

Article

Subscriber access provided by TULANE UNIVERSITY

Optimal Hydrophobicity in ROMP-based Protein Mimics Required for siRNA Internalization

Brittany M. deRonde, Nicholas D. Posey, Ronja Otter, Leah M. Caffrey, Lisa M Minter, and Gregory N. Tew Biomacromolecules, Just Accepted Manuscript • DOI: 10.1021/acs.biomac.6b00138 • Publication Date (Web): 21 Apr 2016 Downloaded from http://pubs.acs.org on April 23, 2016

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



Biomacromolecules is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Optimal Hydrophobicity in ROMP-based Protein Mimics Required for siRNA Internalization Brittany M. deRonde[†], Nicholas D. Posey[†], Ronja Otter[†], Leah M. Caffrey, Lisa M. Minter^{§,‡}, and Gregory N. Tew^{*,†,§,‡} [†] Department of Polymer Science and Engineering, University of Massachusetts Amherst, Amherst, MA 01003 [§] Department of Veterinary and Animal Sciences, University of Massachusetts Amherst, Amherst, MA 01003 ^{*}Molecular and Cellular Biology Program, University of Massachusetts Amherst, Amherst, MA 01003 Exploring the role of polymer structure for the internalization of ABSTRACT.

ABSTRACT. Exploring the role of polymer structure for the internalization of biologically relevant cargo, specifically siRNA, is of critical importance to the development of improved delivery reagents. Herein, we report guanidinium-rich protein transduction domain mimics (PTDMs) based on a ring-opening metathesis polymerization scaffold containing tunable hydrophobic moieties that promote siRNA

internalization. Structure-activity relationships using Jurkat T cells and HeLa cells were explored to determine how the length of the hydrophobic block and the hydrophobic side chain compositions of these PTDMs impacted siRNA internalization. To explore the hydrophobic block length, two different series of diblock copolymers were synthesized: one series with symmetric block lengths and one with asymmetric block lengths. At similar cationic block lengths, asymmetric and symmetric PTDMs promoted siRNA internalization in the same percentages of the cell population regardless of the hydrophobic block length; however, with twenty repeat units of cationic charge, the asymmetric block length had greater siRNA internalization, highlighting the non-trivial relationships between hydrophobicity and overall cationic charge. To further probe how the hydrophobic side chains impacted siRNA internalization, an additional series of asymmetric PTDMs was synthesized that featured a fixed hydrophobic block length of five repeat units that contained either dimethyl (dMe), methyl phenyl (MePh), or diphenyl (**dPh**) side chains and varied cationic block lengths. This series was further expanded to incorporate hydrophobic blocks consisting of diethyl (dEt), diisobutyl (diBu), and dicyclohexyl (dCy) based repeat units to better define the hydrophobic window for which our PTDMs had optimal activity. HPLC retention times quantified the relative hydrophobicities of the non-cationic building blocks. PTDMs containing the MePh, diBu, and dPh hydrophobic blocks were shown to have superior siRNA internalization capabilities compared to their more and less hydrophobic counterparts, demonstrating a critical window of relative hydrophobicity for optimal internalization. This better understanding of how hydrophobicity impacts PTDM-induced internalization efficiencies will help guide the development of future delivery reagents.

Biomacromolecules

1 Introduction

Intracellular delivery of therapeutics, particularly siRNA, continues to be a challenge for the biomedical community.^{1,2} Transient gene knockdown plays an important role in the exploration of molecular pathways and in the development of more advanced treatment options; the field, however, needs a clearer understanding of how to efficiently and reliably deliver bioactive molecules across cellular membranes, particularly for primary human cells.²⁻⁸ Nevertheless, nature is already capable of designing proteins that can perform these functions.⁹⁻¹¹ One example is HIV-1 TAT, which is partly responsible for the spread of the viral genome of HIV.^{9,10} This peptide contains a region referred to as a protein transduction domain (PTD) which enables the protein to enter cells.¹²⁻¹⁴ These regions in proteins are generally cation-rich, containing lysine and arginine residues, which aid in cellular uptake. Structure-activity relationships (SARs) related to this protein, as well as others such as the Antennapedia homeodomain protein, led to the development of a field referred to as cell-penetrating peptides (CPPs), which are capable of delivering cargo including small molecules, siRNA, pDNA, and proteins into cells.¹⁵⁻²⁰ Three classical examples of CPPs include TAT49-57, Pep-1, and MPG, where TAT is an arginine rich peptide based on the PTD of the HIV-1 TAT protein and PEP-1 and MPG are lysine rich, primary, amphipathic peptides.^{13,14,20-24}

Although extensive work has been devoted to exploring CPPs for siRNA delivery applications,²⁴⁻²⁷ the extension of design principles learned from these systems to the development of synthetic mimics, referred to as cell-penetrating peptide mimics (CPPMs) or protein transduction domain mimics (PTDMs), offers many distinct advantages.^{28,29} By breaking out of the synthetic confinement of amino acids, a wider range of

chemistries can be used to manipulate key features of CPPs, including hydrophobic segregation as well as cationic charge content.²⁸ This field of mimetic polymer chemistry has already demonstrated a range of polymer scaffolds for the development of siRNA delivery reagents,^{28,30,31} including those based on polyoxanorbornene.^{28,30,32} polymethacrylamide,³³ arginine-grafted bioreducible polydisulfide,^{34,35} and oligocarbonate backbones.³⁶⁻³⁸ Similar design principles were previously used for the creation of antimicrobial peptide mimics, where the facially amphiphilic structures of natural peptides were successfully recapitulated using highly modular synthetic scaffolds.³⁹⁻⁴³ In order to realize the full potential of these PTDM materials and continue to improve internalization and delivery efficiencies, extensive SARs studies are necessary to elucidate key design parameters.

To this end, our research group has devoted an extensive amount of research into understanding how the structures of ring-opening metathesis (ROMP) based protein mimics influence their membrane interactions,^{29,40,44-48} cellular uptake efficiencies,^{29,49} and siRNA delivery.^{30,32} We have demonstrated the utility of the platform for the successful internalization of siRNA and for the knockdown of active biological genes in T cells.^{30,32} Previous SARs established that there was a critical cationic block length required for efficient siRNA delivery, which, not surprisingly, showed some cell-type dependencies.³² Additionally, the incorporation of a fixed-length, segregated, hydrophobic segment into the PTDM platform improved siRNA internalization efficiencies by six fold compared to their homopolymer counterparts with the same relative cationic block lengths.³² From our preliminary studies, a great deal was learned

Biomacromolecules

about the cationic block length;³² however, further studies were needed to understand how the amount and type of hydrophobic content influenced siRNA internalization.

Many literature reports demonstrate that adding hydrophobicity, either through direct incorporation or through the use of bulky counter ions, generally improves membrane interactions, cellular uptake, and delivery efficiencies of CPPs and their synthetic mimics.^{30,32,37,47,48,50-56} Pep-1 and MPG, two common CPPs used for protein and siRNA delivery, respectively, have amphipathic structures where their hydrophobic components improve membrane interactions and internalization efficiencies.^{23,24} In light of these trends and the wide variety of hydrophobic groups available, it is important to understand how the length of the hydrophobic block within the PTDMs and the nature of the hydrophobic moiety impacts siRNA internalization. At the same time the cationic block length particularly in relation to the corresponding hydrophobic block, should not be overlooked as it also influences PTDM activity.

Structure activity relationships using Jurkat T cells and HeLa cells were used to probe how the length and the nature of the hydrophobic block impacted siRNA internalization. To explore the length of the hydrophobic block, two series of diblock copolymers (n and m, where n is the hydrophobic block length and m is the cationic block length) with symmetric (n = m) or asymmetric (n \neq m) block lengths were synthesized (Figure 1). In a separate series of PTDMs, the hydrophobic side chain was varied to probe how the relative hydrophobicity impacted siRNA internalization. After initial testing, this set of PTDMs was expanded strategically to incorporate additional hydrophobic side chains to more specifically understand how hydrophobicity influences Relative hydrophobicities of the hydrophobic building blocks were internalization.

estimated using HPLC retention times to guide the selection of structures used for
 analysis. Probing the effects of hydrophobicity on siRNA internalization efficiencies
 enabled us to better elucidate the essential design principles for our PTDMs.

EXPERIMENTAL SECTION

Monomer Synthesis and Characterization. Monomers were synthesized using a two-step process previously reported by our group.³² In brief, oxanorbornene anhydride was ring-opened using the desired alcohol and 4-dimethylaminopyridine (DMAP) as a catalyst to yield the half-ester intermediates. The intermediates that formed precipitates were isolated using vacuum filtration whereas those that did not were recrystallized from a mixture of chloroform/hexanes (3/1, v/v). The half-esters were then further reacted with a second desired alcohol using 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC) coupling to yield the difunctional monomer. A one-pot synthesis was used for monomers designed to display two of the same functionalities. All monomers were purified using a CombiFlash system using ethyl acetate/dichloromethane (CH₂Cl₂) (1/9, v/v) and subsequently analyzed by ¹H NMR spectroscopy, ¹³C NMR spectroscopy, and mass spectrometry (MS) to assess their chemical compositions and purity. Detailed synthetic procedures and all characterization data are provided in the supporting information.

PTDM Synthesis and Characterization. All block copolymer PTDMs were 20 synthesized by ROMP using Grubbs' third generation catalyst following previously 21 described methods.³² In brief, the monomers and catalyst were dissolved separately in 22 CH₂Cl₂ and degassed using freeze-pump-thaw methods. To initiate the polymerization,

Page 7 of 29

Biomacromolecules

the hydrophobic monomer was first cannulated into the catalyst solution followed by the Boc-protected guanidinium monomer. The cationic monomer was polymerized in protected form in order to prevent premature termination of polymerizations due to catalyst coordination and to allow for sufficient solubility in organic solvents. Polymers were quenched with ethyl vinyl ether, precipitated, and subsequently deprotected using a 1:1 ratio of CH₂Cl₂:trifluoroacetic acid (TFA). Excess TFA was removed by azeotropic distillation with methanol and polymers were then dialyzed against reverse osmosis (RO) water using dialysis membranes with a molecular weight cutoff (MWCO) of 100-500 g/mol until the water conductivity was $< 0.2 \ \mu$ S. All polymers were characterized by ¹H NMR and gel permeation chromatography (GPC) to assess chemical compositions and molecular weight distributions, respectively. Detailed synthetic procedures and all characterization data are provided in the supporting information.

Jurkat T Cell Culture and FITC-siRNA Internalization. Jurkat T cells were cultured in RPMI 1640 supplemented with 10% (v/v) fetal bovine serum (FBS), 100 U/mL nonessential amino acids, 100 U/mL sodium pyruvate, 100 U/mL penicillin, and 100 U/mL streptomycin. These cells were incubated at 37 °C with 5% CO₂ and passaged 24 hours prior to experimentation. On the day of the experiment, PTDMs and siRNA were mixed at an N:P ratio of 8:1 (50 nM siRNA/well) and incubated for 30 minutes at room temperature prior to adding them drop-wise to the cells $(4x10^5 \text{ cells/well}; 1 \text{ mL total})$ volume) in a 12 well plate. Cells were incubated at 37 °C with 5% CO₂ in serum-containing media for four hours prior to analysis by flow cytometry. Cell viability was assessed using 7-AAD staining.

HeLa Cell Culture and FITC-siRNA Internalization. HeLa cells were cultured in DMEM supplemented with 10% (v/v) FBS, 100 U/mL non-essential amino acids, 100U/mL sodium pyruvate, 100 U/mL penicillin, and 100 U/mL streptomycin. For these experiments, cells $(5x10^4 \text{ cells/well}; 1 \text{ mL total volume})$ were cultured in 12-well plates using serum-containing media for 48 hours. This led to cell populations that were 70-90% confluent on the day of the experiment. On the day of the experiment, PTDMs were and siRNA were mixed at an N:P ratio of 4:1 (50 nM siRNA/well) and incubated for 30 minutes at room temperature. The cell media was replaced with fresh, complete media prior to adding the PTDM/siRNA complexes carefully to the top of the sample wells. Cells were incubated at 37 °C with 5% CO₂ for four hours in serum-containing media prior to analysis by flow cytometry. Cell viability was assessed using 7-AAD staining.

HPLC Assessment of Hydrophobicity. HPLC experiments were performed using an HP 5890 HPLC system equipped with a photodiode array detector using an Agilent Zorbax SB-C₈, 80 Å, 4.6 x 150 mm ID (5 µm) column. Samples were eluted using a linear gradient of 100% water with 0.1% TFA to 100% acetonitrile with 0.1% TFA over 60 minutes at a flow rate of 1 mL/min, and were detected using a wavelength of 215 nm. HPLC retention times were compared as a means to assess relative hydrophobicity, with more hydrophobic monomers eluting at higher retention times (RTs). These methods were previously reported in the literature.⁵⁷⁻⁵⁹

RESULTS AND DISCUSSION

22 Initial PTDM Design and Characterization

Page 9 of 29

Biomacromolecules

In this report, we document the use of ring-opening metathesis polymerization (ROMP) for the synthesis of block copolymer (BCP) PTDMs with varying hydrophobic block lengths. ROMP is a facile polymerization method that is functional group tolerant and allows for good control over molecular weights and dispersities.⁶⁰⁻⁶⁷ In addition, we continue to exploit the versatile, dual-functional oxanorbornene-based monomer platform as a means to tailor the overall polymer properties;^{29,30,32,44,47-49,68,69} in this case with a focus on the hydrophobic block length and side chain structure. The monomers and corresponding polymers for this study are summarized in Figure 1. The guanidinium-rich monomer (dG), shown in its Boc-protected form and meant to mimic arginine residues, was used as the cationic component of the BCP PTDMs. This selection was made based on previous literature that demonstrates the superior performance of guanidinium moieties over their ammonium counterparts, which are reminiscent of lysine and ornithine residues.⁷⁰



Figure 1. Monomer and polymer structures used for this study. A) Monomer structures
B) Polymer Structures. Blue represents cationic moieties and green represents
hydrophobic moieties.

The three initial hydrophobic monomers were designed to contain either two methyl substituents (dimethyl, **dMe**), one methyl and one phenyl substituent (methyl phenyl, MePh), or two phenyl substituents (diphenyl, dPh) (Figure 1A) and were selected because HPLC retention times indicated that they spanned a range of hydrophobicities. The phenyl-based hydrophobic groups have also demonstrated useful activities in prior studies.^{30,32,44,47} MePh, which has traditionally been selected by our group as the hydrophobic component of PTDMs for siRNA delivery, was used to synthesize BCPs with symmetric (n = m = 5, 10, 20, or 40) or asymmetric ($n \neq m$, with n being fixed at five for all polymers in the series and m = 10, 20, or 40 block sizes. This allowed us to explore how the length of the hydrophobic block impacts internalization at a given cationic charge length.

Page 11 of 29

Biomacromolecules

In addition, asymmetric BCPs containing a fixed block length (n = 5) of all three hydrophobic monomers and a cationic charge block of five, ten, or twenty **dG** units were synthesized as a way to further probe the relationship between the hydrophobic side chain and the length of the charged block for siRNA internalization. Previous work in this area documented the effect that the cationic charge block length and the addition of a hydrophobic block had on siRNA internalization and delivery.³² In that study, BCP PTDMs significantly outperformed their homopolymer counterparts, recapitulating the importance of an added hydrophobic component.³² The current study builds further from these initial findings to explore how the quantity and type of hydrophobic repeat units used impacts siRNA internalization.

FITC-

FITC-siRNA Internalization: Symmetric vs. Asymmetric

Initially, both the symmetric and asymmetric MePh_n-b-dG_m BCP PTDMs were studied. A total of seven PTDMs were designed; four contained equal hydrophobic and cationic block lengths (symmetric, n = m = 5, 10, 20, and 40) and three asymmetric ($n \neq 1$ m) ones contained a fixed hydrophobic block length (n = 5) but cationic charge blocks of 10, 20, and 40 repeat units. Fluorescein isothiocyanate (FITC) labeled siRNA (FITC-siRNA) was used to establish trends in siRNA internalization for both series in Jurkat T cellsand HeLa cells using complete media. In this report, only FITC-labeled siRNA was used since we previously demonstrated the internalization relationship for our PTDMs between FITC-siRNA in Jurkats and siRNA for NOTCH1 in human peripheral blood mononuclear cells (PBMCs). The N:P ratios for these experiments, which are the ratios of the number of positively charged guanidinium groups in the PTDMs to the number of negatively charged phosphate atoms in the FITC-siRNA, were set at 8:1 for Jurkat T cells

and 4:1 for HeLa cells and were previously optimized by our group.³² The selection of these N:P ratios were further supported by gel retardation assays, which demonstrated that all PTDMs fully bound siRNA at N:P ratios of 4:1 or greater (Figures S17-S27).

A summary of FITC-siRNA internalization for these four symmetric and three asymmetric PTDMs is shown in Figure 2, where Figures 2A and 2B present the percentage of the cell population with internalized FITC-siRNA and Figures 2C and 2D present the median fluorescence intensity (MFI) of the cell populations.







2 Figure 2. FITC-siRNA internalization into Jurkat T cells and HeLa cells using symmetric 3 and asymmetric ROMP-based protein mimics. Jurkat T cells (cell density = 4×10^5 4 cells/mL) treated with PTDM/FITC-siRNA complexes with an N:P ratio = 8:1 in 5 complete media for four hours at 37°C and compared cells only receiving FITC-siRNA. HeLa cells (cell density = 5×10^4 cells/mL 48 hours prior to experiment; 70-90% 6 7 confluent on the day of the experiment) treated with PTDM/FITC-siRNA complexes with 8 an N:P ratio = 4:1 in complete media for four hours at 37° C and compared cells only 9 receiving FITC-siRNA. All data was compared to an untreated control. A) Percent 10 positive Jurkat T cells. B) Percent positive HeLa cells. C) MFI of the Jurkat T cell population. D) MFI of the HeLa cell population. Each data point represents the mean ± 11 SEM of three independent experiments. * = p < 0.05 and ns = not significant, as 12 13 calculated by the unpaired two-tailed student *t*-test. Statistics represents significance 14 between symmetric and asymmetric block copolymer PTDMs with the same cationic 15 charge block length.

~	
3	
4	
-	
С	
6	
_	
ſ	
Q	
-	
9	
1	Λ
I	U
1	1
Å	0
I	Z
1	3
	4
ľ	4
1	5
ż	č
1	6
1	7
1	'
1	8
1	a
1	J
2	0
<u>_</u>	4
2	1
2	2
~	ົ
2	3
2	4
~	-
2	5
2	6
~	2
2	7
ი	0
~	0
2	9
م	<u> </u>
S	U
3	1
~	ò
3	2
3	3
2	4
3	4
ຊ	5
2	~
3	6
Q	7
0	1
3	8
S	0
J	9
4	0
,	1
4	1
4	2
,	2
4	J
4	4
,	F
4	S
4	6
ź	-
4	1
Δ	8
ļ	ž
4	9
5	0
2	
5	1
ᄃ	2
ں ۔	2
5	3
ᄃ	Λ
J	4
5	5
Ē	ĉ
С	о
5	7
_	
	0
5	8
55	8 9
55	8 9

1

2 All PTDM-treated cells showed greater than 85% viability using 7-AAD staining 3 (Figure S32 and Figure S38). For both cell types, symmetric and asymmetric PTDMs 4 containing the same cationic block length were able to facilitate siRNA internalization in 5 the same percentages of the cell populations (Figures 2A and 2B) regardless of the 6 different hydrophobic block lengths. When looking at the MFI data, PTDMs with five, 7 ten, and forty cationic repeat units lead to similar internalization amounts regardless of 8 the hydrophobic block length; however, for PTDMs with twenty cationic repeat units, the 9 asymmetric PTDM MePh₅-b-dG₂₀, significantly outperformed its symmetric counterpart 10 (Figures 2C and 2D). For both cell types, samples using MePh₅-b-dG₂₀ had double the 11 fluorescence intensity of samples using MePh₂₀-b-dG₂₀. These results demonstrate that 12 the relationship between hydrophobicity and cationic block length is not always trivial 13 and that increasing the hydrophobic block length further does not guarantee superior performance.³⁷ In addition, these results suggest that understanding the interplay 14 15 between hydrophobic and cationic block lengths as illustrated by MePh₅-b-dG₂₀ is 16 necessary for optimal internalization.

17 FITC-siRNA Internalization: Varying the Hydrophobic Block Side Chains

To further probe the relationship between hydrophobic side chain and cationic block length, PTDMs with a fixed hydrophobic block length of five repeat units of either low (**dMe**), moderate (**MePh**), or high hydrophobicity (**dPh**) were prepared and studied. The relative hydrophobicities of these monomers were determined using HPLC and can be found in Table 1. The given retention times were 14.2 min, 27.8 min, and 36.1 min, respectively, where larger HPLC retention times reflect increased hydrophobicity. Based

Page 15 of 29

Biomacromolecules

on previous results and those shown in Figure 2, the hydrophobic block length was held constant (n = 5) while the cationic block length was varied (m = 5, 10, 20). In addition, based on results from the previous section, BCPs with m = 40 were not studied due to insufficient internalization regardless of hydrophobic content. This generated a total of nine PTDMs. FITC-siRNA was again used to establish trends in siRNA internalization in Jurkat T cells and HeLa cells using the same protocols as previously described.

A summary of FITC-siRNA internalization for the PTDMs with variable hydrophobic side chains is shown in Figure 3, where Figures 3A and 3B present the percentage of the cell population with internalized FITC-siRNA and Figures 3C and 3D present the MFIs of the cell populations. All PTDM-treated cells showed greater than 85% viability using 7-AAD staining (Figure S33 and Figure S39).



dMe₅-b-dG_m, MePh₅-b-dG_m, and dPh₅-b-dG_m PTDM Series

Figure 3. FITC-siRNA internalization into Jurkat T cells and HeLa cells using ROMP-based protein mimics with different hydrophobic blocks. Jurkat T cells (cell density = $4x10^{\circ}$ cells/mL) treated with polymer/FITC-siRNA complexes with an N:P ratio = 8:1 in complete media for four hours at 37°C and compared cells only receiving FITC-siRNA. HeLa cells (cell density = 5×10^4 cells/mL 48 hours prior to experiment; 70-90% confluent on the day of the experiment) treated with PTDM/FITC-siRNA complexes with an N:P ratio = 4:1 in complete media for four hours at 37° C and compared cells only receiving FITC-siRNA. All data was compared to an untreated control. A) Percent positive Jurkat T cells. B) Percent positive HeLa cells. C) MFI of the Jurkat T cell population. D) MFI of the HeLa cell population. Each data point represents the mean ± SEM of three independent experiments. * = p < 0.05, ** = p < 0.01, *** = p < 0.001 as calculated by the unpaired two-tailed student *t*-test. Statistics represents significance between dMe-containing and MePh-containing PTDMs.

Page 17 of 29

Biomacromolecules

For both cell types, **MePh-** and **dPh-**containing PTDMs with the same cationic block length were able to facilitate siRNA internalization to the same percentages of the cell populations; however, in all cases, the **dMe**-containing PTDMs facilitate siRNA delivery to a significantly smaller percentage of the population, particularly at the lower cationic block lengths. This percentage increased from 20% (dMe₅-b-dG₅) to 80% (dMe_5-b-dG_{20}) , supporting the idea that cationic charge blocks larger than five (up to 20) improve FITC-siRNA internalization. In addition, for Jurkat T cells, the MFIs for the samples that used MePh- and dPh-containing PTDMs were not statistically different from each other but were significantly higher when compared to samples that used the dMe-containing PTDMs. Specifically, samples that used MePh₅-b-dG₁₀ and dPh₅-b- dG_{10} had four times the fluorescence intensity of samples that used dMe_5-b-dG_{10} , and at higher cationic block lengths, samples that used MePh₅-b-dG₂₀ and dPh₅-b-dG₂₀ had double the fluorescence intensity of samples that used dMe₅-*b*-dG₂₀. This observed trend was similar for HeLa cells when comparing samples that used the MePh₅-b-dG_m series to samples that used the dMe_5-b-dG_m series, with samples that used the MePh-containing PTDMs having roughly double the fluorescence intensity regardless of the cationic block length. Taken together, this data suggests that there may be a minimum hydrophobic threshold necessary for better siRNA internalization, but there also seems to be a limit to which increasing the hydrophobicity improves siRNA internalization (see red circles in Figure S42).

21 Expanding the Hydrophobic Monomer Set

To better understand the required hydrophobicity for optimal internalization and to gain more insight regarding the tunability of the hydrophobic domain, several

additional PTDMs were designed. In order to target the appropriate PTDMs, a series of hydrophobic monomers was synthesized and analyzed using HPLC to assess their relative hydrophobicities (Table 1, Figure S1). Additionally, logP values were calculated using MarvinSketch (ChemAxon Ltd).⁷¹ Monomers with longer retention times required more organic component in the mobile phase in order to be eluted and are considered to be more hydrophobic. A plot of HPLC RT as it relates to logP value can be found in the supporting information (Figure S41). The linear relationship between HPLC RT and logP value supports the use of HPLC RTs as a viable, experimental method for assessing relative monomer hydrophobicity.⁵⁷⁻⁵⁹ It also allows more confidence with calculated logP values for this monomer class going forward. The monomer names in Table 1 reflect the substituents used for the monomer R groups, R₁ and R₂, which can either be the same or different. The wide variety of hydrophobic monomers synthesized in Table 1 also highlights the versatility of the diester monomer platform.⁶⁹

4	
1	
2	
3	
Δ	
5	
5	
6	
7	
8	
0	
9	
10	
11	
12	
13	
10	
14	
15	
16	
17	
18	
10	
19	
20	
21	
22	
<u>~~</u>	
23	
24	
25	
26	
27	
21	
28	
29	
30	
21	
31	
32	
33	
34	
35	
20	
30	
37	
38	
39	
10	
40	
41	
42	
43	
11	
 15	
45	
46	
47	
48	
40	
49	
50	
51	
52	
52	
55	
54	
55	
56	
57	
57	
58	
59	

1 Table 1. Summary of Monomers and Their Corresponding HPLC Retention Times and

2 LogP Values.

				1	^
Monomer Name	Monomer Abbreviation	HPLC Rt ^a (min)	LogP ^b		0 0 0 0
Dimethyl	dMe	14.2	0.05	0	R ₁
Methyl ethyl	MeEt	17.3	0.41	5	
Diethyl	dEt	21.1	0.77	Dicit	
Methyl Propyl	MePr	21.8	0.93	hok	
Methyl Butyl	MeBu	25.8	1.38	rop	
Methyl Phenyl	MePh	27.8	1.78	Hyd	
Dipropyl	dPr	28.6	1.81	ng	
Diisobutyl	diBu	34.8	2.54	asi	
Dibutyl	dBu	34.9	2.70	Icre	
Diphenyl	dPh	36.1	3.50		
Dicyclohexyl	dCy	44.2	4.28		

^aHPLC data collected using a linear gradient from 100% water containing 0.1% TFA to
100% CH₃CN containing 0.1% TFA at a flow rate of 1 mL/min. HPLC detection
wavelength = 215 nm. ^bLogP is the octanol/water partition coefficient. All values were
calculated using MarvinSketch (ChemAxon Ltd.)

8

3

9 The original hydrophobic monomers used for this study, **dMe**, **MePh**, and **dPh**, 10 had HPLC RTs of 14.2 min, 27.8 min, and 36.1 minutes, respectively. From the eleven 11 monomers in Table 1, three new candidates were selected based on their measured 12 hydrophobicities: **dEt** falls in the hydrophobic region between **dMe** and **MePh**, **diBu** has 13 similar hydrophobicity to **MePh** and **dPh**, and **dCy** is the most hydrophobic of the eleven 14 monomers. These three new monomers and their corresponding polymers are 15 summarized in Figure 4.



Figure 4. Additional monomer and polymer structures used for this study. A) Monomer
structures B) Polymer Structures. Blue represents cationic moieties and green represents
hydrophobic moieties.

Due to the high siRNA internalization facilitated by **MePh₅-b-dG₂₀** in Jurkat T cells, these three additional polymers were synthesized with the same cationic (m = 20) and hydrophobic (n = 5) block lengths. Zeta potential and size data were also collected for all six PTDM/siRNA complexes used for these studies in order to ensure trends observed were due to PTDM composition and not due to surface charge and size effects. The zeta potential data is shown in Figure 5.



Figure 5. Zeta potential (A) and size distribution data (B) for PTDM/siRNA complexes at an N:P
 ratio of 8:1 in 10 mM Tris Buffer. Each data point represents the mean ± SEM of three
 independent experiments.

5 This data demonstrates that there is no difference in zeta potential for all of the 6 complexes tested. In addition, the size distributions are similar with no trends between 7 update efficiency and size or size distribution. Although some samples appear to be 8 clearly bimodal while others are monomodal, there is <u>no</u> correlation to internalization. 9 For example, the three best polymers (MePh₅-*b*-dG₂₀, dPh₅-*b*-dG₂₀, diBu₅-*b*-dG₂₀) show 10 one very bimodal and two mainly monomodal distributions while the less active PTDMs 11 include two bimodal and one monomodal distribution. Taken together, it appears that



A summary of the FITC-siRNA internalization data in Jurkat T cells can be found in the supporting information (Figure S38) while the MFI values for these PTDMs, plotted as a function of the monomers' HPLC RTs, is shown in Figure 6. This shows the initial three PTDMs (**dMe**₅-*b*-**dG**₂₀, **MePh**₅-*b*-**dG**₂₀, and **dPh**₅-*b*-**dG**₂₀, red circles) with the three new PTDMs (**dEt**₅-*b*-**dG**₂₀, **diBu**₅-*b*-**dG**₂₀, and **dCy**₅-*b*-**dG**₂₀, blue squares).



Figure 6. Plot of relative FITC fluorescence in Jurkat T cells as it relates to monomer
HPLC retention times. Green dashed lines indicate the hydrophobic window for optimal
PTDM performance. Red data points represent hydrophobic monomers initially used.
Blue data points represent hydrophobic monomers added after monomer hydrophobicity
assessment by HPLC.

Page 23 of 29

Biomacromolecules

From the initial studies, there appeared to be a hydrophobic threshold for optimal internalization ($dMe_{MFI} < MePh_{MFI} \sim dPh_{MFI}$) shown by the green dashed line near RT = 24 minutes. The new **dEt**-containing PTDM appears to support this hypothesis given that it facilitates similar amounts of FITC-siRNA internalization compared to the dMe-containing PTDM despite being more hydrophobic. Increasing the monomer hydrophobicity with **diBu**-containing PTDMs beyond the "threshold value" yields siRNA internalization values similar to **MePh**- and **dPh**-containing PTDMs; however, a further increase in the monomer hydrophobicity to dCy yields a PTDM with decreased FITC-siRNA internalization. This would suggest a window of optimal hydrophobicity from RTs of 27.8 minutes to 36.1 minutes when using these PTDMs for siRNA internalization. Furthermore, polymers that fall within this critical hydrophobic window are comprised of PTDMs with both aromatic and non-aromatic side chains, suggesting that overall hydrophobicity may be more important than monomer side chain structure.47,48

15 CONCLUSIONS

Understanding the structural components of carrier molecules necessary for siRNA internalization is critical for the development of better delivery reagents. In efforts to better determine appropriate design principles for our ROMP-based PTDMs, several new PTDMs were designed to understand how the length and relative hydrophobicity of the non-charged block as well as the length of the charged block had on FITC-siRNA internalization efficiencies in Jurkat T cells and HeLa cells. Initially, a set of symmetric (n = m) and asymmetric (n \neq m, with n being fixed at five for all polymers in the series) BCP PTDMs were tested with varied cationic block lengths (m =

5, 10, 20, or 40). At fixed cationic block lengths, the percentage of the cell population receiving FITC-siRNA remained the same regardless of the hydrophobic block length; however, the samples that used asymmetric **MePh₅-b-dG**₂₀ had twice the fluorescence intensity of samples that used MePh₂₀-b-dG₂₀, demonstrating the complex relationship between hydrophobic and cationic block lengths. In a separate series of polymers, the hydrophobic block length was held constant at five repeat units, but the hydrophobic component was varied from dMe-, to MePh-, to dPh-based repeat units, which represented a range of hydrophobicities. In this series, the **dMe**-based PTDMs exhibited diminished internalization in comparison to their more hydrophobic counterparts (MePh-and **dPh**-based PTDMs). This suggested there was a minimum hydrophobicity required for improved internalization. HPLC retention times were used to assess the relative hydrophobicity of new monomers and to select several for the design and preparation of additional PTDMs, (dEt₅-b-dG₂₀, diBu₅-b-dG₂₀, and dCy₅-b-dG₂₀) to more fully explore this optimal hydrophobic window. An approximately 10 minute retention time window between 27 and 37 (or logP values of 1.78 and 3.50) was shown to yield optimal PTDM siRNA internalization. Below this threshold, PTDMs lack sufficient hydrophobicity to promote optimal internalization and the PTDM above this hydrophobicity also showed diminished internalization capabilities. Overall, optimization of PTDM hydrophobicity led to a better understanding of the structural components necessary for siRNA internalization, which will be used in the future to guide the development of superior delivery reagents.

23 ASSOCIATED CONTENT

Biomacromolecules

2		
3		
4		
5		
6		
7		
8		
a		
1	^	
1	2	
1	1	
1	2	
1	3	
1	4	
1	5	
1	6	
1	7	
1	א	
1	a	
י ר	0	
2	0	
2	1	
2	2	
2	3	
2	4	
2	5	
2	6	
2	7	
2 2	0	
2	0	
2	9	
3	0	
3	1	
3	2	
3	3	
3	4	
3	5	
ч 2	6	
2	7	
ა ი	0	
3	ð	
3	9	
4	0	
4	1	
4	2	
4	3	
4	4	
4	5	
л Л	е А	
4	7	
4	1	
4	8	
4	9	
5	0	
5	1	
5	2	
5	3	
5	⊿	
ᄃ	F	
о г	ວ ດ	
о г	0	
5	1	
5	8	
5	9	

60

1 All detailed synthetic procedures, molecular characterization, biological assays, and 2 cellular viability data are provided in the supporting information. This material is 3 available free of charge via the internet at http://pubs.acs.org.

4 AUTHOR INFORMATION

5 Corresponding Author:

6 *E-mail: tew@mail.pse.umass.edu.

7 Notes:

8 The authors declare no competing financial interest.

9 Author Contributions

10 The manuscript was written through contributions of all authors. All authors have given11 approval to the final version of the manuscript.

12 ACKNOWLEDGMENT

13 This work was funded by the NIH (T32 GMO8515) and NSF (CHE-0910963 and DMR-14 1308123). The authors would like to thank Ms. Angie Korpusik, and Ms. Salimar 15 Cordero-Mercado for help with monomer synthesis and Ms. Kelly McLeod for feedback 16 on early drafts of this manuscript. The authors would also like to thank Mr. Joe A. Torres 17 for invaluable discussions about data and for assistance with gel electrophoresis and Dr. 18 Rachel Letteri for advice on size and zeta potential sample preparation. Mass spectral 19 data were obtained at the University of Massachusetts Mass Spectrometry Facility, which 20 is supported in part by NSF. Flow cytometry data were obtained using the Flow

1	Cytometry Core Facility at the University of Massachusetts Amherst, which is supported
2	in part by NSF.
3	REFERENCES
4	(1) Lorenzer, C.; Dirin, M.; Winkler, A. M.; Baumann, V.; Winkler, J. J
5	<i>Control Release</i> 2015 , <i>203</i> , 1-15.
6	(2) Whitehead, K. A.; Langer, R.; Anderson, D. G. Nat Rev Drug Discov
7	2009 , <i>8</i> , 129-138.
8	(3) Deshayes, S.; Morris, M. C.; Divita, G.; Heitz, F. Cell Mol Life Sci 2005,
9	62, 1839-1849.
10	(4) Fonseca, S. B.; Pereira, M. P.; Kelley, S. O. Adv Drug Deliv Rev 2009, 61,
	953-964.
12	(5) Heitz, F.; Morris, M. C.; Divita, G. Brit J Pharmacol 2009 , 157, 195-206.
13	(6) Park, I. G.; Jeong, J. H.; Kim, S. W. Adv Drug Deliv Rev 2006, 58, 467-
14	480. (7) Survey E_{1} : Downly, S_{2} = <i>Dharm</i> , <i>Dec</i> 2004 , 21, 280, 202
15	(7) Silydel, E. L., DOWdy, S. F. Pharm Res 2004, 21, 589-595. (8) Thomas M: Klibanov, A. M. Appl Microbiol Piotochyal 2003, 62, 27, 24
10 17	(8) I nomas, M., Kilbanov, A. M. <i>Appl Microbiol Biolechnol</i> 2003 , 02 , $27-34$.
1 / 1 Q	(9) Frankel, A. D., Pado, C. O. Cell 1988 , <i>33</i> , 1189-1195. (10) Green M: Leewanstein D. M. Cell 1988 , 55, 1170, 1188
10	(10) Oleell, M., Loewellstelli, F. M. Cell 1966 , 55, 1179-1166. (11) Joliot A: Dornalla C: Dagagetinibazin H: Drachiantz A Drac Natl
19 20	(11) Johot, A., Felhene, C., Deagostinioazin, H., Flochaniz, A. Froc. Nat.
20	Acua. Sci. U. S. A. 1991, 00, 1804-1800. (12) Vivas \mathbf{E} : Prodin \mathbf{D} : Labley \mathbf{P} Lowmal of Piological Chamistry 1007
$\frac{21}{22}$	(12) VIVES, E., BIOUIII, F., LEDIEU, B. JOURNAL OF BIOLOGICAL CHEMISTRY 1997, 272–16010–16017
22	(12) Mitchell D. I. Kim D. T. Steinman I. Fathman C. G. Rothbard I. B.
23 24	Iournal of Pantide Research 2000 56 318-325
2 4 25	(14) Wender P A · Mitchell D I · Pattabiraman K · Pelkey F T · Steinman
25	I · Rothbard I B Proceedings of the National Academy of Sciences of the United States
20	of America 2000 97 13003-13008
28	(15) Lindgren M. Langel U Cell-Penetrating Pentides. Methods and
29 29	Protocols 2011 683 3-19
30	(16) Siprashvili, Z.; Reuter, J. A.; Khavari, P. A. <i>Molecular Therapy</i> 2004 , 9.
31	721-728.
32	(17) Opalinska, J. B.; Gewirtz, A. M. <i>Nature Reviews Drug Discovery</i> 2002 , <i>1</i> ,
33	503-514.
34	(18) Patel, L. N.; Zaro, J. L.; Shen, W. C. Pharmaceutical Research 2007, 24,
35	1977-1992.
36	(19) Futaki, S.; Suzuki, T.; Ohashi, W.; Yagami, T.; Tanaka, S.; Ueda, K.;
37	Sugiura, Y. Journal of Biological Chemistry 2001, 276, 5836-5840.
38	(20) Morris, M. C.; Depollier, J.; Mery, J.; Heitz, F.; Divita, G. Nature
39	Biotechnology 2001, 19, 1173-1176.
40	(21) Morris, M. C.; Deshayes, S.; Heitz, F.; Divita, G. <i>Biology of the Cell</i> 2008 ,
41	100, 201-217.
42	(22) Morris, M. C.; Vidal, P.; Chaloin, L.; Heitz, F.; Divita, G. Nucleic Acids
43	<i>Research</i> 1997 , <i>25</i> , 2730-2736.

Biomacromolecules

2		
3	1	(23) Morris, M. C.: Depollier, J.: Mery, J.: Heitz, F.: Divita, G. Nat Biotechnol
4	2	2001 . <i>19</i> . 1173-1176.
5	3	(24) Simeoni, F.; Morris, M. C.; Heitz, F.; Divita, G. Nucleic Acids Res. 2003,
7	4	31, 2717-2724.
8	5	(25) Boisguerin, P.: Deshaves, S.: Gait, M. J.: O'Donovan, L.: Godfrey, C.:
9	6	Betts C A Wood M J Lebleu B Adv Drug Deliv Rev 2015 87 52-67
10	7	(26) Crombez L. Morris M C. Deshaves S. Heitz F. Divita G Curr.
11	8	Pharm Design 2008 14 3656-3665
12 13	9	(27) Huang Y W Lee H J Tolliver L M Aronstam R S <i>Biomed Res Int</i>
14	10	2015 2015 834079
15	11	(28) deRonde B M Tew G N <i>Biopolymers</i> 2015 104 265-280
16	12	(29) Sociastra F : Minter L M: Osborne B A: Tew G N
17	12	<i>Riomacromolecules</i> 2014 <i>15</i> 812-820
18	13	(30) Tezgel A O : Gonzalez-Perez G : Telfer I C : Oshorne B A : Minter
19	14	$I M \cdot T_{ew} G N Mol Thor 2013 21 201-209$
20	15	(31) Vader $P: van der A = I = I:$ Storm $G:$ Schiffelers $P:$ M: Enghersen I F
22	10	(31) Value, 1., Van dei Aa, E. J., Storm, G., Semmeters, R. W., Engbersen, J. T.
23	17	$(32) \text{deR onde } \mathbf{R} \mathbf{M} : \text{Torres } \mathbf{I} \mathbf{A} : \text{ Minter } \mathbf{I} \mathbf{M} : \text{Terres } \mathbf{N}$
24	10	(32) definite, D. M., Tones, J. A., Minter, L. M., Tew, O. N. <i>Biomagromologulas</i> 2015 <i>16</i> 2172 2170
25	19 20	$\begin{array}{c} \text{Diomacromolecules 2013, 10, 51/2-51/7.} \\ (22) \text{Tabujaw I: Fraidal C: Kriag P: Halm M: Kaynay K: Mullan K: \\ \end{array}$
20 27	20	(55) Tabujew, I., Freuer, C., Kneg, D., Heini, M., Koynov, K., Munch, K., Denovo V. Maanomol Panid Commun. 2014, 25, 1101, 1107
28	21	(24) Vim S H: Joong J H: Vim T J: Vim S W: Dull D A Mol Dham
29	22	(34) КIII, S. П., Jeolig, J. П., КIII, I. I., КIII, S. W., Dull, D. A. <i>Mol. Pharm.</i> 2000 6 719 726
30	23	(25) Vim S. II: Loong I. II: On M: Vooleman I. W: Vim S. W: Dull D. A.
31	24	(35) Kim, S. H.; Jeong, J. H.; Ou, M.; Yockman, J. W.; Kim, S. W.; Bull, D. A. P_{iouv} interview 2009, 20, 4420, 4446
32	25	Biomaterials 2008, 29, 4439-4446.
33 34	26	(36) Cooley, C. B.; Irantow, B. M.; Nederberg, F.; Klesewetter, M. K.;
35	27	Hedrick, J. L.; Waymouth, R. M.; Wender, P. A. J. Am. Chem. Soc. 2009, 131, 16401-
36	28	
37	29	(37) Geine, E. I.; Cooley, C. B.; Simon, J. R.; Kiesewetter, M. K.; Edward, J.
38	30	A.; Hickerson, K. P.; Kaspar, R. L.; Hedrick, J. L.; Waymouth, K. M.; Wender, P. A.
39	31	Proc. Natl. Acad. Sci. U. S. A. 2012, 109, 13171-13176.
40 41	32	(38) Wender, P. A.; Huttner, M. A.; Staveness, D.; Vargas, J. R.; Xu, A. F.
41	33	Mol. Pharm. 2015, 12, 742-750.
43	34	(39) Brogden, K. A. <i>Nat. Rev. Microbiol.</i> 2005, <i>3</i> , 238-250.
44	35	(40) Gabriel, G. J.; Som, A.; Madkour, A. E.; Eren, T.; Tew, G. N. Mater Sci
45	36	Eng R Rep 2007, 57, 28-64.
46	37	(41) Kuroda, K.; Caputo, G. A. <i>WIREs Nanomed. Nanobiotechnol.</i> 2013 , <i>5</i> , 49-
47 48	38	66.
40	39	(42) Som, A.; Vemparala, S.; Ivanov, I.; Tew, G. N. <i>Biopolymers</i> 2008 , <i>90</i> , 83-
50	40	93.
51	41	(43) Tew, G. N.; Scott, R. W.; Klein, M. L.; Degrado, W. F. Acc. Chem. Res.
52	42	2010 , <i>43</i> , 30-39.
53 54	43	(44) deRonde, B. M.; Birke, A.; Tew, G. N. <i>Chemistry</i> 2015 , <i>21</i> , 3013-3019.
04 55	44	(45) Gabriel, G. J.; Madkour, A. E.; Dabkowski, J. M.; Nelson, C. F.; Nusslein,
56	45	K.; Tew, G. N. Biomacromolecules 2008, 9, 2980-2983.
57		
58		
59		

1	(46) Hennig, A.; Gabriel, G. J.; Tew, G. N.; Matile, S. J Am Chem Soc 2008,
2	130, 10338-10344.
3	(47) Som, A.; Reuter, A.; Tew, G. N. Angew Chem Int Ed Engl 2012, 51, 980-
4	983.
5	(48) Som, A.; Tezgel, A. O.; Gabriel, G. J.; Tew, G. N. Angew Chem Int Ed
6	Engl 2011, 50, 6147-6150.
7	(49) Tezgel, A. O.; Telfer, J. C.; Tew, G. N. <i>Biomacromolecules</i> 2011, 12, 2079 2092
8	50/8-3083. (50) Futaki S.: Ohashi W.: Suzuki T.: Niwa M.: Tanaka S.: Uada K.:
9 10	(50) Fulaki, S., Ollasili, W., Suzuki, I., Niwa, M., Tallaka, S., Ueda, K., Harashima H.: Sugiura V <i>Biogoniug Cham</i> 2001 12 1005 1011
10	(51) Katayama S: Hirose H: Takayama K: Nakase I: Eutaki S I Control
11	<i>Release</i> 2011 149 29-35
13	(52) Rothbard J B · Jesson T C · Lewis R S · Murray B A · Wender P A
14	J Am Chem Soc 2004 , 126, 9506-9507.
15	(53) Takeuchi, T.; Kosuge, M.; Tadokoro, A.; Sugiura, Y.; Nishi, M.; Kawata,
16	M.; Sakai, N.; Matile, S.; Futaki, S. ACS Chem Biol 2006, 1, 299-303.
17	(54) Nishihara, M.; Perret, F.; Takeuchi, T.; Futaki, S.; Lazar, A. N.; Coleman,
18	A. W.; Sakai, N.; Matile, S. Org Biomol Chem 2005, 3, 1659-1669.
19	(55) Sakai, N.; Futaki, S.; Matile, S. Soft Matter 2006, 2, 636-641.
20	(56) Nelson, C. E.; Kintzing, J. R.; Hanna, A.; Shannon, J. M.; Gupta, M. K.;
21	Duvall, C. L. ACS Nano 2013, 7, 8870-8880.
22	(57) Thaker, H. D.; Cankaya, A.; Scott, R. W.; Tew, G. N. ACS Med Chem Lett
23	2013 , <i>4</i> , 481-485.
24	(58) Thaker, H. D.; Som, A.; Ayaz, F.; Lui, D.; Pan, W.; Scott, R. W.; Anguita,
25	J.; Tew, G. N. J Am Chem Soc 2012, 134, 11088-11091.
26	(59) Thaker, H. D.; Sgolastra, F.; Clements, D.; Scott, R. W.; Tew, G. N. J
27	Med Chem 2011, 54, 2241-2254.
28	(60) Bielawski, C. W.; Benitez, D.; Grubbs, R. H. J Am Chem Soc 2003, 125,
29 20	6424-6423. (61) Diolowski C. W. Crubba D. H. Anamy Cham Int Ed Engl 2000 20 2002
50 21	(01) Dielawski, C. W., Olubos, K. H. Angew Chem Int Ed Engi 2000, 59, 2905-
31	(62) Bielawski C. W.: Grubbs R. H. Prog. Polym. Sci. 2007 32 1-29
32	(62) Dictawski, C. W., Orubbs, R. H. $170g$, $10tym$, Sci. 2007, 52, 1-2). (63) Cannizzo I F: Grubbs R H Macromolecules 21 1961-1967
34	(64) Love I A · Morgan I P · Trnka T M · Grubbs R H Angew Chem Int
35	Ed Engl 2002 41 4035-4037
36	(65) Schwab, P.: France, M. B.: Ziller, J. W.: Grubbs, R. H. Angew Chem Int
37	<i>Ed Engl</i> 1995 , <i>34</i> , 2039-2041.
38	(66) Singh, R.; Czekelius, C.; Schrock, R. R. Macromolecules 2006, 39, 1316-
39	1317.
40	(67) Trnka, T. M.; Grubbs, R. H. Acc Chem Res 2001, 34, 18-29.
41	(68) Lienkamp, K.; Madkour, A. E.; Kumar, K. N.; Nusslein, K.; Tew, G. N.
42	Chem Eur J 2009 , 15, 11715-11722.
43	(69) Lienkamp, K.; Madkour, A. E.; Musante, A.; Nelson, C. F.; Nusslein, K.;
44	Tew, G. N. <i>J Am Chem Soc</i> 2008 , <i>130</i> , 9836-9843.
45	(70) Mitchell, D. J.; Kim, D. T.; Steinman, L.; Fathman, C. G.; Rothbard, J. B.
46	J Pept Res 2000, 56, 318-325.

