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Suzuki-Miyaura cross-coupling of 3,4-disubstituted 5-bromoisoxazoles: An efficient access to trisubstituted isoxazoles



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Introduction

Isoxazole is a five-membered heteroaromatic compound containing a nitrogen-oxygen bond in its ring. The ring structure has been recognized as a privileged framework in medicinal chemistry, because the isoxazole ring itself exhibits critical pharmacophores for biological activity [1]. This high utility as pharmaceuticals has encouraged the development of various synthetic approaches towards functionalized isoxazoles [2]. Most of them, however, are based on conventional ring construction strategies that often suffer from insufficient regioselectivity and poor availability of linear components. On the other hand, the direct functionalization approach has not been established mainly due to its lability under basic conditions. To resolve this situation, we have developed direct functionalization methods such as 4-isoxazolyl anion species [3], gold(I)-catalyzed S_EAr type reactions,[4] and rhodium (III)-catalyzed C–H functionalizations [5]. Turning our attention to cross-coupling strategy which enables regioselective installation of substituents on heteroarenes using readily available building blocks [6]. 3,5-disubstituted isoxazoles were successfully synthesized via Suzuki-Miyaura coupling (Scheme 1a) [7]. However, the synthesis of trisubstituted isoxazoles via cross-coupling reactions at the C5 position has not been well developed [8,9]. Furthermore, Jurberg et al. clearly pointed out the difficulty in cross-coupling of

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ABSTRACT

The Suzuki-Miyaura cross-coupling of 3,4-disubstituted 5-bromoisoxazoles **1** at the C5 position has successfully proceeded in the presence of $Pd_2(dba)_3$ and $P(t-Bu)_3 \cdot HBF_4$ catalysts to give the corresponding trisubstituted isoxazoles **3** in good to high yields while suppressing the formation of ketone **4** as a byproduct. The use of bulky phosphine ligand $P(t-Bu)_3 \cdot HBF_4$ is essential for the current transformation, and the formation of ketone **4**, which was a major product in the previous report, was able to be suppressed under the current conditions.

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3,4-disubstituted 5-(pseudo)halogenated isoxazole [10]. According to their report, 3,4-disubstituted 5-bromoisoxazoles such as **1a** did not participate in Suzuki-Miyaura coupling, and desired trisubstituted isoxazoles were not obtained at all (Scheme 1b). Intrigued by their report, we decided to investigate the cross-coupling reactions of 5-bromoisoxazoles **1** with aryl boronic acids **2** in detail and succeeded in synthesizing trisubstituted isoxazoles **3** (Scheme 1c).

Results and discussion

Based on the previous report by Jurberg and coworkers, we began with an investigation on Suzuki-Miyaura cross-coupling of 4-benzyl-5-bromo-3-phenylisoxazole (1a) and phenylboronic acid (2a) (Table 1). First, we treated 1a and 2a in combination with Pd (PPh₃)₄ (10 mol%) and Na₂CO₃ (4.0 equiv.) in 1,4-dioxane at 100 °C for 16 h, as following the previously reported combination. As a result, the desired product 3a was not observed, though the ringopening product 4a was obtained in 22% yield along with 15% recovery of bromoisoxazole 1a. This ring-opening process was also observed by Jurberg et al. when they employed isoxazolyl triflate instead of **1a** [10]. The ketone **4a** was presumably given by N–O bond cleavage of the intermediate generated via oxidative addition, followed by hydrolysis and decarboxylation [11]. Surprisingly, by simply changing the base from Na₂CO₃ to K₃PO₄, the desired trisubstituted isoxazole 3a was produced in 67% yield (entry 2). In this case, ring-opening of the isoxazole ring also proceeded, resulting in the undesired ketone 4a in a non-negligible yield (24%). To suppress the generation of the ring-opening product 4a, condition screening was performed. The use of Pd(OAc)₂ without



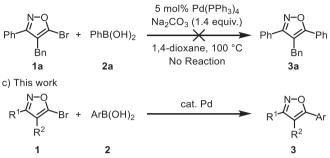
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a) Synthesis of 3,5-disubstituted isoxazoles (well established)

$$R^{1} \xrightarrow{M-O}_{H} X + R^{2}B(OH)_{2} \xrightarrow{\text{cat. Pd}} R^{1} \xrightarrow{N-O}_{H} R^{2}$$

X = Halogen
or OTf

b) Previous trial to synthtesis trisubstituted isoxazole (ref. 10)



Scheme 1. Suzuki-Miyaura cross-coupling of the isoxazoles at the C5 position.

ligands resulted in low conversion of 1a with a poor yield of 3a (entry 3, 13% yield). In contrast, the combination of Pd₂(dba)₃ and $P(t-Bu)_3$ ·HBF₄ [12] gave only traceable amount of ketone **4a**, though approximately 40% of substrate **1a** was recovered (entry 4). With this promising palladium salt and ligand system, various solvents were investigated (entries 5-7 and see supporting information, Table S1). As a result, the combination of K₂CO₃ and DMF was found to be the best condition, and the trisubstituted isoxazole 3a was obtained in 88% NMR yield (73% isolated yield) while suppressing the formation of ketone 4a in 5% yield. (entry 7). Next, various phosphine ligands were examined (entries 8-16). Among them, bulky ligands such as $P(o-Tol)_3$ was found to exhibit an efficient catalytic activity to afford **3a** in high yield (entry 10, 87%) yield) accompanied by 6% yield of ketone 4a.

Table 1

Optimization of the Suzuki-Miyaura cross-coupling reaction for the synthesis of 3a.ª

ough coupling reaction with phenylboronic acid 2a (Scheme 2). Interestingly, the corresponding ring-opening products 4 were not observed in all cases. With a benzyl group as R^2 , 3-(hetero)aryl isoxazoles **3 k-m** were obtained in good to high yields (64–77%). Not only those 3-aryl isoxazoles, but also the 3-alkyl isoxazole

3n was provided in 92% yield. Substituents on a benzyl group did not show any drastic effects, and the product 30 was obtained in 64% yield. Finally, methyl or *n*-propyl group on the C4-position also gave the products **3p** and **3q** in good yields (84% and 52%, respectively).

Conclusion

In this study, we succeeded in the Suzuki-Miyaura cross-coupling of 3,4-substituted 5-bromoisoxazoles 1 at C5 position by optimization of reaction conditions, giving trisubstituted isoxazoles **3** while suppressing the formation of ketone **4** as a byproduct. The use of bulky phosphine ligand $P(t-Bu)_3 \cdot HBF_4$ is essential for the

	N-O Ph	∽Br + PhB(OH)₂	[Pd], Lignad Base	Ph + Ph + Ph		
	''' T Bn 1a	2a	Solvent 100 °C, 16 h	Bn Ph 3a 4a		
Entry	Catalysts	Base	Solvent	Recovery of 1a (%) ^b	3a (%) ^b	4a (%) ^b
1	$Pd(PPh_3)_4$	Na ₂ CO ₃	1,4-dioxane	15	0	22
2	$Pd(PPh_3)_4$	K ₃ PO ₄	1,4-dioxane	0	67	24
3	$Pd(OAc)_2$	K_3PO_4	1,4-dioxane	62	13	8
4	$Pd_2(dba)_3$, $P(t-Bu)_3 \cdot HBF_4$	K ₃ PO ₄	1,4-dioxane	39	40	trace
5	$Pd_2(dba)_3$, $P(t-Bu)_3 \cdot HBF_4$	K ₃ PO ₄	DMF	0	77	2
6	$Pd_2(dba)_3$, $P(t-Bu)_3 \cdot HBF_4$	Na ₂ CO ₃	DMF	9	80	5
7	$Pd_2(dba)_3$, $P(t-Bu)_3 \cdot HBF_4$	K ₂ CO ₃	DMF	0	88 (73) ^c	5
8	Pd ₂ (dba) ₃ , PCy ₃	K_2CO_3	DMF	0	68	6
9	Pd ₂ (dba) ₃ , P(CH ₂ CH ₂ CH ₃) ₃	K_2CO_3	DMF	0	79	4
10	Pd ₂ (dba) ₃ , P(o-Tol) ₃	K ₂ CO ₃	DMF	0	87	6
11	Pd ₂ (dba) ₃ , P(OPh) ₃	K ₂ CO ₃	DMF	9	17	6
12	Pd ₂ (dba) ₃ , JohnPhos	K ₂ CO ₃	DMF	0	50	5
13	Pd ₂ (dba) ₃ , XPhos	K ₂ CO ₃	DMF	0	77	8
14	Pd ₂ (dba) ₃ , Xantphos	K ₂ CO ₃	DMF	0	75	5
15	Pd ₂ (dba) ₃ , DPEphos	K ₂ CO ₃	DMF	0	78	10
16	$Pd_2(dba)_3$, dppf	K ₂ CO ₃	DMF	0	57	17

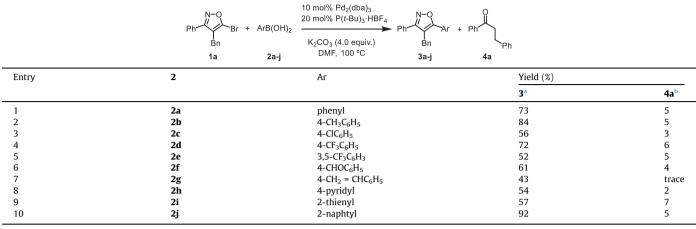
^a Reagents and conditions: 1a (0.10 mmol), 2a (0.12 mmol), Pd-catalyst (10 mol%), ligand (20 mol%), base (0.40 mmol), solvent (2 mL), 16 h. ^b NMR yield using 1,1dibromomethane as an internal standard. ^c Isolated yield in parenthesis. Ph = phenyl, Bn = benzyl, Ac = acetyl, dba = dibenzylideneacetone, Bu = butyl, THF = tetrahydrofuran, DMF = N, N-dimethylformaldehyde, Cy = cyclohexyl, Tol = tolyl, JohnPhos = (2-biphenylyl)di-tert-butylphosphine, XPhos = 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl, Xantphos = 4,5-bis(diphenylphosphino)-9,9-dimethylxanthene, DPEphos = bis[2-(diphenylphosphino)phenyl] ether, dppf = 1,1'-bis(diphenylphosphino)ferrocene.

examined the scope of Suzuki-Miyaura cross-coupling of 3,4-disubstituted 5-bromoisoxazole **1a** with various arvl boronic acids (Table 2). Aryl groups having both an electron donating group (entry 2) and electron withdrawing groups (entries 3-6) were introduced into C5 position of the isoxazole ring, and corresponding trisubstituted isoxazoles **3b-f** were provided in moderate to good vields (52–84%). The styrene derivative **3 g** was also obtained via the cross-coupling of **1a** with 4-vinylphenylboronic acid **2** g, though the yield was rather sluggish (entry 6). Heteroaryl boronic acids, such as 4-pyridylboronic acid **2** h and 2-thiopheneboronic acid 2i, were coupled with substrate 1a to afford 3 h and 3i in 54% and 57% yields, respectively (entries 7 and 8). Further, the cross-coupling using 2-naphthaleneboronic acid 2i afforded the corresponding coupling product 3j in 92% yield (entry 9). In all cases in Table 2, the formation of ketone **4a** via ring-opening of isoxazole ring was suppressed less than 7% yield. Next, the scope of 5-bromoisoxazole 1 was investigated thor-

With the best conditions in our hand (Table 1, entry 7), we next

Table 2

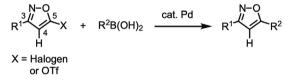
Suzuki-Miyaura cross-coupling of 3a with various aryl boronic acids.



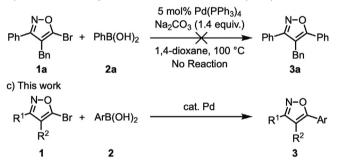
^a Isolated yield.

^b NMR yield using 1,1-dibromomethane as an internal standard.

a) Synthesis of 3,5-disubstituted isoxazoles (well established)



b) Previous trial to synthesis trisubstituted isoxazole (ref. 9)



Scheme 2. Suzuki-Miyaura cross-coupling of various 5- bromoisoxazoles **3** with phenylboronic acid.

current transformation. We are now in a position to synthesize a variety of trisubstituted isoxazoles, some of which are difficult to synthesize via the previously known methods. Further studies on cross-coupling for modular synthesis of multi-functionalized isoxazoles are ongoing in our laboratory.

Declaration of Competing Interest

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: Taiki Morita reports financial support was provided by Japan Society for the Promotion of Science (JSPS). Hiroyuki Nakamura reports a relationship with Japan Society for the Promotion of Science (JSPS) that includes: funding grants. Hiroyuki Nakamura reports a relationship with AMED that includes: funding grants.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.tetlet.2021.153185.

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