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PII: S0040-4039(18)31527-2
DOI: <https://doi.org/10.1016/j.tetlet.2018.12.067>
Reference: TETL 50530

To appear in: *Tetrahedron Letters*

Received Date: 6 November 2018
Revised Date: 26 December 2018
Accepted Date: 27 December 2018

Please cite this article as: Swapnaja, K.J., Yennam, S., Chavali, M., Design and Synthesis of Spirobiisoxazoline Dibenzoquinone Derivatives via [3+2] Double 1,3-Dipolar Cycloaddition Reaction, *Tetrahedron Letters* (2018), doi: <https://doi.org/10.1016/j.tetlet.2018.12.067>

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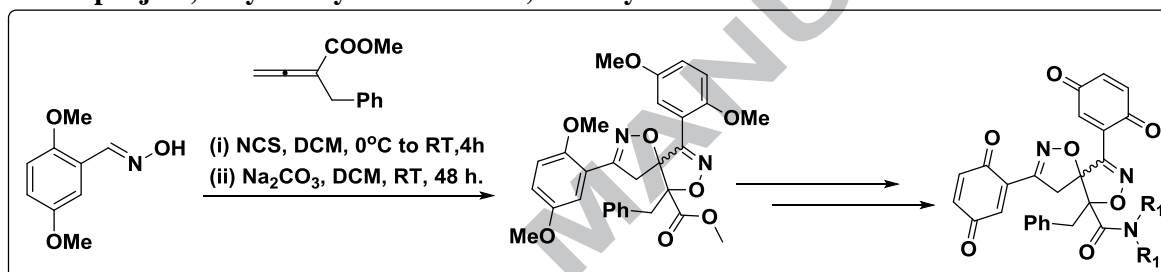
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Design and Synthesis of Spirobiisoxazoline Dibenzoquinone Derivatives via [3+2] Double 1,3-Dipolar Cycloaddition Reaction

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K. Jones. M. Swapnaja^{a,b}, Satyanarayana Yennam^{a*}, Murthy Chavali^c





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Design and Synthesis of Spirobiisoxazoline Dibenzoquinone Derivatives via [3+2] Double 1,3-Dipolar Cycloaddition Reaction

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Abstract: “A series of novel spirobiisoxazoline dibenzoquinone derivatives were synthesized starting from 2,5-dimethoxybenzaldehyde in a six-step synthetic sequence”. The key step [3+2] double 1,3-dipolar cycloaddition of oxime chloride with allenolate was performed under mild reaction conditions using sodium carbonate at ambient temperature. This is the first innovative synthesis of Spirobiisoxazoline Dibenzoquinone system where quinone ring is alkylated to isoxazoline moiety.

Spiro compounds are a particular class of naturally occurring substances characterized by highly pronounced biological properties [1-2]. Keeping in view of diverse biological activities associated with spiro heterocyclic compounds, it was thought to construct a novel system by the conjugation of these bioactive rings together in a single molecular framework to see the additive effects towards their biological activities.

Natural and nonnatural isoxazolines and spirocyclic isoxazoline derivatives have emerged as promising drug candidates due to their significant biological activities [3] resulting in an overall antiproliferative effect on a variety of cancer cell lines. The natural product 11-deoxyfistularin-3 [4] contains two spirocyclic isoxazoline moieties and is cytotoxic toward estrogen dependent MCF-7 breast cancer cells (LD50= 17 mg/L). Other closely related spirocyclic isoxazoline natural products such as fistularin-3, [7] psammaphysins A, [5] exhibits cytotoxic activity and aeriothionin, homoaeriothio-nin [6], Fig. 1 are known for antimicrobial activity. Quinones have attained great interest from a medical and toxicological perspective due to their unique

reactivity and high prevalence in the environment. Natural and synthetic quinone derivatives are well known for antitumor activity [9]. Many drugs such as daunorubicin, doxorubicin Fig.2, epirubicin, mitomycin, mitoxantrones, and saintopin, which are used clinically in the therapy of solid cancers [10]. The quinone anticancer compounds undergo enzymatic reduction via one or two electrons to give the corresponding semiquinone radical or hydroquinone. These semiquinone and the superoxide radical anion will generate the hydroxyl radical and which is the cause of DNA strand breaks [11].

In recent years, the strategy employing to design new bioactive hybrid molecules gained more importance due to their improved biological activity compared to the standard drugs. Such dual action compounds, or hybrid compounds, offer the possibility to overcome the current resistance and reduce the appearance of new resistant strains [12]. Based on this hypothesis, previously we have synthesized quinone five membered phenyl/heterocyclic hybrids [13-16].

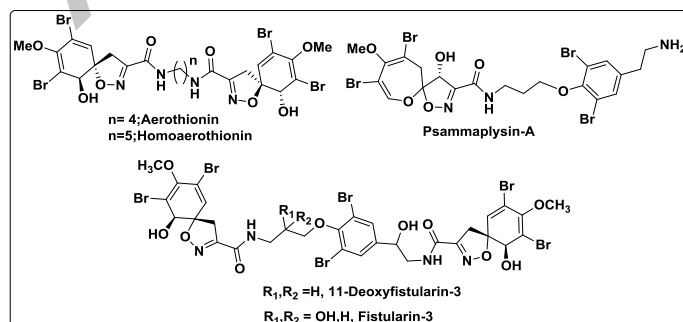


Fig. 1. Spiroisoxazoline Natural Products

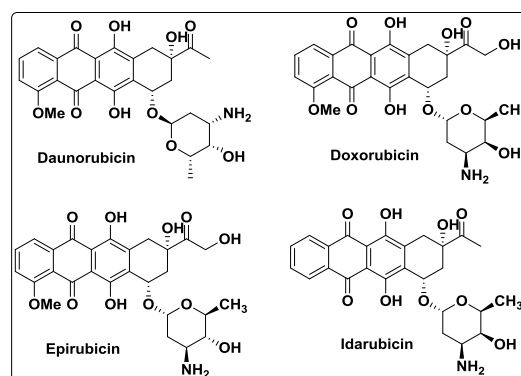


Fig. 2. Known Quinone Drugs

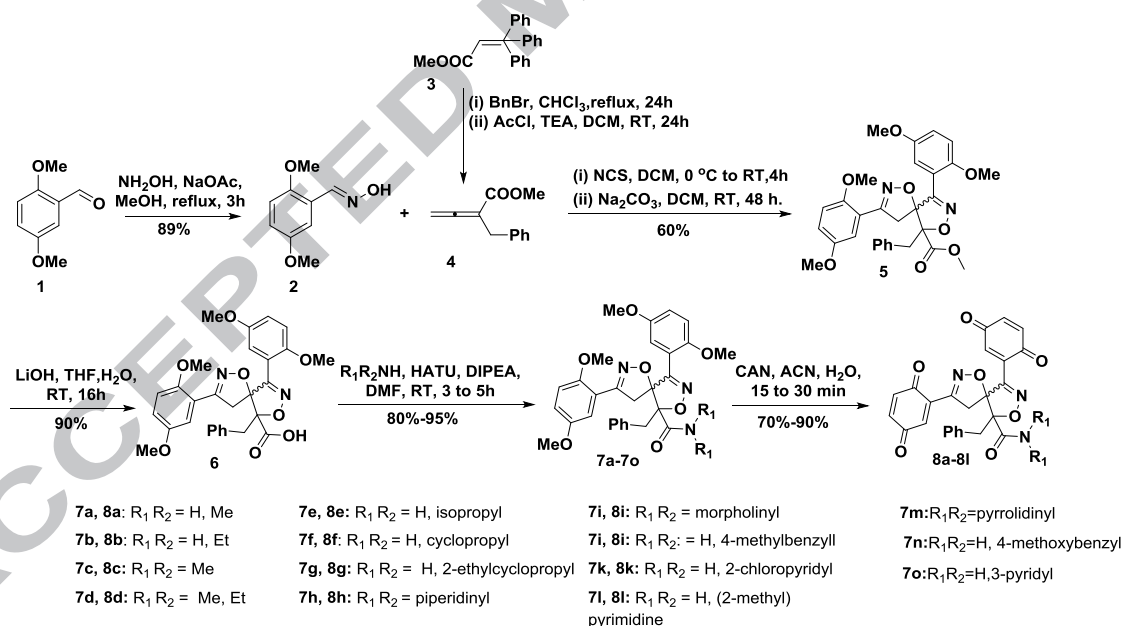
In this rationale, we are encouraged to discover rarely addressed new class of spirobiisoxazoline derivatives by double 1,3-dipolar cycloaddition reaction and the conjugates of these new molecules with biologically active quinones. Herein we have designed and synthesized spirobiisoxazoline dibenzoquinone derivatives as new hybrid molecules to identify more potent biologically active compounds.

Several cycloaddition reactions were reported to approach isoxazoline and spiroisoxazoline skeleton over the years. [17] The strategy based on the cycloaddition of nitrile oxide, has been revealed to be very feasible and trustworthy by many research groups. [18] Whereas, in the above reports, limited reaction partners (alkenes) were reacted with nitrile oxides to construct the isoxazoline frameworks [19] and very few reports were addressed the formation of spirocyclic biisoxazolines *via* intramolecular double 1,3-dipolar cycloaddition of dienes. [20]. Therefore, evolving further approaches is still highly desirable.

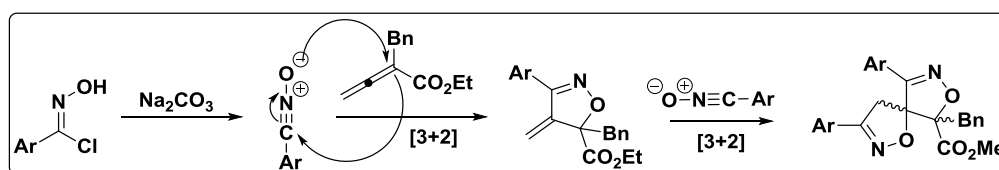
Many cycloaddition reactions of allenates described the formation of only carbon-carbon double bond.[21] Zecchiet al. reported the minor formation of spirobiisoxazoline from the cycloaddition between nitrile oxide and substituted allene. [22] Guo et al. have reported

an efficient double 1,3-dipolar cycloaddition of allenates with nitril-imines for the construction of spirobidiisoxazolines. [23] Recently Xinye Shang et al. addressed the double 1,3-dipolar cycloaddition involved DABCO combined with Et₃N, 2-substituted buta-2,3-dienoates reacted with oxime chlorides to afford spirobiisoxazolines.[24]

Herein we are interested to develop a mild, simple “thermal free” methodology to synthesize spirobiisoxazolines by inorganic base (Na₂CO₃) mediated double “1, 3 dipolar” cycloaddition reaction at room temperature for easy workup and isolation in very good yield. Further, a literature survey reveals no report on the synthesis of spirobiisoxazoline with dibenzoquinone novel nucleus yet so far. As part of our ongoing program to develop efficient compounds and robust methods for the preparation of biologically relevant molecules from readily available building blocks, which are novel yet resemble known biological activity by virtue of the presence of some critical structural features, we have developed a facile, simple and mild methodology for the synthesis of novel spiro biisoxazoline dibenzoquinone derivatives.



Scheme1. Synthesis of Spirobiisoxazoline Dibenzoquinone Derivatives (8a-8m).



Scheme 2. Plausible mechanism of double 1,3 Dipolar cycloaddition reaction

For the synthesis of spirobiisoxazoline dibenzoquinone compounds **8a-8l**, (Scheme 1) commercially available 2,5-dimethoxybenzaldehyde (**1**) was reacted with hydroxylamine hydrochloride in MeOH at reflux to afford oxime **2** (89%). Oxime **2** was treated with NCS in DCM to afford oxime chloride which on further treatment with allenolate **4** resulted in dimethoxy spirobiisoxazoline **5**. Therefore, a reactive intermediate nitrile oxide could be easily generated in situ from oxime chloride in the presence of Na_2CO_3 in DCM and the nitrile oxide was expected to react with allenolate to furnish the desired dimethoxy spirobiisoxazoline **5** (Scheme 2) as a diastereomeric mixture (9:1, 60 %) by [3+2] by double 1,3-dipolar cycloaddition. In this investigation, the reaction of oxime chloride precursor with allenolate **4** was chosen as the model reaction with the use of Na_2CO_3 as the base in dichloromethane at RT for 48 h to give the corresponding product spirobiisoxazoline **5** in 60% yield (Table 1, entry 5). Further investigation was carried out to optimize reaction conditions by taking the combination of different solvents such as ACN, DCM, CHCl_3 and THF with different bases like Na_2CO_3 , K_2CO_3 , Cs_2CO_3 and NaOH. The results clearly indicated that the reaction media played an important role in the process (entries 1–

16). An enhancement of the yield was achieved when the reaction was carried out in DCM with Na_2CO_3 (60%) at RT for 48h. After that, the reaction with K_2CO_3 in CHCl_3 gave good yield (52%), the yields with remaining systems were not desirable. The reaction with NaOH as base is poor yielding. The base catalyzed hydrolysis of compound **5** gave the key intermediate carboxylic acid derivative **6** (90%), which was subjected to amide coupling with various aliphatic, aromatic, acyclic and cyclic, primary/secondary amines. Acid amine coupling was carried out by using HATU as coupling reagent, which afforded corresponding amide derivatives (**7a-7o**) in 80-95% yield. These amide derivatives (**7a-7o**) upon oxidation with ceric ammonium nitrate in ACN/Water at 0 °C resulted in spirobiisoxazoline dibenzoquinone derivatives (**8a-8l**) as yellow solids with very good yield (70-90%) (Fig.3). However, quinones (**8m, 8n and 8o**) obtained with some of the amides **m** (pyrrolidine), **n** (4-methoxybenzyl), **o** (3-pyridyl) were found to be unstable even at 0 °C and thus we were unsuccessful in isolating these compounds. The synthesized stable compounds (**8a-8l**) were well characterized by “IR, ^1H NMR, ^{13}C NMR & HRMS spectrometry”.

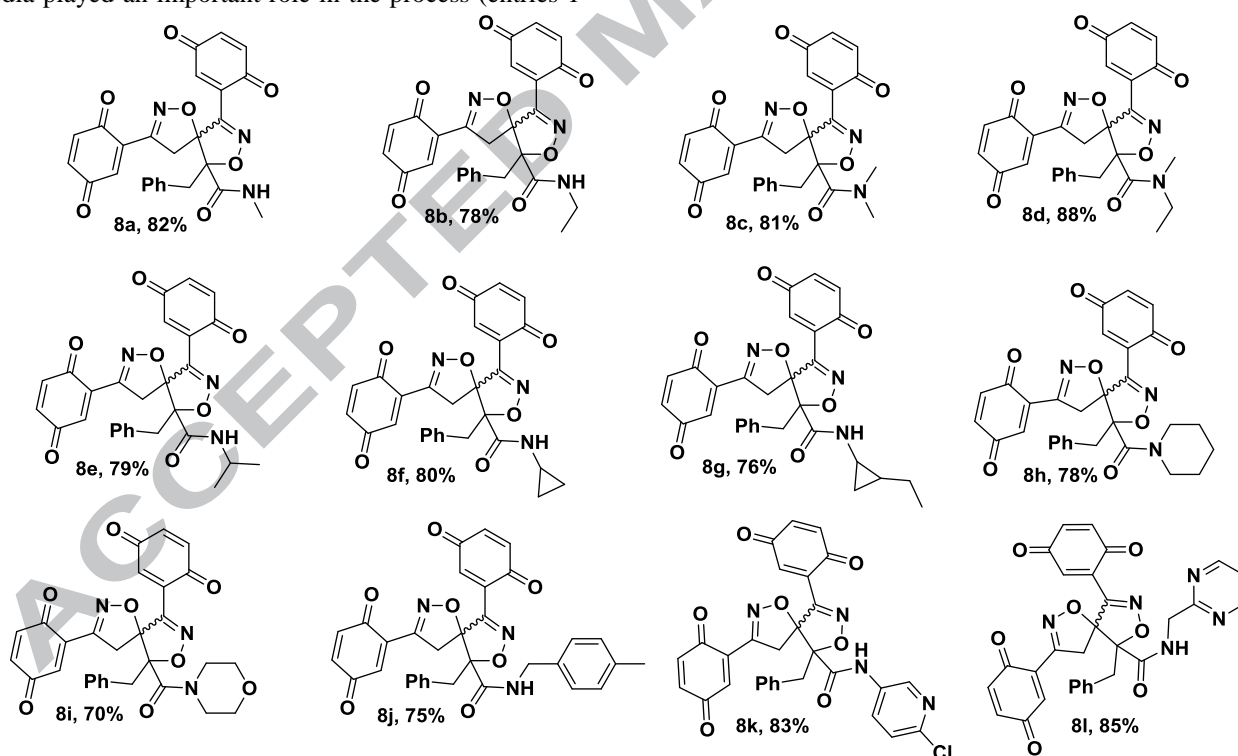


Fig. 3. Spirobiisoxazoline Dibenzoquinone Derivatives (**8a-8l**)

Table 1. Optimization of reaction conditions

Sl.No	Solvent	Conditions	Yield
1	ACN	Na ₂ CO ₃ , RT, 64h	32%
2	ACN	CS ₂ CO ₃ , RT, 48h	22%
3	ACN	K ₂ CO ₃ , RT, 48h	32%
4	ACN	NaOH, RT, 48h	10%
5	DCM	Na₂CO₃, RT, 48h	60%
6	DCM	CS ₂ CO ₃ , RT, 48h	25%
7	DCM	K ₂ CO ₃ , RT, 48h	48%
8	DCM	NaOH, RT, 48 h	15%
9	CHCl ₃	Na ₂ CO ₃ , RT, 48h	40%
10	CHCl ₃	CS ₂ CO ₃ , RT, 48 h	28%
11	CHCl ₃	K ₂ CO ₃ , RT, 48h	52%
12	CHCl ₃	NaOH, RT, 48h	12%
13	THF	Na ₂ CO ₃ , RT, 48h	25%
14	THF	CS ₂ CO ₃ , RT, 48h	10%
15	THF	K ₂ CO ₃ , RT, 48h	10%
16	THF	NaOH, RT, 48h	traces

In conclusion we synthesized a series of novel spirobioxazoline dibenzoquinone derivatives via [3+2] double 1, 3 dipolar cycloaddition reaction. Hence quinones of spirobiisoxazoline scaffolds are featured in a range of natural and synthetic products with wide-reaching biological activities and useful synthetic intermediates for organic synthesis, spirobiisoxazoline dibenzoquinone based compounds have the ability to become drugs of immense use.

Acknowledgements

Authors are grateful to GVK Biosciences Pvt. Ltd. for the financial support and encouragement. Help from the analytical department for the analytical data is appreciated. We thank Dr. Sudhir Kumar Singh for his invaluable support and motivation.

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Research Highlights

- Novel spirobiisoxazoline dibenzoquinone derivatives were synthesized.
- [3+2] Double 1,3-dipolar cycloaddition was performed using allenolate and nitrile oxide.
- The new molecules were successfully separated and characterized.