Synthesis of Z-Protected Aib- and Phe(2Me)-Containing Pentapeptides and Their Crystal Structures

by Franziska S. Arnhold¹), Anthony Linden, and Heinz Heimgartner*

Institut für Chemie der Universität Zürich, Winterthurerstrasse 190, CH-8057 Zürich (phone: +41446354282; fax: +41446356812; e-mail: heinz.heimgartner@chem.uzh.ch)

A series of pentapeptide derivatives containing a,a-disubstituted a-amino acids have been prepared by a combination of the 'azirine/oxazolone method' and segment condensations. X-Ray crystal-structure determinations of the molecular structures confirmed the presence of helical conformations stabilized by β -turns of type *III* or *III'*. Pentapeptides containing (*R*)-Phe(2Me) form a right-handed helix, whereas those containing (*S*)-Phe(2Me) adopt a left-handed helical structure.

1. Introduction. – The 'azirine/oxazolone method', established as a convenient protocol for the synthesis of Aib-containing peptides [1-3], has been used successfully for the preparation of model peptides [2c][3-6], sterically congested oligopeptides [7][8], and peptaibols [2d][9-11]. It has also been shown that this procedure can be adapted to solid-phase methodology [12]. Another modification allowed the synthesis of endothiopeptides containing Aib units [13-16].

In the present study, a series of Z-protected (Z = (benzyloxy)carbonyl) pentapeptides containing a,a-disubstituted a-amino acids were prepared according to this method. All a,a-disubstituted a-amino acids were introduced *via* coupling of amino or peptide acids with 2,2-disubstituted 2*H*-azirin-3-amines. The synthon for Phe(2Me) was the racemic 2-benzyl-2,*N*-dimethyl-*N*-phenyl-2*H*-azirin-3-amine leading to mixtures of (*R*)- and (*S*)-Phe(2Me)-peptides²). In the case of diastereoisomers, they were separated chromatographically.

Based on the studies of *Ramachandran et al.* [18], it is well-known that the introduction of α,α -disubstituted α -amino acids³) into peptides stabilizes β -turn and 3_{10} -helical conformations [19][20]. For the crystalline state, this has been established by X-ray crystallography (*cf., e.g.,* [21][22]). In spite of this knowledge, current interest in conformations of small peptides is documented by several recent reports (*e.g.,* for pentapeptides, [23][24]). For this reason, we determined the crystal structures of some of the prepared pentapeptides.

2. Results and Discussion. – 2.1. *Synthesis of Pentapeptides.* The syntheses of the Z-protected pentapeptides were carried out *via* combinations of the coupling of amino or

³) The prototype is α -aminoisobutyric acid (Aib).

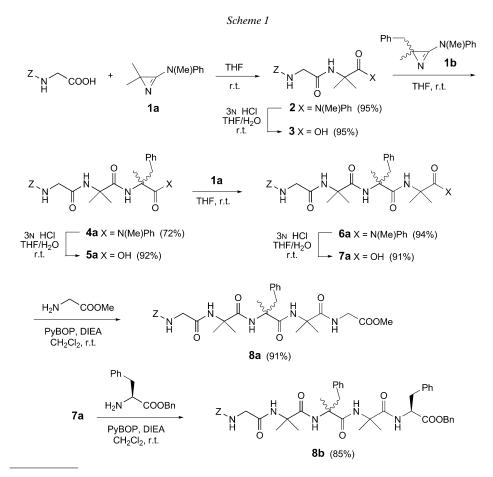
© 2014 Verlag Helvetica Chimica Acta AG, Zürich

¹) In part from the Ph.D. thesis of *F. S. A.*, Universität Zürich, 1997. Present address: Bachem AG, Hauptstrasse 144, CH-4416 Bubendorf

²) Synthons for enantiomerically pure (*R*)- and (*S*)-Phe(2Me) are now also available [4a][17].

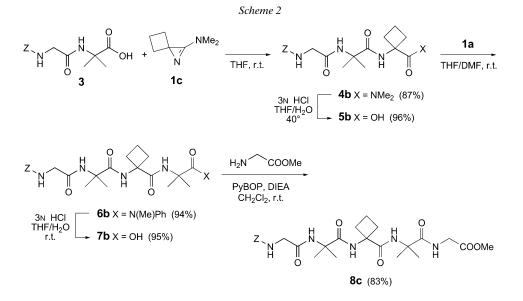
peptide acids with 2*H*-azirin-3-amines, and standard coupling with amino acid esters or peptide segments. Two examples are shown in *Scheme 1*: as described in [11][25], Z-Gly-OH in THF was reacted with 2,2,*N*-trimethyl-*N*-phenyl-2*H*-azirin-3-amine (**1a**; Aib synthon [26]) leading to dipeptide amide **2**, followed by selective hydrolysis to give Z-Gly-Aib-OH (**3**). Reaction of the latter with 2-benzyl-2,*N*-dimethyl-*N*-phenyl-2*H*azirin-3-amine (**1b**) [26b] afforded the tripeptide amide **4a** [25a], which was hydrolyzed (\rightarrow **5a**), coupled with **1a** (\rightarrow **6a**), and again hydrolyzed (\rightarrow **7a**). The coupling of **7a** with H-Gly-OMe and H-Phe-OBn, respectively, was carried out in CH₂Cl₂ using PyBOP/ DIEA⁴), as the coupling reagent, leading to pentapeptide **8a**⁵) and **8b**, respectively. The latter was obtained as a 1:1 mixture of diastereoisomers.

Similar to the preparation of **8a** and **8b**, pentapeptide **8c** containing 1-aminocyclobutanecarboxylic acid (Acb) was synthesized (*Scheme 2*). Coupling of **3** with 2-



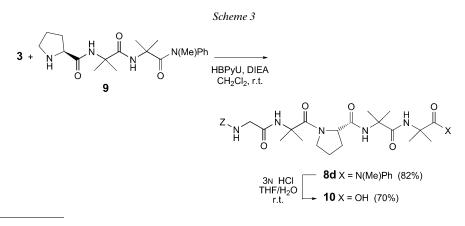
PyBOP = [(Benzotriazol-1-yl)oxy]tripyrrolidinophosphonium hexafluorophosphate; DIEA = ethyl(diisopropyl)amine (EtNⁱPr₂, *Hünig* base).

⁵) The coupling of **7** with H-Gly-OMe was also achieved by treatment with *N*,*N*'-dicyclohexylcarbodiimide (DCC), ZnCl₂, and Et₃N in DMF [27][28] leading to **8a** in 70% yield.



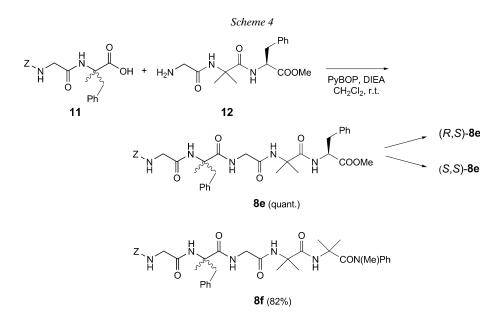
(dimethylamino)-1-azaspiro[2.3]hex-1-ene (1c) [2b] gave the tripeptide amide 4b in 87% yield. The hydrolysis of the terminal amide group was achieved in 3N HCl (THF/H₂O) at 40° leading to 5b (96%). Subsequent azirine coupling with 1a (\rightarrow 6b), selective hydrolysis at room temperature (\rightarrow 7b), and coupling with H-Gly-OMe by treatment with PyBOP/DIEA gave 8c.

The Pro-containing pentapeptide **8d** was obtained by segment coupling of **3** and H-Pro-Aib-Aib-N(Me)Ph (**9**) in CH₂Cl₂ with HBPyU⁶)/DIEA as the coupling reagent [29] (*Scheme 3*). The tripeptide **9** [25b] was prepared by subsequent coupling of Z-Pro-OH with azirine **1a**, selective hydrolysis, again coupling with **1a**, and deprotection of the N-terminus by hydrogenolysis. Selective hydrolysis of **8d** under the usual conditions gave the pentapeptide acid **10** in 70% yield.



⁶) HBPyU = O-[(Benzotriazol-1-yl)oxy]dipyrrolidinocarbenium hexafluorophosphate.

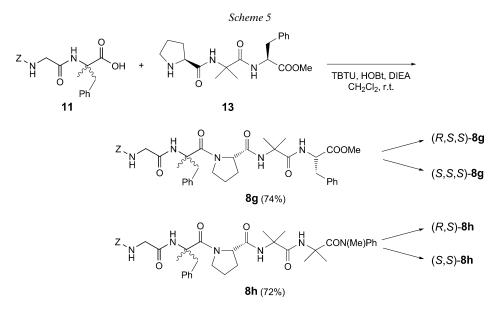
The pentapeptides **8e** and **8f** were prepared *via* coupling of the dipeptide segment **11** [25a] with the tripeptide segment **12** and the corresponding H-Gly-Aib-Aib-N(Me)Ph [25a], respectively (*Scheme 4*). The coupling was performed by treatment with PyBOP/DIEA in CH₂Cl₂ at room temperature and led to **8e** as a 1:1 mixture of diastereoisomers in almost quantitative yield. The corresponding **8f** was isolated in 82% yield as a racemate. The segment **12** was obtained by coupling of **3** with H-Phe-OMe (PyBOP/DIEA; 85%), followed by hydrogenolytic deprotection of the NH₂ group (98%). The diastereoisomers of **8e** were separated chromatographically (SiO₂, AcOEt/MeOH), and the pure isomers were isolated in 47 ((*S*,*S*)-**8e**) and 44% ((*R*,*S*)-**8e**) yield. The configuration of (*S*,*S*)-**8e** was determined by X-ray crystallography (see *Sect. 2.3*).



Finally, the dipeptide segment **11** was coupled with the tripeptides H-Pro-Aib-Phe-OMe (**13**) and the corresponding H-Pro-Aib-Aib-N(Me)Ph [25b], respectively, leading to pentapeptides **8g** and **8h**, respectively (*Scheme 5*). These couplings proved to be rather difficult. In the case of **8g**, the reaction with PyBOP/DIEA in CH₂Cl₂ for 19.5 h gave the product in only 39% yield, whereas with TBTU/HOBt/DIEA⁷), after 19.5 h, **8g** was obtained in 74% yield. Also in the case of **8h**, TBTU/HOBt gave slightly better results than PyBOP/DIEA. The tripeptide **13** was prepared by coupling of Z-Pro-Aib-OH [25b] with H-Phe-OMe by treatment with PyBOP/DIEA.

In both cases, 8g and 8h, mixtures of diastereoisomers were formed, which could be separated by column chromatography (SiO₂; AcOEt/MeOH). The pure epimers

⁷⁾ TBTU = O-(1*H*-Benzotriazol-1-yl)-*N*,*N*,*N'*,*N'*-tetramethyluronium tetrafluoroborate; HOBt = 1-hydroxybenzotriazole.



(R,S,S)-8g and (S,S,S)-8g were obtained in a 1.6:1 ratio⁸). As the starting dipeptide 11 was a racemate, the coupling of (R)-11 with 13 was obviously more efficient than that of (S)-11. A similar result was obtained in the case of 8h: the epimers (R,S)-8h and (S,S)-8h⁸) were obtained in a 2:1 ratio.

2.2. Conformation of the Pro-Peptide Bonds in the Pentapeptides in Solution. Whereas in general the *trans*-conformation of the peptide bond is highly preferred, *cis*and trans-conformations of Xaa-Pro peptide bonds show comparable thermodynamic stability. This was shown first for short peptides such as (S-Bn)-Cys-Pro-Leu-Gly-NH₂ on the basis of their ¹H-NMR spectra [30]. Later, it has been demonstrated that ¹³C-NMR spectroscopy is a convenient method for distinguishing between the *cis*- and trans-conformation of the Xaa-Pro bond [31]. Typically, the chemical shifts of $C(\beta)$ (Pro) and $C(\gamma)$ (Pro) in *cis*-Xaa-Pro peptides are *ca.* 31.3 and 22.5 ppm, respectively, whereas the corresponding values for *trans*-Xaa-Pro peptides are 29.5 and 24.2 ppm [31c]. Obviously, the chemical shift difference, $\Delta \delta$, is a reliable indicator for the conformation, with values of ca. 8.8 for the cis- and 5.3 ppm for the transconformation. For this reason, we analyzed the ¹³C-NMR spectra of the prepared Procontaining pentapeptides 8d, (R,S,S)- and (S,S,S)-8g, (R,S)- and (S,S)-8h, and 10 (*Table 1*). In all cases, only one conformation was observed with $\Delta \delta$ values between 2.3 and 2.7 ppm, indicating the *trans*-conformation. This result is in accordance with the crystal-structure determination (cf. Sect. 2.3), i.e., the Xaa-Pro peptide bond adopts the same conformation in solution and in the crystal.

2.3. Crystal Structures of Pentapeptides. Suitable crystals of the tetrapeptide Z-Gly-Aib-(RS)-Phe(2Me)-Aib-N(Me)Ph (6a) were obtained from CH₂Cl₂/Et₂O/hexane.

⁸) The configurations of (R,S,S)-8g and (R,S)-8h were established by X-ray crystallography (see *Sect. 2.3*).

Table 1. ¹³C-NMR Chemical Shifts [ppm] of $C(\beta)$ - and $C(\gamma)$ -Atoms of the Pro Residue in Pro-Containing Pentapeptides

Peptide	Solvent	$C(\beta)$	$C(\gamma)$
Z-Gly-Aib-Pro-Aib-Aib-N(Me)Ph (8d)	CDCl ₃	28.8	26.2
Z-Gly- (R) -Phe $(2Me)$ -Pro-Aib-Phe-OMe $((R,S,S)$ - 8g)	CDCl ₃	28.5	26.0
Z-Gly- (S) -Phe $(2Me)$ -Pro-Aib-Phe-OMe $((S,S,S)$ - 8g)	CDCl ₃	28.3	26.0
Z-Gly- (R) -Phe $(2Me)$ -Pro-Aib-Aib-N (Me) Ph $((R,S)$ - 8h)	CDCl ₃	28.7	26.3
Z-Gly-(S)-Phe(2Me)-Pro-Aib-Aib-N(Me)Ph ((S , S)-8h)	CDCl ₃	28.6	26.2
Z-Gly-Aib-Pro-Aib-Aib-OH (10)	(D ₆)DMSO	28.2	25.5

The crystals are racemic, and the asymmetric unit contains one peptide molecule plus one half of a disordered molecule of hexane, which sits across a center of inversion. In *Fig. 1*, the peptide molecule with the (*S*)-configured amino acid Phe(2Me) is shown. The molecule forms a β -turn of type *III*' [32] with the Phe(2Me) unit in position (*i*+3) and an intramolecular H-bond N(4)–H…O(11) (H…O 2.13(3) Å, N…O 2.963(3) Å, N–H…O 169(3)°). This interaction has a graph set motif [34] of *S*(10). A second intramolecular H-bond is formed between N(10)–H and N(13). The other two NH groups are involved in intermolecular H-bonds with an adjacent molecule (N(7)– H…O(2'), N(13)–H…O(5')); these intermolecular interactions link the molecules into extended chains which run parallel to the [010] direction and can be described by graph set motifs of *C*(8) and *C*(11), respectively.

All peptide bonds show the *trans*-conformation, but the Phe(2Me)-Aib bond deviates significantly from planarity (ω 155.5(2)°). The values of the torsion angles ω (Gly-Aib), ω (Aib-Phe(2Me)), and ω (Aib-N(Me)Ph) are close to 180° (175.0(2), -176.6(2), and -177.5(2)°, resp.). The torsion angles ϕ and ψ of Aib(2) (61.0(4) and 25.3(4)°) and Phe(2Me) (60.2(3) and 23.2(3)°) correspond well with the typical values

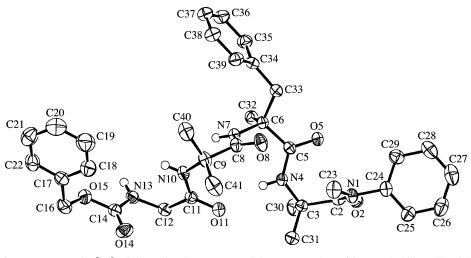


Fig. 1. ORTEP Plot [33] of the molecular structure of the tetrapeptide **6a** (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity, and only the major conformation of the disordered central benzyl ring is shown; solvent molecule not included).

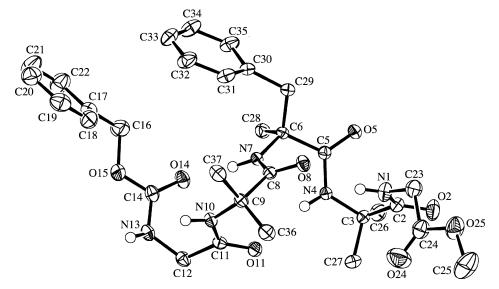
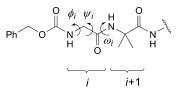


Fig. 2. ORTEP Plot [33] of the molecular structure of the pentapeptide **8a** (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity).

Table 2. Torsion Angles ϕ , ψ , and ω [°] of the Backbone of Compounds 8a, (S,S)-8e, 8f, (R,S,S)-8g, (R,S)-8h, and 10 in the Crystal (amino acid (i) = Gly(1))



	8a	(<i>S</i> , <i>S</i>)-8e	8f	(<i>R</i> , <i>S</i> , <i>S</i>)- 8g	(<i>R</i> , <i>S</i>)-8h	10
$\phi_{(i)}$	53.7(2)	-96.3(3)	-88.7(5)	70.6(5)	64(1); 62(1)	82.2(6)
$\psi_{(i)}$	32.4(2)	-5.6(5)	-33.9(6)	179.0(4)	-170.3(7); -172.3(7)	155.3(5)
$\omega_{(i)}$	-178.7(1)	173.2(3)	178.7(4)	179.4(4)	175.1(7); 177.9(7)	- 169.2(4)
$\phi_{(i+1)}$	55.8(2)	53.9(4)	49.4(6)	-49.6(5)	-51.5(9); -51.0(9)	-53.8(6)
$\psi_{(i+1)}$	21.8(2)	34.3(5)	42.2(6)	-43.4(5)	-36(1); -40.5(9)	-37.8(6)
$\omega_{(i+1)}$	-178.7(1)	179.5(3)	173.1(4)	-173.7(4)	-178.9(7); -177.8(6)	- 176.9(4)
$\phi_{(i+2)}$	51.9(2)	61.3(5)	61.3(6)	-66.2(5)	-55(1); -57(1)	-61.2(6)
$\psi_{(i+2)}$	29.6(2)	13.5(5)	29.4(6)	-13.6(6)	-30(1); -30(1)	-31.7(6)
$\omega_{(i+2)}$	178.6(1)	-170.8(3)	-177.5(4)	171.5(4)	-179.0(7); -178.1(7)	178.7(4)
$\phi_{(i+3)}$	57.0(2)	55.8(4)	59.4(6)	-55.6(5)	-64.8(9); -58.4(9)	-54.1(7)
$\psi_{(i+3)}$	30.4(2)	30.6(5)	32.4(6)	-32.0(5)	-30(1); -32.6(9)	-39.5(5)
$\omega_{(i+3)}$	166.3(1)	163.6(4)	-170.7(4)	-172.5(4)	-162.1(8); -154.6(7)	175.4(4)
$\phi_{(i+4)}$	-83.0(2)	-81.0(5)	-47.4(7)	-111.0(5)	-57(1); -52(1)	49.4(6)
$\psi_{(i+4)}$	162.8(1)	-179.1(4)	- 57.4(7)	179.4(4)	-50(1); -54(1)	45.2(6)

Compound	$N(1)-H\cdots O(8)$		$N(4)-H\cdots O(11)$		$N(7)-H\cdots O(14)$	
	N…O [Å]	N–H···O [°]	N…O [Å]	N–H···O [°]	N…O [Å]	N–H···O [°]
8a	2.928(2)	163(2)	3.052(2)	171(2)	3.013(2)	166(1)
(<i>S</i> , <i>S</i>)-8e	3.007(4)	165(4)	2.872(4)	166(4)	-	
8f	2.976(5)	130	2.977(5)	141	_	
(<i>R</i> , <i>S</i> , <i>S</i>)-8g	2.960(5)	162(4)	3.001(5)	161(4)	_	
(<i>R</i> , <i>S</i>)-8h	2.974(8);	163; 168	3.036(8);	149; 147	_	
	3.055(7)		3.067(8)			
10	3.034(5)	126	3.000(5)	140	_	

Table 3. Intramolecular H-Bonds of Compounds 8a, (S,S)-8e, 8f, (R,S,S)-8g, (R,S)-8h, and 10 (atom numbering refers to Figs. 1-7)

of $+60^{\circ}$ and $+30^{\circ}$ for a left-handed helical structure (β -turn of type III') (*e.g.* [7][35]).

The crystals of the pentapeptide Z-Gly-Aib-(*RS*)-Phe(2Me)-Aib-Gly-OMe (**8a**), obtained from CH₂Cl₂/hexane, are also racemic. The extension of the peptide chain of **6a** by an additional C-terminal Gly-OMe results in the formation of a 3_{10} -helix, *i.e.*, three consecutive β -turns of type III/III'. The molecule containing (*S*)-Phe(2Me) again forms a left-handed helix with β -turns of type III' (*Fig. 2* and *Table 2*). Three intramolecular H-bonds stabilize the 3_{10} -helix: N(1)–H…O(8), N(4)–H…O(11), and N(7)–H…O(14) (*Table 3*). All these interactions form formal ten-membered rings (graph set motif *S*(10)), the last one involving the C=O group of the Z group. The other two NH groups (Gly(1) and Aib(2)) are involved in intermolecular H-bonds with the same O(5')-atom of Phe(2Me) of an adjacent molecule. This O(5')-atom, therefore, accepts two H-bonds, although one of these is quite weak. The intermolecular H-bonds link the molecules into extended chains which run parallel to the [101] direction and can be described by graph set motifs of *C*(11) and *C*(8) for the interactions involving the donors Gly(1) and Aib(2), respectively.

Racemic crystals of Z-Gly-(RS)-Phe(2Me)-Gly-Aib-Aib-N(Me)Ph (8f) were grown from AcOEt/hexane. In Fig. 3, the molecule containing (S)-Phe(2Me) is depicted. The asymmetric unit contains one molecule of 8f and approximately one third of a highly disordered hexane molecule, which sits across a center of inversion. Four of the five NH groups act as donors for H-bonds. Two of them, N(1)-H and N(4)-H, form intramolecular H-bonds with amide O-atoms that are seven atoms along the peptide backbone to give a graph set motif of S(10) in each case (*Table 3*). These interactions form two β -turns of type III' (Table 2), leading to a left-handed helical structure for the molecule containing (S)-Phe(2Me). Surprisingly, Gly(3) does not continue this helical structure, and the Z group is turned away from the helix. Rather, N(7)-H of Gly(3) forms an intermolecular H-bond with the terminal amide O(24) of a neighboring molecule. The N(13)-H group of Gly(1) forms an intermolecular H-bond to O(2) of the same neighboring molecule. These intermolecular interactions link the peptide molecules into extended chains, which run parallel to the [010] direction (graph set motifs C(11) and C(14), resp.). Interestingly, N(10)–H is not involved in any Hbonding. Finally, N(1)-H, N(4)-H, N(7)-H, and N(10)-H exhibit the usual weak 'sideways' contacts with one of their neighboring N-atoms. All peptide bonds are in the

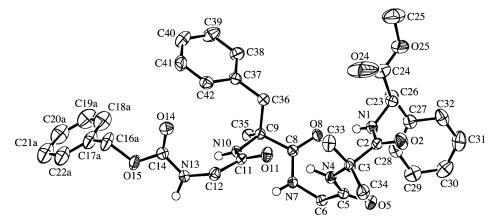


Fig. 3. ORTEP Plot [33] of the molecular structure of the pentapeptide **8f** (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity; solvent molecule not included).

trans-conformation and differ only slightly from planarity, but the terminal amide group deviates significantly from planarity (torsion angle $\omega - 153.1(8)^\circ$; *Table 2*).

The crystals of Z-Gly-(S)-Phe(2Me)-Gly-Aib-Phe-OMe ((S,S)-**8e**) are enantiomerically pure; however, the absolute configuration of the molecule has not been determined. The enantiomer used in the refinement was based on the (S)-configuration at C(23), resulting from (S)-configured H-Phe-OMe used in the synthesis. Based on this, the configuration at C(9) of Phe(2Me) is also (S) (*Fig. 4*). The Ph ring of the

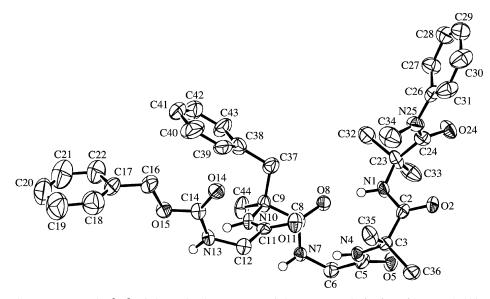


Fig. 4. ORTEP Plot [33] of the molecular structure of the pentapeptide (*S*,*S*)-**8e** (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity, and only the major conformation of the disordered terminal Bn group is shown; H₂O molecules not included).

terminal Bn group is disordered over two orientations, with a ratio of the conformers of 0.33:0.67. The crystal lattice also contains one disordered H₂O molecule per peptide molecule. All five peptide NH groups act as donors for H-bonding. Two of the H-bonds are intramolecular interactions with, in each case, the amide O-atom two peptide units further along the peptide chain, forming two consecutive β turns of type III' (graph set motif: S(10); Table 3). As a result of the presence of (S)-Phe(2Me), a left-handed helical structure is formed (see, e.g., [36]). The presence of (S)-Phe, an amino acid which usually forms right-handed helices, probably has no influence because of its position at the C-terminus of the peptide. As in 8f, the helix is not continued by Gly(3), and the Z group is turned away. The NH group of Gly(3) forms an intermolecular Hbond with O(2') of Aib(4) of an adjacent molecule. A second intermolecular H-bond, $N(13)-H\cdots O(5')$, is formed with the same neighboring molecule. These two interactions link the peptide molecules into extended chains, which run parallel to the [100] direction (graph set motifs C(8) and C(11), resp.). The fifth interaction, from N(10)-H of Phe(2Me), involves a H₂O molecule acting as an acceptor. Furthermore, analogous to 8f, N(1)-H, N(4)-H, N(7)-H, and N(10)-H are close enough to adjacent N-atoms in the peptide chain to form weak interactions.

The asymmetric unit of Z-Gly-(R)-Phe(2Me)-Pro-Aib-Phe-OMe ((R,S,S)-8g) contains one peptide molecule, one molecule of AcOEt, and one site for a H_2O molecule, which is partially occupied (50%). The crystals are enantiomerically pure; however, the absolute configuration of the molecule has not been determined. The enantiomer used in the refinement was based on the known (S)-configuration at C(6) and C(23) of the Pro and Phe residues, respectively. Based on this assumption, C(9) of Phe(2Me) has the (R)-configuration. The molecule forms a right-handed β_{10} -helix, as expected for a peptide containing two protein amino acids and (R)-Phe(2Me) (Fig. 1 and Table 2). Again, the Z group is turned away from the helix, in this case because of the presence of Pro in position 3 of the peptide. The N(1)-H and N(4)-H groups of Phe(5) and Aib(4), respectively, form intramolecular H-bonds with amide O-atoms that are seven atoms along the peptide backbone (graph set motif S(10) for each interaction, Table 3). The NH groups of Phe(2Me)(2) and Gly(1), N(10)-H and N(13)–H, respectively, form intermolecular H-bonds with the amide O-atoms, O(5')and O(2''), respectively, of different neighboring molecules. The N(10)-H interaction links the molecules into extended chains, which run parallel to the [100] direction (graph set motif C(8)), while the N(13)–H interaction links the molecules into a second type of extended chains, which also run parallel to the [100] direction (graph set motif C(14)). The H₂O molecule acts as a donor for two H-bonds to amide O-atoms of two different peptide molecules (graph set motif D for each interaction). As a result, the amide O-atoms O(2) of Aib(4) and O(11) of Gly(1) are each acceptors of two H-bonds. Neither the H₂O molecule nor the AcOEt molecule acts as an acceptor for H-bonds. The combination of all intermolecular interactions links the molecules into extended sheets, which lie parallel to the (010) plane. Finally, N(1)-H and N(4)-H exhibit the usual weak 'sideways' contacts with one of their neighboring N-atoms.

The crystals of Z-Gly-(R)-Phe(2Me)-Pro-Aib-Aib-N(Me)Ph ((R,S)-8h) are enantiomerically pure; however, the absolute configuration of the molecule has not been determined. The enantiomer used in the refinement was based on the known (S)configuration of the Pro residue (C(6)). The configuration of Phe(2Me), at (C(9)), is

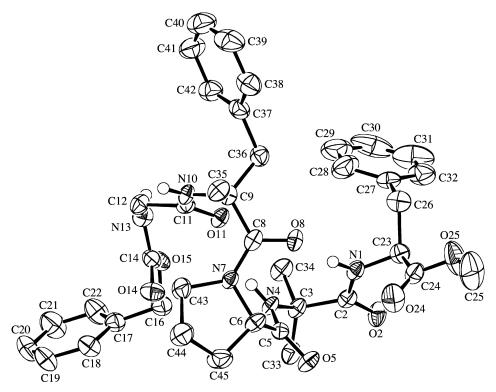


Fig. 5. ORTEP Plot [33] of the molecular structure of the pentapeptide (R,S,S)-8g (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity, solvent molecules not included).

therefore (R) (Fig. 6). The asymmetric unit contains two independent peptide molecules, one hexane molecule, and a H₂O molecule, which lies on a twofold axis. The two peptide molecules have generally similar conformations with the major difference being in the orientation of the Ph rings at each end of the peptide chain. One of the peptide molecules (molecule B) has disorder in the Ph ring of the protecting group Z. In molecule A, the NH groups of the two Aib units, N(1)-H and N(4)-H, form intramolecular H-bonds with O(8) of Phe(2Me) and O(11) of Gly, respectively, resulting in two β -turns of type III (graph set motifs of S(10); Table 3). The same interactions occur in molecule B, involving N(51)-H and N(54)-H. These are the normal intramolecular interactions found for peptides of this type and help to form a right-handed 3_{10} -helical structure (*Table 2*). These NH groups also have weak 'sideways' interactions with the nearest neighboring N-atom in the chain. The other NH groups and the H_2O molecule act as donors for intermolecular H-bonds: N(10)-H in molecule A forms an H-bond with the amide O-atom, O(74), at the opposite end of molecule B. The corresponding group, N(60)-H, in molecule B, however, interacts with a different O-atom in another molecule A, namely O(2). These two interactions form spiralling $\cdots A \cdots B \cdots A \cdots B \cdots$ chains of molecules which run parallel to the [001]

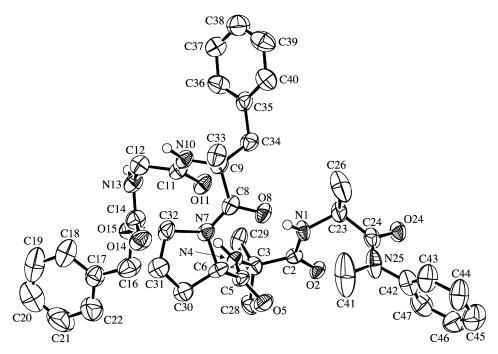


Fig. 6. ORTEP Plot [33] of the molecular structure of one of the two symmetry-independent molecules of the pentapeptide (R,S)-**8h** (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity; solvent molecules not included).

direction and have a binary graph set motif of $C_2^2(25)$. Furthermore, N(13)–H of molecule A forms an intermolecular H-bond with the amide O-atom of Pro(3), O(55), of molecule B and the same interaction, involving N(63)–H, occurs to link molecule B to atom O(5) of a different molecule A, thus forming extended $\cdots A \cdots B \cdots A \cdots B \cdots$ chains of molecules, which run parallel to the [010] direction (graph set motif $C_2^2(22)$). The H₂O molecule sits on a two-fold axis and, therefore, acts as a donor for two equivalent H-bonds to the amide O-atom O(52) of Aib(4) in two different B molecules (graph set motif D). The combination of the intermolecular interactions links the molecules into three-dimensional supramolecular frameworks.

The crystals of Z-Gly-Aib-Pro-Aib-Aib-OH (10) are enantiomerically pure. However, the absolute configuration has not been determined. The enantiomer used in the refinement was based on the known (S)-configuration of Pro (C(6)) (Fig. 7). The asymmetric unit contains one peptide and one disordered MeOH molecule. Two intramolecular H-bonds, N(1)–H···O(8) and N(4)–H···O(11), form two β -turns of type *III* (graph set motif S(10); *Table 3*), *i.e.*, a right-handed 3₁₀-helix, as expected for a peptide containing (S)-Pro as the only chiral amino acid. The other two NH groups, N(10)–H and N(13)–H, as well as the carboxy OH group, O(25)–H, form intermolecular H-bonds with C=O O-atoms in three different neighboring molecules. The O(25)–H···O(5') interaction links the molecules into extended chains which run parallel to the [100] direction and have a graph set motif of C(10). The

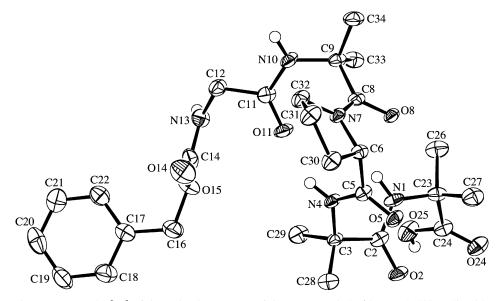


Fig. 7. ORTEP Plot [33] of the molecular structure of the pentapeptide **10** (50% probability ellipsoids, arbitrary numbering of atoms, H-atoms bonded to C-atoms omitted for clarity; solvent molecule not included).

 $N(10)-H\cdots O(2'')$ interaction links the molecules into extended chains which run parallel to the [010] direction (graph set motif C(11)), while the $N(13)-H\cdots O(8''')$ interaction links the molecules into extended chains which also run parallel to the [100] direction (graph set motif C(8)). The combination of these intermolecular interactions links the molecules into a two-dimensional network, which lies parallel to the (001) plane. The O(8)-atom acts as an acceptor for two H-bonds, an intra- and an intermolecular one. Both positions for the disordered MeOH molecule indicate the presence of a H-bond to the C=O O(14)-atom.

3. Conclusions. – The presented results confirm once more the usefulness of the 'azirine/oxazolone method', combined with traditional segment coupling, in the synthesis of peptides containing α, α -disubstituted α -amino acids. All of the Aib and Phe(2Me) residues, as well as the 1-aminocyclobutane carboxylic acid, have been added to *N*-protected amino acids or peptides *via* the reaction with the corresponding 2,2-disubstituted 2*H*-azirin-3-amine **1**. As reported previously [7][8][25][37], the coupling of two α, α -disubstituted α -amino acids does not suffer from steric hindrance. This is of importance, because coupling reactions with N-terminal Aib residues are known to be difficult (see, *e.g.*, [38]). As the synthon for Phe(2Me), **1b** was used as a racemate²), racemic pentapeptides such as **8a** or mixtures of epimers such as **8b** were formed. The pure epimers were obtained after chromatographic separation in the cases of **8e**, **8g**, and **8h**.

It is also important to note that the segment coupling with dipeptides with a Cterminal Aib or Phe(2Me) residue could be performed in good yields, even in the case of tripeptides **9** and **13** with a N-terminal Pro. Surprisingly, the coupling with TBTU/ HOBt yielded the pentapeptides **8g** and **8h** in significantly higher yields than with PyBOP, which was the preferred coupling reagent in the cases without Pro.

The X-ray crystal-structure determinations revealed that all of the prepared pentapeptides adopt 3_{10} -helical conformations, as expected for Aib-containing peptides [39]. As Aib is achiral, the handedness of the helix is determined by the chiral amino acid present. It has been shown that a right-handed helix is preferred, when the poly-Aib peptide contains a N-terminal L-amino acid or a L-amino acid within the peptide chain [40]. On the other hand, a C-terminal L-amino acid leads to the inverse relationship, *i.e.*, a left-handed helix. Furthermore, it was found that the chiral α,α -disubstituted α -amino acid Phe(2Me) has the opposite effect, *i.e.*, an internal L-Phe(2Me) residue induces a left-handed helix [36a][41]. Recently, a similar reversal of the screw-sense induction was described for the amino acids Val and Val(2Me): the hexapeptides Ac-L-Xaa-(Aib)₄-Gly-NH₂ and pentapeptides Ac-L-Xaa-(Aib)₄-OR form left-handed helices with Xaa = Val, but right-handed helices with Xaa = Val(2Me) [42]. The study of the influence of the type and the position of L- and D-amino acids on the helical screw sense of Aib-containing peptides is a topic of current interest (see, *e.g.*, [43]).

The crystal structures of the prepared pentapeptides are in full accordance with the results reported in [36][41]. Whereas the peptide **10** with (S)-Pro as the only chiral amino acid adopts a right-handed \mathcal{J}_{10} -helical structure, as is 'normal' for protein amino acids, the enantiomers of the racemic peptides **8a** and **8f**, containing (S)-Phe(2Me) in position 3 and 2, respectively, exist in a left-handed \mathcal{J}_{10} -helical conformation in the crystal. Furthermore, in the crystal of the racemic tetrapeptide **6a**, the enantiomer with (S)-Phe(2Me) forms a β -turn of type *III*, *i.e.*, a turn with a lefthanded helical structure. On the other hand, both (*R*,*S*,*S*)-**8g** and (*R*,*S*)-**8h** form righthanded \mathcal{J}_{10} -helices, supported by the presence of (*R*)-Phe(2Me) as well as (S)-Pro in position 3.

It is worth mentioning that in **8a** three intramolecular $(1 \leftarrow 4)$ H-bonds are formed, stabilizing the 3_{10} -helix (*Table 3*). One of them is the interaction of N(7)–H of Phe(2Me) with C=O(14) of the Z protecting group. Surprisingly, only two intramolecular $(1 \leftarrow 4)$ H-bonds are present in the analogous pentapeptide **8f**, and the Z group is not involved in an intramolecular H-bond. The analogous H-bond pattern with only two $(1 \leftarrow 4)$ interactions, is observed in the enantiomerically pure (S,S)-**8e**. In both cases, N(7)–H of Gly(3) does not form intramolecular H-bonds. In the two remaining pentapeptides, (R,S,S)-**8g** and (R,S)-**8h**, which possess a Pro(3) unit, only two intramolecular $(1 \leftarrow 4)$ H-bonds, N(1)–H…O(8) and N(4)–H…O(11), are possible.

We thank the analytical sections of our institute for spectra and analyses, and the *Stipendienfonds der* Basler Chemischen Industrie and F. Hoffmann-La Roche AG, Basel, for financial support.

Experimental Part

1. *Abbreviations*. Acb, 1-Aminocyclobutane carboxylic acid; Aib, 2-aminoisobutyric acid (2-methylalanin); CSA, camphor-10-sulfonic acid; DCC, *N*,*N*'-dicyclohexylcarbodiimide; DEPC, diethyl cyanophosphonate; DIEA, ethyl(diisopropyl)amine (EtNⁱPr₂; *Hünig* base); DPPA, diphenyl phosphor-

azidate; HBPyU, O-[(benzotriazol-1-yl)oxy]dipyrrolidinocarbenium hexafluorophosphate; HOBt, 1-hydroxybenzotriazole; PyBOP = [(benzotriazol-1-yl)oxy]tripyrrolidinophosphonium hexafluorophosphate; TBTU, O-(1*H*-benzotriazol-1-yl)-*N*,*N*,*N*',*N*'-tetramethyluronium tetrafluoroborate; TFA, tri-fluoroacetic acid; Z, (benzyloxy)carbonyl.

2. General. See [11][25]. Amino acids were purchased from Novabiochem and Bachem, and all of them were L-configured; other reagents and solvents were from Aldrich, Bachem, Fluka, and Merck, resp. Solvents were purified by standard procedures. TLC: Merck TLC glass plates, silica gel 60 F_{254} . Prep. TLC: Merck glass plates, silica gel 60 F_{254} . Column chromatography (CC): Uetikon-Chemie, silica gel C-560 (0.04–0.063 mm). M.p.: Mettler-FP-5 apparatus; uncorrected. [a]_D Values: PerkinElmer-241 polarimeter at 21°. IR Spectra: PerkinElmer-781 spectrometer, in KBr. ¹H- and ¹³C-NMR spectra: Bruker AC-300 or Bruker ARX-300 spectrometer at 300 (¹H) and 75.5 MHz (¹³C), in CDCl₃, CD₃OD, or (D₆)DMSO; multiplicities of ¹³C signals determined by the DEPT technique. ESI-MS: Finnigan TSQ-700 instrument; and CI-MS (with NH₃ or isobutane): Finnigan MAT-90 or SSQ-700 instrument; m/z (rel. %).

General Procedure 1 (GP 1; Coupling with 2H-Azirin-3-amines). To a soln. of an N-protected amino acid or N-protected peptide (1 mmol) in abs. THF (5 ml) at 0°, 1.1 equiv. of the corresponding 2H-azirin-3-amine in THF were added, and the mixture was stirred at r.t. for several h (under Ar). After completion of the reaction, the solvent was evaporated, and the product was purified by crystallization from CH_2Cl_2 /hexane or AcOEt/hexane. In the case of incomplete reactions, the residue was dissolved in CH_2Cl_2 , the remaining acid was extracted with 1N aq. NaOH, the solvent was evaporated, and the residue was crystallized.

General Procedure 2 (GP 2; Selective Hydrolysis of Peptide N-Methyl-N-phenylamides). A soln. of the peptide amide (1 mmol) in 3N HCl (THF/H₂O 1:1; 10 ml) was stirred at r.t. for 1-60 h. Then, aq. 2N HCl was added, and the mixture was extracted with Et₂O. The org. layers were combined and dried (Na₂SO₄), and the solvent was evaporated.

General Procedure 3 (GP 3; Hydrogenolytic Deprotection). The Z-protected peptide (1 mmol) was dissolved in MeOH, and 10 weight-% Pd/C (10%) was added to the soln. The mixture was stirred at r.t. under H_2 overnight. Then, the soln. was filtered through a pad of *Celite*, and the solvent was evaporated. The product was dried in high vacuum (*i.v.*).

General Procedure 4 (GP 4; Deprotection via Catalytic Hydrogen Transfer [44]). To a soln. of Zprotected peptide (1 mmol) in MeOH were added 10 weight-% Pd/C (10%) and HCO₂NH₄ (5 mmol). The mixture was heated at reflux (10 min) and filtered through a pad of *Celite*, and washed with MeOH, and the solvent was evaporated.

General Procedure 5 (GP 5; Coupling of Amino Acids). To a soln. of a Z-protected peptide (1 mmol), an amino acid ester (1.1 mmol), and a coupling reagent (1 mmol) in CH_2Cl_2 (1 ml) was added dropwise DIEA (2 mmol; 3 mmol by using the hydrochloride of the amino acid ester) at r.t. The soln. was stirred for 1 h, and then the solvent was evaporated. The residue was dissolved in AcOEt (20 ml), washed with 5% aq. KHSO₄ (3 ×), 5% aq. NaHCO₃ (3 ×), and sat. aq. NaCl soln., and purified by CC.

3. Synthesis of the Pentapeptides **8**. 3.1. Z-Gly-Aib-(RS)-Phe(2Me)-Aib-Gly-OMe (**8a**). 3.1.1. Z-Gly-Aib-(RS)-Phe(2Me)-N(Me)Ph (**4a**). According to GP 1, to a soln. of Z-Gly-Aib-OH (**3** [25]; 4.49 g, 15.3 mmol) in abs. THF (70 ml) at 0° was added 2-benzyl-2,N-dimethyl-N-phenyl-2H-azirin-3-amine (**1b** [26b]; 15.1 g of a *ca*. 3:4 mixture of **1b** and 2-benzyl-N-methyl-N-phenylpropanamide (*ca*. 25.9 mmol **1b**)); stirring at r.t. for 70 h. The solvent was evaporated, and the product was purified by CC (CH₂Cl₂/MeOH 97:3): 5.95 g (72%) of **4a**. Colorless foam. M.p. 72.2–73.1°. IR (KBr): 3320m, 3060m, 3030m, 2980m, 2940m, 1640s, 1590s, 1515s, 1495s, 1450s, 1380m, 1365m, 1260m, 1240m, 1190m, 1155m, 1105m, 1080m, 1050m, 770m, 735m, 700s. ¹H-NMR (CDCl₃): 7.46 (*s*, NH); 7.40–7.20 (*m*, 11 arom. H); 7.15–7.05 (*m*, 4 arom. H); 6.85 (*s*, NH); 5.62 (br. *s*, NH); 5.11 (*s*, PhCH₂O); 3.77 (*d*, J = 5.4, CH₂(Gly)); 3.48, 3.06 (*AB*, $J_{AB} = 14.2$, PhCH₂); 3.31 (*s*, MeN); 1.42 (*s*, Me₂C); 1.33 (*s*, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 172.6, 172.5, 168.1 (3s, 3 CO(amide)); 156.5 (*s*, CO(urethane)); 144.1, 136.4, 136.3 (3s, 3 arom. C); 129.8, 129.5, 128.5, 128.2, 128.1, 128.0, 127.0 (7d, 15 arom. CH); 67.0 (*t*, PhCH₂O); 63.2, 57.5 (2*s*, C(2)(Aib), C(2)(Phe(2Me))); 44.9 (*t*, CH₂(Gly)); 42.3 (*t*, PhCH₂); 41.8 (*q*, MeN); 24.9, 24.6, 23.5 (3*q*, Me₂C, Me(Ph(2Me))). CI-MS: 545 (7, [*M*+1]⁺), 439 (13), 438 (53, [*M*-Ph(Me)N]⁺), 437 (100, [*M*-

 $PhCH_2O]^+$), 412 (8), 411 (15). Anal. calc. for $C_{31}H_{36}N_4O_5$ (544.65): C 68.36, H 6.66, N 10.29; found: C 67.96, H 6.90, N 10.35.

3.1.2. *Z*-*Gly*-*Aib*-(RS)-*Phe*(2*Me*)-*OH* (**5a**). According to *GP* 2, a soln. of **4a** (5.86 g, 10.8 mmol) in 3N HCl was stirred at r.t. for 7 h. The solvent was evaporated, and the precipitate was filtered: 4.50 g (92%) of **5a**. Colorless crystals. M.p. 192.6–193.8°. IR (KBr): 3310*m*, 3040*m*, 2980*m*, 2970*m*, 1720*s*, 1660*s*, 1650*s*, 1560*m*, 1525*s*, 1460*m*, 1450*m*, 1440*m*, 1285*m*, 1275*m*, 1245*s*, 1195*m*, 1165*m*, 1130*m*, 700*m*. ¹H-NMR ((D₆)DMSO): 12.6 (br. *s*, COOH); 7.96 (*s*, NH); 7.49 (*t*, *J* = 5.8, NH(Gly)); 7.35–7.15 (*m*, 8 arom. H, NH); 7.10–7.05 (*m*, 2 arom. H); 5.00 (*s*, PhCH₂O); 3.58 (*d*, *J* = 6.0, CH₂(Gly)); 3.17 (*s*, PhCH₂); 1.39, 1.35, 1.30 (3*s*, Me₂C, Me(Phe(2Me))). ¹³C-NMR ((D₆)DMSO): 175.2, 173.5, 169.0 (3*s*, 2 CO(amide), COOH); 156.8 (*s*, CO(urethane)); 137.4, 137.0 (2*s*, 2 arom. C); 130.7, 128.7, 128.1, 128.0, 126.8 (5*d*, 10 arom. CH); 65.8 (*t*, PhCH₂O); 59.5, 56.6 (2*s*, C(2)(Aib), C(2)(Phe(2Me)))); 44.2 (*t*, CH₂(Gly)); 41.0 (*t*, PhCH₂); 25.4, 24.8, 22.7 (3*q*, *Me*₂C, Me(Phe(2Me))). CI-MS: 457 (6), 456 (24, [*M* + 1]⁺), 439 (28), 438 (100, [*M* – H₂O + 1]⁺), 348 (12), 330 (8), 312 (12), 295 (47), 277 (15, [*M* – Phe(2Me)]⁺, 251 (18), 180 (30), 134 (26). Anal. calc. for C₂₄H₂₉N₃O₆ (455.51): C 63.28, H 6.42, N 9.22; found: C 62.99, H 6.31, N 9.05.

3.1.3. *Z*-*Gly*-*Aib*-(**RS**)-*Phe*(2*Me*)-*Aib*-*N*(*Me*)*Ph* (**6a**). According to *GP* 1, **5a** (931 mg, 2.04 mmol) was reacted with 2,2,N-*trimethyl*-N-*phenyl*-2H-*azirin-3-amine* (**1a** [26]; 395 mg, 2.27 mmol) in THF/DMF (20:1; 21 ml) for *ca*. 70 h: 1.21 g (94%) of **6a**. Colorless crystals. M.p. $91.3-93.0^{\circ}$. IR (KBr): 3420*m*, 3320*m*, 3280*m*, 1720*s*, 1680*s*, 1650*s*, 1635*s*, 1590*m*, 1530*s*, 1495*s*, 1455*m*, 1395*m*, 1380*m*, 1365*m*, 1280*m*, 1240*s*, 1215*m*, 1195*m*, 1170*m*, 1100*m*, 1090*m*, 1050*m*, 740*m*, 705*s*. ¹H-NMR (CDCl₃): 7.66 (*s*, NH); 7.35 – 7.15 (*m*, 13 arom. H); 7.10–7.05 (*m*, 2 arom. H); 6.87, 6.56, 6.18 (3*s*, 3 NH); 5.03, 4.98 (*AB*, J_{AB} = 12.2, PhCH₂O); 3.65–3.55 (*m*, CH₂(Gly), 1 H of PhCH₂); 3.35 (*s*, MeN); 3.09 (*d*, J_{AB} = 13.7, 1 H of PhCH₂); 1.46 (*s*, Me₂C); 1.42, 1.38, 1.27 (3*s*, Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 174.0, 173.5, 172.9, 170.3 (4*s*, 4 CO(amide)); 157.1 (*s*, CO(urethane)); 145.6, 136.8, 136.2 (3*s*, 3 arom. C); 130.9, 129.2, 128.5, 128.2, 128.0, 127.9, 127.5, 127.0, 126.7 (9*d*, 15 arom. CH); 67.0 (*t*, PhCH₂O); 60.2, 57.6, 57.2 (3*s*, 2 C(2)(Aib), C(2)(Phe(2Me))); 45.3 (*t*, CH₂(Gly)); 40.5 (*q*, MeN); 40.3 (*t*, PhCH₂); 25.8, 25.5, 24.3 (3*q*, 2 *Me*₂C, Me(Phe(2Me))). ESI-MS: 668 (29, [*M*+K]⁺, 652 (100, [*M*+Na]⁺, 523 (47, [*M* – N(Me)Ph]⁺). Anal. calc. for C₃₅H₄₃N₅O₆ (629.76): C 66.75, H 6.88, N 11.12; found: C 66.55, H 6.67, N 11.00.

Suitable crystals for the X-ray crystal-structure determination were grown from $CH_2Cl_2/Et_2O/$ hexane by slow evaporation of the solvent.

3.1.4. Z-Gly-Aib-(RS)-Phe(2Me)-Aib-OH (**7a**). According to GP2, **6a** (4.12 g, 6.55 mmol) was hydrolyzed for 3.5 h: 3.21 g (91%) of **7a**. Colorless solid. M.p. 193.5 – 195.7°. IR (KBr): 3400*m*, 3300*m*, 2980*m*, 1760*s*, 1680*s*, 1650*s*, 1530*s*, 1500*m*, 1470*m*, 1455*m*, 1390*m*, 1315*m*, 1280*m*, 1240*m*, 1160*m*, 700*m*. ¹H-NMR ((D₆)DMSO): 8.27 (*s*, NH); 7.51 (*t*, *J* = 5.5, NH(Gly)); 7.47 (*s*, NH); 7.40 – 7.20 (*m*, 8 arom. H); 7.15 – 7.10 (*m*, 2 arom. H); 6.97 (*s*, NH); 5.01, 4.98 (*AB*, J_{AB} = 13.4, PhCH₂O); 3.70 – 3.50 (*ABX*, J_{AB} = 16.3, CH₂(Gly)); 3.40, 2.99 (*AB*, J_{AB} = 13.4, PhCH₂); 1.36, 1.35, 1.33, 1.30, 1.22 (*5s*, 2 Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 175.5, 172.82, 172.78, 169.9 (4*s*, 3 CO(amide), COOH); 156.6 (*s*, CO(urethane)); 137.0, 136.7 (2*s*, 2 arom. C); 130.7, 128.2, 127.73, 127.68, 127.5, 126.1 (6*d*, 10 arom. CH); 65.6 (*t*, PhCH₂O); 58.7, 56.1, 54.7 (3*s*, 2 C(2)(Aib), C(2)(Phe(2Me))); 43.6 (*t*, CH₂(Gly)); 39.2 (*t*, PhCH₂); 25.6, 24.9, 24.3, 24.2, 23.2 (*5q*, 2 *Me*₂C, Me(Phe(2Me))). ESI-MS: 563 (100, [*M*+Na]⁺). Anal. calc. for C₂₈H₃₆N₄O₇ (540.62): C 62.21, H 6.71, N 10.36; found: C 62.15, H 6.70, N 10.34.

3.1.5. *Z*-*Gly*-*Aib*-(RS)-*Phe*(2*Me*)-*Aib*-*Gly*-*OMe* (**8**a). According to *GP* 5, **7a** (2.46 g, 4.54 mmol) was coupled with H-Gly-OMe ·HCl (639 mg, 5.09 mmol) by treatment with PyBOP (2.22 g, 4.63 mmol) and DIEA (1.81 g, 13.97 mmol) in CH₂Cl₂ (7 ml); 1 h. CC (AcOEt) gave 2.54 g (91%) of **8a**. Colorless foam. M.p. 160.3 – 161.3°. IR (KBr): 3340*m*, 3320*s*, 3000*s*, 2960*m*, 1755*s*, 1705*s*, 1675*s*, 1650*s*, 1530*s*, 1460*m*, 1385*m*, 1365*m*, 1305*m*, 1290*m*, 1260*m*, 1240*m*, 1215*m*, 1195*m*, 1180*m*, 1160*m*, 975*m*, 740*m*, 710*m*, 700*m*. ¹H-NMR (CDCl₃): 7.81 (*t*, NH(Gly)); 7.41 (*s*, NH); 7.35 – 7.20 (*m*, 8 arom. H); 7.10 – 7.05 (*m*, 2 arom. H); 6.90, 6.82 (2*s*, 2 NH); 6.08 (*t*-like, NH(Gly)); 4.97 (*s*, PhCH₂O); 3.95 (*d*, *J* = 5.9, CH₂(Gly)); 3.71 – 3.70 (*m*, CH₂(Gly)); 3.56 (*s*, MeO); 3.23, 3.03 (*AB*, *J_{AB}* = 13.7, PhCH₂); 1.49, 1.46, 1.39, 1.37 (4*s*, 2 Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 176.2, 174.8, 173.8, 171.5, 170.5 (5*s*, 4 CO(amide), COOMe); 157.3 (*s*, CO(urethane)); 136.1, 136.0 (2*s*, 2 arom. C); 130.8, 128.6, 128.3, 128.1, 127.9, 127.0 (6*d*, 10 arom. CH); 67.1 (*t*, PhCH₂O); 59.9, 57.3, 57.0 (3*s*, 2 C(2)(Aib), C(2)(Phe(2Me))); 52.2 (*s*, MeO); 45.4, 41.4, 41.1 (3*t*,

2 CH₂(Gly), PhCH₂); 26.2, 25.4, 24.5, 23.4 (4q, 2 Me_2 C, Me(Phe(2Me)))). ESI-MS: 650 (5, $[M + K]^+$), 634 (100, $[M + Na]^+$], 612 (3, $[M + 1]^+$). Anal. calc. for C₃₁H₄₁N₅O₈ (611.69): C 60.87, H 6.76, N 11.45; found: C 60.78, H 6.63, N 11.44.

Suitable crystals for the X-ray crystal-structure determination were grown from CH_2Cl_2 /hexane by slow evaporation of the solvent.

3.2. *Z*-*Gly*-*Aib*-(RS)-*Phe*(*2Me*)-*Aib*-*Phe*-*OBn* (**8b**). According to *GP* 5, **7a** (432 mg, 0.779 mmol) was coupled with H-Phe-OBn · HCl (250 mg, 0.857 mmol) by treatment with PyBOP (373 mg, 0.779 mmol) and DIEA (307 mg, 2.205 mmol) in CH₂Cl₂ (3 ml); 5.5 h. CC (AcOEt/hexane 3 :1) furnished 527 mg (85%) of **8b** (mixture of diastereoisomers). Colorless solid. M.p. 134.0–136.5°. $[a]_D = -20.3 (c = 0.5, EtOH)$. IR (KBr): 3300*m*, 1740*m*, 1700*m*, 1660*s*, 1535*s*, 1500*m*, 1455*m*, 1275*m*, 1240*m*, 700*m*. ¹H-NMR (CDCl₃): 7.78 (*s*, NH); 7.46, 7.36 (2*s*, 1 NH); 7.25–7.15 (*m*, 18 arom. H); 7.10–7.05 (*m*, 2 arom. H); 6.69, 6.93 (2*s*, 1 NH); 6.75, 6.71 (2*s*, 1 NH); 6.22, 6.17 (2 br. *s*, 1 NH); 5.01 (*s*, PhCH₂O); 5.00–4.90 (*m*, PhCH₂O); 4.75–4.65 (*m*, CH(Phe)); 3.65–3.60 (*m*, CH₂(Gly)); 3.37, 3.20–3.10, 3.02 (*t*,*m*,*d*,*J* = 15.6, 13.7, 1:2:1, 2 PhCH₂); 1.46, 1.44, 1.40, 1.37, 1.34, 1.33 (6*s*, 2 Me₂C; Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 175.7, 174.1, 173.0, 172.1, 170.3 (5*s*, 4 CO(amide), COOBn); 157.3 (*s*, CO(urethane)); 137.1, 136.1, 135.5 (3*s*, 3 arom. C); 130.9, 129.3, 129.2, 128.5, 128.4, 128.3, 128.12, 128.09, 128.0, 126.9, 126.6 (11*d*, 20 arom. CH); 67.1, 66.8, 66.7 (3*t*, 2 PhCH₂O); 59.9, 57.0 (2*s*, 2 C(2)(Aib), C(2)(Phe(2Me)))); 54.5 (*d*, CH(2)Phe); 45.4, 37.5 (*2t*, CH₂(Gly), 2 PhCH₂); 25.7, 25.0, 24.5, 24.1, 23.4 (5*q*, 2 *Me*₂C, Me(Phe(2Me)))). ESI-MS: 801 (100, [*M* + Na]⁺). Anal. calc. for C₄₄H₅₁N₅O₈ (777.92): C 67.94, H 6.61, N 9.00; found: C 67.79, H 6.88, N 9.00.

3.3. *Z*-*Gly*-*Aib*-*Acb*-*Aib*-*Gly*-*OMe* (**8c**). 3.3.1. *Z*-*Gly*-*Aib*-*Acb*-*NMe*₂ (**4b**). According to *GP*1, **3** (1.68 g, 5.72 mmol) was reacted with 2-(*dimethylamino*)-1-*azabicyclo*[2.3]*hex*-1-*ene* (**1c** [2b]; 782 mg, 6.30 mmol) in THF (16 ml); *ca*. 1 h. The precipitate was filtered and washed with Et₂O: 1.02 g of **8c**. CC (CH₂Cl₂/MeOH 10:1) of the filtrate gave another 1.06 g of **8c**. Total yield: 2.08 g (87%). Colorless solid. M.p. 212.3 – 213.7°. IR (KBr): 3340m, 3270s, 3050m, 2980m, 2940m, 1730s, 1710m, 1680m, 1660s, 1630s, 1540m, 1530m, 1395m, 1270m, 1245m, 1230m, 750w, 700w. ¹H-NMR ((D₆)DMSO): 7.90, 7.89 (2s, 2 NH); 7.56 (*t*-like, NH); 7.35 – 7.30 (*m*, 5 arom. H); 5.05 (*s*, PhCH₂O); 3.61 (*d*, *J* = 5.8, CH₂(Gly)); 3.33 (*s*, Me₂N); 2.60 – 2.50, 2.20 – 2.10, 1.76 – 1.74, 1.60 – 1.55 (*4m*, 2 : 2 : 1:1, 3 CH₂(Acb)); 1.36 (*s*, Me₂C). ¹³C-NMR ((D₆)DMSO): 175.2, 170.6, 168.4 (3s, 3 CO(amide)); 156.6 (*s*, CO(urethane)); 136.9 (*s*, 1 arom. C); 128.3, 127.7, 127.5 (3*d*, 5 arom. CH); 65.4 (*t*, PhCH₂O); 58.2, 55.7 (2*s*, C(2)(Aib), C(2)(Acb)); 44.0 (*t*, CH₂(Gly)); 36.8, 36.0 (2*q*, Me₂N); 30.9 (*t*, 2 CH₂(Acb)); 24.8 (*q*, Me₂C), 14.3 (*t*, CH₂(Acb)). ESI-MS: 441 (100, $[M + Na]^+$).

3.3.2. *Z*-*Gly*-*Aib*-*Acb*-*OH* (**5b**). According to *GP* 2, **4b** (372 mg, 0.890 mmol) was hydrolyzed for 70 h at 40°: 334 mg (96%) of **5b**. Colorless crystals. M.p. 180.5–182.3°. IR (KBr): 3340s, 3290s, 3060*m*, 2980*m*, 2950*m*, 1720s, 1680s, 1670s, 1650s, 1565*m*, 1550s, 1540s, 1525*m*, 1470*m*, 1430s, 1385*m*, 1280s, 1250s, 1235*m*, 1165*m*, 1110*m*, 750*m*. ¹H-NMR (CD₃OD): 7.40–7.25 (*m*, 5 arom. H); 5.11 (*s*, PhCH₂O); 3.72 (*s*, CH₂(Gly)); 2.65–2.50, 2.40–2.30, 2.05–1.85 (3*m*, 3 CH₂(Acb)); 1.44 (*s*, Me₂C). ¹³C-NMR (CD₃OD): 175.5, 174.8, 170.1 (3*s*, 2 CO(amide), COOH); 157.8 (*s*, CO(urethane)); 136.6 (*s*, 1 arom. C); 128.0, 127.5, 127.3 (3*d*, 5 arom. CH); 66.3 (*t*, PhCH₂O); 58.2, 56.2 (2*s*, C(2)(Aib), C(2)(Acb)); 44.0 (*t*, CH₂(Gly)); 30.5 (*t*, 2 CH₂(Acb)); 23.8 (*q*, *Me*₂C), 14.8 (*t*, CH₂(Acb)). ESI-MS: 430 (51, [*M*+K]⁺), 414 (100, [*M*+Na]⁺). Anal. calc. for C₁₉H₂₅N₃O₆ · 0.5 H₂O (400.43): C 56.99, H 6.54, N 10.49; found: C 57.25, H 6.77, N 10.76.

3.3.3. *Z*-*Gly*-*Aib*-*Aib*-*Aib*-*N*(*Me*)*Ph* (**6b**). According to *GP* 1, **5b** (840 mg, 2.15 mmol) was reacted with **1a** (411 mg, 2.36 mmol) in THF/DMF 28 :1 (29 ml). CC (CH₂Cl₂/MeOH 10 :1) gave 1.14 g (94%) of **6b**. Colorless solid. Phase transitions at 137 and 167°. M.p. 187.8 – 188.3°. IR (KBr): 3300s, 3060m, 2990m, 2940m, 1710s, 1680s, 1675s, 1670s, 1660s, 1650s, 1645s, 1630s, 1605m, 1590m, 1550s, 1540s, 1530s, 1520s, 1505s, 1495s, 1470s, 1395m, 1275m, 1265m, 1240s, 1190m, 1155m, 1090m, 700m. ¹H-NMR (CD₃OD): 7.51 (br. *s*, 2 NH); 7.40 – 7.20 (*m*, 10 arom. H); 6.58 (br. *s*, NH); 6.44 (*s*, NH); 5.15 (*s*, PhCH₂O); 3.70 (*d*, *J* = 5.2, CH₂(Gly)); 3.30 (*s*, MeN); 2.70 – 2.55, 2.15 – 1.95, 1.95 – 1.80 (3m, 2 : 3 : 1, 3 CH₂(Acb)); 1.44, 1.28 (2s, 2 Me₂C). ¹³C-NMR (CDCl₃): 174.0, 173.7, 173.5, 170.7 (4*s*, 4 CO(amide)); 157.6 (*s*, CO(urethane)); 146.0, 136.5 (2*s*, 2 arom. C); 129.2, 128.5, 128.2, 127.8, 127.3, 127.0 (6d, 10 arom. CH); 67.0 (*t*, PhCH₂O); 59.2, 56.8, 56.6 (3*s*, 2 C(2)(Aib), C(2)(Acb)); 45.7 (*t*, CH₂(Gly)); 40.5 (*q*, MeN); 31.6, 31.3 (2*t*, 2 CH₂(Acb)); 25.7, 24.9 (2*q*, 2 *Me*₂C); 16.0 (*t*, CH₂(Acb)). ESI-MS: 588 (100, [*M* + Na]⁺), 459 (5, [*M* – N(Me)Ph]⁺), 374 (26, [*M* – Aib-N(Me)Ph]⁺), 277 (3, [*M* – Acb-Aib-N(Me)Ph]⁺).

3.3.4. *Z*-*Gly*-*Aib*-*Aib*-*OH* (**7b**). According to *GP* 2, **6b** (1.033 g, 1.83 mmol) was hydrolyzed, and the product was extracted with AcOEt: 828 mg (95%) of **7b**. Colorless solid. M.p. 201.4–203.6°. IR (KBr): 3350*m*, 3320*s*, 3270*s*, 3000*m*, 2960*m*, 1730*s*, 1705*s*, 1670*s*, 1660*s*, 1650*s*, 1635*s*, 1605*m*, 1545*s*, 1540*s*, 1530*s*, 1470*m*, 1465*m*, 1395*m*, 1330*m*, 1310*m*, 1290*m*, 1260*s*, 1245*s*, 1215*m*, 1170*m*, 1150*m*, 970*m*, 745*m*. ¹H-NMR (CD₃OD): 8.40 (*s*, 0.3 NH); 7.98 (*s*, NH); 7.34 (*s*, 0.7 NH); 7.40–7.25 (*m*, 10 arom. H); 5.08 (*s*, PhCH₂O); 3.71 (*s*, CH₂(Gly)); 2.70–2.60, 2.25–2.10, 2.00–1.90 (3*m*, 2 H each, 3 CH₂(Acb); 1.47, 1.44 (2*s*, 2 Me₂C). ¹³C-NMR (CD₃OD): 178.2, 176.3, 175.3, 172.5 (4*s*, 3 CO(amide), COOH); 159.5 (*s*, CO(urethane)); 138.1 (*s*, 1 arom. C); 129.5, 129.1, 128.9 (3*d*, 5 arom. CH); 68.1 (*t*, PhCH₂O); 60.5, 57.7, 57.1 (3*s*, 2 C(2)(Aib), C(2)(Acb)); 45.3 (*t*, CH₂(Gly)); 31.9 (*t*, 2 CH₂(Acb)); 25.4 (*q*, 2 *Me*₂C); 16.5 (*t*, CH₂(Acb)). ESI-MS: 499 (100, [*M* + Na]⁺). Anal. calc. for C₂₃H₃₂N₄O₇ (476.53): C 57.97, H 6.77, N 11.76; found: C 57.76, H 6.84, N 11.31.

3.3.5. *Z*-*Gly*-*Aib*-*Aib*-*Aib*-*Gly*-*OMe* (**8c**). According to *GP* 5, **7b** (268 mg, 0.653 mmol) and H-Gly-OMe ·HCl (79 mg, 0.629 mmol) were treated with HBPyU (243 mg, 0.564 mmol) and DIEA (220 mg, 1.702 mmol) for 0.5 h. CC (AcOEt/MeOH 20 :1) gave 255 mg (83%) of **8c**. Colorless solid. M.p. 160.0–161.4°. IR (KBr): 3340s, 3280s, 3060m, 2980m, 2950m, 1740s, 1710s, 1685s, 1650s, 1630s, 1550s, 1540s, 1530s, 1505m, 1470m, 1465m, 1455m, 1440m, 1380m, 1365m, 1310m, 1275s, 1240s, 1220m, 1195m, 1170m, 1155m, 1050m, 700m. ¹H-NMR (CDCl₃): 7.52 (*s*, NH); 7.46 (*t*, *J* = 5.8, NH(Gly)); 7.40–7.35 (*m*, 5 arom. H, 1 NH); 6.90 (*s*, NH); 6.31 (*t*-like, NH); 5.14 (*s*, PhCH₂O); 3.98 (*d*, *J* = 5.8, CH₂(Gly)); 3.75 (*d*, *J* = 5.2, CH₂(Gly)); 3.65 (*s*, MeO); 2.75–2.70, 2.05–1.85 (2m, 1:2, 3 CH₂(Acb); 1.55, 1.46 (2*s*, 2 Me₂C). ¹³C-NMR (CDCl₃): 176.5, 175.3, 173.6, 171.0, 170.6 (5*s*, 4 CO(amide), COOMe); 157.8 (*s*, CO(urethane)); 136.4 (*s*, 1 arom. C); 128.6, 128.2, 127.6 (3*d*, 5 arom. CH); 67.2 (*t*, PhCH₂O); 59.2, 57.0, 56.6 (3*s*, 2 C(2)(Aib), C(2)(Acb)); 52.1 (*q*, MeO); 45.8, 41.4 (2*t*, 2 CH₂(Gly)); 3.09 (*t*, 2 CH₂(Acb)); 25.3, 24.7 (2*q*, 2 *Me*₂C); 15.4 (*t*, CH₂(Acb)). ESI-MS: 570 (100, [*M*+Na]⁺).

3.4. *Z*-*Gly*-*Aib*-*Pro*-*Aib*-*Aib*-*N*(*Me*)*Ph* (**8d**). According to *GP* 5, **3** (1.18 g, 4.00 mmol) and H-Pro-Aib-Aib-N(Me)Ph [25b] (1.65 g, 4.40 mmol) in CH₂Cl₂ (20 ml) were treated with HBPyU (1.72 g, 4.00 mmol) and DIEA (1.03 g, 8.00 mmol) for 15 h. CC (CH₂Cl₂/MeOH 20 : 1) gave 2.15 g (82%) of **8d**. Colorless solid. M.p. 107.6–108.7°. $[a]_D = +34.5$ (c = 1.04, EtOH). IR (KBr): 3420*m*, 3300*s*, 3060*m*, 2980*m*, 2930*m*, 1725*s*, 1660*s*, 1625*s*, 1595*m*, 1540*s*, 1520*s*, 1495*s*, 1470*m*, 1455*m*, 1410*m*, 1395*m*, 1365*m*, 1280*m*, 1240*s*, 1215*m*, 1170*m*, 1090*m*, 1050*m*, 705*m*, 700*m*. ¹H-NMR (CDCl₃): 7.57 (*s*, NH); 7.41 (*s*, NH); 7.40–7.15 (*m*, 10 arom. H, 1 NH); 6.14 (br. *s*, NH); 5.10, 5.08 (*AB*, $J_{AB} = 12.0$, PhCH₂O); 4.20–4.15 (*m*, CH₂(2)(Pro)); 3.80–3.55 (*m*, CH₂(Gly), 1 H of CH₂(5)(Pro)); 3.38 (*s*, MeN); 3.35–3.20 (*m*, 1 H of CH₂(5)(Pro)); 2.35–2.20, 1.95–1.80, 1.75–1.65 (3*m*, CH₂(3), CH₂(4)(Pro)); 1.60, 1.58, 1.49, 1.43, 1.28 (5*s*, 1:1:1:2:1, 3 Me₂C). ¹³C-NMR (CDCl₃): 175.1, 174.3, 173.3, 171.7, 169.7 (5*s*, 5 CO(amide)); 156.9 (*s*, CO(urethane)); 145.5, 136.5 (2*s*, 2 arom. C); 129.1, 128.5, 128.2, 128.0, 127.2, 126.6 (6*d*, 10 arom. CH); 6.9 (*t*, PhCH₂O); 64.2 (*d*, CH(2)(Pro)); 57.1, 57.0, 56.5 (3*s*, 3 C(2)(Aib)); 48.7 (*t*, CH₂(5)(Pro)); 44.1 (*t*, CH₂(Gly)); 40.1 (*s*, MeN); 28.8, 26.2 (2*t*, CH₂(3), CH₂(4)(Pro)); 27.8, 26.1, 25.8, 25.6, 23.9, 23.6 (6*q*, 3 *Me*₂C). ESI-MS: 673 (100, [*M* + Na]⁺).

3.5. *Z*-*Gly*-*Aib*-*Pro*-*Aib*-*Aib*-*OH* (**10**). According to *GP* 2, **8d** (1.03 g, 1.58 mmol) was hydrolyzed for 4.5 h. The precipitate was filtered (549 mg of **10**), and the filtrate was purified by CC (CH₂Cl₂/MeOH 10:1). Total yield of **10**: 629 mg (70%). Colorless solid. M.p. 192.3–193.6°. $[a]_D = +22.4$ (*c* = 1.02, MeOH). IR (KBr): 3460*m*, 3400*m*, 3310*s*, 3290*s*, 3040*m*, 2990*m*, 2940*m*, 1715*s*, 1665*s*, 1650*s*, 1600*s*, 1540*s*, 1520*s*, 1470*m*, 1450*m*, 1425*m*, 1410*m*, 1365*m*, 1310*m*, 1290*m*, 1280*m*, 1250*m*, 1215*m*, 1170*m*, 695*m*. ¹H-NMR ((D₆)DMSO): 8.53 (*s*, NH); 7.55–7.30 (*m*, 5 arom. H, 2 NH); 7.17 (*s*, NH); 5.04, 5.00 (*AB*, *J*_{AB} = 12.6, PhCH₂O); 4.09 (*t*, *J* = 7.8, CH₂(2)(Pro)); 3.85–3.70, 3.70–3.55, 3.55–3.54 (3*m*, 1:2:1, CH₂(Gly), CH₂(5)(Pro)); 2.15–2.00, 1.95–1.70, 1.65–1.45 (3*m*, 1:2:1, CH₂(3), CH₂(4)(Pro)); 1.37, 1.34, 1.32, 1.29 (4*s*, 1:2:2:1, 3 Me₂C). ¹³C-NMR ((D₆)DMSO): 175.4, 173.4, 172.3, 171.1, 168.9 (5*s*, 4 CO(amide), COOH); 156.5 (*s*, CO(urethane)); 136.9 (*s*, 1 arom. C); 128.2, 127.7, 127.6 (3*d*, 5 arom. CH); 65.4 (*t*, PhCH₂O); 62.8 (*d*, CH(2)(Pro)); 55.6, 54.6 (2*s*, 3 C(2)(Aib)); 47.8 (*t*, CH₂(5)(Pro)); 42.7 (*t*, CH₂(Gly)); 28.2, 25.5 (2*t*, CH₂(3), CH₂(4)(Pro)); 25.9, 25.4, 24.9, 24.1, 24.0, 23.9 (6*q*, 3 *Me*₂C). ESI-MS: 584 (100, [*M* + Na]⁺). Anal. calc. for C₂₇H₃₉N₅O₈ · 1.5 H₂O (588.66): C 55.09, H 7.19, N 11.90; found: C 54.99, H 7.23, N 11.88.

Suitable crystals for the X-ray crystal-structure determination were grown from MeOH.

3.6. Z-Gly-(RS)-Phe(2Me)-Gly-Aib-Phe-OMe (**8e**). 3.6.1. Z-Gly-Aib-Phe-OMe. According to GP 5, a mixture of Z-Gly-Aib-OH (**3**) [25] (1.03 g, 3.51 mmol) and H-Phe-OMe · HCl (834 mg, 3.87 mmol) in CH₂Cl₂ (5 ml) was treated with PyBOP (1.68 g, 3.52 mmol) and DIEA (1.34 g, 10.65 mmol). CC (AcOEt/ hexane 3 :1) gave 1.36 g (85%) of Z-Gly-Aib-Phe-OMe. Colorless solid. M.p. 150–158.7°. $[a]_D = +6.6$ (c = 1.01, EtOH). IR (KBr): 3380m, 3260m, 3070m, 1745s, 1710s, 1680s, 1635s, 1555s, 1525m, 1380m, 1275m, 1250m, 1215m, 1205m, 1180m, 1160m, 1110w, 1080w, 1030w, 695m. ¹H-NMR (CDCl₃): 7.35–7.20 (m, 8 arom. H); 7.15–7.10 (m, 2 arom. H); 6.67 (br. d, J = 6.7, NH); 6.54, 5.37 (2 br. s, 2 NH); 5.13 (s, PhCH₂O); 4.90–4.80 (ABX, CH(2)(Phe)); 3.78 (d, J = 5.6, CH₂(Gly)); 3.72 (s, MeO); 3.20–3.05 (ABX, $J_{AB} = 13.9$, PhCH₂); 1.48 (s, Me₂C). ¹³C-NMR (CDCl₃): 173.7, 171.9, 168.6 (3s, 2 CO(amide), COOMe); 156.7 (s, CO(urethane)); 136.1, 135.9 (2s, 2 arom. C); 129.3, 128.52, 128.48, 128.2, 128.0, 127.0 (6d, 10 arom. CH); 67.2 (t, PhCH₂O); 57.2 (s, C(2)(Aib)); 53.4 (q, MeO); 52.3 (d, CH(2)(Phe)); 44.9 (t, CH₂(Gly)); 3.77 (t, PhCH₂); 25.0, 24.9 (2q, Me_2 C). ESI-MS: 478 (7, [M + Na]⁺). Anal. calc. for C₂₄H₂₉N₃O₆ (455.50): C 63.29, H 6.42, N 9.23; found: C 62.99, H 6.64, N 8.99.

3.6.2. *H*-*Gly*-*Aib*-*Phe*-*OMe* (**12**). According to *GP* 3, Z-Gly-Aib-Phe-OMe (158.6 mg, 0.348 mmol) was deprotected: 110 mg (98%) of **12**. Colorless oil. IR (CDCl₃): 3420*m*, 3310*m*, 2930*m*, 1740*m*, 1685*s*, 1660*s*, 1635*s*, 1600*m*, 1540*s*, 1510*s*, 1495*s*, 1450*m*, 1435*m*, 1385*m*, 1360*m*, 1330*m*, 1275*m*, 1175*m*, 690*m*. ¹H-NMR (CDCl₃): 7.64 (*s*, NH); 7.30–7.10 (*m*, 5 arom. H, 1 NH); 4.90–4.80 (*ABX*, CH(2)(Phe)); 3.71 (*s*, MeO); 3.26 (*s*, CH₂(Gly)); 3.20–3.05 (*ABX*, J_{AB} =13.9, PhCH₂); 1.51, 1.49 (2*s*, Me₂C). ¹³C-NMR (CDCl₃): 174.0, 172.8, 172.0 (3*s*, 2 CO(amide), COOMe); 136.0 (*s*, 1 arom. C); 129.3, 128.3, 126.9 (3*d*, 5 arom. CH); 56.8 (*s*, C(2)(Aib)); 53.3 (*d*, CH(2)(Phe)); 52.2 (*q*, MeO); 44.9 (*t*, CH₂(Gly)); 37.6 (*t*, PhCH₂); 25.0 (*q*, *Me*₂C). ESI-MS: 665 (46, [2*M* + Na]⁺), 643 (100, [2*M* + 1]⁺), 633 (82, [2*M* – MeOH + Na]⁺), 611 (26, [2*M* – OMe]⁺), 344 (88, [*M* + Na]⁺), 322 (21, [*M* + 1]⁺).

3.6.3. *Z*-*Gly*-(RS)-*Phe*(2*Me*)-*Gly*-*Aib*-*Phe*-*OMe* (**8e**). According to *GP* 5, a mixture of *Z*-*Gly*-(RS)-*Phe*(2*Me*)-*OH* (**11**) [25a] (734 mg, 1.98 mmol) and **12** (700 mg, 2.18 mmol) in CH₂Cl₂ (10 ml) was treated with PyBOP (949 mg, 1.98 mmol) and DIEA (533 mg, 4.12 mmol). CC ($2 \times$, AcOEt/MeOH 20:1) and PLC (AcOEt/MeOH 10:1) gave 623 mg (*S*,*S*)-**8e** and 586 mg (*R*,*S*)-**8e**; total yield: 1.21 mg (91%).

Z-*Gly*-(S)-*Phe*(2*Me*)-*Gly*-*Aib*-*Phe*-*OMe* ((*S*,*S*)-**8e**). Colorless solid. M.p. 162.5 – 164.1°. $[\alpha]_D = -69.0 \ (c = 1.01, EtOH)$. IR (KBr): 3410*m*, 3280*s*, 1745*s*, 1730*s*, 1670*s*, 1660*s*, 1535*s*, 1455*m*, 1385*m*, 1360*m*, 1325*m*, 1280*m*, 1230*s*, 1190*m*, 1185*m*, 1165*m*, 1150*m*, 1110*w*, 1080*w*, 700*m*. ¹H-NMR (CDCl₃): 7.40–7.05 (*m*, 15 arom. H, 3 NH); 6.49 (*s*, NH); 5.80 (*t*-like, NH); 5.05, 4.98 (*AB*, $J_{AB} = 12$, PhCH₂O); 4.80–4.70 (*ABX*, CH(2)(Phe)); 3.90–3.65 (*m*, 2 CH₂(Gly)); 3.60 (*s*, MeO); 3.29, 3.03 (*AB*, $J_{AB} = 13.6$, PhCH₂); 3.20–3.05 (*m*, PhCH₂); 1.48, 1.44, 1.40 (3*s*, Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 175.3, 175.0, 173.2, 170.7, 170.0 (5*s*, 4 CO(amide), COOMe); 157.3 (*s*, CO(urethane)); 136.8, 136.1, 135.3 (3*s*, 3 arom. C); 130.5, 129.1, 128.6, 128.5, 128.4, 128.3, 128.0, 127.2, 126.8 (9*d*, 15 arom. CH); 67.3 (*t*, PhCH₂O); 59.6, 57.0 (2*s*, C(2)(Aib), C(2)(Phe(2Me))); 54.1 (*d*, CH(2)(Phe)); 52.5 (*q*, MeO); 45.4, 44.3 (2*t*, 2 CH₂(Gly)); 36.9, 30.7 (2*t*, 2 PhCH₂); 25.9, 24.3, 22.7 (3*q*, *Me*₂C, Me(Phe(2Me))). ESI-MS: 696 (100, [*M* + Na]⁺). Anal. calc. for C₃₆H₄₃N₅O₈H₂O (691.78): C 62.50, H 6.56, N 10.12; found: C 62.43, H 6.30, N 10.07.

Suitable crystals for the X-ray crystal-structure determination were grown from AcOEt/hexane by slow evaporation of the solvent.

Z-*Gly*-(**R**)-*Phe*(2*Me*)-*Gly*-*Aib*-*Phe*-*OMe* ((*R*,*S*)-**8e**). Colorless solid. M.p. 116.4–118.2°. $[a]_D = +67.5$ (*c* = 1.05, EtOH). IR (KBr): 3380*m*, 3280*s*, 1730*s*, 1680*s*, 1670*s*, 1660*s*, 1540*s*, 1530*s*, 1455*m*, 1440*m*, 1410*m*, 1390*m*, 1365*m*, 1345*m*, 1310*m*, 1275*m*, 1230*m*, 1195*m*, 1175*m*, 1150*m*, 1120*w*, 1080*w*, 1045*m*, 700*m*. ¹H-NMR (CDCl₃): 8.35 (br. *s*, NH); 7.65 (*s*, NH); 7.42 (*d*, *J* = 7.9, NH); 7.40–7.05 (*m*, 15 arom. H); 7.00 (*s*, NH); 6.18 (br. *s*, NH); 5.12, 5.00 (*AB*, $J_{AB} = 12.3$, PhCH₂O); 4.65–4.55 (*m*, CH(2)(Phe)); 4.00–3.60 (*m*, 2 CH₂(Gly)); 3.47, 3.10 (*AB*, $J_{AB} = 13.7$, PhCH₂); 3.24 (*s*, MeO); 3.25–2.95 (*m*, PhCH₂); 1.43, 1.38, 1.26 (3*s*, Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 176.9, 176.4, 172.3, 171.0, 169.9 (5*s*, 4 CO(amide), COOMe); 157.0 (*s*, CO(urethane)); 136.7, 136.4, 136.2 (3*s*, 3 arom. C); 131.1, 129.0, 128.5, 128.4, 128.2, 128.1, 126.8, 126.6 (8*d*, 15 arom. CH); 67.1 (*t*, PhCH₂O); 59.4, 57.0 (2*s*, C(2)(Aib), C(2)(Phe(2Me))); 54.1 (*d*, CH(2)(Phe)); 52.0 (*q*, MeO); 45.5, 44.3 (2*t*, 2 CH₂(Gly)); 39.5, 37.2 (2*t*, 2 PhCH₂); 27.0, 23.5 (2*q*, 1:2, *Me*₂C, Me(Phe(2Me))). ESI-MS: 696 (100, [*M* + Na]⁺). Anal. calc. for $C_{36}H_{43}N_5O_8$ (673.76): C 64.18, H 6.43, N 10.39; found: C 63.91, H 6.33, N 10.53.

3.7. Z-Gly-(RS)-Phe(2Me)-Gly-Aib-Aib-N(Me)Ph (8f). According to GP 5, a mixture of 11 (100 mg, 0.270 mmol) and H-Gly-Aib-Aib-N(Me)Ph [25a] (99 mg, 0.296 mmol) in CH₂Cl₂ (2 ml) was treated with PyBOP (129 mg, 0.270 mmol) and DIEA (85 mg, 0.658 mmol): 152 mg (82%) of 8f. Colorless solid. M.p. 191.9–192.8°. IR (KBr): 3410m, 3300m, 1725m, 1680s, 1665s, 1650s, 1640s, 1595m, 1535s, 1495m, 1355m, 1375m, 1365m, 1270m, 1250m, 1240m, 1095m, 710m. ¹H-NMR (CDCl₃): 7.86 (br. *s*, NH); 7.40–7.05 (*m*, 15 arom. H, 2 NH); 6.83, 6.54 (2br. *s*, 2 NH); 5.08, 5.02 (*AB*, J_{AB} =12.2, PhCH₂O); 3.80–3.60 (*m*, 2 CH₂(Gly)); 3.23 (*s*, MeN); 3.23, 2.98 (*AB*, J_{AB} =13.6, PhCH₂); 1.49, 1.46, 1.27 (3s, 3:1:1, 2 Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 175.6, 175.2, 174.0, 170.8, 169.2 (5s, 5 CO(amide)); 157.3 (*s*, CO(urethane)); 145.3, 136.2, 135.6 (3s, 3 arom. C); 130.5, 129.1, 128.4, 128.2, 128.1, 127.9, 127.0, 126.9 (8d, 15 arom. CH); 67.0 (*t*, PhCH₂O); 59.3, 57.1 (*2s*, 2 C(2)(Aib), C(2)(Phe(2Me))); 45.4, 44.8, 40.7 (3*t*, 2 CH₂(Gly), PhCH₂); 40.3 (*q*, MeN); 25.91, 25.88, 24.9, 22.7 (4q, 1:2:1:1, 2 Me₂C, Me(Phe(2 Me)))). ESI-MS: 725 (8, [M + K]⁺), 709 (100, [M + Na]⁺), 580 (5, [M - N(Me)Ph]⁺). Anal. calc. for C₃₇H₄₆N₆O₇ (686.81): C 64.71, H 6.75, N 12.24; found: C 64.90, H 6.89, N 12.40.

Suitable crystals for the X-ray crystal-structure determination were grown from AcOEt/hexane by slow evaporation of the solvent.

3.8. Z-Gly-(RS)-Phe(2Me)-Pro-Aib-Phe-OMe (**8g**). 3.8.1. Z-Pro-Aib-Phe-OMe. According to GP 5, a mixture of Z-Pro-Aib-OH [25b] (1.14 g, 3.40 mmol) and H-Phe-OMe ·HCl (808 mg, 3.75 mmol) in CH₂Cl₂ (6 ml) was treated with PyBOP (1.63 g, 3.40 mmol) and DIEA (1.32 g, 10.23 mmol). CC (AcOEt/ hexane 3 :1) gave 1.33 g (79%) of Z-Pro-Aib-Phe-OMe. Colorless solid. M.p. 169.7–170.4°. $[a]_D = -23.8$ (c = 0.817, EtOH). IR (KBr): 3380m, 3280m, 1745s, 1690s, 1665s, 1640s, 1550m, 1525m, 1455m, 1435s, 1355m, 1240m, 1215m, 1205m, 1180m, 1170m, 1120m, 700m. ¹H-NMR (CDCl₃): 7.35–7.10 (m, 10 arom. H, 0.8 NH); 6.86 (br. s, NH); 6.40 (br. s, 0.2 NH); 5.20–5.10 (m, PhCH₂O); 4.80–4.75, 4.25–4.20 (2m, CH(2)(Phe), CH(2)(Pro)); 3.68 (s, MeO); 3.60–3.40 (m, CH₂(5)(Pro)); 3.20–3.00 (ABX, J_{AB} = 13.9, PhCH₂); 2.25–1.25 (m, CH₂(3), CH₂(4)(Pro)); 1.44, 1.26 (2br. s, Me₂C). ¹³C-NMR (CDCl₃): 173.7, 171.9 (2s, 2 CO(amide), COOMe); 156.1 (s, CO(urethane)); 136.3 (s, 2 arom. C); 129.2, 128.4, 128.3, 128.1, 127.8, 126.8 (6d, 10 arom. CH); 67.2 (t, PhCH₂O); 61.0 (d, CH(2)(Pro)); 57.2 (s, C(2)(Aib)); 53.6 (d, CH(2)(Phe)); 52.0 (s, MeO); 47.0 (t, CH₂(5)(Pro)); 37.8 (t, PhCH₂); 28.4, 24.8 (2t, CH₂(3), CH₂(4)(Pro)); 25.5, 24.8 (2q, Me₂C). ESI-MS: 518 (100, [M + Na]⁺). Anal. calc. for C₂₇H₃₃N₃O₆ (495.57): C 65.44, H 6.71, N 8.48; found: C 65.49, H 6.97, N 8.54.

3.8.2. *H-Pro-Aib-Phe-OMe* (13). According to *GP* 3, Z-Pro-Aib-Phe-OMe (1.15 g, 2.32 mmol) in MeOH (25 ml) in the presence of Pd/C (113 mg) was deprotected (2 h): 842 mg (84%) of 13. Colorless oil. $[a]_D = +3.6$ (c = 1.24, CHCl₃). IR (CHCl₃): 3280*m*, 3020*m*, 3000*s*, 2980*m*, 1740*s*, 1670*s*, 1510*s*, 1455*m*, 1440*m*, 1390*m*, 1360*m*, 1215*m*, 1180*m*, 850*m*, 700*m*. ¹H-NMR (CDCl₃): 7.89 (*s*, NH); 7.41 (d, J = 7.5, NH); 7.30 – 7.15 (m, 5 arom. H); 4.80 – 4.75, 3.70 – 3.65 (2m, CH(2)(Pro), CH(2)(Phe)); 3.69 (*s*, MeO); 3.20 – 2.90 (m, CH₂(5)(Pro), PhCH₂, NH(Pro)); 2.15 – 2.05, 1.90 – 1.85, 1.75 – 1.65 (3m, 1:1:2, CH₂(3), CH₂(4)(Pro)); 1.47, 1.45 (2s, Me₂C). ¹³C-NMR (CDCl₃): 175.2, 174.2, 172.1 (3s, 2 CO(amide), COOMe); 136.3 (s, 1 arom. C); 129.3, 128.4, 126.9 (3*d*, 5 arom. CH); 60.7 (d, CH(2)(Pro)); 56.9 (s, C(2)(Aib)); 53.5 (d, CH(2)(Phe)); 52.2 (s, MeO); 47.2 (t, CH₂(5)(Pro)); 37.7 (t, PhCH₂); 30.6, 26.0 (2t, CH₂(3), CH₂(4)(Pro)); 25.2, 25.1 (2q, Me_2 C). CI-MS: 363 (21), 362 (100, [M + 1]⁺).

3.8.3. Z-Gly-(RS)-Phe(2Me)-Pro-Aib-Phe-OMe (8g). According to GP 5, a mixture of 11 (374.5 mg, 1.01 mmol) and 13 (400 mg, 1.11 mmol) in CH_2Cl_2 (10 ml) was treated with TBTU (326.2 mg, 1.02 mmol) and DIEA (301.8 mg, 2.34 mmol) for 16 h. CC (4 × , AcOEt/MeOH 20:1) gave 193.9 mg of (*S*,*S*)-8g and 319.8 mg of (*R*,*S*)-8g; total yield: 513.7 mg (71%).

Z-*Gly*-(S)-*Phe*(2*Me*)-*Pro*-*Aib*-*Phe*-*OMe* ((*S*,*S*,*S*)-**8g**). Colorless lacquer. M.p. 59.1–60.7°. $[a]_D = -17.6$ (c = 1.03, EtOH). IR (KBr): 3300*m*, 2940*m*, 1745*s*, 1740*s*, 1730*s*, 1720*s*, 1670*s*, 1660*s*, 1635*s*, 1625*s*, 1565*m*, 1540*s*, 1520*s*, 1505*s*, 1460*m*, 1455*m*, 1410*m*, 1385*m*, 1240*s*, 1170*m*, 1050*m*, 700*m*. ¹H-NMR (CDCl₃): 7.49 (d, J = 8.1, NH); 7.45–7.05 (m, 15 arom. H, NH); 6.92 (s, NH); 5.71 (t-like, NH); 5.16, 5.10 (*AB*, $J_{AB} = 12.2$, PhCH₂O); 4.80–4.70 (m, CH(2)(Phe)); 4.41 (t, CH(2)(Pro)); 3.90–3.70 (*AB* of *ABX*, CH₂(Gly)); 3.61 (s, MeO); 3.40–3.10, 3.10–2.95 (2m, 1:1, 2 PhCH₂), CH₂(5)(Pro)); 2.25–2.10, 1.85–1.65, 1.65–1.50 ($3m, 1:2:1, CH_2(3), CH_2(4)$ (Pro)); 1.52, 1.50, 1.34 ($3s, Me_2C, Me(Phe(2Me))$)). ¹³C-NMR (CDCl₃): 175.1, 172.2, 171.6, 171.1, 169.0 (5s, 4 CO(amide), COOMe); 156.8 (s, CO(urethane)); 137.1, 136.0, 134.5 (3s, 3 arom. C); 130.2, 129.2, 128.9, 128.6, 128.5, 128.2, 127.7, 126.5 (8*d*, 15)

arom. CH); 67.4 (*t*, PhCH₂O); 63.6 (*d*, C(2)(Pro)); 60.1, 57.1 (2*s*, C(2)(Aib), C(2)(Phe(2Me))); 53.9 (*d*, CH(2)(Phe)); 52.0 (*q*, MeO); 48.6 (*t*, CH₂(5)(Pro)); 44.8 (*t*, CH₂(Gly)); 43.0, 37.9 (2*t*, 2 PhCH₂); 28.3, 26.0 (2*t*, CH₂(3), CH₂(4)(Pro)); 26.7, 24.4, 24.0 (3*q*, Me_2 C, Me(Phe(2Me))). ESI-MS: 736 (100, $[M + Na]^+$).

Z-*Gly*-(**R**)-*Phe*(2*Me*)-*Pro*-*Aib*-*Phe*-*OMe* ((*R*,*S*,*S*)-**8g**). Colorless solid. M.p. 165.5 – 166.5°. $[a]_{\rm D}$ = +68.9 (*c* = 1.04, EtOH). IR (KBr): 3400*s*, 3060*m*, 3030*m*, 2980*m*, 2940*m*, 1750*s*, 1730*s*, 1655*s*, 1625*s*, 1545*s*, 1535*s*, 1500*m*, 1455*m*, 1445*m*, 1405*m*, 1385*m*, 1375*m*, 1365*m*, 1305*m*, 1280*m*, 1240*s*, 1170*m*, 1160*m*, 1130*w*, 1100*m*, 1080*m*, 1045*m*, 1030*m*, 1000*m*, 740*m*, 700*m*. ¹H-NMR (CDCl₃): 7.50 (*d*, *J* = 8.1, NH); 7.39 (*s*, NH); 7.40 – 7.00 (*m*, 15 arom. H); 6.85 (*s*, NH); 5.70 (br. *s*, NH); 5.08 (*s*, PhCH₂O); 4.75 – 4.65 (*m*, CH(2)(Phe)); 4.37 (*t*-like, CH(2)(Pro)); 4.00 – 3.80 (*AB* of *ABX*, CH₂(Gly)); 3.75 – 3.65, 3.35 – 3.20 (2*m*, CH₂(5)(Pro)); 3.61 (*s*, MeO); 3.42, 3.17 (*AB*, *J_{AB}* = 13.8, PhCH₂); 3.25 – 3.10, 3.10 – 2.95 (2*m*, PhCH₂); 2.20 – 2.05, 1.95 – 1.80, 1.80 – 1.60 (3*m*, 1:1:2, CH₂(3), CH₂(4)(Pro)); 1.46, 1.39, 1.26 (3*s*, Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 175.3, 172.5, 172.4, 171.2, 169.3 (5*s*, 4 CO(amide), COOMe); 156.8 (*s*, CO(urethane)); 136.8, 136.6, 136.1 (3*s*, 3 arom. C); 131.1, 129.1, 128.6, 128.3, 128.22, 128.17, 128.1, 126.8, 126.6 (9*d*, 15 arom. CH); 67.2 (*t*, PhCH₂O); 63.7 (*d*, CH(2)(Pro)); 59.1, 57.1 (2*s*, C(2)(Aib), C(2)(Phe(2Me))); 53.6 (*d*, CH(2)(Phe)); 52.0 (*q*, MeO); 48.7 (*t*, CH₂(5)(Pro)); 44.4 (*t*, CH₂(Gly)); 40.9, 37.8 (2*t*, 2 PhCH₂); 28.5, 26.0 (2*t*, CH₂(3), CH₂(4)(Pro)); 27.0, 24.1, 21.1 (3*q*, *Me*₂C, Me(Phe(2Me))). ESI-MS: 736 (100, [*M* + Na]⁺), 714 (10, [*M* + 1]⁺), 535 (6, [*M* – PheOMe]⁺).

Suitable crystals for the X-ray crystal-structure determination were grown from AcOEt/MeOH by slow evaporation of the solvent.

3.9. Z-Gly-(RS)-Phe(2Me)-Pro-Aib-Aib-N(Me)Ph (**8h**). According to GP 5, a mixture of **11** (1.25 g, 3.36 mmol) and H-Pro-Aib-Aib-N(Me)Ph [25b] (1.40 g, 3.73 mmol) in CH₂Cl₂ (30 ml) was treated with TBTU (1.20 g, 3.74 mmol), HOBt (910 mg, 6.73 mmol), and DIEA (916 mg, 7.09 mmol) for 18 h. CC (CH₂Cl₂/MeOH 15:1) gave 1.76 g (72%) of **8h**. Separation of 815.5 mg of the mixture of the diastereoisomers by PLC (AcOEt/MeOH 20:1, $5 \times$ developed) led to 224.6 mg of (*S*,*S*)-**8h** and 470.6 mg of (*R*,*R*)-**8h**.

Z-*Gly*-(S)-*Phe*(2*Me*)-*Pro*-*Aib*-*Aib*-*N*(*Me*)*Ph* ((*S*,*S*)-**8h**). Colorless solid. M.p. 92.4–93.9°. $[a]_D = +14.0 \ (c = 0.57, EtOH)$. IR (KBr): 3310*s*, 2980*m*, 2930*m*, 1730*s*, 1680*s*, 1660*s*, 1630*s*, 1595*m*, 1545*s*, 1540*s*, 1530*s*, 1495*s*, 1465*m*, 1455*m*, 1405*m*, 1390*s*, 1360*m*, 1280*m*, 1240*s*, 1200*m*, 1160*m*, 1090*m*, 1050*m*, 740*m*, 700*s*. ¹H-NMR (CDCl₃): 7.52 (*s*, NH); 7.35 – 7.05 (*m*, 15 arom. H, 2 NH); 6.29 (br. *s*, NH); 5.12, 5.08 (*AB*, $J_{AB} = 12.3$, PhCH₂O); 4.22 (*t*, J = 8.3, CH(2)(Pro)); 3.80 – 3.55 (*AB* of *ABX*, $J_{AB} = 17.2$, CH₂(Gly)); 3.88 (*s*, MeN); 3.20 – 3.00 (*m*, CH₂(5)(Pro)); 2.91 (*s*, PhCH₂); 2.30 – 2.15, 1.85 – 1.70, 1.70 – 1.40 (3*m*, 1:1:2, CH₂(3), CH₂(4)(Pro)); 1.60, 1.57, 1.47, 1.45, 1.40 (5*s*, 2 Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 174.7, 174.1, 171.8, 171.6, 169.8 (5*s*, 5 CO(amide)); 157. (*s*, CO(urethane)); 146.1, 136.4, 134.6 (3*s*, 3 arom. C); 130.1, 129.0, 128.8, 128.6, 128.3, 128.1, 127.6, 127.2, 126.6 (9*d*, 15 arom. CH); 67.0 (*t*, PhCH₂O); 64.4 (*d*, C(2)(Pro)); 60.3, 57.2, 57.0 (3*s*, 2 C(2)(Aib), C(2)(Phe(2Me))); 48.6, 44.6, 43.6 (3*t*, CH₂(Gly), CH₂(5)(Pro), PhCH₂); 40.1 (*q*, MeN); 28.6, 26.2 (*z*, CH₂(3), CH₂(4)(Pro)); 27.8, 26.4, 25.4, 24.3, 23.7 (5*q*, 2 *Me*₂C, Me(Phe(2Me))). ESI-MS: 749 (100, [*M* + Na]⁺). Anal. calc. for C₄₀H₅₀N₆O₇· 0.5H₂O (735.88): C 65.29, H 6.99, N 11.42; found: C 65.28, H 7.17, N 11.15.

Z-*Gly*-(**R**)-*Phe*(2*Me*)-*Pro*-*Aib*-*Aib*-*N*(*Me*)*Ph* ((*R*,*S*)-**8h**). Colorless solid. M.p. 168.7 – 170.2°. $[\alpha]_{D} = +114$ (*c* = 0.51, EtOH). IR (KBr): 3300s, 3060w, 3030w, 2980w, 2940w, 1725s, 1660s, 1625s, 1595m, 1540s, 1495m, 1465m, 1455m, 1410m, 1395m, 1375m, 1360m, 1280m, 1240m, 1160m, 1135m, 1090m, 1050m, 705s. ¹H-NMR (CDCl₃): 7.44 (*s*, NH); 7.35 – 7.05 (*m*, 15 arom. H, 2 NH); 5.95 (br. *s*, NH); 5.09 (*s*, PhCH₂O); 4.23 (*t*, *J* = 8.3, CH(2)(Pro)); 3.90 – 3.60 (*AB* of *ABX*, *J_{AB}* = 17, CH₂(Gly)); 3.70 – 3.60, 3.30 – 3.10 (2*m*, CH₂(5)(Pro)); 3.37, 3.08 (*AB*, *J_{AB}* = 13.7, PhCH₂); 3.29 (*s*, MeN); 2.30 – 2.15, 1.95 – 1.85, 1.85 – 1.60 (3*m*, 1:1:2, CH₂(3), CH₂(4)(Pro)); 1.54, 1.45, 1.39, 1.17 (4*s*, 1:1:2:1, 2 Me₂C, Me(Phe(2Me))). ¹³C-NMR (CDCl₃): 174.8, 173.9, 173.1, 171.3, 169.6 (5*s*, 5 CO(amide)); 156.8 (*s*, CO(urethane)); 145.9, 136.5, 136.2 (3*s*, 3 arom. C); 131.2, 128.9, 128.6, 128.3, 128.2, 127.2, 126.8, 126.6 (8*d*, 15 arom. CH); 67.1 (*t*, PhCH₂O); 64.4 (*d*, CH(2)(Pro)); 58.9, 57.1 (2*s*, 2 C(2)(Aib), C(2)(Phe(2Me)))); 48.8 (*t*, CH₂(5)(Pro)); 44.2 (*t*, CH₂(Gly)); 40.7 (*t*, PhCH₂); 40.1 (*q*, MeN); 28.7, 26.2 (2*t*, CH₂(3), CH₂(4)(Pro)); 27.7, 26.3, 25.6, 23.7, 20.9 (5*q*, 2 *Me*₂C, Me(Phe(2Me))). ESI-MS: 749 (100, [*M* + Na]⁺), 620 (7, [*M* – N(Me)Ph]⁺).

	6a	8a	(<i>S</i> , <i>S</i>)- 8 e
Crystallized from	CH ₂ Cl ₂ /Et ₂ O/hexane	CH ₂ Cl ₂ /hexane	AcOEt/hexane
Empirical formula	$C_{35}H_{43}N_5O_6 \cdot 0.5 C_6H_{14}$	$C_{31}H_{41}N_5O_8$	$C_{36}H_{43}N_5O_8 \cdot H_2O$
Formula weight	672.84	611.69	691.78
Crystal color, habit	colorless, prism	colorless, prism	colorless, irregular prism
Crystal dimensions [mm]	$0.23 \times 0.25 \times 0.45$	$0.33 \times 0.41 \times 0.43$	$0.25 \times 0.27 \times 0.50$
Temp. [K]	173(1)	173(1)	173(1)
Crystal system	monoclinic	monoclinic	orthorhombic
Space group	$P2_1/n$	$P2_1/n$	$P2_{1}2_{1}2_{1}$
Z	4	4	4
Reflections for cell determination	22	25	25
2θ Range for cell determination [°]	32-39	38 - 40	23-33
Unit cell parameters a [Å]	11.642(3)	13.698(3)	15.347(4)
b [Å]	15.135(3)	13.979(4)	24.232(4)
c [Å]	21.307(2)	17.409(3)	10.028(4)
$\alpha [\circ]$	90	90	90
β [°]	97.294(12)	109.133(14)	90
γ[°]	90	90	90
V [Å ³]	3724(1)	3150(1)	3729(2)
$D_x [g \text{ cm}^{-3}]$	1.200	1.290	1.232
$\mu(MoK_a)$ [mm ⁻¹]	0.0817	0.0940	0.0892
Scan type	$\omega/2\theta$	$\omega/2\theta$	ω
$2\theta_{(\text{max})}$ [°]	55	60	55
Transmission factors (min; max)	_	-	-
Total reflections measured	9296	9898	5535
Symmetry-independent reflections	8539	9182	5389
Reflections used $[I > 2\sigma(I)]$	4362	6222	3892
Parameters refined; restraints	484; 164	423;0	548; 202
Final $R(F)$ [$I > 2\sigma(I)$ reflections]	0.0577	0.0454	0.0549
$wR(F^2)$ (all data)	0.1671	0.1262	0.1682
Weighting parameter $(a; b)^{a}$)	0.0830; 0	0.0525; 0.4262	0.0886; 0.8267
Secondary extinction coefficient	0.0022(6)	-	-
Goodness-of-fit	0.927	1.034	1.058
Final Δ_{\max}/σ	0.000	0.000	0.000
$\Delta \rho$ (max; min) [e Å ⁻³]	0.29; -0.24	0.37; -0.18	0.79; -0.26

Table 4. Crystallographic Data for Compounds 6a, 8a, (S,S)-8e, 8f, (R,S,S)-8g, (R,S)-8h, and 10

^a) $w^{-1} = \sigma^2 (F_o^2) + (aP)^2 + bP$, where $P = (F_o^2 + 2F_c^2)/3$.

Suitable crystals for the X-ray crystal-structure determination were grown from AcOEt/MeOH/ hexane by slow evaporation of the solvent.

4. X-Ray Crystal-Structure Determination of 6a, 8a, (S,S)-8e, 8f, (R,S,S)-8g, (R,S)-8h, and 10 (see Table 4 and Figs. $1-7)^{9}$). The measurements were performed using graphite-monochromated MoK_a radiation (λ 0.7107 Å) on a Rigaku AFC5R diffractometer fitted to a 12-kW rotating-anode generator. The intensities were corrected for Lorentz and polarization effects. Empirical absorption corrections based on azimuthal scans of several reflections [45] were applied in the cases of 8f and (R,S,S)-8g.

⁹⁾ CCDC-989272-989278 contain the supplementary crystallographic data for this article. These data can be obtained free of charge from *The Cambridge Crystallographic Data Centre via* www.ccdc.cam.ac.uk/data_request/cif.

8f	(<i>R</i> , <i>S</i> , <i>S</i>)- 8g	(<i>R</i> , <i>S</i>)-8h	10
AcOEt/hexane	MeOH/AcOEt	AcOEt/MeOH/hexane	МеОН
$C_{37}H_{46}N_6O_7 \cdot 0.33C_6H_{14}$	$C_{39}H_{47}N_5O_8 \cdot C_4H_8O_2 \cdot 0.5 H_2O$	$C_{40}H_{50}N_6O_7 \cdot 0.5 C_6H_{14} \cdot 0.25 H_2O$	$C_{27}H_{39}N_5O_8 \cdot CH_4O_{10}$
715.33	810.94	774.46	593.67
colorless, plate	colorless, plate	colorless, hexagonal prism	colorless, prism
$0.10 \times 0.38 \times 0.42$	$0.20 \times 0.45 \times 0.50$	$0.36 \times 0.36 \times 0.38$	$0.18 \times 0.33 \times 0.50$
173(1)	173(1)	173(1)	173(1)
orthorhombic	orthorhombic	trigonal	orthorhombic
Pbca	$P2_{1}2_{1}2_{1}$	P3 ₁ 21	$P2_{1}2_{1}2_{1}$
8	4	12	4
25	25	25	17
21-25	24-35	20-27	20-26
25.465(4)	16.601(5)	15.706(6)	12.983(9)
20.729(6)	26.785(8)	15.706(6)	10.249(5)
14.536(5)	9.720(6)	60.971(6)	23.918(7)
90	90	90	90
90	90	90	90
90	90	120	90
7673(4)	4322(3)	13026(11)	3183(3)
1.239	1.246	1.185	1.239
0.0860	0.0895	0.0813	0.0928
ω	ω	ω	$\omega/2\theta$
50	50	50	55
0.806; 1.000	0.903; 1.000	_	-
8431	5016	29374	4118
6768	4874	15416	4090
2204	3299	8853	2798
457; 0	554;0	1087; 187	407; 36
0.0701	0.0446	0.0787	0.0611
0.1842	0.1104	0.2097	0.1975
0.0781; 0	0.0392; 0.9522	0.0641; 16.507	0.0808; 2.3471
-	_	0.0014(2)	-
0.790	1.013	1.057	1.085
0.000	0.000	0.002	0.000
0.32; -0.25	0.19; -0.21	0.42; -0.38	0.33; -0.36

Equivalent reflections, other than *Friedel* pairs for structures in non-centrosymmetric space groups, were merged. The data collection and refinement parameters are compiled in *Table 4*, and views of the molecules are shown in *Figs. 1–7*. The structures were solved by direct methods using SHELXS86 [46] for **6a**, **8a**, (S,S)-**8e**, **8f**, (R,S,S)-**8g**, and **10**, which revealed the positions of all non-H-atoms, and SnB [47]¹⁰) for (R,S)-**8h**, which revealed the positions of most of the non-H-atoms. In the latter case, the remaining non-H-atoms were located in a subsequent difference electron-density map. For all structures,

¹⁰) We are grateful to *R. Miller* and *C. Weeks* of the *Hauptman-Woodward* Medical Research Institute, Buffalo, who carried out the solution of the structure using what was, at the time in 1996, a new, unreleased version of their program, SnB. Without their assistance, this structure determination would not nave been successful.

the non-H-atoms were refined anisotropically. Unless indicated otherwise for individual structures below, H-atoms on heteroatoms were generally placed in the positions indicated by difference electron-density maps and their positions were allowed to refine together with individual isotropic displacement parameters. All other H-atoms were placed in geometrically calculated positions and refined by using a riding model, with $U_{iso}(H) = 1.2 U_{eq}(parent atom)$ (1.5 U_{eq} for Me groups).

In the case of **6a**, the Ph ring of the middle Bn group of the molecule is disordered over two orientations. The site occupation factor for the major conformation refined to 0.645(19). Similarity restraints were applied to the chemically equivalent bond lengths and angles involving all disordered C-atoms, while neighboring atoms within and between each conformation of the disordered ring were restrained to have similar atomic displacement parameters. There are disordered solvent molecules present in the crystal lattice, and these lie across centres of inversion, so that the ratio of peptide to solvent molecules is 2:1. The solvent appears to be a hexane molecule which has two orientations, although the end atoms in one orientation may be further disordered. As the disorder could not be modelled adequately, the contribution of the solvent molecules to the intensity data was removed by using the SQUEEZE routine [48] of the PLATON program [49]. Omission of the solvent molecules leaves two cavities of 217 Å³ per unit cell. The number of electrons contributing to each void in the structure was calculated by the SQUEEZE routine to be *ca*. 41 e, although this may be an underbound. Allowing for one hexane molecule per cavity yields 50 e and this assumption was used in the subsequent calculation of the empirical formula, formula weight, density, linear absorption coefficient and F(000).

In the case of (S,S)-**8e**, the terminal Bn group is disordered over two orientations with a site occupation factor for the major conformation of 0.669(5). Similarity restraints were applied to the chemically equivalent bond lengths and angles involving all disordered C-atoms, while neighboring atoms within and between each conformation of the disordered group were restrained to have similar atomic displacement parameters. The terminal COOMe group might also be slightly disordered, because the atomic displacement parameters of these atoms, particularly O(24), are large. However, the disorder could not be resolved reasonably and average positions were used for the atoms of this region of the molecule. As a result, the C(24)=O(24) bond appears to be somewhat shorter than it should be. The asymmetric unit also contains two sites which are partially occupied by H₂O molecules. Initial refinement of the site occupation factors of these molecules gave values close to 0.5 and in the final refinement the site occupation factors of the H₂O molecules were set to 0.5. H-Atoms were not included for the H₂O molecules.

The crystals of **8f** rapidly lost included solvent on standing in air. The asymmetric unit contains one peptide molecule plus approximately one third of a highly disordered hexane molecule which sits across a centre of inversion. As the disorder could not be modelled adequately, the contribution of the solvent molecules to the intensity data was removed by using the SQUEEZE routine of the PLATON program. Omission of the solvent molecules leaves four cavities of 128 Å³ per unit cell. The number of electrons contributing to each void in the structure was calculated by the SQUEEZE routine to be *ca.* 31 e. Allowing for two thirds of a hexane molecule per cavity yields 30 e, which leads to a peptide/hexane ratio in the structure of 3:1, and this assumption was used in the subsequent calculations. The amide H-atoms were placed in geometrically calculated positions and refined as described above for riding H-atoms.

In the case of (R,S,S)-**8**g, the asymmetric unit contains one peptide molecule, one AcOEt molecule, and one partially occupied site for a H₂O molecule. The site occupation factor for the O-atom of H₂O was initially refined and then fixed at 0.5. The H-atoms of the H₂O molecule were included in the positions obtained from a difference electron-density map and then constrained to ride on the parent O-atom with $U_{iso}(H) = 1.5 U_{ea}(O)$.

The asymmetric unit of (R,S)-**8h** contains two peptide, one hexane, and one half of a H₂O molecule, where the latter sits on a two-fold axis. One of the peptide molecules has a disordered Ph ring, in the Z group. Two sets of positions were defined for the atoms of this Ph ring and the site occupation factor of the major conformation of the ring refined to 0.504(1). Similarity restraints were applied to the chemically equivalent bond lengths and angles involving all disordered C-atoms, the C–C bonds were restrained to 1.395(5) Å, the two conformations of the ring were restrained to be planar, and neighboring atoms within and between each conformation of the disordered Ph ring were restrained to have similar atomic displacement parameters. The amide H-atoms were placed in geometrically calculated positions

and refined as described above for riding H-atoms. The position of the symmetry-unique water H-atom was refined with $U_{iso}(H) = 1.5 U_{eq}(O)$.

In the case of 10, the asymmetric unit contains one molecule of the peptide plus a disordered molecule of MeOH. The site occupation factor of the major orientation of the MeOH molecule refined to 0.659(14). Neighboring atoms within and between each conformation of the disordered MeOH molecule were restrained to have similar atomic displacement parameters. The amide and peptide OH H-atoms were placed in geometrically calculated positions and refined as described above for riding H-atoms; for the OH H-atom $U_{iso}(H) = 1.5 U_{eq}(O)$. The OH H-atoms of the MeOH molecules were not included in the model.

The refinement of each structure was carried out on F^2 by using full-matrix least-squares procedures, which minimized the function $\Sigma w (F_o^2 - F_c^2)^2$. A correction for secondary extinction was applied in the case of **6a** and (R,S)-**8h**. Neutral atom-scattering factors for non-H-atoms were taken from [50], and the scattering factors for H-atoms were taken from [51]. Anomalous dispersion effects were included in F_c [52]; the values for f' and f'' were those of [53]. The values of the mass attenuation coefficients are those of [54]. All refinements were performed using SHELXL-2013 [55].

REFERENCES

- D. Obrecht, H. Heimgartner, *Helv. Chim. Acta* 1981, 64, 482; D. Obrecht, H. Heimgartner, *Helv. Chim. Acta* 1987, 70, 102.
- [2] a) P. Wipf, H. Heimgartner, *Helv. Chim. Acta* 1986, 69, 1153; b) P. Wipf, H. Heimgartner, *Helv. Chim. Acta* 1987, 70, 354; c) P. Wipf, H. Heimgartner, *Helv. Chim. Acta* 1988, 71, 140; d) P. Wipf, H. Heimgartner, *Helv. Chim. Acta* 1990, 73, 13.
- [3] M. Sahebi, P. Wipf, H. Heimgartner, Tetrahedron 1989, 45, 2999.
- [4] a) C. B. Bucher, A. Linden, H. Heimgartner, *Helv. Chim. Acta* 1995, 78, 935; b) C. B. Bucher, H. Heimgartner, *Helv. Chim. Acta* 1996, 79, 1903.
- [5] C. Strässler, A. Linden, H. Heimgartner, Helv. Chim. Acta 1997, 80, 1528.
- [6] K. A. Brun, A. Linden, H. Heimgartner, Helv. Chim. Acta 2008, 91, 526.
- [7] I. Dannecker-Dörig, A. Linden, H. Heimgartner, Helv. Chim. Acta 2011, 94, 993.
- [8] S. A. Stoykova, A. Linden, H. Heimgartner, *Helv. Chim. Acta* 2012, 95, 1325; S. A. Stoykova, A. Linden, H. Heimgartner, *Helv. Chim. Acta* 2013, 96, 1714.
- [9] R. Luykx, C. B. Bucher, A. Linden, H. Heimgartner, *Helv. Chim. Acta* 1996, 79, 527; R. T. N. Luykx, A. Linden, H. Heimgartner, *Helv. Chim. Acta* 2003, 86, 4093.
- [10] N. Pradeille, H. Heimgartner, J. Pept. Sci. 2003, 9, 827; N. Pradeille, O. Zerbe, K. Möhle, A. Linden, H. Heimgartner, Chem. Biodiversity 2005, 2, 1127; N. Pradeille, M. Tzouros, K. Möhle, A. Linden, H. Heimgartner, Chem. Biodiversity 2012, 9, 2528.
- [11] W. Altherr, A. Linden, H. Heimgartner, *Chem. Biodiversity* 2007, 4, 1144; P. Blaser, W. Altherr, A. Linden, H. Heimgartner, *Chem. Biodiversity* 2013, 10, 920.
- [12] a) S. Stamm, H. Heimgartner, Eur. J. Org. Chem. 2004, 3820; b) S. Stamm, A. Linden, H. Heimgartner, Helv. Chim. Acta 2006, 89, 1; c) S. Stamm, H. Heimgartner, Tetrahedron 2006, 62, 9671.
- [13] J. Lehmann, A. Linden, H. Heimgartner, *Tetrahedron* 1998, 54, 8721; J. Lehmann, A. Linden, H. Heimgartner, *Tetrahedron* 1999, 55, 5359; J. Lehmann, H. Heimgartner, *Helv. Chim. Acta* 1999, 82, 1899.
- [14] R. A. Breitenmoser, H. Heimgartner, Helv. Chim. Acta 2001, 84, 786.
- [15] A. Budzowski, A. Linden, H. Heimgartner, Helv. Chim. Acta 2008, 91, 1471.
- [16] A. Bärtsch, B. Bischof, H. Heimgartner, Pol. J. Chem. 2009, 83, 195.
- [17] K. A. Brun, A. Linden, H. Heimgartner, Helv. Chim. Acta 2001, 84, 1756.
- [18] G. N. Ramachandran, C. Ramakrishnan, V. Sasisekharan, J. Mol. Biol. 1963, 7, 95; G. N. Ramachandran, V. Sasisekharan, Adv. Protein Chem. 1968, 23, 283.
- [19] B. V. V. Prasad, P. Balaram, Crit. Rev. Biochem. Mol. Biol. 1984, 16, 307.
- [20] C. Toniolo, E. Benedetti, ISI Atlas Sci.: Biochem. 1988, 5, 1238; C. Toniolo, E. Benedetti, Macromolecules 1991, 24, 4004; C. Toniolo, Biopolymers 1989, 28, 247.

- [21] I. L. Karle, P. Balaram, Biochemistry 1990, 29, 6747.
- [22] E. Benedetti, M. Saviano, R. Iacovino, C. Pedone, A. Santini, M. Crisma, F. Formaggio, C. Toniolo, Q. B. Broxterman, J. Kamphuis, *Biopolymers* 1998, 46, 433.
- [23] S. Aravinda, N. Shamala, P. Balaram, *Chem. Biodiversity* 2008, 5, 1238; U. S. Raghavender, S. Aravinda, N. Shamala, Kantharaju, R. Rai, P. Balaram, *J. Am. Chem. Soc.* 2009, 131, 15130; U. S. Raghavender, Kantharaju, S. Aravinda, N. Shamala, P. Balaram, *J. Am. Chem. Soc.* 2010, 132, 1075; B. Dinesh, K. Basuroy, N. Shamala, P. Balaram, *Tetrahedron* 2012, 68, 4374; K. Basuroy, V. Karuppiah, N. Shamala, P. Balaram, *Helv. Chim. Acta* 2012, 95, 2589.
- [24] T. Yamada, S. Fujioka, M. Takeda, S. Gohda, T. Murashima, Y. In, *Pept. Sci.* 2010, 47, 136; Y. In, K. Minoura, T. Ishida, S. Fujioka, M. Takeda, T. Murashima, T. Yamada, *Pept. Sci.* 2011, 48, 205.
- [25] a) T. Jeremic, A. Linden, H. Heimgartner, *Chem. Biodiversity* 2004, 1, 1730; b) T. Jeremic, A. Linden, H. Heimgartner, *Helv. Chim. Acta* 2004, 87, 3056.
- [26] a) K. Dietliker, H. Heimgartner, *Helv. Chim. Acta* 1983, 66, 262; b) J. M. Villalgordo, H. Heimgartner, *Helv. Chim. Acta* 1995, 78, 1983.
- [27] P. Wipf, Ph.D. thesis, Universität Zürich, 1987.
- [28] H.-D. Jakubke, C. Klessen, E. Berger, K. Neubert, Tetrahedron Lett. 1978, 1497.
- [29] S. Chen, J. Xu, Tetrahedron Lett. 1992, 33, 647.
- [30] J. V. Hruby, A. I. Brewster, J. A. Glasel, Proc. Natl. Acad. Sci. 1971, 68, 450.
- [31] a) C. Grathwohl, K. Wüthrich, *Biopolymers* 1976, 15, 2025; b) S. K. Sarkar, P. E. Young, C. E. Sullivan, D. A. Torchia, *Proc. Natl. Acad. Sci.* 1984, 81, 4800; c) H. Kessler, *Angew. Chem., Int. Ed.* 1982, 21, 512.
- [32] C. M. Venkatachalam, *Biopolymers* 1968, 6, 1425; P. N. Lewis, F. A. Momany, H. A. Scheraga, *Biochem. Biophys. Acta* 1973, 303, 211.
- [33] C. K. Johnson, ORTEP II, Report ORNL-5138, Oak Ridge National Laboratory, Oak Ridge, Tennessee, 1976.
- [34] J. Bernstein, R. E. Davis, L. Shimoni, N.-L. Chang, Angew. Chem., Int. Ed. 1995, 34, 1555.
- [35] C. Toniolo, G. M. Bonora, M. Crisma, E. Benedetti, A. Bavoso, B. Di Blasio, V. Pavone, C. Pedone, Int. J. Pept. Protein Res. 1983, 22, 603; E. G. Hutchinson, J. M. Thornton, Protein Sci. 1994, 3, 2207.
- [36] a) F. Formaggio, S. Pegoraro, M. Crisma, G. Valle, C. Toniolo, G. Précigoux, W. H. J. Boesten, H. E. Schoemaker, J. Kamphuis, J. Biomol. Struct. Dyn. 1993, 10, 919; b) E. Benedetti, Biopolymers 1996, 40, 3.
- [37] I. Dannecker-Dörig, A. Linden, H. Heimgartner, Coll. Czech. Chem. Commun. 2009, 74, 901.
- [38] L. A. Carpino, A. A. Abdel-Maksoud, E. M. E. Mansour, M. A. Zewail, *Tetrahedron Lett.* 2007, 48, 7407.
- [39] a) G. R. Marshall, E. E. Hodgkin, D. A. Langs, G. D. Smith, J. Zabrocki, M. T. Leplawy, Proc. Natl. Acad. Sci. 1990, 87, 487; b) C. Toniolo, M. Crisma, F. Formaggio, G. Valle, G. Cavicchioni, G. Précigoux, Biopolymers 1993, 33, 167; c) B. Di Blasio, V. Pavone, A. Lombardi, C. Pedone, E. Benedetti, Biopolymers 1993, 33, 1037; d) R. Banerjee, S. Chattopadhyay, G. Basu, Proteins 2009, 76, 184.
- [40] a) C. Toniolo, G. M. Bonora, A. Bavoso, E. Benedetti, B. Di Blasio, V. Pavone, C. Pedone, *Biopolymers* 1983, 22, 205; b) B. Pengo, F. Formaggio, M. Crisma, C. Toniolo, G. M. Bonora, Q. B. Broxterman, J. Kamphuis, M. Saviano, R. Iacovino, F. Rossi, E. Benedetti, *J. Chem. Soc., Perkin Trans.* 2 1998, 1651.
- [41] C. Toniolo, F. Formaggio, M. Crisma, G. Valle, W. H. J. Boesten, H. E. Schoemaker, J. Kamphuis, P. A. Temussi, E. L. Becker, G. Précigoux, *Tetrahedron* 1993, 49, 3641.
- [42] M. De Poli, M. De Zotti, J. Raftery, J. A. Aguilar, G. A. Morris, J. Clayden, J. Org. Chem. 2013, 78, 2248.
- [43] Y. Demizu, M. Tanaka, M. Nagano, M. Kurihara, M. Doi, T. Maruyama, H. Suemune, *Chem. Pharm. Bull.* 2007, 55, 840; Y. Demizu, Y. U. Yabuki, M. Doi, Y. Sato, M. Tanaka, M. Kurihara, *J. Pept. Sci.* 2012, 18, 466.
- [44] S. Ram, L. D. Spicer, Tetrahedron Lett. 1987, 28, 515.
- [45] A. C. T. North, D. C. Phillips, F. S. Mathews, Acta Crystallogr., Sect. A 1968, 24, 351.
- [46] G. M. Sheldrick, Acta Crystallogr., Sect. A 1990, 46, 467.

- [47] R. Miller, S. M. Gallo, H. G. Khalak, C. M. Weeks, J. Appl. Crystallogr. 1994, 27, 613.
- [48] P. van der Sluis, A. L. Spek, Acta Crystallogr., Sect. A 1990, 46, 194
- [49] A. L. Spek, Acta Crystallogr., Sect. D, 2009, 65, 148.
- [50] E. N. Maslen, A. G. Fox, M. A. O'Keefe, 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 6.1.1.1, p. 477.
- [51] R. F. Stewart, E. R. Davidson, W. T. Simpson, J. Chem. Phys. 1965, 42, 3175.
- [52] J. A. Ibers, W. C. Hamilton, Acta Crystallogr. 1964, 17, 781.
- [53] D. C. Creagh, W. J. McAuley, 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 4.2.6.8, p. 219.
- [54] D. C. Creagh, J. H. Hubbell, 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 4.2.4.3, p. 200.
- [55] G. M. Sheldrick, Acta Crystallogr., Sect. A 2008, 64, 112.

Received March 11, 2014