

Lucia Cecchi and Guido Filacchioni*

Istituto di Chimica Farmaceutica e Tossicologica, Università di Firenze,
50121 Firenze, Italy

Received December 14, 1982

The reaction of 2-nitrobenzyl bromide with dimethyl pyrazole-3,5-dicarboxylate gave dimethyl 1-(2-nitrobenzyl)pyrazole-3,5-dicarboxylate which through a few steps procedure afforded the key intermediate 5,10-dihydro-11-oxopyrazolo[5,1-*c*][1,4]benzodiazepine. The latter was reduced and dehydrogenated to yield the new tricyclic system 5*H*-pyrazolo[5,1-*c*][1,4]benzodiazepine.

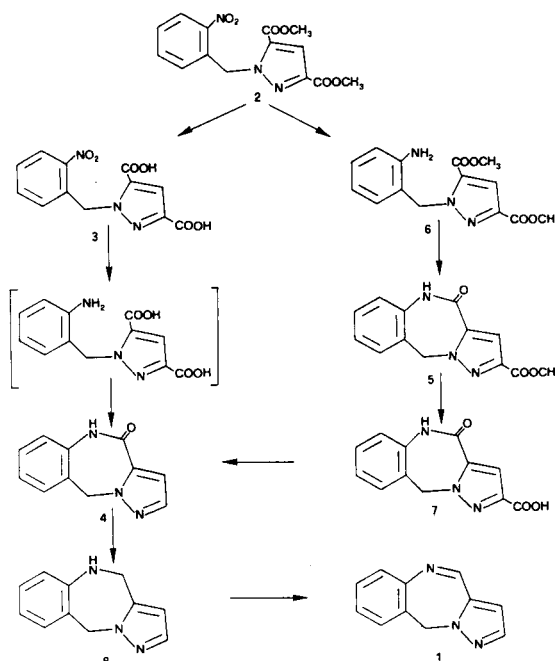
J. Heterocyclic Chem., **20**, 871 (1983).

Antramycin, tomaymycin and sibiromycin [1] are anti-tumor antibiotics containing as common feature the 5*H*-pyrrolo[2,1-*c*][1,4]benzodiazepine tricyclic ring system. In a search for synthetic tumor inhibitors [2-4] related to the cited antibiotics we now describe the synthesis of 5*H*-pyrazolo[5,1-*c*][1,4]benzodiazepine **1**, an isosteric analog of 5*H*-pyrrolo[2,1-*c*][1,4]benzodiazepine [5,6]. This work might represent a useful contribution to a synthesis of a variety of pyrazolo[5,1-*c*][1,4]benzodiazepine derivatives.

Dimethyl 1-(2-nitrobenzyl)pyrazole-3,5-dicarboxylate **2** was obtained (Scheme 1) in good yield by treating, in anhydrous tetrahydrofuran, 2-nitrobenzyl bromide with dimethyl pyrazole-3,5-dicarboxylate as *N*-potassium salt. Acidic hydrolysis of **2** gave the corresponding acid **3**. The direct conversion of **3** to 5,10-dihydro-11-oxopyrazolo[5,1-*c*][1,4]benzodiazepine (**4**) was carried out by catalytic reduction of the nitro group and by heating the resulting product under reduced pressure. These reactions proceeded in low yields and it was not possible to isolate 1-(2-aminobenzyl)pyrazole-3,5-dicarboxylic acid in the pure state. Therefore an alternative profitable procedure, even if longer, was chosen: **2** was transformed into methyl 5,10-dihydro-11-oxopyrazolo[5,1-*c*][1,4]benzodiazepine-2-carboxylate **5** via the amino derivative **6**. Acidic hydrolysis of **5** gave the corresponding acid **7** which, heated in vacuum, provided **4** in good yield. Reduction of the key intermediate **4** with lithium aluminium hydride afforded 10,11-dihydro-5*H*-pyrazolo[5,1-*c*][1,4]benzodiazepine **8** which was dehydrogenated to the expected 5*H*-pyrazolo[5,1-*c*][1,4]benzodiazepine **1**.

Attempts to obtain the pyrazolo[5,1-*c*][1,4]benzodiazepine ring system by condensation of 2-nitrobenzyl bromide with methyl pyrazole-3(5)-carboxylate (**9**) was unsuccessful. Compound **9** was prepared starting from hydrazine and ethoxymethylenepyruvate [7] unlike the procedure described by Reimlinger [8]. When **9** was allowed to react with 2-nitrobenzyl bromide (Scheme 2) only one product, **10** or **11**, was isolated. Application of the chemistry in Scheme 1 to this product failed to produce the lactam **4**. Accordingly we assigned it the structure of methyl 1-(2-nitrobenzyl)pyrazole-3-carboxylate (**10**) rather than **11**. Assignment of the 1,3-disubstituted structure to **10** is consistent even with

SCHEME 1



the literature data [9,10].

We tried then to prepare **1** from **10** via methyl 1-(2-formylaminobenzyl)pyrazole-3-carboxylate **12**, following the procedure described by Artico [6] to synthesize 5*H*-pyrrolo[2,1-*c*][1,4]benzodiazepine. Thus **10** was reduced to the corresponding amino derivative **13**, which, with 99% formic acid, gave **12**. We were unable to cyclize **12** to methyl 5*H*-pyrazolo[5,1-*c*][1,4]benzodiazepine-2-carboxylate using phosphorus oxychloride, polyphosphoric acid and other agents. In any case there was no evidence for ring closure.

All the synthesized compounds were characterized by elemental analysis, ir and pmr spectra. Furthermore the structure of **1** and **8** were confirmed by mass spectra.

EXPERIMENTAL

All melting points were determined on a Büchi 510 capillary melting point apparatus and are uncorrected. The ir spectra were taken in nujol

183 (100) M^+ , 182 (50), 156 (10), 155 (14), 130 (14), 129 (14), 128 (14), 104 (7), 89 (14), 77 (14).

Anal. Calcd. for $C_{11}H_9N_3$: C, 72.11; H, 4.95; N, 22.93. Found: C, 72.20; H, 5.00; N, 22.91.

Methyl Pyrazole-3(5)-carboxylate (**9**).

A solution of 4.9 g (31 mmoles) of methyl ethoxymethylenepyruvate [7] and 2 ml of 85% hydrazine hydrate in 100 ml of methanol was refluxed for 30 minutes. Evaporation of the solvent gave a yellow residue which was recrystallized from water giving 1.6 g (40%) of white crystals, mp 136-138° (lit 142° [8], 139-140° [11]); ir: 3120 and 3000-2300 (NH), 1740 (CO) cm^{-1} ; pmr (deuteriochloroform): 3.95 (s, 3H, CH_3), 6.84 (d, 1H, $J = 2$ Hz), 7.85 (d, 1H, $J = 2$ Hz), 14.3 (br, 1H, NH, it exchanges with deuterium oxide).

Anal. Calcd. for $C_5H_6N_2O_2$: C, 47.62; H, 4.80; N, 22.21. Found: C, 47.93; H, 4.78; N, 22.49.

Methyl 1-(2-Nitrobenzyl)pyrazole-3-carboxylate (**10**).

A mixture of 2.8 g (22 mmoles) of **9** and 0.7 g (18 mmoles) of potassium in 400 ml of anhydrous tetrahydrofuran was refluxed for 5 hours. After cooling 3.9 g (18 mmoles) of 2-nitrobenzyl bromide was added and the mixture was refluxed for 5 hours. The reaction was carried out under a flow of nitrogen. The precipitate was filtered off and the solution was evaporated to give a residue which was recrystallized from ethanol giving 2.8 g (60%) of white crystals, mp 105-107°; ir: 1730 (CO), 1540 and 1340 (NO_2) cm^{-1} ; pmr (deuteriochloroform): 3.93 (s, 3H, CH_3), 5.48 (s, 2H, CH_2), 6.7-7.8 (m, 5H, 3 benzenic protons and 2 pyrazolic protons), 8.12 (m, 1H, benzenic proton).

Anal. Calcd. for $C_{12}H_{11}N_3O_4$: C, 55.17; H, 4.24; N, 16.08. Found: C, 55.10; H, 4.27; N, 16.16.

Methyl 1-(2-Aminobenzyl)pyrazole-3-carboxylate (**13**).

To a suspension of 3.3 g (12.6 mmoles) of **10** in 200 ml of ethyl acetate 1.7 g of 10% Pd/C was added. The mixture was hydrogenated in a Parr apparatus at 50 psi for 6 hours. Removal of the catalyst and evaporation in vacuum of the solvent gave a residue which was recrystallized from diethyl ether/hexane giving 2.3 g (79%) of white crystals, mp 97-99°; ir: 3420 and 3320 (NH_2), 1730 (CO) cm^{-1} ; pmr (deuteriochloroform): 3.92 (s, 3H, CH_3), 4.3 (br, 2H, NH_2 , it exchanges with deuterium oxide), 5.30 (s, 2H, CH_2), 6.5-7.5 (m, 6H, 4 benzenic protons and 2 pyrazolic protons).

Anal. Calcd. for $C_{12}H_{13}N_3O_2$: C, 62.32; H, 5.67; N, 18.17. Found: C, 62.01; H, 5.71; N, 17.95.

Methyl 1-(2-Formylaminobenzyl)pyrazole-3-carboxylate (**12**).

Compound **13** (2.3 g, 10 mmoles) in 20 ml of 99% formic acid was refluxed for 30 minutes. After cooling the solution was poured into crushed ice to give a white residue which was recrystallized from benzene/cyclohexane giving 0.8 g (30%) of white crystals, mp 120-121°; ir: 3230 (NH), 1730 (CO ester), 1650 (CO formyl) cm^{-1} ; pmr (deuteriochloroform): 3.92 (s, 3H, CH_3), 5.30 (s, 2H, CH_2), 6.78 (d, 1H, pyrazolic proton, $J_{4,5} = 2$ Hz), 7.0-8.2 (m, 5H, 4 benzenic protons and 1 pyrazolic proton), 8.5 (br s, 1H, formyl proton), 9.3 (br, 1H, NH, it exchanges with deuterium oxide).

Anal. Calcd. for $C_{13}H_{13}N_3O_3$: C, 60.23; H, 5.05; N, 16.21. Found: C, 60.16; H, 4.71; N, 15.89.

Acknowledgement.

The authors wish to thank Mr. Fabrizio Melani for his assistance in experimental work and for pmr spectra.

REFERENCES AND NOTES

- [1] L. H. Hurley, *J. Antibiot.*, **30**, 349 (1977).
- [2] G. C. Porretta, G. Scicchitano, G. Filacchioni, M. Scalzo and M. Artico, *Il Farmaco, Ed. Sc.*, **34**, 914 (1979).
- [3] M. Artico, G. De Martino, G. Filacchioni and R. Giuliano, *ibid.*, **24**, 276 (1969).
- [4] A. Ermili and G. Filacchioni, *Ann. Chim. (Rome)*, **59**, 770 (1969).
- [5] M. Artico, G. De Martino, R. Giuliano, S. Massa and G. C. Portetta, *Chem. Commun.*, 671 (1969).
- [6] M. Artico, G. De Martino, R. Giuliano, S. Massa and G. C. Porretta, *Il Farmaco, Ed. Sc.*, **24**, 980 (1969).
- [7] R. Reiner and C. H. Eugster, *Helv. Chim. Acta*, **50**, 128 (1967).
- [8] H. Reimlinger, *Chem. Ber.*, **93**, 1857 (1960).
- [9] R. G. Jones, M. J. Mann and K. C. McLaughlin, *J. Org. Chem.*, **19**, 1428 (1954).
- [10] N. W. Gilman, B. C. Holland and R. I. Fryer, *J. Heterocyclic Chem.*, **14**, 1163 (1977).
- [11] K. V. Auwers and Th. Breyhan, *J. Prakt. Chem.*, **143**, 259 (1935).