# ChemComm

## COMMUNICATION

Check for updates

Cite this: Chem. Commun., 2019, 55, 553

Received 20th November 2018, Accepted 8th December 2018

DOI: 10.1039/c8cc09226e

rsc.li/chemcomm

### Facile synthesis of chiral [2,3]-fused hydrobenzofuran *via* asymmetric Cu(ı)-catalyzed dearomative 1,3-dipolar cycloaddition†

Lei Liang, <sup>(b)</sup><sup>a</sup> Hong-Ying Niu,<sup>b</sup> Dong-Chao Wang, <sup>(b)</sup>\*<sup>c</sup> Xin-He Yang,<sup>c</sup> Gui-Rong Qu<sup>c</sup> and Hai-Ming Guo <sup>(b)</sup>\*<sup>ac</sup>

Intermolecular asymmetric dearomative 1,3-dipolar cycloaddition of 2-nitrobenzofurans with azomethine ylides was enabled by using a chiral  $Cu(i)/(S,S_p)$ -<sup>i</sup>Pr-Phosferrox catalyst. As a result, a series of highly stereoselective chiral [2,3]-fused hydrobenzofurans possessing four contiguous stereogenic centers were obtained with good to high yields, diastereoselectivities and enantioselectivities. The reaction has broad substrate scope tolerating various functional groups.

The fused polycyclic hydrobenzofurans are naturally prevalent molecules, some of which exhibit a broad range of biological activities.<sup>1</sup> For example (Fig. 1), rocaglamide is a naturally occurring complex molecule that possesses a tricyclic hydrobenzofuran scaffold and exhibits potent anticancer, anti-inflammatory and neuroprotective activities.<sup>2</sup> (+)-Gynunone isolated from the roots of *Gynura elliptica* possesses anti-platelet aggregation activity.<sup>3</sup> As a member of the [2,3]-fused hydrobenzofurans, CID755673, which was found in a marine sponge, contains a benzofuro[2,3-*c*]azepin core and is a highly selective protein kinase inhibitor.<sup>4</sup> Moreover, azonazine, a novel dipeptide from *Aspergillus insulicola*, exhibits anti-inflammatory activity by inhibiting NF-kB luciferase and nitrite production.<sup>5</sup> In this context, the exploration of an efficient synthesis route for constructing this type of scaffold can be highly valuable.

The catalytic asymmetric dearomatization (CADA) reactions have emerged as a powerful tool for the rapid assembly of enantio-enriched three-dimensional polycyclic molecules from readily available aromatic substrates.<sup>6</sup> The dearomatization of electron-rich arenes (naphthol, indole, pyrrole, *etc.*) has been successfully developed based on their inherent nucleophilicity.<sup>7</sup>

Fig. 1 Representative examples of natural products containing fused polycyclic hydrobenzofuran scaffolds.

On the contrary, studies on the dearomatization of electrondeficient arenes as electrophiles remain relatively scarce.<sup>8,9</sup> Recently, several dearomatization reactions based on the electrophilicity of nitro-substituted arenes were reported. In 2017, You and co-workers<sup>9a</sup> developed a method for the straightforward construction of chiral tetrahydrofurobenzofurans via palladiumcatalyzed dearomative [3+2] cycloaddition of nitrobenzofurans with epoxybutene. Later, Yuan and co-workers9b reported the CADA reaction of nitrobenzofurans with 3-isothiocyanato oxindoles by using bis(oxazoline)/Zn(OTf)2 as a catalyst (Scheme 1A). Despite this elegant progress, the one-step construction of chiral hydrobenzofuran molecules containing multiple chiral centers and functional groups remains a challenging task. In recent years, catalytic asymmetric 1,3-dipolar cycloaddition of azomethine vlides with electron deficient C=C bonds has provided a direct route to highly substituted pyrrolidines with multiple stereogenic centers.<sup>10</sup> However, the merging of CADA reaction with cycloaddition of azomethine ylides to construct chiral-fused polycyclic compounds is rare.<sup>8h</sup> Herein, we report the straightforward construction of new chiral core structures of hydrobenzofuran enabled by copper-catalyzed asymmetric dearomative 1,3dipolar cycloaddition of 2-nitrobenzofurans with azomethine ylides (Scheme 1B).

Initially, we investigated the 1,3-dipolar reaction of 2-nitrobenzofuran 1a with *N*-benzylidene glycine methyl ester 2a by using  $Cs_2CO_3$  as the base in MTBE and  $Cu(MeCN)_4ClO_4/rac-$ BINAP-L1 complex as the catalyst at 0 °C. Gratifyingly, the reaction proceeded well, delivering the desired racemic dearomatized



**View Article Online** 

<sup>&</sup>lt;sup>a</sup> School of Environment, Henan Normal University, Xinxiang, Henan province, 453007, P. R. China. E-mail: ghm@htu.edu.cn

<sup>&</sup>lt;sup>b</sup> School of Chemistry and Chemical Engineering, Henan Institute of Science and Technology, Xinxiang, Henan Province 453003, P. R. China

<sup>&</sup>lt;sup>c</sup> Henan Key Laboratory of Organic Functional Molecules and Drug Innovation, School of Chemistry and Chemical Engineering, Henan Normal University, Xinxiang, Henan Province 453007, P. R. China. E-mail: wangdc@htu.edu.cn

<sup>†</sup> Electronic supplementary information (ESI) available. CCDC 1864355 (**3ah**). For ESI

and crystallographic data in CIF or other electronic format see DOI: 10.1039/c8cc09226e

A) Previous work:



Scheme 1 Different synthetic routes to chiral [2,3]-fused hydrobenzofuran compounds

product 3aa with a moderate yield (Table 1, entry 1). With such an effective method for synthesizing racemic products, we then turned our attention to chiral ligands (entries 2-11). The axially chiral bidentate phosphine ligands including (R)-BINAP-L2 and (R)-segphos-L3 were first examined. The cycloaddition proceeded well, resulting in 3aa as a major product with good yield, however, with poor-to-moderate enantioselectivities (Table 1, entries 2 and 3). The enantioselectivity of this reaction was further improved by screening different types of commercially available chiral ligands, such as (R)-Ph-BOX-L4, (R)-Ph-pybox-L5, (R)-Ph-PHOX-L6, (R,R)-DIOP-L7 and a series of phosferrox ligands L8-L11 (entries 4-13). As shown in Table 1, L4-L5 led to poor enantioselectivities (entries 4 and 5). When (R)-Ph-PHOX-L6 and (R,R)-DIOP-L7 were used as ligands, the chiral cycloaddition product was obtained with moderate results (entries 6 and 7). We further focused on the screening of phosferrox ligands with different substituents, among which, the  $(S,S_p)$ -<sup>i</sup>Pr-phosferrox-L8 was the most promising ligand, generating the desired product 3aa with 82% yield, 98% ee and 9:1 dr (entries 8-11). In the presence of  $(S,S_p)$ -<sup>i</sup>Pr-phosferrox-L8, the catalyst Cu(MeCN)<sub>4</sub>ClO<sub>4</sub> was changed to CuCl, Cu(OTf)<sub>2</sub> and AgCO<sub>2</sub>CF<sub>3</sub>. These central metals failed to further improve the enantioselectivity or diastereoselectivity under identical reaction conditions (entries 12-14). Furthermore, the screening of other solvents and bases also did not show better results (see Table S1, ESI<sup>+</sup>).

Finally, the optimal reaction conditions were obtained as follows: 5 mol% of Cu(MeCN)<sub>4</sub>ClO<sub>4</sub>, 5.5 mol% of L8 and 10 mol% of  $Cs_2CO_3$  in MTBE at 0 °C (entry 8).

Under the optimal reaction conditions, various azomethine ylide precursors,  $\alpha$ -iminoesters 2a-2l, were tested by reacting with 2-nitrobenzofuran 1a using the asymmetric dearomative [3+2] reaction (Scheme 2). The effect of the group at the para position of the benzene ring (3ab-3af) was firstly investigated. Both the electron-rich and electron-deficient substituents were well tolerated and led to the corresponding adducts with good yields (75-85%) and dr values (up to 10:1 dr), and excellent ee values (98–99%). The reaction proceeded smoothly with  $\alpha$ -iminoesters containing an aryl ring with meta substituents (2g, 2h), generating

Optimization of reaction conditions<sup>a</sup> Table 1



4	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L4	63	9:1	5	
5	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L5	55	8:1	3	
6	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L6	80	9:1	53	
7	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L7	63	7:1	56	
8	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L8	83	9:1	98	
9	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L9	76	8:1	83	
10	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L10	80	9:1	89	
11	Cu(MeCN) <sub>4</sub> ClO <sub>4</sub>	L11	81	7:1	78	
12	CuCl	L8	65	6:1	93	
13	$Cu(OTf)_2$	L8	77	7:1	91	
14	AgCO <sub>2</sub> CF <sub>3</sub>	L8	21	1:3	92	

<sup>a</sup> Reaction conditions: 1a (0.2 mmol), 2a (0.3 mmol), Cu(MeCN)<sub>4</sub>ClO<sub>4</sub> (5 mol%), L (5.5 mol%) and Cs<sub>2</sub>CO<sub>3</sub> (10 mol%) in MTBE (2.0 mL), 0 °C, N<sub>2</sub>, 12 h.<sup>b</sup> Yields of isolated product.<sup>c</sup> Determined by <sup>1</sup>H NMR analysis of the crude mixture. <sup>d</sup> Determined by chiral HPLC analysis.

the desired products with good yields and diastereoselectivities, and excellent enantioselectivities (up to 73% yield, 5:1 dr and 97% ee). Moreover, the 2-thienyl- and 2-naphthyl-substituted  $\alpha$ -iminoesters (2i and 2j) also delivered the corresponding compounds 3ai and 3aj with good results. Delightfully,  $\alpha$ -iminoester 2k derived from an aliphatic aldehyde and α-iminoester 2l with an alkenyl substituent were also successfully employed under the same reaction conditions, and the targeted chiral [2,3]-fused benzofuranpyrrolidines 3ak and 3al were obtained with excellent enantioselectivities (up to 96% ee).

Subsequently, we focused on the groups on the benzofuran ring. A wider variety of nitrobenzofurans were prepared and evaluated (Scheme 3). The reaction proceeded smoothly with N-benzylidene glycine methyl ester 2a, furnishing the respective cycloadducts with excellent results (3ba-3la). Substrates bearing various substituents, either with electron-withdrawing or electrondonating groups, at the 5 position (1b-1g) led to the corresponding dearomatized products with good yields (74-82%) and diastereoselectivities (6:1-15:1 dr), and excellent enantioselectivities (94-99% ee). A similar trend was also observed when substituting at the 6 or 7 position (Br, Me, MeO) (3ha-3ka). Notably, the highest enantioselectivities were observed in 5-nitro, 6-methyl or 6-methoxy nitrobenzofuran (1g, 1i and 1j). Furthermore, 4-chloro



Scheme 2 Substrate scope for azomethine ylides. Reaction conditions: **1a** (0.2 mmol), **2a** (0.3 mmol), Cu(MeCN)<sub>4</sub>ClO<sub>4</sub> (5 mol%), **L8** (5.5 mol%) and Cs<sub>2</sub>CO<sub>3</sub> (10 mol%) in MTBE (2.0 mL), 0 °C, N<sub>2</sub>, 12 h. Yields of isolated product. The dr ratios were determined by <sup>1</sup>H NMR analysis of the crude mixture. The ee values were determined by chiral HPLC analysis.



nitrobenzofuran was employed under the same reaction conditions, and the reaction proceeded to deliver the chiral tricyclic product **3la** with good results, but with slightly lower enantioselectivity (92% ee).



Fig. 2 The absolute configuration of the cycloadduct 3ah.

A) Gram-scale experiment



Scheme 4 Gram-scale synthesis and transformation.

The absolute configuration of [2,3]-fused benzofuranpyrrolidine product **3ah** (Fig. 2), determined by single-crystal X-ray diffraction analysis, was unambiguously found to be (1*S*,3*S*,3*aS*,8*bS*).

To further explore the prospective use of this methodology, the gram-scale synthesis of [2,3]-fused benzofuranpyrrolidine was carried out. As shown in Scheme 4A, 3ja was obtained, with 79% yield, 10:1 dr, and 99% ee, by treating 5.0 mmol 2-nitro-6methoxy-benzofuran 1j with N-benzylidene glycine methyl ester 2a in the presences of 5 mol% Cu(MeCN)<sub>4</sub>ClO<sub>4</sub> and 5.5 mol%  $(S,S_{\rm p})^{-i}$ Pr-Phosferrox-L8. The transformations of [2,3]-fused benzofuranpyrrolidine products were carried out. Firstly, zinc dust in a methanolic hydrogen chloride solution was used to reduce the nitro group to the respective amine 4aa (Scheme 4B). After that, the ester group was selectively reduced using LiAlH<sub>4</sub> (2 equiv.) as a reducing agent, and the corresponding product containing a hydroxyl group 4ja was obtained with good results (Scheme 4C). Additionally, the phenyl group was attached to the aromatic ring via the Suzuki-Miyaura cross coupling of 3fa with phenylboronic acid, and 5-phenyl substituted nitrobenzofuranpyrrolidine 4fa was obtained with 88% yield and 95% ee (Scheme 4D).

In summary, we successfully developed a highly enantioselective dearomative 1,3-dipolar cycloaddition reaction of nirtobenzofurans with iminoesters. Using Cu(i)<sup>*i*</sup>Pr-phosferrox complex as the catalyst, a wide range of [2,3]-fused benzofuranpyrrolidines bearing nitro groups on the chiral tetrasubstituted carbon center were obtained with moderate to high yields, diastereoselectivities and enantioselectivities (up to 86% yields, 11:1 dr and 99% ee). The reaction features a general scope for both azomethine ylides and nitrobenzofurans. This work provides a facile synthesis strategy for the construction of a highly substituted chiral tricyclic hydrobenzofuran scaffold.

We are grateful for the financial support from the National Natural Science Foundation of China (21672055 and U1604283), and the 111 Project (D17007) for support.

#### Conflicts of interest

There are no conflicts to declare.

#### Notes and references

- (a) H. Ding, P. L. DeRoy, C. Perreault, A. Larivée, A. Siddiqui, C. G. Caldwell, S. Harran and P. G. Harran, Angew. Chem., Int. Ed., 2015, 54, 4818; (b) N. Ribeiro, F. Thuaud, C. Nebigil and L. Désaubry, Bioorg. Med. Chem., 2012, 20, 1857; (c) R. R. Knowles, J. Carpenter, S. B. Blakey, A. Kayano, I. K. Mangion, C. J. Sinz and D. W. C. MacMillan, Chem. Sci., 2011, 2, 308; (d) S. Kim, A. A. Salim, S. M. Swanson and A. D. Kinghorn, Anticancer Agents Med. Chem., 2006, 6, 319; (e) E. B. Melian and K. L. Goa, Drugs, 2002, 62, 107; (f) P. R. Blakemore and J. D. White, Chem. Commun., 2002, 1159; (g) S. Han, J. E. Sweeney, E. S. Bachman, E. J. Schweiger, G. Forloni, J. T. Coyle, B. M. Davis and M. M. Joullié, Eur. J. Med. Chem., 1992, 27, 673.
- 2 M. A. Arai, Y. Kofuji, Y. Tanaka, N. Yanase, K. Yamaku, R. G. Fuentes, U. K. Karmakar and M. Ishibashi, *Org. Biomol. Chem.*, 2016, **14**, 3061.
- 3 W.-Y. Lin, C.-M. Teng, I.-L. Tsai and I.-S. Chen, *Photochemistry*, 2000, **53**, 833.
- 4 E. Torres-Marquez, J. Sinnett-Smith, S. Guha, R. Kui, R. T. Waldron, O. Rey and E. Rozengurt, *Biochem. Biophys. Res. Commun.*, 2010, 391, 63.
- 5 J.-C. Zhao, S.-M. Yu, Y. Liu and Z.-J. Yao, Org. Lett., 2013, 15, 4300.
  6 (a) Asymmetric Dearomatization Reactions, ed. S.-L. You, Wiley-VCH, 2016; (b) W.-T. Wu, L. Zhang and S.-L. You, Chem. Soc. Rev., 2016, 45, 1570; (c) C. Zheng and S.-L. You, Chem, 2016, 1, 830; (d) C.-X. Zhuo, C. Zheng and S.-L. You, Acc. Chem. Res., 2014, 47, 2558.

- 7 (a) Z.-P. Yang, R. Jiang, C. Zheng and S.-L. You, J. Am. Chem. Soc., 2018, 140, 3114; (b) G. Zhu, Y. Li, G. Bao, W. Sun, L. Huang, L. Hong and R. Wang, ACS Catal., 2018, 8, 1810; (c) Y. Wang, C. Zheng and S.-L. You, Angew. Chem., Int. Ed., 2017, 56, 15093; (d) L. Huang, Y. Cai, C. Zheng, L.-X. Dai and S.-L. You, Angew. Chem., Int. Ed., 2017, 56, 10545; (e) R.-Q. Xu, Q. Gu and S.-L. You, Angew. Chem., Int. Ed., 2017, 56, 755; (f) S. Bera, C. G. Daniliuc and A. Studer, Angew. Chem., Int. Ed., 2017, 56, 7402; (g) Z.-P. Yang, C. Zheng, L. Huang, C. Qian and S.-L. You, Angew. Chem., Int. Ed., 2017, 56, 1530; (h) D.-C. Wang, M.-S. Xie, H.-M. Guo, G.-R. Qu, M.-C. Zhang and S.-L. You, Angew. Chem., Int. Ed., 2016, 55, 11834; (j) K.-Y. Ye, Q. Cheng, C.-X. Zhuo, L.-X. Dai and S.-L. You, Angew. Chem., Int. Ed., 2016, 55, 8113; (k) Q. Cheng, Y. Wang and S.-L. You, Angew. Chem., Int. Ed., 2016, 55, 3496.
- Q. Cheng, F. Zhang, Y. Cai, Y.-L. Guo and S.-L. You, Angew. Chem., Int. Ed., 2018, 57, 2134; (b) D. J. Rivinoja, Y. S. Gee, M. G. Gardiner, J. H. Ryan and C. J. T. Hyland, ACS Catal., 2017, 7, 1053; (c) D.-F. Yue, J.-Q. Zhao, X.-Z. Chen, Y. Zhou, X.-M. Zhang, X.-Y. Xu and W.-C. Yuan, Org. Lett., 2017, 19, 4508; (d) M. Laugeois, J. Ling, C. Férard, V. Michelet, V. Ratovelomanana-Vidal and M. R. Vitale, Org. Lett., 2017, 19, 2266; (e) Y. Li, F. Tur, R. P. Nielsen, H. Jiang, F. Jensen and K. A. Jørgensen, Angew. Chem., Int. Ed., 2016, 55, 1020; (f) J.-Q. Zhao, Z.-J. Wu, M.-Q. Zhou, X.-Y. Xu, X.-M. Zhang and W.-C. Yuan, Org. Lett., 2015, 17, 5020; (g) B. M. Trost, V. Ehmke, B. M. O'Keefe and D. A. Bringley, J. Am. Chem. Soc., 2014, 136, 8213; (h) A. Awata and T. Arai, Angew. Chem., Int. Ed., 2014, 53, 10462.
- 9 (a) Q. Cheng, H.-J. Zhang, W.-J. Yue and S.-L. You, Chem, 2017,
  3, 428; (b) J.-Q. Zhao, X.-J. Zhou, Y. Zhou, X.-Y. Xu, X.-M. Zhang and
  W.-C. Yuan, Org. Lett., 2018, 20, 909; (c) X.-W. Liang, X. Chen,
  Z. Zhang and S.-L. You, Chin. Chem. Lett., 2018, 29, 1212.
- 10 (a) S. Xu, Z.-M. Zhang, B. Xu, B. Liu, Y. Liu and J. Zhang, J. Am. Chem. Soc., 2018, 140, 2272; (b) B. Feng, L.-Q. Lu, J.-R. Chen, G. Feng, B.-Q. He, B. Lu and W.-J. Xiao, Angew. Chem., Int. Ed., 2018, 57, 5888; (c) F. Esteban, W. Cieślik, E. M. Arpa, A. Guerrero-Corrella, S. Díaz-Tendero, J. Perles, J. A. Fernández-Salas, A. Fraile and J. Alemán, ACS Catal., 2018, 8, 1884; (d) H. Deng, W.-L. Yang, F. Tian, W. Tang and W.-P. Deng, Org. Lett., 2018, 20, 4121; (e) F. Tian, F.-S. He, H. Deng, W.-L. Yang and W.-P. Deng, Org. Lett., 2018, 20, 3838; (f) J. Corpas, A. Ponce, J. Adrio and J. C. Carretero, Org. Lett., 2018, 20, 3179; (g) B. Xu, Z.-M. Zhang, B. Liu, S. Xu, L.-J. Zhou and J. Zhang, Chem. Commun., 2017, 53, 8152; (h) Z.-M. Zhang, B. Xu, S. Xu, H.-H. Wu and J. Zhang, Angew. Chem., Int. Ed., 2016, 55, 6324; (i) A. Pascual-Escudero, A. de Cózar, F. P. Cossío, J. Adrio and J. C. Carretero, Angew. Chem., Int. Ed., 2016, 55, 15334.