

Synthesis and anticancer activity of some new s-triazine derivatives

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Received: 10 December 2012 / Accepted: 13 March 2013
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Abstract New s-triazine derivatives **13a–h** were synthesized for the structure–activity relationship studies as potent anticancer agents. The prepared analogues were evaluated for their in vitro inhibitory activity against the growth of PA-1 (Ovarian cancer), A549 (Lung cancer), MCF-7 (Breast cancer), and HT-29 (Colon cancer). Tri-substituted s-triazine derivatives (**13e–h**) with morpholino group on s-triazine scaffold exhibited potent anticancer activities compared to di-substituted s-triazine derivatives. Compounds **13e–h** also showed relatively selective PA-1 and HT-29 cancer cell inhibition over other cancer cell lines. Structure–activity relationships provided useful insights in these classes of compounds and paved the way to design novel analogues with more potency.

Keywords S-triazine derivatives · Anticancer activity · Structure–activity relationships · Hit identification

Introduction

s-Triazine is a six-membered heterocyclic ring, with three nitrogens situated at 1st, 3rd and 5th positions. Its analogues, melamine, cyanuric acid and cyanuric chloride are important starting compounds for various materials with wide range of applications in textile, plastic, pharmaceuticals and rubber industries. These compounds are also used as pesticides, dyestuffs, optical bleaches, explosives and surface active agents (Bartholomew, 1996; Comins and O'Connor, 1988; Quirke, 1984; Smolin and Rapport, 1959). s-Triazine compounds have been studied extensively and are the subject of many reviews (Blotny, 2006; Giacomelli *et al.*, 2004; Sathiakumar *et al.*, 2011; Therrien, 2010). This s-triazine scaffold attracted many researchers, as its symmetrical structure facilitates to synthesize diverse set of analogues (ease of synthesis) such as 2, 4, 6-mono, di- or tri-substituted, symmetrical and nonsymmetrical compounds bearing different substituents and in particular, cyanuric chloride is the most important one in this aspect (Blotny, 2006). The s-triazine scaffold also provides the basis for the design of biologically relevant molecules with widespread application as therapeutics. For example, these compounds possess potent antiprotozoal (Klenke *et al.*, 2001), antimalarial (Manohar *et al.*, 2010; Melato *et al.*, 2008), antiviral (Mahajan *et al.*, 2009; Maarouf *et al.*, 2012; Xiong *et al.*, 2008), anticancer (Baindur *et al.*, 2005; Leftheris *et al.*, 2004; Manohar *et al.*, 2010; Moon *et al.*, 2002; Sun *et al.*, 2010; Zheng *et al.*, 2007), antimicrobial (Patel *et al.*, 2012a, b, c; Raval *et al.*, 2011; Singh *et al.*, 2011; Srinivas *et al.*, 2005, 2006; Desai *et al.*, 2008), antituberculosis (Patel *et al.*, 2012a; Sunduru *et al.*, 2010) etc. Recently, the s-triazine derivatives were extensively investigated for anticancer activity with a particular target to mTOR/PI3K pathway (Menaar *et al.*, 2009; Poulsen *et al.*, 2012; Tanneeru *et al.*, 2012; Venkatesan *et al.*, 2010;

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Verheijen *et al.*, 2010; Zask *et al.*, 2010; Zhang *et al.*, 2011). Some of the compounds possessing s-triazine scaffold reported in the literature are presented in Fig. 1. Our continuous interest in the field of bioactive heterocyclic compounds (Kumar *et al.*, 2010, 2011; Ravinder *et al.*, 2012; Rao *et al.*, 2011; Srinivas *et al.*, 2009), we have got interest in new s-triazine derivatives as cancer therapeutics. In the context of identification of new chemical entities (NCEs) for cancer therapy, we have selected s-triazine as core scaffold as many reports indicate its significance (Fig. 1) and structural modifications were made at three positions (1st, 3rd and 5th positions) with various pharmacophores. In this article, synthesis of new s-triazine derivatives is presented. All the synthesized compounds were screened against four cancer cell lines and discussed to develop the structure–activity relationships (SAR) of this series. Among all, compounds **13e–h** exhibited potent inhibitory activity against ovarian (PA-1) and colon (HT-29) cancer cell lines

Chemistry

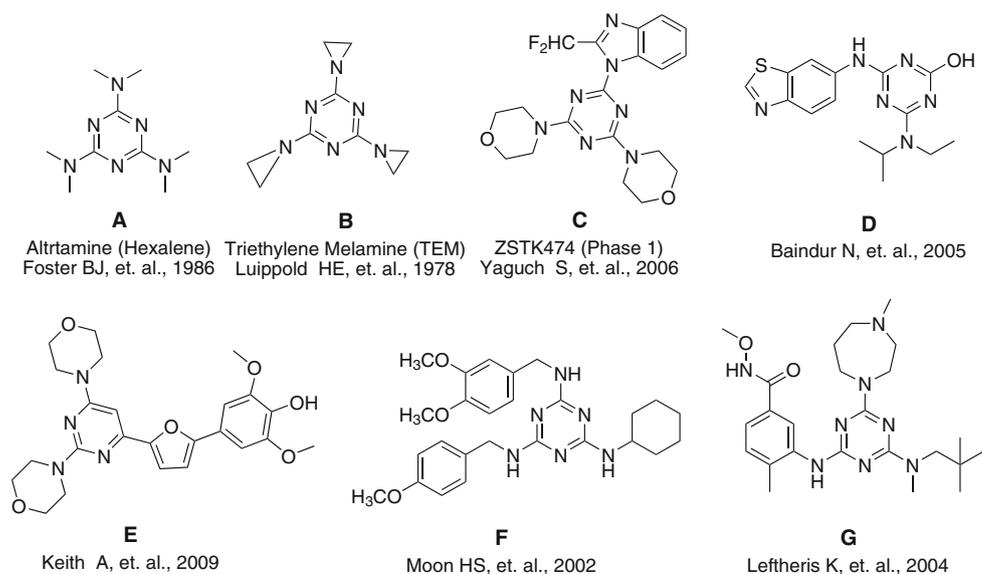
4-(Benzo[d]oxazol-2-yl) aniline (**3**) was prepared from the simple condensation of *o*-amino phenol (**1**) with *p*-amino benzoic acid in the presence of polyphosphoric acid at 200 °C (Scheme 1) (Chua *et al.*, 1999). Nucleophilic substitution of one of the chlorines of the cyanuric acid (**4**) with substituted anilines (**5–7** & **11**) in the presence of diisopropylethylamine (DIPEA) yielded mono-substituted triazine derivatives **8–10** & **12**, respectively, in good yields (mono substitution w.r.t. cyanuric chloride) and the synthetic scheme is presented in Scheme 2 (Zheng *et al.*, 2007). Compounds **8–10** and **12** were reported in the literature (Hunter *et al.*, 1994; McKay *et al.*, 2006; Maga *et al.*, 2011). Synthesis of compounds **13a–d** involves a

coupling reaction between mono-substituted s-triazine derivatives (**8–10** & **12**) and one equivalent of compound **3** in the presence of K_2CO_3 (Scheme 3). Compounds **13e–h** were synthesized by the replacement of third chlorine atom of di-substituted s-triazine derivatives (**13a–d**) in the presence of K_2CO_3 (Scheme 4). All the intermediates and final compounds were purified by column chromatography and characterized by spectroscopic techniques.

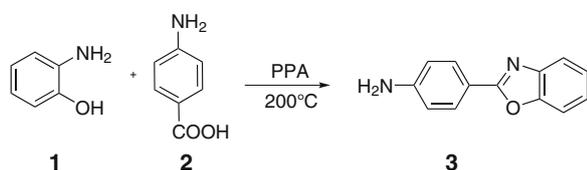
Biology

The anticancer activities of all the synthesized compounds were evaluated by MTT [3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide] method using PA-1 (Ovarian cancer), A549 (Lung cancer), MCF-7 (Breast cancer) and HT-29 (Colon cancer) cell lines. The effect of compounds on the cell viability for each cell line after exposure to different concentrations were depicted in the Fig. 2 and their respective IC_{50} values are presented in Table 1. Synthesized compounds can be classified into two categories based on the presence of the chlorine atom and morpholino group *i.e.* di-substituted s-triazine derivatives (**13a–d**) and tri-substituted s-triazine derivatives (**13e–h**), respectively. Among the two categories, tri-substituted s-triazine derivatives were more active against in all tested human cancer cell lines than di-substituted s-triazine derivatives which suggest the key role of morpholino group for activities. In case of PA-1 cell lines, among **13e–h**, compounds **13e–g** are more potent and are equally potent as Doxorubicin. This indicates the importance of substituted aniline groups on s-triazine compared to benzylamine group. The activity difference in compounds **13e** and **13h** is more than one magnitude (**13e**: $0.61 \pm 0.16 \mu M$ and **13h**: $10.5 \pm 0.51 \mu M$) and this attributes to bridge between

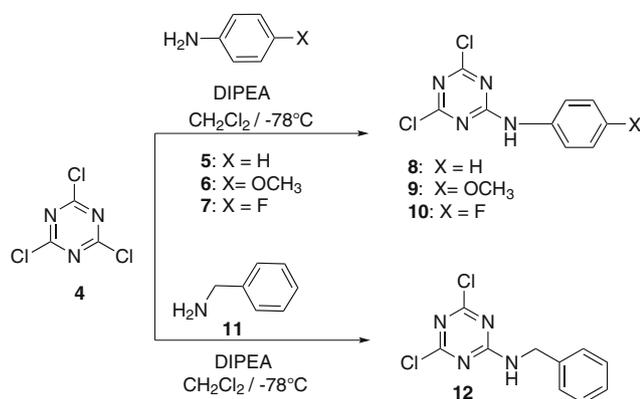
Fig. 1 List of potent anticancer compounds possessing s-triazine scaffold



s-triazine and substituted group i.e. –NH– and –NH–CH₂– groups, respectively. The di-substituted s-triazine derivatives (**13a–d**) were inactive against A549 (lung cancer) cell lines, whereas tri-substituted derivatives have exhibited relatively better inhibitory activity. Among the compounds **13e–h**, compound **13f** is relatively potent with IC₅₀ at 12.9 ± 2.43 μM and other compounds (**13e**, **13g–h**) exhibited twofolds less inhibitory activity. In case of MCF-7 cell lines, compounds **13f** and **13h** are more potent than reference compound, doxorubicin. Among the all compounds in the series, **13h** is most potent. Compounds **13e** and **13g** exhibited moderate anticancer activity against MCF-7 cell lines. Compounds **13e–h** exhibited potent anticancer activity against HT-29 (Colon cancer) cell lines

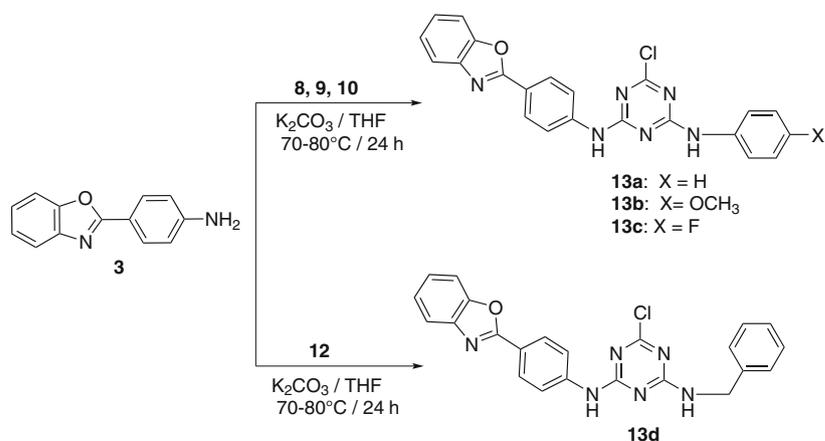


Scheme 1 Synthesis of 4-(benzo[d]oxazol-2-yl) aniline

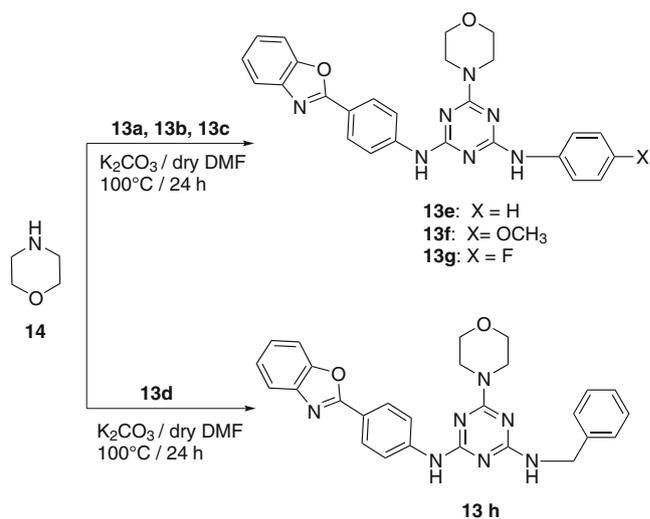


Scheme 2 Synthesis of compounds **8–10** & **12**

Scheme 3 Synthesis of compounds **13a–d**



(Fig. 1) and these are more potent than reference compound, doxorubicin. Among the **13e–h**, compound **13e–g** are more potent with IC₅₀ of 0.27 ± 0.02, 0.14 ± 0.04 and 0.31 ± 0.15 μM, respectively and compound **13h** is relatively less potent with IC₅₀ at 0.92 ± 0.51 μM (Table 1). The disparity in magnitude of activity within **13e–h** is due to the structural dissimilarity i.e. the presence of anilino and substituted anilino groups in case of **13e–g** and the presence of benzylamino group in case of **13h**. In summary, compounds **13e–h** exhibited selective cytotoxicity in HT-29 & PA-1 when compared to other cell line (A549 and MCF-7) and compound **13f** is found to be most potent in respective cell lines. Structure–activity relationships (SAR) reveals that the presence of morpholino group, substituted aniline groups along with benzoxazole moiety on s-triazine ring are essential to exhibit potent anticancer activity. Structural modifications on anilino group and benzoxazole group can provide more insights and give better hits/leads and focus of our future work is on structural modifications on these classes of compounds to



Scheme 4 Synthesis of compounds **13e–h**

Table 1 Chemical structures of compounds **13a–h** and their inhibitory effects on the growth of tumour cell lines

Comp- ound	R ₁	R ₂	R ₃	IC ₅₀ (μM)			
				PA-1 ^a	A549 ^b	MCF-7 ^c	HT-29 ^d
13a			Cl	169±2.78	225±5.64	55.2±3.21	32.6±2.65
13b			Cl	172±6.21	278±3.98	269.7±6.98	139±5.47
13c			Cl	186±3.21	254±6.21	68.4±3.41	47±2.54
13d			Cl	195±5.43	296±2.45	66.1±2.48	44±3.59
13e				0.61±0.16	22.0±1.11	10.56±1.56	0.27±0.02
13f				0.25±0.19	12.9±2.43	6.48±0.21	0.14±0.04
13g				0.45±0.21	26.7±0.67	12.39±1.38	0.31±0.15
13h				10.5±0.51	23.4± 1.70	2.46±0.15	0.92±0.51
Doxorubicin ((Reference compound))				0.64±0.13	1.88±0.56	10.9±1.76	1.76±0.23

^a Ovarian cancer cells; ^b Lung cancer cells; ^c Breast cancer cells, ^d Colon cancer cells

improve the anticancer activity and on understanding of this activity through mechanistic studies.

Conclusions

In conclusion, two series of s-triazine derivatives *i.e.* di-substituted (**13a–d**) and tri-substituted derivatives (**13e–h**) were synthesized as anticancer agents. All the synthesized compounds were screened against four cancer cell lines viz. PA-1 (Ovarian cancer), A549 (Lung cancer), MCF-7 (Breast cancer) and HT-29 (Colon cancer). This preliminary study demonstrated that tri-substituted derivatives **13e–h** exhibited potent anticancer activity over the di-substituted derivatives. Among these derivatives, compound **13f** is the most potent in respective cell lines. Structure–activity relationships in these classes of compounds laid foundations to develop more potent compounds as leads for cancer chemotherapy.

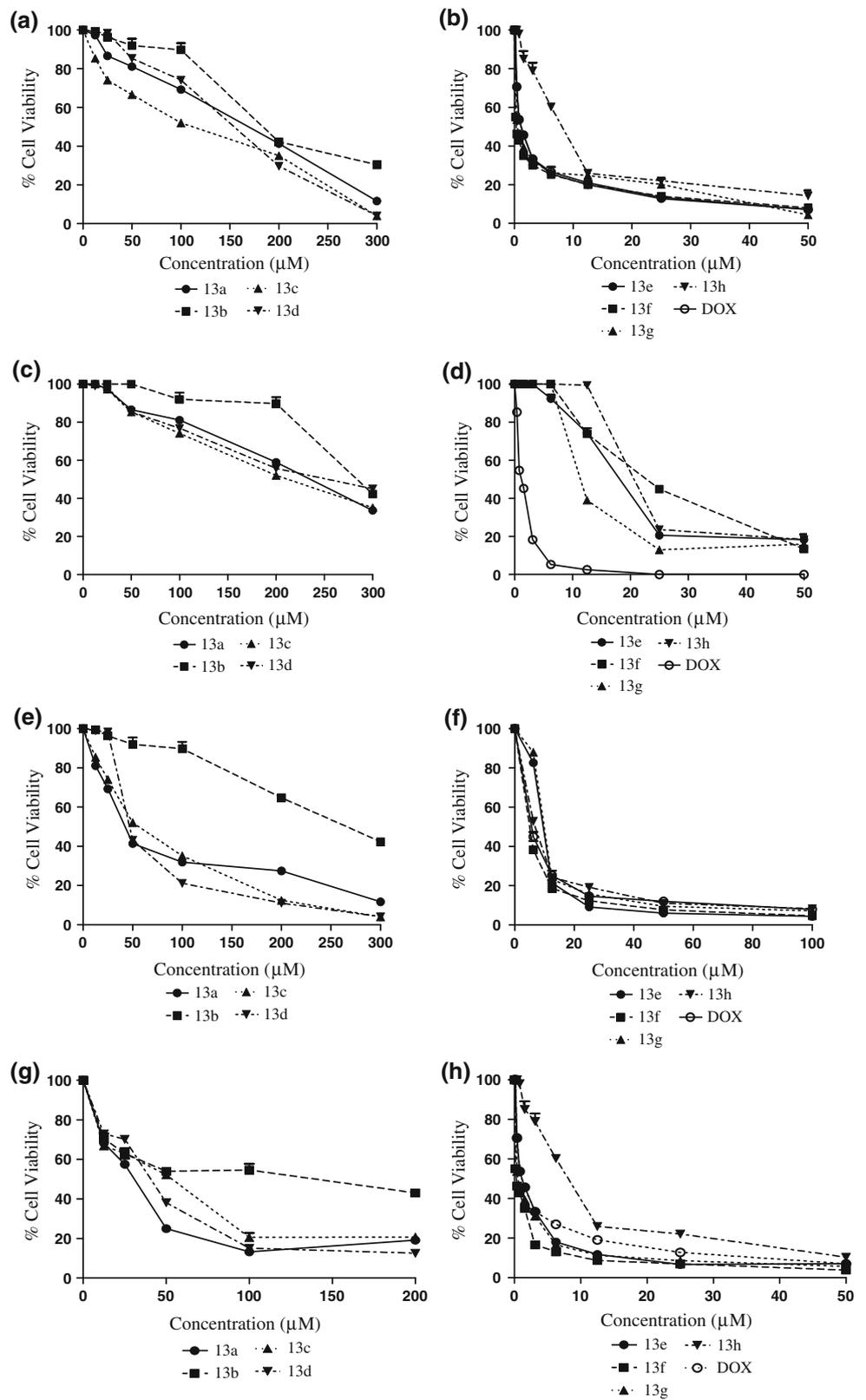
Experimental

Chemistry

Synthesis of 4-(benzo[d]oxazol-2-yl) aniline (3) (Chua *et al.*, 1999; Shi *et al.*, 1996)

To a stirring solution of polyphosphoric acid (PPA; 85 g), 2-Aminothiophenol **1** (5.78 g, 0.053 mol), para amino benzoic acid **2** (7.26 g, 0.053 mol) were added, heated at 220 °C for 4 h. After confirming the reaction completion by TLC (Thin Layer Chromatography), it was cooled, and poured into ice-cold 10 % aqueous sodium carbonate. The solid product was collected, washed with water and recrystallized with methanol water. Yield: 65 %; mp 180–183 °C; ¹H NMR (CDCl₃, 300 MHz): δ 8.07–8.03 (d, 2H, *J* 8.3 Hz), 7.73–7.67 (m, 1H), 7.55–7.49 (m, 1H), 7.31–7.27 (m, 2H), 6.78–6.74 (d, 2H, *J* 9.0 Hz) and 4.05 (brs, 2H); ESI-MS: *m/z* 211 (M + H).

Fig. 2 Dose response of compounds (**13a–13h**) against **a, b** PA-1, **c, d** A549, **e, f** MCF-7, **g, h** HT-29 cancer cell lines



General procedure for synthesis of mono-substituted cyanuric chloride derivatives (Zheng et al., 2007)

Compounds **8**, **9**, **10** and **12** were synthesized using cyanuric chloride **4** (10 g, 0.054 mol) in dichloromethane (DCM), *N,N*-Diisopropyl ethylamine (DIPEA). DIPEA (9.26 ml, 0.054 mol) was added slowly drop wise to the reaction mixture at $-78\text{ }^{\circ}\text{C}$ (maintained by using acetone and dry ice) for 10 min then it was allowed to stir for 10 min. Aniline **5** (4.94 ml, 0.054 mol) or 4-methoxyaniline **6** (6.66 g, 0.054 mol) or 4-fluoroaniline **7** (5.19 ml, 0.054 mol) or benzyl amine **11** (5.9 ml, 0.054 mol) were added respectively and allowed to stir for 15 min. The reaction was monitored by TLC. After completing the reaction, the reaction mixture was filtered, washed with water, extracted with chloroform, the organic layer was dried over Na_2SO_4 and then concentrated to get the desired mono-substituted s-triazines in excellent yield. Compounds **8–10** and **12** were reported in the literature (Hunter et al., 1994; McKay et al., 2006; Maga et al., 2011).

General procedure for synthesis of di-substituted s-triazines

Di-substituted cyanuric chloride derivatives (**13a–d**), were synthesized using mono-substituted cyanuric chloride **8** (1 eq), **9** (1 eq), **10** (1 eq) and **12** (1 eq) in dry THF (30 ml), K_2CO_3 (2 eq). The reaction mixture was allowed to stir for 5 min then the benzoxazolephenylamine **3** (1 eq) was added and refluxed for 24 h at $70\text{--}80\text{ }^{\circ}\text{C}$. After confirming the reaction completion by TLC, K_2CO_3 was decanted and the THF was removed, water was added and extracted with EtOAc. The organic layer was dried over Na_2SO_4 , concentrated and purified to give the desired di-substituted s-triazines.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-6-chloro-*N*⁴-phenyl-1,3,5-triazine-2,4-diamine (**13a**): Yield: 80 %; mp: $263\text{--}265\text{ }^{\circ}\text{C}$; IR (KBr): 3,273, 3,109, 1,610, 1,585, 1,243 and 989 cm^{-1} ; ¹H NMR (300 MHz, DMSO-*d*₆): δ 10.32 (brs, 1H), 10.03 (brs, 1H) 8.11 (m, 2H), 7.93 (m, 2H), 7.74–7.68 (m, 2H), 7.57 (m, 1H) and 7.32 (m, 4H); ¹³C NMR (75 MHz, DMSO-*d*₆): δ 162.1, 150.1, 150.0, 142.0, 141.5, 128.8, 128.5, 127.8, 125.0, 124.7, 120.6, 120.4, 119.4, 110.8, 110.7; ESI-MS: *m/z* 415 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for $\text{C}_{22}\text{H}_{16}\text{ClN}_6\text{O}$ [M + H]⁺ 415.8471, found 415.8342.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-6-chloro-*N*⁴-(4-methoxyphenyl)-1,3,5-triazine-2,4-diamine (**13b**): Yield: 82 %; mp: $275\text{--}276\text{ }^{\circ}\text{C}$; IR (KBr): 3,278, 3,185, 2,835, 1,232 and 992 cm^{-1} ; ¹H NMR (300 MHz, DMSO-*d*₆): δ 10.36 (brs,

1H), 10.00 (brs, 1H), 8.13–8.07 (m, 3H), 7.90 (s, 1H), 7.70–7.55 (m, 4H), 7.5 (s, 1H), 7.35 (m, 2H), 6.90 (m, 2H) and 3.82 (s, 3H); ¹³C NMR (75 MHz, CDCl₃ + DMSO-*d*₆): δ 165.1, 164.3, 164.3, 155.0, 151.9, 142.3, 142.2, 133.3, 133.2, 127.1, 123.4, 122.2, 121.9, 119.8, 119.7, 113.8, 79.6, 79.3, 79.1, 78.7; ESI-MS: *m/z* 445 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for $\text{C}_{23}\text{H}_{18}\text{ClN}_6\text{O}_2$ [M + H]⁺ 445.1102, found 445.1112.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-6-chloro-*N*⁴-(4-fluorophenyl)-1,3,5-triazine-2,4-diamine (**13c**): Yield: 77 %; mp: $268\text{--}269\text{ }^{\circ}\text{C}$; IR (KBr): 3,384, 3,111, 1,508, 1,242 and 995 cm^{-1} ; ¹H NMR (300 MHz, DMSO-*d*₆): δ 9.42 (brs, 1H), 9.16 (brs, 1H), 8.18–8.12 (m, 4H), 7.69–7.67 (m, 2H), 7.59–7.57 (m, 1H), 7.32–7.29 (m, 3H), 7.05–7.01 (t, 2H), *J* 8.5, 17.0 Hz); ¹³C NMR (75 MHz, CDCl₃ + DMSO-*d*₆): δ 162.1, 160.5, 148.5, 140.1, 132.8, 126.0, 122.9, 122.6, 121.5, 119.2, 118.5, 117.6, 113.3, 108.6, 93.9; ESI-MS: *m/z* 433 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for $\text{C}_{22}\text{H}_{15}\text{ClFN}_6\text{O}$ [M + H]⁺ 433.8376, found 433.8362.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-*N*⁴-benzyl-6-chloro-1,3,5-triazine-2,4-diamine (**13d**): Yield: 76 %; mp: $278\text{--}279\text{ }^{\circ}\text{C}$; IR (KBr): 3,268, 3,111, 2,993, 1,240 and 987 cm^{-1} ; ¹H NMR (75 MHz, CDCl₃): δ 8.30–8.26 (m, 1H), 8.22–8.17 (m, 2H), 7.78–7.74 (m, 3H), 7.58–7.56 (m, 1H), 7.40–7.39 (m, 2H), 7.37–7.31 (m, 4H), 6.05 (brs, 1H), 5.77 (brs, 1H) and 4.72–4.66 (t, 2H, *J* 5.2, 10.5 Hz); ¹³C NMR (75 MHz, CDCl₃ + DMSO-*d*₆): δ 148.9, 140.9, 140.6, 126.9, 126.8, 126.4, 125.6, 125.5, 118.4, 118.3, 118.0, 117.9, 117.8, 109.1, 109.0; ESI-MS: *m/z* 429 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for $\text{C}_{23}\text{H}_{18}\text{ClN}_6\text{O}$ [M + H]⁺ 429.8737, found 429.8725.

General procedure for synthesis of tri-substituted s-triazines

Tri-substituted s-triazine derivatives (**13e–h**) were synthesized from di-substituted cyanuric chloride derivatives **13a–d** (1 eq) in dry DMF (20 ml), K_2CO_3 (2 eq). The reaction mixture was allowed to stir for 5 min then morpholine **14** (1 eq) was added and stirred for 24 h at $100\text{ }^{\circ}\text{C}$. The reaction was monitored by TLC. After completion of the reaction, the mixture was poured into ice-cold water, extracted with ethylacetate and dried over Na_2SO_4 . Solvent was evaporated and solid was purified to get desired product in good yield.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-6-morpholino-*N*⁴-phenyl-1,3,5-triazine-2,4-diamine (**13e**): Yield: 82 % mp: $259\text{--}260\text{ }^{\circ}\text{C}$; IR (KBr): 3,402, 3,299, 2,925–2,855 and $1,242\text{ cm}^{-1}$; ¹H NMR (300 MHz, DMSO-*d*₆): δ 9.08 (brs,

1H), 8.67 (brs, 1H), 8.14–8.11 (d, 2H, *J* 8.6 Hz), 7.98–7.96 (d, 2H, *J* 8.4 Hz), 7.72–7.70 (t, 3H, *J* 7.9, 5.3 Hz), 7.61–7.58 (t, 1H, *J* 7.1, 14.3 Hz), 7.35–7.27 (m, 4H), 7.03–6.98 (m, 1H), 3.84 (d, 4H, *J* 3.2 Hz), 3.75 (d, 4H, *J* 3.3 Hz); ¹³C NMR (75 MHz, CDCl₃ + DMSO-d₆): δ 164.5, 163.9, 163.9, 162.4, 150.0, 143.7, 141.6, 139.8, 128.3, 127.8, 124.8, 124.6, 122.1, 122.0, 119.5, 119.3, 118.9, 10.6, 65.9, 43.4; ESI-MS: *m/z* 466 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for C₂₆H₂₄N₇O₂ [M + H]⁺ 466.5065, found 466.5023.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-*N*⁴-(4-methoxyphenyl)-6-morpholino-1,3,5-triazine-2,4-diamine (**13f**): Yield: 80 %; mp: 252–253 °C; IR (KBr): 3,400, 3,307, 2,963, 2,922, 2,855 and 1,227 cm⁻¹; ¹H NMR (300 MHz, DMSO-d₆): δ 9.52 (brs, 1H), 9.08 (brs, 1H), 8.04–7.99 (d, 2H, *J* 8.3 Hz), 7.97–7.92 (m, 2H), 7.72–7.66 (m, 2H), 7.53–7.50 (d, 2H, *J* 8.3 Hz), 7.35–7.26 (m, 2H), 3.8 (m, 7H) and 3.7 (d, 4H, *J* 4.5 Hz); ¹³C NMR (75 MHz, CDCl₃ + DMSO-d₆): δ 164.6, 163.8, 154.4, 151.4, 141.6, 132.6, 126.6, 122.8, 121.6, 121.3, 119.2, 113.3, 79.0, 78.8, 78.6, 78.1, 65.9, 54.9, 43.3; ESI-MS: *m/z* 496 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for C₂₇H₂₆N₇O₃ [M + H]⁺ 496.2325, found 466.2091.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-*N*⁴-(4-fluorophenyl)-6-morpholino-1,3,5-triazine-2,4-diamine (**13g**): Yield: 84 %; mp: 246–247 °C; IR (KBr): 3,424, 3,297, 2,924–2,856, 1,508 and 1,243 cm⁻¹; ¹H NMR (75 MHz, DMSO-d₆): δ 8.22–8.19 (d, 2H, *J* 8.3 Hz), 7.75–7.72 (m, 3H), 7.58–55 (m, 1H), 7.53–7.48 (m, 2H), 7.37–7.31 (m, 2H), 7.08–7.02 (m, 3H), 6.99 (brs, 1H) 6.79 (brs, 1H) 3.84 (d, 4H, *J* 4.5 Hz) and 3.76 (d, 4H, *J* 4.5 Hz); ¹³C NMR (300 MHz, CDCl₃ + DMSO-d₆): δ 164.6, 164.0, 163.9, 151.4, 143.8, 141.7, 139.8, 134.8, 128.3, 126.7, 123.2, 122.0, 121.9, 121.4, 120.1, 119.5, 118.3, 110.9, 65.39, 43.3; ESI-MS: *m/z* 484 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for C₂₆H₂₃FN₇O₂ [M + H]⁺ 484.1970, found 484.1891.

*N*²-(4-(benzo[d]oxazol-2-yl)phenyl)-*N*⁴-benzyl-6-morpholino-1,3,5-triazine-2,4-diamine (**13h**): Yield: 78 %; mp: 240–242 °C; IR (KBr): 3,403, 3,296, 2,924, 2,853 and 1,241 cm⁻¹; ¹H NMR (300 MHz, DMSO-d₆): δ 9.33–9.29 (brs, 2H), 8.19–8.16 (d, 2H, *J* 8.3 Hz), 7.71–7.68 (d, 2H, *J* 8.3 Hz), 7.52–7.49 (d, 2H), 7.29–7.27 (m, 2H), 7.1 (m, 3H), 7.09–7.06 (m, 1H), 6.86 (s, 1H), 6.66–6.62 (d, 2H, *J* 7.9 Hz), 3.78–3.76 (d, 4H, *J* 4.5 Hz), 3.72–3.70 (d, 4H, *J* 4.9 Hz) and 3.55 (s, 2H); ¹³C NMR (75 MHz, CDCl₃ + DMSO-d₆): δ 136.3, 163.1, 150.0, 138.6, 126.3, 125.6, 125.2, 125.0, 121.1, 119.9, 117.6, 117.5, 64.4, 42.0, 41.6; ESI-MS: *m/z* 480 [M + H]⁺; HRMS (ESI) *m/z* Calcd. for C₂₇H₂₆N₇O₂ [M + H]⁺ 480.5331, found 480.2103.

Biological evaluation

Materials

PA-1 (Ovarian cancer), A549 (Lung cancer) cell lines MCF-7 (Breast cancer) and HT-29 (Colon cancer) cell line were obtained from the National center for Cell science (NCCS), Pune, India. MEM, DMEM, RPMI, MTT [3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide], Trypsin–EDTA were purchased from Sigma Chemicals Co (st. Louis, MO), Fetal bovine serum were purchased from Gibco, USA, 96-well flat bottom tissue culture plates were purchased from Tarson.

Maintenance of cell lines

PA-1 (Ovarian cancer) A549 (Lung cancer) cell line were grown as adherent in MEM medium, MCF-7 (Breast cancer) cell line was grown as adherent in DMEM medium and HT-29 (Colon cancer) cell line was grown as adherent in RPMI medium supplemented with 10 % fetal bovine serum, 100 µg/ml penicillin, 200 µg/ml streptomycin, 2 mM L-glutamine, and culture was maintained in a humidified atmosphere with 5 % CO₂.

Preparation of samples for cytotoxicity

Stock solution for compounds **13a–13d** of 30 mM and compounds **13e–13f** of 10 mM stock solution in DMSO were prepared, from the above stock various dilutions were made with sterile PBS to get required concentration.

Cytotoxicity screening using MTT assay

MTT [3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide] assay is a standard colourimetric assay for measuring cellular proliferation. MTT is a tetrazolium salt, which is yellow in colour and is photosensitive. MTT [3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide] is taken by the living cells and reduced by a mitochondrial dehydrogenase enzyme to a purple formazan product that is impermeable to the cell membrane. Solubilisation with solvents like DMSO leads to liberation of product and amount of purple formazan product is directly related to the cell viability. 1 × 10⁴ Cells (counted by Trypan blue exclusion dye method) in 96-well plates were incubated with compounds and standard Cisplatin with series of concentrations for 24 h at 37 °C in MEM with 10 % FBS medium. Then, the above media was replaced with 90 µl of fresh serum free media and 10 µl of MTT reagent (5 mg/ml) and plates were incubated at 37 °C for 4 h, there after the above media was replaced with 200 µl of DMSO and incubated at 37 °C for 10 min. The

absorbance at 570 nm was measured on a spectrophotometer (spectra max, Molecular devices) IC₅₀ values were determined from plot: % cell viability (from control) versus concentration.

Acknowledgments Authors thank Director, CSIR-IICT and Project Director, NIPER-Hyderabad for constant support and encouragement. BJK, BRK and SS thank Department of Pharmaceuticals for fellowship.

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