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years. Various dienophiles such as maleic anhydride, benzyne, dimethyl acetylenedicarboxylate, and methyl acrylate yielded azabicyclo[2.2.2]octane derivatives. Blocking of the nitrogen atom by N-methylation or benzylation is necessary to prevent nucleophilic addition of the amide group to the dienophile. All these reactions proceed with a low yield of cycloaddition products, suggesting that 2-pyridone system behaves as a neutral diene. Three modifications of the pyridone structure can enhance the reactivity: 1. a vinyl moiety at the nitrogen atom, 2. high pressure (up to 15 kbar) and 3. introduction of a substituent in the 6-position, which has a great influence of the rate of cycloaddition due to a steric effect.

Some of these cycloaddition products are interesting synthetic intermediates. The 2-aza-3-oxobicylo[2.2.2]oct-7-ene-5,6-dicarboxylic anhydride can be transformed to 3-methoxy-2-azabarrelene. The cycloaddition of methyl acrylate to 1-benzyl-2-pyridone led to *endo* Diels-Alder adduct 1 in 17% yield, which was transformed to (\pm) epiibogamine. 4

6 a

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Skeleton via [4+2]-Cycloaddition

A Facile Entry to the 2-Azabicyclo[2.2.2]octane-6-one

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2-Aza-bicyclo[2.2.2]octane-6-one derivatives are prepared from 5-hydroxy-2-pyridone via [4+2]-cycloaddition with phenyl vinyl sulfone as an ethylene equivalent. Complete regiospecificity is observed. The *cxo-endo*¹⁵ ratio of the adducts is 7:3. A four step reaction sequence affords 2-aza-6-dimethoxybicyclo[2.2.2]octane, a starting material in the synthesis of desethylibogamine.

There has been considerable interest in the Diels-Alder reaction of N-substituted 2(1H)-pyridones with dienophiles in recent

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Recently, we have shown that the hydroxy function in the 5-position of 2-pyridone serves as a latent ketone functionality. 8 δ -Aminolaevulinic acid can be synthesized in 3 step. 9 Two factors make the dibenzyl derivative 2 attractive as a diene in the Diels-Alder reaction: 1. a donor substituent in the 5-position of the pyridone influences the reaction rate and the regio- and stereose-lectivity of the cycloaddition, and 2. the resulting benzyl enol ether moiety can easily be hydrolized to a ketone.

Here we report that cycloadditon of phenyl vinyl sulfone¹⁰ (3) to 2 in boiling toluene furnishes the diastereomers 4a and 4b (exo, endo).¹⁵ The reaction is completely regiospecific. Isomers 4a and 4b can be separated by flash chromatography providing a 4a: 4b ratio of 7: 3. The epimers can be hydrolyzed in acetic acid with 1 normal hydrochloric acid to give the ketones 5a and 5b, respectively.

Desulfonation with sodium amalgam¹¹ of $\mathbf{5a}$ or $\mathbf{5b}^{12}$ requires the protection of the ketone as the dimethyl acetal $\mathbf{6a}$ or $\mathbf{6b}$. Reduction of $\mathbf{7}$ with lithium aluminum hydride in tetrahydrofuran and subsequent debenzylation by hydrogenation over palladium on carbon in methanol provides $\mathbf{9}$. During desulphonation of the mixture $\mathbf{6a/6b}$ or $\mathbf{6a}$ complete epimerization of $\mathbf{6a}$ to $\mathbf{6b}$ occurs. This entry to the 2-azabicyclo[2.2.2]octane-6-one system is an attractive alternative path way to the common procedure developed by Huffman¹³ for the synthesis of desethylibogamine. In addition, further derivatization of the adducts $\mathbf{6a}$ and $\mathbf{6b}$ is possible via the α -sulfonyl carbanion derivative.

 1 H-NMR were taken using a Bruker AC 200 MHz FT-spectrometer with TMS as internal standard. $R_{\rm f}$ values are quoted for Merck silicagel 60 GF_{2.54} TLC plates of thickness 0.25 mm; detection: UV and iodine vapor. All experiments were carried out under nitrogen atmosphere.

exo/endo-2-Aza-2-benzyl-6-benzyloxy-8-phenylsulfonylbicyclo[2.2.2]oct-5-en-3-one (4a/4b):

To a stirred solution of 1-benzyl-5-benzyloxy-2-pyridone (2; 5.82 g, 20 mmol) in toluene (150 mL) is added phenyl vinyl sulfone (3; 8.4 g, 50 mmol). This mixture is refluxed for 7d. The solvent is evaporated, and the residual brown oil is flash-chromatographed on silica gel (Merck, $0.015-0.04 \, \mu m$, $30 \, \text{cm} \times 4 \, \text{cm}$, $20 \, \text{mL}$ fractions) using CH₂Cl₂/EtOAc (9:1) as eluent. Evaporation of solvent from the fractions containing pure exo-product 4a gives a residue, which is triturated with ether. Product 4a is isolated by suction as colorless powder; yield $3.0 \, \text{g}$ (33%).

A sample for analytic purposes of this exo-product 4a can be recrystallized from EtOH; mp 165-166 °C; $R_f = 0.48$ (CHCl₃/EtOAc, 9:1).

Evaporation of the solvent of the remaining fractions provides an exo/endo-mixture 3 g. Pure endo-product 4b can be isolated by a second flash chromatography under the above mentioned conditions; yield 1.8 g (20%); mp 117-119°C (EtOH); $R_f=0.35$ (CHCl₃/9:1); together with 1.2 g (13%) of further exo product.

exo-Product 4a:

C₂₇H₂₅NO₄S calc. C 70.56 H 5.48 N 3.05 (459.5) found 70.48 5.41 3.02

MS: m/z = 459.

IR (KBr): v = 2998, 1665 (amide), 1640 (C=C-O) cm⁻¹.

¹H-NMR (CDCl₃): $\delta = 2.10-2.17$ (m, 2 H, H-7); 3.57 (ddd, 1 H, J = 9 Hz, 4 Hz, 2 Hz, H-8); 3.70 (dd, 1 H, J = 6.6 Hz, 2.0 Hz, H-4); 4.02 (dd, 1 H, J = 5, 2.5 Hz, H-1); 4.42 (s, 2 H, NCH₂); 4.64 (s, 2 H, OCH₂); 4.95 (dd, 1 H, J = 6.6 Hz, 2.5 Hz, H-5); 7.15-7.69 (m, 15 H, H_{acom}).

endo-Product 4b:

C₂₇H₂₅NO₄S calc. C 70.56 H 5.48 N 3.05 (459.5) found 70.31 5.39 2.98

MS: m/z = 459.

IR (KBr): v = 2933, 1676 (amide), 1642 (C=C-O) cm⁻¹.

¹H-NMR (CDCl₃): δ = 2.10–2.19 (m, 2 H, H-7); 3.14 (ddd, 1 H, J = 7.5 Hz, 6 Hz, 2.8 Hz, H-8); 3.47 (dd, 1 H, J = 7.2 Hz, H-4); 3.87–3.96, 4.56–4.85 (m, 5 H, NCH₂, OCH₂, H-1); 5.28 (dd, 1 H, J = 7 Hz, 2.5 Hz, H-5); 6.98–7.66 (m, 15 H, H_{arom}).

exo/endo-2-Aza-2-benzyl-8-phenylsulfonylbicyclo[2.2.2]octane-3,6-dione (5a/5b):

Enol ether 4a and 4b (1.38 g, 3 mmol), AcOH (30 mL) and 2N HCl (10 mL) are stirred for 10 min at 40 °C to give a clear solution. Stirring is continued for 1 h at room temperature. Water (150 mL) is added, and the mixture is extracted with CH_2Cl_2 (3 × 50 mL). The combined organic phase is washed with sat. aq. NaHCO₃ solution (2 × 20 mL) and with water (20 mL), dried (Na₂SO₄) and evaporated. The clear oily residue crystallizes to give colorless needles, which are recrystallized from EtOH.

exo-Dione 5a; yield: 1.1 g (99%); mp 154-156°C.

C₂₀H₁₉NO₄S calc. C 65.02 H 5.18 N 3.79 (369.4) found 64.98 5.22 3.83

MS: m/z = 369.

IR (KBr): v = 1751 (C=O), 1673 (amide), 1450, 1420 cm⁻¹.

¹H-NMR (CDCl₃) δ = 2.13 (ddd, 1 H, J = 15 Hz, 10.7 Hz, 4.3 Hz, H-7e); 2.46 (ddd, 2 H, J = 15 Hz, 6.3 Hz, 1.5 Hz, H-7a); 2.55 (d, 2 H, J = 1.7 Hz, H-5); 3.25 (dd, 1 H, J = 5 Hz, 2.5 Hz, H-4); 3.54 (ddd, J = 10.7 Hz, 6.3 Hz, 2.5 Hz, H-8); 3.80 (dd, 1 H, J = 4.3 Hz, 1.5 Hz, H-1); 4.24 (d, 1 H, J = 14.6 Hz, NCH₂); 4.79 (d, 1 H, J = 14.6 Hz, NCH₂); 7.15–7.89 (m, 10 H, H_{arom}).

endo-Dione 5b; yield: 1.1 g (99%); mp 178-179°C.

C₂₀H₁₉NO₄S calc. C 65.02 H 5.18 N 3.79 (369.4) found 65.06 5.13 3.70

MS: m/z = 369.

IR (KBr): v = 1737 (C=O), 1700 (amide), 1454, 1445 cm⁻¹.

¹H-NMR (CDCl₃): δ = 2.32–2.50 (m, 2 H, H-7a, H-7e); 2.65–2.67 (m, 2 H, H-5a, H-5e); 3.03–3.13 (m, 2 H, H-4, H-8); 3.43 (d, 1 H, J = 3 Hz, H-1); 3.73 (d, 1 H, J = 14.2 Hz, NCH₂); 5.04 (d, 1 H, J = 14.2 Hz, NCH₂); 6.95–7.76 (m, 10 H, H_{arom}).

exo/endo-2-Aza-2-benzyl-6,6-dimethoxy-8-phenylsulfonylbicyclo[2.2.2] octane-3-one (6a/6b):

To a stirred solution of 5a or 5b (0.74 g, 2 mmol) in abs. MeOH (30 mL) is added trimethylorthoformate (5 mL) and TsOH (0.2 g). The mixture is refluxed for 5 h. The solvent is evaporated, and the colorless oil is dissolved in CH_2Cl_2 (30 mL). The organic layer is washed with sat. aq. NaHCO₃ solution (2 × 30 mL) and with water (30 mL), dried (Na₂SO₄) and evaporated. Treating the residue with EtOH gives a colorless powder, which can be recrystallized from EtOH (6a) or from MeOH (6b).

exo-Acetal 6a; yield: 0.69 g (83 %); mp 176-177 °C.

C₂₂H₂₅NO₅S calc. C 63.59 H 6.07 N 3.37 (415.5) found 63.45 5.93 3.41

MS: m/z = 415.

IR (KBr): $\gamma = 1670$ (amide); 1500, 1360, 950 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.74 (m, 2 H, H-5a, H-5e); 2.55 (m, 1 H, H-7e); 2.80–2.91 (m, 2 H, H-4, H-7a); 3.12 (s, 3 H, OCH₃); 3.24 (s, 3 H, OCH₃); 3.33 (m, 1 H, H-8); 3.62 (dd, 1 H, J = 4.5 Hz, 1.3 Hz, H-1); 3.90 (d, 1 H, J = 14.8 Hz, NCH₂); 5.26 (d, 1 H, J = 14.8 Hz, NCH₂); 7.17–7.89 (m, 10 H, H_{arom}).

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endo-Acetal **6b**; yield: 0.69 g (83 %); mp 157 °C. C₂₂H₂₅NO₅S calc. 63.59 H 6.07 N 3.37 (415.5) found 63.64 6.02 3.42

MS: m/z = 415.

IR (KBr): v = 1690 (amide); 1510, 1370, 960, 940 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.63–2.15 (m, 4 H, H-7, H-5); 2.79 (m, 1 H, H-4); 3.10 (s, 6 H, OCH₃); 3.39 (ddd, 1 H, J = 10 Hz, 6.9 Hz, 1.7 Hz, H-8); 3.56 (t, 1 H, J = 6 Hz, H-1); 3.85 (d, 1 H, J = 14.8 Hz, CH₂N); 5.36 (d, 1 H, J = 14.8 Hz, CH₂N); 7.3–8.0 (m, 10 H, H_{arom}).

2-Aza-2-benzyl-6,6-dimethoxybicyclo[2.2.2]octane-3-one (7):

To a stirred solution of the diastereomeric mixture of 6a/6b (0.48 g, 1 mmol) in THF/MeOH (1:2, 30 mL) is added 5% sodium amalgam (Na: 3.3 g, 5 mmol) in portions over 2 h. Stirring is continued overnight. The solution is filtered and evaporated to dryness. The residue is dissolved in CH₂Cl₂(30 mL). The organic layer is washed with sat. aq. NH₄Cl solution (2 × 20 mL), then with water (50 mL), dried (Na₂SO₄) and evaporated. The resultant colorless oil is distilled on a Kugelrohr apparatus (Büchi) (150–250 °C, 10^{-3} mbar) to give pure product 7; yield: 0.2 g (73%). The oil solidifies after some days; mp 68°C.

C₁₆H₂₁NO₃ calc. C 69.79 H 7.69 N 5.09 (275.2) found 69.30 7.57 4.91

MS: m/z = 275.

IR (KBr): v = 2986, 1686 (amide), 1676 (amide), 1495, 1470, 1452, 1430 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.32-2.09 (m, 6 H, H-5, H-7, H-8); 2.68 (m, 1 H, H-4); 3.10 (s, 3 H, OCH₃); 3.14 (s, 3 H, OCH₃); 3.50 (m, 1 H, H-1); 3.90 (d, 1 H, J = 14.8 Hz, NCH₂); 5.23 (d, 1 H, J = 14.8 Hz, NCH₂); 7.25-7.33 (m, 5 H, H_{arom}).

2-Aza-2-benzyl-6,6-dimethoxybicyclo[2.2.2]octane (8):

A mixture of 7 (2.75 g, 10 mmol) with LiAlH₄ (0.38 g, 11 mmol) in THF (30 mL) is stirred at room temperature for 30 min. A 2N NaOH solution (20 mL) is added, and the mixture extracted with CH_2Cl_2 (3×30 mL). The combined organic phase is washed with water (2×30 mL), dried (Na₂SO₄) and evaporated. The residual oil is distilled on a Kugelrohr apparatus (Büchi) (120–190 °C, 10^{-3} mbar) to give a colorless oil; yield: 2.4 g (92%).

C₁₆H₂₃NO₂ calc. C 73.53 H 8.87 N 5.36 (261.2) found 72.99 8.57 5.15

MS: m/z = 261.

IR (KBr): v = 2986, 1686, 1676, 1495, 1470 cm⁻¹.

¹H-NMR (CDCl₃): δ = 1.4–1.85 (m, 7 H); 2.4–2.53 (m, 1 H, H-4); 2.77–2.8 (m, 2 H, H-3); 3.11 (s, 3 H, OCH₃); 3.14 (s, 3 H, OCH₃); 3.77 (q, 2 H, J = 12.4 Hz, NCH₂); 7.23–7.39 (m, 5 H, H_{arom}).

2-Aza-6,6-dimethoxybicyclo[2.2.2]octane (9):

A solution of **8** (0.23 g, 0.88 mmol) in MeOH (50 mL) is hydrogenated over Pd –C (93.6 mg) for 4 h (H₂: 2 bar). The mixture is filtered and the solution is evaporated to dryness. The residual oil is distilled on a Kugelrohr apparatus (Büchi) (50 °C 10^{-3} mbar); yield: 90 mg (60%).

The oil solidifies after some days in the refrigerator to a waxy solid; mp 25-28°C.

C₉H₁₇NO₂ calc. C 63.13 H 10.01 N 8.18 (171.1) found 59.33 9.58 7.72

MS: m/z = 171.1 (M⁺, 22.9%), 156.0 (M⁺ -CH₃, 100%), 140.1 (M⁺ -OCH₃, 25.2%), 124.0 (73%), 111.1 (M⁺ -OCH₃ -HCN, 38%), 110.1 (73.5%), 81.1 (68.7%).

IR (KBr): $v = 3360, 2960, 1460, 1430 \,\mathrm{cm}^{-1}$.

 1 H-NMR (CDCl₃): $\delta = 1.52 - 1.87$ (m, 7 H); 2.66 (s, 1 H, NH); 2.87 – 2.90 (m, 3 H, H^{1.3}); 3.18 (s, 3 H, OCH₃); 3.21 (s, 3 H, OCH₃).

Note added in proof:

The recently reported ¹⁶ Diels-Alder reactions of 1,3-disulfonyl-2-pyridones represent the first examples of electrophilic [2+4]-cycloadditions in the 2-pyridone series.

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