

Combining Ring-Closing Metathesis and Hydroformylation Strategies: A Novel Approach to Spirocyclic γ -Butyrolactones

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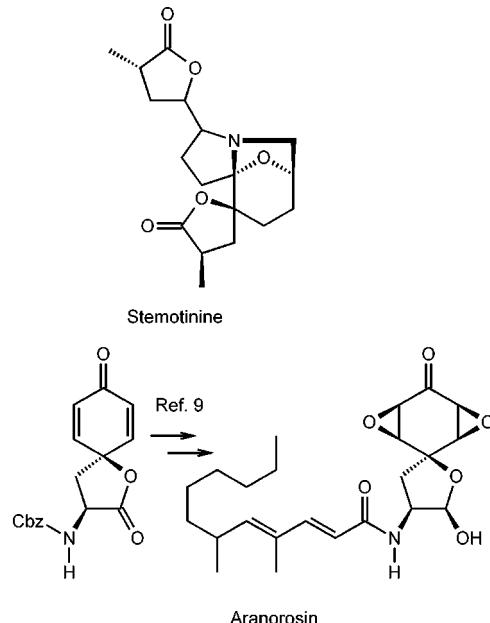
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Di- or tetrahydropyrans with a vinyl side chain are obtained by diastereoselective ring-closing metathesis or by addition of vinylmagnesium chloride to an appropriately functionalized tetrahydropyranone. The resulting allylic alcohols are converted to spirocyclic hemiacetals under hydroformylation conditions. Oxidation yields the corresponding lactones.

Introduction

Spirocyclic γ -butyrolactones not only are widespread in nature^{1–4} but have also played a key role as synthetic intermediates.^{5–7} One example of a natural product with a spirocyclic γ -butyrolactone moiety is the alkaloid stemotinine,⁸ and the antitumor antibiotic aranorosin may serve as an example where a spirocyclic lactone has been used as a key intermediate in the total synthesis⁹ (Chart 1). Furthermore, a variety of methods are available for the conversion of γ -butyrolactones into α -methylene- γ -butyrolactones.¹⁰ Spirocyclic, steroidal derivatives of these compounds often show outstanding biological properties.^{3,4} In principle, one can imagine two different strategies for the construction of spiro compounds: (i) the conversion of a tetrafunctional acyclic compound into a spirocyclic system in a one-pot reaction or (ii) the stepwise formation of two carba- or heterocycles via a monocyclic intermediate. A variety of methods for the synthesis of lactones is available, for example, lactonization after asymmetric dihydroxylation, iodolactonization, palladium-catalyzed cyclization of allenic or homopropargylic carboxylic acids, or samarium-mediated addition of ketones to acrylates. This subject has been reviewed several times over the past few years.^{11–14}

Chart 1



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In this contribution, we present a novel concept for the synthesis of spirocyclic lactones, which is based on a ruthenium-catalyzed olefin metathesis and a rhodium-catalyzed hydroformylation–acetalation sequence (Scheme 1).

We demonstrate the utility of our approach for the example of di- or tetrahydropyrans linked to lactones in a spirocyclic fashion. Although this class of substances might be interesting in itself,^{5,6} they may also serve, after cleavage of the lactone, as intermediates in the synthesis of pyran derivatives with functionalized side chains. This structural element is found in several natural products such as the pseudomonic acids.¹⁵

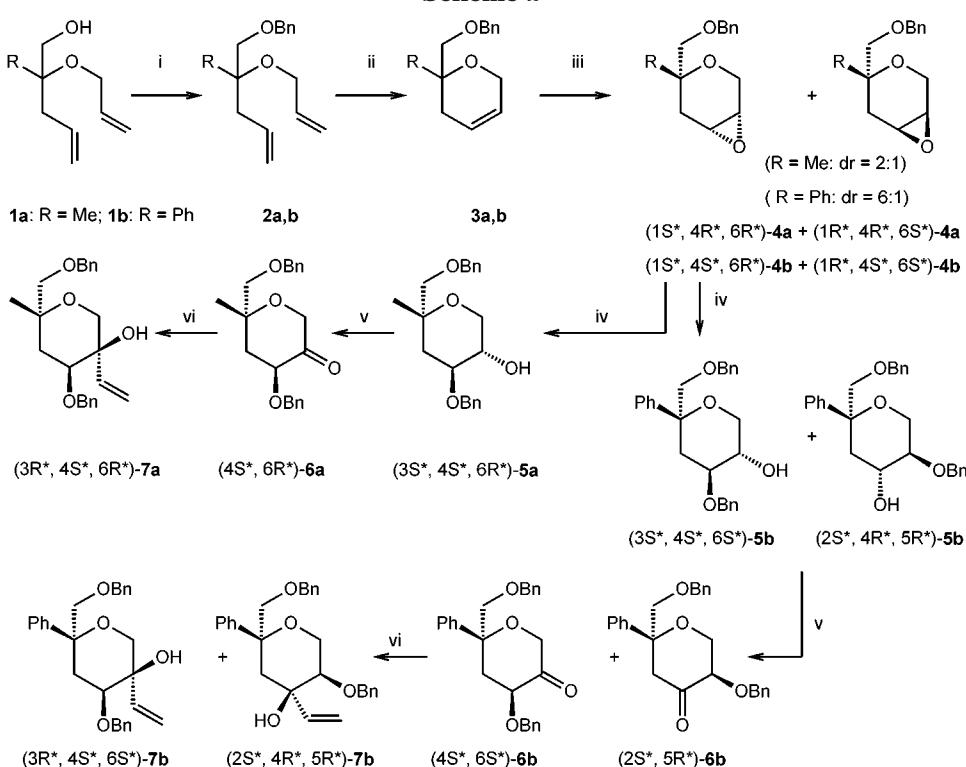
The ring-closing olefin metathesis^{16–18} has become one of the most powerful cyclization reactions since the

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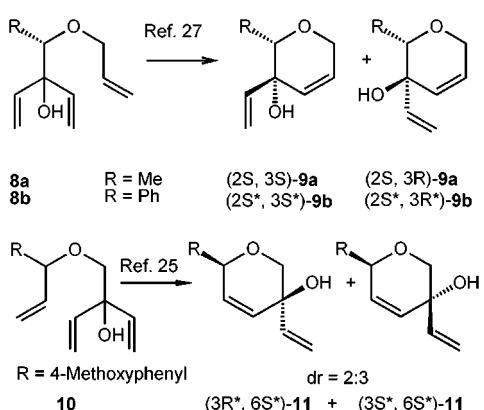
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Scheme 2^a

^a Key: (i) NaH, BnBr, THF (90% of **2a**, 91% of **2b**); (ii) Cl₂(C₅P)₂Ru=CHPh (1 mol %), DCM (81% of **3a**, 88% of **2b**); (iii) MCPBA, DCM (91% of **4a**, 99% of **4b**); (iv) BnOH, BF₃OEt₂ (10 mol %), DCM, 0 °C, chromatography (57% of (3S*,4S*,6R*)-**5a**, 78% of (3S*,4S*,6S*)-**5b** and (2S*,4R*,5R*)-**5b**) (4.5:1); (v) TPAP (5 mol %), NMO, 4 Å molecular sieve, DCM (84% of **6a**, 82% of **6b**); (vi) H₂C=CHMgCl, ether, -78 °C (66% of **7a**, 89% of **7b**).

Scheme 3



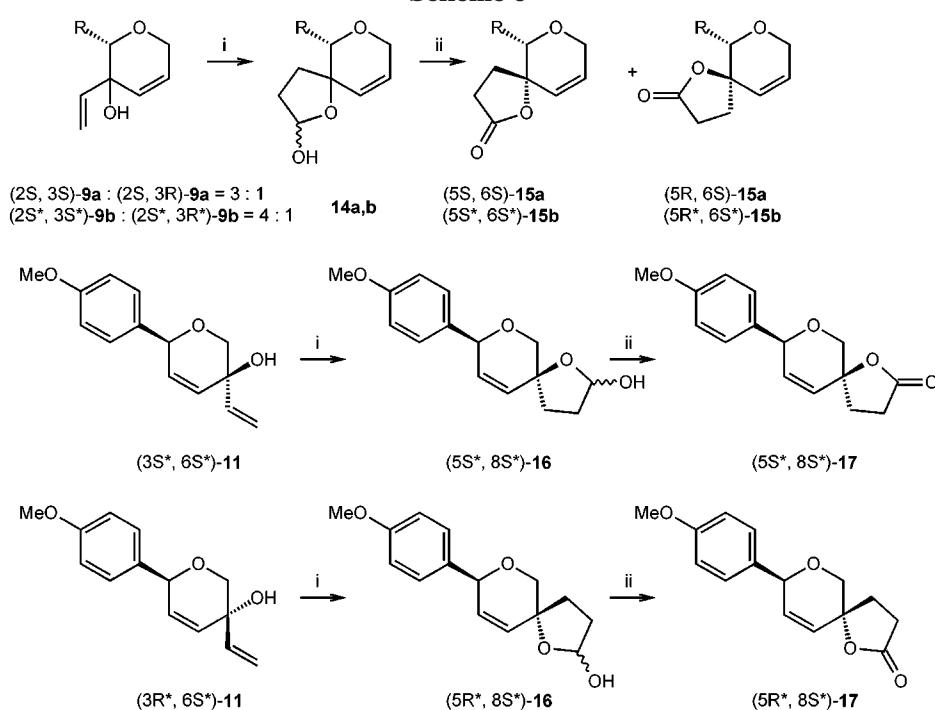
NOE Experiments. The relative configuration of all spirocyclic products was investigated by one- or two-dimensional NOE experiments conducted at 600 or 500 MHz, respectively. For spirocyclic tetrahydropyran **13a**, NOE interactions between the methyl group in the 2-position and the proton H3ax and between the -CH₂-OBn moiety and the proton H6ax indicate that the molecule preferably adopts the conformation depicted in Scheme 6. One proton of the CH₂ group of the lactone moiety shows NOE interactions with both protons H6, whereas the other proton of this CH₂ group shows a NOE interaction with H4. The phenyl analogue (*5R*,8R*,-10S**)-**13b** shows similar NOE interactions. Its regioisomer (*5S*,6R*,9S**)-**13b** adopts a conformation with the phenyl substituent in an axial position: NOE interactions between the *ortho*-H of the phenyl moiety and H6ax as well as H3eq are observed. A NOE interaction between

the axially oriented H5 and one proton of the methylene moiety of the lactone ring is indicative of the relative configuration shown in Scheme 6.

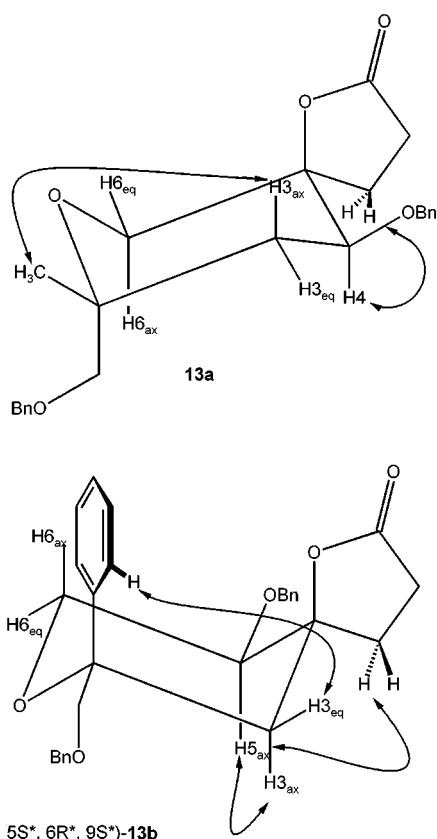
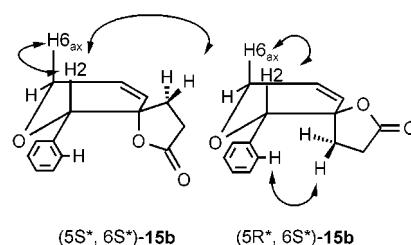
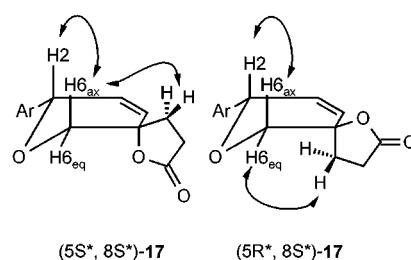
NOE interactions for spirocyclic dihydropyrans **15** are summarized in Scheme 7. A NOE interaction between H6ax and H2 indicates that the phenyl moiety adopts a pseudoequatorial position. In the case of (*5S*,6S**)-**15b**, a NOE interaction between H2 and one proton of the methylene group of the lactone moiety is observed, which is missing for the other diastereomer. For the (*5R*,6S**)-isomer, a NOE interaction between the *ortho* protons of the phenyl group and one proton of the methylene group is indicative for the proposed relative configuration. For the methyl derivative **15a**, analogous NOE interactions were observed for the protons of the methyl group.

For dihydropyrans **17** with the spirocyclic junction in the 5-position, NOE interactions of the protons H6ax and H6eq with the methylene group of the lactone ring are most indicative of the relative configuration. For both diastereomers, a NOE effect is observed between the singlet of H2 and one doublet of the H6-AB-system; assignment of the corresponding signals to H6ax and H6eq, respectively, is made on the basis of this interaction. In (*5S*,8S**)-**17**, a NOE interaction between H6ax and one proton of the methylene group of the lactone is observed, whereas in (*5R*,8S**)-**17**, H6eq interacts with this CH₂ group (Scheme 8).

In conclusion, we have developed a synthetic approach toward bicyclic spiro compounds with a lactone moiety linked to a six-membered oxacycle on the basis of ring-closing metathesis and a hydroformylation–acetalation sequence. Use of the BIPHEPHOS ligand system allows the differentiation between exocyclic and endocyclic C–C

Scheme 5^a

^a Key: (i) Rh(acac)(CO)₂ (1.0 mol %), BIPHEPHOS (4.0 mol %), CO (10 bar), hydrogen (10 bar), dioxane; (ii) TPAP (5 mol %), NMO, 4 Å molecular sieve, DCM (52–81% over two steps).

Scheme 6**Scheme 7****Scheme 8**

4-Benzylxymethyl-4-methyl-3,7-dioxabicyclo[4.1.0]heptane (4a). To a solution of the dihydropyran **3a** (2.65 g, 12.1 mmol) in DCM (50 mL) was added MCPBA (70% dispersion in water, 3.14 g, 18.2 mmol). The mixture was stirred until the starting material was completely consumed as indicated by TLC. The reaction mixture was diluted with MTBE and washed with Na₂SO₃ solution and then Na₂CO₃ solution. The organic layer was dried, filtered, and evaporated, and the residue was purified by flash chromatography on silica to give epoxide **4a** (2.75 g, 97%) as a 2:1 mixture of diastereoisomers. Spectroscopic data for the major (*1S*, 4R*, 6R**)-isomer. IR (neat): 739 s, 1110 s, 1454 m, 2862 m. MS *m/z* (rel intensity): 235 (*M*⁺ + 1, 10), 181 (10), 143 (15), 113 (85), 91 (100), 65 (20). ¹H NMR (CDCl₃, 400 MHz): δ 7.34–7.22 (5H), 4.52 (d, 1H, *J* = 12.3 Hz), 4.45 (d, 1H, *J* = 12.3 Hz), 4.02 (d, 1H, *J* = 13.8 Hz), 3.95 (d, 1H, *J* = 13.8 Hz), 3.32 (d, 1H, *J* = 9.5 Hz), 3.29 (dd, 1H, *J* = 9.8, 4.8 Hz), 3.20 (d, 1H, *J* = 9.5 Hz), 3.04 (m, 1H), 1.89 (d, 1H, *J* = 15.3 Hz), 1.63 (dd, 1H, *J* = 15.3, 5.7 Hz),

(d, 1H, *J* = 10.0 Hz), 3.37 (d, 1H, *J* = 10.0 Hz), 2.70 (ddddd, 1H, *J* = 17.8, 2.5, 2.5, 2.5, 2.5 Hz), 2.55 (ddddd, 1H, *J* = 17.8, 2.5, 2.5, 2.5, 2.5 Hz). ¹³C NMR (CDCl₃, 100 MHz): δ 140.9 (0), 138.2 (0), 128.2 (1), 128.1 (1), 127.5 (1), 127.4 (1), 127.3 (1), 126.8 (1), 125.7 (1), 122.8 (1), 76.3 (0), 77.1 (2), 73.4 (2), 61.7 (2), 28.3 (2). Anal. Calcd for C₁₉H₂₀O₄: C, 81.4; H, 7.2. Found: C, 81.1; H, 7.2.

from (*3R*^{*},*6S*^{*})-**11** (0.50 g, 2.2 mmol) as a colorless solid (0.33 g, 58%). Purification was achieved by flash chromatography on silica and recrystallization from DCM/hexane. Mp = 64 °C. IR (KBr, disk): 1032 s, 1244 s, 1304 s, 1514 s, 1612 s, 1775 s, 2838 m, 2959 s. MS *m/z* (rel intensity): 260 (M⁺, 76), 162 (61), 135 (100). ¹H NMR (CDCl₃, 600 MHz): δ 7.21 (d, 2H, *J* = 8.3 Hz, *ortho*-H, Ar), 6.87 (d, 2H, *J* = 8.3 Hz, *meta*-H, Ar), 5.96 (d, 1H, *J* = 10.7 Hz, H3,4), 5.93 (d, 1H, *J* = 10.7 Hz, H3,4), 5.10 (s, 1H, H2), 3.86 (d, 1H, *J* = 11.2 Hz, H6eq), 3.79 (d, 1H, *J* = 11.2, H6ax), 3.78 (s, 3H, OMe), 2.66–2.56 (2H, (O)CCH₂CH₂–), 2.45 (ddd, 1H, *J* = 13.2, 8.3, 5.4 Hz, (O)CCH₂CH₂–), 2.18 (ddd, 1H, *J* = 13.2, 9.8, 9.3 Hz, (O)CCH₂CH₂–). ¹³C NMR (CDCl₃, 100 MHz): δ 176.2 (0), 159.7 (0), 132.6 (1), 131.4 (0), 128.7 (1), 128.0 (1), 114.0 (1), 79.3 (0), 76.0 (1), 69.2 (2), 55.3 (3), 31.3 (2), 27.8 (2). Anal. Calcd for C₁₅H₁₆O₄: C, 69.2; H, 6.2. Found: C, 68.9; H, 6.2.

(5*S*^{*},8*S*^{*})-8-(4-Methoxyphenyl)-1,7-dioxaspiro[4,5]dec-9-en-2-one ((5*R*^{*},8*S*^{*})-17**).** The title compound was obtained from (*5S*^{*},*8S*^{*})-**17** (0.80 g, 3.4 mmol) as a colorless solid (0.48 g, 54%). Purification was achieved by flash chromatography on silica and recrystallization from DCM/hexane. Mp = 218

°C. IR (KBr, disk): 967 s, 1173 s, 1437 m, 1594 m, 1774 s, 2960 m. MS *m/z* (rel intensity): 260 (M⁺, 100), 229 (28), 162 (45), 135 (90). ¹H NMR (CDCl₃, 600 MHz): δ 7.29 (d, 2H, *J* = 8.5 Hz, *ortho*-H, Ar), 6.87 (d, 2H, *J* = 8.5 Hz, *meta*-H, Ar), 6.03 (d, 1H, *J* = 10.3, 2.0 Hz, H3), 5.90 (d, 1H, *J* = 10.3 Hz, H4), 5.03 (s, 1H, H2), 3.97 (d, 1H, *J* = 11.7 Hz, H6eq), 3.77 (s, 3H, OMe), 3.66 (d, 1H, *J* = 11.7, H6ax), 2.66 (ddd, 1H, *J* = 18.1, 10.0, 7.0 Hz, (O)CCH₂CH₂–), 2.59 (ddd, 1H, *J* = 18.1, 10.0, 7.0 Hz, (O)CCH₂CH₂–), 2.15 (ddd, 1H, *J* = 13.2, 10.0, 7.0 Hz, (O)CCH₂CH₂–), 2.10 (ddd, 1H, *J* = 13.2, 10.0, 7.0 Hz, (O)CCH₂CH₂–). ¹³C NMR (CDCl₃, 100 MHz): δ 176.0 (0), 159.6 (0), 134.0 (1), 131.0 (0), 129.0 (1), 126.0 (1), 113.8 (1), 78.9 (0), 75.9 (1), 69.6 (2), 55.2 (3), 30.1 (2), 28.0 (2). Anal. Calcd for C₁₅H₁₆O₄: C, 69.2; H, 6.2. Found: C, 69.0; H, 6.0.

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