

Asymmetric Synthesis of Unsaturated Pipecolic Acid Derivatives

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Abstract: Two enantiopure pipecolic acid derivatives having an olefinic moiety were synthesized from an intramolecular addition of allyl silanes on an iminium double bond. The iminium moiety was generated via a reaction of chiral β -amino alcohols with glyoxal.

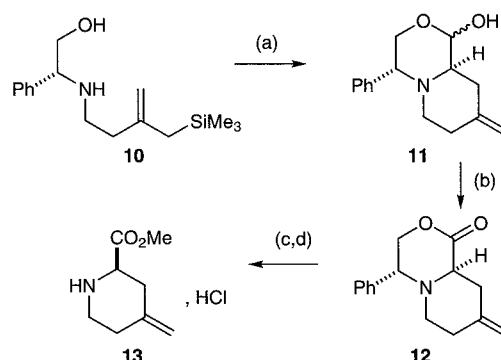
Due to their central role in chemistry and biology, α -amino acids are still synthetic targets of great interest.¹ In particular, much attention has been focused on the asymmetric synthesis of substituted pipecolic acids.² Reports of asymmetric syntheses of unsaturated derivatives of such compounds are very scarce,^{3,4} in spite of the great importance of olefinic amino acids which are interesting synthetic materials due to the possible functionalisation of the double bond.⁵ We describe here an efficient synthesis of two alkenyl derivatives of (*R*)-pipecolic acid.

The present method is based on a totally stereoselective addition of allyl silanes on an iminium moiety,⁶ which results from a condensation between glyoxal and a *N*-substituted (*R*)-phenylglycinol.⁷

Unsaturated ω -hydroxy silane **1**⁸ was reacted, via its mesyl derivative **2**, with (*R*)-phenylglycinol and the resulting product **3** yielded amino thioether **4** by treatment with glyoxal in the presence of thiophenol. Protection of the hemiketal moiety of **4** was followed by Lewis acid (ZnBr_2) mediated formation of an intermediate iminium ion whose intramolecular reaction with the unsaturated silyl moiety, followed by a

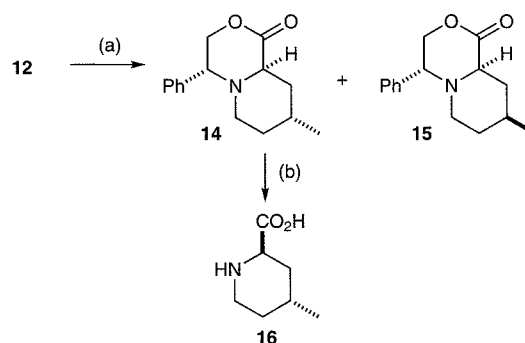
Swern oxidation, yielded lactone **7** whose debenzylation (by formation of the vinylcarbamate **8**) and methanolysis eventually furnished the unsaturated pipecolic derivative **9** (Scheme 1). The structure of the diastereomerically pure lactone **7** was determined by using ^1H NMR measurements.⁹

Methyl (*R*)-4-methylene pipecolate **13** was obtained¹⁰ in a similar way (Scheme 2) starting from substrate **10**. Reaction of this compound¹¹ with glyoxal directly afforded bicyclic derivative **11** which was submitted to a Swern oxidation. The resulting diastereomerically pure lactone **12** was then treated as described above in the case of lactone **7**.



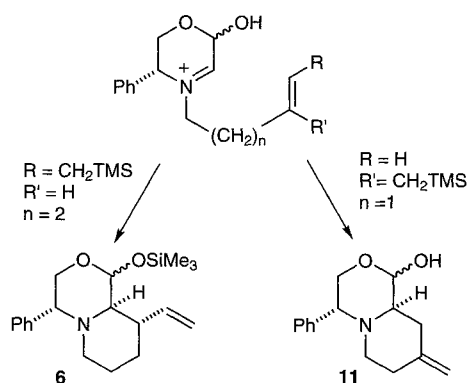
Scheme 2. a) OHC-CHO, THF:H₂O (50:50), rt, 97%; b) (COCl)₂, DMSO, Et₃N, -50 to 0°C, 85%; c) CH₂=CH-OCOCN, CH₂Cl₂, reflux, 70%; d) 5N HCl in MeOH, reflux, 95%.

The absolute configuration of pipecolic derivative **13** was deduced from a chemical correlation with compound **16** which has already been described in enantiopure form.¹² (2*R*,4*R*)-Methylpipecolic acid **16** was obtained from compound **14** (Scheme 3) which is the major (75%) isomer produced by the hydrogenation of lactone **12**.



Scheme 3. a) H₂, PtO₂, AcOEt, 96%; b) H₂, Pd(OH)₂, EtOH, 98%.

Scheme 1. a) MsCl, Et₃N, CH₂Cl₂, 98%; b) (*R*)-phenylglycinol, (*i*-Pr)₂EtN, MeCN, 70%; c) OHC-CHO, THF:H₂O (50:50), then PhSH, 98%; d) TMSCl, Et₃N, THF, 70%; e) ZnBr₂, CH₂Cl₂, 51%; f) *n*-Bu₄NF, THF, 80%; g) (COCl)₂, DMSO, Et₃N, -50 to 0°C, 70%; h) CH₂=CH-OCOCN, CH₂Cl₂, reflux, 85%; i) 5N HCl in MeOH, reflux, 95%.

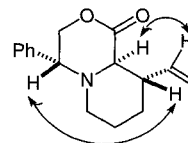


Scheme 4

In conclusion, we have developed a novel and totally stereoselective method for the asymmetric synthesis of olefinic pipecolic acid derivatives.

References and Notes

- Duthaler R.O. *Tetrahedron*, **1994**, 50, 1539.
- See, for recent examples: Jackson, R.F.W.; Graham, L.J.; Rettie, A.B. *Tetrahedron Lett.*, **1994**, 35, 4417; Berrien, J.F.; Royer, J.; Husson, H.P. *J. Org. Chem.*, **1994**, 59, 3769; Golubev, A.; Sewald, N.; Burger, K. *Tetrahedron Lett.*, **1995**, 36, 2037; Davies, C.E.; Heightman, T. D.; Hermitage, S.A.; Moloney, M.G. *Synth. Commun.*, **1996**, 26, 687; Agami, C.; Couty, F.; Mathieu, H. *Tetrahedron Lett.*, **1996**, 37, 4001.
- For the use of baikanine, the naturally occurring Δ^4 -unsaturated (S)-pipecolic acid, see: Fujita, Y.; Irreverre, F.; Witcop, B. *J. Am. Chem. Soc.*, **1964**, 86, 1844; Callens, R.E.A.; Anteunis, M.J.O. *Bull. Soc. Chim. Belg.*, **1982**, 91, 713; Hanson, G.J.; Russel, M.A. *Tetrahedron Lett.*, **1989**, 30, 5751.
- Ojima, I.; Tzamarioudaki, M.; Egushi, M. *J. Org. Chem.*, **1995**, 60, 7078.
- Baldwin, J.E.; Flinn, A. *Tetrahedron Lett.*, **1987**, 28, 3605; Browterman, Q.B.; Kaptein B.; Kamphuis, J.; Schoemaker, H.E. *J. Org. Chem.*, **1992**, 57, 6286; Baumann, H.; Duthaler, R. *Helv. Chim. Acta*, **1988**, 71, 1025.
- Hiemstra, H.; Speckamp, W.N. *Tetrahedron Lett.*, **1983**, 24, 1407; Esch, P.M.; Boska, I.M.; Hiemstra, H.; de Boer, R.F.; Speckamp, W.N. *Tetrahedron*, **1991**, 47, 4039; see also: Tietze, L.F.; Bratz, M. *Synthesis*, **1989**, 439.
- Agami, C.; Bihan, D.; Puchot-Kadouri, C. *Tetrahedron*, **1996**, 52, 9079.
- Hiemstra, H.; Sno, M.H.A.M.; Vijn, R.J.; Speckamp, W.N. *J. Org. Chem.*, **1985**, 50, 4014.
- Data for compound 9: $[\alpha]_{20}^D$: -10 (c 0.5, H₂O); ¹H NMR (250 MHz, D₂O): 1.35-1.82 (m, 4H), 2.39 (dq, J=3.1 and 11.2 Hz, 1H), 2.87 (td, J=3.2 and 12.5 Hz, 1H), 3.21-3.33 (m, 1H), 3.62 (s, 3H), 3.71 (d, J=11.2 Hz, 1H), 4.81-5.04 (m, 2H), 5.50-5.65 (m, 1H); ¹³C NMR (63 MHz, D₂O): 22.3, 29.8, 43.6, 45.0, 54.6, 62.3, 119.2, 142.4, 170.9. ¹H DIFNOE NMR (250 MHz) data of intermediate lactone 7 are as follows:



- Data for compound 13: $[\alpha]_{20}^D$: -26 (c 1.5, H₂O); ¹H NMR (250 MHz, D₂O): 2.30-2.48 (m, 3H), 2.77 (dd, J=4.1 and 14.5 Hz, 1H), 2.98 (td, J=4.1 and 11.8 Hz, 1H), 3.35-3.43 (m, 1H), 3.72 (s, 3H), 4.02 (dd, J=4.1 and 11.4 Hz, 1H), 4.91 (bs, 2H); ¹³C NMR (63 MHz, D₂O): 30.8, 34.5, 45.6, 54.9, 58.5, 114.9, 139.0, 170.7.
- Miguel, D.; Diez, A.; Blache, Y.; Luque, J.; Orozco, M.; Remuson, R.; Gelas-Mialhe, Y.; Rubiralta, M. *Tetrahedron*, **1995**, 51, 7527.
- Data for compound 16: ¹H NMR (250 MHz, D₂O): 0.89 (d, J=6.4 Hz, 3H), 1.28-1.76 (m, 4H), 1.94-2.02 (m, 1H), 3.10-3.14 (m, 2H), 3.78 (dd, J=4.7 and 3.1 Hz, 1H); ¹³C NMR (63 MHz, D₂O): 20.2, 26.5, 30.2, 33.8, 41.9, 56.7, 175.4; $[\alpha]_{20}^D$: -20 (c 0.25, H₂O); [lit.¹³ $[\alpha]_{20}^D$: -18 (c 1, 2N HCl)].
- Okamoto, S.; Hijikata, A.; Kikumoto, R.; Tonomura, S.; Hara, H.; Ninomiya, K.; Maruyama, A.; Sugano, M.; Tamao, Y. *Biochem. Biophys. Res. Commun.*, **1981**, 101, 440.