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Synthesis, cytotoxic activity, ADMET and molecular docking study of quinoline-based hybrid compounds of 1,5-benzothiazepines

Duong Ngoc Toan,^{a,b*} Nguyen Dinh Thanh,^{b*} Mai Xuan Truong,^a Duong Nghia Bang,^c Mai Thanh Nga, ^a Nguyen Thi Thu Huong ^b

Some α , β -unsaturated ketones **4a-g** of 3-acetyl-4-hydroxyquinolin-2(1*H*)-one were prepared by its reaction with (hetero)aromatic aldehydes with yields of 61–87% using piperidine as catalyst. These ketones reacted with *o*-aminothiophenol in the presence of acetic acid to afford a series of new hybrid compounds, quinoline-benzothiazepine, **6a-g**. The yields of benzothiazepines **6a-g** were 62–85%. All the synthesized compounds **6a-g** were screened for their *in vitro* anticancer activity against human hepatocellular carcinoma HepG2 and squamous cell *carcinoma* KB cancer lines. Compounds **6d** and **6g** had the best activity in the series, with IC₅₀ values of 0.25 and 0.27 µg/mL, respectively against HepG2, and of 0.26 and 0.28 µM, respectively, against KB cell lines. ADMET properties showed that compounds **6c** and **6g** possessed the drug-likeness behavior. Cross-docking results indicated that residues GLN778(A), DA12(F), and DG13(F) in the binding pocket as potential ligand binding hot-spot residues for compounds **6c** and **6g**.

represented some bioactive benzothiazepine

Diltiazem, **A**,¹⁴ was used clinically in USA from 1982¹⁵, followed by clentiazem, **B**,¹⁶ for their cardiovascular actions. Other 1,5-

benzothiazepine derivatives were also used clinically for CNS

disorders including thiazesim, C,17 and quetiapine fumarate,

D¹⁸. Therefore, their beneficial properties have prompted

several groups to study these compounds. Owing to their

importance from a pharmacological and synthetic point of

view, several approaches have been reported for the synthesis

of 1,5-benzothiazepines. Some 1,5-benzothiazepine derivatives

were synthesized efficiently and environmentally from 2-

aminobenzenethiols with α , β -unsaturated ketones¹⁹ in good to

excellent yield under solvent-free microwave irradiations using

sulfuric acid on silica or basic alumina respectively as solid

supports²⁰, under microwave irradiation *via* Mannich

condensation²¹, from chalcones and *o*-aminothiophenol in the

presence of 10 mol% catalyst of ammonium cerium(IV) nitrate

under ultrasonic irradiation¹³, via a one-pot thia-Michael-

cyclization sequence by the reaction of various o-

aminothiophenols with chalcones in ionic liquid media, such as

1-octyl-3-methyl imidazolium thiocyanate ([omim]SCN) at 60

°C and 1-octyl-3-methyl imidazolium chloride ([omim]Cl)

medium in room temperature²², in PEG-400 catalyzed by acetic

acid²³. Another pathway to 1,5-benzothiazepine derivatives

was the reaction of ω -bromoacetophenones and aromatic

aldehydes²¹, chalcone analogues of dehydroacetic acid²⁴, 2-

dimethylitacone²⁵, α , β -unsaturated ketones with bis(2-

nitrophenyl)disulphide in the presence of TiClO₄/Sm²⁶,

nitrodisulphides and α , β -unsaturated ketones by SmI₂²⁷,

photochemical reaction of 2-phenylbenzothiazole with ethoxy

acetylene/ethoxy propyne²⁸, microwave activated reaction

between 3-(4'-fluoro-2'-methylbenzoyl)-2-propenoic acid with

itaconic

anhydride

and

with

Introduction

Heterocycles containing nitrogen and sulphur as heteroatoms undoubtedly constitute an important class of highly applicable bioactive molecules because of their interesting biological activities and uses as key structural motif for the synthesis of various products of pharmaceutical interest. Benzothiazepine is a heterocyclic compound that contains a benzene ring fused with a seven membered ring having nitrogen and sulphur atoms. Benzothiazepine derivatives are of three types: 1,4-, 4,1- and 1,5-benzothiazepines^{1, 2}. 1,5-Benzothiazepine is one of these three possible benzo-condensed derivatives. It is known that 1,5-benzothiazepine itself has not hitherto been described in the literature for its pharmacological properties^{1,} ³. However, substituted 1,5-benzothiazepines are of particular interest for lead discovery because they have been found active against different families of targets1, 2. 1,5-Benzothiazepines play a unique role in drug discovery programs, as they display a wide spectrum of biological activities such as antibacterial⁴, antifungal⁵, antiviral⁶, anticancer⁷, antihypertensive⁸ activities, and so on. 1,5-Benzothiazepine derivatives are also used as calcium channel modulators9, either inhibitors for adenosine kinase 10, cholinesterase¹¹, antagonists¹², vasodilators¹³, etc. Fig. 1

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aminophenyldisulfide

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Electronic Supplementary Information (ESI) available: Synthetic procedures and NMR characterization details for.....See DOI: 10.1039/x0xx00000x

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aminophenols (as dinucleophiles) ketones.

DOI: 10.1039/D0NJ04295A niles) with α , β -unsaturated



We have been interested in 1,5-benzodiazepines and 1,5benzothiazepine derivatives for a few years, and recently, have prepared a series of 1,5-benzothiazepines³⁰. Inspired by the manifold applications of 1,5-benzothiazepine nuclei and in continuation of our interest in the synthesis and biological evaluation studies of heterocyclic compounds containing nitrogen and sulphur as heteroatoms, we report herein the synthesis, characterization, cytotoxic activities of some novel benzothiazepines having quinoline ring.

Results and discussion

Based on above-mentioned literature methods, we have chosen the synthetic pathway of target 1,5-benzothiazepines from α , β -unsaturated ketones of 3-acetyl-4-hydroxy-1methylquinolin-2(1H)-one (3). This initial material was prepared from N-methylaniline. First, pyronoquinoline 2, hydroxy-6-methyl-2H-pyrano[3,2-c]quinoline-2,5(6H)-dione, was prepared by using Kappe's and Stadlbauer's procedures^{31,} ³² by reaction *N*-methylaniline with diethyl malonate. In Stadlbauer's procedure diphenyl ether was used as solvent, but Kappe's procedure used excess diethyl malonate and this diester played the reagent and solvent roles. The former gave higher yield of pyronoquinoline 2. The ring opening of 2 by sodium hydroxide in glycerol and subsequent spontaneous decarboxylation gave 3-acetyl-4-hydroxy-1-methylquinolin-2(1H)-one (3). α,β-Unsaturated ketones 5a-g, substituted (E)-4hydroxy-3-(3-(aryl)acryloyl)-1-methylquinolin-2(1H)-ones,

were prepared by Claisen-Schmidt reaction of 3acetylquinolone **3** with some appropriate aromatic and heteroaromatic aldehydes **4a-g** (Scheme 1). The reaction was carried out by heating under reflux in absolute ethanol as solvent for 25–50 h. Piperidine was used as catalyst for this process. The molar ratios of **3** and aldehydes **4a-g** were 1:1. Initially, the starting materials were dissolved completely in reaction solvent, then the product was separated as yellowcolour precipitates during the reaction. Products were obtained with yields of 61–87%.

IR spectra of these α,β -unsaturated ketones **5a-g** had characteristic absorption bands of *trans*-vinyl group in range of 998–967 cm⁻¹. Absorption band appeared at 1680–1634 cm⁻¹ belonged carbonyl group of lactam function. Their ¹H NMR spectra indicated the presence of this trans-vinyl group through two signals at δ = 8.59–8.41 ppm and δ = 7.95–7.87 ppm for protons H-2' and H-3', respectively. These signals had the roof effects that showed the coupling constants J =15.5–16 Hz. These values indicated that the resulting alkene had *trans* configuration. In compounds **5a-g** the phenolic hydroxyl group on position 4 of quinolone ring had no chemical shift downfield in DMSO- d_6 solvent, possibly due to hydrogen-bonding formation of this group to oxygen atom of carbonyl ketone (Fig. 2)³³. In their IR spectra, weak and narrow absorption bands presented in region at 3462-3150 cm⁻¹ that showed this intramolecular hydrogen bonding between this hydroxyl group and carbonyl oxygen atom. ¹H NMR spectrum of compound **5f** had signal at δ = 9.81 ppm that belonged 4"hydroxyl group on benzene ring.

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Scheme 1. Synthetic pathway for 1,5-benzothiazepines. Reaction conditions: (*i*) Diphenyl ether, 5 h, under reflux; (*ii*) 40% NaOH in water, glycerol, under reflux for 1h; (*iii*) Piperidine, abs. EtOH, under reflux for 25–50 h; (*iv*) Glacial acetic acid, abs. EtOH, under reflux for 5–7 h.



Next, substituted benzothiazepines **6a-g** were synthesized by ring-closure condensation reaction of obtained substituted α , β -unsaturated ketones **5a-g** with *o*-aminothiophenol. Reaction was carried out under reflux in absolute ethanol (Scheme 1) via Michael addition. At the beginning of the reaction, the initial materials were dissolved in solvent, and

after about 3-4 h, the product precipitate began to appear. The ring-closure condensation process occurred over 5-7 h with obtained product yields of 65-82%. Structurally, the formation of benzothiazepines **6a-g** from α , β -unsaturated ketones **5a-g** of 3-acetyl-4-hydroxy-1-methylquinolin-2(1*H*)one **3** could be confirmed by spectral data (IR, NMR, and MS). In IR spectra of these compounds, characteristic absorption band for out-of-plane bending vibration of C-H of trans-alkene group in α , β -unsaturated ketones **5a-g** in the region at 998–967 cm⁻¹ disappeared. Hydroxyl group on position 4 of quinoline-2-one ring had chemical shift at δ = 14.78–14.76 ppm. This is the difference when compared to the case of α , β unsaturated ketones above, in which the phenolic hydroxyl group on position 4 of quinolone ring had no chemical shift downfield in DMSO- d_6 solvent, and could be used as evidence to show that benzothiazepine ring was formed in the reaction

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of α , β -unsaturated ketones **5a-g** with 2-aminophenol. In ¹H NMR spectra of these benzothiazepines, the disappearance of signals at δ = 8.59–8.41 ppm and δ = 7.95–7.87 ppm for protons H-2' and H-3', respectively, also suggested that the benzothiazepine ring was formed.

Cytotoxic activity

All the synthesized of 3-(2-amino-6-arylpyrimidin-4-yl)-4hydroxy-1-methylquinolin-2(1H)-ones 6a-g were screened for their in vitro anticancer activity against two representatives: human squamous cell carcinoma (KB) and hepatocellular carcinoma (HepG2) cell lines. Ellipticine was used as reference. Evaluated results for 6a-g were given in Table 1 and Figs. 3 and 4. In general, it has been observed that almost all tested compounds showed remarkable activity against the tested cancer cell lines, KB and HepG2, in comparison with the IC₅₀ values of the reference compound (ellipticine). The results in Table 1 showed that almost all novel molecule exhibited anticancer activity against the tested KB cell line at wide range of concentrations. With the exception of compounds 6d and 6e that displayed negligible inhibitory effect on the KB carcinoma cell line (IC₅₀ >128 μ M), the remaining compounds had a good inhibitory effect. The order of the inhibitory effect of these compounds was 6c ~ 6g > 6f > 6b > 6a > 6d,6e of which compounds **6c** and **6g** had the best activity in the series, with IC_{50} values equal to 0.25 and 0.27 $\mu g/mL$, respectively, when compared to IC_{50} value with 0.28 μM of a positive reference drug ellipticine. Compound 6f had significant inhibitory activity with IC_{50} of 0.8 $\mu M.$

For cancer cell line HepG2, compounds 6a-g exhibited weak or insignificant activity, and two compounds 6c and 6g showed the highest activity in this sequence, with IC₅₀ values equal to 0.26 and 0.28 μ M (Table 1), respectively, when compared to the IC₅₀ value of ellipticine (with IC₅₀ = 0.36 μ M). Compound **6f** exhibited medium activity with IC_{50} value of 2.10 μ M. The remaining compounds in the series had a negligible activity for liver cancer cell lines HepG2. The order of inhibitory activity was as follows: 6c ~ 6g > 6f > 6b > 6e > 6a >6d.

Table 1. Cytotoxic activity against KB and HepG2 cancer cell line in IC₅₀ of compounds 6a-g

npds.	Ar —	IC ₅₀ (μM)		
	3″-CIC₅H₄	80.0	45.71	
	3″-MeC₀H₄	8.0	32.0	
	4"-MeOC ₆ H ₄	0.25	0.26	
	4''-BrC ₆ H ₄	>128	>128	
	$4''-Me_2NC_6H_4$	>128	32.0	
	4"-OH-3"-MeOC ₆ H ₃	0.8	2.10	
	2"-Thienyl	0.27	0.28	
oticine		0.28	0.36	
	% Inhibition against can	cer cell line KB		
100				
90	fi	<u> </u>		
80	/			
70	 			
60				
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20	╷║┝╕──┤╽╎╽┝╕╢┝	7		
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0	6a 6b 6c 6c	6 e 6	f 6g	
0		i 6e 6	f 6g	
0	6a 6b 6c 6c	i 6e 6 L Ω 2 μg/mL Ω	f 6g 0.5 μg/mL	
0 128 re 3. Do besized	6a 6b 6c 6c μg/mL 32 μg/mL 8 μg/ml se-dependent cell growth inhibitio	6e 6 L 2 μg/mL π n percentages agai	f 6g 0.5 μg/mL nst KB cell line by the	
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ADMET studies

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A variety of key ADMET (Absorption, Distribution, Metabolism, Excretion and Toxicity) properties had been calculated *in silico* to estimate the drug-likeness of the compounds (Table 2)³⁴. The predictions were carried out for synthesized compounds **6a-g** were screened using PreADMET online software³⁵ (for HIA, Caco2) and SwissADME online softwares³⁶ (for TPSA, n-ROTB, n-ON, n-OHNH, LogP, Lipinski's violations and Veber's violation). From Table 2, it could be observed that all the synthesized compounds have shown promising human intestinal percentage absorption (HIA = 95.71–98.59%). These values showed that these compounds had good human intestinal absorption. The most active compounds **6c** and **6g** showed 96.90% and 96.92% absorption, respectively. The designed structures were tested for compliance with rules evaluating bioavailability of a compound after oral administration Lipinski's rule of five and Veber's filter. The first one assumed that compounds having LogP_{o/w} (octanol/water partition coefficient) lower than 5 (with values of 3.95-4.98), molecular weight (MW) below 500, less than 10 H-bond acceptors (n-ON), and less than 5 H-bond donors (n-OHNH) were more likely to show favourable bioavailability³⁷. The Veber rule extended the range of parameters by rotatable bonds (preferably n-ROTB <10) and topological polar surface area with values of 79.89–109.35 (preferably TPSA ≤ 140 Å²)³⁸. The Egan rule considered good bioavailability for compounds with TPSA ≤ 132 Å² and $-1 < LogP < 6^{39}$. These compounds had middle cell permeability with Caco2 values in ranges of 24.51–49.87 mm/s.

Table 2. Physicochemical properties, lipophilicity and drug-likeness of compounds 6a-g										
Entry	Ar	MW ^[a]	TPSA ^[b]	n-ROTB ^[c]	n-ON ^[d]	n-OHNH ^[e]	LogP ^[f]	L.V. ^[g]	V.V. ^[h]	Caco2 ^[i]
6a	3"-CIC ₆ H ₄	446.95	79.89	2	3	1	4.94	1	0	49.87
6b	3"-MeC ₆ H ₄	426.53	79.89	2	3	1	4.66	1	0	31.03
6c	4"-MeOC ₆ H ₄	442.53	89.12	3	4	1	4.41	0	0	29.33
6d	$4''-BrC_6H_4$	491.40	79.89	2	3	1	4.98	1	0	37.73
6e	$4''-Me_2NC_6H_4$	455.57	83.13	3	3	1	4.35	0	0	30.31
6f	4"-OH-3"-MeOC ₆ H ₃	458.53	109.35	3	5	2	3.95	0	0	24.51
6g	2"-Thienyl	418.53	108.13	2	3	1	4.41	0	0	34.01

Note. [a] MW: molecular weight (<500, expressed as Dalton); [b] TPSA: topological polar surface area (Å²); [c] n-ROTB: number of rotatable bonds; [d] n-ON: number of hydrogen bond acceptors (<10); [e] n-OHNH: number of hydrogen bond donors (<5); [f] LogP: logarithm of partition coefficient (<5) of compound between n-octanol and water ³⁷; [g] L.V.: Lipinski's violations ³⁷; [h] V.V.: Veber's violation ³⁸; [i] Caco-2: Caco-2 cell permeability (P_{Caco-2} (nm/s), <4: low, 4–70: middle, >70: high); [j] HIA: human intestinal absorption (0–20=poor, 20–70=moderate, 70–100=good).

Molecular Modelling

Molecular docking simulations were performed in order to better understand the molecular basis for the inhibition of topoisomerases by ellipticine (reference drug) and compounds (ligands, herein) 6a-g against above- mentioned cancer cell lines. Based on the results obtained from the enzymatic assays, molecular modelling studies were performed as a step toward understanding the interaction mode of these compounds as inhibitors. It is known that ellipticine (5,11dimethyl-6H-pyrido[4,3-b]carbazole, an alkaloid from extracted from trees of the species Ochrosia elliptica and Rauvolfia sandwicensis ⁴⁰) is a potent antineoplastic agent exhibiting multimodal mechanism of action. This compound inhibited the enzyme topoisomerase II via intercalative binding to DNA. The prevalent mechanisms of ellipticine antitumor, mutagenic and cytotoxic activities were suggested to be intercalation into DNA and inhibition of DNA topoisomerase II activity^{41, 42}. Human topoisomerase II_β in complex with DNA and mitoxantrone was a targeting anticancer drug, and so we have chosen it for docking study^{43, 44}. This enzyme was retrieved with PDB code 4G0V. The structure has a resolution

of 2.6 Å with intercalated drug mitoxantrone, and consisted of 7 chains: A, B, C, D, E, and F, two molecules of mitoxantrone had complexed with these chains. Two of the most active benzothiazepine compounds (**6c** and **6g**), one moderately active compound (**6f**), and one of the least active compound (**6d**) were chosen for docking examination. The favourably docked molecules were ranked according to the XP Glide Score. Obtained docking results (glide score, in kcal/mol) were represented in Table 3. The order of glide scores was **6c** > **6g** > **6f** > **6d**, which indicated that ligands **6c** and **6g** were docked nicely fitted into the active site. The active pocked was formed from chains A, D, F.

able 3. Molecular docking analysis of protein target 6QXG with ligands 6c,6f and 6g						
Ligands	Ar	Glide score ^[a]				
6c	4-MeOC ₆ H ₄	-7.964				
6d	$4-BrC_6H_4$	-7.555				
6f	4-OH-3-MeOC ₆ H ₃	-7.876				
6g	2-Thienyl	-7.948				
Ellipticine		-8.102				

[a] In kcal/mol.

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Figure 5. Docked (superimposed) poses showed ellipticine (magenta), 6c (red), (6d, orange), 6f (blue), and 6g (green), all binding at the same position on molecular pocket of human topoisomerase IIB in complex with DNA (4GOV) and showing common interactions with residues GNL A:778, ARG A:503, DA F:12, DG F:13.

Superimposed poses in Fig. 5 showed that four ligands 6c (red), (6d, orange), 6f (blue), and 6g (green), all bound at the same position that the reference ellipticine (magenta) did on molecular pocket of human topoisomerase IIB in complex with DNA. Ligands 6c (with 4-methoxyphenyl group), 6f (4-hydroxy-3-methoxyphenyl group), and 6g (with 2-thienyl group) gave better glide scores when compared with other five compounds for the target protein, with binding score of -7.964, -7.876, and -7.948 kcal/mol, respectively. These values could be compared with the one of ellipticine (-8.102 kcal/mol). The worst active ligand 6d with 5-bromophenyl group had glide score of -7.555 kcal/mol.

The important intermolecular protein-ligand interactions of compounds 6c and 6g are depicted in Fig. 6 (top). The lowest energies pose the highest active compounds 6c and 6g obtained by Glide 8.1 software are reported in Fig. 6 (bottom), similar to the docking position of ellipticine (Fig. 6, middle). Ellipticine itself had some active interactions with the residues in active site, such as π -cation (ARG503 on chain A), π - π stacking (DA12 and DG13 on chain F; DC8 on chain C; DT9, on chain D). There was not any hydrogen bonding interaction of this drug with the residues in active site. The ligands 6c, 6d, 6f, and 6g had common interactions with the residues ARG503 (on chain A), GLN778(on chain A), DA12 (on chain F), and DG13 (on chain F). There were some intermolecular residue interactions with compounds 6c and 6g, some interactions were the same ones in case of ellipticine: π -cation (ARG503 on chain A), π - π stacking (DA12 and DG13 on chain F). Another ligand-amino acid interactions of residue GLN778 on chain A of enzyme 4G0V took place in these ligands. This interaction was 58 hydrogen bonding of GLN778 with C=O (lactam) group of 4hydroxyquinolin-2(1H)-one moiety (Fig. 6). Additionally, ligand **6g** had more π - π stacking interactions than ligand **6c**. This one could explain the obtained higher glide score value of this ligand (Table 3).

Experimental

Melting points were determined by open capillary method on STUART SMP3 instrument (BIBBY STERILIN, UK) and are uncorrected. IR spectra (KBr disc) were recorded on an Impact 410 FT-IR Spectrometer (Nicolet, USA). ¹H and ¹³C NMR spectra were recorded on Bruker Avance Spectrometer AV500 (Bruker, Germany) at 500 MHz and 125 MHz, respectively, using DMSO d_6 as solvent and TMS as an internal standard. ESI-mass spectra were recorded on LTQ Orbitrap XL[™] (Thermo Fisher Scientific Co., USA) and ESI-MSD-Trap-SL (Agilent Technologies, Inc., USA) mass spectrometers) mass spectrometers. All reactions were monitored by thin layer chromatography, carried out on silica gel 60 WF254S aluminium sheets (Merck, Germany) and was visualized with UV light. Chemical reagents in high purity were purchased from the Merck Chemical Company (in Viet Nam). All materials were of reagent grade for organic synthesis. 4-Hydroxy-6-methyl-2H-pyrano[3,2c]quinoline-2,5(6H)-dione (2, pyronoquinoline) was prepared from N-methylaniline according to known method³². 3-Acetyl-4-hydroxy-1-methylquinolin-2(1*H*)-one was prepared by modified procedure according to literatures⁴⁵, in which glycerol was used instead ethylene glycol, as follows.

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Figure 6. Residues involved in intermolecular interactions of ligands 6c (top left), 6g (top right), and ellipticine (middle left) in active pocked of human topoisomerase II \$\beta\$ in complex with DNA (4G0V). The molecular 4G0V interactions of 6c and 6g (bottom) were depicted as dashed lines (orange – hydrogen bonds, green – π/π stacking).

Synthesis of 3-acetyl-4-hydroxy-1-methylquinolin-2(1H)-one (3)

This compound was prepared by modified procedure⁴⁵. To suspension of pyronoquinoline 2 (0.103 mol, 25 g) in glycerol (321 mL) a 40% aqueous solution of sodium hydroxide (0.515

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mol, 32.1 mL) was added. Reaction mixture was boiled within 1 hour. The obtained solution was cooled in ice bath and poured into cold water (642 mL), and neutralized the solution by concentrated HCl until the precipitate completely separated (to acidic medium, pH 3). The separated precipitates were filtered, washed with water to pH 7, dried at temperature of 80°C, and crystallized from 96% ethanol. Yield: 12.51 g (56%).

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M.p. 141-142°C, ref.³²: 141-142.5°C. IR (KBr, v, cm⁻¹): 3468 (v_{OH}), 1650 (v_{C=O lactam}), 1620 (v_{C=O conj. acetyl}). ¹H NMR (500 MHz, DMSO-*d*₆), δ (ppm): 2.71 (3H, s, 3-COCH₃), 3.54 (s, 3H, *N*-CH₃), 7.31 (t, J = 8.0 Hz, 1H, H-6), 7.50 (d, J = 8.0 Hz, 1H, H-5), 7.79 (td, J = 8.0, 1.5 Hz, 1H, H-7), 8.06 (dd, J = 8.0, 1.5 Hz, 1H, H-8).

General procedure for synthesis of (E)-4-hydroxy-3-(3-(aryl/hetaryl)acryloyl)-1-methylquinolin-2(1H)-ones (5a-g)

To a mixture of 3-acetyl-4-hydroxy-N-methyl-2(1H)-quinolone (3, 5 mmol) and appropriate (un)substituted benzaldehydes 4a-f or thiophene-2-carboxaldehyde 4g (5 mmol) in absolute ethanol (25 mL) was added piperidine (1 mol%, 0.1 mL). The reaction mixture was heated under reflux for 25-50 h. After reaction, solvent was led to evaporate to half a volume. Separated solid product was filtered, washed with a little of 96% ethanol (2×2 mL), recrystallized from appropriate solvents to afford the titled compounds 5a-g.

(E)-4-Hydroxy-3-(3-(3-chlorophenyl)acryloyl)-1-methylquinolin-2(1H)-one (5a)

From 3 (5 mmol, 1085 mg) and 3a (Ar=3"-ClC₆H₄, 5 mmol, 702 mg), under reflux for 35 h. Yield: 1324 mg (78%) of 5a as pale-yellow crystals. M.p.: 168–169°C (96% ethanol/DMF 5:1 by volume); $R_{\rm f}$ = 0.70 (*n*-hexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3102, 1650, 1623, 1596, 1542, 983. ¹H NMR (500 MHz, DMSO-*d₆*), δ (ppm): 8.57 (d, J = 16.0 Hz, 1H, H-3'), 8.17 (dd, J = 8.0 and 1.5 Hz, 1H, H-5), 7.87 (d, J = 16.0 Hz, 1H, H-2'), 7.81 (td, J = 8.0 and 1.5 Hz, 1H, H-7), 7.76 (s, 1H, H-2"), 7.71–7.69 (m, 1H, H-6"), 7.56 (d, J = 8.0 Hz, 1H, H-8), 7.52–7.51 (m, 2H, H-4", H-5"), 7.34 (t, J = 8.0 Hz, 1H, H-6), 3.62 (s, 3H, *N*-CH₃). ESI-MS (+): calcd. for C₁₉H₁₄³⁵CINO₃, M=339.1 Da, found: m/z 399.9 [M]*.

(E)-4-Hydroxy-3-(3-(3-methylphenyl)acryloyl)-1-methylquinolin-2(1H)-one (5b)

From 3 (5 mmol, 1085 mg) and 4b (Ar=3"-MeC₆H₄, 5 mmol, 600 mg), under reflux for 32 h. Yield: 1308 mg (87%) of 5b as yellow crystals. M.p.: 172-173°C (96% ethanol); R_f = 0.82 (nhexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3110, 3034, 2917, 2849, 1633, 1625, 980, 870, 797, 750. ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 8.57 (d, J = 16.0 Hz, 1H, H-3'), 8.13 (d, J = 8.0 Hz, 1H, H-5), 7.88 (d, J = 16.0 Hz, 1H, H-2'), 7.80 (t, J = 8.0 Hz, 1H, H-7), 7.53–7.51 (m, 3H, H-8, H-2", H-6"), 7.38-7.29 (m, 3H, H-6, H-4", H-5"), 3.58 (s, 3H, N-CH₃), 2.36 (s, 3H, 3"-CH₃).

(E)-4-Hydroxy-3-(3-(4-methoxyphenyl)acryloyl)-1-methylquinolin-2(1H)-one (5c)

From 3 (5 mmol, 1085 mg) and 4c (Ar=4"-MeOC₆H₄, 5 mmol, 680 mg), under reflux for 30 h. Yield: 1423 mg (85%) of 5c as yellow crystals. M.p.: 178-179°C (96% ethanol/DMF 5:1 by volume); $R_f = 0.81$ (*n*-hexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3103, 2971, 2933, 2838, 1643, 1599, 1532, 1504, 980. ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 8.51 (d, J = 15.5 Hz, 1H, H-2'), 8.13 (d, J = 8.0 Hz, 1H, H-5), 7.93 (d, J = 15.5 Hz, 1H, H-3'), 7.81 (t, J = 8.0 Hz, 1H, H-7), 7.72 (d, 2H, J = 8.5 Hz, H-3", H-5"), 7.55 (d, J = 8.0 Hz, 1H, H-8), 7.33 (t, J = 8.0 Hz, 1H, H-6), 7.06 (d, 2H, J = 8.5 Hz, H-2", H-6"), 3.83 (s, 3H, 4"-OCH₃), 3.59 (s, 3H, N-CH₃). ESI-MS (+): calcd. for C₂₀H₁₇NO₄, M=335.1 Da, found: m/z 335.9 [M]+.

(E)-4-Hydroxy-3-(3-(4-bromophenyl)acryloyl)-1-methylquinolin-2(1H)-one (5d)

From 3 (5 mmol, 1085 mg) and 4d (R =4'-Br, 5 mmol, 925 mg), under reflux for 25 h. Yield: 1632 mg (80%) of 5d as pale-yellow crystals. M.p.: 200–201°C (96% ethanol/DMF 5:1 by volume); $R_{\rm f}$ = 0.68 (*n*-hexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3088, 3045, 2925, 1680, 1616, 1535, 979, 756. ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 8.58 (d, J = 16.0 Hz, 1H, H-3'), 8.12 (dd, J = 8.0, 1.0 Hz, 1H, H-5), 7.86 (d, J = 16.0 Hz, 1H, H-2'), 7.81 (td, J = 8.0, 1.5 Hz, 1H, H-7), 7.72-7.67 (m, 4H, H-2", H-3", H-5", H-6"), 7.55 (d, J = 8.5 Hz, 1H, H-8), 7.33 (t, J = 8.5 Hz, 1H, H-6), 3.58 (s, 3H, N-CH₃). ESI-MS (-): calcd. for $C_{19}H_{14}^{79}BrNO_3$, M-H=382.0 Da, found: m/z 381.9 [M-H]+.

(E)-4-Hydroxy-3-(3-(4-(N,N-dimethylamino)phenyl)acryloyl)-1methylquinolin-2(1H)-one (5e)

From 3 (5 mmol, 1085 mg) and 4e (Ar=4"-Me₂NC₆H₄, 5 mmol, 820 mg), under reflux for 42 h. Yield: 1061 mg (61%) of 5e as red brown crystals. M.p.: 220-221°C (96% ethanol/DMF 5:2 by volume); $R_f = 0.78$ (*n*-hexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3110, 2910, 2807, 2538, 1636, 1595, 1504, 979, 762. ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 8.43 (d, J = 15.5 Hz, 1H, H-3'), 8.10 (d, J = 8.0 Hz, 1H, H-5), 7.94 (d, J = 15.5 Hz, 1H, H-2'), 7.75 (t, J = 8.0 Hz, 1H, H-7), 7.58 (s, 2H, J = 8.5 Hz, H-3", H-5"), 7.50 (d, J = 8.0 Hz, 1H, H-8), 7.28 (t, J = 8.0 Hz, 1H, H-6), 6.77 (d, 2H, J = 8.5 Hz, H-2", H-6"), 3.56 (s, 3H, N-CH₃), 3.02 [s, 6H, 4'-N(CH₃)₂].

(E)-4-Hydroxy-3-(3-(4-hydroxy-3-methoxyphenyl)acryloyl)-1methylquinolin-2(1H)-one (5f)

From 3 (5 mmol, 1085 mg) and 4f (Ar=4"-OH-3"-MeOC₆H₃, 5 mmol, 706 mg), under reflux for 38 h. Yield: 1440 mg (82%) of 5f as yellow crystals. M.p.: 232-233°C (96% ethanol/DMF 5:1 by volume); $R_f = 0.81$ (*n*-hexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3365, 3116, 3010, 2962, 2838, 1634, 1616, 1597, 1505, 998, 747. ¹H NMR (500 MHz, DMSOd₆), δ (ppm): 9.81 (s, 1H, 4"-OH), 8.43 (d, J = 16.0 Hz, 1H, H-3'), 8.07 (d, J = 8.0 Hz, 1H, H-5), 7.88 (d, J = 16.0 Hz, 1H, H-2'), 7.74 (td, J = 8.0, 1.5 Hz, 1H, H-7), 7.47 (d, J = 8.0 Hz, 1H, H-8), 7.28-7.21 (m, 3H, H-6, H-2", H-5"), 6.88 (d, 1H, J = 8.0 Hz, H-6"), 3.85 (s, 3H, 3"-OCH₃), 3.53 (s, 3H, N-CH₃).

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(E)-4-Hydroxy-3-(3-(2-thienyl)acryloyl)-1-methylquinolin-2(1H)one (5g)

From **3** (5 mmol, 1085 mg) and **4g** (Ar =2"-thienyl, 5 mmol, 701 mg), under reflux for 50 h. Yield: 1078 mg (70%) of **5g** as bright yellow crystals. M.p.: 211–212°C (96% ethanol); $R_f = 0.80$ (*n*-hexane/ethyl acetate = 5:3 by volume). IR (KBr), v (cm⁻¹): 3106, 3086, 1655, 1607, 1597, 1532, 967, 755. ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 8.41 (d, J = 15.5 Hz, 1H, H-3'), 8.12–8.10 (m, 2H, H-5, H-2'), 7.79–7.78 (m, 2H, H-7, H-5"), 7.61 (d, 1H, J = 3.5 Hz, H-3"), 7.53 (d, J = 8.0 Hz, 1H, H-8), 7.31 (t, J = 8.0 Hz, 1H, H-6), 7.20 (td, 1H, J = 3.5, 1.0 Hz, H-4"), 3.58 (s, 3H, N-CH₃).

General procedure for synthesis of 3-(2aryl/hetaryl-1,5-benzothiazepin-4-yl)-4-hydroxy-1methyl-quinolin-2(1*H*)-ones (**6a-g**)

A reaction mixture of appropriate α , β -(un)saturated ketone **5a-g** (1 mmol) and *o*-aminothiophenol (1 mmol, 125 mg, 0.1 mL) in absolute ethanol (30 mL). Glacial acetic acid (0.05 mL, 5 mol%) was added to this mixture. Obtained reaction mixture was heated under reflux for 5–7 h. Separated products were filtered and recrystallized from a mixture of DMF and 96% ethanol (in appropriate volume rations) to afford the compounds **6a-g**.

3-(2-(3-Chorophenyl)-1,5-benzothiazepin-4-yl)-4-hydroxy-1methyl-quinolin-2(1*H*)-one (6a)

From **5a** (Ar=3"-ClC₆H₄, 1 mmol, 339 mg) under reflux for 6 h. Yield: 299 mg (67%) of **6a** as white crystals. M.p.: 222-223°C (from 96% ethanol/DMF=5:1 in volume), $R_f = 0.80$ (TLC solvent system: *n*-hexane/acetone 5:3). IR (KBr), v (cm⁻¹): 3428 (v_{OH}), 3061 (v_{C-H} aryl), 2940, 2887 (v_{C-H} alkyl), 1628 (v_{C=O} lactam), 1590, 1550, 1478 ($v_{C=C}$ arene). ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 14.76 (s, 1H, 4-OH), 8.16-8.12 (m, 1H, H-5), 7.69-7.61 (m, 3H, H-7, H-6', H-8'), 7.53 (d, J = 7.5 Hz, 1H, H-9'), 7.43-7.31 (m, 6H, H-8, H-7', H-2", H-4", H-5", H-6"), 7.23 (t, J = 7.5 Hz, 1H, H-6), 5.33–5.30 (m, 1H, H_c), 4.34–4.32 (m, 1H, H_b), 3.52 (s, 3H, N-CH₃), 2.75-2.70 (m, 1H, H_a); ¹³C NMR (125 MHz, DMSOd₆), δ (ppm): 145.7 (C-4), 134.8 (C-6'), 134.8 (C-3''), 134.0 (C-7), 132.8 (C-2"), 130.3 (C-8'), 130.1 (C-6"), 127.7 (C-7'), 127.0 (C-5"), 126.7 (C-5), 126.3 (C-1"), 125.7 (C-4"), 124.7 (C-9'), 124.4 (C-6), 116.9 (C-4a), 114.3 (C-8), 101.0 (C-3), 54.7 (C-2'), 54.9 (OCH₃), 38.2 (C-3'), 28.4 (N-CH₃); ESI-MS: calcd. for C₂₅H₁₉³⁵ClN₂O₂S/C₂₅H₁₉³⁷ClN₂O₂S, M=446.94/448.94 Da, found: m/z 447 (100%) [M-H]⁺ and 449 (32%) [M+2-H]⁺.

3-(2-(3-Methylphenyl)-1,5-benzothiazepin-4-yl)-4-hydroxy-1methyl-quinolin-2(1*H*)-one (6b)

From **5b** (Ar=3"-MeC₆H₄, 1 mmol, 319 mg) under reflux for 7 h. Yield: 349 mg (82%) of **6b** as white crystals. M.p.: 208–209°C (from 96% ethanol/DMF=5:1 in volume), R_f =0.76 (TLC solvent

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system: *n*-hexane/acetone 5:2). IR (KBr), v (cm⁻¹): 3444 (v_{OH}), 2933 (v_{C-H} alkyl), 1627 (v_{C=O} lactam), 1600, 1548, 1479 (v_{C=C} arene). ¹H NMR (500 MHz, DMSO-*d*₆), δ (ppm): 14.78 (s, 1H, 4-OH), 8.16–8.10 (m, 1H, H-5), 7.69–7.60 (m, 3H, H-7, H-6', H-8'), 7.53 (d, *J* = 7.5 Hz, 1H, H-9'), 7.44–7.40 (m, 2H, H-8, H-7'), 7.24–7.08 (m, 5H, H-6, H-2'', H-4'', H-5'', H-6''), 5.26 (dd, *J* = 13.0, 4.5 Hz, 1H, H_c), 4.44 (dd, *J* = 11.0, 3.0 Hz, 1H, H_b), 3.51 (s, 3H, *N*-CH₃), 2.73-2.67 (m, 1H, H_a), 2.28 (s, 3H, 3''-CH₃); ¹³C NMR (125 MHz, DMSO-*d*₆), δ (ppm): 143.4 (C-4), 134.9 (C-6'), 133.6 (C-7), 130.1 (C-8'), 128.0 (C-4''), 127.7 (C-5''), 127.6 (C-7'), 127.5 (C-3''), 127.2 (C-5), 126.3 (C-1''), 124.6 (C-9'), 122.7 (C2''), 121.0 (C-6), 119.8 (C-4a), 114.2 (C-8), 55.7 (C-2'), 38.4 (C-3'), 28.4 (*N*-CH₃), 20.56 (3''-CH₃); ESI-MS: calcd. for C₂₆H₂₂N₂O₂S, M=426.53 Da, found: *m/z* 427.0 [M+H]⁺.

3-(2-(4-Methoxyphenyl)-1,5-benzothiazepin-4-yl)-4-hydroxy-1methyl-quinolin-2(1*H*)-one (6c)

From 5c (Ar=4"-MeOC₆H₄, 1 mmol, 335 mg) under reflux for 5 h. Yield: 318 mg (72%) of 6c as white crystals. M.p.: 213-214°C (from 96% ethanol/DMF=5:2 in volume), $R_f = 0.68$ (TLC solvent system: *n*-hexane/acetone 5:3). IR (KBr), v (cm⁻¹): 3415 (v_{OH}), 3061 (v_{с-н} aryl), 2948, 2836 (v_{с-н} alkyl), 1631 (v_{с=0} lactam), 1590, 1550, 1478 ($v_{C=C}$ arene). ¹Η NMR (500 MHz, DMSO- d_6), δ (ppm): 14.77 (s, 1H, 4-OH), 8.17 (d, 1H, H-5), 7.68-7.66 (m, 2H, H-7, H-6'), 7.61 (t, J = 7.5 Hz, 1H, H-8'), 7.51 (d, J = 7.5 Hz, 1H, H-9'), 7.42 (t, J = 7.5 Hz, 2H, H-8, H-7'), 7.27 (d, J = 7.5 Hz, 2H, H-2", H-6"), 7.23 (t, J = 8.0 Hz, 1H, H-6), 6.89 (d, J = 7.5 Hz, 2H, H-3", H-5"), 5.28 (d, 1H, H_c), 4.44-4.36 (m, 1H, H_b), 3.75 (s, 3H, 4"-OCH₃), 3.54 (s, 3H, N-CH₃), 2.73-2.71 (m, 1H, H_a); ¹³C NMR (125 MHz, DMSO- d_6), δ (ppm): 162.5 (C-2 & C-4'), 158.4 (C-4''), 139.9 (C-4), 139.5 (C-9a'), 135.9 (C-1"), 135.0 (C-6'), 133.7 (C-7), 130.2 (C-8'), 127.8 (C-7'), 127.2 (C-5a'), 126.9 (C-2", C-6"), 125.9 (C-5), 124.7 (C-9'), 121.2 (C-6), 120.1 (C-4a), 114.4 (C-8), 113.7 (C-3", C-5"), 91.5 (C-3), 55.5 (C-2'), 54.9 (4"-OCH₃), 39.4 (C-3'), 28.5 (N-CH₃); ESI-MS: calcd. for C₂₆H₂₂N₂O₃S, M=442.52 Da, found: m/z 441 (100%) [M-H]+.

3-(2-(4-Bromophenyl)-1,5-benzothiazepin-4-yl)-4-hydroxy-1methyl-quinolin-2(1*H*)-one (6d)

From **5d** (Ar=4"-BrC₆H₄, 1 mmol, 384 mg) under reflux for 6 h. Yield: 382 mg (78%) of **6d** as white crystals. M.p.: 221–222°C (from 96% ethanol/DMF=5:1 in volume), $R_f = 0.60$ (TLC solvent system: *n*-hexane/acetone 5:2). IR (KBr), v (cm⁻¹): 3525, 3320 (v_{OH}), 3189, 3061 (v_{C-H} aryl), 2977, 2858 (v_{C-H} alkyl), 1628 (v_{C=0} lactam), 1593, 1541, 1463 (v_{C=C} arene). ¹H NMR (500 MHz, DMSO-*d*₆), δ (ppm): 14.76 (s, 1H, 4-OH), 8.18 (d, 1H, H-5), 7.70–7.67 (m, 2H, H-7, H-6'), 7.62 (t, *J* = 7.5 Hz, 1H, H-8'), 7.55–7.52 (m, 3H, H-3", H-5", H-9'), 7.44–7.41 (m, 2H, H-7', H-8), 7.36 (d, *J* = 8.0 Hz, 2H, H-2", H-6"), 7.24 (t, *J* = 7.5 Hz, 1H, H-6), 5.34 (1H, d, H_c), 4.42 (1H, dd, H_b), 3.56 (s, 3H, *N*-CH₃), 2.75 (1H, t, H_a); ¹³C NMR (125 MHz, DMSO-*d*₆), δ (ppm): 172.6 (C-4), 162.3 (C-4"), 143.2 (C-2), 141.0 (C-8a), 139.7 (C-9a'), 135.4 (C-1"), 134.1 (C-6'), 131.5 (C-3", C-5"), 130.8 (C-8'), 128.3 (C-2", C-

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6"), 127.0 (C-5a'), 126.5 (C-7'), 125.3 (C-5), 125.2 (C-9'), 121.6 (C-6), 120.5 (C-7), 120.1 (C-4a), 114.8 (C-8), 101.4 (C-3), 55,2 (C-2'), 39.5 (C-3'), 28.7 (*N*-CH₃); ESI-MS: calcd. for $C_{25}H_{19}^{79}BrN_2O_2S/C_{25}H_{19}^{81}BrN_2O_2S$, M=490.52/492.52 Da, found: *m/z* 489 (9.7%) [M-H]⁺ and 491 (9.5%) [M+2-H]⁺.

3-(2-(4-(N,N-Dimethylamino)phenyl)-1,5-benzothiazepin-4-yl)-4hydroxy-1-methyl-quinolin-2(1H)-one (6e)

From 5e (Ar=4"-Me₂NC₆H₄, 1 mmol, 348 mg) under reflux for 7 h. Yield: 354 mg (78%) of 6e as red brown crystals. M.p.: 228–229°C (from 96% ethanol/DMF=5:2 in volume), $R_{\rm f} = 0.81$ (TLC solvent system: *n*-hexane/acetone 5:3). IR (KBr), v (cm⁻¹): 3450 (v_{OH}), 2895, 2794 (v_{C-H} alkyl), 1630 (v_{C=O} lactam), 1610, 1542, 1482 ($v_{C=C}$ arene); ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 14.77 (s, 1H, 4-OH), 8.17-8.11 (m, 1H, H-5), 7.69-7.52 (m, 4H, H-7, H-6', H-8', H-9'), 7.46-7.41 (m, 2H, H-8, H-7'), 7.23 (1H, t, J = 7.5 Hz, H-6), 7.13 (d, 2H, J = 8.5 Hz, H-3", H-5"), 6.66 (d, 2H, J = 8.5 Hz, 2H, H-2", H-6"), 5.25 (dd, 1H, J = 12.0, 4.0 Hz H_c), 4.42-4.28 (m, 1H, H_b), 3.52 (s, 3H, N-CH₃), 2.87 [s, 6H, N(CH₃)₂], 2.73–2.63 (m, 1H, H_a); ¹³C NMR (125 MHz, DMSO-d₆), δ (ppm): 149.6 (C-4"), 139.9 (C-4), 134.9 (C-6'), 133.4 (C-7), 131.2 (C-1"), 129.9 (C-8'), 127.5 (C-7'), 126.2 (C-3", C-5"), 125.9 (C-5), 124.5 (C-9'), 121.1 (C-6), 114.2 (C-8), 111.9 (C-2", C-6"), 56.0 (CH₃), 55.9 (C-2'), 38.7 (C-3'), 28.4 (N-CH₃).

3-(2-(4-Hydroxy-3-methoxyphenyl)-1,5-benzothiazepin-4-yl)-4hydroxy-1-methyl-quinolin-2(1*H*)-one (6f)

From 5f (Ar=4"-OH-3"-MeOC₆H₃, 1 mmol, 351 mg) under reflux for 6 h. Yield: 284 mg (62%) of 6f as white crystals. M.p.: 241-242°C (from 96% ethanol/DMF=5:1 in volume), R_f = 0.72 (TLC solvent system: *n*-hexane/acetone 5:2). IR (KBr), v (cm⁻¹): 3475 (v_{OH}), 2935, 2837 (v_{C-H} alkyl), 1628 (v_{C=O} lactam), 1595, 1552 ($v_{C=C}$ arene). ¹H NMR (500 MHz, DMSO- d_6), δ (ppm): 14.77 (s, 1H, C4-OH), 8.99 (s, 1H, C4" -OH), 8.17-8.11 (m, 1H, H-5), 7.70-7.66 (m, 2H, H-7, H-6'), 7.61 (t, J=7.5 Hz, 1H, H-8'), 7.55 (d, J=7.5 Hz, 1H, H-9'), 7.47-7.41 (m, 2H, H-8, H-7'), 7.23 (1H, t, J=7.5 Hz, H-6), 6.90 (m, 1H, H-6"), 6.72 (m, 2H, 2H, H-2", H-3"), 5.26-5.22 (m, 1H, Hc), 4.33-4.31 (m, 1H, Hb), 3.71 (s, 3H, 4"-OCH₃), 3.52 (s, 3H, N-CH₃), 2.75-2.60 (m, 1H, Ha); ¹³C NMR (500 MHz, DMSO-d₆), δ (ppm): 180.5 (C-4'), 172.8 (C-4), 162.3 (C-4"), 147.4 (C-5"), 146.0 (C-2), 141.0 (C-8a), 139.7 (C-9a'), 135.4 (C-1"), 135.0 (C-6'), 134.0 (C-7), 130.5 (C-8'), 128.1 (C-7'), 127.6 (C-5a'), 126.1 (C-5), 125.0 (C-9'), 121.4 (C-6), 120.0 (C-4a), 118.2 (C-6"), 115.1 (C-3"), 114.7 (C-8), 110.2 (C-2"), 101.4 (C-3), 55.4 (C-2'), 55.9 (OCH₃), 39.3 (C-3'), 28.6 (N-CH₃).

5051523-(2-(2-Thienyl)-1,5-benzothiazepin-4-yl)-4-hydroxy-1-methyl-99</tr

53From **5g** (Ar=2"-thienyl, 1 mmol, 312 mg) under reflux for 7 h.54Yield: 356 mg (85%) of **6g** as white crystals. M.p.: 216–217°C55(from 96% ethanol/DMF=5:1 in volume), $R_f = 0.81$ (TLC solvent56system: n-hexane/acetone 5:2). IR (KBr), v (cm⁻¹): 3450 (v_{OH}),573059 (v_{C-H} aryl), 1630 (v_{C=0} lactam), 1600, 1548, 1475 (v_{C=C}

arene). ¹H NMR (500 MHz, DMSO-*d*₆), δ (ppm): 8.17 (s, 1H, H-5), 7.70–7.61 (m, 3H, H-7, H-6', H-8'), 7.52 (dd, *J* = 8.0, 2.0 Hz, 1H, H-9'), 7.45–7.40 (m, 3H, H-8, H-7', H-5''), 7.24 (td, *J* = 8.0, 1.0 Hz, 1H, H-6), 7.05 (s, 1H, H-3''), 6.98 (td, *J* = 5.0, 3.5 Hz, 1H, H-4''), 5.60 (dd, *J* = 12.0, 4.5 Hz, 1H, H_c), 4.56 (s, 1H, H_b), 3.55 (s, 3H, *N*-CH₃), 2.69–2.63 (m, 1H, H_a); ¹³C NMR (125 MHz, DMSO-*d*₆), δ (ppm): 148.5 (C-4), 147.1 (C-2''), 139.5 (C-9a'), 135.5 (C-6'), 133.8 (C-7), 130.5 (C-8'), 126.4 (C-5''), 126.3 (C-5), 126.0 (C-5a'), 125.9 (C-5), 124.7 (C-9'), 124.6 (C-3''), 123.6 (C-4''), 121.2 (C-6), 114.4 (C-8), 90.6 (C-3), 51.4 (C-2'), 39.0 (C-3'), 28.5 (*N*-CH₃); ESI-MS: calcd. for C₂₃H₁₈N₂O₂S₂, M=418.53 Da, found: *m/z* 419 [M+H]⁺.

Cytotoxicity assay

Dilution series (128, 32, 8, 2, and 0.5 µg/mL of each compound 6a-g) were prepared and used for 3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide (MTT) assay ⁴⁶. Two cancer cell lines were seeded at a density of 3×10⁴ cells/well and treated with a range of concentrations in triplicate in 96-well cell culture plates, whereupon cell proliferation was assessed using a standard MTT assay. Specifically, the growth inhibitory activity of benzothiazepines was determined using MTT, which correlates the cell number with the mitochondrial reduction of MTT to a blue formazan precipitate. In brief, the cells were plated in 96-well plates and allowed to attach overnight. The medium was then replaced with serum-free medium containing the test compounds and cells were incubated at 37°C for 72 h. The medium was then replaced with fresh medium containing 1 mg/mL MTT. Following incubation at 37°C for 2-4 h, the wells were aspirated, the dye was solubilized in DMSO and the absorbance was measured at 540 nm using a Tecan[™] GENios[®] Microplate Reader (Conquer Scientific, USA). The viability of cells was compared with that of the control cells. The slope of the absorbance change was used for calculating the reaction rate. Negative controls were performed in the absence of enzyme and compound, and positive controls in the presence of enzyme and 100% DMSO. The percentage of residual activity was calculated as the difference in absorbance between the time 6 and 2 min, obtained by the average of two experiments carried out in triplicate. The obtained rate was related to the rate when the inhibitor was absent. IC₅₀ values were calculated from linear extrapolations of reaction rate (as a function of the logarithm of the concentration). The IC₅₀ values were determined with increasing concentrations of inhibitor (128, 32, 8, 2, and 0.5 μ g/mL) versus % of inhibition, in triplicate in two independent experiments. The experimental data were analysed with TableCurve 2D Software (Systat Software, Inc.) and the IC₅₀ values determined by linear regression. It is important to stress the fact that all compounds are soluble in the assay mixtures at the described experimental conditions.

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In silico physicochemical property calculation and druglikeness evaluation

PreADMET online software (<u>https://preadmet.bmdrc.kr/</u>)³⁵ and SwissADME online (<u>http://www.swissadme.ch/</u>)³⁶ prediction tools were applied for determination of physicochemical properties, lipophilicity, water solubility, pharmacokinetics, drug-likeness and medicinal chemistry parameters, and also Topological Polar Surface Area (TPSA), number of rotatable bonds, LogP, violations of Veber's rule³⁸, and violations of Lipinski's rule of five³⁷.

Molecular docking

The two-dimensional structures (.mae) of compounds 6a-g (ligands), were drawn and the structure was analysed by using 2D sketcher and 3D builder of Maestro 11.8 (Schrödinger, LLC, New York, NY, USA) 47. The three-dimensional structures of these compounds (ligands) were generated from threedimensional structures prepared first using LigPrep 3.6 using OPLS-2005 force field. The tautomeric isomers for the ligands, as displayed in Scheme 1, were searched and energy minimizations were carries out by applying the OPLS 2005 force fields, at pH 7.0 ± 2.0. The Epik v.4.3 methodology was used when preparing the ligands. Then geometrically minimized with MacroModel 12.2 followed by conformational analysis using MMFFs force field. Monte Carlo Multiple Minimum (MCMM) conformational search was used with 2500 iterations and convergence threshold of 0.05 kJ/mol. Water was chosen as solvent. Truncated Newton Conjugate Gradient minimization was used with 2500 iterations and convergence threshold of 0.05 kJ/mol. Other parameters were used as default. Crystal structure of human topoisomerase IIB in complex with DNA and mitoxantrone (code: 4G0V) was RCSB Protein retrieved from the Data Bank (https://www.rcsb.org/structure/4G0V). This structure was solved by X-ray crystallography at 2.6 Å resolution. Coordinates of the protein-ligand complex were fixed for errors in atomic representations and optimized using Protein Preparation Wizard Maestro v. 11.5 (Maestro, v. 11.5: Schrödinger, LLC, New York, NY, USA). The bond orders were assigned to residues, hydrogen atoms were added at pH 7.0 \pm 2.0. The restrained minimizations were carried out using the OPLS 2005 force field with an RMSD cut-off value of 0.3 Å for heavy atom convergences. The molecular docking was accomplished and analysed via the Glide v. 8.1 docking tool⁴⁸. The receptor grid was located in the centre based on the active site of the protein, using the receptor grid generation tool. The ligands were flexibly docked in grid box using Monte Carlobased simulation algorithm and an XP (Extra Precision) method without any constraints was employed that generated binding poses based on energy. For preparation of protein, water

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molecules (559 molecules) and 6 magnesium ions were removed. Mitoxantrone molecule complexed with chains A, C, D, F was chosen for Protein Preparation Wizard used OPLS-2005 force field for structural optimization and minimization and for Receptor Grid Generation tool of Glide v.8.1. The Glide HTVS 8.1 algorithm (High-Throughput Virtual Screening Mode) was employed using a grid box volume of $10 \times 10 \times 10$ Å. Briefly, Glide approximates a systematic search of positions, orientations and conformations of the ligand in the receptor binding site using a series of hierarchical filters. Docking progress used constraints of two H-bond donor on Arg503, GLU522 and GLU778, one H-bond acceptor on ASN520. Upon completion of each docking calculation, 32 poses per ligand were generated and the best docked structure was chosen using a GlideScore (Gscore) function. A flow chart for molecular docking calculation using Schrödinger suite was implemented Supplementary Information file in online version of this article.

Conclusions

A series of (*E*)-4-hydroxy-3-(3-(aryl)acryloyl)-1-methylquinolin-2(1*H*)-ones (**5a-g**) were prepared from 3-acetyl-4-hydroxy-1-methylquinolin-2(1*H*)-one with yields of 61–87%. These α , β -unsaturated ketones were converted into 3-(2-amino-6-arylpyrimidin-4-yl)-4-hydroxy-1-methylquinolin-2(1*H*)-ones

(**6a-g**) by reaction with *o*-aminothiophenol in the presence of piperidine as catalyst. The yields of benzothiazepines **6a-g** were 62–85%. All synthesized compounds **6a-g** were evaluated for *in vitro* anticancer activity against two cancer cell lines, human squamous cell carcinoma (KB) and hepatocellular carcinoma (HepG2). Compounds **6c** and **6g** exhibited remarkable inhibitory activity against the tested cancer cell lines with MIC values of 0.25–0.27 and 0.26–0.28 μ M, respectively. ADMET properties as well as drug-likeness behaviours showed that of compounds 6c, 6f, and 6g were evaluated. Docking results indicated that residues GLN778(A), DA12(F), and DG13(F) in the binding pocket as potential ligand binding hot-spot residues for compounds **6c** and **6g**.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Synthesis, cytotoxic activity, ADMET and molecular docking study of quinoline-based hybrid compounds of 1,5benzothiazepines

Duong Ngoc Toan,^{a,b*} Nguyen Dinh Thanh,^{b*} Mai Xuan Truong,^a Duong Nghia Bang,^c Mai Thanh Nga, ^a Nguyen Thi Thu Huong ^b

Some α,β -unsaturated ketones **4a-g** of 3-acetyl-4-hydroxyquinolin-2(1*H*)-one were prepared by its reaction with (hetero)aromatic aldehydes with yields of 61–87% using piperidine as catalyst. These ketones reacted with *o*-aminothiophenol in the presence of acetic acid to afford a series of new hybrid compounds, quinoline-benzothiazepine, **6a-g**. The yields of benzothiazepines **6a-g** were 62–85%. All the synthesized compounds **6a-g** were screened for their *in vitro* anticancer activity against human hepatocellular carcinoma HepG2 and squamous cell *carcinoma* KB cancer lines. Compounds **6d** and **6g** had the best activity in the series, with IC₅₀ values of 0.25 and 0.27 µg/mL, respectively against HepG2, and of 0.26 and 0.28 µM, respectively, against KB cell lines. ADMET properties showed that compounds **6c** and **6g** possessed the drug-likeness behavior. Cross-docking results indicated that residues GLN778(A), DA12(F), and DG13(F) in the binding pocket as potential ligand binding hot-spot residues for compounds **6c** and **6g**.



