ChemComm

COMMUNICATION



Cite this: Chem. Commun., 2014, 50, 15067

Received 3rd July 2014, Accepted 8th October 2014

DOI: 10.1039/c4cc05085a

www.rsc.org/chemcomm

Synthesis aided structural determination of amyloid- $\beta(1-15)$ glycopeptides, new biomarkers for Alzheimer's disease[†]

Peng Wang, a Jonas Nilsson, b Gunnar Brinkmalm, c Göran Larson*b and Xuefei Huang*a

Unique tyrosine glycosylated amyloid- $\beta(1-15)$ glycopeptides were synthesized with well-defined stereochemistry at the glycosidic linkages. Aided by these glycopeptides and tandem mass spectrometry analysis, the naturally existing amyloid- β glycopeptides, isolated from Alzheimer's disease patients, were determined to contain an α -linked *N*-acetyl galactosamine at the modified tyrosine 10 residue. Glycosylation can significantly impact the properties of amyloid- β as the glycopeptide has much lower affinity for Cu⁺ ions.

Alzheimer's disease (AD) is the most common form of dementia, which is characterized by the formation of amyloid plaques composed of aggregates of amyloid- β (A β) polypeptides. AD is a serious concern for modern society as it is estimated that 5.4 million Americans had AD resulting in a cumulative cost of approximately 200 billion dollars in 2013.¹ One of the major challenges in AD treatment is accurate diagnosis at an early stage. A β and tau proteins have been widely viewed as biomarkers for AD.² However, they are of limited utility in diagnosis at the presymptomatic stage. There is an urgent need to identify AD biomarkers with higher sensitivity and specificity before cognitive decline appears.³ Recently, a series of novel Aß glycopeptides, corresponding to the N-terminal fragments (A β 1–X, X ranging from 15 to 20 residues) of A β , were isolated from the cerebrospinal fluid of patients.⁴ It was found that these Aβ glycopeptides are present in AD patients at markedly higher levels compared to those from non-AD control patients.⁴ Moreover, cognitively impaired patients considered as non-AD patients due to below threshold $A\beta(1-42)$ levels actually exhibited $A\beta$ glycopeptide profiles similar to those of AD patients, which may have indicated

early AD development. Therefore, the A β glycopeptides can be potential new biomarkers to facilitate early AD diagnosis.

The structures of $A\beta$ glycopeptides were analyzed by mass spectrometry (MS).⁴ While glycoproteins and glycopeptides with a mammalian origin commonly bear carbohydrate chains on serines/threonines (O-linked glycans) or on asparagine residues (*N*-linked glycans),^{5,6} A β glycopeptides contain glycosylation on tyrosine 10 (Tyr10).⁴ Although tyrosine glycosylations have been found with a glucose- α (1-*O*)Tyr or a galactose- β (1-*O*)Tyr linkage,⁷⁻⁹ the carbohydrates on A β glycopeptides were determined to have a core N-acetyl hexosamine (HexNAc) moiety linked to the hydroxyl group of Tyr. The HexNAc was further extended with one hexose and two or three N-acetylneuraminic acid residues (Neu5Ac). In analogy with other O-linked glycans, the core HexNAc was proposed to be N-acetyl galactosamine (GalNAc). An ambiguity in the structure is that the stereochemistry of the HexNAc-Tyr linkage could not be defined by the LC-MS/MS analysis alone. The uncertainty in structure has hindered the understanding of its function as well as the efforts in generating glycopeptide specific monoclonal antibodies for its detection.

To enable structural determination, herein, we report the total synthesis of $A\beta(1-15)$ glycopeptides bearing GalNAc on Tyr10 with well-defined stereochemistry, which joins the few examples of chemical synthesis of *O*-glycosyl tyrosine amino acid¹⁰⁻¹³ and glycopeptide¹⁴ reported to date. Aided by the synthetic $A\beta(1-15)$ glycopeptides, the HexNAc in the natural glycopeptide from AD patients was determined to be α -linked GalNAc through MS. Glycosylation of Tyr10 can significantly change the properties of the A β peptide as evident from the much reduced affinity of the glycopeptide for Cu⁺ ions.

To access the A β glycopeptide, we began with the preparation of the glycosylated Tyr building block from the D-galactosamine hydrochloride **1**. As stereochemistry at the glycosyl linkage is not known, we aimed for both α and β anomers. To facilitate the formation of the α glycosyl linkage, the amine moiety of galactosamine **1** was converted to azide through the copper catalyzed diazo transfer reaction (Scheme 1a).¹⁵ Global acetylation followed by selective removal of the anomeric acetate (Ac) and addition of



View Article Online

^a Department of Chemistry, Michigan State University, 578 S. Shaw Lane,

East Lansing, Michigan 48824, USA. E-mail: xuefei@chemistry.msu.edu

^b Department of Clinical Chemistry and Transfusion Medicine,

Sahlgrenska Academy at the University of Gothenburg, Sweden. E-mail: goran.larson@clinchem.gu.se

^c Department of Psychiatry and Neurochemistry,

Sahlgrenska Academy at the University of Gothenburg, Sweden

[†] Electronic supplementary information (ESI) available: Synthesis procedures, characterization data for all new compounds and supplementary figures. See DOI: 10.1039/c4cc05085a



Scheme 1 Reagents and conditions: (a) (i) triflic azide, K_2CO_3 , $CuSO_4$, H_2O /toluene/MeOH, r.t.; (ii) Ac₂O, pyridine, DMAP, 0 °C to r.t., 71% for 2 steps; (b) hydrazine acetate, DMF, r.t., 82%; (c) trichloroacetonitrile, K_2CO_3 , DCM, r.t., 75%; (d) **3**, TMSOTf, 4 Å MS, DCM/Et₂O (1:1), -30 °C (α only); (e) Zn, Ac₂O, AcOH, THF, r.t., 56% for 2 steps; (f) H_2 , Pd/C, AcOH, DCM/MeOH, r.t., 94%. (g) (i) H_2 , Pd/C, *p*-toluenesulfonic acid monohydrate, THF, r.t., ii. 2,2,2-trichloroethyl chloroformate, TEA, THF, 0 °C, 80% for 2 steps; (h) *p*-toluenethiol, BF₃·Et₂O, DCM, r.t., 85%; (i) NIS, trifluoromethanesulfonic acid, **3**, 4 Å MS, DCM, -20 °C; (j) Zn, Ac₂O, AcOH, THF, r.t., 52% for 2 steps (**7**: 26%; **4**: 26%); (k) H_2 , Pd/C, AcOH, DCM/MeOH, r.t., 90%.

trichloroacetonitrile led to the trichloroacetimidate donor 2. Trimethylsilyl triflate (TMSOTf) promoted glycosylation of Fmoc protected tyrosine 3 by donor 2 forming the glycosylated tyrosine, which was then reduced with simultaneous acetylation leading to 4 as a single stereoisomeric product in 56% yield for the two steps. NMR analysis of 4 (${}^{3}J_{H1-H2} = 3.0$ Hz) suggested that the newly formed glycosidic linkage was α . Catalytic hydrogenolysis of 4 generated tyrosine carboxylic acid 5, which was ready for glycopeptide synthesis.

In order to form β -linked glycosyl tyrosine, trichloroethyl carbamate (Troc) protected galactosamine thioglycosyl donor **6** was prepared from D-galactosamine hydrochloride **1** (Scheme 1b). Using *N*-iodosuccinimide (NIS) and trifluoromethane sulfonic acid as the promoters, donor **6** was coupled with tyrosine acceptor **3**, which was followed by reduction of the 2-*N*-trichlorethoxycarbonyl (Troc) group and acetylation leading to two anomeric isomers 7 and **4** in a **1** : **1** ratio. 7 and **4** were separated by silica gel chromatography and 7 was determined to be the β anomer (${}^{3}J_{H1-H2} = 8.5 \text{ Hz}$) by 1 H-NMR spectroscopy. The formation of a large quantity of **4** in the presence of the 2-*N*-Troc moiety capable of neighboring group participation in donor **6** was presumably due to the higher thermodynamic stability of **4** resulting from the anomeric effect. 7 was subjected to catalytic hydrogenolysis yielding the β -linked tyrosine carboxylic acid **8** (Scheme 1c).

With the glycosyl tyrosines **5** and **8** in hand, glycopeptide syntheses were performed using the solid phase method on 2-chlorotrityl resin (Scheme 2). Iterative peptide coupling using



Scheme 2 Synthesis of A β glycopeptides **9** (3%), **10** (5%) and peptide **11** (9%).

Fmoc protected amino acid building blocks including the glycosylated tyrosines followed by Fmoc deprotection produced glycopeptides, which were cleaved off the resin using a cocktail of trifluoroacetic acid (TFA), water and triisopropylsilane (TIPS). Treatment of the resulting peptides with base removed all *O*-acetates on GalNAc and the N-terminal Fmoc. Subsequent HPLC purification generated pure $A\beta(1-15)$ glycopeptides **9** and **10** containing α and β linked GalNAc, respectively. In a similar manner, the unglycosylated $A\beta(1-15)$ **11** was synthesized.

The synthetic glycopeptides provide valuable standards to establish MS methods for stereochemical assignment. A β (1–15) glycopeptides 9 and 10 were analyzed by nano-flow liquid chromatography electrospray ionization tandem mass spectrometry (LC-MS/MS or MS2) using a hybrid linear ion trap-Fourier transform ion cyclotron resonance mass spectrometer. Collision induced dissociation (CID) was utilized to achieve glycosidic and partial peptide fragmentation and electron capture dissociation (ECD) was utilized for determining the amino acid position of the glycan. The CID-MS/MS (MS2) spectra were collected in the linear ion trap (Fig. 1A for glycopeptide 9 and Fig. 1B for glycopeptide 10) resulting from the $[M + 4H]^{4+}$ precursors (Fig. S1, ESI⁺). The theoretical mass (monoisotopic) for the $[M + 4H]^{4+}$ precursor of compounds 9 and 10 is m/z 508.2213. The measured masses for compound 9 and compound 10 were m/z 508.2207 (1.14 ppm off) and 508.2210 (0.63 ppm off), respectively. Both compounds lost GalNAc + H⁺ from the $[M + 3H]^{3+}$ peptide ion (*m*/*z* 609.6). Although qualitatively the spectra were similar, quantitative analysis revealed key differences distinguishing the two structures. For 9, the $[M + 4H]^{4+}$ peptide ion (*m*/*z* 457.7), resulting from the neutral loss of GalNAc, had lower intensity than a number of GalNAc substituted b- and y-ions (e.g., $[y_{14} + GalNAc]^{2+}$ at m/z 479.8) (Fig. 1A). The reverse was observed for 10 where the $[M + 4H]^{4+}$ ion was more prominent compared to the $[y_{14} + GalNAc]^{2+}$ ion (Fig. 1B). The ratios for the m/z 457.7 and 479.8 ions were found to be 0.5 \pm 0.1 and 2.6 \pm 0.7 (m \pm SD) for 9 and 10 respectively.



Fig. 1 CID-MS2 of A β (1–15) glycopeptides **9** (A) and **10** (B); and CID-MS3 of HexNAc substituted A β (1–15) from human cerebrospinal fluid (C). The spectra are intensity zooms of the corresponding CID-MS2 and CID-MS3 spectra (see Fig. S1 and S2, ESI†). Squares symbolize free or glycopeptide-linked HexNAc and horizontal lines represent the peptide ¹DAEFRHDSGYEVHHQ.¹⁵

Thus, GalNAc was more easily cleaved from the Tyr residue during CID when it was attached to the peptide in β -configuration (glycopeptide **10**) compared to the α -configuration (glycopeptide **9**). This is most likely due to the higher stability of the glycosidic linkage in **9** as a result of the anomeric effect. Therefore, MS can be used to reliably differentiate these two isomeric glycopeptides.

The native $A\beta(1-15)$ glycopeptide isolated from human cerebrospinal fluid (CSF) was examined next by MS. A β peptides and glycopeptides were immunopurified using the 6E10 antibody and any terminal Neu5Ac residues were selectively hydrolyzed using 0.1 M formic acid at 353 K for 30 minutes resulting in A β glycopeptides rich in de-sialylated HexHexNAc (Hex denotes a hexose).¹⁶ The [M + 4H]⁴⁺ precursor of A $\beta(1-15)$ + HexHexNAc was observed at m/z 548.7350 (Fig. S2A, ESI[†]) and CID-MS2 resulted in an abundant neutral loss of Hex into A $\beta(1-15)$ + HexNAc at m/z 508.4 (Fig. S2B, ESI[†]) isomeric to **9** and **10**. Further consecutive fragmentation (CID-MS3) led to a prominent loss of HexNAc + H⁺ and a spectrum closely matching that of CID-MS2 of **9** (Fig. 1C). Specifically, CID-MS3 spectra of the native glycopeptide possessed a more abundant $[y14 + \text{GalNAc}]^{2+}$ ion (m/z 479.8) and a less abundant $[M + 4H]^{4+}$ peptide ion (m/z 457.7) with the intensity ratio of 0.4 ± 0.2 for the peak at m/z 457.7 over that at m/z 479.8. Thus, we conclude that the native A β (1–15) glycopeptides from CSF are composed of an α -linked HexNAc moiety, most likely an α -GalNAc moiety. An electron capture dissociation (ECD) MS spectrum of the $[M + 4H]^{4+}$ ion for compound **9** also confirmed the peptide sequence and pinpointed that the GalNAc residue was attached to Tyr10 (Fig. S3, ESI⁺).

A peptides can bind with metal ions, 17-19 which can facilitate nucleation contributing to plaque formation. At the same time, as ions such as free Cu⁺ are highly toxic to cells, it has been proposed that AB may play a neuroprotective role by scavenging metal ions.²⁰ Thus, the Cu⁺ binding abilities of A β may influence plaque biology. This prompted us to analyze the affinity of glycopeptide 9 with Cu⁺ ions through a competitive binding assay.²¹ Cu(NO₃)₂ was reduced by sodium ascorbate to Cu⁺, which formed an orange complex with disodium bathocuproinedisulfonic acid (BC) with an absorbance maximum at 483 nm. When increasing amounts of Aß glycopeptide 9 were added to a solution of the BC-Cu complex, it competed with BC for Cu⁺ binding thus serially reducing the absorbance at 483 nm (Fig. 2). Based on absorbance changes, the dissociation constant of the [Cu⁺–9] complex was calculated to be $1.69 \pm 0.84 \times 10^{-14}$ M. In comparison, the dissociation constant of the unglycosylated Aß peptide 11 with Cu⁺ was measured to be 2.72 \pm 1.26 \times 10⁻¹⁵ M, which was similar to the reported value of A β (1–16) binding with Cu⁺.²¹ The Aβ peptide binds Cu⁺ in a linear bis-His geometry,²⁰ thus most likely the Cu⁺ ion is ligated to His13 and His14 in unglycosylated A β peptide **11**. The presence of a bulky GalNAc on Tyr10 in glycopeptide 9 could possibly sterically hinder Cu⁺ binding. The impact of reduced Cu⁺ affinity of the glycopeptide on plaque formation and toxicity will need to be established in vivo in the future.

In conclusion, we developed viable synthetic routes to the first synthesis of Tyr *O*-glycosylated $A\beta$ peptides. Aided by these welldefined synthetic samples and tandem MS analysis, we identified



Fig. 2 Competitive chelation of BC and A β glycopeptide **9** with Cu⁺. Increasing concentrations of glycopeptide **9** disrupted the [CuBC₂]³⁻ complex resulting in reduction of the absorbance at 483 nm.

that Tyr10 and O-glycan were most likely linked through an α -GalNAc linkage in natural A β glycopeptide fragments isolated from AD patients. Glycosylation could significantly impact the properties of the glycopeptide such as interactions with Cu⁺ ions. The determination of the glycopeptide structure can enable the monoclonal antibody generation study and lay the groundwork towards further understanding of their roles as biomarkers.

This work was supported by Michigan State University, the National Science Foundation (CHE 1111550), the National Institute of General Medical Sciences, NIH (R01GM072667) and the Swedish Research council (8266).

Notes and references

- 1 Alzheimer's Association 2013 Alzheimer's disease facts and figures, http://www.alz.org/downloads/facts_figures_2013.pdf, 2013.
- 2 B. Dubois, H. H. Feldman, C. Jacova, H. Hampel, J. L. Molinuevo, K. Blennow, S. T. DeKosky, S. Gauthier, D. Selkoe, R. Bateman, S. Cappa, S. Crutch, S. Engelborghs, G. B. Frisoni, N. C. Fox, D. Galasko, M. O. Habert, G. A. Jicha, A. Nordberg, F. Pasquier, G. Rabinovici, P. Robert, C. Rowe, S. Salloway, M. Sarazin, S. Epelbaum, L. C. de Souza, B. Vellas, P. J. Visser, L. Schneider, Y. Stern, P. Scheltens and J. L. Cummings, *Lancet Neurol.*, 2014, 13, 614–629.
- 3 K. Blennow, H. Hampel, M. Weiner and H. Zetterberg, Nat. Rev. Neurol., 2010, 6, 131-144.
- 4 A. Halim, G. Brinkmalm, U. Rüetschi, A. Westman-Brinkmalm, E. Portelius, H. Zetterberg, K. Blennow, G. Larson and J. Nilsson, *Proc. Natl. Acad. Sci. U. S. A.*, 2011, **108**, 11848–11853.

- 5 M. J. Grogan, M. R. Pratt, L. A. Marcaurelle and C. R. Bertozzi, *Annu. Rev. Biochem.*, 2002, **71**, 593–634.
- 6 R. A. Dwek, Chem. Rev., 1996, 96, 683-720.
- 7 K. Zarschler, B. Janesch, M. Pabst, F. Altmann, P. Messner and C. Schäffer, *Glycobiology*, 2010, **20**, 787–798.
- 8 P. Messner, R. Christian, C. Neuninger and G. Schulz, J. Bacteriol., 1995, 177, 2188–2193.
- 9 M. D. Alonso, J. Lomako, W. M. Lomako and W. J. Whelan, *FEBS Lett.*, 1994, 342, 38–42.
- 10 N. E. Fahmi, L. Dedkova, B. Wang, S. Golovine and S. M. Hecht, J. Am. Chem. Soc., 2007, 129, 3586–3597.
- 11 A. I. Khodair, G. A. Winterfeld and R. R. Schmidt, *Eur. J. Org. Chem.*, 2003, 1847–1852.
- 12 L. A. Salvador, M. Elofsson and J. Kihlberg, *Tetrahedron*, 1995, **51**, 5643–5656.
- 13 A. Vargas-Berenguel, M. Meldal, H. Paulsen, K. J. Jensen and K. Bock, J. Chem. Soc., Perkin Trans. 1, 1994, 3287–3294.
- 14 A. M. Jansson, K. J. Jensen, M. Meldal, J. Lomako, W. M. Lomako, C. E. Olsen and K. Bock, J. Chem. Soc., Perkin Trans. 1, 1996, 1001–1006.
- 15 P. B. Alper, S.-C. Hung and C.-H. Wong, *Tetrahedron Lett.*, 1996, 37, 6029–6032.
- 16 J. Nilsson, U. Rüetschi, A. Halim, C. Hesse, E. Carlsohn, G. Brinkmalm and G. Larson, *Nat. Methods*, 2009, 6, 809–811.
- 17 M. G. Savelieff, S. Lee, Y. Liu and M. H. Lim, ACS Chem. Biol., 2013, 8, 856–865.
- 18 P. Faller, C. Hureau and O. Berthoumieu, Inorg. Chem., 2013, 52, 12193-12206.
- 19 K. P. Kepp, Chem. Rev., 2012, 112, 5193-5239.
- 20 J. Shearer and V. A. Szalai, J. Am. Chem. Soc., 2008, 130, 17826-17835.
- 21 H. A. Feaga, R. C. Maduka, M. N. Foster and V. A. Szalai, *Inorg. Chem.*, 2011, **50**, 1614–1618.