A Convenient Synthesis of Oxazolidinone Derivatives Linezolid and Eperezolid from (S)-Glyceraldehyde Acetonide

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ABSTRACT: A new convenient method for the synthesis of oxazolidinone antibacterials Linezolid and Eperezolid from readily available (S)-glyceraldehyde acetonide was developed. The key steps include reductive amination of arylamine and (S)-glyceraldehyde acetonide in the presence of NaBH₄ and 4 Å sieve, followed by hydrolysis and regioselective cyclization. © 2008 Wiley Periodicals, Inc. Heteroatom Chem 19:316–319, 2008; Published online in Wiley InterScience (www.interscience.wiley.com). DOI 10.1002/hc.20435

INTRODUCTION

Oxazolidinones as a new class of synthetic antimicrobial agents are active against numerous multidrug-resistant Gram-positive organisms [1]. Linezolid **1a** and Eperezolid **1b** (Fig. 1), which are well known as the promising candidates of this family, work effectively against numerous serious Grampositive human pathogens such as MRSA (Methicillin resistant Staphylococcus aureus) and VRE (Vancomycin-resistant Enterococci) [2–4]. Linezolid has recently received approval for the treatment of community- and hospital-acquired pneumonia and skin structure infections.

Several methods are available for the synthesis of Linezolid and Eperezolid. Brickner et al. [2] and Schaus and Jacobson [5] reported separately a route to oxazolidinones with good yields. However, in these methods for the synthetic key step to oxazolidinone ring, severe conditions with low temperature (-78° C) and an air-sensitive base (*n*-BuLi) were required, which limit the large-scale production in the industry. Lohray et al. [6] offered another possibility to oxazolidinones via asymmetric bis-epoxide using D-mannitol as a starting material. However, the synthesis of asymmetric bis-epoxide suffers from a long-synthetic route and was reported without any optical data.

We wish to report herein a very convenient and efficient method for the synthesis of Linezolid and Eperezolid, using readily available (*S*)glyceraldehyde acetonide as a chiral-building block.

RESULTS AND DISCUSSION

The synthesis of Linezolid and Eperezolid share a common route, as detailed in Scheme 1. (S)-Glyceraldehyde acetonide (3) was easily obtained



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FIGURE 1 Linezolid (1a) and Eperezolid (1b).

from L-ascorbic acid (vitamin C) via hydrogenation [7], acetonation, and oxidative cleavage [8,9]. Reductive amination of fluoroarylamine 2 with 3 in the presence of NaBH₄ and 4 Å molecular sieve at room temperature afforded dioxolanes 4 with good yields. The yields were considerably poor when Pd/C or Raney Ni was used as a catalyst in the reductive amination step, because unstable (S)-glyceraldehyde acetonide may polymerize during long reaction time. The addition of molecular sieve to NaBH₄ system could shorten the reaction time and minimize the polymerization of glyceraldehyde acetonide in the solvent. After hydrolysis of dioxolanes 4 in aqueous HCl/methanol, the regioselective cyclization with triphosgene in the presence of K₂CO₃ afforded oxazolidinones 5. However, if NaOH [10] or triethylamine was used as an acid scavenger, the byproduct 5c could form. The key intermediate 5 was activated as mesylate and then displaced with NaN₃ to give 6. Hydrogenation of the azide 6a over Pd/C and then treatment with Ac₂O provided the targeted antibacterial Linezolid. In our laboratory, we tried to synthesize 1a via direct displacement of mesylate by acetamide anion in DMF or THF, and the vield was very low (10%) because 5a mesylate was unstable and degraded under this condition. Reductive acetvlation of azide **6b** with thioacetic acid [11] gave acetamide 7. The compound 7 after side-chain manipulation [2] provided the targeted antibacterial Eperezolid with good yield. The final product's melting points and specific rotation data were identical to those reported in the literature.

In summary, we describe herein a simple, mild, and facile synthesis of Linezolid and Eperezolid from (*S*)-glyceraldehyde acetonide with reductive amination and regioselective cyclization as key steps with good yields.

EXPERIMENTAL

Melting points were determined with a Büchi 510 melting point apparatus and are uncorrected. ¹H and ¹³C NMR spectra were recorded in CDCl₃ on a Bruker-300 NMR spectrometer, and chemical shifts δ are in ppm (TMS was used as an internal standard). EI mass spectra were accomplished at 70 eV, using a MAT 90 spectrometer. Microanalyses were performed on a Leco CHN-2000 elemental analyzer. Optical rotations were taken on a Perkin-Elmer 341 polarimeter at the sodium-D line. Purification by column chromatography was carried out with silica gel 60 (70–230).

N-[4(R)-(2,2-Dimethyl-1,3-dioxolane-4-yl)methyl]-3-fluoro-4-morpholinylaniline (**4a**). A solution of **2a** (5.00 g, 25.5 mmol) and **3** (6.00 g, 46.1 mmol) in 25 mL THF and 25 mL methanol was added 6.25 g of molecular sieves (4 Å), then NaBH₄ (0.97 g, 25.6 mmol) was added portionwise. The mixture was stirred for 10 h at room temperature, ice water (150 mL) was added, and the molecular sieves were removed by filtration. The filtrate was extracted



SCHEME 1 Reagents and conditions: (a) NaBH₄, MeOH, 4 Å MS, r.t., 78%; (b) (*i*) H_3O^+ , MeOH, r.t.; (*ii*) K_2CO_3 , triphosgene, CH₂Cl₂, 30°C, 80%; (c) CH₃SO₂Cl, triethylamine, CH₂Cl₂, 0°C \rightarrow r.t., 88%; (d) NaN₃, DMF, 80°C, 90%–92%; (e) (*i*) Pd/C, H₂, EtOH, r.t.; (*ii*) pyridine, Ac₂O, r.t., 51%; (f) CH₃COS H, r.t., 75%; (g) Pd/C, MeOH/EtOAc, r.t.; (h) ClCOCH₂OBn, triethylamine, CH₂Cl₂, 0°C \rightarrow r.t.; (i) Pd/C, H₂, MeOH, 73%.

with ethyl acetate $(50 \times 3 \text{ mL})$ was and dried with Na_2SO_4 . After removal of the solvent, 6.74 g of 4a as a pale yellow oil was obtained without further purification. The analytic sample was obtained by chromatography on a silica-gel column (mobile phase PE/EtOAc from 6:1 to 4:1). $[\alpha]_{\rm D}^{20}$ -1.7° (*c* 4.54, CHCl₃); ¹H NMR(CDCl₃): δ 6.84 (t, 1H, J = 8.2 Hz), 6.39 (m, 2H), 4.35 (m, 8 lines, 1H), 4.09 (dd, 1H, J = 7.1, 8.3 Hz), 3.84 (t, 4H, J = 4.7 Hz), 3.75 (dd, 1H, *J* = 6.0, 8.2 Hz), 3.24 (dd, 1H, *J* = 3.7, 12.3 Hz), 3.13 (dd, 1H, J = 6.6, 12.5 Hz), 2.95 (br s, 4H), 1.44 (s, 3H), 1.37 (s, 3H); ¹³C NMR (CDCl₃): δ 25.24, 26.86, 46.92, 51.73, 67.07, 67.12, 74.32, 101.70 (d, *J* = 24.35 Hz), 108.50, 109.50, 120.25, 130.99, 144.77 (d, J = 9.56 Hz), 156.94 (d, J = 244.71 Hz); MS: m/z310 (61, M⁺), 219 (12), 209 (100), 151 (16).

Compounds **4b** was prepared by using the method described for the preparation of **4a** except that the **4b** was recrystallized from petroleum ether and ethyl acetate.

N-[4(R)-(2,2-Dimethyl-1,3-dioxolane-4-yl)methyl]-3-fluoro-4-(N-carbobenzoxypiperazinyl) Aniline (4b). Yield 78%, mp 92.5—94°C, $[\alpha]_{D}^{20}$ –1.0° (*c* 1.14, CHCl₃); ¹H NMR (CDCl₃): δ 7.35 (m, 5H), 6.81 (t, 1H, J = 7.6Hz), 6.36 (m, 2H), 5.16 (s, 2H), 4.35 (m, 8 lines, 1H), 4.10 (dd, 1H, *J* = 6.4, 8.3 Hz), 3.91 (br s, 1H), 3.76 (dd, 1H, *J* = 6.1, 8.2 Hz), 3.63 (br s, 4H), 3.25 (dd, 1H, J = 4.2, 12.5 Hz), 3.13 (dd, 1H, J = 6.7, 12.6 Hz), 2.89 (br s, 4H), 1.45 (s, 3H), 1.37 (s, 3H); ¹³C NMR (CDCl₃): δ 25.24, 26.86, 44.00, 46.86, 51.33, 67.06, 67.16, 74.29, 101.57 (d, J = 24.35Hz), 108.51, 109.52, 120.88, 127.89, 128.02, 128.48, 130.71, 136.61, 145.09, 155.18, 157.03 (d, *J* = 245.06 Hz); MS: *m*/*z* 443 (96, M⁺), 342 (25), 245 (66), 187 (27), 151 (24), 127 (20), 91 (100); Anal. calcd for C₂₄H₃₀FN₃O₄: C, 64.99; H, 6.82; N, 9.47. Found: C, 65.02; H, 6.79; N, 9.39.

(R)-[N-3-(3-Fluoro-4-morpholinylphenyl)-2-oxo-5-oxazolidinyl]methanol (5a). A solution of benzamine (4a, 4.56 g) in dry HCl methanol (4 M, 40 mL) was added 0.5 mL of water, the mixture was stirred at room temperature for 8 h, and concentrated with rotary evaporator, the residue was diluted with methyl chloride (35 mL) then added aq. Na₂CO₃ solution (15%, 40 mL, 56.5 mmol). The mixture was cooled to 0°C and treated dropwise with a solution of triphosgene (1.30 g, 4.38 mmol) in methyl chloride (35 mL) for 30 min. After being stirred for 24 h at room temperature, the water phase was separated and extracted with methyl chloride $(50 \times 2 \text{ mL})$. The combined organic phase was washed with water $(25 \times 2 \text{ mL})$ and dried with anhydrous Na₂SO₄. The solvent was evaporated at reduced pressure, and

the residue was purified by crystallization from ethyl acetate and petroleum ether to provide 2.84 g (56% from **2a**) of **5a** as a light gray amorphous solid. mp 114–116°C, $[\alpha]_D^{20}-53^\circ$ (*c* 0.69, CHCl₃) (lit. [2] mp 112–114°C, $[\alpha]_D^{20}-54^\circ$ (*c* 0.990, CHCl₃)); ¹H NMR(CDCl₃): δ 7.47 (dd, 1H, J = 2.1, 14.4 Hz), 7.12 (dt, 1H, J = 2.4, 8.9 Hz), 7.00 (br s, 1H), 4.75 (m, 1H), 3.95 (m, overlapping, 3H), 3.88 (t, 4H, J = 4.7Hz), 3.76 (d, 1H, J = 13.1 Hz), 3.08 (t, 4H, J = 4.2Hz), 2.35 (br s, 1H); MS: m/z 296 (100, M⁺), 238 (55), 149 (20), 57 (36); Anal. calcd for C₁₄H₁₇FN₂O₄: C, 56.75; H, 5.78; N, 9.45. Found: C, 57.02; H, 5.78; N, 9.28.

(*R*)-[*N*-3-[3-Fluoro-4-[*N*-1-(4-carbobenzoxy)piperazinyl]-phenyl]-2-oxo-5-oxazolidinyl]methanol (**5b**). Yield 78%, mp 155–157°C, $[\alpha]_D^{20}-35^\circ$ (c 1.01, CHCl₃), (lit. [2] mp 156–157°C, $[\alpha]_D^{20}-32^\circ$ (c 0.991, CHCl₃)); ¹H NMR (CDCl₃): δ 7.52 (d, 1H, *J* = 14.4 Hz), 7.36 (m, 5H), 7.12 (m, 2H), 5.16 (s, 2H), 4.75 (m, 1H), 4.00 (m, 3H), 3.73 (m, 5H), 3.07 (br s, 4H).

6a and **6b** were prepared according to the method described in [2].

(*R*)-[*N*-3-(3-Fluoro-4-morpholinylphenyl)-2-oxo-5-oxazolidinyl]methyl Azide (**6a**). Yield 79% (from **5a**). ¹H NMR (CDCl₃): δ 7.45 (dd, 1H, *J* = 2.5, 14.3 Hz), 7.13 (ddd, 1H, *J* = 1.1, 2.6, 8.9 Hz), 6.95 (t, 1H, *J* = 9.1 Hz), 4.78 (m, 1H), 4.05 (t, 1H, *J* = 8.8 Hz), 3.88 (t, 4H, *J* = 4.6 Hz), 3.83 (dd, 1H, *J* = 6.2, 8.8 Hz), 3.71 (dd, 1H, *J* = 4.6, 13.3 Hz), 3.59 (dd, 1H, *J* = 4.3, 13.1 Hz), 3.06 (t, 4H, *J* = 4.6 Hz).

(*R*)-[*N*-3-[3-Fluoro-4-[*N*-1-(4-carbobenzoxy)piperazinyl]-phenyl]-2-oxo-5-oxazolidinyl]methyl Azide (**6b**). Yield 81% (from **5b**). ¹H NMR (CDCl₃): δ 7.46 (dd, 1H, *J* = 2.6, 14.3 Hz), 7.35 (m, 5H), 7.12 (ddd, 1H, *J* = 0.8, 2.5, 8.6 Hz), 6.93 (t, 1H, *J* = 9.1 Hz), 4.78 (m, 8 lines, 1H), 4.05 (t, 1H, *J* = 8.8 Hz), 3.83 (dd, 1H, *J* = 6.3, 8.9 Hz), 3.68 (m, 5H), 3.59 (dd, 1H, *J* = 4.4, 13.2 Hz), 3.01 (br s, 4H); MS: *m*/*z* 454 (14, M⁺), 424 (38), 329 (32), 289 (16), 165 (16), 91 (100), 56 (37).

(S)-*N*-[[3-(3-*Fluoro*-4-*morpholinylphenyl*)-2-oxo-5-oxazolidinyl]*methyl*]*acetamide* (**1a**, *Linezolid*). **1a** was prepared according to the method described in [2], except that the solvent was replaced by ethanol when reductive hydrogenation of azide **6a**. Yield 51% (from **6a**). mp 179–180.5°C, $[\alpha]_D^{20}$ -10° (*c* 1.52, CHCl₃) (lit. [2] mp 181.5–182.5°C, $[\alpha]_D^{20}$ -9° (*c* 0.919, CHCl₃); ¹H NMR (CDCl₃): δ 7.46 (dd, 1H, *J* = 2.2, 14.6 Hz), 7.08 (dd, 1H, *J* = 1.8, 9.1 Hz), 6.96 (t, 1H, *J* = 9.1 Hz), 6.05 (t, 1H, *J* = 6.0 Hz), 4.77 (m, 1H), 4.02 (t, 1H, J = 8.9 Hz), 3.88 (t, 4H, J = 4.5 Hz), 3.73 (dd, 1H, J = 1.8, 7.0 Hz), 3.69 (dd, 1H, J = 3.2, 5.9 Hz), 3.61 (dt, 1H, J = 5.5, 8.5 Hz), 3.07 (t, 4H, J = 4.6 Hz), 2.02 (s, 3H); MS: m/z 337 (27, M⁺), 293 (25), 234 (20), 209 (26), 149 (50), 91 (100); Anal. calcd for C₁₆H₂₀FN₃O₄: C, 56.97; H, 5.98; N, 12.46. Found: C, 57.16; H, 6.03; N, 12.22.

(S)-N-[[3-[3-Fluoro-4-[N-1-(4-hydroxyacetyl)piperazinvl]-phenvl]-2-oxo-5-oxazolidinvl]methvl]acetamide (1b, Eperezolid). 1b was prepared from 6b according to the method described in [2,11], yield 55% (from **6b**). mp 172–174°C, $[\alpha]_{\rm D}^{20}$ –20° (c 0.93, DMSO), (lit. [2] mp 175–176°C, $[\alpha]_{D}^{20}$ –21° (*c* 0.853, DMSO); ¹H NMR (CDCl₃): δ 7.50 (dd, 1H, J = 2.5, 14.0 Hz), 7.08 (d, 1H, J = 8.7 Hz), 7.00 (t, 1H, J = 9.0Hz), 6.06 (t, 1H, J = 2.7 Hz), 4.77 (m, 1H), 4.21 (s, 2H), 4.02 (t, 1H, J = 9.0 Hz), 3.87 (t, 2H, J = 5.1Hz), 3.75 (dd, 1H, J = 6.8, 9.1 Hz), 3.62 (m, 2H), 3.48 (t, 2H, J = 5.0 Hz), 3.09 (t, 4H, J = 4.3 Hz), 2.01 (s, 3H); MS: *m*/*z* 394 (28, M⁺), 350 (56), 306 (27), 165 (26), 56 (100). Anal. calcd for C₁₈H₂₃FN₄O₅ C, 54.82; H, 5.88; N, 14.21. Found: C, 54.86; H, 5.99; N, 13.98.

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