This article was downloaded by: [Georgetown University] On: 20 May 2013, At: 12:35 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: <u>http://www.tandfonline.com/loi/lsyc20</u>

Epoxides from Myrcene: New Versatile Tools for the Synthesis of Functionalized Acyclic Terpenoids

V. Fauchet $^{\rm a}$, B. Arreguy-San Miguel $^{\rm b}$, M. Taran $^{\rm b}$ & B. Delmondo $^{\rm a}$

^a Laboratoire de Chimie des Substances Végétales, Institut du Pin, Université Bordeaux 1, 351, cours de la Liberation, 33405, Talence Cedex, France

^b Unité d'Enseignement et de Recherche des Sciences Pharmaceutiques Universite, Bordeaux 2 Published online: 17 Sep 2007.

To cite this article: V. Fauchet , B. Arreguy-San Miguel , M. Taran & B. Delmondo (1999): Epoxides from Myrcene: New Versatile Tools for the Synthesis of Functionalized Acyclic Terpenoids, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 29:21, 3673-3684

To link to this article: http://dx.doi.org/10.1080/00397919908086005

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

EPOXIDES FROM MYRCENE: NEW VERSATILE TOOLS FOR THE SYNTHESIS OF FUNCTIONALIZED ACYCLIC TERPENOIDS

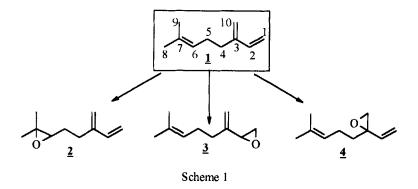
V. Fauchet^a, B. Arreguy-San Miguel^b, M. Taran^b and B. Delmond^{*a}

 ^{a)} Laboratoire de Chimie des Substances Végétales, Institut du Pin, Université Bordeaux 1, 351, cours de la Libération, 33405 Talence Cedex France.
 ^{b)} Unité d'Enseignement et de Recherche des Sciences Pharmaceutiques Université Bordeaux 2.

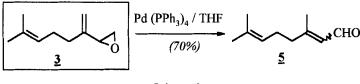
Abstract: Mono-epoxides prepared from myrcene, were used as synthetic intermediates to obtain citral, linalool, geranylacetone and pseudo-ionone.

Due to its easy availability and good reactivity, myrcene $\underline{1}$ is frequently used to access to high value products¹ (perfumes, cosmetics, vitamins, retinoids,...). Epoxides being versatile synthetic tools², we wish to report in this communication new aspects of the utilization of mono-epoxides $\underline{2-4}$, previously obtained³ from myrcene $\underline{1}$ (Scheme 1).

^{*} To whom correspondence should be addressed

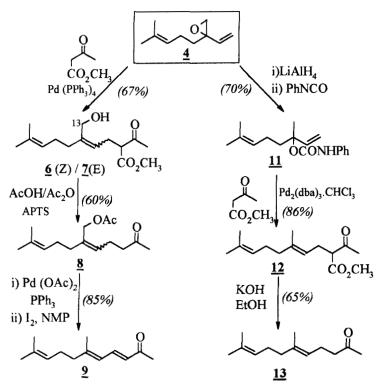


Under the influence of a catalytic amount of tetrakis (triphenylphosphine)palladium, myrcene-1,2-epoxide $\underline{3}$ undergoes isomerization⁴ to citral $\underline{5}$ in 70% yield (Scheme 2). Citral is a very important compound in perfumery for citrus flavors. It is also industrially used for the large-scale manufacture of vitamin A and carotenoids⁵.



Scheme 2

The myrcene-3,10-epoxide $\underline{4}$ (Scheme 3) with its vinyl-epoxide moiety could be selectively alkylated on carbon atom-1 with methyl acetoacetate, using Pd(PPh₃)₄ catalyzed neutral reactions⁶. We obtained in 67% isolated yield a mixture (86/14) of 13-hydroxy- β -keto esters $\underline{6}(Z)$ and $\underline{7}(E)$ which could be separated by liquid chromatography. The stereochemistry of the 5,6-double bond was established⁷ by ¹³C-NMR, comparing chemical shifts of the hydroxylated



Scheme 3

carbon atom (C-13). The (Z)-configuration was assigned to compound <u>6</u>, whose C-13 carbon atom (δ =60ppm) is more shielded than its (E)-isomer <u>7</u> (δ =66.5ppm).

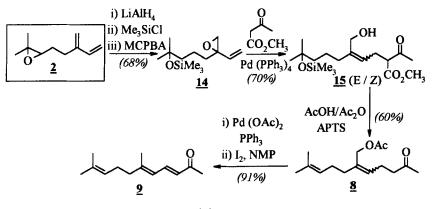
Decarbomethoxylation of the mixture of both isomers 6/7 with the AcOH/Ac₂O system⁸ using p-toluenesulfonic acid catalysis resulted in the formation of keto acetate **8** in 60% yield, without loss of configurational purity. Treatment⁹ of **8** with Pd(OAc)₂/PPh₃ gave a mixture which treated with iodine catalyst in 1-methyl-2-pyrrolidinone (NMP)¹⁰, led to pseudo-ionone **9** in 85% yield. Pseudo-ionone is a precursor of ionones¹¹, which are components of perfume compositions (violet fragrance) and serve as intermediates in vitamin A synthesis^{1d}.

The reduction of myrcene-3,10-epoxide $\underline{4}$ by means of lithium aluminium hydride led to linalool <u>10</u> in 78% yield. Linalool is used in perfumery (fruity notes) and as an intermediate in the production of vitamin E^{1e}. From linally phenyl carbamate¹² <u>11</u> prepared from linalool <u>10</u> (90% yield) we performed palladiumcatalyzed homologation reaction¹³. The alkylation with methyl acetoacetate produced in 86% yield β -keto ester <u>12¹⁴</u>. A decarbomethoxylation reaction (KOH/EtOH)¹⁵ led in 65% yield to geranyl acetone <u>13</u>. Geranylacetone used in perfumery and in soap perfumes is also an important intermediate in isophytol and vitamin E synthesis¹⁶.

From myrcene-6,7-epoxide $\underline{2}$ we prepared in 68% overall yield the 7trimethyl-silyloxy-3,10-epoxide $\underline{14}$ (Scheme 4). Treatment of this ene-oxide $\underline{14}$ with methyl acetoacetate in the presence of Pd(PPh₃)₄ catalyst gave in 70% yield a mixture (E/Z) of hydroxy β -keto ester $\underline{15}$ in a 3:7 ratio. Decarbomethoxylation (AcOH/Ac₂O;APTS) also induced a dealkoxysilylation to give 13-acetoxy geranyland nerylacetone $\underline{8}$. Deacetoxylation in the presence of Pd(OAc)₂/PPh₃ and isomerization (I₂, NMP) led to pseudo-ionone $\underline{9}$ in 91% yield.

CONCLUSION

Thus, mono-epoxides 2-4, easily and selectively obtained from myrcene 1, an industrial raw material, are useful precursors for a new access to citral 5, linalool 10, pseudo-ionone 9 and geranyl acetone 13. These functionalized terpenoids, are key intermediates in the synthesis of perfumes and vitamins.



Scheme 4

EXPERIMENTAL SECTION

Separation and purification of the products were achieved by flashchromatography (silica gel Merck, 70-230 mesh ASTM). ¹H-NMR and ¹³C-NMR spectra were recorded on a Bruker AC 250 (250MHz for ¹H and 62.89MHz for ¹³C) spectrometer using CDCl₃ as solvent. Chemical shifts are recorded as δ values in ppm related to internal tetramethylsilane. Mass spectra were obtained with a Micromass VG 7070 F instrument. High-resolution mass spectra (HRMS) data were recorded at an ionizing voltage of 70eV on a VG Micromass 16F. Solvents were freshly distilled from drying agent in a nitrogen atmosphere, before use.

Isomerization of myrcene-1,2-epoxide 3.

To a solution of tetrakis(triphenylphosphine)palladium (23mg, 0.02mmol) and triphenylphosphine (10mg, 0.04mmol) in THF (2mL) was added myrcene-1,2epoxide $\underline{3}$ (305mg, 2mmol). The reaction mixture was refluxed for 22 hours. After filtration, the crude mixture was purified by flash-chromatography; elution with petroleum ether-ether (6:4) gave citral $\underline{5}$ (213mg, 70% yield).

Alkylation of myrcene-3,10-epoxide 4.

To a solution of tetrakis(triphenylphosphine)palladium (298mg, 0.24mmol) in THF (100mL), was added myrcene-3,10-epoxide <u>4</u> (786mg, 5.17mmol). The mixture was stirred for 5min, then methyl acetoacetate (0.75mL, 6.02mmol) was added. After stirring at room temperature for 24 hours, brine was added. The organic layer was dried (MgSO₄) and concentrated under reduced pressure. The crude product was separated by flash-chromatography.

Elution with petroleum ether-ether (5:5) gave (Z)-isomer <u>6</u> (800mg): ¹H-NMR (δ): 1.52 (s, 3H), 1.60 (s, 3H), 2.18 (s, 3H, H-1), 3.48 (t, 1H, H-3), 3.66 (s, 3H, -CO₂C<u>H₃</u>), 4.05 (m, 2H, -C<u>H</u>₂OH), 4.99 (m, 1H, H-9), 5.09 (t, 1H, H-5). ¹³C-NMR (δ): 17.6 (C-12), 25.5 (C-11), 26.3 (C-8), 26.6 (C-4), 29.3 (C-1), 35.4 (C-7), 52.4 (-CO₂C_{H₃}), 59.3 (C-3), 60.0 (C-13), 123.1 (C-5), 123.7 (C-9), 131.6 (C-10), 141.8 (C-6), 169.9 (-<u>C</u>O₂CH₃), 202.7 (C-2). SM, m/z (relative intensity): 250 (M'-H₂O, 2%), 69 (100%). Anal. calcd. for C₁₅H₂₄O₄: C, 67.16; H, 8.96. Found: C, 67.42; H, 9.07.

Elution with petroleum ether-ether (4:6) gave (E)-isomer 7 (130mg): ¹H-NMR (δ): 1.53 (s, 3H), 1.60 (s, 3H), 2.16 (s, 3H, H-1), 3.42 (t, 1H, H-3), 3.66 (s, 3H, -CO₂C<u>H₃</u>), 3.93 (m, 2H, -C<u>H</u>₂OH), 4.98 (m, 1H, H-9), 5.24 (t, 1H, H-5). ¹³C-NMR (δ): 17.7 (C-12), 25.7 (C-11), 26.5 (C-8), 26.8 (C-4), 28.2 (C-7), 29.5 (C-1), 52.7 (-CO₂CH₃), 59.6 (C-3), 66.5 (C-13), 121.2 (C-5), 124.0 (C-9), 132.3 (C-10), 142.1 (C-6), 170.2 (-CO₂CH₃), 203.1 (C-2). SM, m/z (relative intensity): 250 (M⁺-H₂O, 1.5%), 69 (100%). Anal. calcd. for C₁₅H₂₄O₄: C, 67.16; H, 8.96. Found: C, 68.02; H, 9.13.

Decarbomethoxylation of hydroxy- β -keto ester <u>6/7</u>.

To a solution of crude hydroxy-\(\beta\)-keto esters (6/7)(531mg, 1.98mmol) in acetic acid (10mL) and acetic anhydride (0.2mL) was added APTS (45mg, 0.26mmol). The reaction mixture was heated at 70°C for one hour, and then hydrolyzed with water (20mL). After extraction with Et₂O, the organic layers were washed with a NaCl saturated solution and dried (MgSO₄). The solvent was evaporated and the residue was purified by flash-chromatography. Elution (7:3 petroleum ether-ether) gave 13-acetoxy geranyl-and nervlacetone¹⁷ 8 (300mg, 60% yield). To a solution of 8 (300mg, 1.19mmol) in dioxane (3mL) was added Pd(OAc)₂ (3mg) and triphenylphosphine (37mg, 0.15mmol). The reaction mixture was stirred for one hour at 80°C. The solvent was evaporated under reduced pressure and the residue was heated at 160°C for one hour with 1-methyl-2pyrrolidinone (1,16g) and iodine (2mg). The reaction mixture was diluted with Et₂O (100mL), washed with 10% Na₂S₂O₃ solution (100mL),10% HCl solution (100mL), brine and dried. After elimination of solvent the residue was purified by flash-chromatography. Elution with petroleum ether-ether (9:1) gave pseudoionone¹⁸ 9 (194mg, 85%): ¹H-NMR (δ): 1.59 (s, 3H), 1.66 (s, 3H), 1.88 (s, 3H), 2.25 (s, 3H), 5.07 (m, 1H), 5.96-6.09 (m, 2H), 7.33-7.47 (m, 1H). ¹³C-NMR (δ): 17.5 (C-13), 17.7 (C-12), 25.7 (C-11), 26.3 (C-8), 27.5 (C-1), 40.5 (C-7), 123.2 (C-9), 123.7 (C-5), 128.2 (C-3), 132.3 (C-10), 139.5 (C-4), 151.2 (C-6), 198.2 (C-2). SM, m/z (relative intensity): 192 (M⁺, 9%), 69 (100%).

Alkylation of linalool 10.

To a suspension of LiAlH₄ (152mg, 4mmol) in Et₂O (2mL) was added at

0°C myrcene-3, 10-epoxide <u>4</u> (152mg, 1mmol). The reaction mixture was stirred for 16 hours at room temperature. After hydrolysis with H₂O (5mL) and then 5% HCl (2mL), extraction with Et₂O, the solvent was evaporated to give linalool <u>10</u> (120mg, 78% yield).

To tris(dibenzylideneacetone)dipalladium-chloroform adduct (103mg, 0.10mmol) and triphenylphosphine (104mg, 0.4mmol) in THF (6mL) were added linalyl phenylcarbamate $11^{12}(1.1g; 4.02mmol)$ in THF (4mL) and methyl acetoacetate (233mg, 2.01mmol). The reaction mixture was stirred for 12 hours at 40°C. After elimination of solvent under reduced pressure, the residue was treated with Et₂O (10mL). The organic layer was washed with brine, dried (MgSO₄) and concentrated. The crude product was purified by flash-chromatography; elution with 9:1 petroleum ether-ether gave <u>12</u>(E) (437mg, 86% yield): ¹H-NMR (δ): 1.57 (s, 3H), 1.59 (s, 3H), 1.61 (s, 3H), 2.21 (s, 3H, H-1), 3.39 (t, 1H, H-3), 3.71 (s, 3H, -CO₂CH₃), 4.96 (m, 2H, H-5, H-9). ¹³C-NMR (δ): 17.7 (C-13), 17.8 (C-12), 25.8 (C-11), 26.5 (C-4), 27.1 (C-8), 29.3 (C-1), 39.8 (C-7), 52.4 (-CO₂CH₃), 59.9 (C-3), 119.7 (C-5), 124.0 (C-9), 131.6 (C-10), 138.6 (C-6), 170.1 (-<u>CO₂CH₃</u>), 203.2 (C-2). SM, m/z (relative intensity): 252 (M⁺, 2%), 69 (100%). HRMS : calcd. for C₁₅H₂₄O₃: 252.1725; found : 252.1726.

Obtention of geranylacetone 13(E).

To EtOH (0.58mL) was added a solution of KOH (174mg) in H₂O (0.29mL) and β -keto-ester <u>12</u> (368mg, 1.48mmol). The reaction mixture was stirred at room temperature for 4 hours, acidified with HCl and heated to reflux for 2 hours. After addition of water, the mixture was extracted with Et₂O. The organic

layers were washed with a NaCl saturated solution and dried (MgSO₄). The solvent was evaporated and the residue purified by flash-chromatography (9:1 petroleum ether-ether) to give geranylacetone <u>13(E)</u> (184mg, 65% yield): ¹H-NMR (δ): 1.52 (s, 3H), 1.54 (s, 3H), 1.61 (s, 3H), 2.07 (s, 3H, H-1), 5.00 (m, 2H, H-5, H-9). ¹³C-NMR (δ): 16.0 (C-13), 17.7 (C-12), 22.3 (C-4), 25.7 (C-11), 26.6 (C-8), 30.0 (C-1), 39.6 (C-7), 44.0 (C-3), 122.5 (C-9), 124.2 (C-5), 131.4 (C-10), 136.4 (C-6), 207.8 (C-2).

Obtention of 3,10-epoxy-7-trimethylsilyloxy myrcene 14.

To a suspension of LiAlH₄ (811mg, 21.3mmol) in Et₂O (10mL) was added at 0°C, myrcene-6,7-epoxide 2 (2.163g, 14.2mmol) in Et₂O (15mL). The solution was stirred for 4 hours at room temperature, hydrolyzed with a 5% H_2SO_4 solution (50mL) and extracted with Et₂O. The organic layers were washed with a NaCl saturated solution, dried (MgSO₄) and evaporated under reduced pressure to give myrcenol (1.76g, 80% yield). To this crude product was added at 0°C, CH₂Cl₂ (40mL), triethylamine (5mL), 4-dimethylaminopyridine (209mg, 1.7mmol) and trimethylchlorosilane (2.88mL, 22.8mmol). The stirring was kept for 24hours at room temperature. The reaction mixture was washed with brine, the organic layers were dried (MgSO₄) and concentrated to give 7-trimethylsilyloxy myrcene (2.27g, 88% yield). From 7-trimethylsilyloxy myrcene (815mg, 3.6mmol), the epoxidation with m-chloroperbenzoic acid, according to usual work-up, gave after purification 3.10-epoxy-7flash-chromatography (98:2 petroleum ether-ether) by trimethylsilyloxy myrcene <u>14</u> (840mg, 96% yield): ¹H-NMR (δ): 0.03 (s,9H, (CH₃)₃Si), 1.14 (s, 6H), 2.64 (m, 2H, H-10), 5.20 (m, 2H, H-1), 5.71 (m, 1H, H-

2). ¹³C-NMR (δ): 2.5 ((<u>C</u>H₃)₃Si), 19.8 (C-5), 29.6 (C-8), 29.7 (C-9), 33.8 (C-4),
44.2 (C-6), 54.7 (C-10), 58.5 (C-3), 73.6 (C-7), 116.0 (C-1), 137.4 (C-2). SM,
m/z (relative intensity): 227 (M²-CH₃, 1%), 131 (100%).

Alkylation of 3,10-epoxy-7-trimethylsilyloxy myrcene 14.

From 3,10-epoxy-7-trimethylsilyloxy myrcene <u>14</u> (1.210mg, 5mmol), tetrakis-(triphenylphosphine)palladium (289mg, 0.25mmol) and methyl acetoacetate (1.16g, 10mmol) in THF solution (80mL), according to the previously described procedure, we obtained 10-trimethylsilyloxy- β -keto ester <u>15</u> (E/Z 3:7) (125 mg, 70% yield):

15 (E):¹H-NMR (δ): 0.00 (s, 9H), 1.10 (s, 3H), 2.12 (s, 3H, H-1), 3.27 (m, 1H, H-3), 3.72 (s, 3H, -CO₂CH₃), 3.91 (m, 2H, H-13), 5.23 (t, 1H, H-5). ¹³C-NMR (δ): 2.6 ((<u>C</u>H₃)₃Si), 23.0 (C-8), 26.0 (C-4), 28.3 (C-7), 29.8 (C-1), 29.9 (C-11, C-12), 44.6 (C-9), 52.9 (-CO₂<u>C</u>H₃), 60.6 (C-3), 66.1 (C-13), 73.8 (C-10), 121.2 (C-5), 141.9 (C-6), 168.7 (-<u>CO₂</u>CH₃), 203.1 (C-2). <u>15</u> (Z): ¹H-NMR (δ): 0.00 (s, 9H), 1.10 (s, 6H), 2.14 (s, 3H, H-1), 3.36 (m, 1H, H-3), 3.70 (s, 3H, -CO₂<u>C</u>H₃), 4.02 (m, 2H, H-13), 5.07 (t, 1H, H-1). ¹³C-NMR (δ): 2.6 ((<u>C</u>H₃)₃Si), 22.7 (C-8), 26.3 (C-4), 29.8 (C-1), 29.9 (C-11, C-12), 35.8 (C-7), 44.3 (C-9), 53.0 (-CO₂<u>C</u>H₃), 59.8 (C-13), 60.6 (C-3), 73.9 (C-10), 123.3 (C-5), 141.9 (C-6), 168.8 (-<u>CO₂</u>CH₃), 203.2 (C-2).

Obtention of pseudo-ionone 9.

From 7-trimethylsilyloxy- β -keto ester <u>15</u> (531mg, 1.33mmol), according to the previously described protocols (decarbomethoxylation and deacetoxylation), we obtained pseudo-ionone <u>9</u> (279mg, 54% overall yield).

REFERENCES

- (a) Dorsky, J., in "Perfumes, Art, Science and Technology", Muller, P.M. and Lamparsky Eds., Elsevier, London, 1991. (b) Ohloff, G., "Scent and Fragrances", Springer-Verlag, Berlin, 1994. (c) Isler, O., " Carotenoids", Birkauser Verlag, Basel, 1971. (d) Van Arnum, S.D., in "Kirk-Othmer Encyclopedia of Chemical Technology", Vol. 25, 4th. Edition., John Wiley and Sons, New York, 1998. (e) Casani, R. in "Kirk-Othmer Encyclopedia of Chemical Technology", Vol. 25, 4th. Edition, John Wiley and Sons, New York, 1998. (e) Casani, R. in "Kirk-Othmer Encyclopedia of Chemical Technology", Vol. 25, 4th. Edition, John Wiley and Sons, New York, 1998. (f) Sporn, M.B.; Roberts, A.B. and Goodman, D.S. Eds., "The Retinoids", 2nd ed., Raven Press, New York, 1994.
- (a) Rosovsky, A., "The Chemistry of Heterocyclic Compounds", Weissberger,
 A. Ed., Vol.19, Wiley, 1964. (b) Malinovskii, M.S., "Epoxides and Their Derivatives", Israel Program for Scientific Translations, Jerusalem, 1965. (c) Bartok, M. and Lang, K.L., "The Chemistry of Ethers, Crown Ethers, Hydroxyl Groups and Their Sulfur Analogues", Patai, S. Ed., Part 2, Wiley, 1980.
- (a) Mousseron-Canet, M.; Mousseron, M. and Levallois, C., Bull. Soc. Chim. Fr., 1963, 376. (b) Fauchet, V.; Arreguy-San Miguel, B.; Taran, M. and Delmond, B., Synth. Commun., 1993, 23, 2503.
- 4. Suziki, M.; Oda, Y. and Noyori, R., J. Am. Chem. Soc., 1979, 101, 1623.
- 5. Isler, O., Pure Appl Chem., 1979, 51, 447.
- (a) Trost, B.M. and Molander, G. A., J. Am .Chem. Soc., 1981, 103, 5969. (b)
 Tsuji, J.; Kataoka, H. and Kobayashi, Y., Tetrahedron Lett., 1981, 22, 2575.
- 7. De Hann, J.W. and Van de Ven, L.J.M., Org. Magn. Reson., 1973, 5, 147.

- (a) Fonken, G.S. and Johnson, W.S., J. Am. Chem. Soc., 1952, 74, 831. (b) Kieczykowski, G.R.; Roberts, M.R. and Schlessinger, R.H., J. Org. Chem., 1978, 43, 788.
- Tsuji, J.; Yamakana, T.; Kaito, M. and Mandoi, T., Tetrahedron Lett., 1978, 24, 2075.
- 10. Fujita, Y.; Onishi, T.; Hino, K. and Nishida, T., Tetrahedron Lett., 1980, 21, 1347.
- 11. Bedoukian, P.Z., "Fragrance Chemistry", Theimer, E.T. Ed, Ch. 8, Academic Press, 1982.
- 12. Nikiforov, A.; Jirovetz, L. and Buchbauer, G., Liebigs Ann. Chem., 1989, 489.
- Tsuji, J.; Shimizu, I.; Minami, I.; Ohashi, Y.; Sugiura, T. and Takanashi, K., J. Org. Chem., 1985, 50, 1523.
- Mercier, C.; Mignani, G.; Aufrand, M. and Allmang, G., *Tetrahedron Lett.*, 1991, 32, 1433.
- 15. Patrick, T.M., J. Org. Chem., 1952, 17, 1009.
- (a) Kasparek, S., "Vitamin E: A Comprehensive Treatise", Machlin, L.S. Ed.,
 M.Dekker, 1980. (b) Heathcock, C.H., in "The Total Synthesis of Natural Products", Apsimon, J. Ed., Vol. 2, John Wiley and Sons, New York, 1973.
- 17. Pandey, U.C.; Sarmah, P. and Sharma, R.P., Tetrahedron, 1984, 40, 3739.
- 18. Englert, G., Helv. Chim. Acta, 1975, 58, 2367.

Received in Exeter 8 February 1999; accepted 29 March 1999