# AN IRIDOID GLUCOSIDE FROM NYCTANTHES ARBORTRISTIS\*

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Key Word Index—Nyctanthes arbortristis; Verbenaccae; iridoid glucosides; arbortristoside A;  $\delta\beta$ -hydroxyloganin.

Abstract—A new iridoid glycoside along with the known compounds, nyctanthic acid, oleanolic acid, friedelin,  $\beta$ -sitosterol glucoside,  $6\beta$ -hydroxyloganin and arbortristoside A have been isolated from Nyctanthes arbortristis and characterized by spectral and chemical means.

## INTRODUCTION

Nyctanthes arbortristis (Hindi-Harsingar) is used in the Ayurvedic system of medicine for the treatment of various diseases, such as fever, rheumatism and intestinal worm infections. A decoction of the leaves is recommended as a specific treatment for obstinate sciatica [1, 2]. The powdered seeds are used to cure scurfy infections of the scalp. The inflorescence and young fruits powdered in water is used for relieving cough by tribal people of central India. Bark of the plant mixed with that of *Terminalia arjuna* is rubbed on the body to cure internal injuries and to repair broken bones [3].

A 50% ethanolic extract of the aerial parts of the plant exhibited encouraging antileishmanial activity of the order of 85.89% at a dose level of 1 g/kg/day  $\times$  5 on the 28th day post-treatment against hamsters infected with *Amastigotes* parasites (Guru, P. Y., personal communication). The ethanolic extract of the leaves of the plant showed significant amoebicidal activity in rats (Prasad, B. N. K., personal communication).

Earlier workers have reported the isolation of crocin 1, 2 and 3 from corolla tubes [4], nyctanthic acid,  $\beta$ sitosterol, oleanolic acid, friedlin, lupeol, astragalin and nictoflorin from leaves [5, 6] and nyctanthoside [7], arbortristosides A and B and polysaccharides from seed kernels [8, 9] of N. arbortristis.

### **RESULTS AND DISCUSSION**

The *n*-butanol soluble fraction of the seeds of N. arbortristis after column chromatography resulted in the isolation of a new iridoid glucoside (1), arbortristoside A (2) and  $6\beta$ -hydroxyloganin (3).

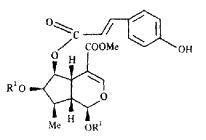
Compound 1 was obtained as a white amorphous powder,  $C_{26}H_{32}O_{13}$ , mp 200–202°,  $[\alpha]_D - 78^\circ$  (MeOH). The characteristic bands in the IR spectrum showed the presence of an  $\alpha,\beta$ -unsaturated carbonyl (1710), C=C (1650), aromatic (1620–1455) and hydroxyl (3400) cm<sup>-1</sup> functions in the molecule. In addition to the UV absorption maxima at 229 nm, typical of an iridoid enol ether system conjugated with a C-4 carbonyl group, the UV spectrum showed peaks at 301 (sh) and 312 nm indicating the presence of a *p*-substituted aromatic ring [10, 11]. The EI mass spectrum of 1 exhibited typical features expected from the fragmentation pattern of a  $C_{10}$ -iridoid glucoside [12, 13], with peaks at m/z 226, 198, 178, 164, 161, 143 and 139. The <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) displayed singlets at  $\delta$ 7.40 and 3.75 unambiguously assigned to H-3 and a methoxy group, thereby suggesting the presence of a conjugated 4-carbomethoxy enol ether system of an iridoid. The occurrence of signals at  $\delta$ 6.36 and 7.56 due to a simple AB system ( $J_{AB} = 16$  Hz) for *trans*-olefinic protons, an  $A_2B_2$  system at  $\delta$ 6.75 and 7.51 ( $J_{A_2B_2} = 10$  Hz) and a broad singlet at  $\delta$ 9.98 indicated the presence of a *p*hydroxycinnamoyl group in compound 1.

Since the <sup>1</sup>H NMR of its hexaacetate derivative (1a) (Ac<sub>2</sub>O/pyridine), showed a phenolic acetate peak at  $\delta 2.34$ it was evident that the p-substituent in the aromatic nucleus was a hydroxyl group. A doublet at  $\delta 0.94$  (J = 6 Hz) correlated well with the placement of a methyl group at C-8. The existence of H-7 as a triplet at  $\delta 4.06 (J$ =4 Hz) and H-6 as a multiplet at  $\delta$  4.98 was in accordance with the placement of a hydroxyl group at C-7 and a pcoumaroyl group at C-6 [10, 11]. Of the remaining signals, doublets at  $\delta$ 4.42 (J = 9 Hz) and 4.94 (J = 7.5 Hz) were assigned to CH<sub>2</sub>OH and to the anomeric proton of the  $\beta$ -D-glucopyranosyl moiety, respectively, and for other sugar protons there was a complex absorption in the region  $\delta$ 3.5-4.5. The presence of a  $\beta$ -glucoside was confirmed by the <sup>13</sup>CNMR (SFORD) spectrum which showed characteristic resonances at  $\delta 100.21$  (d, C-1'), 74.72 (d, C-2'), 78.04 (d, C-3'), 71.64 (d, C-4'), 77.64 (d, C-5'), 62.83 (t, C-6'). It is clear from both the constancy of the signals for glucose that compound 1 has a  $\beta$ -D-configuration at C-1'

Mannich hydrolysis of 1 afforded glucose and an aglucone (16). The <sup>1</sup>H NMR of 1b showed that it retained the *p*-coumaroyl moiety. It also showed a pair of doublets at  $\delta 6.82$  and 7.35 (J = 10 Hz), assigned to an  $A_2B_2$  system of aromatic protons of a *p*-coumaroyl group and two trans-olefinic protons resonated at  $\delta 6.27$  and 7.82 (d, J = 16 Hz). Alkaline hydrolysis (MeOH-KOH) of 1 afforded *p*-hydroxycinnamic acid and another compound which on subsequent methylation with diazomethane yielded a methyl ester which proved to be identical with another iridoid glycoside (3) isolated from the same plant and characterized as  $\delta\beta$ -hydroxyloganin [14]. Furthermore structure 1 was confirmed by the <sup>13</sup>C NMR [SFORD] spectrum (Table 1).

<sup>\*</sup>C.D.R.I. communication number 4110.

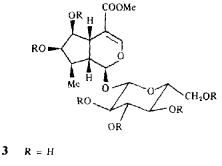
2



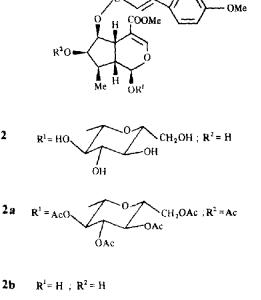
$$\mathbf{I} \qquad \mathbf{R}^{1} = \mathbf{OH} \qquad \mathbf{OH} \qquad \mathbf{OH} \qquad \mathbf{CH}_{2}\mathbf{OH} ; \mathbf{R}^{2} = \mathbf{H}$$

1a 
$$R^{1} = AcO \xrightarrow{O} OAc OAc ; R^{2} = Ac$$

**1b** 
$$R^1 = H$$
;  $R^2 = H$ 



3a  $\mathbf{R} = \mathbf{A}\mathbf{c}$ 



Compound 2,  $C_{27}H_{34}O_{13}$ , mp 220–222° (ethanol), [x]<sub>D</sub> -92.5° (MeOH) was found to be identical to arbortristoside A, on the basis of its spectroscopic data. [IR:  $v_{max}^{KBr}$ 3400 (OH), 1710 (>C=O), 1650 (C=C); EI mass spectrum m/z 226, 198, 178, 161, 139; UV:  $\lambda_{max}^{MeOH}$  227 [C-(4)COOMe], 300 (sh), 308 nm.] The <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) spectrum contained signals at  $\delta 5.27$  (d, J = 6 Hz), 7.40 (s), 4.13 (t, J = 4 Hz), 4.98 (m) for H-1, H-3, H-7 and H-6. A singlet at  $\delta$  3.75 for a methoxy group with a pair of doublet centred at  $\delta 6.94$  and 7.68 (J = 9 Hz each) assigned to the A2B2 system of aromatic protons and another pair of doublet for two trans-olefinic proton resonances at  $\delta 6.74$  and 7.62 (J = 16 Hz) confirmed the presence of a p-methoxy cinnamoyl group in 2. Compound 2 was found to be the p-methoxycinnamoyl derivative of 1, as the methyl ester of 1 obtained by methylation  $(CH_2N_2)$  was identical with 2 (co-TLC, spectral data).

Compound 3, C17H26O11, mp 221-223° was obtained as white needles (EtOH- $H_2O$ ). Its IR (1705 and 1650 cm<sup>-1</sup>) and UV spectrum (235 nm) were characteristic of carbomethoxy enol ether system, which was confirmed by the presence of an olefinic signal at  $\delta$ 7.54 (s, H-3) and a methoxy signal at  $\delta$  3.82 in its <sup>1</sup>H NMR  $(DMSO-d_6)$  spectrum. The <sup>13</sup>C NMR spectrum displayed signals for 17 carbon atoms and correlated well with that of 6<sup>β</sup>-hydroxyloganin [14].

Structures 1-3 were also unambiguously assigned by their <sup>13</sup>C NMR (SFORD and DEPT) data (Table 1). The chemical shifts were assigned on the basis of structurally similar compounds [15, 16].

## EXPERIMENTAL

General, Mps uncorr. EIMS: 70 eV. 'H NMR: 80, 90 and 400 MHz; <sup>13</sup>C NMR: 100 MHz. TMS int. std. CC: silica gel

с	1 (CD <sub>3</sub> OD)	la (CDCl <sub>3</sub> )	2 (CD <sub>3</sub> OD)	<b>2a</b> (CDCl <sub>3</sub> )	3 (DMSO-d <sub>6</sub> )
1	97.80 d	94.40	97.42	94.32	95.42
3	154.00 d	150.64	154.49	150.56	151.38
4	110.52 s	110.39	109.85	110.35	109.33
5	39.67 d	36.15	39.23	36.10	37.57
6	78.36 d	77.00	76.86	77.25	77.12
7	79.04 d	72.37	70.74	72.26	72.58
8	37.45 d	35.51	36.88	35.42	36.60
9	46.07 d	44.91	41.94	44.84	44.04
10	14.87 g	13.19	14.36	13.12	13.96
11	170.13 s	170.32	169.73	170.25	167.34
12	52.01 q	51.30	51.94	51.22	50.92
1'	100.21 d	96.00	99.33	95.91	98.50
2'	74.72 d	70.72	73.73	70.63	73.03
- 3'	78.04 d	72.54	77.06	72.45	77.74
4′	71.64 d	68.42	70.74	68.32	70.04
5'	77.64 d	74.96	77.84	74.61	76.63
5' 6'	62.83 t	61.71	62.20	61.71	61.08
1″	127.33 s	129.22	127.64	128.19	
2″	131.13 d	131.97	130.25	129.72	
- 3″	116.85 d	122.09	114.81	114.33	_
4″	161.16 s	152.35	162.09	161.54	_
5″	116.85 d	122.09	113.89	114.33	
<i>6</i> ″	131.13 d	131.97	129.27	129.72	
α	115.30 d	117.64	115.60	114.88	
Γ β	146.66 d	144.18	145.57	144.89	_
ço	168.99 s	168.79	168.01	168.87	<u> </u>
-ОМе		-	55.63 g	55.25	
6×CO		165.86 s	55.05 q	166.23	
0.00	_	165.38 s	_	166.33	
		168.94 s		169.17	
	-	169.23 s		169.46	
	_	169.52 s		169.85	
	_	169.92 s		107.85	
6×-0C0		20.02 q	_	19.95	
Me		20.02 q 20.40 a		20.34	
1412	·	20.40 q 20.45 q		20.34 20.48	
		20.43 q 20.59 q		20.48	
		-		20.34	
		20.95 q			

Table 1. <sup>13</sup>CNMR data of 1, 1a, 2, 2a and 3 (100 MHz)

60-120. TLC and prep-TLC: silica gel 60, spots and bands were detected by  $I_2$  vapour and spraying reagents for iridoids: (i) 1% ceric sulphate in 1 M H<sub>2</sub>SO<sub>4</sub> (ii) vanillin (3%) and H<sub>2</sub>SO<sub>4</sub> (1%) in 100 ml EtOH followed by heating at 100-110° for 5-10 min.

Plant material. Twigs of N. arbortristis L. were collected from Lucknow (Uttar Pradesh) in January, 1985. A voucher specimen (1176 CDRI) is deposited in the Herbarium of Medicinal Plants of the Central Drug Research Institute, Lucknow.

Isolation of iridoids. Seeds of N. arbortristis (7 kg) were exhaustively extracted with 50% EtOH ( $5 \times 1.5$  l.) at room temp. The combined extract were evapd in vacuo below 45° to give a residue (850 g). The concd EtOH extract (400 g) on subsequent fractionation with hexane, CHCl<sub>3</sub> and *n*-BuOH gave hexane (10 g), CHCl<sub>3</sub> (60 g) and *n*-BuOH (140 g) sol fractions.

The concd *n*-BuOH ext (70 g) was chromatographed on a silica gel (1.5 kg) column and eluted with EtOAc and EtOAc (satd with  $H_2O$ )-MeOH of increasing MeOH content; this afforded fraction A (8 g) and fraction B (4.6 g), respectively. Fraction A was rechromatographed on silica gel (250 g) and eluted with CHCl<sub>3</sub>-MeOH, when 7% MeOH-CHCl<sub>3</sub> gave

compound 2 (4 g) as white needles, mp 220-222° (EtOH) and 10% MeOH-CHCl<sub>3</sub> gave compound 1 (0.965 g) as a white amorphous powder from CHCl<sub>3</sub>-MeOH, mp 200-202°. Fraction B was also chromatographed on silica gel (100 g) using EtOAc-MeOH when 2% MeOH-EtOAc yielded compound 3 (90 mg) which was further purified by prep. TLC (EtOAc-MeOH-H<sub>2</sub>O; 14:3:3). It was recrystallized from EtOH-H<sub>2</sub>O as colourless needles, mp 221-223°.

Compound 1.  $[\alpha]_{2}^{28} - 78.0^{\circ}$  (MeOH; c 1.0): UV  $\lambda_{max}^{MeOH}$  nm: 229 (O-C=C-CO<sub>2</sub>Me), 301 (sh), 312 (*p*-substituted benzene ring); IR  $\nu_{max}^{KBr}$  cm<sup>-1</sup>: 3450 (OH), 2970, 1710 (conjugated carbonyl) 1650 (C=C), 1620, 1530, 1450 (aromatic), 1300, 1220, 1190, 1090, 890, 855, 790; FD-MS: m/z 553 [M + H]<sup>+</sup>, 552 [M]<sup>+</sup>; EI-MS 70 eV, m/z 390 [M]<sup>+</sup> [(M - glucose)]<sup>+</sup>, 389, 372, 344, 244, 226, 198, 195, 164, 161, 148, 139; <sup>1</sup>H NMR (400 MHz) see Table 2; <sup>13</sup>C NMR (100 MHz) see Table 1.

Acetylation of 1. 1 (100 mg) was acetylated with  $Ac_2O$ -pyridine (0.25 ml each) at room temp. and kept for 14 hr. The reaction mixt after usual work-up and recrystallization from EtOH afforded a hexaacetate (1a, 40 mg) as colourless needles,

Н	1 (DMSO- <i>d</i> <sub>6</sub> )	la (CDCl <sub>3</sub> )	1b (CDCl <sub>3</sub> )
1	5.32 <i>d</i> , $J = 6$ Hz	5.35 m	5.42 t, $J = 5.5$ Hz
3	7.40 s	7.47 s	7.41 s
5	3.09 m	3.04  dd, $J = 1.5  and  8  Hz$	3.07  dd,  J = 1.5  and  8  Hz
6	4.98 m	5.35 m	4.96 dd, $J = 2.5$ and 5 Hz
7	4.06 hr t, $J = 4$ Hz		4.54 t, $J = 6$ Hz
8	2.01 m	2.50 m	2.16 m
9	2.14 m	2.59 t, $J = 11$ Hz	2.16 m
10	0.94 d, J = 8 Hz	1.1 d, J = 7.27 Hz	1.13 d, $J = 7$ Hz
12	3.61 s	3.68 s	3.76 s
α	6.36 d, $J = 16$ Hz	6.42 d, $J = 16$ Hz	6.27 d, $J = 16$ Hz
β	7.56 d, $J = 16$ Hz	7.68 $d$ , $J = 16$ Hz	7.60 d, $J = 16$ Hz
3" and 5"	6.75 d, $J = 10$ Hz	7.14 $d$ , $J = 10$ Hz	6.82 d, J = 10 Hz
2" and 6"	7.51 <i>d</i> , $J = 10$ Hz	7.58 d, $J = 10$ Hz	7.35 d, $J = 10 \text{ Hz}$
ОН	9.98 br s		9.95 br s
H-1'	4.94 d, J = 7.5 Hz	5.29 m	
-CH <sub>2</sub> OH	4.42 d, $J = 9$ Hz	4.32 m	
remaining sugar protons	3.5-4.5 m		
6×OAc		1.92, 2.02, 2.06	
group protons	_	2.1 s	~
Phenolic -OAc protons	······	2.34 s	
> CHOAc of glucose	THE Real	4.85, 5.0, 5.12 m	

Table 2. <sup>1</sup>H NMR Spectral data of compounds 1, 1a, 1b, (400 MHz)

mp 76–78°, C<sub>38</sub>H<sub>44</sub>O<sub>19</sub>; IR  $\nu _{max}^{B_3}$  cm<sup>-1</sup>: 2970, 1760, 1725, 1650, 1620, 1515, 1440, 1380, 1220, 1170, 1090, 1040, 960, 920; FD-MS: m/z 805 [M+H]<sup>+</sup>, 331 (acetylated glucose fragment); <sup>1</sup>H NMR (400 MHz) and <sup>13</sup>C NMR, (100 MHz) see Tables 1 and 2.

Mannich hydrolysis of 1. 1 (100 mg) dissolved in Me<sub>2</sub>CO (15 ml) was treated with cone HCl (0.15 ml) and kept at room temp. After 3 weeks the reaction mixt was extd with EtOAc × 4 and H<sub>2</sub>O. The organic layer was washed × 3 with H<sub>2</sub>O, dried (Na<sub>2</sub>SO<sub>4</sub>), coned and purified by prep. TLC (CHCl<sub>3</sub>-MeOH; 19:1) to obtain the aglycone (1b) as a viscous mass (15 mg),  $C_{20}H_{22}O_8$ ; UV  $\lambda_{max}^{MeOH}$  nm: 228, 305 (sh), 312; IR  $v_{max}^{KP}$  cm<sup>-1</sup>: 3400, 2970, 1720, 1710, 1645, 1615, 1530, 1460, 1400, 1230, 1180, 770; EI-MS: m/z 390 [M]<sup>+</sup> 389 [M-H]<sup>+</sup>, 385, 369, 355, 226, 185, 160, 147; <sup>1</sup>H NMR see Table 2. The aq layer was neutralized with Amberlite IR 410 CO<sub>3</sub><sup>-1</sup> resin and subjected to PC with authentic sugar samples in *n*-BuOH-HOAc-H<sub>2</sub>O; (4:1:5); glucose was identified as the sugar present in compound 1.

Alkaline hydrolysis of 1. 1 (100 mg) was dissolved in 10 ml of 2% KOH in MeOH and refluxed. After 2 hr the reaction mixt was coned, neutralized with dil HCl and immediately extracted with  $2 \times 10$  ml EtOAc and  $2 \times 15$  ml *n*-BuOH, successively. The EtOAc layers were combined, dried (Na<sub>2</sub>SO<sub>4</sub>) and coned under red. pres. The mixt. was examined by TLC and the product isolated by prep. TLC in hexane-EtOAc (7:3, with 3 drops HOAc). The chromatographic behaviour and <sup>1</sup>H NMR spectrum of the major product was identical to that of *p*-hydroxy-cinnamic acid (4), mp 210-212° (EtOH).

The *n*-BuOH extract on concn *in vacuo* was dissolved in MeOH and treated with  $CH_2N_2$  at 0° for 14–16 hr. The reaction product after removal of solvent was purified by prep. TLC (CHCl<sub>3</sub>-MeOH, 3:1, with a trace of HOAc) and recrystallized from EtOH as white needles of 3 (14 mg); mp 221–223°,  $C_{17}H_{26}O_{11}$ ; UV  $\lambda_{meN}^{MeOH}$  nm: 235; IR  $v_{mat}^{KBr}$  cm<sup>-1</sup>: 3350 (OH), 2990, 1705, 1650, 1300, 1210, 1190, 1090, 890 and 760; FD-MS: *m/z* 406 [M]<sup>+</sup>; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$ 1.21 (3H, *d*, *J*=7 Hz, Me), 1.96 (1H, *m*, H-8), 2.24 (1H, *m*, H-9), 2.75 (1H, *m*, H-5), 3.85 (3H, s, COOMe), 3.98 (2H, *br* t, *J*=5 Hz, H-6 and H-7), 3.5-4.5 (m, sugar H), 4.65 (2H, d, J = 6.5 Hz,  $-CH_2OH$ ), 5.42 (1H, d, J = 4.5 Hz, H-1), 7.50 (1H, s, H-3).

Compound 2.  $[\alpha]_{D}^{28} - 92.5^{\circ}$  (MeOH; c 1.0); UV  $\lambda_{max}^{\text{moh}}$  nm: 227, 300 (sh), 308; IR  $\nu_{max}^{\text{KBr}}$  cm<sup>-1</sup>: 3400, 2950, 1710, 1650, 1620, 1520, 1455, 1300, 1290, 1260, 1200, 1160, 1090, 1030, 975, 960, 880, 845, 810, 790, 635; FD-MS: m/z 565  $[M-H]^+$ ; <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta 0.94$  (3H, d, J = 6 Hz, Me), 2.04 (1H, m, H-8), 214 (1H, m, H-9), 3.10 (1H, m, H-5), 3.60 (3H, s, COOMe), 3.75 (3H, s,OMe), 4.13 (1H, t, J = 4 Hz, H-7), 4.42 (2H, d, J = 9 Hz,  $-CH_2OH$ ), 4.95 (1H, d, J = 7 Hz, H-1'), 4.98 (1H, m, H-6), 4.5–5.0 (m, sugar protons), 5.27 (1H, d, J = 6 Hz H-1), 6.45 (1H, d, J = 16 Hz, H- $\alpha$ ), 6.94 (2H, d, J = 10 Hz, H-3" and H-5"), 7.40 (1H, s,H-3), 7.62 (1H, d, J = 16 Hz, H- $\beta$ ), 7.68 (2H, d, J = 10 Hz, H-2" and H-6"). <sup>13</sup>C NMR (100 MHz) see Table 1.

Arbortristoside A pentaacetate (2a). Acetylation of 2 was carried out using a standard procedure, mp 85° (EtOH) (lit. 85–87°),  $C_{37}H_{44}O_{18}$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$ 1.90, 2.00, 2.02, 2.09 (15H, s, 5 × OAc), 4.23 (2H, m, --CH<sub>2</sub>OAc), 5.35 (1H, m, H-1'); <sup>13</sup>C NMR (100 MHz) see Table 1.

*Mannich hydrolysis* of 2 afforded the aglucone 2b in small yield and glucose as the sugar moiety identified by PC. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): 1.17 (3H, d, J = 6.5 Hz, - Me), 2.10 (1H, m, H-8), 2.16 (1H, m, H-9), 3.08 (1H, dd, J = 2, 8 Hz, H-5), 3.76 (3H, s, COOMc), 3.83 (3H, s, OMe), 4.20 (1H, d, J = 4.5 Hz, H-7), 4.96 (1H, dd, J = 2, 4 Hz, H-6), 5.42 (1H, t, J = 5.5 Hz, H-1), 6.37 (1H, d, J = 16 Hz, H-2), 6.87 (2H, d, J = 10 Hz, H-3" and H-5"), 7.37 (2H, d, J = 10 Hz, H-2" and H-6"), 7.43 (1H, s, H-3), 7.57 (1H, d, J = 16 Hz, H- $\beta$ ).

Alkaline hydrolysis of 2. Methanolic KOH yielded p-methoxycinnamic acid, identified by co-TLC, mmp and superimposable IR with an authentic sample and another compound which on methylation with  $CH_2N_2$  resulted in the formation of  $6\beta$ hydroxy loganin.

Compound 3. Prep. TLC (EtOAc-MeOH- $H_2O$ ; 14:3:3) of a portion of fraction B yielded 3 ( $R_f$  0.5), recrystallized as colourless needles (20 mg) from EtOH- $H_2O$ , mp 221-223° (lit. [8] 220-222°),  $C_{17}H_{26}O_{11}$ .

Conversion of 1 to arbortristoside A. To a soln of 1 (50 mg) in MeOH was added a small excess of  $CH_2N_2$ -Et<sub>2</sub>O and the reaction mixt. kept overnight. Solvent was evapd in vacuo and the product purified by prep. TLC (CHCl<sub>3</sub>-MeOH; 43:7) to give the Me ester of 1 (23 mg) as a white amorphous powder, which on recrystallization from EtOH was identified as arbortristoside A (co-TLC, mp, superimposable IR).

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