

Erythrina Alkaloid Analogues as nAChR Antagonists-A Flexible Platform for Leads in Drug Discovery

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■ INTRODUCTION

nAChRs.

Extracts from the seeds of Erythrina plants have been utilized as hypnotics, sedatives, and arrow poison by indigenous South Americans for the past centuries. Many alkaloids of this natural product family possess a curare-like neuromuscular blocking effect, and comprehensive studies in the early 1940s concluded that a majority of the isolated Erythrinanes exert their principal pharmacological activity via the central nervous system by acting as competitive antagonists of nicotinic acetylcholine receptors (nAChRs), a heterogeneous class of pentameric ligand-gated ion channels mediating fast cholinergic transmission.^{1–3}

(+)-Dihydro- β -erythroidine [DH β E (1), Figure 1B] was previously used in the treatment of Parkinson's disease to relieve tetanus and spastic disorders. As one of the most potent members of the Erythrina alkaloid family, DH β E (1) has become a key pharmacological tool in the nAChR field as it displays pronounced subtype-selectivity for β 2-subunit-containing nAChRs (e.g., $\alpha 4\beta 2$) over $\beta 4$ -containing (e.g., $\alpha 3\beta 4$) and other nAChR subtypes (e.g., α 7).^{4,5,6a-c} 1 and the other Erythrina alkaloids act as competitive antagonists at the nAChR, and their binding modes to the receptors have been elucidated by a co-crystal structure of 1 and the ACh-binding protein (AChBP), a mollusk protein homologous to the extracellular domain of the nAChR.60

The Erythrina alkaloids all possess a structurally rigid spiroamine scaffold, where the A-, B-, and C-rings are generally decorated in an oxidatively similar manner-Erythrivarine B (2) represents an exception to this. Most subcategories of Erythrina alkaloids distinguish themselves from each other in the nature of their D-ring, where lactones, aryls, and heteroaryls comprise the majority of the reported moieties (Figure 1B).⁷⁻

The structural complexity and pharmacological profile of Erythrina alkaloids have made them popular targets for total synthesis, resulting in several notable approaches.^{10a-m,11} However, only a few of these efforts have produced optically pure natural products.^{7,10a,11}

(+)-Cocculidine (R=Me)

Moreover, the oxidative diversity found in the D-ring of Erythrinanes has also made a unifying synthetic approach to various subcategories very challenging. In the present work,^{12,14,16} we expand upon our previously developed synthetic route to the Erythrina alkaloids and demonstrate that our divergent synthetic strategy allows access to the lactonic and aromatic Erythrina natural products for the first time.^{10k,l}

We also characterize the binding and functional properties of this series of Erythrina alkaloids at nAChRs. In a previous investigation, we found that the AB-ring system of DH β E (1) is bound deep in the nAChR-binding pocket with the lactonic D-ring protruding from it (Figure 2A).^{6c} In the present study, we take advantage of our recently disclosed total synthesis of 1 to access a library of ligands.^{10k,1} These ligands maintain the structural features of 1 embedded in the binding pocket, while allowing for easy manipulation of the part protruding from the binding pocket (Figure 2B). The MeO-appendage on the Aring established an interaction with a water molecule in the binding pocket, and we also wanted to probe the importance of this interaction by varying the stereochemical orientation at

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Figure 1. (A) General overview of a divergent approach to various Erythrina alkaloid subcategories and (B) representative members of the alkaloid family.



Figure 2. (A) DH β E (1) bound in the nAChR model.^{6b} (B) Outline of the contributions of different parts of DH β E (1) in its nAChR binding and the targeted library, wherein R and the stereochemistry at C3 varied.

this carbon–given that our synthetic route provides access to both diastereisomers of the DH β E-scaffold.

RESULTS AND DISCUSSION

Our synthesis commenced with revisiting our total synthesis of **1** (see Scheme 1). Asymmetric allylic allylation using Trost ligand 7 delivered allyl prolinone **8**.^{12a,b} The terminal double bond was then oxidatively cleaved, and the product isolated as the corresponding dimethyl acetal **9**.^{13,14a,b} Methenylation of the ketone moiety proved problematic, and all attempts at using titanium carbenoids^{15a-g} or addition of organometallic sources of methylene^{16a,b} failed. Wittig procedures using potassium *tert*-butoxide furnished the desired olefin **10**; however, the reaction suffered from solubility issues and the yield was consistently around 50%.¹⁷ Both aspects were resolved by forming the ylide in a separate flask and then transferring the said ylide-solution to the neat ketone,¹⁸ which

resulted in faster reaction time and increased scalability and yield. This is a notable improvement to the previously reported total synthesis.

Subsequent allylation of the C3 with (+)-allylboronic acid pinanediol ester 11¹⁹ generated the desired diastereomer 12 in total 60% yield. Methylation and ring closing metathesis^{20a,b} furnished the bicyclic ester 15, and this compound could be further enantioenriched through the means of chiral chromatography. The final carbons of the C-ring were introduced through a one pot deprotection-reductive amination.^{20,28-31} The tricyclic ring system could then be assembled by refluxing bis-ester 17 with potassium tertbutoxide, furnishing 18 in 98% yield. Further functional group manipulation produced the silyl-protected triflate 19 in 73% (see Scheme 2). Installation of the final two carbons of 1 proved problematic as the triflate 19 rapidly decomposed under basic coupling conditions.^{21a-d} Eventually, we found that a base-free decarboxylative α -Cyanation formed the desired carbon-carbon bond, and the process was ultimately telescoped to 1, using HCl.²

Total Synthesis of Aromatic Erythrinanes (+)-Cocculine and (+)-Cocculidine. We then turned our attention to the aromatic Erythrinanes looking to utilize intermediate **18** to access representative members from this structural class. The required carbons of the D-ring could be introduced *via* a Michael addition with methyl-vinyl-ketone **22** in quantitative yield (see Scheme 2).

Decarboxylation-condensation of 23 with potassium hydroxide in two iterations delivered enone 24/24' as a 1.8:1 mixture of C-12 epimers. Enone 24 could later be

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Scheme 1. Previous Synthesis of Intermediate 18^{10k,l,a}



^aReagents and conditions: (a) **6**, 7, $[PdCl(allyl)]_2$, K_2CO_3 , DMF, -40-0 °C. (b) OsO₄, NMO, THF:H₂O, rt, 12 h, then NaIO₄, DCM, rt, 4 h. (c) CH(OMe)_3, CSA, toluene, rt, 2 h. (d) MePPh₃Br, NaH, toluene, 90 °C, 10 h, then **9**. (e) **11**, HCl, DCM: H₂O, rt, 8 h, d.r. 1.8:1. (f) Me₃OBF₄, proton-sponge, DCM, rt, 3 h. (g) **14**, toluene, 80 °C. (h) TFA:DCM, rt, 30 min, then **16**, AcOH, NaBH₃CN, THF:MeOH, rt. (i) KOtBu, toluene, 95 °C, 1 h.





^{*a*}Reagents and conditions: (a) **20** (3 equiv), $[PdCl(allyl)]_2$ (0.2 equiv), S-Phos (0.4 equiv), mesitylene, 110 °C, 30 min, then HCl (aq), MeOH, 85 °C, 30 min, 54%. (b) Et₃N (5 equiv), **22** (5 equiv), MeOH, 70 °C, 14 h, >99%. (c) KOH (5 equiv), water, 105 °C, 2 h, then KOH (15 equiv), 105 °C, 2h, 78%. (d) HCl (6.3 equiv), CHCl₃, room temperature, 24 h, 74%. (e) CuBr₂ (4 equiv), MeCN, 40 °C, 5 h, 60%. (f) CuBr₂ (2 equiv), CH(OMe)₃ (15 equiv), methanol, 80 °C, 14 h, then BF₃·OEt₂ (2 equiv), DCM, 0 °C-rt, 2 h, 62%. S-Phos = 2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl. DCM = dichloromethane.

obtained as a single epimer in 56% yield following treatment with HCl. The final oxidation of the D-ring proved difficult (see the Supporting Information) but was ultimately accomplished using stoichiometric amounts of CuBr_2 ,²³ delivering the phenolic natural product (+)-Cocculine **25** in 60% yield. Standard O-methylation procedures (methyl iodide or diazomethane) did not produce (+) Cocculidine (**3**), but upon treatment of enone **24** with CuBr_2 in the presence of trimethyl orthoformate, the bromo-ketal **26** (tentatively assigned) was formed as the major component. Subsequent Lewis acid-mediated elimination delivered (+)-Cocculidine (**3**) in good 62% yield from **24** (see the Supporting Information for a complete list of attempted oxidations). This approach represents the first synthesis of (+)-Cocculine (25) as well as the shortest and first asymmetric synthesis of (+)-Cocculidine (3), both in 12 steps, respectively.

Synthesis of Erythrina Library. Using key intermediate **18** as the starting point, we set out to access a library of Erythrina alkaloid analogues (see Scheme 3). To be able to investigate the significance of the stereochemistry of the C3 methyl ether, both epimers **18** and **epi-18** were decarboxylated to ketones **29** and **epi-29** using 4 and 3 molar aqueous HCl, respectively. Interestingly, when exposed to 4–6N HCl, the C3-epimer underwent O-demethylation and subsequent lactonization to form the tetracyclic lactone **31**. Triflation of **29** and **epi-29** with KHMDS and bis-triflimide²⁴ generated the

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Scheme 3. Synthesis of Erythrina Analogues^a



"Reagents and conditions: (a) Hydrazine-hydrate (10 equiv), 1,4-dioxane, 100 °C, 72 h, 30%. (b) Thiourea (2.6 equiv), KOtBu (2.6 equiv), MeOH, 90 °C, 12 h, 5%. (c) HCl (4N), 110 °C, 3 h, 68%. (d) HCl (3N), 110 °C, 3 h, 56%. (e) HCl (6M), 115 °C, 16 h, 30%. (f) KHMDS (2 equiv), PhN(OTf)₂ (1.8 equiv), THF, -78-0 °C, 3.5 h, 68%, resp. 56%. (g) Pd(dppf)Cl₂ (0.1 equiv), R-B(OH)₂ (1.7 equiv), NaOH (1.8 equiv), DME, 85 °C, 75 min. DME = dimethoxyethane. KHMDS = potassium *bis*(trimethylsilyl)amide. THF = tetrahydrofuran.

Scheme 4. Structures of Ligands in the Erythrina Library Accessed in Scheme 3



corresponding triflates **30** and **epi-30**, which set the stage for the assembly of the final carbon–carbon bond—consequently introducing the D-ring as a non-fused, appended ring system. Following a standard Suzuki coupling²⁴ of the triflates, a library containing over 22 unnatural Erythrina alkaloids bearing various functionalities in their D-ring was synthesized.

With the exception of 33 and 38, both epimers were obtained from the analogues 32-43. Furthermore, 18 was condensed with hydrazine and thiourea to give heterocyclic compounds 27 and 28, respectively. See Scheme 4 for the structure of all the ligands in the Erythrina library.

Pharmacological Characterization of the Erythrina Alkaloids at nAChRs. The binding properties of the Erythrina alkaloids as nAChR ligands were determined at two β 2-containing ($\alpha 4\beta 2$ and $\alpha 6/\alpha 3\beta 2\beta 3$ V9'S, a surrogate for the $\alpha 6\beta 2\beta 3$ receptor) and two β 4-containing ($\alpha 3\beta 4$ and $\alpha 4\beta 4$) subtypes in a [³H]epibatidine competition binding assay, as previously described (see Table 1).^{25a,b} The functional properties of the compounds as nAChR antagonists were investigated at the $\alpha 4\beta 2$ and $\alpha 3\beta 4$ receptors in a Ca²⁺ imaging assay using the Ca2+ fluorophore Fluo-4.^{25b} Full details on all of the synthesized compounds are given in Tables 1 and 2.

In agreement with the literature, 1 was found to exhibit pronounced β 2-over- β 4 selectivity at the nAChRs, both in terms of its binding affinities and functional antagonist potencies (see Table 1). In striking contrast and highly

| Table 1. Binding Affinities of the <i>Erythrina</i> Alkaloid Analogues at $h\alpha 4\beta 2$ -, $h\alpha 6/\alpha 3\beta 2\beta 3^{V9S}$ -, $r\alpha 3\beta 4$ -, and $r\alpha 3\beta 4$ -HEK293 C | Cell |
|--|------|
| Membranes in the [³ H]epibatidine Competition Binding Assay ^{<i>a</i>} | |

| | $h\alpha 4\beta 2$ | $h\alpha 6/\alpha 3\beta 2\beta 3^{V9'S}$ | $r\alpha 3\beta 4$ | rα4β4 |
|----------------------------|--------------------------------------|---|--------------------------------------|--------------------------------------|
| | $K_i [pK_i \pm \text{S.E.M.}]^{(n)}$ | $K_i [pK_i \pm \text{S.E.M.}]^{(n)}$ | $K_i [pK_i \pm \text{S.E.M.}]^{(n)}$ | $K_i [pK_i \pm \text{S.E.M.}]^{(n)}$ |
| $(+)$ -DH β E (1) | $0.27 \ [6.57 \pm 0.03]^{(4)}$ | $0.60 \ [6.22 \pm 0.10]^{(4)}$ | ~50 [~4.3] ^{(3),c} | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| $(-)$ -DH β E $(1')$ | ~30 [~4.5] ^{(3),c} | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 3 | $0.0049 \ [8.31 \pm 0.08]^{(3)}$ | $0.075 \ [7.13 \pm 0.01]^{(3)}$ | $\sim 20 \ [\sim 4.7]^{(3),b}$ | $3.4 [5.47 \pm 0.08]^{(3)}$ |
| 18 | $0.60 \ [6.22 \pm 0.05]^{(4)}$ | $1.6 [5.80 \pm 0.12]^{(3)}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| epi-18 | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 24 | $0.59 [6.23 \pm 0.03]^{(3)}$ | $4.0 [5.39 \pm 0.02]^{(3)}$ | $IC_{50} > 100^{(3),b}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| 25 | $0.0018 \ [8.75 \pm 0.13]^{(3)}$ | $0.042 \ [7.37 \pm 0.11]^{(3)}$ | ~10 [~5.0] ^{(3),c} | $1.2 \ [5.91 \pm 0.05]^{(3)}$ |
| 27 | ~10 [~5.0] ^{(3),c} | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 28 | ~20 [~4.8] ^{(3),c} | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 29 | 4.3 $[5.37 \pm 0.09]^{(4)}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| epi-29 | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 31 | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 32 | $4.8 \ [5.32 \pm 0.06]^{(4)}$ | 4.6 $[5.34 \pm 0.10]^{(3)}$ | ~50 [~4.3] ^{(3),c} | ~50 [~4.3] ^{(3),c} |
| epi-32 | ~10 [~5.0] ^{(3),c} | 7.0 $[5.16 \pm 0.09]^{(3)}$ | ~50 [~4.3] ^{(3),c} | $IC_{50} > 100^{(3),b}$ |
| 33 | $2.0 \ [5.70 \pm 0.04]^{(3)}$ | $3.4 \ [5.47 \pm 0.10]^{(3)}$ | ~50 [~4.3] ^{(3),c} | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| 34 | $3.4 [5.46 \pm 0.07]^{(4)}$ | 2.9 $[5.54 \pm 0.10]^{(3)}$ | ~50 [~4.3] ^{(3),c} | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| epi-34 | $\sim 10 \ [\sim 5.0]^{(3),c}$ | $\sim 10 \ [\sim 5.0]^{(3),c}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 35 | $0.39 \ [6.40 \pm 0.04]^{(3)}$ | 1.9 $[5.71 \pm 0.13]^{(3)}$ | ~50 [~4.3] ^{(3),c} | 4.9 $[5.31 \pm 0.07]^{(3)}$ |
| epi-35 | $4.8 \ [5.32 \pm 0.10]^{(3)}$ | 7.1 $[5.15 \pm 0.08]^{(3)}$ | ~50 [~4.3] ^{(3),c} | $IC_{50} > 100^{(3),b}$ |
| 36 | $1.2 \ [5.92 \pm 0.02]^{(3)}$ | $1.8 \ [5.75 \pm 0.09]^{(4)}$ | ~50 [~4.3] ^{(3),c} | $\sim 30 \ [\sim 4.5]^{(3),c}$ |
| epi-36 | ~30 [~4.5] ^{(3),c} | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 37 | $0.22 \ [6.66 \pm 0.03]^{(4)}$ | $0.88 \ [6.06 \pm 0.06]^{(4)}$ | ~30 [~4.5] ^{(3),c} | $2.9 \ [5.54 \pm 0.09]^{(n)}$ |
| epi-37 | $5.5 [5.26 \pm 0.05]^{(3)}$ | 4.2 $[5.38 \pm 0.07]^{(3)}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 38 | $1.3 [5.90 \pm 0.05]^{(3)}$ | $0.96 [6.01 \pm 0.11]^{(3)}$ | $\sim 50 \ [\sim 4.3]^{(3),c}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| 39 | $0.87 [6.06 \pm 0.06]^{(4)}$ | $0.70 \ [6.16 \pm 0.11]^{(3)}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ | $\sim 50 \ [\sim 4.3]^{(3),c}$ |
| epi-39 | $0.63 \ [6.20 \pm 0.01]^{(3)}$ | $0.33 [6.48 \pm 0.08]^{(3)}$ | $\sim 10 \ [\sim 5.0]^{(3),c}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| 40 | $1.2 [5.94 \pm 0.08]^{(4)}$ | $1.4 [5.86 \pm 0.09]^{(3)}$ | $\sim 50 [\sim 4.3]^{(3),c}$ | $\sim 10 \ [\sim 5.0]^{(3),c}$ |
| epi-40 | $2.8 [5.55 \pm 0.08]^{(3)}$ | $5.2 [5.29 \pm 0.06]^{(3)}$ | ~30 [~4.5] ^{(3),c} | $IC_{50} > 100^{(3),b}$ |
| 41 | $0.81 \ [6.09 \pm 0.06]^{(3)}$ | $1.8 [5.75 \pm 0.08]^{(3)}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ | $\sim 10 \ [\sim 5.0]^{(3),c}$ |
| epi-41 | $6.5 [5.19 \pm 0.01]^{(3)}$ | $4.7 [5.33 \pm 0.05]^{(3)}$ | $\sim 30 \ [\sim 4.5]^{(3),c}$ | $IC_{50} > 100^{(3),b}$ |
| 42 | $3.0 [5.52 \pm 0.02]^{(4)}$ | $1.4 [5.87 \pm 0.12]^{(2)}$ | $IC_{50} > 100^{(3),b}$ | $\sim 50 \ [\sim 4.3]^{(3),c}$ |
| epi-42 | $\sim 10 \ [\sim 5.0]^{(3),c}$ | $6.0 [5.22 \pm 0.01]^{(3)}$ | $IC_{50} > 100^{(3),b}$ | $IC_{50} > 100^{(3),b}$ |
| 43 | $0.83 \ [6.08 \pm 0.04]^{(3)}$ | $1.1 [5.97 \pm 0.06]^{(3)}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ |
| epi-43 | 1.0 $[5.98 \pm 0.03]^{(3)}$ | $0.23 \ [6.63 \pm 0.10]^{(3)}$ | $\sim 10 \ [\sim 5.0]^{(3),c}$ | $\sim 50 \ [\sim 4.3]^{(3),c}$ |

^{*a*}The binding affinities of the compounds are given as K_i values in μ M [with $pK_i \pm$ standard error of mean (S.E.M.) in brackets] with the number of experiments (*n*) indicated in superscript. ^{*b*}At 100 μ M, the compound mediated less than 50% inhibition of specific [³H]epibatidine binding (*i.e.*, IC₅₀ > 100 μ M). ^{*c*}The concentration–inhibition relationship for the compound was not complete at 100 μ M. Thus, the K_i value for the compound was calculated based on an IC₅₀ value estimated by visual inspection from the data.

interestingly, however, an overall difference was observed for these properties exhibited by most of the 33 other Erythrina alkaloids. However, all active analogues analogously to 1 displayed significantly higher binding affinities to $\alpha 4\beta 2$ and $\alpha 6/\alpha 3\beta 2\beta 3V9'S$ than to $\alpha 3\beta 4$ and $\alpha 4\beta 4$, they displayed substantially lower degrees of $\alpha 4\beta 2$ -over- $\alpha 3\beta 4$ selectivity in the Ca²⁺/Fluo-4 assay, so much so that many of the analogues were equipotent antagonists at these two receptors. Since the binding affinities and antagonist potencies exhibited by the analogues at the $\alpha 4\beta 2$ nAChR correlated fairly well (Figure S9), the observed loss of functional β 2-over- β 4 selectivity arises from the analogues possessing higher $\alpha 3\beta 4$ activity in this Ca2+/Fluo-4 assay than that in the binding assay. Since data from radioligand-binding competition binding experiments predominantly are believed to reflect the binding affinity of the ligand to a desensitized nAChR conformation, the antagonist potency of the competitive antagonist in a functional assay mainly reflects its binding to the resting receptor conformation. The apparent discrepancy between the

 β 2-over- β 4 selectivity exhibited by the Erythrina alkaloids in the two assays could thus be speculated to be a reflection of different binding properties of the compounds to different receptor conformations, in particular in the case of $\alpha 3\beta 4$ and other β 4-containing nAChRs. Overall, the accessed Erythrina ligands retained the high affinity/antagonist potency of 1 at the two β 2-containing nAChRs. Among the exceptions was the significant loss of activity in the heterocyclic derivatives 27, 28, **36**, and **42**. Strikingly, with low nanomolar K_i values at $\alpha 4\beta 2$ and $\alpha 6/\alpha 3\beta 2\beta 3V9'S$, natural products (+)-Cocculine (25) and (+)-Cocculidine (3) displayed 10-100-fold higher binding affinities at the two β 2-containing nAChRs than 1 itself, and these two natural products exhibited pronounced β 2-over- β 4 selectivity in both binding and functional assays. Finally, comparison of the binding affinities exhibited by the 10 epimeric pairs at the two β 2-containing nAChRs reveals very similar K_i values for the 32/epi-32, 40/epi-40, and 43/epi-43 epimers, whereas for the seven other epimeric pairs, it is the natural (S)-epimer that is a somewhat better binder than the

Table 2. Functional Properties Displayed by the *Erythrina* Alkaloid Analogues at $h\alpha 4\beta 2$ -HEK293 and $r\alpha 3\beta 4$ -HEK293 Cell Lines in the Ca²⁺/Fluo-4 Assay^a

| | $h\alpha 4\beta 2$ | $r\alpha 3\beta 4$ | |
|----------------------------|---------------------------------------|---------------------------------------|--|
| | $IC_{50} [pIC_{50} \pm S.E.M.]^{(n)}$ | $IC_{50} [pIC_{50} \pm S.E.M.]^{(n)}$ | |
| (+)-DHβE (1) | $0.35 \ [6.45 \pm 0.06]^{(6)}$ | >100 [<4.0] ^{(6),b} | |
| $(-)$ -DH β E $(1')$ | ~30 [~4.5] ^{(3),c} | ~100 [~4.0] ^{(3),b} | |
| 3 | $0.26 \ [6.59 \pm 0.10]^{(4)}$ | 4.2 $[5.37 \pm 0.12]^{(4)}$ | |
| 18 | $4.8 \ [5.32 \pm 0.08]^{(3)}$ | >100 [<4.0] ^{(3),b} | |
| epi-18 | >100 [<4.0] ^{(3),b} | >100 [<4.0] ^{(3),b} | |
| 24 | $2.0 \ [5.69 \pm 0.08]^{(3)}$ | ~100 [~4.0] ^{(3),c} | |
| 25 | $0.12 \ [6.92 \pm 0.05]^{(4)}$ | $\sim 20 \ [\sim 4.7]^{(3),c}$ | |
| 27 | ~100 [~4.0] ^{(3),c} | >100 [<4.0] ^{(3),b} | |
| 28 | $\sim 20 \ [\sim 4.7]^{(3),c}$ | >100 [<4.0] ^{(3),b} | |
| 29 | >100 [<4.0] ^{(3),b} | >100 [<4.0] ^{(3),b} | |
| epi-29 | >100 [<4.0] ^{(3),b} | >100 [<4.0] ^{(3),b} | |
| 31 | >100 [<4.0] ^{(3),b} | >100 [<4.0] ^{(3),b} | |
| 32 | $7.4 \ [5.13 \pm 0.07]^{(3)}$ | $2.6 [5.59 \pm 0.11]^{(3)}$ | |
| epi-32 | $2.1 \ [5.68 \pm 0.07]^{(3)}$ | 5.0 $[5.30 \pm 0.07]^{(3)}$ | |
| 33 | $0.49 \ [6.31 \pm 0.08]^{(4)}$ | 1.6 $[5.79 \pm 0.14]^{(4)}$ | |
| 34 | $1.5 [5.81 \pm 0.10]^{(3)}$ | $2.6 [5.89 \pm 0.08]^{(3)}$ | |
| epi-34 | $0.99 \ [6.00 \pm 0.07]^{(3)}$ | 4.2 $[5.38 \pm 0.07]^{(3)}$ | |
| 35 | $0.33 \ [6.49 \pm 0.02]^{(3)}$ | $0.79 \ [6.10 \pm 0.05]^{(3)}$ | |
| epi-35 | $0.33 \ [6.49 \pm 0.12]^{(3)}$ | $1.2 \ [5.91 \pm 0.06]^{(3)}$ | |
| 36 | $\sim 20 \ [\sim 4.7]^{(3),c}$ | $\sim 30 \ [\sim 4.5]^{(3),c}$ | |
| epi-36 | $\sim 20 \ [\sim 4.7]^{(3),c}$ | $\sim 50 \ [\sim 4.3]^{(3),c}$ | |
| 37 | $1.8 \ [5.75 \pm 0.11]^{(3)}$ | 11 $[4.96 \pm 0.06]^{(3)}$ | |
| epi-37 | $5.6 \ [5.25 \pm 0.13]^{(3)}$ | $\sim 30 \ [\sim 4.5]^{(3),b}$ | |
| 38 | ~20 [~4.7](3), ^c | ~100 [~4.0](3), ^c | |
| 39 | $1.3 \ [5.89 \pm 0.11]^{(3)}$ | 5.7 $[5.25 \pm 0.09]^{(3)}$ | |
| epi-39 | $1.1 \ [5.94 \pm 0.11]^{(4)}$ | 5.6 $[5.25 \pm 0.06]^{(3)}$ | |
| 40 | $0.52 \ [6.29 \pm 0.10]^{(3)}$ | $2.9 [5.54 \pm 0.03]^{(3)}$ | |
| epi-40 | $0.14 \ [6.84 \pm 0.12]^{(3)}$ | $1.6 \ [5.80 \pm 0.11]^{(4)}$ | |
| 41 | $0.24 \ [6.62 \pm 0.07]^{(3)}$ | $0.78 \ [6.11 \pm 0.07]^{(3)}$ | |
| epi-41 | $0.44 \ [6.36 \pm 0.08]^{(3)}$ | $1.5 [5.82 \pm 0.04]^{(3)}$ | |
| 42 | $1.2 [5.91 \pm 0.09]^{(4)}$ | $2.6 [5.59 \pm 0.09]^{(3)}$ | |
| epi-42 | $1.3 \ [5.89 \pm 0.08]^{(4)}$ | $2.5 [5.61 \pm 0.10]^{(3)}$ | |
| 43 | $0.29 \ [6.54 \pm 0.07]^{(3)}$ | $1.5 \ [5.82 \pm 0.06]^{(3)}$ | |
| eni-43 | $0.54 [6.27 \pm 0.13]^{(3)}$ | $11[596 \pm 0.10]^{(3)}$ | |

^{*a*}The antagonist properties were determined using (S)-nicotine $\sim EC_{80}$ ($EC_{70}-EC_{90}$) as an agonist. Antagonist potencies are given as IC_{50} values in μM [with $pIC_{50} \pm$ standard error of mean (S.E.M.) in brackets] with the number of experiments (*n*) indicated in superscript. ^{*b*}No significant inhibition observed at 100 μ M. ^{*c*}The concentration–inhibition relationship was not complete at 100 μ M. The IC_{50} value was estimated from the fitted curve by visual inspection.

unnatural (*R*)-epimer. The most prominent example of the latter is the 30–100-fold higher binding affinities exhibited by 36 compared to epi-36 at $\alpha 4\beta 2$ and $\alpha 6/\alpha 3\beta 2\beta 3V9'S$.

Elucidation of nAChR SAR Properties Using $\alpha 4\beta 2$ Homology Model. In order to aid the interpretation of the pharmacological properties exhibited by the alkaloids at the nAChRs, all analogues were docked in the homology model of the $\alpha 4\beta 2$ nAChR reported by Yu *et al*^{6b} (see the Supporting Information for details). Docking of 1 into this homology model replicated the binding mode exhibited by the compound in the AChBP/DH βE co-structure.^{6c} All synthesized ligands could be accommodated in the binding pocket, where they adopted a similar conformation to 1 (see Figure 3a): (1) a Hbond from the protonated amine to the backbone carbonyl of W156 in the α 4-subunit; (2) a H-bond from the methoxy pubs.acs.org/joc

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group to the water molecule; and (3) the varying substituents protruding toward the $\beta 2$ subunit. In the proposed binding mode from the docking of **3** and **25**, the aromatic ring forms π stacking interactions to W57 of the $\beta 2$ (t-shaped) and possibly also to Y197 of the $\alpha 4$ subunit (parallel) (see Figure 3b). This interaction is stronger compared to the van der Waals interactions of **1** and could explain the very high binding affinity of these two compounds compared to **1**. In contrast, the two heterocyclic derivatives **27** and **28** have a hydroxyl group protruding into this area, and the unfavorable interactions arising from this could be the root of the significantly reduced binding affinity exhibited by these analogues (see Figure 3c).

A total of 10 epimeric ligand pairs were accessed. In general, the inversion of the C3-stereocenter does not have a significant impact to the affinity of the ligands for $\alpha 4\beta 2$ -with the notable exception of 36/epi-36, where the epimer displayed a substantially lower binding affinity. When docked into the $\alpha 4\beta 2$ model, most epimers exhibited the highest scoring binding pose, where the fused ring system is overlapping with 1 and the methoxy group is adopting a pseudo-axial position with no H-bond to the water molecule (Figure 3e). This is independent of including the water molecule found in the Xray structure in the docking. Thus, the importance of the interaction with the water molecule found in the X-ray structure for binding of Erythrina alkaloids may be overestimated. As the size of the substituent in the Erythrina alkaloid is increased (39 and 43), binding modes wherein the AB-ring systems rotate in the binding pocket become predominant for the epimeric derivatives, as illustrated with epi-43 in Figure 3f. This suggests that the styryl derivatives epi-39 and epi-43 may have a different binding mode than the rest of the epimer analogues.

CONCLUSIONS

A library of Erythrina alkaloid derivatives were accessed in a concise and convergent manner from a common intermediate, giving access to ligands with a complex molecular architecture. The enantioselective total synthesis of two Erythrina alkaloids was completed. Subsequently, the pharmacological properties of the synthesized analogues at nAChRs were determined in binding and functional assays. The natural products (+)-Cocculidine (3) and (+)-Cocculine (25) exhibited remarkably higher affinities at $\alpha 4\beta 2$ and $h\alpha 6/\alpha 3\beta 2\beta 3V9'S$ and higher β 2-over- β 4 selectivity than (+)-DH β E (1) in the binding assay and thus constitute promising leads for future nAChR ligand development. Also, notably was the loss of the inherent $\alpha 4\beta 2$ -over- $\alpha 3\beta 4$ selectivity observed for most of the synthetic analogues in the functional assay as this indicates that the nAChR selectivity profile of the Erythrina alkaloids can be tweaked by structural modifications to its scaffold. Finally, a set of epimeric ligands were used to investigate a specific interaction with a water molecule in the receptor, which showed that the binding pocket can accommodate both epimers with comparable affinity. In conclusion, the Erythrina alkaloids constitute a promising scaffold for ligand development in the nAChR and potentially other fields, and the divergent synthetic strategy presented in this work will enable the exploration of this potential by allowing access to diverse series of derivatives.

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Figure 3. Docking poses exhibited by (a) 1, (b) 3, (c) 27, (d) 37, (e) epi-37, and (f) epi-43 in the $\alpha 4\beta 2$ nAChR homology model.^{6b}

EXPERIMENTAL SECTION

General Information. Unless otherwise stated, all reactions were performed in an oven-dried or flame-dried glassware under an atmosphere of dry nitrogen or argon. Dry tetrahydrofuran (THF), N.N-dimethylformamide (DMF), dichloromethane (DCM), toluene, hexane, acetonitrile, and diethyl ether were used. Reagents were purchased at the highest commercial quality and used without further purification, unless otherwise stated. Reactions were monitored by thin-layer chromatography (TLC) on Merck TLC-Silica gel 60 F₂₅₄ Aluminum sheets $(5 \times 7.5 \text{ cm})$ and visualized by UV irradiation and staining with the potassium permanganate developing agent. LC-MS data were acquired using a waters acquity UPLC-MS consisting of a waters acquity system including column manager, binary solvent manager, sample organizer, Photodiode-Array Detection (PAD) detector (operating at 254 nm), Evaporative light scattering (ELS) detector, and Triple Quad Mass Spectrometry (TQ-MS) equipped with atmospheric pressure photoionization-source operating in the positive ion mode. LC conditions were as follows: the column was Acquity UPLC BEH C18 1.7 μ m; 2.1 \times 50 mm operating at 60 °C with 1.2 mL/min binary gradient consisting of H_2O + 0.05% trifluoroacetic acid (TFA) (A) and MeCN + 5% H₂O + 0.05% TFA (B). Gradient: 0.00 min: 10% B; 1.00 min: 100% B; 1.01 min: 10% B; and 1.15 min: 10% B. The retention times provided in the Experimental Section shall be compared to the total run time of 1.20 min. Yields refer to chromatographically and spectroscopically ($^1\mathrm{H}$ NMR and $^{13}\mathrm{C}$ NMR) homogeneous material, unless otherwise stated. Structural assignments were made with additional information from gradient correlation spectroscopy (gCOSY), gradient heteronuclear single quantum coherence (gHSQC), and gradient heteronuclear multiple bond correlation (gHMBC) experiments. Volatile solvents were removed under reduced pressure using a rotary evaporator. Flash column chromatography was performed using RediSep silica gel (60 Å, 230-400 mesh, 35-70 µm). Ethyl acetate and heptane were purchased from Fisher Chemical and used for chromatography without further purification. NMR data were collected with a Bruker 600-AVANCE-III spectrometer equipped with a 5 mm TCI cryoprobe operating at 600 MHz for ¹H and 151 MHz for ¹³C. Chemical shifts are reported in parts per million with respect to the residual solvent signal $CDCl_3$ (¹H NMR: δ = 7.26; ¹³C NMR: $\delta = 77.16$) or D₂O (¹H NMR: $\delta = 4.79$) or CD₃OD (¹H NMR: $\delta = 3.31$; ¹³C NMR: $\delta = 49.00$) or C₆D₆ (¹H NMR: $\delta = 7.16$; ^{13}C NMR: δ = 128.06), as well as using tetramethylsilane as an internal reference. Peak multiplicities are reported as follows: s = singlet, d = doublet, t = triplet, q = quartet, p = pentet, dd = doublet of doublets, td = triplet of doublets, qd = quartet of doublets, dt =

doublet of triplets, ddd = doublet of doublet of doublets, ddt = doublet of doublet of triplets, ddq = doublet of doublet of quartets, dddd = doublet of doublet of doublets, m = multiplet, and app = apparent. Melting points were determined using a Büchi Melting Point B-540 apparatus. IR spectra were recorded on a Bruker Platinum ATR, Tensor 27 spectrometer. HRMS data were acquired with a Bruker Daltonic MicroTOF with internal calibration using electrospray ionization (ESI) in the positive mode. Optical rotations were measured on an Anton Paar MCP 300 polarimeter. Preparative supercritical fluid chromatography (SFC) was performed on a Berger Multigram II operating at 50 mL/min at 35 °C and 100 bar backpressure using stacked injections. The column was a Diacel IA CHIRALPAK (25 cm, 5μ) (250 × 20 mm). The eluent was CO₂ (95%) and ethanol + 0.1% diethylamine in ethanol (5%). UV detection was performed at 230 nM. Enantiomeric excess (ee) was determined on an Aurora Fusion A5/Agilent SFC system operating at 4 mL/min at 40 °C and 150 bar backpressure. The column was an AD (3× AD 3 columns in serial for compound 6) 3 μ , 15cm (150 × 4.6 mm). The eluent was CO₂ (95%) and 2-propanol (for compound 7) or ethanol (for compounds 6 and 1) + 0.1% diethylamine in ethanol (10%). For the synthesis and characterization of compounds 5-21, and 1, see ref 10k.

(2S,13bS)-2,12-Dimethoxy-2,3,5,6,8,9-hexahydro-1H-indolo-[7a,1-a]isoquinoline (+)-Cocculidine (3). An oven-dried 5 mL microwave vial equipped with a stir bar was charged with enone 24' + 24 (20.0 mg, 0.073 mmol) and methanol (1.5 mL). Copper (II) bromide (33.0 mg, 0.15 mmol) and trimethylorthoformate (0.12 mL, 1.10 mmol) were added at room temperature. The vial was capped, and the reaction mixture was heated to 80 °C (using an oil bath) and stirred for 14 h. At this point, the starting material was completely consumed and the intermediate bromo-ketal is primarily formed, as indicated by TLC and LC-MS (see $[M + 2]^+$ in Figure S1). The reaction mixture was concentrated in vacuo and subsequently diluted with DCM (1.5 mL). Boron trifluoride diethyl etherate (18.0 μ L, 0.15 mmol) was added dropwise at 0 °C, and the reaction mixture was slowly warmed to room temperature and stirred for 2 h. LC-MS at this point shows complete conversion of the intermediate to the desired mass (see $[M + H]^+$ in Figure S2). Aqueous HCl (91.0 μ L, 0.37 mmol, 4 M) was added dropwise, and the resulting solution was stirred for 30 min at room temperature. The mixture was poured into a separation funnel charged with saturated aqueous NaHCO₃ (50 mL) and EtOAc (100 mL). The organic layer was separated, and the aqueous layer was extracted with EtOAc (50 mL \times 2), dried over MgSO₄, and concentrated in vacuo. The resulting crude material was purified by column chromatography [EtOAc (w/10% MeOH and 5% Et_3N):heptane = 0:1 to 8:2], affording (+)-Cocculidine (3) (13.0 mg,

0.046 mmol, 62%) as a white solid. ¹H NMR (600 MHz, C_6D_6): δ 6.97 (d, J = 2.8 Hz, 1H), 6.90 (d, J = 8.3 Hz, 1H), 6.67 (dd, J = 8.3, 2.7 Hz, 1H), 5.41–5.37 (m, 1H), 3.91 (dddd, J = 12.5, 8.7, 6.5, 4.1 Hz, 1H), 3.44 (ddd, J = 14.2, 10.8, 7.0 Hz, 1H), 3.38 (s, 3H), 3.01 (s, 3H), 2.90 (ddd, J = 14.2, 7.7, 2.1 Hz, 1H), 2.83 (td, J = 8.8, 3.2 Hz, 1H), 2.76 (ddd, J = 17.9, 10.8, 7.7 Hz, 1H), 2.63–2.56 (m, 2H), 2.52 (dd, J = 11.3, 4.0 Hz, 1H), 2.34–2.22 (m, 2H), 2.19–2.07 (m, 2H), 1.95 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, C_6D_6): δ 157.9, 142.0, 140.6, 130.6, 126.0, 117.8, 114.5, 112.0, 74.1, 64.8, 55.7, 54.8, 47.5, 42.2, 41.4, 32.6, 27.7, 21.7. HRMS (ESI) m/z: [M + H]⁺ Calcd for C₁₈H₂₄NO₂, 286.1802; Found, 286.1803. Rf = 0.18 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_D^{25} = +260^{\circ}$ (c 1.0, MeOH).] mp = 80–82 °C. [Reference $1-[\alpha]_D^{25} = +260^{\circ}$ (c 1.0, MR spectra are in congruent with previously reported spectra.^{10b}

The olifination reaction leading to 10 was optimized in the following way.

1-(Tert-butyl) 2-Methyl (S)-2-(2,2-dimethoxyethyl)-3-methylenepyrrolidine-1,2-dicarboxylate (10). An oven-dried 250 mL roundbottom flask equipped with a stir bar was charged with sodium hydride (3.62 g, 91 mmol 60% w/w), methyltriphenylphosphonium bromide (32.3 g, 90 mmol), and toluene (160 mL). The flask was equipped with a reflux condenser, and the reaction mixture was heated to 90 °C (using an oil bath) and stirred for 12 h. The reaction mixture was allowed to reach room temperature, allowing the solids to settle. The yellow supernatant was added to a toluene solution (75 mL) of acetal 9 (2.75 g, 8.3 mmol) in an oven-dried 500 mL roundbottom flask, and the resulting solution was stirred at room temperature for 12 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was poured into a separation funnel charged with water (200 mL) and EtOAc (300 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (150 mL \times 3), and the combined organic phase was dried over MgSO4 and concentrated in vacuo. The resulting crude material was purified by column chromatography (EtOAc/ heptane = 0:1 to 7:3), affording olefin 10 (2.11 g, 6.42 mmol, 78%) as a pale yellow oil. For analytical data of olein 10, see ref 10k.

Methyl (105,11aS)-1-hydroxy-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole-2-carboxylate (18). The title compound was synthesized, as previously described.^{10k}

Methyl (10R,11a\$)-1-hydroxy-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole-2-carboxylate (**epi-18**). The title compound was synthesized, as previously described.^{10k}

The enol-esters 18 and epi-18 were usually carried on without any purification but were on occasion purified using column chromatography [EtOAc (w/10% MeOH and 5% Et_3N)/heptane = 0:1 to 8:2].

Methyl (2R,10S,11aS)-10-methoxy-1-oxo-2-(3-oxobutyl)-1,2,3,4,7,9,10,11-octahydro-6H-pyrido[2,1-i]indole-2-carboxylate (23). An oven-dried 25 mL microwave vial equipped with a stir bar was charged with enol-ester 18 (100 mg, 0.36 mmol) and methanol (7 mL). Et₃N (0.25 mL, 1.79 mmol) and methyl vinyl ketone (0.15 mL, 1.79 mmol) were added at room temperature. The vial was capped, and the reaction mixture was heated to 70 $^{\circ}\mathrm{C}$ (using an oil bath) and stirred for 14 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was cooled to room temperature and concentrated in vacuo. The resulting crude material was purified by column chromatography [EtOAc (w/ 10% MeOH and 5% Et_3N)/heptane = 0:1 to 8:2], affording dione 23 (125 mg, 0.36 mmol, >99%) as a colorless oil. ¹H NMR (600 MHz, CDCl₂): δ 5.64–5.60 (m, 1H), 3.68 (s, 3H), 3.54–3.44 (m, 2H), 3.29 (s, 3H), 3.09 (td, J = 9.3, 5.7 Hz, 1H), 3.05–2.98 (m, 2H), 2.87 (ddd, J = 14.3, 9.7, 6.0 Hz, 1H, 2.72–2.60 (m, 2H), 2.41–2.28 (m, 4H), 2.25-2.21 (m, 2H), 2.13 (s, 3H), 1.97-1.89 (m, 1H), 1.81 (dt, J = 14.3, 4.6 Hz, 1H), 1.45 (t, J = 12.2 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 211.0, 207.4, 172.2, 138.6, 120.8, 73.4, 71.9, 58.7, 56.3, 52.5, 48.9, 40.3, 39.2, 36.8, 31.5, 30.3, 30.1, 29.0, 28.1. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₁₉H₂₈NO₅, 350.1962; Found, 350.1965. **Rf** = 0.72 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_D^{25} = +16.3^\circ$ (c 1.0, CHCl₃). >99% ee. The stereochemistry could not be established through 2D NMR, and the structure is therefore

tentatively assigned. The 1 H NMR shows a presence (less than 5%) of the other epimer at C-12. The peaks of the major epimer are reported.

(2S,9aR,13bS)-2-Methoxy-2,3,5,6,9,9a,10,11-octahydro-1Hindolo[7a,1-a]isoquinolin-12(8H)-one (24' + 24). An oven-dried 25 mL microwave vial equipped with a stir bar was charged with dione 23 (125 mg, 0.36 mmol) and water (7 mL). Potassium hydroxide (100 mg, 1.79 mmol) in water (2.3 mL) was added at room temperature. The vial was capped, and the reaction mixture was heated to 105 °C (using an oil bath) and stirred for 2 h. At this point, the starting material was completely consumed, as indicated by TLC. Potassium hydroxide (300 mg, 5.37 mmol) in water (2.3 mL) was added. The reaction mixture was heated to 105 °C (using an oil bath) and stirred for another 3 h. At this point, the intermediate product was completely consumed, as indicated by TLC. The mixture was cooled to room temperature and subsequently poured into a separation funnel charged with saturated aqueous $\dot{\rm NH}_4 Cl$ (50 mL), brine (50 mL), and EtOAc (100 mL). The organic layer was separated, and the aqueous layer was extracted with EtOAc (100 mL \times 2), dried over MgSO₄, and concentrated *in vacuo*. The resulting crude material was purified by column chromatography [EtOAc (w/ 10% MeOH and 5% Et_3N /heptane = 0:1 to 8:2], affording enone 24' + 24 as an inseparable 1:2 diastereomeric mixture (76.0 mg, 0.28 mmol, 78% total yield) as a colorless oil. The crude diastereomeric mixture was proceeded with the reactions below.

(2S,9aS,13bS)-2-Methoxy-2,3,5,6,9,9a,10,11-octahydro-1Hindolo[7a,1-a]isoquinolin-12(8H)-one (24). An oven-dried 5 mL microwave vial equipped with a stir bar was charged with enone 24' + 24 (20.0 mg, 0.073 mmol) and chloroform (1.5 mL). HCl (0.37 mL, 0.46 mmol, 1.25 M) in water methanol was added at room temperature. The vial was capped, and the reaction mixture was stirred for 24 h at room temperature. At this point, only one compound is visible, as indicated by crude ¹H NMR. The mixture was poured into a separation funnel charged with saturated aqueous NaHCO₃ (50 mL), brine (50 mL), and EtOAc (100 mL). The organic layer was separated, and the aqueous layer was extracted with EtOAc (100 mL \times 2), dried over MgSO₄, and concentrated *in vacuo*. The resulting crude material was purified by column chromatography [EtOAc (w/10% MeOH and 5% Et₃N)/heptane = 0:1 to 8:2], affording enone 24 in over 20:1 d.r. (14.8 mg, 0.054 mmol, 74%) as a colorless oil. ¹H NMR (600 MHz, CDCl₃): δ 5.86-5.84 (m, 1H), 5.60-5.56 (m, 1H), 3.36-3.28 (m, 5H), 3.00 (dd, J = 14.5, 4.3 Hz, 1H), 2.98-2.91 (m, 1H), 2.89-2.85 (m, 1H), 2.77-2.71 (m, 1H), 2.57-2.50 (m, 2H), 2.45-2.29 (m, 4H), 2.18-2.12 (m, 1H), 1.99-1.88 (m, 2H), 1.72-1.63 (m, 2H), 1.46 (t, J = 12.0 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 199.6, 164.9, 139.3, 127.2, 119.8, 73.8, 67.1, 56.1, 46.7, 43.1, 37.6, 36.5, 32.9, 31.8, 29.5, 28.7, 26.7. HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{17}H_{24}NO_2$, 274.1802; Found, 274.1802. **Rf** = 0.26 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]_{D}^{25}$ = +85.6° (c 0.5, CHCl₃). >99% ee.

(2S,13bS)-2-Methoxy-2,3,5,6,8,9-Hexahydro-1H-indolo-[7a,1-a]isoquinolin-12-ol (+)-Cocculine (25). An oven-dried 5 mL microwave vial equipped with a stir bar was charged with enone 24 + 24' (20.0 mg, 0.073 mmol) and acetonitrile (1.5 mL). Copper(II) bromide (65.0 mg, 0.29 mmol) was added at room temperature. The vial was capped, and the reaction mixture was heated to 40 °C (using an oil bath) and stirred for 5 h. At this point, the starting material was completely consumed, as indicated by TLC and LC-MS [see $[M + H]^+$ in Figure S3]. The reaction mixture was concentrated in vacuo directly purified by column chromatography [EtOAc (w/10% MeOH and 5% Et₃N)/heptane = 0:1 to 8:2], affording (+)-Cocculine (25) (12.0 mg, 0.044 mmol, 60%) as a white solid. ¹H NMR (600 MHz, CDCl₃): δ 7.01 (d, I = 8.3 Hz, 1H), 6.84 (dd, J = 8.3, 2.5 Hz, 1H), 6.66 (d, J = 2.6 Hz, 1H), 5.80-5.77 (m,1H), 3.81-3.75 (m, 1H), 3.69 (ddd, J = 14.1, 10.0, 7.2 Hz, 1H), 3.43-3.37 (m, 1H), 3.30-3.23 (m, 4H), 3.00 (ddd, J = 17.7, 10.3, 6.3 Hz, 1H), 2.93-2.79 (m, 2H), 2.78-2.70 (m, 1H), 2.63-2.55 (m, 1H), 2.38-2.28 (m, 2H), 2.19-2.12 (m, 1H), 2.08-2.02 (m, 1H). ¹³C{1H} NMR (151 MHz, methanol- D_4): δ 155.7, 136.7, 134.8, 130.5, 121.9, 121.4, 115.5, 113.3, 73.0, 67.1, 55.2, 47.3, 40.9, 38.9,

31.2, 25.9, 20.6. **HRMS (ESI)** m/z: $[M + H]^+$ Calcd for C₁₇H₂₂NO₂, 272.1645; Found, 272.1645. **Rf** = 0.15 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_D^{25} = +215.5^\circ$ (*c* 1.0, MeOH). >99% *ee*. (Reference 27)– $[\alpha]_D^{25} = +217-218^\circ$ (*c* 1.0, MeOH). **mp** = 207-212 °C (Reference 27)–m.p. 216–217).

(2S,12bS)-2-Methoxy-2,3,5,6,9,11-hexahydro-1H,8H-pyrazolo-[3',4':3,4]pyrido[2,1-i]indol-10-ol (27). An oven-dried 5 mL microwave vial equipped with a stir bar was charged with crude enol-ester 18 (22.0 mg, 0.079 mmol) and 1,4-dioxane (1 mL). Hydrazinehydrate (64% wt, 60 μ L, 0.79 mmol) was added at room temperature. The vial was capped, and the reaction mixture was heated to 100 °C (using an oil bath) and stirred for 72 h. At this point, the starting material was completely consumed, as indicated by TLC. The reaction mixture was concentrated in vacuo directly purified by column chromatography (MeOH/EtOAc = 0:1 to 1:1), affording pyrazole 27 (6.00 mg, 0.02 mmol, 30%) as a clear colorless oil. ¹H **NMR** (600 MHz, $CDCl_3$): δ 5.57–5.54 (m, 1H), 3.75–3.69 (m, 1H), 3.31 (s, 3H), 3.24-3.20 (m, 2H), 2.94 (td, J = 9.2, 3.2 Hz, 1H), 2.75-2.65 (m, 2H), 2.63-2.58 (m, 1H), 2.47-2.39 (m, 2H), 2.28-2.22 (m, 2H), 2.09-2.03 (m, 1H), 1.53 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 160.9, 143.7, 139.1, 119.1, 97.5, 74.2, 61.9, 56.3, 46.2, 40.9, 40.2, 31.8, 27.1, 14.0. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₁₄H₂₀N₃O₂⁺, 262.1550; Found, 262.1555 Rf = 0.28 (silica gel, 1:1 EtOAc/MeOH) $[\alpha]_{D}^{25} = +203.8^{\circ}$ (c 0.5, CHCl₃).

(2S,13bS)-12-Mercapto-2-methoxy-2,3,5,6,8,9-hexahydro-1Hpyrimido[4',5':3,4]pyrido[2,1-i]indol-10-ol (28). An oven-dried 5 mL microwave vial equipped with a stir bar was charged with crude enolester 18 (60.0 mg, 0.21 mmol), thiourea (41.0 mg, 0.54 mmol), and potassium tert-butoxide (120 mg, 0.54 mmol). MeOH (1.6 mL) was added at room temperature. The vial was capped, and the reaction mixture was heated to 90 $^\circ C$ (using an oil bath) and stirred for 12 h. At this point, the starting material was almost completely consumed, as indicated by TLC. The solution was poured into a separation funnel containing brine (20 mL) and 2-methyl-THF (20 mL). The organic layer was separated, and the pH of the water phase was adjusted to 7, using 1 M HCl before extracting it with 2-methyl-THF $(3 \times 20 \text{ mL})$. The combined organic phases were dried over MgSO₄ and concentrated in vacuo. The resulting crude was purified by column chromatography (MeOH/EtOAc = 0:1 to 1:1), affording pyrimidine 27 (3.30 mg, 0.001 mmol, 5%) as a clear colorless oil. ¹H NMR (600 MHz, CDCl₃): δ 8.32 (s, 1H), 5.87–5.83 (m, 1H), 3.69– 3.62 (m, 1H), 3.38 (s, 3H), 3.36-3.26 (m, 2H), 3.18 (dd, J = 14.8, 7.6 Hz, 1H), 3.05-3.00 (m, 1H), 2.82-2.71 (m, 2H), 2.66-2.58 (m, 1H), 2.56-2.49 (m, 1H), 2.46-2.35 (m, 2H), 2.29 (dd, J = 18.4, 6.6 Hz, 1H), 2.18–2.11 (m, 1H), 1.66 (t, J = 12.3 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 173.7, 161.1, 150.0, 136.7, 123.1, 110.1, 72.8, 62.6, 56.6, 46.7, 40.6, 39.3, 31.6, 27.1, 15.3. HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{15}H_{20}N_3O_2S^+$, 306.1271; Found, 306.1271 Rf = 0.36 (silica gel, 8:2 EtOAc/MeOH) $[\alpha]_{D}^{25} = +184.8^{\circ}$ (*c* 0.3, CHCl₃).

(10S,11aS)-10-Methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1i]indol-1(2H)-one (29). An oven-dried 100 mL flask equipped with a stir bar was charged with bis-ester 17 (920 mg, 2.96 mmol) and toluene (33 mL). Potassium tert-butoxide (763 mg, 6.80 mmol) was added at room temperature, the vial was equipped with a reflux condenser, and the reaction mixture was heated to 95 °C (using an oil bath) and stirred for 1 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was cooled to room temperature and was poured into a separation funnel charged with saturated aqueous NH₄Cl (50 mL), brine (50 mL), and 2-methyl-THF (100 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (100 mL \times 2), and the combined organic phase was washed with brine (200 mL), dried over MgSO4, and concentrated in vacuo. To an oven-dried 100 mL flask equipped with a stir bar were added the resulting crude oil and 4N aqueous HCl (29 mL). The flask was equipped with a reflux condenser under argon, and the reaction mixture was heated to 110 °C (using an oil bath) for 3 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was poured into a separation funnel charged with saturated aqueous

NaHCO₃ (100 mL) and 2-methyl-THF (100 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (100 mL \times 2), and the combined organic phase was washed with brine (200 mL), dried over MgSO₄, and concentrated in vacuo. The resulting crude material was purified by column chromatography [EtOAc (w/ 10% MeOH and 5% Et₃N)/heptane = 0:1 to 8:2], affording ketone 29 (435 mg, 1.97 mmol, 66%) as a colorless oil. ¹H NMR (600 MHz, $CDCl_3$: δ 5.62–5.57 (m, 1H), 3.45 (ddd, J = 14.6, 13.3, 3.4 Hz, 1H), 3.32 (s, 3H), 3.31-3.26 (m, 1H), 3.03-2.94 (m, 3H), 2.80 (td, J = 13.8, 6.8 Hz, 1H), 2.67 (ddd, J = 12.0, 3.4, 0.9 Hz, 1H), 2.57 (tddd, J = 18.2, 13.5, 8.2, 4.4 Hz, 2H), 2.41 (ddt, J = 8.7, 6.0, 2.7 Hz, 2H), 2.34 (ddt, J = 14.2, 4.6, 2.0 Hz, 1H), 2.05-2.00 (m, 1H), 1.93 (ddq, J = 17.2, 9.3, 3.2 Hz, 1H), 1.54 (t, J = 12.2 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 214.6, 137.7, 119.7, 73.6, 73.6, 56.2, 47.3, 43.0, 37.2, 36.4, 31.8, 26.9, 25.9. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₁₃H₂₀NO₂, 222.1489; Found, 222.1494. Rf = 0.21 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_{D}^{25} = +11.7^{\circ}$ (c 1.0, CHCl₃). >99% ee.

(10R,11aS)-10-Methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1i]indol-1(2H)-one (epi 29). An oven-dried 100 mL flask equipped with a stir bar was charged with bis-ester methyl (6R,7aS)-6-methoxy-1-(4-methoxy-4-oxobutyl)-1,2,3,5,6,7-hexahydro-7aH-indole-7a-carboxylate (191 mg, 0.61 mmol) and toluene (7 mL). Potassium tertbutoxide (158 mg, 1.41 mmol) was added at room temperature, the vial was equipped with a reflux condenser, and the reaction mixture was heated to 95 °C (using an oil bath) and stirred for 1 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was cooled to room temperature and was poured into a separation funnel charged with saturated aqueous NH₄Cl (50 mL), brine (50 mL), and 2-methyl-THF (100 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (100 mL \times 2), and the combined organic phase was washed with brine (200 mL), dried over MgSO₄, and concentrated in vacuo. To an oven-dried 100 mL flask equipped with a stir bar were added the resulting crude oil and 3N aqueous HCl (15 mL). The flask was equipped with a reflux condenser under argon, and the reaction mixture was heated to 110 °C (using an oil bath) for 3 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was poured into a separation funnel charged with saturated aqueous NaHCO₃ (50 mL) and 2-methyl-THF (50 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (50 mL \times 2), and the combined organic phase was washed with brine (100 mL), dried over MgSO₄, and concentrated in vacuo. The resulting crude material was purified by column chromatography [EtOAc (w/10% MeOH and 5% Et₃N)/heptane = 0:1 to 8:2], affording C3-epi-ketone 29 (74 mg, 0.34 mmol, 55%) as a colorless oil. ¹H NMR (600 MHz, CDCl₃): δ 5.56-5.54 (m, 1H), 3.68-3.65 (m, 1H), 3.42 (td, J = 13.9, 3.5 Hz, 1H), 3.23 (s, 3H), 2.94-2.86 (m, 5H), 2.61-2.50 (m, 2H), 2.42 (dddd, J = 18.5, 10.9, 6.2, 2.2 Hz, 1H), 2.37-2.32 (m, 1H), 2.28-2.18 (m, 2H), 2.07-2.02 (m, 1H), 1.67 (dd, J = 13.4, 2.9 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): *δ* 212.2, 138.7, 118.0, 73.7, 69.6, 56.6, 46.3, 43.2, 37.3, 35.7, 29.6, 28.8, 27.6. HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{13}H_{20}NO_{21}$ 222.1489; found, 222.1489. Rf = 0.3 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_{D}^{25} = +88.6^{\circ} (c \ 0.9, \text{ CHCl}_{3}). >99\% ee.$

(10S,11aS)-10-Methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1i]indol-1-yl Trifluoromethanesulfonate (30). An oven-dried 25 mL microwave vial equipped with a stir bar was charged with ketone 29 (100 mg, 0.46 mmol) and THF (3.8 mL). The mixture was cooled to -78 °C before KHMDS (1.80 mL, 0.90 mmol, 0.5 M) was added. The resulting mixture was left stirring at the same temperature for 1.5 h. PhN(OTf)₂ (300 mg, 0.84 mmol) in 1 mL THF was added dropwise before the dry ice bath was replaced with an ice bath. The mixture was allowed to warm to 0 °C and was stirred at this temperature for 2 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was quenched with ice-cold water and was subsequently poured into a separation funnel charged with saturated aqueous NH₄Cl (50 mL), brine (50 mL), and EtOAc (100 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (100 mL \times 2), and the combined organic phase was washed with brine (200 mL), dried over MgSO₄, and

concentrated *in vacuo*. The resulting crude material was purified by column chromatography [EtOAc (w/10% MeOH and 5% Et₃N)/ heptane = 0:1 to 8:2], affording alkenyl triflate **30** (131 mg, 0.37 mmol, 82%) as a colorless oil. ¹H NMR (600 MHz, CDCl₃): δ 5.75 (dd, J = 5.2, 2.8 Hz, 1H), 5.71–5.68 (m, 1H), 3.87 (dddd, J = 12.4, 8.0, 6.7, 4.4 Hz, 1H), 3.37 (s, 3H), 3.18 (ddd, J = 14.6, 11.7, 5.7 Hz, 1H), 3.02–2.90 (m, 3H), 2.76–2.56 (m, 3H), 2.50–2.41 (m, 2H), 2.12 (dt, J = 18.7, 5.4 Hz, 1H), 1.98 (ddtd, J = 17.7, 7.6, 3.6, 2.4 Hz, 1H), 1.52 (t, J = 12.0 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 152.1, 136.1, 123.3, 118.4 (q, J = 319.7 Hz), 116.8, 73.2, 63.9, 56.4, 46.6, 41.4, 39.8, 31.7, 27.2, 21.0. HRMS (ESI) m/z: [M + H]⁺ Calcd for C₁₄H₁₉F₃NO₄S, 354.0981; Found, 354.0980. Rf = 0.25 (silica gel, 7:1 EtOAc/MeOH). [α]D²⁵ = +58.4° (c 1.0, CHCl₃). >99% ee.

(10R,11aS)-10-Methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1i]indol-1-yl Trifluoromethanesulfonate (epi-30). An oven-dried 5 mL microwave vial equipped with a stir bar was charged with epi-29 (33.0 mg, 0.15 mmol) and THF (1 mL). The mixture was cooled to -78 °C before KHMDS (0.60 mL, 0.30 mmol, 0.5 M) was added. The resulting mixture was left stirring at the same temperature for 1.5 h. PhN(OTf)₂ (80.0 mg, 0.22 mmol) in 0.5 mL THF was added dropwise before the dry ice bath was replaced with an ice bath. The mixture was allowed to warm to 0 °C and was stirred at this temperature for 2 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was quenched with ice-cold water and was subsequently poured into a separation funnel charged with saturated aqueous NH₄Cl (50 mL), brine (50 mL), and EtOAc (100 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (100 mL \times 2), and the combined organic phase was washed with brine (200 mL), dried over MgSO4, and concentrated in vacuo. The resulting crude material was purified by column chromatography [EtOAc (w/10% MeOH and 5% Et₃N)/ heptane = 0:1 to 8:2], affording C3-epi-alkenyl triflate 30 (44.0 mg, 0.13 mmol, 84%) as a colorless oil. ¹H NMR (600 MHz, CDCl₂): δ 5.79 (dtd, J = 7.9, 2.5, 1.3 Hz, 1H), 5.76 (dd, J = 5.6, 2.5 Hz, 1H), 3.35 (s, 4H), 3.32 (tdd, J = 10.2, 5.8, 4.6 Hz, 1H), 3.14 (ddd, J = 14.5, 11.5, 5.2 Hz, 1H), 3.00 (td, J = 9.2, 5.5 Hz, 1H), 2.96–2.90 (m, 2H), 2.73 (dddd, J = 18.1, 11.5, 6.3, 2.5 Hz, 1H), 2.67-2.60 (m, 1H), 2.47-2.41 (m, 1H), 2.40-2.34 (m, 1H), 2.27 (ddd, J = 14.0, 5.8, 2.1 Hz, 1H), 2.09–2.04 (m, 1H), 2.04–1.97 (m, 1H), 1.74 (dd, J = 14.0, 10.2 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 152.2, 137.8, 122.5, 118.3 (q, J = 319.6 Hz), 115.9, 75.0, 63.7, 56.2, 46.5, 39.8, 38.1, 29.4, 27.3, 20.1. HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{14}H_{19}F_{3}NO_{4}S_{1}$ 354.0981; Found, 354.0980. Rf = 0.25 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_D^{25} = +46.3^\circ$ (c 1.0, CHCl₃). >99% ee. Note: It was not possible to remove all traces of solvent from this compound without significant decomposition.

Methyl (10R,11aS)-1-Hydroxy-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole-2-carboxylate (31). To an oven-dried 25 mL microwave vial equipped with a stir bar were added epi-18 (95.0 mg, 0.34 mmol) and 6N aqueous HCl (10 mL). The vial was capped, and the reaction mixture was submerged in an oil bath that was preheated to 115 °C (using an oil bath) and stirred for 16 h. At this point, the starting material was completely consumed, as indicated by TLC, and the mixture was poured into a separation funnel charged with saturated aqueous NaHCO₃ (50 mL) and 2-methyl-THF (50 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (50 mL \times 2), and the combined organic phase was washed with brine (100 mL), dried over MgSO₄, and concentrated in vacuo. The resulting crude material was purified by column chromatography [EtOAc (w/10% MeOH and 5% Et₃N)/heptane = 0:1 to 8:2], affording lactol 31 (21.0 mg, 0.10 mmol, 30%) as a colorless oil. ¹H NMR (600 MHz, CDCl₃): δ 5.61-5.58 (m, 1H), 4.32 (dtd, J = 5.5, 2.6, 1.3 Hz, 1H), 3.29 (dt, J = 10.3, 7.9 Hz, 1H), 3.07-2.97 (m, 2H), 2.91 (td, J = 8.3, 1.2 Hz, 1H), 2.74 (s, 1H), 2.68 (dd, J = 10.6, 5.8 Hz, 1H), 2.65-2.56 (m, 1H), 2.54-2.48 (m, 1H),2.44-2.30 (m, 2H), 2.08 (dtd, J = 13.4, 3.3, 1.5 Hz, 1H), 1.95 (tddd, J = 13.8, 12.2, 4.7, 3.4 Hz, 1H), 1.73 (td, J = 13.7, 4.0 Hz, 1H), 1.50 (d, J = 10.6 Hz, 1H), 1.31–1.24 (m, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 145.2, 117.1, 102.9, 72.8, 67.9, 49.2, 44.5, 36.6, 35.6, 35.1, 28.3, 16.6. HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{13}H_{20}NO_{21}$ 208.1332; Found, 208.1338. **Rf** = 0.11 (silica gel, 7:1 EtOAc/MeOH). $\lceil \alpha \rceil_D^{25} = +11.6^{\circ}$ (c 0.5, CHCl₃). >99% ee.

General Procedure for the Suzuki Coupling. An oven-dried 5 mL microwave vial equipped with a stir bar was charged with alkenyl triflate **30** or C3-epi-30 (20.0 mg, 0.06 mmol), Pd(dppf)Cl₂ DCM (4.60 mg, 0.006 mmol), and boronic acid or boronic ester (0.10 mmol, 1.7 equiv). Degassed DME (0.6 mL) and NaOH (0.05 mL, 2 M) were added, and the resulting solution was further degassed for 10 min. The vial was capped, and the reaction mixture was submerged in an oil bath that was preheated to 85 °C. The reaction mixture was stirred for 75 min. At this point, the starting material was completely consumed, as indicated by TLC. The mixture was cooled to room temperature and concentrated *in vacuo*. The resulting crude material was directly purified by column chromatography (EtOAc/MeOH = 1:0 to 4:6), affording Suzuki-adducts **32**-**43** and **epi-32**-**epi-43** as pale yellow oils. Note: The same elution gradient was used for the purification of compounds **32**-**43**.

3-((105,11aS)-10-Methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido-[2,1-i]indol-1-yl)benzonitrile (**32**). The title compound was derived from **30** and (3-cyanophenyl)boronic acid to give the desired product (15.0 mg, 0.049 mmol, 86%) ¹H NMR (600 MHz, CDCl₃): δ 7.54– 7.51 (m, 1H), 7.42–7.41 (m, 1H), 7.37–7.34 (m, 2H), 5.73–5.69 (m, 1H), 5.61–5.58 (m, 1H), 3.29 (ddd, *J* = 14.4, 11.8, 5.9 Hz, 1H), 3.09–3.03 (m, 6H), 2.76 (dddd, *J* = 12.5, 8.8, 6.2, 3.9 Hz, 1H), 2.64– 2.53 (m, 3H), 2.41 (ddd, *J* = 11.6, 4.0, 0.9 Hz, 1H), 2.10–2.05 (m, 1H), 1.98 (dt, *J* = 19.2, 5.4 Hz, 1H), 1.82 (ddtd, *J* = 17.6, 8.9, 3.5, 2.0 Hz, 1H), 1.51 (t, *J* = 11.9 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 145.0, 141.1, 138.8, 133.5, 132.5, 130.5, 128.8, 128.1, 121.7, 119.0, 112.0, 73.7, 64.6, 56.1, 46.1, 42.8, 40.0, 31.7, 28.2, 19.6. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₂₀H₂₃N₂O, 307.1805; Found, 307.1807. **Rf** = 0.32 (silica gel, 6:2 EtOAc/MeOH). [*α*]D²⁵ = +97.6° (*c* 1.0, CHCl₃). >99% *ee*.

3-((10*R*,11*a*S)-10-*M*ethoxy-3,4,7,9,10,11-*h*exahydro-6*H*-pyrido-[2,1-*i*]*i*ndol-1-yl)*b*enzonitrile (*epi-32*). The title compound was derived from *epi-30* and (3-cyanophenyl)boronic acid to give the desired product (6.00 mg, 0.02 mmol, 30%) ¹H NMR (600 MHz, CDCl₃): δ 7.51 (d, *J* = 7.7 Hz, 1H), 7.45 (s, 1H), 7.40 (d, *J* = 7.8 Hz, 1H), 7.36–7.30 (m, 1H), 5.74–5.70 (m, 1H), 5.60 (d, *J* = 6.8 Hz, 1H), 3.31–3.22 (m, 5H), 3.07–2.99 (m, 3H), 2.67–2.51 (m, 2H), 2.44 (dt, *J* = 14.8, 7.1 Hz, 1H), 2.29 (dd, *J* = 14.4, 5.4 Hz, 1H), 2.23–2.16 (m, 1H), 1.90 (dt, *J* = 18.9, 5.4 Hz, 1H), 1.50–1.43 (m, 1H), 1.09–1.01 (m, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 143.0, 141.9, 141.0, 133.9, 132.9, 130.5, 128.5, 127.3, 120.4, 119.2, 111.7, 75.2, 64.9, 56.3, 46.2, 39.9, 38.1, 29.6, 28.1, 18.7. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₂₀H₂₃N₂O, 307.1805; Found, 307.1805. **Rf** = 0.19 (silica gel, 6:2 EtOAc/MeOH). [*α*]D²⁵ = +47.2° (*c* 0.55, CHCl₃). >99% *ee*.

(10S,11aS)-1-(3-Fluorophenyl)-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (33). The title compound was derived from 30 and (3-fluorophenyl)boronic acid to give the desired product (9.00 mg, 0.050 mmol, 53%) ¹H NMR (600 MHz, $CDCl_3$): δ 7.20 (m, 1H), 6.95–6.89 (m, 2H), 6.84 (dt, J = 10.3, 2.1 Hz, 1H), 5.70 (dd, J = 5.2, 2.6 Hz, 1H), 5.60-5.58 (m, 1H), 3.29 (ddd, J = 14.3, I)11.9, 6.0 Hz, 1H), 3.12–3.04 (m, 6H), 2.87 (tdd, J = 12.5, 6.2, 3.9 Hz, 1H), 2.68–2.52 (m, 3H), 2.42 (dd, J = 11.6, 3.9 Hz, 1H), 2.14–2.07 (m, 1H), 1.98 (dt, J = 19.1, 5.5 Hz, 1H), 1.87–1.79 (m, 1H), 1.52 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 162.4 (d, J =245.5 Hz), 145.8 (d, J = 7.6 Hz), 141.5, 138.7, 129.3 (d, J = 8.5 Hz), 126.7, 124.9 (d, J = 2.7 Hz), 121.7, 115.9 (d, J = 21.3 Hz), 113.7 (d, J = 21.1 Hz), 73.7, 64.9, 56.0, 46.1, 42.6, 40.1, 31.7, 28.1, 19.6. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₁₉H₂₃NOF, 300.1758; Found, 300.1760. Rf = 0.29 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]_D^{25} = +88.9^\circ$ $(c 0.9, CHCl_2)$, >99% ee.

(105,11aS)-1-(3-(Tert-Butyl)phenyl)-10-methoxy-3,4,7,9,10,11hexahydro-6H-pyrido[2,1-i]indole (**34**). The title compound was derived from **30** and (3-(*tert*-butyl)phenyl)boronic acid to give the desired product (10.0 mg, 0.030 mmol, 52%) ¹H NMR (600 MHz, CDCl₃): δ 7.24 (ddd, J = 7.8, 2.1, 1.1 Hz, 1H), 7.17 (t, J = 7.7 Hz, 1H), 7.13 (t, J = 1.9 Hz, 1H), 6.92 (dt, J = 7.4, 1.4 Hz, 1H), 5.69 (ddd, J = 5.1, 2.6, 1.0 Hz, 1H), 5.55 (dd, J = 3.7, 2.2 Hz, 1H), 3.31

(ddd, J = 14.3, 11.9, 6.0 Hz, 1H), 3.13 (ddd, J = 10.4, 9.1, 6.5 Hz, 1H), 3.11–3.02 (m, 5H), 2.78 (tdd, J = 12.5, 6.3, 3.9 Hz, 1H), 2.72–2.64 (m, 1H), 2.57 (dddd, J = 18.9, 11.7, 7.1, 2.6 Hz, 2H), 2.44 (dd, J = 11.4, 3.9 Hz, 1H), 2.06–2.00 (m, 1H), 1.97 (dt, J = 18.9, 5.4 Hz, 1H), 1.79 (ddtd, J = 17.4, 8.7, 3.6, 1.7 Hz, 1H), 1.45 (t, J = 11.8 Hz, 1H), 1.29 (s, 9H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 150.5, 143.1, 143.0, 139.5, 127.6, 126.3, 126.2, 125.3, 123.6, 121.0, 73.8, 65.0, 56.0, 46.2, 42.7, 40.3, 34.7, 31.9, 31.4, 28.2, 19.7. HRMS (ESI) m/z: [M + H]⁺ Calcd for C₂₃H₃₂NO, 338.2478; Found, 338.2477. Rf = 0.34 (silica gel, 7:1 EtOAc/MeOH). [α]D²⁵ = +105.0° (*c* 1.0, CHCl₃). >99% *ee*.

(10R,11aS)-1-(3-(Tert-Butyl)phenyl)-10-methoxy-3,4,7,9,10,11hexahydro-6H-pyrido[2,1-i]indole (epi-34). The title compound was derived from epi-30 and (3-(tert-butyl)phenyl)boronic acid to give the desired product (8.80 mg, 0.026 mmol, 42% crude yield)¹H **NMR** (600 MHz, CDCl₃): δ 7.24 (d, J = 0.9 Hz, 1H), 7.21–7.14 (m, 2H), 6.98 (dt, J = 7.6, 1.4 Hz, 1H), 5.74 (dd, J = 5.5, 2.3 Hz, 1H), 5.64 (d, J = 8.0 Hz, 1H), 3.37-3.30 (m, 2H), 3.28 (s, 3H), 3.22-3.08 (m, 3H), 2.71 (tdd, J = 13.4, 6.4, 3.3 Hz, 1H), 2.59 (dddd, J = 18.7, 11.8, 6.8, 2.3 Hz, 1H), 2.48 (ddd, J = 14.6, 8.0, 4.9 Hz, 1H), 2.34 (dd, *J* = 14.2, 3.6 Hz, 1H), 2.17 (dddd, *J* = 11.9, 7.7, 4.0, 2.3 Hz, 1H), 2.00 (dt, J = 18.7, 5.5 Hz, 1H), 1.61 (dd, J = 14.3, 11.0 Hz, 1H), 1.28 (s, 9H), 1.15–1.07 (m, 1H). ${}^{13}C{1H}$ NMR (151 MHz, CDCl₃): δ 150.3, 142.8, 140.3, 139.7, 127.6, 126.8, 126.0, 124.8, 124.1, 121.5, 74.9, 66.1, 56.3, 46.2, 40.0, 37.1, 34.7, 31.4, 29.3, 27.9, 18.9. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₂₃H₃₂NO, 338.2478; Found, 338.2482. **Rf** = 0.29 (silica gel, 7:1 EtOAc/MeOH). $[\alpha]D^{25} = +60.9^{\circ}$ (c 0.09, CHCl₂). >99% ee. Note: NMR contains 10% of an unknown impurity.

. (10Ś,11aS)-1-(Benzo[d][1,3]dioxol-5-yl)-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (35). The title compound was derived from 30 and benzo[d][1,3]dioxol-5-ylboronic acidto give the desired product (11.0 mg, 0.034 mmol, 60%) ¹H NMR (600 MHz, CDCl₃): δ 6.70 (d, J = 7.8 Hz, 1H), 6.62–6.57 (m, 2H), 5.93 (dd, J = 8.2, 1.5 Hz, 2H), 5.64 (dd, J = 5.2, 2.6 Hz, 1H), 5.59-5.54 (m, 1H), 3.26 (ddd, J = 14.3, 11.9, 6.0 Hz, 1H), 3.13 (s, 3H), 3.10–3.00 (m, 3H), 2.98 (dddd, J = 15.6, 8.8, 6.6, 3.5 Hz, 1H), 2.65– 2.50 (m, 3H), 2.41 (dd, J = 11.6, 3.8 Hz, 1H), 2.18-2.12 (m, 1H), 1.94 (dt, J = 18.9, 5.2 Hz, 1H), 1.83 (ddtd, J = 17.4, 8.8, 3.5, 1.8 Hz, 1H), 1.48 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 147.0, 146.5, 142.1, 139.0, 137.7, 125.7, 122.5, 121.3, 109.6, 107.8, 101.0, 73.9, 64.9, 56.0, 46.0, 42.6, 40.1, 31.9, 28.2, 19.8. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₂₀H₂₄NO₃, 326.1751; Found, 326.1754. Rf = 0.21 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]D^{25}$ = +104.1° (c 1.0, CHCl₃). >99% ee.

(10R,11aS)-1-(Benzo[d][1,3]dioxol-5-yl)-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (epi-35). The title compound was derived from epi-30 and benzo[d][1,3]dioxol-5ylboronic acid to give the desired product (11.0 mg, 0.034 mmol, 60%) ¹H NMR (600 MHz, CDCl₃): δ 6.68–6.66 (m, 2H), 6.64 (dd, J = 7.8, 1.7 Hz, 1H), 5.91 (dd, 2H), 5.67-5.65 (m, 1H), 5.59-5.56 (m, 1H), 3.29 (s, 3H), 3.29-3.23 (m, 2H), 3.08-2.98 (m, 3H), 2.65-2.58 (m, 1H), 2.54 (dddd, J = 18.6, 11.7, 6.8, 2.5 Hz, 1H), 2.42-2.36 (m, 1H), 2.28 (ddd, J = 14.1, 5.6, 2.3 Hz, 1H), 2.23-2.17 (m, 1H),1.87 (dt, J = 18.7, 5.4 Hz, 1H), 1.57 (dd, J = 14.0, 11.0 Hz, 1H), 1.27–1.23 (m, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 146.8, 146.5, 142.8, 141.2, 135.7, 125.5, 122.8, 120.1, 109.9, 107.6, 100.9, 75.5, 65.3, 56.2, 46.1, 39.9, 37.7, 29.7, 28.1, 18.9. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₂₀H₂₄NO₃, 326.1751; Found, 326.1754. Rf = 0.20 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]D^{25} = +43.3^{\circ}$ (c 0.9, CHCl₃). >99% 00

Methyl 5-((105,11a5)-10-*Methoxy*-3,4,7,9,10,11-*hexahydro*-6*Hpyrido*[2,1-*i*]*indo*[-1-*y*]*picolinate* (**36**). The title compound was derived from **30** and (6-(methoxycarbonyl)pyridin-3-yl)boronic acid to give the desired product (5.00 mg, 0.015 mmol, 29%) ¹H NMR (600 MHz, CDCl₃): δ 8.51 (d, J = 2.2 Hz, 1H), 8.02 (d, J = 8.1 Hz, 1H), 7.58 (dd, J = 8.1, 2.2 Hz, 1H), 5.81–5.77 (m, 1H), 5.61–5.59 (m, 1H), 4.01 (s, 3H), 3.38–3.29 (m, 1H), 3.12–3.03 (m, 6H), 2.79 (tdd, J = 12.4, 6.1, 3.9 Hz, 1H), 2.64–2.53 (m, 3H), 2.45 (dd, J = 11.6, 4.0 Hz, 1H), 2.13–2.07 (m, 1H), 2.03–2.00 (m, 1H), 1.83 (ddt, pubs.acs.org/joc

J = 17.4, 8.6, 3.0 Hz, 1H), 1.53 (t, *J* = 12.0 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 165.8, 149.7, 146.1, 141.0, 139.0, 138.7, 136.9, 129.5, 124.3, 121.9, 73.7, 64.6, 56.2, 53.1, 46.2, 42.7, 39.9, 31.7, 28.1, 19.7. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₂₀H₂₅N₂O₃, 341.1860; Found, 341.1864. **Rf** = 0.11 (silica gel, 6:2 EtOAc/MeOH). [α]_D²⁵ = +80.3° (*c* 0.5, CHCl₃). >99% *ee*.

Methyl 5-((10*R*, 11*a*S)-10-*Methoxy*-3,4,7,9,10,11-*hexahydro*-6*Hpyrido*[2,1-*i*]*indo*]-1-*y*]*picolinate* (*epi*-36). The title compound was derived from epi-30 and (6-(methoxycarbonyl)pyridin-3-yl)boronic acid to give the desired product (2.70 mg, 0.008 mmol, 13%) ¹H NMR (600 MHz, CDCl₃): δ 8.51 (d, J = 2.3 Hz, 1H), 8.01 (d, J = 8.0 Hz, 1H), 7.66 (dd, J = 8.1, 2.2 Hz, 1H), 5.81–5.79 (m, 1H), 5.61– 5.58 (m, 1H), 4.00 (s, 3H), 3.34–3.24 (m, 1H), 3.23 (s, 3H), 3.07– 3.03 (m, 3H), 2.65–2.54 (m, 2H), 2.48–2.42 (m, 1H), 2.29 (dd, J = 14.3, 5.4 Hz, 1H), 2.23–2.18 (m, 1H), 1.94 (dt, J = 19.1, 5.4 Hz, 1H), 1.50 (dd, J = 14.2, 10.4 Hz, 1H), 1.13–1.08 (m, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 165.9, 150.1, 146.2, 141.0, 140.8, 139.8, 137.1, 128.4, 124.2, 120.6, 74.9, 64.6, 56.2, 53.0, 46.1, 39.8, 38.3, 29.7, 28.1, 18.7. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₂₀H₂₅N₂O₃, 341.1860; Found, 341.1867. **Rf** = 0.10 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]_D^{25}$ = not enough material for a reliable measurement.

(105, $\overline{1}1a$ 5)-10-Methoxy-1-(1-methyl-1H-pyrazol-4-yl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (**37**). The title compound was derived from **30** and (1-methyl-1H-pyrazol-4-yl)boronic acid to give the desired product (12.0 mg, 0.042 mmol, 74%) ¹H NMR (600 MHz, CDCl₃): δ 7.35 (s, 1H), 7.21 (s, 1H), 5.72 (t, J = 3.8 Hz, 1H), 5.61 (dt, J = 4.0, 1.9 Hz, 1H), 3.84 (s, 3H), 3.31–3.21 (m, 2H), 3.16 (s, 3H), 3.06–2.96 (m, 3H), 2.56–2.46 (m, 3H), 2.39 (dd, J = 11.3, 4.2 Hz, 1H), 2.34–2.29 (m, 1H), 1.96–1.90 (m, 2H), 1.49 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 140.2, 138.3, 132.5, 127.9, 124.3, 124.2, 120.4, 74.1, 64.2, 56.0, 46.1, 42.6, 39.9, 39.0, 31.9, 27.9, 19.7. HRMS (ESI) m/z: [M + H]⁺ Calcd for C₁₇H₂₄N₃O, 286.1914; Found, 286.1917. **Rf** = 0.09 (silica gel, 6:2 EtOAc/MeOH). [α]_D²⁵ = +44.3° (c 1.0, CHCl₃). >99% ee.

(10*R*, 11*a*S)-10-*M*ethoxy-1-(1-methyl-1*H*-pyrazol-4-yl)-3,4,7,9,10,11-hexahydro-6*H*-pyrido[2,1-i]indole (*epi-37*). The title compound was derived from *epi-30* and (1-methyl-1*H*-pyrazol-4yl)boronic acid to give the desired product (6.30 mg, 0.022 mmol, 33%) ¹**H NMR** (600 MHz, CDCl₃): δ 7.42 (s, 1H), 7.31 (s, 1H), 5.85–5.83 (m, 1H), 5.63–5.61 (m, 1H), 3.82 (s, 3H), 3.37–3.33 (m, 1H), 3.32 (s, 3H), 3.24 (ddd, *J* = 14.4, 11.5, 5.7 Hz, 1H), 3.00–2.95 (m, 3H), 2.58–2.46 (m, 2H), 2.38 (dddd, *J* = 14.8, 8.0, 4.3, 2.3 Hz, 1H), 2.32–2.26 (m, 2H), 1.84 (dt, *J* = 18.9, 5.4 Hz, 1H), 1.72–1.66 (m, 1H), 1.51 (dd, *J* = 14.0, 11.4 Hz, 1H). ¹³C{1H} **NMR** (151 MHz, CDCl₃): δ 142.7, 138.4, 133.8, 127.9, 123.7, 121.8, 119.0, 75.3, 64.5, 56.2, 45.9, 39.9, 39.0, 37.2, 30.1, 28.1, 18.8. **HRMS** (ESI) *m*/*z*: [M + H]⁺ Calcd for C₁₇H₂₄N₃O, 286.1914; Found, 286.1921. **Rf** = 0.10 (silica gel, 6:2 EtOAc/MeOH). [*α*]D²⁵ = +77.1° (*c* 0.55, CHCl₃). >99% *ee*.

(105, 11*a*S)-10-*Methoxy*-1-(1-*methyl*-1*H*-*imidazole*-5-*yl*)-3,4,7,9,10,11-*hexahydro*-6*H*-*pyrido*[2,1-*i*]*indole* (**38**). The title compound was derived from **30** and 1-methyl-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-*y*l)-1*H*-*imidazole* to give the desired product (6.00 mg, 0.021 mmol, 37%) ¹**H** NMR (600 MHz, CDCl₃): δ 7.26 (s, 1H), 6.53 (s, 1H), 5.59–5.55 (m, 1H), 5.43–5.39 (m, 1H), 3.39 (s, 3H), 3.14 (ddd, *J* = 14.4, 11.7, 5.9 Hz, 1H), 2.98 (s, 3H), 2.93 (dd, *J* = 14.5, 7.1 Hz, 1H), 2.89–2.85 (m, 2H), 2.49–2.32 (m, 4H), 2.16 (dd, *J* = 11.7, 4.0 Hz, 1H), 2.07 (dd, *J* = 17.7, 4.6 Hz, 1H), 1.87 (dt, *J* = 19.4, 5.4 Hz, 1H), 1.69–1.61 (m, 1H), 1.39 (t, *J* = 12.0 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 138.3, 137.6, 131.8, 130.4, 130.1, 128.7, 121.3, 73.9, 64.8, 56.1, 46.2, 42.6, 40.0, 31.9, 31.7, 27.8, 19.5. HRMS (ESI) *m/z*: [M + H]⁺ Calcd for C₁₇H₂₄N₃O, 286.1914; Found, 286.1914. Rf = 0.1 (silica gel, 6:2 EtOAc/MeOH). [α]D²⁵ = +31.9° (*c* 0.6, CHCl₃). >99% *ee*.

(105, 11*a*5)-10-Methoxy-1-((E)-4-(trifluoromethyl)styryl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (**39**). The title compound was derived from **30** and (E)-(3-(trifluoromethyl)styryl)boronic acid to give the desired product (11.0 mg, 0.029 mmol, 52%) ¹H NMR (600 MHz, CDCl₃): δ 7.55 (d, J = 8.1 Hz, 2H), 7.42 (d, J = 8.0 Hz, 2H), 6.68 (d, J = 16.0 Hz, 1H), 6.60 (d, J = 16.0 Hz, 1H), 6.11 (t, J = 4.0 Hz, 1H), 5.69–5.64 (m, 1H), 3.80 (tdd, J = 12.5, 6.3, 4.0 Hz, 1H), 3.33–3.24 (m, 4H), 3.05 (dd, J = 14.5, 7.4 Hz, 1H), 2.98 (m, 1H), 2.89 (m, 1H), 2.63–2.53 (m, 2H), 2.51–2.36 (m, 3H), 2.09–2.02 (m, 1H), 1.99 (dt, J = 19.9, 5.5 Hz, 1H), 1.52 (t, J = 11.8 Hz, 1H). ¹³**C** NMR{1H} (151 MHz, CDCl₃): δ 141.3, 139.9, 138.0, 132.1, 129.1 (q, J = 32.4 Hz), 126.5, 125.9, 125.7 (q, J = 3.8 Hz), 124.5 (d, J = 271.8 Hz), 124.0, 119.6, 73.8, 64.1, 56.2, 46.1, 42.2, 39.9, 32.1, 27.5, 20.0. HRMS (ESI) m/z: [M + H]⁺ Calcd for C₂₂H₂₅F₃NO, 376.1883; Found, 376.1891. **Rf** = 0.33 (silica gel, 6:2 EtOAc/MeOH). [α]_D²⁵ = +85.7° (c 1.0, CHCl₃). >99% ee.

(10R, 11aS)-10-Methoxy-1-((E)-4-(trifluoromethyl)styryl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (epi-39). The title compound was derived from epi-30 and (E)-(3-(trifluoromethyl)styryl)boronic acid to give the desired product (12.0 mg, 0.032 mmol, 54%) ¹**H NMR** (600 MHz, CDCl₃): δ 7.52 (d, J = 8.1 Hz, 2H), 7.41 (d, J = 8.1 Hz, 2H), 6.73 (d, J = 16.1 Hz, 1H), 6.58 (d, J = 16.1 Hz, 1H)1H), 6.14-6.12 (m, 1H), 5.68-5.65 (m, 1H), 3.43 (tt, J = 9.8, 4.9Hz, 1H), 3.34 (s, 3H), 3.25 (ddd, J = 14.1, 11.2, 5.6 Hz, 1H), 3.03-2.97 (m, 2H), 2.90 (ddd, J = 10.9, 9.4, 4.5 Hz, 1H), 2.60-2.51 (m, 2H), 2.44–2.38 (m, 1H), 2.32–2.26 (m, 2H), 2.04–1.97 (m, 1H), 1.93 (dt, J = 19.5, 5.5 Hz, 1H), 1.66 (dd, J = 14.2, 10.5 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 141.9, 141.2, 139.0, 130.4, 128.8 (q, J = 32.4 Hz), 127.0, 126.7, 126.5, 126.3, 125.5 (q, J = 3.8Hz), 124.3 (q, J = 271.7 Hz), 124.1, 118.4, 75.1, 63.6, 56.1, 45.8, 39.8, 37.8, 30.5, 27.6, 19.0. HRMS (ESI) m/z: [M + H]⁺ Calcd for $C_{22}H_{25}F_{3}NO$, 376.1883; Found, 376.1886. Rf = 0.23 (silica gel, 6:2) EtOAc/MeOH). $[\alpha]_D^{25} = +96.5^\circ$ (c 0.4, CHCl₃). >99% ee.

(10S,11aS)-10-Methoxy-1-(1-methyl-1H-indol-3-yl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (40). The title compound was derived from 30 and 1-methyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-indole to give the desired product (17.0 mg, 0.051 mmol, 90%) ¹H NMR (600 MHz, CDCl₃): δ 7.67 (d, J = 8.0 Hz, 1H), 7.28 (d, J = 8.0 Hz, 1H), 7.21 (t, J = 7.6 Hz, 1H), 7.11 (t, J = 7.4 Hz, 1H), 6.80 (s, 1H), 5.88-5.84 (m, 1H), 5.63-5.58 (m, 10H), 5.63-5.58 (m, 10H), 5.63-5.58 (m, 10H), 5.63-5.58 (m, 10H), 5.53-5.58 (1H), 3.74 (s, 3H), 3.43 (ddd, J = 14.3, 11.8, 6.1 Hz, 1H), 3.18-3.07 (m, 3H), 2.90-2.83 (m, 4H), 2.71-2.63 (m, 1H), 2.63-2.54 (m, 2H), 2.51 (dd, J = 11.3, 3.9 Hz, 1H), 2.13 (dt, J = 18.9, 5.4 Hz, 1H), 2.09-2.06 (m, 1H), 1.88-1.80 (m, 1H), 1.48 (t, J = 11.7 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 139.7, 136.5, 133.3, 128.1, 126.7, 125.4, 121.8, 120.9, 119.9, 119.4, 116.4, 109.1, 74.1, 65.3, 55.8, 46.2, 42.3, 40.2, 32.9, 32.1, 28.0, 19.7. HRMS (ESI) m/z: [M + H]⁺ Calcd for C₂₂H₂₇N₂O, 335.2118; Found, 335.2123. Rf = 0.24 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]_D^{25} = +172.4^{\circ} (c \ 1.0, CHCl_3). >99\% \ ee. (10R, 11aS)-10-Methoxy-1-(1-methyl-1H-indol-3-yl)-$

(10*R*, 11*a*5)-10-Methoxy-1-(1-methyl-1H-indol-3-yl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (*epi*-40). The title compound was derived from *epi*-30 and 1-methyl-3-(4,4,5,5tetramethyl-1,3,2-dioxaborolan-2-yl)-1H-indole to give the desired product (11.6 mg, 0.035 mmol, 63%) ¹H NMR (600 MHz, CDCl₃): δ 7.62 (d, *J* = 7.9 Hz, 1H), 7.25 (d, *J* = 3.7 Hz, 1H), 7.19 (t, *J* = 7.5 Hz, 1H), 7.10–7.07 (m, 2H), 5.99–5.97 (m, 1H), 5.64 (d, *J* = 7.8 Hz, 1H), 3.72 (s, 3H), 3.42–3.35 (m, 2H), 3.27 (s, 3H), 3.15–3.06 (m, 3H), 2.71–2.60 (m, 2H), 2.40–2.35 (m, 2H), 2.31–2.26 (m, 1H), 2.08–2.02 (m, 1H), 1.57–1.52 (m, 2H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 142.0, 136.7, 134.6, 128.7, 126.2, 125.4, 121.7, 120.0, 119.8, 119.3, 114.3, 109.1, 75.2, 65.7, 56.2, 46.1, 40.1, 37.2, 32.9, 30.0, 28.2, 18.9. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₂₂H₂₇N₂O, 335.2118; Found, 335.2118. **Rf** = 0.17 (silica gel, 6:2 EtOAc/ MeOH). [α]_D²⁵ = +200.9° (*c* 0.22, CHCl₃). >99% *ee*.

(105,11a5)-1-(Benzo[b]thiophen-3-yl)-10-methoxy-3,4,7,9,10,11hexahydro-6H-pyrido[2,1-i]indole (41). The title compound was derived from 30 and benzo[b]thiophen-3-ylboronic acid to give the desired product (9.60 mg, 0.028 mmol, 50%) ¹H NMR (600 MHz, CDCl₃): δ 7.83 (dddd, J = 15.2, 8.0, 1.2, 0.7 Hz, 2H), 7.41–7.31 (m, 2H), 6.99 (s, 1H), 5.84 (dd, J = 4.7, 2.8 Hz, 1H), 5.61 (d, J = 3.3 Hz, 1H), 3.46 (ddd, J = 14.3, 11.9, 6.0 Hz, 1H), 3.22–3.14 (m, 3H), 2.77–2.67 (m, 4H), 2.67–2.57 (m, 2H), 2.56–2.47 (m, 2H), 2.17 (dt, J = 19.2, 5.5 Hz, 1H), 2.08–2.00 (m, 1H), 1.84–1.76 (m, 1H), 1.54 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 139.7, 139.7, 138.3, 136.6, 134.5, 127.5, 124.4, 124.4, 123.5, 122.9, 122.8, 122.2, 73.6, 65.7, 55.9, 46.3, 41.6, 40.2, 32.0, 27.9, 19.4. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₂₁H₂₄NOS, 338.1573; Found, 338.1584. **Rf** = 0.23 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]_D^{25} = +68.7^\circ$ (*c* 0.96, CHCl₃). >99% *ee.*

(10R, 11aS)-1-(Benzo[b]thiophen-3-yl)-10-methoxy-3, 4, 7, 9, 10, 11hexahydro-6H-pyrido[2,1-i]indole (epi-41). The title compound was derived from epi-30 and benzo[b]thiophen-3-ylboronic acid to give the desired product (4.50 mg, 0.013 mmol, 24%) ¹H NMR (600 MHz, C_6D_6): δ 7.79 (d, J = 8.1 Hz, 1H), 7.58 (d, J = 8.0 Hz, 1H), 7.21 (t, J = 7.6 Hz, 1H), 7.13 (s, 1H), 7.08 (t, J = 7.5 Hz, 1H), 5.68-5.65 (m, 1H), 5.31–5.28 (m, 1H), 3.39–3.33 (m, 1H), 3.23 (ddd, J = 14.3, 11.6, 5.5 Hz, 1H), 3.02 (s, 3H), 3.00-2.95 (m, 1H), 2.87 (td, J = 10.1, 9.5, 4.5 Hz, 1H), 2.79 (dd, J = 14.5, 6.7 Hz, 1H), 2.54-2.47 (m, 2H), 2.26-2.19 (m, 1H), 2.18-2.13 (m, 1H), 2.10-2.04 (m, 1H), 1.74 (dd, J = 13.9, 10.7 Hz, 1H), 1.51 (dt, J = 18.7, 5.4 Hz, 1H), 1.39–1.34 (m, 1H). ¹³C{1H} NMR (151 MHz, C_6D_6): δ 142.0, 140.9, 140.1, 137.0, 136.3, 128.4, 124.5, 124.3, 123.2, 123.0, 122.9, 120.2, 75.7, 65.7, 55.7, 46.5, 40.3, 38.8, 30.0, 28.7, 18.7. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₂₁H₂₄NOS, 338.1573; Found, 338.1572. Rf = 0.17 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]_D^{25} = +91.6^\circ$ (c 0.24, CHCl₃). >99% ee.

(105,11a5)-10-Methoxy-1-(thiophen-3-yl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (42). The title compound was derived from **30** and thiophen-3-ylboronic acid to give the desired product (12.5 mg, 0.044 mmol, 77%) ¹H NMR (600 MHz, CDCl₃): δ 7.17 (dd, *J* = 5.0, 3.0 Hz, 1H), 7.04 (dd, *J* = 3.0, 1.2 Hz, 1H), 6.94 (dd, *J* = 5.0, 1.3 Hz, 1H), 5.81–5.79 (m, 1H), 5.60–5.57 (m, 1H), 3.28 (ddd, *J* = 14.3, 11.8, 6.1 Hz, 1H), 3.11 (s, 3H), 3.08–2.98 (m, 3H), 2.92 (dddd, *J* = 12.5, 8.8, 6.2, 3.9 Hz, 1H), 2.64–2.50 (m, 3H), 2.40 (dd, *J* = 11.4, 3.9 Hz, 1H), 2.26–2.18 (m, 1H), 1.94 (dt, *J* = 19.1, 5.4 Hz, 1H), 1.86 (dddd, *J* = 17.4, 8.7, 3.5, 1.9 Hz, 1H), 1.46 (t, *J* = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 144.1, 139.9, 137.2, 128.7, 125.6, 124.6, 121.6, 120.8, 74.1, 64.5, 56.1, 46.1, 43.0, 40.1, 32.0, 28.0, 19.7. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₁₇H₂₂NOS, 288.1417; Found, 288.1420. **Rf** = 0.18 (silica gel, 6:2 EtOAc/ MeOH). [α]_D²⁵ = +50.2° (*c* 0.5, CHCl₃). >99% *ee*.

(10*R*,11*a*5)-10-Methoxy-1-(thiophen-3-yl)-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (epi-42). The title compound was derived from epi-30 and thiophen-3-ylboronic acid to give the desired product (11.0 mg, 0.038 mmol, 45%). ¹H NMR (600 MHz, CDCl₃): δ 7.18 (s, 1H), 7.16–7.14 (m, 1H), 7.02 (d, *J* = 5.0 Hz, 1H), 5.90– 5.88 (m, 1H), 5.63–5.60 (m, 1H), 3.36–3.24 (m, 5H), 3.04–2.98 (m, 3H), 2.60–2.53 (m, 2H), 2.37–2.28 (m, 3H), 1.89 (dt, *J* = 18.9, 5.5 Hz, 1H), 1.56 (dd, *J* = 14.3, 11.5 Hz, 1H), 1.53–1.48 (m, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 142.0, 141.6, 137.9, 128.5, 125.4, 124.3, 121.8, 119.6, 75.3, 64.8, 56.2, 46.0, 39.9, 37.5, 29.8, 28.1, 18.91. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₁₇H₂₂NOS, 288.1417; Found, 288.1426. **Rf** = 0.21 (silica gel, 6:2 EtOAc/ MeOH). [α]D²⁵ = +83.9° (*c* 0.03, CHCl₃). >99% ee.

(10S,11aS)-1-((E)-3-Fluorostyryl)-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (43). The title compound was derived from 30 and (E)-(3-fluorostyryl)boronic acid to give the desired product (17.0 mg, 0.052 mmol, 92%) ¹H NMR (600 MHz, CDCl₃): δ 7.26 (td, J = 8.0, 6.0 Hz, 1H), 7.09 (d, J = 7.7 Hz, 1H), 7.03 (dt, J =10.3, 2.1 Hz, 1H), 6.94–6.87 (m, 1H), 6.62 (d, J = 15.9 Hz, 1H), 6.51 (d, J = 15.9 Hz, 1H), 6.07 (t, J = 4.0 Hz, 1H), 5.67-5.64 (m, 1H),3.81 (dddd, J = 12.6, 8.8, 6.4, 4.0 Hz, 1H), 3.26 (s, 4H), 3.05 (dd, J = 14.4, 7.3 Hz, 1H), 2.98 (td, J = 9.0, 3.8 Hz, 1H), 2.89 (td, J = 9.5, 9.1, 6.6 Hz, 1H), 2.62-2.52 (m, 2H), 2.50-2.44 (m, 1H), 2.43-2.36 (m, 2H), 2.08-2.01 (m, 1H), 2.01-1.94 (m, 1H), 1.52 (t, J = 11.8 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 163.3 (d, J = 245.0 Hz), 140.2 (d, J = 7.7 Hz), 137.9, 130.8, 130.2 (d, J = 8.3 Hz), 126.3 (d, J = 2.7 Hz), 123.3, 122.4 (d, J = 2.8 Hz), 119.6, 114.2 (d, J = 21.5 Hz), 112.7 (d, J = 21.9 Hz), 73.8, 64.1, 56.2, 46.2, 42.2, 39.9, 32.0, 27.4, 20.0. HRMS (ESI) m/z: $[M + H]^+$ Calcd for C₂₁H₂₅NOF, 326.1915; Found, 326.1923. **Rf** = 0.35 (silica gel, 6:2 EtOAc/MeOH). $[\alpha]_D^{25}$ = +41.3° (c 1.0, CHCl₃). >99% ee.

(10R, 11aS)-1-((E)-3-Fluorostyryl)-10-methoxy-3,4,7,9,10,11-hexahydro-6H-pyrido[2,1-i]indole (epi-43). The title compound was derived from epi-30 and (E)-(3-fluorostyryl)boronic acid to give the desired product (6.50 mg, 0.020 mmol, 35%). ¹H NMR (600 MHz,

CDCl₃): δ 7.23 (td, *J* = 8.0, 6.0 Hz, 1H), 7.09 (d, *J* = 8.0 Hz, 1H), 7.03 (dt, *J* = 10.4, 2.1 Hz, 1H), 6.88 (td, *J* = 8.5, 2.6 Hz, 1H), 6.67 (d, *J* = 16.1 Hz, 1H), 6.49 (d, *J* = 16.1 Hz, 1H), 6.10–6.06 (m, 1H), 5.68–5.63 (m, 1H), 3.41 (tt, *J* = 9.8, 4.7 Hz, 1H), 3.35 (s, 3H), 3.24 (ddd, *J* = 14.3, 11.4, 5.7 Hz, 1H), 3.02–2.93 (m, 2H), 2.90 (ddd, *J* = 11.0, 9.4, 4.5 Hz, 1H), 2.61–2.48 (m, 2H), 2.45–2.36 (m, 1H), 2.32–2.24 (m, 2H), 2.08–1.95 (m, 1H), 1.90 (dt, *J* = 19.4, 5.4 Hz, 1H), 1.64 (dd, *J* = 14.2, 10.7 Hz, 1H). ¹³C{1H} NMR (151 MHz, CDCl₃): δ 163.3 (d, *J* = 245.0 Hz), 142.3, 140.3 (d, *J* = 7.7 Hz), 139.3, 130.1 (d, *J* = 8.7 Hz), 129.4, 126.9 (d, *J* = 2.7 Hz), 123.7, 122.4 (d, *J* = 2.6 Hz), 118.2, 114.0 (d, *J* = 21.5 Hz), 112.6 (d, *J* = 21.6 Hz), 75.4, 63.7, 56.2, 45.9, 39.9, 38.0, 30.6, 27.8, 19.1. HRMS (ESI) *m*/*z*: [M + H]⁺ Calcd for C₂₁H₂₅NOF, 326.1915; Found, 326.1917. Rf = 0.33 (silica gel, 6:2 EtOAc/MeOH). [α]D²⁵ = +74.4° (*c* 0.54, CHCl₃). >99% *ee*.

Pharmacology. *Materials.* (*S*)-Nicotine and all chemicals for the buffers were purchased from Sigma-Aldrich (St. Louis, MO), and Dulbecco's Modified Eagle Medium Glutamax-I (DMEM) medium, serum, Hanks Buffered Salt Solution (HBSS), and antibiotics were obtained from Invitrogen (Paisley, UK). [³H]epibatidine and the Fluo-4/AM dye were purchased from PerkinElmer (Waltham, MA) and Molecular Probes (Eugene, OR), respectively. The stable $h\alpha 4\beta 2$ - and $h\alpha 6/\alpha 3\beta 2\beta 3^{V9'S}$ -HEK293 cell lines were obtained from Dr. Tino Dyhring (Saniona A/S, Ballerup, Denmark), and the stable $r\alpha 3\beta 4$ -

and r α 4 β 4-HEK293 cell lines were obtained from Dr. Kenneth J. Kellar (Georgetown University Medical Center, Washington, D.C.).² *Cell Culture*. The four nAChR-expressing HEK293 cell lines were cultured in a humidified atmosphere at 37 °C and 5% CO₂ in DMEM

cultured in a humidified atmosphere at 37 °C and 5% CO₂ in DMEM supplemented with penicillin (100 U/mL), streptomycin (100 μ g/mL), 5% dialyzed fetal bovine serum, and, for the $r\alpha 3\beta 4$ - and $r\alpha 4\beta 4$ -HEK293 cell lines, 1 mg/mL G-418.

[³H]epibatidine-Binding Assay. The binding affinities of the test compounds were determined at membranes from the h α 4 β 2-, h α 6/ $\alpha 3\beta 2\beta 3^{V9'S}$ -, $r\alpha 3\beta 4$ -, and $r\alpha 4\beta 4$ -HEK293 cell lines in a [³H]epibatidine competition binding assay, as previously described.^{25a,l} Cells were collected at ~90% confluency by scraping into ice-cold assay buffer (140 mM NaCl, 1.5 mM KCl, 2 mM CaCl₂, 1 mM MgCl₂, 25 mM HEPES, pH 7.4), homogenized for 10 s, and centrifuged for 30 min at 47,810 ×g at 4 °C. The pellet was resuspended in ice-cold fresh assay buffer, homogenized, and centrifuged for another 30 min under the same conditions. The supernatant was decanted, and the cell membranes were stored at -80 °C until use. On the day of the experiment, cell membranes were resuspended in assay buffer and incubated with [3H]epibatidine and various concentrations of test compounds at room temperature for 3-4 h while shaking. In the competition binding experiments, membranes were incubated with a fixed concentration of [³H]-

epibatidine h $\alpha 4\beta 2$ and h $\alpha 6/\alpha 3\beta 2\beta 3^{V9'S}$: ~30 pM (range: 20–40 pM) and $r\alpha 3\beta 4$ and $r\alpha 4\beta 4$: ~150 pM (range: 100-200 pM) in the presence of various concentrations of the test compounds or (S)nicotine in a total reaction volume of 3 mL. In saturation binding experiments performed in parallel, the membranes were incubated with 12 different concentrations of [³H]epibatidine (spanning from 1 pM to 30 nM) in the absence (total binding) or presence of 100 μ M (S)-nicotine (non-specific binding). In both saturation and competition binding experiments, the amount of cell membranes used and total reaction volumes were adjusted so that the bound/free ratios of the radioligand always were <10%. Prior to harvesting, Whatman GF/ C filters (PerkinElmer) were treated with 0.2% polyethyleneimine solution for 30 min. The reaction was terminated by filtration through the presoaked filters using a Brandell M-48T cell harvester (Alpha Biotech, London, UK) and washed three times with 5 mL ice-cold harvesting buffer (0.9% NaCl, 10 mM Tris-HCl, pH 7.4). Filters were subsequently allowed to dry at room temperature and then transferred to scintillation vials, and 3 mL OptiFluor (PerkinElmer) was added to each sample. The filter-retained radioactivity and the exact [3H]epibatidine concentrations used in the experiments were determined by liquid scintillation counting using a Tri-Carb 4910 TR

(PerkinElmer). The test compounds were tested at least three times at each receptor.

 $Ca^{2+}/Fluo-4$ Assay. The functional properties of the test compounds were characterized at the h α 4 β 2-HEK293 and r α 3 β 4-HEK293 cell lines in the fluorescence-based Ca²⁺/Fluo-4 assay performed essentially, as previously described.^{25b} Briefly, the cells were split into poly-D-lysine-coated black 96-well plates with clear bottom (6 \times 10⁴ cells/well). The following day the culture medium was aspirated, and the cells were incubated in 50 μ L of incubation buffer (HBSS containing 20 mM HEPES, 1 mM CaCl₂, 1 mM MgCl₂, and 2.5 mM probenecid, pH 7.4) supplemented with 6 mM Fluo-4/ AM at 37 °C for 1 h. Then, the buffer was aspirated, the cells were washed once with 100 μ L of incubation buffer, once with 100 μ L of assay buffer (140 mM N-methyl-D-glucamine, 5 mM KCl, 1 mM MgCl₂, 10 mM CaCl₂, 10 mM HEPES, and 2.5 mM probenecid, pH 7.4), and then, 100 μ L of assay buffer was added to the cells (in the antagonist experiments, the test compound was added to the buffer at this point). The 96-well plate was assayed in a FLEXStation³ (Molecular Devices, Crawley, UK) measuring emission [in fluorescence units (FU)] at 525 nm caused by excitation at 485 nm before and up to 90 s after addition of 33.3 μ L agonist solution in assay buffer. The compounds were initially tested as agonists at the two receptors, and none of them displayed significant agonist activity at concentrations up to 100 μ M. The compounds were characterized as antagonists in duplicate at least three times at the two receptors using (S)-nicotine ~ EC_{80} (range EC_{70} - EC_{90}) as an agonist.

Data Analysis. Data analysis was performed using GraphPad Prism 7.0c (GraphPad Software, Inc., La Jolla, CA). For the saturation [³H]epibatidine-binding data, specific binding was analyzed with a nonlinear regression one site-specific binding model given by the equation: $Y = B_{\text{max}} \times X/(K_{\text{d}} + X)$, where Y is the specific $[{}^{3}H]$ epibatidine binding, B_{max} is the total amount of binding, X is the $[^{3}H]$ epibatidine concentration, and K_{d} is the dissociation constant. The competition binding data were analyzed with a nonlinear regression one site-fit log $\bar{\text{IC}}_{50}$ model given by the following equation: $Y = bottom + (top - bottom)/(1 + 10((logIC_{50} - X)*n_H))$ where Y is the specific $[{}^{3}H]$ epibatidine binding, top and bottom are the plateau values of the curves, X is the test compound concentration, IC₅₀ is the equilibrium affinity of the test compound, and $n_{\rm H}$ is the Hill slope. K_i values were calculated from the determined IC₅₀ values using the Cheng–Prusoff equation [32]: $K_i =$ $IC_{50}/(1 - ([RL]/K_d))$, where [RL] is the [³H]epibatidine concentration used for the specific experiment and $K_{\rm D}$ is the dissociation constant determined in the saturation binding experiments.

The data from the Ca²⁺/Fluo-4 assay were extracted as the difference in relative fluorescence units (Δ RFU) between the maximum fluorescence level measured after the agonist application and the basal level measured before the agonist application. The concentration—inhibition relationships of the test compounds as antagonists were fitted to a nonlinear regression curve fit with variable slope *Y* = bottom + (top – bottom)/(1 + 10((logIC₅₀ – *X*)**n*_H), where top and bottom values are plateaus in the units of the response axis, *X* is the concentration of the ligand, IC₅₀ is the concentration of the ligand that gives a response half way between bottom and top, and *n*_H is the Hill slope.

Molecular Modeling. The modeling study was performed using the Drug Discovery Suite 2020–3 from Schrodinger Inc.

The structure from a study by Yu *et al.*⁶⁶ was processed using the standard protocol for the protein preparation wizard in standard settings. A model including the water molecule seen in the X-ray structure of the AChBP (PDB: 4alx) was also made for docking in the same way. All ligands were treated with ligprep before docking, however, in the protonated state of the amine and without generating other stereoisomers or tautomers. Two grids were generated in Glide in standard settings: with and without a water molecule included. All ligands were docked with Glide in both models allowing up to five poses for each ligand. The best scoring poses in the model including water are provided in the Supporting Information. Figures were generated using Pymol 1.8.0.4.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.joc.1c00707.

Experimental details and docking poses of investigated ligands (PDF)

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The authors declare no competing financial interest.

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