ylphenyl)-2-imino-1,3,4-selenadiazoline-5-carboxylate (4c)

Method A. A mixture of the appropriate 2a,b (0.005 mol) and the appropriate amount of potassium thiocyanate (or potassium selenocyanate) (0.006 mol) in ethanol (25 mL) was stirred at room temperature for 24 h. The resulting solid was collected, washed with water, and crystallized from ethanol to give 4a,b and 4c, respectively (Tables I and II).

Method B. A mixture of the appropriate 2a,b (0.005 mol) and thiourea (0.38 g, 0.005 mol) in ethanol (25 mL) was refluxed for 3 h. The solid product that formed after cooling was collected and crystallized from ethanol to give products identical in all respects (mp, mixed mp, and spectra) with 4a and 4b, respectively.

1869

Table II	¹ H NMR spectroscopic data of some synthesized compounds

Spectral data	Compound
IR: 3258 (NH), 3055 (CH, aromatic), 2981 (CH, aliphatic), 1747 (CO), 1605 (C=C), 1476	2a
(CH ₂), 1363 (CH ₃)	
¹ H NMR: $\delta = 1.37$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.35 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.40 (d,	
2H, J = 8Hz, ArH's), 7.50 (d, 2H, J = 8Hz, ArH's), 8.00 (d, 2H, J = 8Hz, ArH's), 8.07 (d,	
2H, J = 8Hz, ArH's) and 10.89 (s, br., $1H, NH$).	
IR: 3132 (NH), 3055 (CH, aromatic), 2926 (CH, aliphatic), 1743 (CO), 1605 (C=C), 1476	2b
(CH ₂), 1355 (CH ₃)	
¹ H NMR: $\delta = 1.37$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.35 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.40 (d,	
2H, J = 8Hz, ArH's), 7.50 (d, 2H, J = 8Hz, ArH's), 8.00 (d, 2H, J = 8Hz, ArH's), 8.07 (d,	
2H, J = 8Hz, ArH's) and 10.89 (s, br., 1H, NH).	
IR: 3462 (NH), 3057 (CH, aromatic), 2983 (CH, aliphatic), 1725 (CO, ester), 1628 (C=N), 1603	4 a
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.41(t, 3H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 4.35 (q, 2H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 7.36-8.36$	
(m, 9H, ArH's and NH protons)	
IR: 3460 (NH), 30.55 (CH, aromatic), 2981 (CH, aliphatic), 1729 (CO, ester), 1636 (C=N), 1603	4b
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.41(t, 3H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 4.34 (q, 2H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 7.36-8.36$	
(m, 9H, ArH's and NH protons)	
IR: 3463 (NH), 30.55 (CH, aromatic), 2981 (CH, aliphatic), 1725 (CO, ester), 1636 (C=N), 1603	4c
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.26$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.31 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.38–8.28	
(m, 8H, ArH's), 9.85 (s, br., 1H, NH, exchangable)	
IR: 3055 (CH,aromatic), 2984 (CH, aliphatic), 1720 (CO, ester), 1636 (C=N), 1550 (NO), 1603	5a
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.41(t, 3H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 4.35 (q, 2H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 7.36-8.36$	
(m, 8H, ArH's)	
IR: 3055 (CH,aromatic), 2980 (CH, aliphatic), 1722 (CO, ester), 1636 (C=N), 1603 (C=C),	5b
1535 (NO), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.41(t, 3H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 4.35 (q, 2H, J = 7.5 \text{ Hz}, \text{CH}_2\text{CH}_3), 7.36-8.36$	
(m, 8H, ArH's)	
IR: 3057 (CH,aromatic), 2983 (CH, aliphatic), 1725 (CO, ester), 1628 (C=N), 1603 (C=C),	5c
1537 (NO), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.35$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.38 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.39–8.36	
(m, 8H, ArH's)	
IR: 3055 (CH, aromatic), 2981 (CH, aliphatic), 1725 (CO, ester), 1685 (CO), 1660 (CO,	7a
conjugated), 1636 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.44$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.47 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.27–8.36	
(m, 8H, ArH's)	
IR: 3055 (CH,aromatic), 2981 (CH, aliphatic), 1725 (CO, ester), 1683 (CO), 1636 (C=N), 1603	7b
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.44$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.47 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.27–8.38	
(m, 8H, ArH's)	
IR: 3057 (CH,aromatic), 2983 (CH, aliphatic), 1725 (CO, ester), 1685 (CO), 1628 (C=N), 1603	7c
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.44$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.47 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.27–8.36	
(m, 8H, ArH's)	
IR: 3057 (CH,aromatic), 2983 (CH, aliphatic), 1725 (CO, ester), 1685 (CO), 1628 (C=N), 1603	8a
$(C=C)$, 1476 (CH_2) , 1663 (CH_3) .	
¹ H NMR: $\delta = 1.45$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.41 (s, 3H, CH ₃ CO=N), 4.47 (q, 2H, J = 7.5	
Hz, CH ₂ CH ₃), 7.27–8.44 (m, 8H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1683 (CO), 1628 (C=N), 1603	8b
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.45$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.41 (s, 3H, CH ₃ CO=N), 4.47 (q, 2H, J = 7.5	
Hz, CH₂CH₃), 7.27–8.44 (m, 8H, ArH's)	
,, , , , , , , , , , , , , ,	

 Table II
 ¹H NMR spectroscopic data of some synthesized compounds (Continued)

Spectral data	Compound
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1683 (CO), 1628 (C=N), 1603	8c
$(C=C), 1476 (CH_2), 1663 (CH_3).$	
¹ H NMR: $\delta = 1.45$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.32 (s, 3H, CH ₃ CO=N), 4.41 (q, 2H, J = 7.5	
Hz, CH ₂ CH ₃), 7.44–8.42 (m, 8H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1675 (CO), 1628 (C=N), 1603	9a
(C=C), 1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.46$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.43 (s, 3H, 4-CH ₃ C ₆ H ₄), 4.53 (q, 2H, J = 7.5	
Hz, CH ₂ CH ₃), 7.26–8.50 (m, 12H, ArH's)	
IR: 3050 (CH,aromatic), 2981 (CH, aliphatic), 1725 (CO, ester), 1678 (CO), 1628 (C=N), 1603	9b
$(C=C)$, 1476 (CH_2) , 1663 (CH_3) .	
¹ H NMR: $\delta = 1.46$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.43 (s, 3H, 4-CH ₃ C ₆ H ₄), 4.53 (q, 2H, J = 7.5	
Hz, CH₂ CH ₃), 7.26–8.50 (m, 12H, ArH's) IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1674 (CO), 1628 (C=N), 1603	9c
	90
$(C=C)$, 1476 (CH_2) , 1663 (CH_3) .	
¹ H NMR: δ = 1.34 (t, 3H, <i>J</i> = 7.5 Hz, CH ₂ CH ₃), 2.36 (s, 3H, 4-CH ₃ C ₆ H ₄), 4.41 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.32–8.48 (m, 12H, ArH's)	
IR: 3055 (CH, aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	13a
1476 (CH ₂), 1663 (CH ₃).	13a
¹⁴⁷⁰ (CH ₂), 1005 (CH ₃). ¹ H NMR: $\delta = 1.36$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.39 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.48–8.33	
(m, 13H, ArH's), 8.57 (s, 1H, CH=) $(4, 2H, 3)$ (q, 2H, $3 = 7.5$ Hz, CH2CH3), 7.46–8.55	
IR: 3055 (CH, aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	13b
1466 (CH ₂), 1653 (CH ₃).	150
¹ H NMR: $\delta = 1.37$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.41 (q, 2H, $J = 7.5$ Hz, CH ₂ CH ₃), 6.10 (s,	
2H, OCH ₂ O), $7.01-8.30$ (m, 11H, ArH's), 8.44 (s, 1H, CH=)	
IR: 3055 (CH, aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	13c
1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.28$ (d, 6H, (CH ₃) ₂ CH, 1.36 (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.99 (sex, 1H,	
(CH ₃) ₂ CH, 4.39 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.23–8.35 (m, 12H, ArH's), 8.46 (s, 1H, CH=)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	13d
1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: insoluble	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	13e
1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.37$ (t, 3H, $J = 7.5$ Hz, CH₃CH₂), 2.35 (s, 3H, CH ₃), 4.35 (q, 2H, J = 7.5 Hz,	
CH ₂ CH ₃), 6.35 (s, 1H, furan H-4), 7.48–8.38 (m, 10H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1680 (CO), 1628 (C=N), 1603	13f
$(C=C), 1476 (CH_2), 1663 (CH_3).$	
¹ H NMR: $\delta = 1.36$ (t, 3H, CH ₃ CH ₂ , 4.42 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 6.85–8.38 (m, 12H,	
ArH's), 8.46 (s, br., 1H, NH)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1695 (CO), 1628 (C=N), 1603	13g
$(C=C), 1476 (CH_2), 1663 (CH_3).$	
¹ H NMR: $\delta = 1.35$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.38 (q, 2H, $J = 7.5$ Hz, CH ₂ CH ₃), 7.42–8.51	
(m, 12H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	14a
1476 (CH ₂), 1663 (CH ₃).	
¹ H NMR: $\delta = 1.36$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.39 (q, 2H, $J = 7.5$ Hz, CH ₂ CH ₃), 7.48–8.33	
(m, 13H, ArH's), 8.57 (s, 1H, CH=)	1.0
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1662 (CH ₂)	14b
1476 (CH ₂), 1663 (CH ₃). ¹ H NMP: $\delta = 1.25$ (t. 2H, $L = 7.5$ Hz, CH, CH ₂) 4.41 (a. 2H, $L = 7.5$ Hz, CH, CH ₂) 6.10 (a.	
¹ H NMR: $\delta = 1.35$ (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 4.41 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 6.10 (s, 2H, OCH-O) 6.98 8.23 (m, 11H, ArH's) 8.42 (s, 1H, CH=)	
2H, OCH ₂ O), 6.98–8.23 (m, 11H, ArH's), 8.42 (s, 1H, CH=) IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	140
1476 (CH ₂), 1663 (CH ₃).	14c
¹⁴ / ₁₀ (CH ₂), 1005 (CH ₃). ¹ H NMR: $\delta = 1.28$ (d, 6H, (CH ₃) ₂ CH), 1.36 (t, 3H, $J = 7.5$ Hz, CH ₂ CH ₃), 2.99 (sex, 1H,	
$(CH_3)_2CH$, 4.39 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.23–8.35 (m, 12H, ArH's), 8.46 (s, 1H, CH=)	
$(C_{113})_2 C_{11}, \tau_{,,,,,,,,,$	

 Table II
 ¹H NMR spectroscopic data of some synthesized compounds (Continued)

Spectral data	Compound
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	14d
¹ H NMR: insoluble IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	14e
1476 (CH ₂), 1663 (CH ₃). ¹ H NMR: δ = 1.37 (t, 3H, <i>J</i> = 7.5 Hz, (CH₃CH₂), 2.35 (s, 3H, CH ₃), 4.35 (q, 2H, J = 7.5 Hz, CH (CH) (25 (c, 1H) from H (4) 7.48, 8.28 (m, 10H) A-Hz))	
CH ₂ CH ₃), 6.35 (s, 1H, furan H-4), 7.48–8.38 (m, 10H, ArH's) IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1682 (CO), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	14f
^(C) C), 1470 (CH ₂), 1005 (CH ₃). ¹ H NMR: $\delta = 1.36$ (t, 3H, J = 7.5 Hz, CH₃CH₂), 4.42 (q, 2H, J = 7.5 Hz, CH₂CH₃), 6.85–8.38 (m, 12H, ArH's), 8.46 (s, br., 1H, NH)	
 (III, 121, 11113), 0.10 (9, 01, 11, 111) IR: 3055 (CH, aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1690 (CO), 1628 (C=N), 1603 (C=C), 1476 (CH₂), 1663 (CH₃). 	14g
$^{(0)}$ C), 110 (CH ₂), 100 (CH ₃), 1 H NMR: $\delta = 1.30$ (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 4.29 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.48–8.48 (m, 12H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	20a
¹ H NMR: $\delta = 1.23$ (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 1.42 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 2.88 (s, 3H, CH ₃), 4.11 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 4.45 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 6.67 (s, 1H, pyrimidine H-4), 6.89–8.43 (m, 13H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	20b
¹ H NMR: δ = 1.23 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 1.42 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 2.88 (s, 3H, CH ₃), 3.47 (s, 3H, OCH ₃), 3.72(s, 3H, OCH ₃), 4.11 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 4.45 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 6.67 (s, 1H, pyrimidine H-4), 6.89–8.43 (m, 11H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	20c
¹ H NMR: δ = 1.23 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 1.42 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 2.88 (s, 3H, CH ₃), 4.11 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 4.45 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 5.91 (s, 2H, OCH ₂ O), 6.67 (s, 1H, pyrimidine H-4), 6.89–8.43 (m, 11H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	21 a
¹ H NMR: $\delta = 1.23$ (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 1.42 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 2.88 (s, 3H, CH ₃), 4.11 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 4.45 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 6.67 (s, 1H, pyrimidine H-4), 6.89–8.43 (m, 13H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	21b
¹ H NMR: $\delta = 1.23$ (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 1.42 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 2.88 (s, 3H, CH ₃), 3.47 (s, 3H, OCH ₃), 3.72(s, 3H, OCH ₃), 4.11 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 4.45 (q,	
2H, J = 7.5 Hz, CH ₂ CH ₃), 6.67 (s, 1H, pyrimidine H-4), 6.89–8.43 (m, 11H, ArH's) IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C),	21c
¹⁴⁷⁶ (CH ₂), 1663 (CH ₃). ¹ H NMR: δ = 1.23 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 1.42 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 2.88 (s, 3H, CH ₃), 4.11 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 4.45 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 5.91 (s, 2H, CH ₃) (c) (c) (c) (c) (c) (c) (c) (c) (c) (c	
OCH ₂ O), 6.67 (s, 1H, pyrimidine H-4), 6.89–8.43 (m, 11H, ArH's) IR: 3382 (NH ⁾ , 3050 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1682 (CO), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	26
¹ H NMR: δ = 1.39 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 4.42 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.42–8.35 (m, 13H, ArH's), 11.42 (s, br., 1H, NH)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	29a
¹ H NMR: $\delta = 1.39$ (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 4.42 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.06–8.38 (m, 13H, ArH's)	
IR: 3055 (CH,aromatic), 2983 (CH, aliphatic), 1728 (CO, ester), 1628 (C=N), 1603 (C=C), 1476 (CH ₂), 1663 (CH ₃).	29b
¹ H NMR: δ = 1.39 (t, 3H, J = 7.5 Hz, CH ₃ CH ₂), 4.44 (q, 2H, J = 7.5 Hz, CH ₂ CH ₃), 7.08–8.23 (m, 13H, ArH's)	

Synthesis of Ethyl 2-(Azanitrosomethylene)-3-(benzoxazol-2-ylphenyl)-1,3,4-thiadiazoline-5-carboxylate (5a), Ethyl 2-(Azanitrosomethylene)-3-(benzothiazol-2-ylphenyl)-1,3,4-thiadiazoline-5-carboxylate (5b), and Ethyl 2-(Azanitrosomethylene)-3-(benzoxazol-2-ylphenyl)-1,3,4-selenadiazoline-5-arboxylate (5c)

A cold saturated solution of sodium nitrite (10 mL) was added dropwise to a solution of the appropriate $4\mathbf{a}-\mathbf{c}$ (1 g) in acetic acid (20 mL) in an ice bath while stirring. The reaction mixture was stirred for 30 min. The resulting solid was collected, washed with water, and crystallized from acetone to give $5\mathbf{a}-\mathbf{c}$, respectively (Tables I and II).

Synthesis of Ethyl 3-(4-Benzoxazol-2-ylphenyl)-2-oxo-1,3,4-thiadiazoline-5-carboxylate (7a,) 3-(4-Benzothiazol-2-ylphenyl)-2-oxo-1,3,4-thiadiazoline-5-carboxylate (7b), and Ethyl 3-(4-Benzoxazol-2-ylphenyl)-2-oxo-1,3,4-selenadiazoline-5-carboxylate (7c)

A solution of the appropriate 5a-c (0.5 g) in xylene (20 mL) was refluxed for 15 min. Then the solvent was evaporated under reduced pressure. The residue oil was triturated with petroleum ether (40–60°C), and the solid formed was collected and crystallized from the proper solvent to give 1,3,4-thiadiazolinone **7a,b** and 1,3,4- selenadiazolinone **7c**, respectively (Tables I and II).

Acylation of 4a-c

Acetylation. A mixture of the appropriate $4\mathbf{a}-\mathbf{c}$ (1 g) in acetic acid (10 mL) and acetic anhydride (5 mL) was warmed for 5 min at 70°C. The reaction mixture was poured onto ice water (40 mL). The solid was collected and crystallized to give the *N*-acetyl derivatives **8a–c**, respectively (Tables I and II).

Benzoylation. 4-Methybenzoyl chloride (1 mL) was added to a solution of the appropriate 4a-c (0.5 g) in pyridine (15 mL), and the mixture was refluxed for 10 min, then poured onto ice water (50 mL) and acidified with hydrochloric acid. The resulting product was collected and washed several times with boiling water. The solid was recrystallized from *N*,*N*-dimethylformamide to give the *N*-4-methylbenzoyl derivatives **9a–c**, respectively (Tables I and II).

Synthesis of Thiadiazolines 13a-g and 14a-g

Triethylamine (0.75 mL, 0.005 mol) was added dropwise with stirring to a mixture of the appropriate alkyl carbodithioates 10a-g(a,b) or 101a-g(a,b) (0.005 mol) and compound 2 (1.8 g, 0.005 mol) in ethanol (20 mL). The resulting solid, which formed after 30 min, was collected and recrystallized from the proper solvent and gave the corresponding thiadiazolines 13a-g and 14a-g, respectively, in a good yield (Tables I and II).

Synthesis of 1,2,4-Triazolo[4,3-a]pyrimidines 20a-c and 21a-c

Method A. A mixture of the appropriate hydrazonoyl chlorides **2a,b** (0.005 mol) and the appropriate **17a–c** (1.9 g, 0.005 mol) in chloroform (20 mL) containing triethylamine (0.75 mL, 0.005 mol) was refluxed for 10 h. Chloroform was evaporated under reduced

pressure, and the residue solid was recrystallized from chloroform/methanol mixture to give **20a–c** and **21a–c**, respectively (Tables I and II).

Method B. Equimolar amounts of the appropriate hydrazonoyl chlorides **2a,b**, **22a–c**, and sodium ethoxide (0.005 mol each) in ethanol (20 mL) were refluxed for 3 h. The reaction mixture was cooled, and the resulting solid was collected and recrystallized from the proper solvent to give products identical in all respects (mp, mixed mp, and spectra) with corresponding products obtained by method A.

Synthesis of Ethyl 4-(4-(Benzo[d]oxazol-2-yl)phenyl)-5-(benzoylimino)-4,5-dihydro-1,3,4-thiadiazole-2-carboxylate (26)

Method A. Triethylamine (0.75 mL, 0.005 mol) was added dropwise with stirring to a mixture of methyl benzoylcarbodithioate (**24**) (0.005 mol) and compound **2a** (1.8 g, 0.005 mol) in ethanol (20 mL). The resulting solid, which formed after 30 min, was collected and recrystallized from ethanol to give **26** (Tables I and II).

Method B. Triethylamine (0.75 mL, 0.005 mol) was added dropwise with stirring to a mixture of 5-phenyl-1,3,4-oxadiazole-2-thiol (**27**) (0.005 mol) and compound **2a** (1.8 g, 0.005 mol) in ethanol (20 mL). The resulting solid, which formed after 30 min, was collected and recrystallized from ethanol to give **26** (Tables I and II).

Synthesis of Ethyl 4-(4-(Benzo[d]oxazol-2-yl)phenyl)-4,5-dihydro-5-(phenylimino)-1,3,4-thiadiazole-2-carboxylate (29a) and Ethyl 4-(4-(Benzo[d]thiazol-2-yl)phenyl)-4,5-dihydro-5-(phenylimino)-1,3,4-thiadiazole-2-carboxylate (29b)

Method A. Triethylamine (0.75 mL, 0.005 mol) was added dropwise with stirring to a mixture of methyl phenylcarbamodithioate (**28**) (0.005 mol) and the appropriate amount of hydrazonoyl chlorides **2a** and **2b** (1.8 g, 0.005 mol) in ethanol (20 mL) for 30 min. The resulting solid was collected and recrystallized from acetic acid to give **29a** and **29b**, respectively (Tables I and II).

REFERENCES

- 1. A. O. Abdelhamid, E. A. K. Abdelall, and Y. H. Zaki, J. Heterocycl. Chem., 47, 477 (2010).
- 2. A. Padwa, Angew. Chem., Int. Ed. Engl., 15, 123 (1976).
- 3. R. Huisgen, R. Sustmann, and G. Wallbillich, Chem. Ber., 100, 1786 (1976).
- 4. A. O. Abdelhamid and F. A. Attaby, J. Heterocycl. Chem., 28, 41 (1991).
- D. I. Kornis, In *Comprehensive Heterocyclic Chemistry*, A. R. Katritzky, C. W. Rees, and E. F. V. Scriven, Eds., (Pergamon, Oxford, UK, 1996), vol. 4.
- I. Hutchinson, M.-S. Chua, H. L. Browne, V. Trapani, T. D. Bradshaw, A. D. Westwell, and M. F. G. Stevens, *J. Med. Chem.*, 44, 1446 (2001).
- D.-F. Shi, T. D. Bradsaw, S. Wrigley, C. J. McCall, P. Lelieveld, I. Fichtner, and M. F. G. Stevens, J. Med. Chem., 39, 3375 (1996).
- T. D. Bradsaw, S. Wrigley, D.-F. Shi, R. J. Schuitz, K. D. Paull, and M. F. G. Stevens, *Br. J. Cancer*, 77, 745 (1998).
- T. D. Bradsaw, D.-F. Shi, R. J. Schuitz, K. D. Paull, L. Kelland, A. Wilson, H. H. Fiebig, S. Wrigley, and M. F. G. Stevens, *Br. J. Cancer*, 78, 421 (1998).
- M.-S. Chua, D.-F. Shi, S. Wrigley, T. D. Bradsaw, L. Hutchinson, P. N. Shaw, D. A. Barrett, L. A. Stanley, and M. F. G. Stevens, *J. Med. Chem.*, 42, 381 (1999).

- 11. A. O. Abdelhamid, H. F. Zohdi, and N. A. Ali, Molecules, 5, 961 (2001).
- 12. H. F. Zohdi, N. M. Rateb, M. M. M. Sallam, and A. O. Abdelhamid, J. Chem. Res. (S), 472 (1998); (M), 3329 (1998).
- A. O. Abdelhamid, S. M. Abdelgawad, and S. F. El-Sharnoby, *Phosphorus, Sulfur, and Silicon*, 177, 2699 (2002).
- A. O. Abdelhamid, M. A. Mahmoud, and Y. H. Zaki. *Phosphorus, Sulfur, and Silicon*, 183, 1746 (2008).
- 15. A. O. Abdelhamid and M. A. Afifi, Synth. Commun., 40, 1539 (2010).
- N. M. Rateb, N. A. Abdel-Riheem, A. A. Al-Atoom, A. O. Abdelhamid, *Phosphorus, Sulfur, and Silicon*, **178**, 1101 (2002).
- 17. W. F. Mclimans, E. V. Davis, F. L. Glover, and G. W. Rake, J. Immunol., 79, 428 (1975).
- 18. M. Busch and M. Starke, J. Prakt. Chem., 93, 49 (1916).
- 19. L. J. Rubenstein, J. Chem. Soc., 127, 1998 (1925).