PROSTAGLANDINS: TOTAL SYNTHESIS OF PGD2 "via" 1,3 CYCLOPENTANEDIONE.

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Abstract: A practical total synthesis of PGD₂ starting from an acyclic precursor is reported. A novel efficient inversion of stereochemistry at C_9 tosylate group "via" a SN₂ displacement by tetrabutylammonium nitrate is described.

Prostaglandins (PGs) of the D series and their metabolites, whose biological significance was little appreciated earlier, have recently been revaluated in view of their potent ability to inhibit blood platelet aggregation^{1a}, their antitumor activity^{1b}, and other unique physiological properties^{1c,d}. Although much effort has been devoted to their synthesis², few of them avoid an oxidative step for the introduction of the C₁₁ carbonyl function. Of particular synthetic interest, from this point of view, is, in our opinion, the base-induced cyclization of γ -keto-esters which lead directly to cyclopentanones suitably functionalized with proper side chains at the desired position for further elaboration into PGD₂.



The drawback of this method is the lack of reliable chemoselective methods of discriminating between the two ring carbonyls. Recently we have reported an attractive solution to this problem which allowed the synthesis of the highly substituted cyclopentane system as in $(\underline{A})^3$. The solution proposed includes a sequential bis-acetalization-selective mono-de-acetalization of the cyclopentanedione (\underline{A}) . We wish to report here an application of this synthesis of PGD₂.

The readily available cyclopentanedione $(\underline{2})$, easily obtained from γ -ketoester $(\underline{1})$ via base-induced cyclization^{3C}, was easily converted to the monoethylene acctal $(\underline{3})$ in 70% overall yield by the following sequence:

a) acetalization by exposure to an excess of ethylene glycol in the presence of a catalytic amount of p-toluenesulfonic acid (PTSA) in benzene at reflux for 8 hours, b) selective deprotection of the C₉ carbonyl group by stirring the product with SiO₂ and a catalytic 1386

amount of aqueous $H_{2}SO_{1}$ in methylene dichloride for 5 hours⁴.

The 9-keto group of $(\underline{3})$ was reduced with K-Selectride in THF at -78°C, conditions which selectively produce the 9 α -hydroxyl in similar compounds⁵, to give exclusive formation of the desired alcohol $(\underline{4})$ in 70% yield. The configuration of the hydroxyl group, resulting from the attack of hydride from the less hindered side, was shown by ¹³C and ¹H NMR (see below).

An alternative route to compound (<u>4</u>) starting from (<u>3</u>) involves a convenient procedure for the inversion of stereochemistry at C₉ recently developed in these laboratories. Thus reduction of (<u>3</u>) in ethanol at O°C with sodium borohydride proceeds stereospecifically to give the β -alcohol (<u>5</u>) in quantitative yield.⁶ Analysis of the 90 MHz ¹H NMR and FT 80 ¹³C NMR spectra shows the absence of the corresponding 9 - α - alcohol.

Indeed the (C-9) H in 9- α -hydroxy PG's is known to occur at approx. O.3 ppm upfield from the corresponding signal in 9- β -hydroxy PG's⁸. Moreover the ¹³C NMR spectra of (<u>4</u>) and (<u>5</u>) showed the signals of C-8 and C-9 to be more shielded in (<u>4</u>) than in (<u>5</u>) (Prostane numbering)⁹. The β -hydroxy-acetal thus obtained was converted into its corresponding tosylate (<u>6</u>) by p-toluenesulfonyl chloride in pyridine. Exposure of the tosylate (<u>6</u>) to But₄N⁺NO₃⁻ in pentane at 120°C in a sealed tube for 5 hours afforded the nitric ester (<u>7</u>) "via" SN₂ displacement in quantitative yield.¹⁰ It is worth mentioning that no trace of elimination products could be detected. Reduction of the 9- α -nitric ester by Zn/CH₃COOH gave the α -hydroxyacetal (<u>4</u>), whose structure was confirmed by comparison with authentic sample obtained by an alternative route^{3c}, in 70% yield starting from (<u>3</u>).

Protection of the hydroxy group (t-butyldimethylsilylchloride--imidazole--dimethylformamide-(DMF))¹¹, gave the silyl ether ($\underline{8}$) which upon reduction with lithium aluminum hydride (LAH) in anhydrous ethyl ether afforded the alcohol ($\underline{9}$) as an oil in 90% yield. Finally acetylation of ($\underline{9}$) by treatment with a slight excess of acetic anydride in pyridine and a catalytic amount of dimethylamine pyridine (DMAP) at 0°C for 2 h, furnished quantitatively the acetate ($\underline{10}$) as an oil.

The next phase of the synthesis is the attack of the necessary ring appendages. To this purpose the key intermediate (10) was oxidatively converted into the aldehyde (11) by exposure to a stream of ozone in $\operatorname{CH}_2\operatorname{Cl}_2$ solution at -78°C until a deep blue solution was obtained, followed by the reduction of the resulting ozonide with $\operatorname{Zn/CH}_3\operatorname{COOH}$. The aldehyde (11), without further purification, was then allowed to react in dimethylsulfoxide with the sodium salt of (4-carboxybutyl)triphenyl-phosphonium bromide¹² obtained by treatment with sodium methylsulfinylmethide, to give the coupled product (12) which was subjected to mild basic hydrolysis and sequential esterification by diazomethane to afford the ester (14) in 43% yield starting from (11). $\operatorname{CrO}_3/\operatorname{Pyridine}$ oxidation of (14) led to the aldehyde (15) which was then converted to the $\alpha - \beta$ unsaturated ketone (16) by means of the sodium salt of dimethyl (2-oxohepthyl)phosphonate following a well known standard procedure. Completion of the synthesis was then effected by a three step procedure: a) sodium borohydride reduction of the 15-ketone which afforded a mixture of epimeric C(15) alcohols (17) and (18); b) separation by preparative t.l.c. of the most polar alcohol; c) hydrolysis of the latter to PGD₂ methyl ester by exposure to HF_{ao}/CH₃CN¹³.

In order to complete the synthesis of PGD_2 purely from optically active precursors, we have now developed a practical and efficient method for resolving cyclopentanedione bis-acetal^{3c} (<u>21</u>) "via" the formation of the (+) ephedrine salt of the corresponding acid (<u>22</u>) and

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conversion to the optically active intermediate (23) (SCHEME 11), which can be transformed into the optically active natural PGD, following the route reported in this paper. Details of this process will be reported shortly elsewhere.

SCHEME II



EXPERIMENTAL

l.r. spectra were recorded as film on a Perkin-Elmer 710 B spectrophotometer 1 H and 13 C n.m.r. spectra were determined in CDCl or $C_0 D_6$ solutions on Varian EM 390 and Varian FT 80 instruments respectively; chemical shifts are expressed as δ -values in p.p.m. from internal standard SiMe₂. Mass spectra were taken on a Varian Mat 112 S instrument (70 eV). T.l.c. was performed on silica gel sheets (Kieselgel 60 F₂₅₂ - Merck) and column chromatography on a Chromatospac Prep. 10 (Jobin-Ivon instrument) using silica gel (H 60 Merck).

Materials .-- Commercially available starting materials were used without prior purification, unless otherwise stated. THF was obtained anhydrous and oxygen-free by distillation over sodium benzophenone ketyl under argon. Methylene dichloride was distilled over P_{25}^{0} . Disopropylamine was refluxed over molecolar sieves (Type 4A, Fluka) and distilled $\frac{1}{2}$ atmospheric pressure. Dimethyl sulfoxide (DMSO) was distilled at reduced pressure over Call₂. (4-Carboxybutyl) triphenyl-phosphonium bromide and dimethyl-(2-oxo-heptyl)phosphonate were purchased from Aldrich.

Preparation of compounds.

2,3-trans-3,4-trans-2-Ethoxycarbonyl-3-(prop-2-enyl)-4-hydroxy-cyclopentanone Ethylene Acetal (5)

--A solution of ketone $(\underline{3})$ (1g, 3.9 mmol) in ethanol (5 ml) was added dropwise at 0°C to a stirred suspension of NaBH (0.6 g,15 mmol) in ethanol (20 ml., 95% solution). The resulting solution was stirred at this temperature for 1 h, then the excess of the reagent was destroyed with 1% HCl. The mixture was extracted with ether (3 x 30 ml). The combined extracts were dried (MgSO₂) and evaporated in vacuo to give quantitatively the crude acetal (5) as an oil whose purity was estimated by t.l.c. and ¹C n.m.r. analysis to be 100%.

m/z 256 (M⁺), 238 (M⁺-H₂O), (film) 3 500 and 1^2 735s cm⁻¹

 $\psi_{m,a}$ (film) 3 500 and 1²735s cm⁻¹ $\delta_{H}^{(90)}$ MHz; CDCl₃) 6.0-4.9 (m 3 H), 4.2 (q 2 H), 3.9 (m 5 H), 2.8 (bs 1 H OH), 2.7-2 (complex pattern 6 H), 1.3 (t 3 H).

2,3-trans-3,4-trans-2-Ethoxycarbonyl-3-(prop-2-enyl)-4-p-toluen-sulfonyloxy cyclopentanone Ethylene Acetal (6).

--To a solution of hydroxy ketone (5) (0.6g, 2.3 mmol) in dry methylene dichloride (10 ml) was added at 0°C triethyl amine ($\overline{1}$ ml, 3.5 mmol), p-toluenesulfonyl chloride (0.5g, 2.6 mmol), 4-dimethylaminopyridyne (DMAP) (0.030g). The mixture was stirred overnight at room temperature. After filtration, the reaction mixture was quenched with cold dil. HCl and extracted with methylene dichloride. The organic layers were washed several times with water, dried (Na_2SO_4) , the solvent was removed in vacuo and the residue was

chromatographed at medium pressure (Cyclohexane/ethyl acetate 1/1) to give the tosylate (6) (0.9g 95%). m/z 410 (M^+), 369, 365. (film) 1 730s, 1 600, 1 370, 1 180, 1 090 cm⁻¹. ν (film) 1 730s, 1 600, 1 3/0, 1 100, 1 050 cm . $m_{a,v}$ (max) MHz, CDCl₂) 7.85 (d 2 H), 7.35 (d 2 H), 5.45 (m 1 H), 4.85 (m 2 H), 4.2 (q 2 H), $\delta_{\rm H}^{\rm (90~MHz, CDCl_3)}$ 7.85 (a 2 n/, 7.85 (a 2 n/, 7.85 (a 3 H), 1.2 (t 3 H), 3.9 (m 5 H), 2.8-1.65 (m 6 H), 2.4 (s 3 H), 1.2 (t 3 H). 2,3-trans-3,4-cis-2-Ethoxycarbonyl-3-(prop-2-enyl)-4-nitroxy-cyclopentanone Ethylene Acetal (7). --A solution of tosylate ($\underline{6}$) (0.5g, 1.2 mmol) and tetrabutylammonium nitrate (1.1g 3.6 mmol) in pentane was heated in sealed tube at 120°C for 5 h. The pentane solution was separated from oily-residue and the solvent removed in vacuo to give essentially pure $(\underline{7})$ (0.35g 97%). m/z 301 (M^+), 260, 256, 255. (film) 1 730s, 1 630, 1 280 cm⁻¹. δ_{H} (90MHz CDCl) 5.75 (m 1 H), 5.4 (bs 1 H), 5.15 (m 2 H), 4.2 (q 2 H), 3.9 (m 4 H), 2.9 (m 2 H), 2.3 (m 4 H), 1.2 (t 3 H). 2,3-trans-3,4-cis-2-Ethoxycarbonyl-3-(prop-2-enyl)-4-hydroxy cyclopentanone Ethylene Acetal (4) (From reduction of nitric ester). --A solution of nitric ester (7) (0.35g 1.16 mmol) in CH_COOH (5 ml) was treated with Zn powder in small portions, at room temperature, until no more starting material was detected by t.l.c. The reaction mixture was filtered, diluted with ether and the organic layers were washed several times with aq. (5%) NaHCO₃, brine, and dried (Na₂SO₂). The solvent was stripped off and the residue chromatographed at medium pressure (cyclohexane/ethyl acetate 1/1) to give (4) (0.22g 74%) identical in all respects to the product obtained from reduction of ketone (3) with K-Selectride^{3C}.

m/z 256 (M^+), 238 (M^- -H₂O)

 \boldsymbol{v}_{max} (film) 3 500 and $1^{2}735$ cm⁻¹. $\boldsymbol{\delta}_{11}^{(90)}$ MHz C₆D₆) 5.8 (m 1 H), 5.15 (m 2 H), 4.2 (m 1 H), 4.1 (q 2 H), 3.9 (m 4 H), 3-1.5 (complex pattern 7 H), 1.3 (t 3 H).

2,3-trans-3,4-c1s-2-Ethoxycarbonyl-3-(prop-2-enyl)-4-t-butyldimethylsilyloxy cyclopentanone Ethylene Acetal (8).

--To a solution of alcohol (4) (5g, 19.5 mmol) in dry dimethylformamide (10 ml) t-butyldimethylsilyl chloride (3.45g, 23.6 mmol) and freshly crystallized imidazole (3.5g 51 mmol) were added. The mixture was stirred overnight at room temperature, poured into ice-water and extracted with ethylacetate. The extracts were washed with water, dried over MgSO₄ and concentrated in vacuo to give (8) (7.03g, 97%) with a good degree of purity for further elaboration.

m/z 370 (M^+), 325, 313 γ_{m} (film) 1 750, 1 650, 1 250, 1 100, 1 040 cm⁻¹. δ_{H} (90 MHz CDC1₃) 6.0-4.7 (m 3 H), 4.05 (q 2 H), 3.8 (m 5 H), 3.1-1.9 (m 6 H), 1.3 (t 3 H), 0.9 (s 9 H), 0.08 (s 6 H).

2,3-trans-3,4-cis-2-(<u>Hydroxy-methyl</u>)-3-(<u>prop-2-enyl</u>)-4-<u>t-butyldimethylsilyloxy</u> cyclopentanone Ethylene Acetal (9).

--LiAlH₂ (0.759g 20 mmol) was added in small portions to a solution of ester ($\underline{8}$) (7g, 18.9 mmol) in anhydrous ethyl ether (60 ml) at O°C. After stirring for 1 h, the mixture was treated with saturated aqueous solution of NH₂Cl and extracted with ether (3 x 150 ml). Usual work-up gave ($\underline{9}$) (5.6g, 90.3%) homogeneous on t.l.c. (silica gel, Hexane/Ethyl acetate 1/1).

m/z 328 (M⁺), 271. ν_{max} (film) 3 530, 1 650 and 1 250 cm⁻¹. δ_{μ} (90 MHz CDCl₂) 6.0-4.8 (m 3 H), 4.15 (m 1 H), 3.85 (bs 4 H), 3.7 (d 2 H), 2.65 (bs 1 H OH), 2.4-1.8 (m 6 H), 0.88 (s 9 H), 0.05 (s 6 H).

2,3-trans-3,4-cis-2-(Methyl-acetyloxy)-3-(prop-2-enyl)-4-t-butyldimethylsilyloxy-cyclopentanone Ethylene Acetal (10).

--A mixture of alcohol (9) (5.4g, 16.4 mmol), pyridine (5 ml), acetic anhydride (5 ml), and DMAP (0.2g) is allowed to stand for 1 h at O°C. The solution is partitioned between ethylacetate and cold HCl (30 ml 1 N), the organic phase is washed with saturated NaHCO₃,

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dried over MgSO₄ and the solvent removed in vacuo. Was obtained (<u>10</u>) as an oil in 98% yield (6g).

m/z 3/0 (M), 313, 233 **a** (film) 1 750, 1 650 and 1 240 cm⁻¹ **b** (90 MHz CDCl₂) 6.0-4.7 (m 3 H), 4.25 (m 1 H), 4.15 (d 2 H), 3.8 (bs 4 H), 2.5-1.6 (m 6 H), 2 (s 3 H), 0.88 (s 9 H), 0.05 (s 6 H).

2,3-trans-3,4-cis-2-(Methyl-acetyloxy)-3-(methyl-formyl)-4-t-butyldimethylsililoxy-cyclopentanone Ethylene Acetal (11).

<u>--A solution of (10) (6g 16.1 mmol) in</u> methylene dichloride (50 ml) was cooled to -78° C. Ozone was passed through the solution at the rate of 0.5 ml/min until deep blue solution. While still at -78° C the system was flushed with nitrogen until colourless solution, the temperature was then allowed to rise to 0°C and acetic acid (10 ml) was added, followed by Zn powder in small portions at room temperature monitoring by t.l.c. (SiO₂, Hexane/Ethylacetate 6/4) until the reduction of ozonide was complete. The resulting mixture was filtered, the pH adjusted at 7 with NaHCO₃ (5%), washed several times with water. The organic layer was dried (Na₂SO₄) and the solvent removed in vacuo to give the oily aldehyde (5.3g 88.5%) homogeneous by t.l.c. and C n.m.r., which was employed as such in successive step. m/z 372 (M⁺), 315, 313

 v_{max} (film) 2 700, 1 740, 1 725, 1 250, 1 040 cm⁻¹. δ_{H} (90 MHz CDC1₃) 9.8 (s 1 H CHO), 4.2 (m 1 H), 4.15 (d 2 H), 3.9 (s 4 H), 2.7-2.2 (m 6 H), 2.05 (s 3 H), 0.9 (s 9 H), 0.05 (s 3 H), 0.025 (s 3 H).

2,3-trans-3,4-c1s-2-(llydroxy-methyl)-3-(6-methoxycarbonyl-hex-2(Z)-enyl)-4-t-butyld1methyls1lyloxy-cyclopentanone Ethylene Acetal (14).

--A mixture of 50% sodium hydride-dispersion-in-oil (4.2g) and dimethylsulfoxide (50 ml freshly distilled over calcium hydride) was stirred under argon at 75°C until gas evolution ceased (1 h).

To the resulting solution (33.6 ml, 67 mmol, 2 M in sodium methylsulfinylmethide) cooled to room temperature, (4-carboxybutyl)-triphenyl-phosphonium bromide (15.6g, 35 mmol) in DMSO (40 ml) was added dropwise. After the deep red liquid had been stirred for a further 20 min, a solution of the aldehyde (<u>11</u>) (5.3g, 14 mmol) in DMSO (40 ml) was added. The temperature was raised to 50°C, and stirring continued for 1 h. The mixture was poured into ice-water and extracted with ethyl acetate. The cooled aqueous layer was acidified with dil. HCl and extracted with ethyl acetate (3 x 100 ml). The combined organic extracts were washed with water, dried (MgSO₄) and concentrated. The resulting crude product (5g) was dissolved in methanol (100 ml) and treated with K₂CO₃ (3g) at reflux for 1 h. The mixture diffied with dil. HCl, extracted with ethyl acetate, washed with brine, dried and concentrated. To the oil-residue ethereal diazomethane was added and the solution stirred at 0°C for 1 h. To this solution were added few drops of acetic acid in order to destroy the excess of reagent and the solvent was removed in vacuo. The final product was purified by chromatography at medium pressure on silica gel eluting with ethyl acetate/hexane 1/1 to yield the alcohol (<u>14</u>) as colourless oil (2.6g, 43% from <u>11</u>).

 $\overline{m/z}$ 428 (M⁺), 410, 397, 371.

 ψ_{max} 3 530, 1 735, 1 250, 1 080, 1 040 cm⁻¹. δ_{H} (90 MHz CDCl₂) 5.3 (m 2 H), 4.02 (m 1 H), 3.75 (m 7 H), 3.5 (s 3 H), 2.4(bs 1 H OH), 2.2-1.3 (m 11 H), 0.8 (s 9 H), 0.05 (s 6 H).

2,3-trans-3,4-c1s-2-Formyl-3-/6-methoxycarbonyl-hex-2(Z)-enyl/-4-t-butyldimethylsilyloxy-cyclopentanone Ethylene Acetal (15).

--To a solution of chromium trioxide (0.87g)-pyridine (1.5 ml)-complex in anhydrous methylene dichloride (20 ml), was added at 0°C with stirring under argon, a solution of alcohol (14) (0.5g 1.2 mmol) in methylene dichloride (10 ml). After 20 min the mixture was filtered on dry diatomaceous earth (Celite 20g), washed with ice-cooled HCl (1 N) and the solvent removed in vacuo to give the aldehyde (15) (0.45g 90%). m/z 426 (M⁺), 395, 369.

 y_{max} (film) 2 710, 1 735, 1 720, 1 250, 1 100 and 1 040 cm⁻¹. δ_{11} (90 MHz CDCl₃) 9.7 (d 1 H), 5.4 (m 2 H), 4.25 (m 1 H), 3.9 (m 4 H), 3.65 (s 3 H), 2.9-1.5 (m 12 H), 0.8 (s 9H), 0.05 (s 6 H).

<u>Methyl</u>-(8,9-cis-9,12-trans)-9-<u>t-butyldimethylsilyloxy-11,15-dioxo-prosta-5(2),13(E)-dienoate-11-</u>

ethylene Acetal (16).

-- To a stirred suspension of sodium hydride (50% dispersion-in-oil, 46 mg) in anhydrous tetrahydrofuran (5 ml), at 0°C was added a solution of dimethyl-(2-oxoheptyl)-phosphonate (0.222g, 1 mmol) in the same solvent (5 ml). After stirring at 0°C for 30 min a solution of the aldehyde ($\underline{15}$) (0.42g, 1 mmol) in THF (5 ml) was added. The temperature was allowed to rise 20°C and stirring was continued for 4 h. The reaction was quenched by addition of ice-cooled HCl (0.1 N 5 ml). The resulting mixture was extracted with ethyl acetate (3 x 30 ml), and the combined extracts were washed with brine, dried (Na_2SO_2) and evaporated to give an oil which was chromatographed on medium pressure $(SiO_2$ Hexane/Ethyl acetate 60/30) to yield (<u>16</u>) (0.36g, 69%)

m/z 522 (M^{+}), 507, 491, 465.

 v_{μ} (film) 1 735, 1 715, 1 665, 1 625, 1 250, 1 040 cm⁻¹. v_{μ} (film) 1 735, 1 715, 1 665, 1 625, 1 250, 1 040 cm⁻¹. S_{μ} (90 MHz CDC1₃) 6.76 (dd J=16 Hz and J=8.5 Hz 1 H), 6.25 (d J=16 Hz 1 H), 5.35 (m 2 H J cis), 4.25 (m 1 H), 3.8 (bs 4 H), 3.5 (s 3 H), 2.7-1 (m 23 H), 0.9 (s 9 H), 0.055 (s 6 H). δ_H

-hydroxy<u>-prosta</u>-5(2),13(E)-Methyl(8,9-cis-9,12-trans)-9-t-butyldimethylsilyloxy-11-oxo-15a dienoate-11-ethylene Acetal (18) and Methyl(8,9-cis-9,12-trans)-9-t-butyldimethylsilyloxy-11- $\infty - 15 \beta$ -hydroxy-prosta-5(Z), 13(E)-dienoate-11-ethylene Acetal (17).

--To a solution of (16) (0.65g, 1.24 mmol) in methanol (5 ml) cooled to -10° C, was added a solution of sodium borohydride (100mg) in ice-cooled methanol (3 ml). After 1 h an aqueous solution of satured sodium hydrogen phosphate (10 ml) was added and the mixture was extracted with ethyl acetate. The extracts were dried $(MgSO_{1})$ and the solvent was removed to yield a colourless oil residue. Medium pressure chromatography (hexane/ethyl acetate 70/30) of this oil yield a 1/1 mixture (0.25g each) of (17) and (18) for a total yield of 77%.

(data identical for both compounds):

m/z 524 (M^{+}), 506, 493, 467, 449. \boldsymbol{v} (film) 3 500, 1 750, 1 250, 1 040 cm⁻¹. ess polar compound (<u>17</u>)

Less

 $(90MHz CDCl_{2}) 5.63$ (dd J=7Hz and J=16Hz 1 H), 5.45 (m 3 H), 4.16 (m 2 H), 3.83 (m 4 H), 3.66 (s 3 H), 2.66 (m 1 H), 2.5-1.06 (m 23 H), 0.9 (s 9 H), 0.055 (s 6 H). More polar compound (18)

ð_u 5.63 (dd J=7Hz and J=16Hz 1 H), 5.45 (m 3 H), 4.16 (m 2 H), 3.83 (m 4 H), 3.66 (s 3 H), 2.66 (m 1 H), 2.5-1.06 (m 23 H), 0.9 (s 9 H), 0.055 (s 6 H).

(\pm) PGD₂ methyl ester (20).

--The most polar silyl ether (18) (0.1g 0.19 mmol) was dissolved in a mixture of 40% hydrofluoric acid-acetonitrile (1-3) (3 ml), the solution was stirred at room temperature for 3 h and then poured into 5% aqueos sodium hydrogen carbonate (5 ml) and rapidly extract with methylene dichloride (3 x 30 ml). The combined extracts were washed in turn with water and brine, dried (MgSO₂) and evaporated at 5° C to give a brown oil which was chromatographed on preparative t.l.c. (SiO₂, ether) to give (20) (0.05g 72%) m/z 366 (M⁺)

 v_{max}^{-3} 3 450, 1 735 cm⁻¹ v_{max}^{-1} (300MHz Bruker CXP 300)5.63 (dd J=7 and 16 Hz, becomes a d upon irradiation at 4.09, $\delta_{\rm H}$ (300MHz Bruker CXP 300)5.63 (dd J=7 and 16 Hz, becomes a d upon irradiation at 4.09, J=16 Hz, H(C-15)), 5.45-5.58 (m 2 H (C-5,C-6), 5.43 (dd J=16 and 8.5 Hz, becomes d upon irradiation at 2.87 J=16 Hz 1 H(C-13), 4.47 (bm, becomes a d upon irradiation at 2.45 J=4Hz, 1 H (C-9)), 4.09 (bq J=7 Hz 1 H(C-15)), 3.66 (s 3 H), 2.87 (dd, J=12 and 8.5Hz, becomes a d upon irradiation at 5.43, J=12Hz, and becomes a d upon irradiation at 1.95, J=8.5, 1 H(C-12)), 2.45 (m 2 H (C-10)), 1.95 (bs 1 H (C-8)), 2.8-0.9 (m 21 H). (These values are perfectly in accord with the values reported in reference 2d).

(±)-15-<u>epi-PGD</u> methyl ester (19)

--This product was produced from the less polar silyl ether (17) using the same method as for (20). Chromatography with ether on preparative t.l.c. (SiO₂) gave (\pm)epi-PGD₂ methyl ester in 60% yield.

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Table-13C N.m.r. chemical shifts and assignements for cyclopentanoids derivative^{a,b}

		Compounds										
	4	5	8	9	10	11	14	15	16	17	18	
C(n)												
1							173.3	173.1	173.1	173.1	173.1	
2							(25.1	(25.0	(27.0	(26.9	(27.0	
3							c -26.0	c-26.4	c-25.5	c -25.7	c -25.6	
4							27.0	26.9	\24.9	24.9	\25.2	
5			115.7	115.5	115.7		129.7	128.6	129.1	129.4	129.4	
6			137.3	137.9	137.6	200.3	129.7	130.6	129.8	128.0	128.5	
7			33.3	32.6	32.7	43.3	33.4	33.3	33.3	33.1	33.3	
8	46.8	48.7	47.1	44.8	47.2	41.8	45.6	45.8	50.4	50.9	50.6	
9	71.0	73.5	71.6	71.3	71.1	71.0	71.5	71.9	71.5	71.0	71.0	
10	47.1	45.7	47.9	47.1	47.2	47.2	47.1	47.8	48.3	48.3	48.2	
11	117.5	116.5	117.0	117.7	116.1	115.8	117.6	117.5	117.7	117.6	117.5	
12	57.8	58.2	57.8	51.3	48.8	48.6	51.6	63.0	55.1	55.0	54.9	
13	171.2	171.7	170.8	59.3	63.0	63.1	60.0	200,1	144.3	138.1	137.5	
14									132.9	130.0	129.8	
15									198.3	72.9	72.4	
16									40.14	37.9	37.9	
17									31.8	32.3	32.3	
18									c_{24.2	c√25.1	c-(25.0	
19									22.8	23.1	- 23.0	
20									14.1	14.3	14.2	

Chemical shifts in p.p.m. downfield from Me Si. Spectra were taken in $C_6 D_6$ at 20.00 MHz in the fourier mode with a Varian FT 80 spectrometer. b) Prostane numbering. c) Assignements may be reversed.

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SCHEME I





(2)



(1)



(4): R = OH;

(5): R = H;

(6): R = H;

(7): R = ONO2;

(8): R = OSIMe2;

ሖ



(9) : R = H

 $(10): R = COCH_{2}$



(11): R = COCH3



 $\mathbf{R}' = \mathbf{H}$

R' = OH

R' = OTS

 $\mathbf{R'} = \mathbf{H}$

 $\mathbf{R'} = \mathbf{H}$

(12): $R = COCH_3$; R' = H(13): R = H; R' = H(14): R = H; R' = Me



(16): R = R' = O(17): R = H; R' = OH(18): R = OH; R' = H



(15)



(19): R = H; R' = OH(20): R = OH R' = H