

Tetrahedron 54 (1998) 14377-14400

TETRAHEDRON

Hemisynthesis of Some Biogenetically Anomalous 17β-Neoclerodane Diterpenoids

María J. Domínguez, Ekkehard Mössner, María C. de la Torre and Benjamín Rodríguez^{*}

Instituto de Química Orgánica, CSIC, Juan de la Cierva 3, E-28006 Madrid, Spain

Received 13 July 1998; revised 9 September 1998; accepted 24 September 1998

Abstract

The reactivity of several natural neoclerodane diterpenoids (eriocephalin, isoeriocephalin, 7,8didehydroeriocephalin, picropolin and picropolinone) has been investigated. The transformations catalysed by basic reagents (epimerisations at C-8, transacetylations from the C-7 α to C-6 α positions, formation of stable 7 α ,19-hemiacetals and tautomeric mixtures of 7 α ,19-hemiacetal and 19-hydroxy-7-keto forms) were rationalised in each case by the influence of steric and strain effects caused by the functionality and stereochemistry at the C-20 position of the neoclerodane framework. The acid catalysed nucleophilic substitution of a 20-O-acetyl group in 20,12-hemiacetals by a 20-O-methyl group was also studied. Other reactions (α -ketol rearrangements, formation of enol esters and reduction of diosphenol groups) as well as chemical correlations between several neoclerodanes were also carried out, providing useful data for the chemistry of these compounds. Finally, the hemisynthesis of teuvincentin C, starting from 6-O-acetylisoeriocephalin, supports our previous hypothesis on the formation of the biogenetically anomalous 17 β -neoclerodane diterpenoids. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: Biomimetic reactions; epimerisation; steric and strain effects; tautomerism.

Introduction

The neoclerodane diterpenes have attracted interest in the last few years on account of their useful biological activities and challenging structures [1-4].

It is well established [5-7] that the neoclerodanes' are biogenetically derived from *ent*-labdanes through an 8,4-friedo backbone rearrangement, as result of that the C-17 methyl

^{*} Author to whom correspondence should be addressed. Phone 34 91 5622900, Fax 34 91 5644853, e-mail iqor107@fresno.csic.es

¹ Although the hydrocarbon skeleton of these diterpenoids is biogenetically derived from an *ent*-labdane and they should be named *ent*clerodanes, we prefer to use the term neoclerodane proposed by Ley, S. V. and co-workers [8] because it is the nomenclature used in the majority of the articles published on this subject since 1979.



group adopts an 8α -configuration. Consequently, the vast majority of the neoclerodanes isolated from natural sources possess the C-17 methyl group in an α -configuration, and among the few exceptions reported with the opposite stereochemistry [9-12] we have recently amended [13] the structures of crotocaudin and isocrotocaudin, for which a 17 β neoclerodane framework had been claimed [10,11]. Moreover, we have suggested [12] that the well established 17 β -neoclerodane structures of teuvincentin B and teuvincentin C (1 and 2, respectively, see Figure 1) [12] could arise from the

biogenetically correct 17α -neoclerodane precursors by an epimerisation at C-8, as a consequence of the acid character of the 8 β -hydrogen when C-7 bears a carbonyl function.

In this paper, we describe the preparation of some 17β -neoclerodane derivatives, including teuvincentin C (2), starting from suitable 7-oxo-neoclerodanes, and the results of some transformations designed to provide information concerning the driving forces that cause the epimerisation at the C-8 asymmetric centre in these compounds.

Results and discussion

Among the available natural neoclerodanes, isoeriocephalin (3)[14] seemed to be a suitable compound for obtaining 17 β -neoclerodane derivatives by epimerisation at C-8. However, when 3 (Figure 2) and its acetates 4 and 5 (Scheme 1)[14] were treated with some basic reagents (Na₂CO₃, 'BuOK and MeONa), very complex mixtures of products were always obtained, probably due to side reactions caused by the lability of the 20-O-acetyl 20,12-lactol of these compounds, which is unstable even under very mild acidic or basic conditions [14-16].

Figure 2



Oxidation of 4 with Jones' reagent (see Scheme 1) gave a compound (6, 89% yield) identical to the peracetyl derivative of picropolin (7, Figure 2)[17-20], a neoclerodane diterpene isolated for the first time from *Teucrium polium* [17]. The formation of 6 from 4

can be rationalised by an initial acid catalysed hydrolysis of the 20-O-acetyl group [14,15] and subsequent oxidation of the 20,12-hemiacetal. As previously described for picropolin (7, Figure 2)[17], treatment of 6 with sodium carbonate in methanol solution yielded the 7α ,19hemiacetal 8 (85%, Scheme 1), which was identical to deacetylisopicropolin prepared by Brieskorn and Pfeuffer from 7 [17]. Compound 8 was also obtained in 88% yield by basic treatment of 6-O-acetylpicropolin (9)[17], which in turn was now prepared by oxidation of 5 with Jones' reagent (93% yield, Scheme 1). NOE experiments showed that 8 possessed its C-17 methyl group in an α -configuration, a structural feature not rigorously established in the previous work [17]. As shown in Table 1, the H-6 β proton of 8 showed NOE correlations with the H-8 β , H-10 β and H_B-18 protons, as well as with both the C-6 α and C-7 β hydroxyl protons, thus establishing that H-6 β , H-8 β and H-10 β protons are on the same side of the



i: Jones' reagent. ii: Na2CO3, MeOH, reflux (2 h).

plane defined by the decalin part and, consequently, that the Me-17 group is α -oriented. Thus epimerisation at C-8 did not occur in the transformation of 6 and 9 into 8.

Since epimerisation at C-8 of 7-oxo-neoclerodan-20,12-olide derivatives (6 and 9) was unsuccessful, we supposed that the 20S-O-acetyl substituent (20 β -position in the formula) of 1 and 2, which is spatially close to the Me-17 group, could be a decisive steric factor for

promoting the epimerisation, but the 20-O-acetyl 20,12-lactol grouping is an extremely labile function which causes undesirable side reactions (see above). For this reason, we decided to transform the 20,12-(20-O-acetyl)hemiacetal of 4 into the more stable 20,12-(20-O-methyl)acetal. It is known [12,16] that the C-20 acetoxyl group of compounds such as 4 is easily and stereoselectively substituted by alkoxide or acyloxy groups under acid catalysis, probably via an oxonium ion [16]. Treatment of a methanolic solution of 4 with catalytic amounts of toluene-4-sulfonic acid at 50 °C for 2 hours gave minor quantities of 10 (4% yield, Scheme 2) together with starting material (4, 70% recovered) and several decomposition products. Attempts at improving the yield of 10 by this procedure were unsuccessful. Total transformation of 4 into 10 (80% yield) and 11 (7.6%, Scheme 2) was achieved, however, when the reaction was carried out in a mixture of toluene and methanol (10:1) and traces of toluene-4-sulfonic acid under microwave irradiation for 4 minutes (see Experimental). When the microwave irradiation was continued for a further 16 minutes only

Table 1											
Significant	NOE	data	for	com	pounds	8-10,	12,	25-29,	32, 3	5 and	i 36'

Compound	Observed	Observed NOE cross peaks with proton(s)	Config	uration
	proton(s)		<u>C-8</u>	C-20
8	Η-6β	6α-OH, 7β-OH, H-8β, H-10β, H _B -18	8α-Me	-
9	Η-6β	H-8β, H-10β, H _B -18	8α-Me	-
10	Me-17	20-OMe (no NOE with H-20)	-	R
	H-20	H _A -19, H _B -19 (no NOE with Me-17)	-	R
12	Η-6β	6α-OH, 7β-OH, H-10β, Me-17	8β-Me	
	Me-17	H-6β, H-10β (no NOE with H-20)	8β-Me	R
	H-20	H-1 α , H-1 β , H _A -19 (no NOE with Me-17)	•	R
25	Η-8β	Η-7β, Η-10β	8α-Me	
	Me-17	H-20		S
26	Η-1β	H-12, H-20		R
	Η-8β	Η-7β, Η-10β	8α-Me	
	H-20	$H-1\alpha$, $H_{A}-19$, $H_{B}-19$		R
27	Η-8β	Н-6β, Н-10β	8α-Me	
	Η-10β	Η-2β, Η-8β	8α-Me	
	Me-17	H-20		S
	H-20	Me-17		S
28	Η-6β	H-8β, H-10β, H _B -18	8α-Me	
	Η-8β	Η-6β, Η-10β	8α-Me	
	Me-17	20-OMe (no NOE with H-20)		R
	H-20	H-1 α , H-1 β , H _A -19, H _B -19 (no NOE with Me-17)		R
29	H-20	Me-17	-	S
32	Η-6β	6α -OH, H-10 β , Me-17, H _B -18	8 B -Me	
	Η-8α	H_{A} -19, Me-17 (no NOE with H-6 β and H-10 β)	8В-Ме	
	Me-17	H-6 β (no NOE with H-20)	8β-Me	R
	H-20	$H-1\alpha$, $H_{A}-19$	P	R
35	Η-6β	H-7β, H-8β, H-10β, H _P -18	8α-Me	
	Me-17	H-20		S
36	Η-6β	H-8β, H-10β, H _n -18	8α-Me	2
	Me-17	H-20		S
	H-20	Me-17, H _a -19		S

"All these data were obtained from the NOESY spectra and, in some cases (9, 10, 12, 25-28, 32 and 36), also by 1D NOE experiments.

compound 11 was obtained in almost quantitative yield. As was to be expected [12,16], this reaction gave 10 through a nucleophilic substitution of the C-20 acetoxyl group by a methoxy group (10 and 11) and, in the case of 11, caused an additional nucleophilic opening of the 4α , 18-oxirane [21]. The 20R stereochemistry of 10 (and therefore of 11) was firmly



i: Toluene + MeOH (10:1), p-TsOH, μ w (4 min). ii: Toluene + MeOH (10:1), p-TsOH, μ w (20 min). iii: Na₂CO₃. MeOH, reflux (3 h).

supported by NOE experiments, because irradiation at the Me-17 protons (δ 1.63) caused NOE enhancement in the signal of the 20-methoxy group (δ 3.38) and not in that of the H-20 proton, whereas irradiation at δ 5.02 (H-20 proton) produced an NOE at both the C-19 methylene protons (δ 4.60 and 4.58, see Table 1). These results established [22] that the Me-17 and 20-methoxy groups are on the same side of the plane defined by the 20,12-acetal. Therefore, the formation of **10** from **4** occurred by a stereoselective attack of the nucleophile (MeO) from the less hindered *si* face of the 20-oxonium ion [16].

Treatment of 10 (Scheme 2) with sodium carbonate in methanol solution gave the 7α , 19-hemiacetal 12 (86% yield) and minor quantities of 13 (7.7%), whereas 11 yielded 13 (77%) by the same reaction. The C-17 methyl group of 12 and 13 is β -oriented, as was

evidenced by their spectroscopic data [12,23](see Experimental) and particularly by the NOESY spectrum of 12 (Table 1), which showed NOE correlations between the C-17 methyl group and the axial H-6 β and H-10 β protons.

The different behaviour of compounds 6 and 10 (Schemes 1 and 2) under basic treatment must be attributed to the presence in 10 of a bulky 20*R*-methoxy substituent, which favours the epimerisation at C-8 via an enolate anion intermediate, producing the thermodynamically more stable 17β -neoclerodane derivative 12, in which the steric

Scheme 3





i: NaBH₄, MeOH-dioxane (3:1).

interactions between the Me-17 and 20*R*-methoxy groups are minimised.

The influence of the functionality at C-20 the reactivity of these neoclerodane on diterpenoids was also corroborated by other chemical transformations, such as the reduction of 7-enol acetates of 6,7-dioxo compounds. When compound 14 (Scheme 3) {obtained for work this by acetylation of 6.7didehydroeriocephalin (15, Figure 2)[24]} was treated with sodium borohydride it yielded 16 (90%), a compound previously known as a derivative of isoeriocephalin (3)[14]. This reaction caused, apart from a stereoselective reduction of the diosphenol grouping from the less hindered β face, a transacetylation reaction from the C-7 α to the C-6 α positions. This transacetylation can be attributed to the existence of strong steric interactions between the 20S-acetoxyl group (20 β position in the formula) and the C-7 α axial substituent. On the contrary, identical treatment of the 20,12lactone derivative 18 (Scheme 3) {prepared for this work from picropolinone (19, Figure 2[18-20]} gave 20 (quantitative yield), an

already known compound [25]. In this case, the O-acetyl group was maintained at the C-7 α position, probably due to the absence of C-20 β - C-7 α steric interactions.²

In order to provide more data on the above mentioned transformations, we decided to study the behaviour of other neoclerodane diterpenoids possessing a different arrangement of the functionalities in ring B, like eriocephalin (22, Scheme 4)[14,27]. Reaction of 22 (Scheme 4) with methanol and traces of toluene-4-sulfonic acid under microwave irradiation quantitatively yielded a mixture of two compounds (23 and 24, 2:1 ratio). Attempts at

² It is of interest to indicate that 14 was transformed into 5 (86% yield) by NaBH₄-CeCl₃.7H₂O reduction, and 5 was correlated with 7-O-acetyleriocephalin (17, Figure 2)[14] by a thermal α -ketol rearrangement [26] of this last compound. Moreover, reduction of 18 with NaBH₄-H₃BO₃ gave 9 (84% yield), a substance previously obtained from capitatin (21, Figure 2)[19,20] by thermal rearrangement [26]. (See Experimental).

isolating these substances by chromatography were unsuccessful and only minor quantities of the less polar constituent (23) were isolated (see Experimental). Acetic anhydride-pyridine treatment of the mixture 23 and 24 gave the corresponding acetyl derivatives 25 and 26 (2:1 ratio), which were easily separated by column chromatography. Compounds 25 and 26 are epimers at the C-20 chiral centre and the NOE data shown in Table 1 rigorously established a 20S stereochemistry for 25 and a 20R configuration for 26.

The drastic change observed in the stereoselectivity of the reaction of eriocephalin (22) with methanol (Scheme 4), as compared with the result achieved with peracetyl isoeriocephalin (4, see above and Scheme 2), may be attributed to the distorted boat conformation (7,10 B) of ring B in 22 [27], in which the *si* face of the 20-oxonium ion intermediate [16] is less accessible to the nucleophile attack than in 4, thus favouring the approach of the nucleophile from the *re* face to give predominantly the 20S derivative 23.



i: Toluene-MeOH (5:1), p-TsOH, μ w (6 min). ii: Silica gel, EtOAc, r.t. (55 h). iii: Ac₂O-pyridine (1:1), r.t. (48 h). The mixture of unstable compounds 23 and 24 was almost quantitatively transformed into the corresponding stable regioisomers 27 and 28 by a silica gel catalysed α -ketol rearrangement (Scheme 4).³ Acetylation of 27 yielded 30 (90%, Scheme 4), whereas in the case of 28 the same reaction gave the 6-O-acetyl derivative 31 (78% yield) and minor quantities of 10 (9.5%, Scheme 4). The different reactivity of 27 and 28 under acetylation conditions may be attributed to their opposite stereochemistry at C-20, because the formation of a 7-enol acetate (10) from 28 minimises the 20*R*-methoxy - Me-17 steric interactions.



i: Na₂CO₃, MeOH, reflux (3 h). ii: DABCO, toluene, 120 °C (6-7 h), under Ar.

³ It is noteworthy that reaction of isoeriocephalin (3) with MeOH and traces of *p*-TsOH under microwave irradiation yielded minor quantities of 27 (10%) and 28 (6%) together with several decomposition products (see Experimental). This may be due to the different conformation of ring B in 3 and 22 [14,27].

In agreement with the behaviour of 10 and 11 under basic treatment (see above, Scheme 2), reaction of 28 (or its 6-O-acetyl derivative 31) with sodium carbonate in methanol solution also yielded the 17 β -neoclerodane derivatives 12 and 13 (93% overall yield, Scheme 5). Similarly, treatment of 28 (Scheme 5) with 1,4-diazabicyclo[2.2.2]octane (DABCO) gave, together with starting material (52% recovered), the C-8 epimer 32 (43% yield) whose 17 β -neoclerodane structure was firmly supported by its NOESY spectrum (Table 1). Additionally, when the 7-oxo-neoclerodan-20,12-olide derivatives picropolin (7) and its 6-acetate (9), as well as the 20S-methoxy derivative 27, were treated with DABCO in the same conditions as those for 28 no reaction was observed.⁴ Finally, reaction of 6-O-acetylisoeriocephalin (5, Scheme 5) with DABCO gave teuvincentin C (2, 76% yield), one of the biogenetically anomalous 17 β -neoclerodane diterpenoids isolated from *Teucrium polium* subsp. *vincentinum* [12]. In the transformation of 5 into 2, the use of DABCO as a basic catalyst prevents side reactions of the 20-o-acetyl 20,12-lactol grouping.

Unfortunately, it was not possible to synthesise teuvincentin B (1) or its 6-O-acetyl derivative [12] because suitable natural substrates are not available [1,2] and attempts at removing selectively the C-19 acetyl group in compounds such as 3, 5 and 15 were unsuccessful. It is highly probable, however, that treatment of the unknown 19-deacetyl derivative of isoeriocephalin (3) with DABCO will produce 1. In any case, this natural diterpenoid and the derivative 12 (see above, Schemes 2 and 5) are structurally very close.

We next decided to investigate the reaction of the 20S-methoxy derivative 27 with a basic reagent different from DABCO. In principle, it may be expected that the steric interactions between the 20S-methoxy and the C-1 methylene groups in 27 could be at least of the same magnitude than those shown between the 20R-methoxy and Me-17 groups of 28. For this reason, we supposed that, apart from the epimerisation at C-8 in 28 and not in 27 (see above), the behaviour of these compounds under reaction with sodium carbonate in methanol solution should be slightly different. Treatment of 27 with that reagent (Scheme 6) yielded an apparently single product, but its 'H NMR spectrum revealed that it was a mixture of the tautomers 33 and 34 in 1.3:1 ratio (see Experimental). Reduction of this mixture with sodium borohydride gave 35 (86% yield), for which the NOESY spectrum (Table 1) strongly supported that, as was to be expected, no epimerisation at C-8 occurred in the initial basic treatment of 27. Acetylation of the tautomeric mixture (33 and 34, Scheme 6) yielded three compounds: 30 (46%, see above and Scheme 4), 36 (16%) and another mixture of tautomers (37 and 38, 32%). Obviously, compound 30 originated from the 19-hydroxy-7-keto form (33) and 36 from the 7α , 19-hemiacetal form (34), whereas the tautomers 37 and 38 were still found among the reaction products because of the existence of the hemiacetal form and the difficult acetylation of its tertiary hydroxyl group.⁵ The NOESY spectrum of 36 (Table 1), as well as other spectroscopic data (see Experimental), rigorously supported the structure and relative stereochemistry assigned to this compound.

It is of interest to indicate that crystallisation of the mixture of tautomers 37 and 38 from EtOAc - *n*-hexane slowly yielded the 7α , 19-hemiacetalic form 37, melting point 145-

⁵ For the conditions of acetylation of the tautomers 33 and 34, see Experimental.

⁴ When the reaction of **27** with DABCO was carried out without an Ar atmosphere minor quantities of **29** (6.3% yield. Figure 2) were obtained, probably by air oxidation or by a process in which N-oxide of DABCO takes part (see Experimental).



i: Na₂CO₃, MeOH, r.t. (4 h). ii: NaBH₄, MeOH, r.t. (45 min). iii: Ac₂O-pyridine (1:1), r.t. (10 days).

147 °C. The identification of the crystalline form as **37** was supported by the following results. i) The IR spectrum of **37** (in KBr disk) showed only one carbonyl absorption at 1735 cm⁻¹, assigned to the 6 α -acetoxyl group. ii) The ¹H NMR spectrum of **37** revealed that it was in equilibrium with **38** in 1:1.3 ratio (CDCl₃ solution at 20 °C) and this ratio changed by increasing the temperature: 1:1.5 ratio at 35 °C and 1:1.6 ratio at 45 °C (see Experimental). iii) Compound **37** showed mutarrotation from $[\alpha]_D^{19}$ -0.8° (1 minute after preparing its solution in CHCl₃) to $[\alpha]_D^{19}$ +22.1° (after 15 minutes). This mutarrotation from a negative value (major tautomer **37**) to a positive one (major tautomer **38**, see the ¹H NMR data quoted above) is in agreement with the fact that stable 7 α ,19-hemiacetal derivatives, such as **12** and **36**, possess negative specific rotations (-17.5° and -52.4°, respectively), whereas 7-oxoneoclerodanes like **27** and **30** show positive values (+67.5° and +72.9°, respectively).

In contrast to the 7α , 19-hemiacetal derivatives **8**, **12** and **13** (Schemes 1 and 2), which are stable substances in normal conditions, compounds **34** and **37** (Scheme 6) are in a tautomeric equilibrium with their 19-hydroxy-7-keto forms (**33** and **38**, respectively). The instability of these compounds must be attributed to strong steric interactions between the 20S-methoxy and the C-1 methylene groups, together with strain effects in the decalin moiety. In compound **36**, where the tautomerism is blocked by acetylation of the hemiacetal hydroxyl group, ring A possesses a conformation very different from those of the stable 7α , 19hemiacetals **8** and **12**, as was revealed by several ¹H and ¹³C NMR spectroscopic data (see Experimental, Tables 2, 3 and 5). In particular, the proton-proton coupling values $J_{2\alpha,3\beta}$ = 1.0 Hz, $J_{2\beta,3\beta}$ = 8.4 Hz and $J_{3\alpha,18B}$ = 0 Hz observed for **36**, as compared with those of **8** (3.3, 3.5 and 2.2 Hz, respectively) and **12** (3.3, 3.3 and 2.2 Hz, respectively), evidenced that the conformation of ring A in **36** is different from the ¹⁰C₃ chair conformation of **8** and **12**.

In summary, we have demonstrated the influence of the functionality and stereochemistry at C-20 on the reactivity of several 19-acetoxy-7-oxo-neoclerodane derivatives when they were subjected to treatment with bases. All the differences observed in epimerisation reactions at C-8, transesterification from the C-7 α to the C-6 α positions and in the formation and stability of 7 α ,19-hemiacetals may be attributed predominantly to steric and strain effects. Moreover, we have improved a method for obtaining neoclerodane-20,12-(20-*O*-methyl)acetals by a nucleophilic substitution reaction on 20,12-(20-*O*-acetyl) hemiacetals, justifiying the stereoselectivity of the reaction by the influence of steric hindrance factors. Finally, we have obtained teuvincentin C (2) starting from 6-*O*-acetylisoeriocephalin (5), thus demonstrating the validity of our hypothesis [12] on the formation of the biogenetically anomalous 17 β -neoclerodane diterpenoids.

Experimental section

General experimental procedures.

Mps were determined on a Kofler-type block and are uncorrected. Optical rotations were measured on a Perkin-Elmer 241 MC polarimeter. IR spectra (KBr disk) were obtained on a Perkin-Elmer 681 spectrophotometer. ¹H NMR spectra were recorded using a Bruker AM200, Varian INOVA-300 or Varian Unity-500 instrument at 200, 300 or 500 MHz, in CDCl₃ solution, and chemical shifts are reported with respect to residual CHCl₃ (δ 7.25). ¹³C NMR spectra were recorded at 50.3 or 125.7 MHz in CDCl₃ and chemical shifts are reported with respect to solvent signals (δ_{CDCl_3} 77.00). ¹³C NMR assignments were determined by DEPT pulse sequences, HMQC and, in some cases, by HMBC spectra. MS were recorded in the IE mode on a VG 12-250 instrument (70 eV, direct inlet). Elemental analyses were made with a Carlo Erba EA1108 apparatus. The purity of the compounds was checked by TLC on precoated plates (Merck, Si gel 60F₂₅₄). Merck Si gel No. 7734 (70-230 mesh) deactivated with 10% H₂O, w/v, was used for column chromatography.

Starting materials, isoeriocephalin (3)[14], 7,8-didehydroeriocephalin (15)[24], picropolinone (19)[18-20] and eriocephalin (22)[14,27], were available from previous works.

Peracetylisoeriocephalin (4) and 6-O-acetylisoeriocephalin (5) were obtained as described previously [14].

Preparation of peracetylpicropolin (6) from peracetylisoeriocephalin (4).

A solution of 4 (200 mg, 0.366 mmol)[14] in Me₂CO (100 mL) at 0 °C was treated with Jones' reagent (0.3 mL) for 20 min with stirring. Then EtOH (10 mL) was added and, after a further 10 min, the reaction mixture was diluted with H₂O (200 mL) and extracted with CHCl₃ (4 x 50 mL). The organic extract was dried over Na₂SO₄, filtered and evaporated to dryness giving a residue from which **6** (colourless crystalline solid, 165 mg, 0.328 mmol, 89% yield) was obtained by crystallisation from MeOH. Compound **6** was identical [17] with the peracetyl derivative of picropolin (**7**)[17-20]. Since all the physical and spectroscopic data of **6** have not previously been reported [17], we include here some additional data obtained by us for this compound: mp 204-205 °C (lit. [17] mp 204 °C); $[\alpha]_D^{17}$ +58.9° (*c* 0.112, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3150, 1505, 870 (furan), 3045 (oxirane), 1765 (γ -lactone), 1755, 1740, 1250 (OAc), 2960, 2870, 1450, 1370, 1170, 1090, 1040, 1015, 915, 800. 'H NMR: see Table 2. EIMS *m/z* (rel. int.): 502 [M]⁺ (1.5), 442 (1), 340 (4), 328 (4), 312 (9), 246 (8), 218 (10), 189 (7), 159 (8), 145 (8), 135 (8), 122 (16), 94 (64), 91 (9), 81 (17), 55 (7), 43 (100).

Preparation of deacetylisopicropolin (8) from peracetylpicropolin (6).

To a solution of **6** (100 mg, 0.199 mmol) in MeOH (50 mL) was added Na₂CO₃ (50 mg, 0.471 mmol) and the reaction mixture was refluxed for 2 h. The reaction was allowed to reach room temperature and then acidified to pH 5 with 1N H₂SO₄. Extraction with CHCl₃ (3 x 25 mL) and work-up in the usual manner yielded **8** (colourless crystalline solid, 64 mg, 0.170 mmol, 85% after crystallisation from EtOAc - *n*-hexane). Compound **8** has previously been obtained from picropolin (**7**)[17] and only some physical and spectroscopic data have been reported for it. We have now obtained the following data: mp 226-229 °C (lit. [17] mp 202-205 °C, from H₂O); $[\alpha]_D^{24}$ +5.4° (*c* 0.11, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3480, 3380 (OH), 3150, 3110, 1500, 875 (furan), 1765 (γ -lactone), 2980, 2940, 2920, 2880, 1450, 1400, 1315, 1240, 1170, 1160, 1125, 1035, 1020, 995, 980, 925, 800. ¹H NMR: see Table 2. ¹³C NMR: see Table 3. (Anal. Found: C, 63.89; H, 6.39%. C₂₀H₂₄O₇ requires: C, 63.82; H, 6.43%).

Preparation of 6-O-acetylpicropolin (9) from 6-O-acetylisoeriocephalin (5) and its transformation into deacetylisopicropolin (8).

Treatment of 5 (200 mg, 0.396 mmol)[14] with Jones' reagent as described above for 4 gave 9 (colourless crystalline solid, 171 mg, 0.371 mmol, 93% yield): mp 224-226 °C (EtOAc - *n*-hexane); $[\alpha]_D^{21}$ +19.3° (*c* 0.238, CHCl₃), identical to the previously described compound [17]: mp 221-224 °C (from CHCl₃-Et₂O); $[\alpha]_D^{20}$ +17.5° (*c* 0.5, CHCl₃); for the ¹H NMR spectrum see Table 2.

Compound 9 (120 mg, 0.260 mmol) was transformed into 8 (87 mg, 0.231 mmol, 88% yield) by treatment with Na_2CO_3 as previously described for 5.

Preparation of $(12S,20R)-6\alpha,7,19$ -triacetoxy-4 $\alpha,18$;15,16-diepoxy-neocleroda-7,13(16),14-triene-20,12-(20-O-methyl)acetal (10) and $(12S,20R)-6\alpha,7,19$ -triacetoxy-15,16-epoxy-4 α -

H NMR	pectral data	for compound:	s 6 and 8-13"												
н	9	20	9	10	11	12	13	J _{III} (Hz)	6	8	6	10	=	12	<u>-</u>
10	4	4	1.94 dddd	4	ti.	4	1	1α,1β	"	"	13.4	4		4	
8	4	4	þ	h	4	<i>יי</i>	4	1α.2α	4	3.4	4.3	÷	-	3.3	-
20	4	1.99 bbbb 1	h	4	"	Dubbb 69.1	"	1α.2β	4	13.2	12.9	4	÷	13.1	4
2B	4	1.49 at	1.64 at	ł,	h	l.41 m	"	1α,10B	4	12.1	12.8	4	4	4	4
ļ	4	· F · · · · · · · · · · · · · · · · · ·	2.22 dddd	\$	4	1.71 dddd	h	1 β. 2α	4	3.4	4	4	4	3.3	÷
8	4	1.13 dt	1.11 dold	h	4	1.12 ddd	4	18.28	4	3.9	4.8	4	4	4	4
4 9	5 42 hr a	3.36 hr s	5.17 d	5.38 hr a	5.63 br a	3.68 ď	4.09 br s	18,108	4	4.4	3.4	4	÷	4	4
2 S			,			3.11 g	3.24 q	20,2B	4	13.2	12.9	4	4	12.9	4
e esta	,	1910	2 56 ad	,	Ţ	-	- '	2α.3α	4	3.4	4.5	4	4	3.6	4
4 <u>8</u>	4	1.92 dd	2.38 dd	h	4	\$	<i>n</i>	20,3B	4	3.3	3.8	4	4	3.3	4
ALI	2.48 dd	2.43 dd	2.49 dd	p	"	bb 16.1	bb 16.1	2B.30	4	13.2	13.6	4	4	13.2	4
IB	2.68 dd	2.58 dd	2.67 dd	2.44 dd	2.47 dd	2.22 dd	2.21 dd	28.38	4	3.5	4.8	4	4	3.3	9
12	5.50 t	5.33 br t	5.44 br t	4.91 t	4.95 (5.01 br t	5.04 br t	3α,3β	4	12.9	13.6	4	ų	12.8	4
14	6.36 dd	6.39 dd	6.36 dd	6.39 t	6.40 t	6.41 dd	6.44 dd	6 β.8 β		0	0.9	•			,
5	7.44 t	7.43 (7.43 (7.37"	7.37"	7.36 t	7.37 t	6B.17	1.9	0	0	8.1	6.1	0	0
16	7.45 m	7.45 m	7.45 m	7.37	7.37^{h}	7.34 m	7.39 m	8a,17	•	ı.	•		,	7.3	7.3
Me-17	1.51 d	1.17 d	1.13 d	1.63 d	l.64 d	1.00 d	1.07 d	8B.17		7.1	6.6	•	,		ı
18A"	2.30 d	2.74 d	2.36 d	2.28 d	3.46 br d	2.72 d	3.73 d	IIA, IIB	14.3	14.0	14.4	13.9	13.9	13.5	13.4
18B'	3.00 dd	3.22 dd	2.92 dd	3.05 dd	3.58 d	3.22 dd	3.92 d	11A,12	8.5	9.3	9.0	4	4	8.1	8.3
A91	4.76 br d	4.14 br d	4.49 br d	4.58 br d	4.63 br d	4.14 d	4.09 d	118,12	8.5	8.0	8.1	8.6	8.3	8.4	8.4
19B	4.84 d	4.56 d	4.42 d	4.60 d	4.98 d	4.22 d	4.80 d	14,15	8.1	1.7	1.8	1.3	1.4	1.8	1.7
20			ı	5.02 s	5.11 s	4.91 s	5.04 s	14,16	0.8	0.9	0.9	1.3	1.4	0.8	0.8
6a-OH		4.68 br s	,	,	Į	3.93 d	s	15,16	1.8	1.7	1.8	4	4	1.8	1.7
78-OH	,	3.90 s	,	i	ı	4.49 s	×	18A,18B	3.8	3.7	3.8	3.6	9.4	3.6	9.9
OAc	2.07 s	,	2.10 s	2.08 s	2.10 s	·	4	18B,3a	2.5	2.2	2.5	2.2	0	2.2	0
	2.05 s	,	2.04 s	2.00 s	2.03 s		,	19A, 19B	12.8	8.3	12.9	11.7	12.3	8.3	8.4
	2.05 s	,	ı	2.00 s	2.02 s	ı	ı	19A,6ß	<0.3	<0.3	<0.3	<0.3	<0.3	0	0
18-OMe	,	,	,	ţ	3.32 s	1	3.41 s	6B,6-OH'	ŗ	0	ı			4.0	0
20-OMe	,	•	-	3.38 s	3.32 s	3.35 s	3.35 s								
"At 200 N	1Hz (6, 10, 1	1 and 13) or 5(00 MHz (8, 9	and 12) in CL	Cl, solution, ¢	except for 13 [CDCI ₃ -pyridi	ne-d, (1:1)]. Cl	vemical shi	ifts (ð va	lues) are	c report	ed with re	espect to	the
signal of I	esidual CHC	l ₃ (δ 7.25).													
"Overlapp	ed signal.														

Table 2

"Collapsed into a singlet after addition of D_2O . "*Exo* hydrogen with respect to ring B, except for 11 and 13. "*Endo* hydrogen with respect to ring B, except for 11 and 13. "Disappeared after addition of D_2O ."

14389

C NMR	Spectral	data of cu	mpounds	8, 10, 12	-14, 18, 23	9, 25-32. 3	55 and 36										
C	×	10	12	13	14	18	23'	25	26	27	28	29	30	31	32	35	36
-	25.61	23.4 (24.3 t	22.91	22.1.1	23.3 (23.21	23.5 1	22.6 t	23.7 1	23.01	23.4 1	23.8 (22.9 (22.4 1	23.2 1	20.8 1
7	24.3 1	24.7 1	24.9 1	21.11	24.8 1	24.4 1	25.9 (25.9 (25.7 1	25.5 1	25.0 t	25.7 1	25.4 (24.8 (25.0 (26.1 t	21.7 (
er.,	32.11	32.8 1	32.51	30.61	31.8.0	32.01	31.11	31.5 (30.6 (32.71	32.2 (32.8 (32.8 1	32.11	31.71	32.11	26.1 (
4	63.8 s	63.5 s	63.7 s	73.7 s	61.1 s	60.3 s	s	60.4 s	60.7 s	65.6 s	65.1 s	60.8 s	64.5 s	64.1 s	65.4 s	67.3 s	58.1 s
S	50.2 s	43.9 s	51.1 s	52.8 s	50.5 s	50.3 s	s	54.4 s	54.3 s	50.4 s	49.4 s	49.6 s	49.9 s	48.9 s	49.2 s	45.7 s	52.8 s ^c
6	78.7 d	69.5 d	72.4 d	70.6 d	185.4 s	186.8 s	s	200.2 s	200.1 s	78.0 d	78.7 d	190.0 s	75.9 d	76.6 d	72.8 d	76.6 d	78.8 d
7	104.3 s	138.0 s	107.8 s	106.7 s	143.8 s'	142.5 s'	69.6 d	75.3 d	77.1 d	206.8 s	205.4 s	133.4 s	201.9 s	200.8 s	209.6 s	74.2 d	111.5 s
90	46.9 d	128.4 s	41.0 d	41.3 d	145.1 s ^c	139.2 s'	44.9 d [°]	42.7 d	40.9 d	52.5 d	51.1 d	124.5 s	52.9 d	51.6 d	46.9 d	44.6 d	48.5 d
6	52.6 s	54.0 s	55.9 s	54.9 s	55.0 s	54.1 s	s	51.1 s	53.6 s	57.6 s	56.0 s	54.6 s	56.8 s	56.0 s	56.9 s	52.3 s	53.7 s ^c
01	51.8 d	48.5 d	47.9 d	44.5 d	50.3 d	49.5 d	44.8 ď	45.2 d	50.7 d	50.2 d	50.6 d	49.0 d	50.6 d	51.2 d	45.4 d	51.6 d	51.1 d
=	48.3 t	45.1 t	44.5 (43.7 t	42.8 t	43.8 t	46.7 1	46.7 t	49.3 I	45.91	46.4 t	44.2 t	45.5 t	45.8 t	44.4 1	45.8 t	46.8 t
12	71.1 d	71.1 d	72.0 d	72.0 d	74.7 d	72.6 d	73.0 d	69.6 d	71.6 d	70.4 d	68.7 d	71.5 d	70.4 d	69.5 d	72.4 d	69.2 d	P 0.69
13	124.8 s	126.6 s	129.0 s	128.5 s	127.2 s	124.4 s	s	127.7 s	127.9 s	124.7 s	125.5 s	124.6 s	124.7 s	126.1 s	128.3 s	125.3 s	125.2 s
14	108.0 d	108.7 d	109.4 d	108.3 d	108.4 d	107.7 d	108.3 d	108.5 d	109.2 d	108.6 d	108.7 d	108.6 d	108.6 d	108.7 d	109.1 d	108.8 d	108.7 d
15	144.2 d	143.5 d	143.5 d	142.1 d	143.7 d	144.5 d	143.5 d	143.6 d	143.5 d	143.5 d	143.6 d	143.6 d	143.4 d	143.6 d	143.7 d	I43.2 d	i43.4 d
16	139.6 d	139.3 d	139.6 d	138.2 d	139.0 d	139.6 d	138. I d	138. I d	139.8 d	139.5 d	139.6 d	139.5 d	139.3 d	139.4 d	139.8 d	139.4 d	139.5 d
17	9.2 q	14.8 q	15.5 q	14. l q	17.1 q	14.5 q	8.4 q	10.0 q	12.4 q	9.7 q	10.4 g	14.2 q	9.7 q	10.5 q	18.2 q	14.3 q	8.9 q
18	53.2 t	51.61	53.2 t	74.6 t	50.8 t	48.2 t	50.0 t	50.2 (52.4 1	51.41	49.9 t	50.8 t	51.51	50.01	51.5 t	49.3 t	48.0 t
61	66.0 t	63.2 t	66.6 t	64.5 t	63.2 (62.8 t	62.2 t	62.5 t	62.4 t	62.0 t	62.2 t	64.0 t	61.5 t	61.8 t	61.3 t	63.61	67.5 1
20	173.9 s	106.6 d	106.6 d	105.6 d	98.9 d	174.4 s	106.9 d	106.7 d	106.3 d	106.8 d	106.0 d	110.6 d	106.9 d	106.0 d	104.7 d	107.4 d	106.8 d
OAc		170.3 s	,	ı	169.9 s	170.5 s	s	170.3 s	170.5 s	169.9 s	170.5 s	170.2 s	169.9 s	170.2 s	169.7 s	•	169.4 s
		170.2 s	•	4	169.1 s	168.0 s	20.8 q	l 69.9 s	170.2 s	20.7 q	21.0 q	20.8 q	169.8 s	169.9 s	20.3 q	,	167.9 s
		167.9 s	ı		168.3 s	20.8 q	•	20.9 q	21.0 q	,	,	'	20.4 q	20.7 q			20.8 q
	ı	21.1 q	•		20.9 q	20.0 q	ı	20.5 q	20.5 q			,	20.4 q	20.5 q	ı	ı	20.7 q
	,	20.6 q	·	ı	20.4 q	•	,	,		,	•	ł	ı	ı	ı	•	
		20.2 q		,	20.1 q	,	۱	•	,		•	•	,	,	•		•
18-OMe	·	•		58.1 q	•		,	,	•	ı	ı	·	·	•	•	•	•
20-OMe	•	57.0 g	54.9 q	53.3 q	-	, i	53.7 q	53.8 q	54.3 q	54.0 q	57.3 9	54.6 q	53.8 9	56.5 q	54.69	54.39	54.19
*At 125.7	MHz (8,	12, 27, 28	3 and 32)	or 50.3 M	Hz (10, 1:	3, 14, 18,	23, 25, 26	, 29-31, 3	5 and 36)	in CDCI,	solution,	except fo	r 13 [CD	Cl ₃ -pyridi	ne-d ₅ (1:1)). Chemi	cal shifts
(S values)	are renor	ted with re	spect to th	e solvent	signals (§ 7	77.00). All	these assi	gnments v	vere in agr	cement with	th HMOC	and DEP	T spectra	and, in son	ne cases (1	8, 12, 27, 2	8 and
12) alen v	with HMB	AC shectra			þ			2)		,						

32), also with HMBC spectra. ⁴For this compound only protonated carbons were measured by a DEPT pulse sequence experiment (see Discussion of results). ⁴Assignments within the same column may be interchanged, but those given here are considered to be the most likely.

Table 3

hydroxy-18-methoxy-neocleroda-7,13(16),14-triene-20,12-(20-O-methyl)acetal (11) from peracetylisoeriocephalin (4).

To a solution of 4 (130 mg, 0.238 mmol) in a mixture of toluene (50 mL) and MeOH (5 mL) was added *p*-TsOH.H₂O(2 mg, 0.01 mmol) and the reaction mixture was subjected to microwave irradiation in a domestic microwave oven for 4 successive periods of 1 min each, using a potence of 100 W [28] and an open vessel. CAUTION: between each irradiation period the reaction mixture was cooled to 15 °C. Then the reaction was diluted with EtOAc (75 mL), washed with an aqueous solution of NaHCO₃ (10%, w/v) and the organic layer dried (Na₂SO₄) and evaporated to dryness yielding a residue (112 mg) which was subjected to column chromatography (Si gel, 15 g). Elution with EtOAc-petrol (2:1) gave 10 (99 mg, 0.191 mmol, less polar constituent, 80% yield) and 11 (10 mg, 0.018 mmol, 7.6%). When the irradiation was continued for a further 20 min only compound 11 was obtained.

Compound **10**: amorphous white solid, mp 55-70 °C; $[\alpha]_{D}^{18}$ +19.6° (*c* 0.424, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3140, 3120, 1500, 870 (furan), 3020 (oxirane), 1760, 1740, 1250, 1230 (OAc), 2950, 2860, 1445, 1370, 1200, 1085, 1050, 1025, 915. ¹H NMR: see Table 2. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 518 [M]⁺ (0.02), 487 (0.1), 458 (0.6), 427 (0.2), 416 (4), 374 (3), 314 (6), 296 (7), 283 (12), 265 (10), 173 (5), 145 (7), 95 (17), 94 (12), 91 (6), 85 (15), 81 (13), 55 (8), 47 (9), 43 (100). (Anal. Found: C, 62.47; H, 6.53%. C₂₇H₃₄O₁₀ requires: C, 62.54; H, 6.61%).

Compound **11**: colourless crystalline solid, mp 152-154 °C (EtOAc - *n*-hexane): $[\alpha]_{10}^{17}$ +13.6° (*c* 0.022, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3460 (OH), 3140, 1505, 875 (furan), 1760, 1740, 1250, 1230 (OAc), 2940, 2860, 2830, 1450, 1370, 1200, 1110, 1050, 1025, 905, 815, 800. ¹H NMR: see Table 2. EIMS *m/z* (rel. int.): 550 [M]⁺ (0.1), 519 (0.2), 490 (0.2), 448 (6). 430 (4), 406 (6), 388 (9), 370 (19), 343 (18), 328 (32), 297 (49), 283 (34), 265 (31), 213 (42), 161 (28), 115 (20), 95 (38), 94 (27), 91 (15), 81 (32), 55 (17), 45 (35), 43 (100). (Anal. Found: C, 60.87; H, 6.99%. C₂₈H₃₈O₁₁ requires: C, 61.08; H, 6.96%).

Preparation of $(12S,20R)-4\alpha,18;7\alpha,19;15,16$ -triepoxy- $6\alpha,7\beta$ -dihydroxy- 17β -neocleroda-13(16),14-diene-20,12-(20-O-methyl)acetal (12) and (12S,20R)- $7\alpha,19;15,16$ -diepoxy- $4\alpha,6\alpha,7\beta$ -trihydroxy-18-methoxy- 17β -neocleroda-13(16),14-diene-20,12-(20-O-methyl) acetal (13) from compound 10.

A mixture of **10** (350 mg, 0.675 mmol) and Na_2CO_3 (150 mg, 1.415 mmol) in MeOH solution (200 mL) was refluxed for 3 h. The solvent was evaporated under reduced pressure and low temperature (35 °C) and the residue was digested with CHCl₃ (4 x 10 mL). The chloroform extract was evaporated to dryness yielding a residue (280 mg) which was sujected to column chromatography (Si gel, 40 g, CHCl₃-MeOH 97:3 as eluent) giving **12** (228 mg, 0.581 mmol, less polar constituent, 86% yield) and **13** (22 mg, 0.051 mmol, 7.7%).

Compound 12: colourless crystalline solid, mp 214-217 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{17}$ -17.5° (*c* 0.143, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3540, 3390 (OH), 3140, 3110, 1500, 870 (furan), 2980, 2950, 2860, 2820, 1465, 1380, 1330, 1220, 1100, 1035, 1020, 1005, 975, 955, 910, 890, 850, 800. ¹H NMR: see Table 2. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 392 [M]⁺ (0.8), 374 (3), 361 (11), 343 (10), 313 (11), 283 (10), 245 (17), 239 (10), 220 (19), 219 (15), 203 (12), 190 (23), 175 (22), 173 (20), 163 (52), 161 (36), 145 (39), 135 (41), 124

(58), 95 (85), 94 (100), 91 (71), 81 (93), 55 (60), 53 (44), 43 (31). (Anal. Found: C, 64.52; H, 7.31%. $C_{21}H_{28}O_7$ requires: C, 64.27; H, 7.19%).

Compound **13**: colourless crystalline solid, mp 202-204 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{17}$ -33.8° (*c* 0.514, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3480, 3360, 3160 (OH), 1500, 870 (furan), 2960, 2880, 2820, 1450, 1385, 1375, 1325, 1200, 1195, 1165, 1105, 1060, 1045, 1020, 1000, 980, 930, 915, 835, 820, 790, 755. ¹H NMR: see Table 2. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 424 [M]⁺ (2), 406 (2), 393 (12), 375 (9), 347 (45), 329 (12), 317 (11), 314 (11), 299 (15), 283 (15), 265 (11), 233 (15), 189 (28), 177 (26), 175 (26), 163 (100), 159 (25), 147 (50), 145 (45), 135 (68), 124 (39), 121 (47), 105 (38), 95 (74), 94 (60), 93 (49), 91 (58), 81 (92), 79 (62), 67 (47), 55 (78), 45 (100), 43 (48), 41 (47). (Anal. Found: C, 62.09; H, 7.49%. C₂₂H₃₂O₈ requires: C, 62.25; H, 7.60%).

Compound 13 from compound 11.

Treatment of 11 (5 mg, 0.009 mmol) with Na_2CO_3 in MeOH solution as described above for 10 yielded 13 (3 mg, 0.007 mmol, 77%), identified by its mp, ¹H NMR and MS. Comparison (mixed mp, TLC) with an authentic sample confirmed the identity.

7-O-Acetyl-7,8-didehydroeriocephalin (14) from 7,8-didehydroeriocephalin (15).

Ac₂O-pyridine (20 mL, 1:1) treatment of **15** (200 mg, 0.434 mmol)[24] for 24 h at room temperature and work-up in the usual manner yielded **14** (218 mg, 0.434 mmol, quantitative yield): colourless crystalline solid, mp 182-184 °C decomp. (EtOAc - *n*-hexane); $[\alpha]_D^{19}$ -89.1° (*c* 0.311, CHCl₃). UV (MeOH) λ_{max} nm (log ε): 212 (3.81), 246.5 (4.09). IR (KBr) v_{max} cm⁻¹: 3140, 3120, 1505, 875 (furan), 1760, 1730, 1245, 1215 (OAc), 1685, 1640 (diosphenol), 2940, 2910, 2870, 1440, 1370, 1150, 1100, 1060, 1020, 970, 960, 910, 800. 'H NMR: see Table 4. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 502 [M]⁺ (0.3), 443 (6), 383 (2), 341 (6), 312 (5), 299 (6), 281 (8), 253 (5), 215 (5), 203 (5), 177 (5), 161 (5), 145 (5), 121 (5), 105 (7), 95 (11), 94 (36), 91 (11), 81 (15), 43 (100). (Anal. Found: C, 62.18; H, 6.11%. C₂₆H₃₀O₁₀ requires: C, 62.14; H, 6.02%).

Sodium borohydride reduction of compound 14 to give compound 16.

Compound 14 (33 mg, 0.065 mmol) in MeOH-dioxane solution (8 mL, 3:1) was treated with an excess of NaBH₄ (30 mg, 0.793 mmol) at room temperature for 15 min. Work-up in the usual manner gave 16 (30 mg, 0.059 mmol, 90% yield): colourless crystalline solid, mp 217-218 °C (EtOAc - *n*-hexane); $[\alpha]_D^{18}$ -45.1° (*c* 0.162, CHCl₃). The derivative 16 was identical (IR, ¹H NMR and MS) to a compound previously obtained [14] by reduction of 5; (lit. [14] 16: mp 210-212 °C, from CHCl₃ - *n*-hexane; $[\alpha]_D^{26}$ -46.3°, *c* 0.564, CHCl₃). Comparison (mixed mp, TLC) with an authentic sample confirmed the identity.

Sodium borohydride-cerium trichloride reduction of compound 14 to give compound 5.

Compound 14 (150 mg, 0.3 mmol) in MeOH solution (30 mL) was treated with NaBH₄ (45 mg, 1.2 mmol) and CeCl₃.7H₂O (447 mg, 1.2 mmol) for 3 h at room temperature. Workup in the usual manner yielded 5 (130 mg, 0.26 mmol, after crystallisation from EtOAc - *n*-hexane, 86%), identical in all respects (mp, $[\alpha]_D$, IR, ¹H NMR, MS, TLC) with 6-O-

Table 4 'H NMR	Spectral data	for compound	ls 14, 18, 23,	and 25-28"											
E	14	18	23	25	26	27	28	<i>J</i> _{ни} (Hz)	14	18	23	25	26	27	28
12	"	h	4	1.86 dddd	1.62 dddd	2.47 dddd	1.92 dddd	1α,1β	4	4	4	12.5	13.2	13.6	13.7
BI	4	h	4	''	1.96 hr d	2.06 dddd	4	Ια.2α	4	4	4	3.5	3.9	3.6	3.5
20	4	'n	4	4	2.05 ddddd	"	h	1α.2β	4	4	4	13.5	13.4	13.4	13.4
۶ ۲	4	4	4	1.34 at	1.50 qt	1.37 qt	1.57 qt	1a, 10B	1.1	12.0	4	12.0	12.3	12.2	10.6
÷ ځ	4	2.38 dddd	4	•	2.27 dddd	. "	. 4	1 β.2 α	4	4	4	4	3.4	3.8	4
, e	1.16 m	1.02 ddd	4	1.15 ddd	1.17 ddd	1.19 ddd	1.21 dt	18,28	ų	4	4	4.4	4.9	3.8	3.2
, 8			1	1	,	4.03 ť	4.02 br t'	18,108	4.4	4.0	4	2.0	2.3	2.7	5.0
18			4.80 dd ^c	5.65 d	5.57 d	,	I	2a,2B	4	4	4	13.5	13.4	13.4	13.4
- 6	,	,	ħ	2.18 ad	2.48 qd	2.39 qd	2.37 qd	2α,3α	4	5.0	q	4	5.4	4	4
9 90	2.47 dd	2.48 dd	4	2.62 dd	2.40 dd	2.18 dd	2.15 dd	2α,3β	4	4.8	4	3.1	1.7	2.9	2.9
AII	2.34 dd	2.59 dd	pp 16.1	bb 68.1	2.42 dd	1.92 dd	2.22 dd	2β,3α	4	12.0	4	13.5	13.9	13.4	13.4
E	2.74 dd	2.76 dd	2.78 dd	2.82 dd	2.57 dd	2.47 dd	2.27 dd	2 β.3 β	4	2.1	4	4.4	4.9	3.8	3.8
12	5.31 br (5.56 br t	5.09 dd	5.11 dd	5.08 1	4.97 dd	4.70 dd	3α,3β	13.0	14.2	4	13.0	13.9	13.3	13.4
4	6.39 dd	6.36 dd	6.30 dd	6.32 dd	6.41 dd	6.41 dd	6.43 dd	6 β ,8β	,	1	,	,	,	1.6	1.5
15	7.41 (7.43 (7.37 1	7.40 t	7.37 t	7.40 t	7.39 t	7 β.8 β	ı	,	6.9	6.4	4.9	•	ſ
16	7.36 m	7.46 m	7.35 m	7.37 m	7.39 m	7.42 m	7.40 m	8β,17	ı	,	7.4	7.3	7.3	6.8	6.6
Me-17	2.00 s	1.74 s	0.91 d	1.07 d	1.14 d	1.18 d	1.40 d	11A,11B	14.0	14.6	12.2	12.3	12.7	13.2	13.7
18A/	2.41 d	2.25 d	2.41 d	2.37 d	2.38 d	2.56 d	2.53 d	11A,12	8.7	8.6	4.4	4. I	8.3	10.4	7.4
18.8%	2.75 dd	2.91 dd	2.31 dd	3.03 dd	2.87 dd	3.28 dd	3.20 dd	11B,12	6.9	8.2	9.5	9.7	8.1	6.7	9.3
¥61	4.47 d	4.84 d	4.61 d	4.59 d	4.63 d	4.23 d	4.31 d	14,15	1.8	1.9	1.8	1.8	1.8	1.8	1.8
19B	4.69 d	4.95 d	4.67 d	4.64 d	4.68 d	4.70 d	4.55 br d	14,16	0.8	0.9	0.9	0.8	0.9	0.8	0.8
20	6.40 s		4.93 s	4.96 s	4.80 s	4.60 s	4.84 s	15,16	1.8	1.9	1.8	1.8	1.8	1.8	1.8
60-OH		,	,	ı	1	3.67 d	3.79 d	18A,18B	4.5	5.0	3.8	3.8	3.9	3.5	3.5
7α-OH		ı	3.42 d	ı			•	18B,3α	2.0	2.2	2.1	2.3	2.2	2.3	2.3
OAc	2.22 s	2.16 s	2.08 s	2.11 s	2.13 s	s 86.1	2.07 s	19A, 19B	11.7	12.5	11.8	11.7	12.2	11.4	12.2
	1.94 s	2.02 s	ı	2.08 s	2.10 s	•		19B,6β	ł	,	,	•	,	0	<0.3
	1.93 s	,	1	ı	,		,	68,6-OH [*]	ı	r	ı	r	ł	6.1	1.5
20-OMe		,	3.27 s	3.29 s	3.29 s	3.19 s	3.35 s	7B,7-OH*	,	'	5.2	,	•		·
"At 200 N	14, 18 a	and 23) or 500	MHz (25-28)) in CDCl ₃ solu	tion. Chemica	ul shifts (8 val	ues) are repor	ted with respec	it to the sig	gnal of re	sidual C	CHCI ₅ (8	7.25).		
"Overlapp	ed signal.														
	Linto o doubl	of (-1 6 U-)	ofter addition	of D.O											

"Collapsed into a doublet (J= 1.6 Hz) after addition of D₂O. "Collapsed into a broad doublet (J= 1.5 Hz) after addition of D₂O. "Collapsed into a doublet (J= 6.9 Hz) after addition of D₂O. "*Exo* hydrogen with respect to ring B. "*Endo* hydrogen with respect to ring B.

acetylisoeriocephalin (5)[14]. The derivative 5 was also obtained from 7-O-acetyleriocephalin (17)[14] by a thermal α -ketol rearrangement as previously described [26] for other 6-oxo- 7α -acetoxy-neoclerodane derivatives. This rearrangement was carried-out by heating a small sample of 17 (2 mg) on a Kofler block at 130 °C for 3 min. TLC analysis revealed the presence of 5 and starting material (17) in 1:1 ratio.

7-O-Acetylpicropolinone (18) from picropolinone (19).

Ac₂O-pyridine (10 mL, 1:1) treatment of **19** (150 mg, 0.360 mmol)[18-20] for 24 h at room temperature and work-up in the usual manner gave **18** (138 mg, 0.301 mmol, 83% yield): colourless crystalline solid, mp 197-199 °C (EtOAc - *n*-hexane); $[\alpha]_D^{20}$ +38.3° (*c* 0.478, CHCl₃). UV (MeOH) λ_{max} nm (log ε): 209 (3.97), 246 (4.06). IR (KBr) ν_{max} cm⁻¹: 3160, 3130, 1505, 875 (furan), 1765 (γ-lactone), 1760, 1730, 1260 (OAc), 1700, 1660 (diosphenol), 2960, 2880, 1470, 1370, 1170, 1160, 1025, 920, 730, 690. ¹H NMR: see Table 4. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 458 [M]⁺ (0.05), 399 (0.1), 398 (0.05), 338 (0.2), 326 (4), 250 (4), 205 (4), 177 (7), 121 (12), 95 (14), 94 (18), 91 (11), 81 (16), 55 (8), 43 (100). (Anal. Found: C, 62.91; H, 5.64%. C₂₄H₂₆O₉ requires: C, 62.87; H, 5.72%).

Sodium borohydride reduction of compound 18 to give compound 20.

Treatment of a solution of **18** (85 mg, 0.185 mmol) in MeOH-dioxane (10 mL, 3:1) with NaBH₄ (50 mg, 1.321 mmol) at room temperature for 15 min quantitatively yielded **20**: colourless crystalline solid, mp 275-278 °C (EtOAc); $[\alpha]_D^{21}$ +19.6° (*c* 0.136, CHCl₃-MeOH 1:1), IR, ¹H NMR and MS identical to those reported for the derivative obtained by NaBH₄ reduction of **21** (lit. [25] **20**: mp 275-278 °C; $[\alpha]_D^{21}$ +19.8°, *c* 0.121, CHCl₃-MeOH 1:1).

Sodium borohydride-boric acid reduction of compound 18 to give 6-O-acetylpicropolin (9).

To a solution of 18 (40 mg, 0.087 mmol) and H_3BO_3 (80 mg, 1.293 mmol) in MeOH (20 mL) an excess of NaBH₄ (100 mg, 2.642 mmol) was added in portions with stirring and the reaction mixture was kept for 30 min at room temperature. Then, the reaction was diluted with H_2O (50 mL) and extracted with EtOAc (4 x 20 mL). Work-up in the usual manner gave 9 (34 mg, 0.074 mmol, 84% yield), identical in all respects (mp, IR, ¹H NMR, MS and TLC) with the compound described above and previously obtained by acetylation of picropolin (7)[17] and by thermal rearrangement of capitatin (21)[26].

Preparation of (12S,20S)- (23) and (12S,20R)-19-acetoxy-4 α ,18;15,16-diepoxy-7 α -hydroxy-6-oxo-neocleroda-13(16),14-diene-20,12-(20-O-methyl)acetal (24) and their corresponding acetates (25 and 26) from eriocephalin (22).

A solution of 22 (160 mg, 0.346 mmol)[14,27] and p-TsOH.H₂O (2 mg, 0.01 mmol) in a mixture of toluene (15 mL) and MeOH (3 mL) was irradiated in a domestic microwawe oven for 3 successive periods of 2 min each in the conditions described above for 4. Work-up as described above yielded a crude of reaction (140 mg) which showed two spots on TLC, both less polar than 22. Column chromatography of this mixture (Si gel, EtOAc-petrol 1:1 as eluent) allowed the isolation of the less polar constituent (23, 36 mg, 0.083 mmol, 24% yield) and a mixture of other two compounds, both of them different from the second constituent (24) of the initial mixture. Compound 23 was an unstable substance which was slowly transformed into another compound (27, see below) on storage or in CHCl₃, Me₂CO or MeOH solutions. For the ¹H and ¹³C NMR spectra of 23 see Tables 4 and 3, respectively.

A fresh amount (160 mg) of the crude of the above reaction was treated with Ac₂Opyridine (10 mL, 1:1) for 48 h at room temperature giving **25** (92 mg, 0.193 mmol, less polar constituent, 52% yield) and **26** (52 mg, 0.109 mmol, 29%) after column chromatography (Si gel, petrol-EtOAc 7:3 as eluent).

Acetylation of 23 (1 mg) in the same conditions yielded 25, identified by TLC.

Compound **25**: amorphous white solid, mp 80-87 °C; $[\alpha]_{D}^{18}$ +135.1° (*c* 0.339, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3140, 1500, 875 (furan), 1750, 1250, 1230 (OAc), 1730 (ketone), 2940, 2880, 1450, 1370, 1160, 1110, 1070, 1030, 970, 915, 865, 790, 755. ¹H NMR: see Table 4. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 476 [M]⁺ (0.05), 445 (0.08), 371 (0.7), 283 (2), 190 (3), 189 (3), 163 (5), 161 (6), 147 (5), 121 (8), 95 (25), 94 (93), 91 (11), 81 (21), 55 (11), 43 (100). (Anal. Found: C, 62.81; H, 6.93%. C₂₅H₃₂O₉ requires: C, 63.01; H, 6.77%).

Compound **26**: colourless crystalline solid, mp 219-221 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{20}$ +43.4° (*c* 0.214, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3140, 3120, 1505, 875 (furan), 1755, 1745, 1230 (OAc), 1735 (ketone), 2960, 1475, 1455, 1390, 1370, 1160, 1090, 1050, 1020, 800, 750, 640, 630. ¹H NMR: see Table 4. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 476 [M]⁺ (0.1), 445 (0.7), 416 (0.3), 371 (0.4), 343 (0.8), 329 (0.7), 283 (1), 265 (1), 190 (2), 175 (2), 173 (2), 163 (4), 161 (5), 125 (22), 95 (17), 94 (100), 91 (7), 81 (16), 43 (69). (Anal. Found: C, 62.93; H, 6.59%. C₂₅H₃₂O₉ requires: C, 63.01; H, 6.77%).

Preparation of (12S,20S)- (27) and (12S,20R)-19-acetoxy-4 α ,18;15,16-diepoxy-6 α -hydroxy-7-oxo-neocleroda-13(16),14-diene-20,12-(20-O-methyl)acetal (28) from the mixture of compounds 23 and 24.

To a solution of the mixture 23 and 24 (450 mg, 1.036 mmol) in EtOAc (250 mL) was added Si gel (Merck, No. 7734, previously activated at 140 °C for 2 h, 30 g) and the mixture was stirred at room temperature for 55 h. Then, Si gel was removed by filtration through a pad of Celite and the solvent evaporated. The residue (443 mg) was subjected to column chromatography (Si gel, EtOAc-petrol 3:2 as eluent) giving 27 (266 mg, 0.613 mmol, 59% yield, less polar constituent) and 28 (164 mg, 0.377 mmol, 36%).

Compound 27: colourless crystalline solid, mp 204-206 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{18}$ +67.5° (*c* 0.504, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3440 (OH), 3160, 3130, 3120, 1510, 880 (furan), 3050 (oxirane), 1730, 1240 (OAc), 1715 (ketone), 2980, 2920, 2860, 1450, 1370, 1130, 1035, 1020, 1000, 930, 830, 800, 760. ¹H NMR: see Table 4. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 434 [M]⁺ (0.1), 374 (0.5), 361 (1.7), 301 (0.4), 287 (2.5), 255 (3), 227 (4), 219 (4), 189 (5), 175 (6), 163 (16), 161 (12), 145 (16), 135 (13), 133 (12), 121 (10), 105 (12), 95 (24), 94 (84), 91 (19), 81 (48), 79 (27), 67 (18), 55 (16), 53 (12), 43 (100), 41 (14). (Anal. Found: C, 63.21; H, 6.93%. C₂₃H₃₀O₈ requires: C, 63.58; H, 6.96%).

Compound 28: colourless crystalline solid, mp 219-222 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{18}$ -43.3° (*c* 0.374, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3490 (OH), 3140, 3120, 1500, 880 (furan), 3050 (oxirane), 1735, 1260 (OAc), 1720 (ketone), 2950, 2880, 1450, 1370, 1160, 1125, 1050, 1025, 1005, 925, 895, 810. ¹H NMR: see Table 4. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.):

[M]⁺ absent, 374 [M-AcOH]⁺ (1.2), 361 [M-CH₂OAc]⁺ (0.5), 301 [M-AcOH-CH₂OAc]⁺ (0.5), 287 (3), 255 (3), 233 (3), 227 (3), 219 (3), 209 (4), 189 (5), 173 (7), 163 (11), 161 (10), 147 (13), 145 (15), 135 (12), 131 (11), 123 (12), 119 (13), 107 (14), 105 (19), 95 (62), 94 (100), 91 (38), 81 (68), 79 (32), 77 (22), 67 (16), 55 (15), 53 (12), 43 (65), 41 (19). (Anal. Found: C, 63.18; H, 6.93%. $C_{23}H_{30}O_8$ requires: C, 63.58; H, 6.96%).

Microwave irradiation of isoeriocephalin (3) to give compounds 27 and 28.

Treatment of 3 (100 mg, 0.216 mmol)[14] as previously described for 22 gave 27 and 28 in low yield (10 and 6%, respectively), together with several decomposition products. Attempts at improving this reaction were unsuccessful.

Treatment of **27** with 1,4-diazabicyclo[2.2.2]octane to give (12S,20S)-19-acetoxy-4 α , 18; 15, 16-diepoxy-7-hydroxy-6-oxo-neocleroda-7, 13(16), 14-triene-20, 12-(20-O-methyl) acetal (**29**).

A solution of 27 (160 mg, 0.368 mmol) and DABCO (30 mg, 0.267 mmol) in toluene (100 mL) was heated at 120 °C for 10 h. Evaporation of the solvent and column chromatography (Si gel, petrol-EtOAc 1:1 as eluent) yielded 29 (10 mg, 0.023 mmol, less polar compound, 6.3%) and starting material (27, 136 mg, 0.313 mmol, 85%). When this reaction was carried out under Ar only starting material (27) was recovered.

Compound **29**: colourless crystalline solid, mp 163-165 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{18}$ +49.2° (*c* 0.413, CHCl₃). UV (EtOH) λ_{max} nm (log ϵ): 207 (3.81), 281 (4.05). IR (KBr) ν_{max} cm⁻¹: 3380 (OH), 3140, 3120, 1505, 880 (furan), 3060 (oxirane), 1740, 1240 (OAc), 1675, 1640 (diosphenol), 2940, 1450, 1390, 1370, 1290, 1160, 1140, 1050, 1030, 1020, 985, 805, 785, 700, 650. ¹H NMR: see Table 5. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 432 [M]⁺ (0.03), 372 [M-AcOH]⁺ (1.5), 357 (1), 312 (3), 299 (10), 281 (14), 263 (7), 219 (17), 177 (19), 159 (12), 131 (11), 105 (10), 95 (19), 94 (20), 91 (22), 81 (37), 77 (21), 69 (10), 55 (20), 43 (100). (Anal. Found: C, 64.17; H, 6.63%. C₂₃H₂₈O₈ requires: C, 63.88; H, 6.53%).

Acetic anhydride-pyridine treatment of 27: derivative 30.

Ac₂O-pyridine (10 mL, 1:1) treatment of **27** (80 mg, 0.184 mmol) for 48 h at room temperature and work-up as usual (elimination of the volatiles in rotavapor at 70 °C) yielded **30** (79 mg, 0.166 mmol, 90%) as an amorphous white solid: mp 55-65 °C; $[\alpha]_{D}^{19}$ +72.9° (*c* 0.837, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3140, 1505, 875 (furan), 3050 (oxirane), 1740, 1250 (OAc), 1730 (ketone), 2940, 1450, 1370, 1120, 1050, 1030, 1000, 980, 925. ¹H NMR: see Table 5. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 476 [M]⁺ (0.04), 416 [M-AcOH]⁺ (1.3), 401 (0.4), 385 (1), 361 (4), 296 (5), 283 (4), 268 (4), 239 (5), 227 (5), 187 (5), 175 (6), 173 (6), 163 (10), 159 (12), 145 (14), 105 (12), 95 (32), 94 (73), 91 (32), 81 (26), 77 (11), 55 (16), 43 (100). (Anal. Found: C, 63.12; H, 6.58%. C₂₅H₃₂O₉ requires: C, 63.01; H, 6.77%).

Acetic anhydride-pyridine treatment of 28: compounds 10 and 31.

Treatment of 28 (70 mg, 0.161 mmol) with Ac_2O -pyridine (8 mL, 1:1) as in the case of 27 (see above) gave a mixture of two compounds (TLC), which were separated by column chromatography (Si gel, petrol-EtOAc 1:1 as eluent) yielding 10 (8 mg, 0.015 mmol, 9.5%)

and **31** (60 mg, 0.126 mmol, most polar constituent, 78%). The derivative **10** was identified by its $[\alpha]_{D}$, ¹H NMR and MS and by comparison (TLC) with the compound described above.

HINMR	spectral da	ta for com	pounds 29-	32, 33 and	30-				~~~				
<u>н</u>	29	30	31	32	35	36	J _{нн} (Hz)	29	30	31	32	35	36
lα	2.44 ^b	b	b	1.89 dddd	2.34 dddd	ь	1α,1β	h	b	Ь	13.9	13.6	þ
1β	b	Ь	b	1.81 dddd	1.87 dddd	ь	Ια,2α	h	ь	h	3.9	3.5	þ
2α	h	b	b	2.07 br d	1.97 br d	ь	1α,2β	þ	13.4	13.3	13.4	13.3	Ь
2β	1.32 m	1.37 qt	1.55 qt	1.53 qt	1.30 qt	ь	1α,10β	þ	12.3	þ	12.5	12.6	11.8
3α	2.11 tdd	b	ь	2.00 tdd	2.46 tdd	b	1β,2α	þ	ь	þ	3.4	3.8	Ь
3β	1.18 dm	b	1.10 ddd	1.18 ddd	1.10 ddd	1.11 ddd	1β,2β	h	3.8	3.7	4.4	3.9	Ь
6β	-	5.03 br s	5.01 d	4.18 d ^c	3.51 dd	5.60 s	1β,10β	b	2.6	þ	3.7	3.0	b
7β	-	-	-	-	3.89 dd	-	2α,2β	b	13.4	13.4	13.4	13.3	b
8α	-	-	-	3.38 q	•	•	2α,3α	4.5	b	þ	4.4	4.3	b
8β	-	b	2.40 qd	-	1.43 qd	2.47 q	2α,3β	h	b	2.9	2.9	2.8	1.0
ιόβ	b	2.22 dd	ь	2.20 dd	1.54 dd	2.06 br d	2β,3α	13.4	13.4	13.4	13.4	13.3	1,
IIA	2.25 dd	1.92 dd	2.20 dd	2.12 dd	1.76 dd	1.83 dd	2β,3β	ь	3.8	3.8	4.4	3.9	8.4
1 I B	2.34 dd	2.47 dd	2.32 dd	2.17 dd	2.22 dd	2.25 dd	3α,3β	13.4	þ	13.4	13.4	13.3	14.5
12	5.12 dd	4.99 dd	4.78 br t	5.05 t	4.86 dd	4.93 dd	6β,7β	-	-	-	~	3.6	-
14	6.46 dd	6.41 dd	6.41 dd	6.38 dd	6.42 dd	6.41 dd	6β,8β	-	0.3	1.0	-	0	0
15	7.44 t	7.41 t	7.36 t	7.36 t	7.39 t	7.40 t	7β,8β	-	-	-	-	2.7	-
16	7.47 m	7.42 m	7.37 m	7.34 m	7.41 m	7.42 m	8α,17	-	· -	-	7.6	-	-
Me-17	1.99 s	1.15 d	1.33 d	1.24 d	1.19 d	0.97 d	8β,17	-	6.8	6.6	-	7.2	7.2
18A″	2.40 d	2.40 d	2.34 d	2.61 d	2.43 d	2.44 d	11A,11B	13.5	13.2	13.8	13.4	13.1	13.2
18 B ′	2.56 dd	3.06 dd	2.95 dd	3.27 dd	3.19 dd	2.88 dd	11A,12	10.0	10.2	7.8	8.1	10.4	9.9
19A	4.52 d	4.35 d	4.38 d	4.29 d	4.33 dd	3.49 dd	11B,12	7.1	6.8	8.8	8.1	6.8	6.8
19 B	4.64 d	4.83 d	4.64 d	4.73 d	4.58 d	4.15 d	14,15	1.9	1.7	1.7	1.7	1.8	1.8
20	4.79 s	4.62 s	4.87 s	4.83 s	5.36 s	5.29 s	14,16	0.8	0.8	0.9	0.7	0.9	0.8
6α-OH'	-	-	-	3.60 d	3.97 s	•	15,16	1.9	1.7	1.7	1.7	1.8	1.8
7α-OH	-	-	-	-	3.02 s	-	18A,18B	4.6	3.4	3.7	3.7	3.5	4.5
7-OH′	6.40 s	-	-	-	-	-	18B,3α	2.2	2.7	2.1	2.4	2.4	0
19-OH′		-	-	•	3.02 s	-	19A,19B	11.2	11.0	12.1	12.0	12.7	8.8
OAc	1.99 s	2.07 s	2.06 s	1.97 s	-	2.10 s	19A,6β	-	0	0	0	1.2	0
	-	1.95 s	2.02 s	-	-	2.09 s	19A,10β	0	0	0	0	0	0.8
20-0Me	3.26 s	3.18 s	3.28 s	3.22 s	3.32 s	3.32 s	6β,6-OH	-	-	-	1.9	0	-

Table 5 'H NMR Spectral data for compounds 29-32, 35 and 36^a

"All at 300 MHz, except for 32 (500 MHz), in CDCl₃ solution. Chemical shifts (δ values) are reported with respect to the signal of residual CHCl₃ (δ 7.25).

"Overlapped signal.

'Collapsed into a singlet after addition of D₂O.

^dExo hydrogen with respect to ring B.

"Endo hydrogen with respect to ring B.

'Disappeared after addition of D_2O .

Compound 31: colourless crystalline solid, mp 208-210 °C (EtOAc - *n*-hexane); $[\alpha]_{D}^{20}$ -34.4° (*c* 0.776, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3140, 3120, 1510, 875 (furan), 3040 (oxirane), 1745, 1260 (OAc), 1725 (ketone), 2940, 2860, 1375, 1160, 1100, 1030, 995, 920, 890, 800, 740, 680. ¹H NMR: see Table 5. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 476 [M]⁺ (0.5), 416 (0.5), 343 (0.6), 287 (3), 200 (3), 173 (4), 171 (4), 162 (5), 159 (5), 147 (7), 145 (5), 134 (8), 124 (16), 105 (9), 95 (23), 94 (37), 91 (19), 81 (14), 79 (15), 67 (11), 55 (8), 43 (100), 41 (15). (Anal. Found: C, 62.81; H, 6.82%. C₂₅H₃₂O₉ requires: C, 63.01; H, 6.77%).

Preparation of compounds 12 and 13 starting from 28 and 31.

Treatment of 28 (100 mg, 0.230 mmol) or 31 (50 mg, 0.105 mmol) with Na_2CO_3 (50 mg and 25 mg, 0.471 mmol and 0.235 mmol, respectively) in MeOH (50 mL and 20 mL, respectively) solution as described above for 10 also yielded 12 and 13 (in 10:1 ratio, overall yield 93% and 91% from 28 or 31, respectively).

Preparation of (12S,20R)-19-acetoxy-4α,18;15,16-diepoxy-6α-hydroxy-7-oxo-17β-neocleroda -13(16),14-diene-20,12-(20-0-methyl)acetal (**32**) from compound **28**.

To a solution of **28** (150 mg, 0.345 mmol) in anhydrous toluene (20 mL) was added DABCO (8 mg, 0.071 mmol) and the reaction mixture was heated at 120 °C for 6 h under Ar. Evaporation of the solvent and column chromatography of the residue (Si gel, EtOAcpetrol 3:2 as eluent) gave **32** (65 mg, 0.149 mmol, less polar compound, 43% yield) and starting material (**28**, 78 mg, 0.180 mmol, 52%). Compound **32** (colourless crystalline solid) had mp 192-194 °C (EtOAc - *n*-hexane); $[\alpha]_D^{20}$ -35.7° (*c* 0.611, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3440 (OH), 3140, 3120, 1610, 1550, 1505, 875 (furan), 3080 (oxirane), 1750, 1260, (OAc), 1730 (ketone), 2960, 2860, 1450, 1365, 1150, 1115, 1080, 1045, 1015, 1000, 980, 910, 810, 750. ¹H NMR: see Table 5. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): 434 [M]⁺ (0.4), 403 [M-OMe]⁺ (0.4), 374 [M-AcOH]⁺ (2), 283 (3), 255 (5), 219 (5), 191 (6), 189 (7), 163 (21), 161 (18), 147 (18), 125 (15), 121 (15), 105 (16), 95 (26), 94 (100), 91 (21), 81 (25), 67 (17), 55 (26), 43 (82). (Anal. Found: C, 63.63; H, 7.01%. C₂₃H₃₀O₈ requires: C, 63.58; H, 6.96%).

Treatment of picropolin (7) and 6-O-acetylpicropolin (9) with 1,4-diazabicyclo[2.2.2]octane.

Treatment of compound 7 and 9 with DABCO in the same conditions as those reported above for 28 was unsuccessful and starting materials were quantitatively recovered.

Preparation of teuvincentin C(2) starting from 6-O-acetylisoeriocephalin (5).

A solution of **5** (30 mg, 0.059 mmol)[14] and DABCO (5 mg, 0.044 mmol) in anhydrous toluene (30 mL) was heated at 120 °C for 7 h under Ar. Evaporation of the solvent and column chromatography of the residue (Si gel, EtOAc-petrol 7:3 as eluent) gave **2** (23 mg, 0.045 mmol, less polar constituent, 76% yield) and starting material (**5**, 4 mg, 0.008 mmol, 13%). Compound **2** had mp 221-223 °C (EtOAc); $[\alpha]_D^{20}$ -8.3° (*c* 0.361, CHCl₃) and IR, ¹H NMR and MS identical with those of teuvincentin C [12](lit. mp 220-225 °C; $[\alpha]_D^{22}$ -8.1°, *c* 0.246, CHCl₃). Comparison (mixed mp, TLC) with an authentic sample confirmed the identity.

Treatment of 27 with sodium carbonate: derivatives 33 and 34.

To a solution of 27 (200 mg, 0.460 mmol) in MeOH (20 mL) was added Na₂CO₃ (200 mg, 1.886 mmol) and the reaction mixture was stirred at room temperature for 4 h. Work-up in the usual manner gave 153 mg (0.390 mmol, 84% yield) of a mixture of the tautomers 33 (ketol form) and 34 (hemiacetal form) in 1.3:1 ratio, as was established from the ¹H NMR spectrum (signals for the C-19 methylene protons in 33: δ 3.75 and 4.56, both d, J_{gem} 11.0 Hz; for 34: δ 4.14 and 4.44, both d, J_{gem} 7.4 Hz). Attempts at isolating these compounds by chromatography or by crystallisation were unsuccessful.

Sodium borohydride reduction of the tautomeric mixture (**33** and **34**) to give (12S,20S)- 4α , 18; 15, 16-diepoxy- 6α , 7α -19-trihydroxy-neocleroda-13(16), 14-diene-20, 12-(20-O-methyl) acetal (**35**).

A solution of the mixture **33** and **34** (150 mg, 0.382 mmol) in MeOH (15 mL) was treated with NaBH₄ (80 mg, 2.144 mmol) at room temperature for 45 min. Then, the reaction mixture was diluted with H₂O (50 mL) and extracted with EtOAc (3 x 50 mL). Work-up as usual gave **35** as the sole reduction product (130 mg, 0.330 mmol, 86% yield): mp 75-85 °C, amorphous white solid; $[\alpha]_D^{18}$ +44.3° (*c* 0.316, CHCl₃). IR (KBr) ν_{max} cm⁻¹: 3450 (OH), 3140, 1505, 875 (furan), 2940, 2880, 1450, 1160, 1100, 1070, 1025, 1000, 980, 920, 840, 800. ¹H NMR: see Table 5. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): [M]⁺ absent, 344 [M-H₂O-MeOH]⁺ (3.5), 314 (4), 245 (10), 203 (14), 192 (18), 174 (17), 163 (100), 159 (36), 147 (35), 145 (29), 135 (42), 121 (57), 117 (31), 105 (33), 95 (71), 94 (37), 91 (77), 79 (73), 67 (47), 55 (58), 41 (76). (Anal. Found: C, 63.64; H, 7.55%. C₂₁H₃₀O₇ requires: C, 63.94; H, 7.66%).

Acetic anhydride-pyridine treatment of the tautomeric mixture (33 and 34) to give 30, (12S,20S)-6 α ,7 β -diacetoxy-4 α ,18;7 α ,19;15,16-triepoxy-neocleroda-13(16),14-diene-20,12-(20-O-methyl)acetal (36) and the tautomers 37 and 38.

Ac₂O-pyridine (4 mL, 1:1) treatment of the mixture **33** and **34** (153 mg, 0.390 mmol) for 10 days at room temperature and work-up in the usual manner gave a residue, from which the derivatives **36** (30 mg, 0.063 mmol, 16% yield), **30** (see above, 86 mg, 0.180 mmol, 46%) and the mixture **37** and **38** (55 mg, 0.126 mmol, 32%) were successively eluted from a chromatographic column (Si gel, EtOAc-petrol 2:1 as eluent).

Compound **36**: amorphous white solid, mp 95-105 °C; $[\alpha]_D^{19}$ -52.4° (*c* 0.225, CHCl₃). IR (KBr) v_{max} cm⁻¹: 3140, 1600, 1500, 880 (furan), 1745, 1250 (OAc), 2950, 1450, 1370, 1100, 1060, 1030, 980, 960, 940, 790, 740. ¹H NMR: see Table 5. ¹³C NMR: see Table 3. EIMS *m/z* (rel. int.): [M]⁺ absent, 416 [M-AcOH]⁺ (2), 402 (5), 314 (4), 313 (3), 193 (5), 175 (9), 163 (20), 161 (12), 145 (16), 121 (17), 105 (15), 94 (90), 91 (10), 81 (30), 55 (13), 43 (100), 41 (12). (Anal. Found: C, 62.83; H, 6.78%. C₂₅H₃₂O₉ requires: C, 63.01; H, 6.77%).

Tautomers 37 and 38.

From the mixture of these tautomers the hemiacetal form (**37**) crystallised slowly from EtOAc - *n*-hexane: mp 145-147 °C. IR (KBr) v_{max} cm⁻¹: 3400 (OH), 3150, 3130, 1505, 875 (furan), 1735, 1250 (OAc), 2960, 1440, 1370, 1300, 1160, 1110, 1100, 1070, 1030, 1020, 1010, 980, 975, 960, 920, 910, 890, 855, 810, 690. EIMS *m/z* (rel. int.): 434 [M]⁺ (1.2), 402 [M-MeOH]⁺ (0.7), 372 (1.5), 342 (2), 312 (3), 300 (4), 255 (3), 245 (12), 163 (29), 147 (13), 145 (14), 121 (15), 105 (14), 94 (51), 91 (30), 81 (38), 55 (22), 43 (100). (Anal. Found: C, 63.77; H, 7.10%. C₂₃H₃₀O₈ requires: C, 63.58; H, 6.96%). Compound **37** showed mutarrotation: $[\alpha]_{D}^{19}$ -0.8° (after 1 min), +7.8° (3 min), +22.1° (15 min), +22.1° (1 h)(*c* 0.515, CHCl₃). The ¹H NMR spectrum of **37** revealed that it was in equilibrium with the ketol form (**38**) in 1:1.3 ratio (CDCl₃ or C₆D₆ solution, at 20 °C). This ratio was modified by the temperature (1:1.3 at 20 °C, 1:1.5 at 35 °C, 1:1.6 at 45 °C) and the open form was always the main tautomer at those temperatures. Some significant ¹H NMR spectral data for the mixture **37** and **38** are the following (500 MHz, CDCl₃): **37** δ 5.31 (1H, br s, H-6β), 1.13 (3H, d, J

6.9 Hz, Me-17), 2.87 (1H, d, J 4.6, H_B-18), 4.84 (1H, s, H-20), 3.32 (3H, s, 20-OMe); **38** δ 5.13 (1H, br s, H-6β), 1.11 (3H, d, J 6.6, Me-17), 2.96 (1H, dd, J 3.6, 2.6, H_B-18), 4.55 (1H, s, H-20), 3.27 (3H, s, 20-OMe). ¹H NMR (500 MHz, C₆D₆): **37** δ 5.56 (1H, br s, H-6β), 1.41 (3H, d, J 7.1, Me-17), 2.67 (1H, d, J 4.7, H_B-18), 4.85 (1H, s, H-20), 3.10 (3H, s, 20-OMe); **38** δ 5.08 (1H, br s, H-6β), 1.13 (3H, d, J 6.8, Me-17), 2.86 (1H, dd, J 3.5, 2.3, H_B-18), 4.70 (1H, s, H-20), 3.08 (3H, s, 20-OMe).

Acknowledgements

This work was supported by the Dirección General de Enseñanza Superior (grants PB94-0104, PB96-0830) and the Consejería de Educación y Cultura de la Comunidad de Madrid (project 06G/001/96).

References

- [1] Merritt AT, Ley SV. Nat. Prod. Rep. 1992; 9: 243-287.
- [2] Rodríguez-Hahn L, Esquivel B, Cárdenas J. Prog. Chem. Org. Nat. Prod. 1994; 63: 107-196.
- [3] Rodríguez B, de la Torre MC, Jimeno ML, Bruno M, Fazio C, Piozzi F, Savona G, Perales A. Tetrahedron 1995; 51: 837-848.
- [4] Rodríguez B, de la Torre MC, Jimeno ML, Bruno M, Vassallo N, Bondì ML, Piozzi F, Servettaz O. J. Nat. Prod. 1997; 60: 348-355.
- [5] Nabeta K, Ishikawa T, Kawae T, Okuyama H. J. Chem. Soc., Chem. Commun. 1995; 681-682.
- [6] Nabeta K, Ishikawa T, Okuyama H. J. Chem. Soc., Perkin Trans. 1. 1995; 3111-3115.
- [7] Akhila A, Rani K, Thakur RS. Phytochemistry 1991; 30: 2573-2576.
- [8] Rogers D, Unal GG, Williams DJ, Ley SV, Sim GA, Joshi BS, Ravindranath KR. J. Chem. Soc., Chem. Commun. 1979; 97-99.
- [9] Kitagawa I, Simanjuntak P, Hori K, Nagami N, Mahmud T, Shibuya H, Kobayashi M. Chem. Pharm. Bull. 1994; 42:1050-1055.
- [10] Chatterjee A, Banerjee A, Bohlmann F. Tetrahedron 1977; 33: 2407-2414.
- [11] Chatterjee A, Banerjee A, Bohlmann F. Phytochemistry 1978; 17: 1777-1779.
- [12] Carreiras MC, Rodríguez B, Piozzi F, Savona G, Torres MR, Perales A. Phytochemistry 1989; 28: 1453-1461.
- [13] Lourenço A, de la Torre MC, Rodríguez B. Tetrahedron Lett. 1991; 32: 7305-7308.
- [14] Fernández-Gadea F, Rodríguez B, Savona G, Piozzi F. Phytochemistry 1984; 23:1113-1118.
- [15] Domínguez G, de la Torre MC, Rodríguez B. J. Org. Chem. 1991; 56: 6595-6600.
- [16] de la Torre MC, Domínguez G, Rodríguez B, Perales A, Simmonds MSJ, Blaney WM. Tetrahedron 1994; 50: 13553-13566.
- [17] Brieskorn CH, Pfeuffer T. Chem. Ber. 1967; 100: 1998-2010.
- [18] Márquez C, Valverde S. J. Chem. Soc., Perkin Trans. 1. 1979; 2526-2527.
- [19] Márquez C, Rabanal RM, Valverde S, Eguren L, Perales A, Fayos J. Tetrahedron Lett. 1980; 21: 5039-5042.
- [20] Fernández P, Rodríguez B, Savona G, Piozzi F. Phytochemistry 1986; 25: 181-184.
- [21] Rodríguez B, de la Torre MC, Perales A, Malakov PY, Papanov GY, Simmonds MSJ, Blaney WM. Tetrahedron 1994; 50: 5451-5468.
- [22] Pascual C, Fernández P, García-Alvarez MC, Marco JL, Fernández-Gadea F, de la Torre MC, Hueso-Rodríguez JA, Rodríguez B, Bruno M, Paternostro M, Piozzi F, Savona G. Phytochemistry 1986; 25: 715-718.
- [23] de la Torre MC, Bruno M, Piozzi F, Savona G, Rodríguez B, Omar AA. Phytochemistry 1991; 30: 1603-1606.
- [24] Hueso-Rodríguez JA, Fernández-Gadea F, Pascual C, Rodríguez B, Savona G, Piozzi F. Phytochemistry 1986; 25:175-180.
- [25] Mössner E, de la Torre MC, Rodríguez B. J. Nat. Prod. 1996; 59: 367-373.
- [26] de la Torre MC, Fernández P, Rodríguez B. Tetrahedron 1987; 43: 4679-4684.
- [27] Fayos J. Martínez-Ripoll M, Paternostro M, Piozzi F, Rodríguez B, Savona G. J. Org. Chem. 1979; 44:4992-4994.
- [28] Watkins KW. J. Chem. Educ. 1983; 60: 1043-1044.