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Syntheses of spacer-armed carbohydrate model compounds ^{1,2}

János Kerékgyártó, Zoltán Nagy, Zoltán Szurmai *

Institute of Biochemistry, L. Kossuth University, P.O. Box 55, H-4010 Debrecen, Hungary

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Abstract

Commercially available chemicals, such as diethylene glycol, 1,9-nonanediol, 9-decen-1-ol, 1,2,6-hexanetriol, and *p*-nitrophenol were used to prepare spacer-armed carbohydrate derivatives. Glycosides of D-glucose, *N*-acetyl-D-glucosamine and 3-*O*-methyl-D-glucose have been synthesized, carrying reactive groups at the end of the spacer-arms. These glycosides are capable of forming neoglycoproteins. When bromo sugars were reacted with highly apolar aglycones in the presence of mercuric cyanide or mercuric bromide, a considerable amount of '2-OH-compounds' (such as 9-hydroxynonyl 3,4,6-tri-*O*-acetyl- β -D-glucopyranoside) were formed. Some spacer-armed derivatives (such as 9,10-epoxydecyl β -D-glucopyranoside) are theoretically a mixture of diastereomers, but this fact does not mirror in the ¹H and ¹³C NMR spectra. © 1997 Elsevier Science Ltd.

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1. Introduction

Spacer-armed carbohydrate derivatives capable of forming neoglycoconjugates have been found to be useful in biological research and in the last few decades a great number of neoglycoproteins have been prepared for immunological studies [1-12]. Carbohydrate-specific antibodies raised against these semi-artificial antigens have been investigated [13,14]. Purification of antibodies and lectins were possible

by means of affinity chromatography after incorporating spacer-armed glycosides into solid matrixes [15]. The chemical syntheses of oligosaccharides related to different bacteria is one of the ongoing research projects in this laboratory [16–19]. Oligosaccharides which are only haptens must be attached to suitable immunogens, in most cases to proteins to prepare 'artificial' antigens.

In order to create covalent linkages the spacers must have two functional groups. The carbohydrate moiety can be easily attached to the bridge molecule as a glycoside and the other end of the spacer must contain a reactive group. There are many methods in the literature [1] for the preparation of neoglycoproteins, but in the case of complex oligosaccharides the number of methodologies is limited. In the past twenty years the 8-methoxycarbonyloctyl [3] and the *p*-iso-

^{*} Corresponding author.

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thiocyanatophenyl glycosides [2] were the most popular for attaching oligosaccharide components of bacterial cell-wall or blood group substances. The first one has the ideal length and can react with amino groups via the corresponding acyl-azide and the latter can form a thiocarbamide linkage between the spacer and the protein. Bridge molecules bearing aldehydo terminal-groups seem to be the most useful. Reductive alkylation of amino groups is a rather simple procedure but the syntheses of suitable aldehydo derivatives are difficult [4,5]. The reducing end of an oligosaccharide can serve itself as a spacer by losing one monosaccharide unit [7].

Various reactive groups can be formed by applying short spacer-arms (e.g. 2-bromoethyl [20] or 2azidoethyl [21] glycosides) which may be elongated in a later stage of the reaction sequence. Over the last decade it has been demonstrated that 'oxa' or 'dioxa' spacers prepared from diethylene or triethylene glycols [8,9,22–24] are very useful for the preparation of neoglycoproteins.

In this paper we report on the syntheses of a series of interesting spacer-armed glycosides related to D-glucose, *N*-acetyl-D-glucosamine, and 3-*O*-methyl-D-glucose.

2. Results and discussion

Commercially available chemicals, such as diethylene glycol (1), 1,9-nonanediol (2), 9-decen-1-ol (3), 1,2,6-hexanetriol (4), and p-nitrophenol (5) were used to prepare spacer-armed carbohydrate derivatives. Us-

Table 1

| ¹³ C [| NMR | data | for | spacered | D-glucose | derivatives | in | CD, | OD |
|-------------------|-----|------|-----|----------|-----------|-------------|----|-----|----|
|-------------------|-----|------|-----|----------|-----------|-------------|----|-----|----|

ing 1-5, various reactive end-groups, such as methoxycarbonyl, carboxyl, carbonyl, epoxide, and p-isothiocyanate, have been developed in order to produce neoglycoproteins. Monoglycosylation of 1 and 2, followed by Jones oxidation would give methoxycarbonyl-alkyl spacers, respectively. After glycosylation of 3 and removal of the protecting groups, the double bond can be ozonolyzed [5] to give an aldehyde. In an alternative route an epoxide may be formed and used directly in the neoglycoprotein synthesis. The reaction of the appropriately protected 4 with a suitable glycosyl donor, followed by deprotection may give a derivative bearing a vicinal diol portion. Oxidation with sodium periodate then may produce an aldehyde for the reductive alkylation reaction with proteins [1,4,5]. Conversion of pnitrophenyl glycosides to neoglycoproteins via p-isothiocyanatophenyl derivatives are well-documented [2,25].

Diols 1 and 2 were reacted, separately, with 2,3,4,6-tetra-*O*-acetyl- α -D-glucopyranosyl bromide (6). The mercuric bromide-promoted reaction [26] of 1 gave the crystalline monoglucoside 8 in an acceptable yield (61%). Compound 2 was not soluble in dichloromethane, so the reaction was performed under Helferich conditions at 50 °C. It should be noted that during our investigations Japanese authors also used 1,9-nonanediol to form an 8-methoxycarbonyloctyl spacer [27]. Column chromatography of the crude product gave 9 (62%) and an unexpected by-product which turned out to be 9-hydroxynonyl 3,4,6-tri-*O*-acetyl- β -D-glucopyranoside (10). It is known from the literature that under Koenigs–Knorr condi-

| Carbon | Compound | | | | |
|---------------------|----------|-------------|-------------|-------------|--|
| | 18 | 19 | 20 | 21 | |
| C-1 | 104.37 | 104.49 | 104.43 | 104.32 | |
| C-2 | 74.98 | 75.24 | 75.14 | 75.08 | |
| C-3 | 77.86 | 78.24 | 78.14 | 78.09 | |
| C-4 | 71.78 | 71.79 | 71.67 | 71.64 | |
| C-5 | 77.86 | 78.00 | 77.90 | 77.85 | |
| C-6 | 62.71 | 62.96 | 62.88 | 62.76 | |
| $-COOCH_{2}$ | 52.26 | 52.28 | | | |
| C=O , | 172.77 | 176.14 | | | |
| -OCH ₂ - | 71.53 | 71.04 | 71.06 | 70.85 | |
| 4- | 69.68 | | | | |
| | 69.04 | | | | |
| $-CH_2-$ | | 35.01-26.21 | 35.07-27.27 | 33.54-27.02 | |
| $-CH^{2}$ | | | 140.21 | | |
| $=CH_2$ | | | 115.05 | | |
| –CH(Ô)– | | | | 53.48 | |

tions the '2-OH' compounds may be formed [28] because of the rearrangement of the corresponding orthoesters [29]. When **3** was coupled with **6** in the presence of mercuric bromide in dichloromethane, the expected product **11** and the 'by-product' **12** were isolated in almost equal amounts (32 and 29.9%). It is noteworthy that under the same conditions 2,4,6-tri-*O*-acetyl-3-*O*-allyl- α -D-glucopyranosyl bromide [30] and octanol yielded octyl 4,6-di-*O*-acetyl-3-*O*-allyl- α -D-glucopyranoside as the side-product [31].

The free hydroxyl group at the end of a spacer (e.g. in 8 and 9) can be converted into different reactive groups. Tosylation, azide displacement and reduction to amine is a well-known reaction sequence (the first 'dioxa-spacers' were prepared by Lipták in Munich, 1971 [8,9]). Oxidation seems to be a very simple way to generate an aldehyde function, but before deprotection of the sugar unit the reactive function has to be masked [10]. Jones oxidation of 8 and 9 gave 4-methoxycarbonyl-3-oxa-butyl and 8methoxycarbonyloctyl glycosides 13 and 14, respectively. The double bond in 11 was converted into the corresponding epoxide (15) with *m*-chloroperbenzoic acid. The β -anomeric configuration in compounds 8–15 was determined from their ¹H NMR data; all $J_{1,2}$ coupling constants were in the range of 7.4–8 Hz. Compound 15 is theoretically a mixture of diastereoisomers, but this fact is not mirrored in the ¹H NMR spectrum. Conversion of the allyl glucoside 16 [32] with *m*-chloroperbenzoic acid to the epoxide 17 vielded the mixture of the 2'S and 2'R diastereoisomers [33] in a molar ratio of 4:1, as determined from the ¹H NMR spectrum. The 'extra' chiral carbon (2')is so close to the sugar part that the isomers gave

different spectra. Zemplén deacylation of 13, 14, and 11 yielded 18, 19, and 20, respectively. In the case of the epoxides 15 and 17 *O*-deacylation, yielding 21 and 22, respectively, was performed with magnesium oxide in methanol [34]. For the 13 C NMR data of 18–21, see Table 1.

Chemical syntheses of oligosaccharide components of mycobacterial antigens is one of the ongoing research projects in this laboratory. 3-O-Methyl-Dglucose is a frequent building block in the surface antigens of different mycobacteria. In this context, spacer-armed derivatives of this sugar have also been prepared. Reaction of 2,4,6-tri-O-acetyl-3-O-methyl- α -D-glucopyranosyl bromide (7) [35] with 3,6-dioxa-7-ethoxycarbonylheptan-1-ol (23) [8,9] under Helferich conditions gave the corresponding β -glycoside (24). Subsequent deacylation with sodium methoxide (0.35 equiv) in methanol [16] yielded 3,6dioxa-7-methoxycarbonylheptyl 3-O-methyl- β -D-glucopyranoside (25). Saponification of 25 (0.05 M NaOH), followed by neutralization with Amberlite IR 120 H⁺ resin resulted in compound 26. The 13 C NMR spectra of 25 and 26 verified the structures: the chemical shifts of the anomeric carbons clearly demonstrated the β -anomeric configurations (103.3) and 102.1 ppm, respectively). Both 25 and 26 can be coupled to proteins by the acyl-azide [3] or one of the active-ester methods [6]. In parallel experiments, pnitrophenyl 3-O-methyl- β -D-glucopyranoside (27) [17] was hydrogenated over palladium on carbon, and converted into the crystalline *p*-isothiocyanatophenyl compound 28 with thiophosgene [25]. The conjugation of this type of glycoside to proteins is well-documented [2,11].



For the preparation of the glucoside **33** the hexanetriol **4** was converted into 5,6-diacetoxy-hexanol (**31**) as follows: compound **4** was reacted with 2,2-dimethoxypropane in the presence of *p*-toluenesulfonic acid and after the workup procedure TLC showed two spots. One (**29**) had the same chromatographic mobility as the reference compound (5*S*)-5,6-isopropylidene-dioxyhexanol, prepared according to [36]. It was obvious that the by-product was the mixed acetal **29a**. The mixture of **29** and **29a** was treated with acetic acid in dichloromethane to yield **29** (77%). Conventional benzylation, hydrolysis and then acety-

lation gave 1,2-diacetoxy-6-benzyloxy-hexane (**30**) which was hydrogenated (Pd–C, H₂) to afford 1,2diacetoxy-hexanol **31**. Compounds **30** and **31** were purified by column chromatography. 2,4,6-Tri-*O*acetyl-3-*O*-methyl- α -D-glucopyranosyl bromide (**7**) was coupled with **31** and the product (**32**) was deacylated to yield **33** (for ¹³C NMR data, see Table 2). Oxidation of the *vicinal* diol into the *aldehydo* derivative **34** allows to prepare neoglycoproteins by the reductive alkylation method [1]. The structures of the deprotected compounds **18–21**, **25**, **26**, and **33** were confirmed by ¹³C NMR spectroscopy.

| Table 2 | | | | | |
|---------------------|----------|------------|-----------------|-------|-------------|
| ¹³ C NMR | data for | - spacered | 3-O-methyl-D-gl | ucose | derivatives |

| Carbon | Compound | | | | | | |
|---------------------|-------------------------|--------------------------------------------------------|--------------------------|--------------------------|--|--|--|
| | 25 CDCl ₃ | 26 CDCl ₃ -CD ₃ OD 1:1 | 28 CD ₃ OD | 33 CD ₃ OD | | | |
| C-1 | 103.27 | 102.09 | 102.37 | 104.30 | | | |
| C-2 | 73.77 | 72.66 | 74.68 | 74.91 | | | |
| C-3 | 85.67 | 85.56 | 87.72 | 87.84 | | | |
| C-4 | 70.21 | 69.26 | 70.97 | 71.16 | | | |
| C-5 | 75.64 | 75.91 | 78.14 | 77.74 | | | |
| C-6 | 62.03 | 60.76 | 62.49 | 62.60 | | | |
| -OCH 3 | 60.36 | 59.61 | 61.08 | 61.04 | | | |
| $-COOCH_3$ | 51.67 | | | | | | |
| -OCH ₂ - | 70.75-68.72 | 69.26-67.75 | | 70.72 | | | |
| -CH ₂ - | | | | 34.04-23.06 | | | |
| -CH ₂ OH | | | | 67.31 | | | |
| -CH(OH)- | | | | 73.15 | | | |
| Aromatic | | | 127.74 | | | | |
| | | | 119.08 | | | | |



Preparation of neoglycoproteins using glycosides described in this paper will be published elsewhere.

3. Experimental

General.—Melting points (uncorrected) were determined on a Kofler hot-stage apparatus. Optical rotations were measured with a Perkin–Elmer 241 polarimeter. NMR spectra were recorded with a Bruker WP-200 SY spectrometer for solutions in CDCl₃ (internal Me₄Si) or in CD₃OD. The reactions were monitored by TLC on Kieselgel 60 F_{254} (Merck, Darmstadt) with detection by charring with sulfuric acid. Kieselgel 60 (Merck) was used for short-column chromatography. 1,9-Nonanediol (2), 9-decen-1-ol (3), and 1,2,6-hexanetriol (4) were ordered from Aldrich.

5-Hydroxy-3-oxa-pentyl 2,3,4,6-tetra-O-acetyl-β-D-glucopyranoside (8).—A mixture of 2,3,4,6-tetra-*O*-acetyl- α -D-glucopyranosyl bromide (**6**; 4.11 g, 10 mmol) and diethylene glycol (1; 10.6 g, 100 mmol) in dry CH₂Cl₂ (50 mL), containing activated powdered 4 Å molecular sieves (5 g), was stirred under Ar for 15 min. Then, HgBr₂ (3.60 g, 10 mmol) was added and stirring was continued at room temperature overnight. The mixture was diluted with CH₂Cl₂ (300 mL), filtered through Celite, and the filtrate was washed with aq 5% KI (3×30 mL) and water (3×30 mL), dried, and concentrated. The obtained crystalline mass was recrystallized from a mixture of EtOAc and cyclohexane to give 8 (2.68 g, 61%); mp 86–88 °C; lit. mp 87–88 °C [8,9]; $[\alpha]_{\rm D}$ –13.5° (c 0.8, CHCl₃); lit. $[\alpha]_{D} - 14^{\circ}$ (CHCl₃) [8,9].

9-Hydroxynonyl 2,3,4,6-tetra-O-acetyl- β -D-glucopyranoside (9).—A mixture of 1,9-nonanediol (2; 16

g, 100 mmol), $Hg(CN)_2$ (6.3 g, 25 mmol), and $HgBr_2$ (0.90 g, 2.5 mmol) in 1:1 dry toluene-nitromethane (300 mL) was concentrated to half its volume at atmospheric pressure. After cooling to 50 °C under Ar, compound 6 (8.22 g, 20 mmol) was added and the mixture was stirred for 2 h. After cooling, the solution was diluted with CH_2Cl_2 (150 mL), filtered, and concentrated. The residue was taken up in CH_2Cl_2 (300 mL) and the solution was filtered, washed with aq 5% KI (2×50 mL) and water (2×50 mL), dried, and concentrated. The product was subjected to column chromatography using 8:2 CH₂Cl₂-acetone as the eluent. The first product that eluted was 9 (6.12 g, 62.4%); $[\alpha]_{\rm D} = -7^{\circ}$ (c 0.76, CHCl₃); ¹H NMR (CDCl₃): δ 5.26–4.90 (m, 3 H, H-2,3,4), 4.48 (d, 1 H, $J_{1,2}$ 8 Hz, H-1), 4.32–4.03 (m, 2 H, H-6a,6b), 3.94-3.40 (m, 5 H, H-5 and 2 $-CH_2-O_-$), 2.10 (s, 3 H, Ac), 2.06 (bs, 1 H, OH, deuterable), 2.04, 2.03, and 2.01 (3 s, each 3 H, 3 Ac), 1.68–1.20 (m, 14 H, 7 –CH₂–). Compound 9 was used for the next step without further characterization. After mixed fractions (1.74 g), the next compound that eluted was 9-hydroxynonyl 3,4,6-tri-Oacetyl- β -D-glucopyranoside (10) (501 mg, 5.6%); mp 75–76 °C (from EtOH); $[\alpha]_{\rm D}$ + 5° (*c* 1.02, CHCl₃); ¹H NMR (CDCl₃): δ 5.11 and 5.02 (2 t, 2 H, H-3,4), 4.36 (d, 1 H, J_{1,2} 8 Hz, H-1), 4.32–4.03 (m, 2 H, H-6a,6b), 3.97–3.49 (m, 6 H, H-2,5 and 2 –CH₂–O–), 2.10 and 2.03 (2 s, 6,3 H, 3 Ac), 1.73-1.22 (m, 14 H, 7 –CH₂–). Anal. Calcd for $C_{21}H_{36}O_{10}$: C, 56.23; H, 8.09. Found: C, 56.35; H, 8.11.

9-Decen-1-yl 2,3,4,6-tetra-O-acetyl- β -D-glucopyranoside (11).—Compound 6 (2.06 g, 5 mmol) was coupled with 9-decen-1-ol (3; 1.09 g, 7 mmol) in the presence of HgBr₂ as described for the preparation of 8. The products were separated by means of column chromatography. The first compound that eluted as a syrup was 11 (778 mg, 32%); $[\alpha]_{D}$ -15.7° (c 2.2, CHCl₃); ¹H NMR (CDCl₃): δ 5.72 (m, 1 H, -CH=), 5.18-4.78 (m, 5 H, H-2,3,4 and =CH₂), 4.49 (d, 1 H, $J_{1,2}$ 7.4 Hz, H-1), 4.24–4.00 (m, 2 H, H-6a,6b), 3.78 (m, 1 H, $-O-CH_2-$), 3.61(m, 1 H, H-5), 3.38 (m, 1 H, -O-CH₂-), 1.98, 1.95, 1.92, and 1.90 (4 s, each 3 H, 4 Ac), 1.56–1.12 (m, 14 H, 7 $-CH_2$ -). Anal. Calcd for $C_{24}H_{38}O_{10}$: C, 59.24; H, 7.87. Found: C, 59.35; H, 7.97. The second product that eluted was 9-decen-1-yl 3,4,6-tri-Oacetyl- β -D-glucopyranoside (12) (665 mg, 29.9%); mp 58–60 °C (from EtOH); $[\alpha]_{D}$ + 12.7° (c 1.03, CHCl₃); ¹H NMR (CDCl₃): δ 5.82 (m, 1 H, –CH=), 5.23–4.88 (m, 4 H, H-3,4 and CH₂=), 4.36 (d, 1 H, J_{1,2} 7.8 Hz, H-1), 4.33–4.07 (m, 2 H, H-6a,6b), 3.92 $(m, 1 H, -O-CH_2-), 3.68 (m, 1 H, H-5), 3.63-3.48$ (m, 2 H, H-2 and -O-CH₂-), 2.39 (bs, 1 H, OH, deuterable), 2.09 and 2.03 (2 s, 6,3 H, 3 Ac), 1.71-1.22 (m, 14 H, 7 –CH₂–). Anal. Calcd for $C_{22}H_{36}O_9$: C, 59.44; H, 8.16. Found: C, 59.48; H, 8.20.

4-Methoxycarbonyl-3-oxa-butyl 2,3,4,6-tetra-Oacetyl- β -D-glucopyranoside (13).—To a solution of 8 (873 mg, 2 mmol) in 1:1 CH_2Cl_2 -acetone (40 mL) was added dropwise a solution of CrO₃ (500 mg, 5 mmol) in 3.5 M H_2SO_4 (2.5 mL) during 30 min at 0 °C. After an additional 30 min, the chilling was terminated and the mixture stirred for 2 h (TLC, 8:2 CH_2Cl_2 -acetone). EtOH (10 mL) was then added at 0 °C and, after 30 min the solid separated was filtered off and washed with acetone. NaHCO₃ (1 g) was added to the combined organic phases in small portions, and the suspension was concentrated. The residue was partitionated between CH₂Cl₂ (30 mL) and water (30 mL), and Bu_4NBr (1 g) and MeI (5 mL) were added. The mixture was vigorously stirred for 48 h, diluted with CH_2Cl_2 (30 mL), and the organic layer was separated, washed with water (3 \times 15 mL), dried, and concentrated. The residue was purified by column chromatography to yield 13 (643) mg, 69.2%); mp 69–70 °C (from EtOH); [α]_D – 25.2° $(c \ 0.49, \ CHCl_3);$ ¹H NMR $(CDCl_3): \delta \ 5.27-4.97$ (m, 3 H, H-2,3,4), 4.61 (d, 1 H, J_{1,2} 8 Hz, H-1), 4.31–3.63 (m, 9 H, H-5,6a,6b, 3 –CH₂–), 3.74 (s, 3 H, COOCH₃), 2.08, 2.04, 2.02, and 2.00 (4 s, each 3 H, 4 Ac). Anal. Calcd for $C_{19}H_{28}O_{13}$: C, 49.14; H, 6.08. Found: C, 49.04; H, 6.02.

8-Methoxycarbonyloctyl 2,3,4,6-tetra-O-acetyl-β-D-glucopyranoside (14).—Compound 9 (981 mg, 2 mmol) was converted into 14 (736 mg, 71%) as described for the preparation of 13; mp 61–63 °C (from EtOH); $[\alpha]_D - 14^\circ$ (*c* 1, CHCl₃); ¹H NMR (CDCl₃): δ 5.22–4.83 (m, 3 H, H-2,3,4), 4.40 (d, 1 H, $J_{1,2}$ 7.5 Hz, H-1), 4.24–3.99 (m, 2 H, H-6a,6b) 3.87–3.30 (m, 3 H, H-5 and –O–CH₂–), 3.58 (s, 3 H, COOCH₃), 2.21 (t, 2 H, –CH₂–COOMe), 2.03, 1.99, 1.98, and 1.94 (4 s, each 3 H, 4 Ac), 1.60–1.11 (m, 12 H, 6 –CH₂–). Anal. Calcd for C₂₄H₃₈O₁₂: C, 55.59; H, 7.39. Found: C, 55.47; H, 7.42.

9,10-Epoxydecyl 2,3,4,6-tetra-O-acetyl-β-D-glucopyranoside (15).—To a solution of 11 (973 mg, 2) mmol) in CH₂Cl₂ (20 mL) was added 3-chloroperbenzoic acid (800 mg). The mixture was stirred at room temperature for 3 days, then it was diluted with CH_2Cl_2 (100 mL), washed with aq 5% NaHCO₃ $(2 \times 20 \text{ mL})$ and water $(3 \times 20 \text{ mL})$, dried, and concentrated. The residue was subjected to column chromatography to give syrupy 15 (862 mg, 72%); $[\alpha]_{D}$ -15° (c 1.36, CHCl₃); ¹H NMR (CDCl₃): δ 5.18– 4.82 (m, 3 H, H-2,3,4), 4.39 (d, 1 H, $J_{1,2}$ 8 Hz, H-1), 4.23–3.98 (m, 2 H, H-6a,6b), 3.78 (m, 1 H, -O-CH₂-), 3.60 (m, 1 H, H-5), 3.38 (m, 1 H, -O-CH₂-), 2.82 (m, 1 H, -CH-O-), 2.68 and 2.38 (2 dd, each 1 H, -CH₂-O-), 2.00, 1.98, 1.96, and 1.93 (4 s, each 3 H, 4 Ac), 1.52-1.12 (m, 14 H, 7 -CH₂-). Anal. Calcd for C₂₄H₃₈O₁₁: C, 57.36; H, 7.62. Found: C, 57.56; H, 7.57.

2,3-Epoxypropyl 2-acetamido-3,4,6-tri-O-acetyl-2deoxy-β-D-glucopyranoside (17).—Allyl 2-acetamido-3,4,6-tri-O-acetyl-2-deoxy-β-D-glucopyranoside [32] (16; 775 mg, 2 mmol) was converted into 17 (600 mg, 74.4%) as described for the preparation of 15; mp 167–168 °C (from EtOAc-hexane); lit. mp 162–163 °C (from MeOH–Et₂O) [37]; ¹H NMR (CDCl₃): the product was a 1:4 mixture of the 2'*R* and 2'*S* diastereoisomers [33]. Characteristic data: δ 5.61 (d, 1 H, NH), 5.30–4.99 (m, 2 H, H-3,4), 4.76 (d, 0.2 H, $J_{1,2}$ 9 Hz, H-1 of the 2'*R* isomer), 4.62 (d, 0.8 H, $J_{1,2}$ 9 Hz, H-1 of the 2'*S* isomer), 4.33–4.08 (m, 2 H, H-6a,6b), 2.10, 2.04, and 1.98 (3 s, 3,6,3 H, 4 Ac). Anal. Calcd for C₁₇H₂₅NO₁₀: C, 50.62; H, 6.25. Found: C, 50.45; H, 6.30.

4-Methoxycarbonyl-3-oxa-butyl β -D-glucopyranoside (18).—To a solution of 13 (232 mg, 0.5 mmol) in dry MeOH (10 mL) was added a catalytic amount of NaOMe. After standing overnight (TLC, 85:15 CH₂Cl₂–MeOH) the solution was neutralized with Amberlite IR 120 (H⁺) resin, filtered, and concentrated, to yield syrupy 18 (139 mg, 94%); [α]_D – 25.6° (c 0.52, CHCl₃); for ¹³C NMR data, see Table 1. Anal. Calcd for C₁₁H₂₀O₉: C, 44.59; H, 6.80. Found: C, 44.75; H, 6.88.

8-Methoxycarbonyloctyl β -D-glucopyranoside (19).—Zemplén O-deacylation of 14 (259 mg, 0.5 mmol) as described for the preparation of **18** gave crystalline **19** (169 mg, 96%); mp 71–73 °C (from EtOAc–hexane); $[\alpha]_D = -24.1^\circ$ (*c* 1.14, MeOH); for ¹³C NMR data, see Table 1; Anal. Calcd for C₁₆H₃₀O₈: C, 54.84; H, 8.63. Found: C, 54.77; H, 8.59.

9-Decen-1-yl β -D-glucopyranoside (20).— Zemplén O-deacylation of compound 11 (122 mg, 0.25 mmol), followed by column chromatography (85:15 CH₂Cl₂-MeOH) of the crude product yielded 20, isolated as a syrup (72 mg, 90%); $[\alpha]_D - 19.9^\circ$ (*c* 0.98, MeOH); for ¹³C NMR data, see Table 1. Before recording the spectrum the sample was co-concentrated with CD₃OD.

9,10-Epoxydecyl β -D-glucopyranoside (21).—To a solution of 15 (503 mg, 1 mmol) in dry MeOH (40 mL) was added MgO (500 mg) [34]. The mixture was stirred for 80 h at room temperature, then filtered, and the filtrate was concentrated to give crystalline 21 (302 mg, 90%); mp 80–84 °C (from EtOAchexane); [α]_D – 24° (*c* 0.9, MeOH); for ¹³C NMR data, see Table 1. Anal. Calcd for C₁₆H₃₀O₇: C, 57.46; H, 9.04. Found: C, 56.67; H, 9.18.

2,3-Epoxypropyl 2-acetamido-2-deoxy-β-D-glucopyranoside (22).—Compound 17 (202 mg, 0.5 mmol) was deacylated with MgO as described for the preparation of 21. The crude product was subjected to column chromatography (7:3 CH₂Cl₂–MeOH) to yield crystalline 22 (72 mg, 52%); mp 152–155 °C (from EtOAc–MeOH); $[\alpha]_D - 32^\circ$ (*c* 0.70, MeOH); lit. mp 175–178 °C (acetone–H₂O); $[\alpha]_D - 37.4^\circ$ (H₂O) [37].

3,6-Dioxa-7-ethoxycarbonylheptyl 2,4,6-tri-Oacetyl-3-O-methyl- β -D-glucopyranoside (24).—A mixture of 3,6-dioxa-7-ethoxycarbonylheptan-1-ol [8,9] (**23**; 1.92 g, 10 mmol) and Hg(CN)₂ (2.53 g, 10 mmol) in 1:1 dry toluene-nitromethane (60 mL) was concentrated to half its volume at atmospheric pressure. After cooling, 2,4,6-tri-O-acetyl-3-O-methyl- α -D-glucopyranosyl bromide (7) (1.92 g, 5 mmol) [35] was added, and the mixture was stirred overnight at room temperature, then diluted with CH_2Cl_2 (150) mL), filtered, and the filtrate was washed with aq 5% KI $(2 \times 20 \text{ mL})$ and water $(3 \times 20 \text{ mL})$, dried (MgSO₄), and concentrated. Column chromatography (9:1 CH₂Cl₂-acetone) of the residue gave 24, isolated as a syrup (1.89 g, 76.4%); $[\alpha]_{\rm D} = 15.3^{\circ}$ (c 1.35; CHCl₃); ¹H NMR (CDCl₃): δ 5.15–4.97 (m, 2 H, H-2,4), 4.56 (d, 1 H, H-1), 4.35–4.13 (m, 6 H, CH_2 -COOEt, O-C H_2 -C H_3 , and H-6a,6b), 4.03-3.90 (m, 1 H, H-5), 3.85–3.47 (m, 9 H, 4 –CH₂– and H-3), 3.43 (s, 3 H, OMe), 2.13, 2.11, and 2.10 (3

s, each 3 H, 3 Ac), 1.31 (t, 3 H, $O-CH_2-CH_3$). Anal. Calcd for $C_{21}H_{34}O_{13}$: C, 51.01; H, 6.93. Found: C, 51.23; H, 6.96.

3,6-Dioxa-7-methoxycarbonylheptyl 3-O-methyl- β -D-glucopyranoside (25).—To a solution of 24 (1.30 g) in dry MeOH (50 mL) was added NaOMe (50 mg) [16], and the mixture was stored for 3 days at room temperature, then neutralized with HOAc, and concentrated. The residue was purified by column chromatography (85:15 CH₂Cl₂–MeOH) to give 25, isolated as a syrup (656 mg, 70.4%); [α]_D – 14.9° (*c* 1.03, MeOH); for ¹³C NMR data, see Table 2. Anal. Calcd for C₁₄H₂₆O₁₀: C, 47.45; H, 7.40. Found: C, 47.30; H, 7.38.

7-Carboxyl-3,6-dioxa-heptyl 3-O-methyl-β-D-glucopyranoside (**26**).—Compound **25** (177 mg, 0.5 mmol) was dissolved in 0.05 M NaOH (5 mL), and the solution was kept for 2 h at 60 °C. After cooling, the solution was neutralized with Amberlite IR 120 (H⁺) resin, concentrated, and the residue (121 mg, 71%) was purified by column chromatography (1:1 CH₂Cl₂–MeOH) to yield **26**, isolated as a syrup (60 mg, 35%); $[\alpha]_D - 13^\circ$ (*c* 0.44, MeOH); for ¹³C NMR data, see Table 2.

p-Isothiocyanatophenyl 3-O-methyl- β -D-glucopyranoside (28).—p-Nitrophenyl 3-O-methyl- β -Dglucopyranoside [17] (27; 63 mg, 0.2 mmol) was hydrogenated over Pd–C (20 mg) in aq 80% EtOH (10 mL) for 2 h, then the catalyst was filtered off. The pH of the filtrate was adjusted to 8 with BaCO₃ and maintained by adding BaCO₃ whilst thiophosgene (0.2 mL) was added, and the mixture was stirred for 1 h at room temperature. Filtration and concentration yielded a slightly coloured syrup which was purified by column chromatography (9:1 CH₂Cl₂– MeOH) to give 28 (49 mg, 75%); mp 153–156 °C; [α]_D – 63.6° (*c* 0.50, MeOH); for ¹³C NMR data, see Table 2. Anal. Calcd for C₁₄H₁₇NO₆S: C, 51.36; H, 5.24. Found: C, 51.50; H, 5.29.

5,6-Isopropylidene-dioxyhexanol (29).—A mixture of 1,2,6-hexanetriol (4; 1.34 g, 10 mmol), 2,2-dimethoxypropane (8 mL, 65 mmol) and *p*-toluenesulfonic acid (50 mg) was stirred at room temperature for 3 h. The reaction was terminated with aq 5% NaHCO₃ (20 mL). The mixture was extracted with CH₂Cl₂ (100 mL), and the organic phase was washed with water (3 × 15 mL), dried, and concentrated. After this workup procedure, TLC (9:1 CH₂Cl₂– MeOH) of the product showed two spots, but only one had the same chromatographic mobility as the reference compound (5*S*)-5,6-isopropylidene-dioxyhexanol, prepared according to [36]. The product was dissolved in CH₂Cl₂ (30 mL), aq 60% HOAc was added, and the mixture was stirred at room temperature until the mixed acetal (**29a**) had been converted into **29**. The solution was then neutralized as described above, dried, and concentrated to give **29**, isolated as a syrup (1.35 g, 77.5%); ¹H NMR (CDCl₃): δ 4.18–3.98 (m, 2 H, H-1a,1b), 3.68–3.44 (m, 3 H, H-5,6a,6b), 2.75 (bs, 1 H, OH, deuterable), 1.77–1.43 (m, 6 H, 3 –CH₂–), 1.42 and 1.36 (2 s, each 3 H, CMe₂).

1,2-Diacetoxy-6-benzyloxy-hexane (30).—A mixture of 29 (3.48 g, 20 mmol), dry DMF (15 mL), powdered KOH (2.8 g), and benzyl bromide (2.85 mL, 24 mmol) was stirred vigorously at room temperature for 3 h, then diluted with CH₂Cl₂ (150 mL). Inorganics were filtered off, and the filtrate was washed with water $(3 \times 40 \text{ mL})$, dried, and concentrated. The residue (5.1 g) was suspended in aq 60% HOAc (30 mL) and kept for 30 min at 60 °C (TLC, 8:2 CH_2Cl_2 -acetone). After concentration and coconcentration with toluene $(3 \times 20 \text{ mL})$, the crude product was conventionally acetylated in 1:1 pyridine-Ac₂O (60 mL) at room temperature. After standing overnight, the mixture was concentrated and co-concentrated with toluene $(3 \times 20 \text{ mL})$, and the crude product was purified by column chromatography (97:3 CH₂Cl₂-EtOAc) to yield syrupy **30** (2.53 g, 41%); [']H NMR (CDCl₃): δ 7.38–7.22 (m, 5 H, Ph), 5.05 (m, 1 H, -CH-OAc), 4.48 (s, 2 H, PhCH₂), 4.27-3.96 (m, 2 H, -CH₂-OAc), 3.47 (t, 2 H, O-CH₂-), 2.04 (s, 6 H, 2 Ac), 1.70-1.32 (m, 6 H, 3 -CH ,-).

1,2-Diacetoxyhexan-6-ol (**31**).—A mixture of **30** (617 mg, 2 mmol), EtOH (10 mL), HOAc (1 mL), and Pd–C (50 mg) was stirred in an H₂ atmosphere overnight. The mixture was filtered through Celite, the filtrate was concentrated and co-concentrated with toluene (3×10 mL), and the residue was purified by column chromatography (97:3 CH₂Cl₂–EtOAc) to give syrupy **31** (360 mg, 82%); ¹H NMR (CDCl₃): δ 5.08 (m, 1 H, –CH–OAc), 4.28–3.97 (m, 2 H, –CH₂–OAc), 3.64 (t, 2 H, –CH₂–O–), 2.08 (s, 6 H, 2 Ac), 1.70–1.30 (m, 7 H, 3 –CH₂– and OH, deuterable).

5.6-Diacetoxyhexyl 2,4.6-tri-O-acetyl-3-O-methyl- β -D-glucopyranoside (**32**).—Compound **7** (383 mg, 1 mmol) was reacted with **31** (327 mg, 1.5 mmol) as described for the preparation of **8**. Column chromatography of the crude product gave **32**, isolated as a syrup (162 mg, 31%); $[\alpha]_D - 11^\circ$ (*c* 0.60, CHCl₃); ¹H NMR (CDCl₃): δ 5.15–4.91 (m, 3 H, H-2,4 and -CH–OAc), 4.38 (d, 1 H, $J_{1,2}$ 8 Hz, H-1), 3.39 (s, 3 H, OMe), 2.14–2.04 (m, 15 H, 5 Ac), 1.70–1.22 (m, 6 H, 3 –CH₂–). Anal. Calcd for $C_{23}H_{36}O_{13}$: C, 53.07; H, 6.97. Found: C, 52.80; H, 6.93.

5,6-Dihydroxyhexyl 3-O-methyl-β-D-glucopyranoside (**33**).—To a solution of **32** (104 mg, 0.2 mmol) in dry MeOH (10 mL) was added NaOMe (27 mg, 0.5 mmol). The mixture was kept at room temperature for 2 days, then neutralized with Amberlite IR 120 (H⁺) resin, filtered, and concentrated. Column chromatography of the crude product gave **33** (49 mg, 79%); $[\alpha]_D = 10.5^\circ$ (*c* 1.6, MeOH); for ¹³C NMR data, see Table 2. Anal. Calcd for C₁₃H₂₆O₈: C, 50.31; H, 8.45. Found: C, 50.52; H, 8.41.

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