

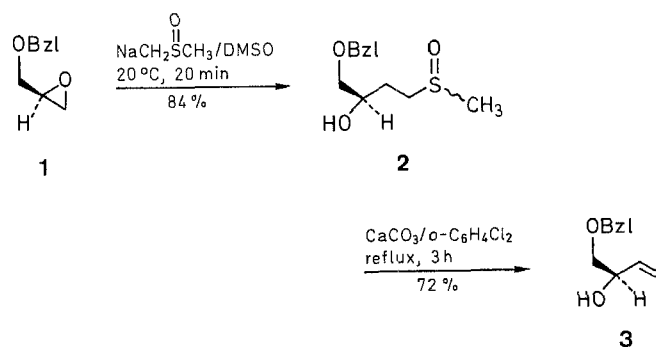
A Facile Synthesis of (*R*)-1-Benzoyloxy-3-buten-2-ol

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A functionalized optically active allyl alcohol, (*R*)-1-benzyloxy-3-buten-2-ol, is efficiently prepared from (*S*)-*O*-benzylglycidol in two steps.

Molecules possessing a chiral allyl alcohol moiety are useful starting materials for the construction of a variety of optically active compounds as they can serve as substrates for various chemical conversions such as sigmatropic reactions.¹ We report here a facile two-step preparation of (*R*)-1-benzyloxy-3-buten-2-ol (**3**),² a potentially useful functionalized chiral allyl alcohol, using (*S*)-*O*-benzylglycidol (**1**).³ Stirring a solution of (*S*)-2-(benzyloxymethyl)oxirane [(*S*)-*O*-benzylglycidol] (**1**) in dimethyl sulfoxide containing two equivalents of sodium methylsulfinylmethide⁴ generated *in situ* in the same reaction medium affords (*R*)-4-benzyloxy-3-hydroxybutyl methyl sulfoxide (**2**) in 84% yield after 20 min. Although thermolysis of



the sulfoxide **2** with calcium carbonate in boiling toluene⁵ is found to proceed very sluggishly to give the desired (*R*)-1-benzyloxy-3-buten-2-ol (**3**), the reaction occurs in more facile and cleaner manner to furnish **3** in 73% yield after 3 h when boiling *o*-dichlorobenzene is used as solvent in place of boiling toluene. The enantiomeric (*S*)-1-benzyloxy-3-buten-2-ol (**3**) may also be obtained by employing the same sequence starting from (*R*)-*O*-benzylglycidol (**1**), which is readily accessible.⁶

Optical rotations were measured with a JASCO-DIP-4 automatic polarimeter. Mass spectra were recorded with a JEOL-OISG-2 instrument, IR spectra with a JASCO A-102 spectrophotometer, and ¹H-NMR spectra on JEOL-JNM-FX90A (90 MHz) and JEOL-JNM-GX500 (500 MHz) spectrometers.

Reactions were carried out under argon.

(*R*)-4-Benzyloxy-3-hydroxybutyl Methyl Sulfoxide (2):

NaH (washed with hexane; 2.68 g, 67.1 mmol) is added portionwise to DMSO (50 mL) at 0 °C with stirring, the solution is warmed to 60 °C for 70 min, then cooled to room temperature. (*S*)-2-(Benzyloxymethyl)oxirane (**1**; 5.0 g, 30.5 mmol) in DMSO (10 mL) is added dropwise to the stirred solution and stirring is continued for 20 min at room temperature. The mixture is diluted with 5% aqueous NaHCO₃ (30 mL), then extracted with CH₂Cl₂ (3 × 50 mL). The wet CH₂Cl₂ extract is evaporated by water pump followed by rotary pump (< 60 °C) to remove most of the solvents. The oily residue is chromatographed on a silica gel column (90 g) using Et₂O as eluent to give **2** as a pale yellow oil; yield: 6.2 g (84%).

C₁₂H₁₈O₃S calc. C 59.48 H 7.49 S 13.23
(242.2) found 59.32 7.60 13.59

MS (70 eV): *m/z* = 242 (M⁺), 170, 91 (100%).

IR (neat): ν = 3400 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 1.78–2.30 (m, 2H); 2.58 (s, 3H); 3.32–3.63 (m, 2H); 2.70–3.20 (m, 3H, ¹H exchangeable); 3.76–4.15 (m, 1H); 4.56 (s, 2H); 7.33 (s, 5H).

(*R*)-1-Benzyloxy-3-buten-2-ol (3):

In a solution of the sulfoxide **2** (3.63 g, 15 mmol) in *o*-dichlorobenzene (55 mL), CaCO₃ (4.5 g, 45 mmol) is suspended and this suspension is refluxed for 3 h with stirring. After cooling, the mixture is filtered through Celite and the filtrate is directly chromatographed on a silica gel column (100 g) using hexane/Et₂O (1:1) as eluent to give **3** as a colorless oil; yield: 1.94 g (72.7%); bp 95–100 °C/0.5 Torr (Kugelrohr); [α]_D²⁴ + 4.72° (*c* = 1.10, CHCl₃) [100% e.e. by ¹H-NMR (500 MHz) of the MTPA ester] (Lit. [α]_D + 5.3° (*c* = 0.37, CHCl₃);^{2b} [α]_D + 6.2° (*c* = 1.6, CHCl₃).^{2c}

C₁₁H₁₄O₂ calc. C 74.13 H 7.92
(178.2) found 74.40 7.72

MS (70 eV): *m/z* = 178 (M⁺), 160, 91 (100%).

IR (neat): ν = 3450 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 2.39 (s, 1H exchangeable); 3.26–3.63 (m, 2H); 4.57 (s, 2H); 4.24–4.44 (m, 1H); 5.11–6.04 (m, 3H); 7.33 (s, 5H).

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- (1) For example, see: Hill, R.K., in: *Asymmetric Synthesis*, Morrison, J.D. (ed.), Vol. 3, Academic Press, New York, 1984, p. 503.
- (2) Recently, three alternative syntheses of (*R*)-1-benzyloxy-3-buten-2-ol (**3**) were reported.
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- (3) Takano, S., Akiyama, M., Ogasawara, K. *Synthesis* **1985**, 503.
- (4) Corey, E.J., Chaykovsky, M. *J. Am. Chem. Soc.* **1962**, 84, 866.
- (5) Trost, B.M., Saltzman, T.N. *J. Am. Chem. Soc.* **1973**, 95, 6840.
- (6) Takano, S., Seya, K., Goto, E., Hiram, M., Ogasawara, K. *Synthesis* **1983**, 116.