# Isocytosine $H_2$ -receptor histamine antagonists II. Synthesis and evaluation of biological activity at histamine $H_1$ - and $H_2$ -receptors of 5-(heterocyclyl)methylisocytosines

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Summary — A series of 2-[2-(5-methyl-4-imidazolylmethylthio)ethylamino]-4-pyrimidones was synthesised based on oxmetidine 2, in which the methylenedioxyphenyl group of 2 was replaced by a heterocyclic ring. Good H<sub>2</sub>-receptor antagonist activity was retained over a range of basic and neutral heterocyclic substituents. Replacement of the 5-methyl-4-imidazolyl ring in selected compounds with 2-thiazolyl and, particularly, 3-bromo-2-pyridyl rings gave a series of compounds which have both H<sub>1</sub>- and H<sub>2</sub>-receptor histamine antagonist activities. Some structure—activity and structure—toxicity correlations are discussed. Compound 6d, 2-[2-(5-methyl-4-imidazolylmethylthio)ethylamino]-5-(6-methylpyrid-3-yl)-4-pyrimidone, has the most favourable combination of properties for H<sub>2</sub>-antagonism and has been evaluated further as an anti-secretory agent.

Résumé — Les isocytosines comme antagonistes des récepteurs histamine- $H_2$  II. Synthèse et évaluation de l'activité biologique des (hétérylméthyl)-5 isocytosine sur les récepteurs histamine- $H_1$  et  $-H_2$ . On a synthétisé une série de [(méthyl-5 imidazolyl-4 méthylthio)-2 éthylamino]-2 pyrimidones-4 basées sur l'oxmétidine 2, dans laquelle le groupe méthylènedioxyphényle est remplacé par un noyau hétérocyclique. Une gamme de groupes hétérocycliques basiques et neutres possèdent tous une bonne activité vis-à-vis des récepteurs histaminiques- $H_2$ . Si l'on remplace dans certains produits le noyau méthyl-5 imidazolyl-4 par le noyau thiazolyl-2, et en particulier, bromo-3 pyridyl-2, on obtient une série de composés qui possèdent à la fois une activité antagoniste envers les récepteurs histaminiques  $H_1$  et  $H_2$ . Certaines corrélations structure—activité et structure—toxicité sont discutées. Le produit 6d, [(méthyl-5 imidazolyl-4 méthylthio)-2] ethylamino-2 (méthyl-6 pyridyl-3)-5 pyrimidone-4, possède la combinaison de propriétés la plus favorable pour l'antagonisme des récepteurs histaminiques- $H_2$  et a de plus été évalué comme agent anti-sécrétoire.

H2-receptor antagonists / combined H1- and H2-receptor antagonists / imidazoles / thiazoles / bromopyridines / 2-amino-pyrimidin-4-ones (isocytosines)

# Introduction

The introduction of H<sub>2</sub>-receptor histamine antagonists has revolutionised drug therapy for hypersecretory states. Cimetidine 1 (Tagamet®) was the first clinically effective agent in this class and was introduced in 1976. Recently we described [1] a series of H<sub>2</sub>-receptor antagonists in which the cyanoguanidine group of cimetidine was replaced by a 2-amino-4-pyrimidone (isocytosine) ring. Manipulation of substituents in the pyrimidone ring led to the clinically effective agent oxmetidine 2, containing a 5-(3,4-methylenedioxy)benzyl group.



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Because of the excellent  $H_2$ -antagonist activity of oxmetidine and its analogues [1] together with the possibility that the relatively high lipophilicity of oxmetidine (oxmetidine octanol—water partition P at pH 8.5 and 37.5°C = 130; cimetidine = 2.5 [2]) might lead to solubility problems in injectable drug administration, we describe in this paper the replacement of the methylenedioxyphenyl moiety with heterocyclic rings. Reduced lipophilicity and potentially increased solubility is achieved by the nature of the heterocycle itself and/or by virtue of it containing an additional basic centre for protonation at physiological pH.

In addition, we describe the replacement of the imidazole ring in the side chain of these new molecules, with alternative heterocyclic rings as previously described in the cyanoguanidine series [3]. This leads to compounds with an interesting combination of histamine  $H_2$ - and  $H_1$ -antagonist activities.

### Chemistry

The general synthetic strategy for the preparation of the desired compounds was similar to that previously described for the preparation of oxmetidine 2 [1], although some improvements were incorporated and modifications were necessary for specific compounds. The synthetic sequences for all the compounds are outlined in Schemes 1-3 and described below. Physical properties and analytical data



Sequence used for precursors of compounds 3a-e and 3g.



Scheme 1. Preparation of ethyl  $\beta$ -(heterocyclyl)propionates.



(Compounds 6a-c, 6e-g, 7d-e, 8a-d) Identities of Het and R as shown in Table V. Scheme 2. Preparation of isocytosines (first route).



(Compounds 6d, 6h, 7a-b, 9-11)

Identities of Het and R as shown in Table V.

Scheme 3. Preparation of isocytosines (second route).

of the new compounds prepared are outlined in Tables I-IV.

A two-carbon homologation of the heterocyclic aldehydes (Het-CHO in Scheme 1), using malonic acid gave  $\beta$ -substituted acrylic acids which, in the majority of cases, were esterified and reduced to the appropriate ethyl 3-(heterocyclyl) propionates (Scheme 1a). For the 2-thienyl analogue, reduction of the propionate ester was extremely slow, even at elevated temperatures, so it was preferable to carry out the reduction at the acrylic acid stage [4] and esterify the propionic acid (Scheme 1b). The acid-labile  $\beta$ -(2-furyl)-acrylic acid was esterified under basic conditions using the method of Stodola [5]. Attempted reduction of ethyl  $\beta$ -(2-furyl)acrylate with the usual 5% palladium catalyst

# Table I. — 2-Thiouracils.



Compd. No.ª	R	Molecular formula <sup>b</sup>	Crystallisation solvent	mp (°C)	Yield¢ (%)
 3a	4-pyridyl	C10H9N3OS	precipitated washed with EtOH	320—324 (dec.)	48
3b	3-pyridyl	C <sub>10</sub> H <sub>9</sub> N <sub>3</sub> OS	MeCOOH/H <sub>2</sub> O	271–274 (dec.)	47
3c	2-pyridyl	C10H9N3OS	MeCOOH/H <sub>2</sub> O	262—267 (dec.)	29
3d	6-methyl-3-pyridyl	C <sub>11</sub> H <sub>11</sub> N <sub>3</sub> OS	MeCOOH/H <sub>2</sub> O	240-241	47
3e	3-quinolinyl	$C_{14}H_{11}N_3OS$	MeCOOH/H <sub>2</sub> O	281—286 (dec.)	31
3f	2-thienyl	$C_9H_8N_2OS_2^d$	EtOH	213-215	38
3g	2-thiazolyl	$C_8H_7N_3OS_2$	МеСООН	275—280 (dec.)	29

<sup>a</sup>Numbers correspond to compounds in Table II. Thus, for example, **3a** is the precursor of compound **4a** in Table II. <sup>b</sup>Problems were encountered in obtaining satisfactory elemental analyses due to residual solvent (glacial acetic acid). Identity and purity of the compounds were checked by NMR and thin—layer chromatography and they were then methylated to give compounds of Table II which were fully analysed. <sup>c</sup>Yields from substituted ethyl propionates. <sup>d</sup>Anal. C<sub>9</sub>H<sub>8</sub>N<sub>2</sub>OS<sub>2</sub> (C, H, N, S).

Table II. — 2-Methylthio-4-pyrimidones.



Compd. No.	R	Molecular formula <sup>a</sup>	Crystallisation solvent	mp (°C)	Yield <sup>b</sup> (%)
4a	4-pyridyl	C <sub>11</sub> H <sub>11</sub> N <sub>3</sub> OS	EtOH/H <sub>2</sub> O	174—175	79
4b	3-pyridyl	$C_{11}H_{11}N_3OS$	EtOH/MeCOOH	247249	44
4c	2-pyridyl	C <sub>11</sub> H <sub>11</sub> N <sub>3</sub> OS·HI	EtOH	195—197.5	69
4d	6-methyl-3-pyridyl	C <sub>12</sub> H <sub>13</sub> N <sub>3</sub> OS	EtOH/MeCOOH	197—198.5	59
<b>4</b> e	3-quinolinyl	C <sub>15</sub> H <sub>13</sub> N <sub>3</sub> OS	EtOH	215.5-218	75
<b>4</b> f	2-thienyl	$C_{10}H_{10}N_2OS_2$	EtOH	170.5171.5	89
<b>4</b> g	2-thiazolyl	C9H9N3OS2	EtOH	181-182.5	77

\*All micro-analytical data for C, H, N and S to within  $\pm 0.4\%$  of theory. >Yields from 2-thiouracils in Table I.



Compd. No.	R	Molecular formula <sup>a</sup>	Crystallisation solvent	mp (°C)	Yield <sup>b</sup> (%)
5a	4-pyridyl	$C_{10}H_9N_5O_3$	MeCOOH/H <sub>2</sub> O	240—243	45
5b	3-pyridyl	$C_{10}H_9N_5O_3$	MeCOOH/H <sub>2</sub> O	230.5–232 (dec.)	39
5d	6-methyl-3-pyridyl	$C_{11}H_{11}N_5O_3$	precipitated from alkaline solution with MeCOOH at pH = 5	229—231	33
5h	2-furyl	$C_9H_8N_4O_4$	MeCOOH	183—184	22

\*Micro-analytical data for C, H and N to within  $\pm 0.4\%$  of theory. <code>bYields</code> from substituted ethyl propionate.

Table	IV	Chemical	characteristics	of	isocytosines	in	Table	V.
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Compd. No.	Molecular formula	Crystallisation solvent	mp (°C)	Yield <sup>a</sup> (%)	Route by scheme
	CrattionNaOSi3 HClb	MeOH/EtOH	228 233	40	2
04 6h	C17H2014605 3 HCl	MeOH/H <sub>2</sub> O	220-255	69	2
60	$C_{17}H_{20}N_{8}OS^{-3}HCl:0.5H_{2}O$	EtOH/H <sub>2</sub> O	207-210	36	2
6d	$C_{19}H_{29}N_{e}OS^{-3}HCl$	MeOH/EtOH	209-212	67	3
6e	CyrHysNeOS 2 8 HCl-1.2 HyO	EtOH/H <sub>2</sub> O	184-189	68	2
6f	$C_{16}H_{19}N_5OS_2\cdot 2$ HBr	EtOH/MeCN	199—203	68	2
6g	$C_{15}H_{18}N_6OS_2\cdot 3$ HC	EtOH/H <sub>2</sub> O	201-205	34	2
6h	$C_{16}H_{19}N_5O_2S$	MeOH/EtOH	180—183	59	3
7a	$C_{16}H_{17}N_5OS_2 \cdot 3 HCld \cdot 0.5 H_2O$	MeOH	190—195 (Sinters)	75	3
7b	C <sub>16</sub> H <sub>17</sub> N <sub>5</sub> OS <sub>2</sub> ·3 HCl	EtOH/H <sub>2</sub> O	178—180.5	76	3
7d	C17H19N5OS2·3 HCl	H <sub>2</sub> O	187	57	2
7e	$C_{20}H_{19}N_5OS_2$ ·3 HCl <sup>e</sup>	EtOH/H <sub>2</sub> O	202—205	50	2
8a	C <sub>18</sub> H <sub>18</sub> BrN <sub>5</sub> OS·3 HBr	MeOH/H <sub>2</sub> O	258—264 (dec.)	57	2
8b	C <sub>18</sub> H <sub>18</sub> BrN₅OS·3 HBr	MeOH/H₂O	218.5-221.5	58	2
8c	$C_{18}H_{18}BrN_5OS \cdot 3 HBr^{f}$	MeOH/H <sub>2</sub> O	225—230 (dec.)	45	2
8d	C <sub>19</sub> H <sub>20</sub> BrN <sub>5</sub> OS·3 HCl	$H_2O$	193—196	67	2
9	C <sub>18</sub> H <sub>22</sub> N <sub>6</sub> O·3 HCl	$H_2O$	242246	70	3
10	C18H22N6OS·3 HCl	EtOH/MeOH	233-236	45	3
11	C <sub>16</sub> H <sub>18</sub> N <sub>6</sub> O·3 HCl·0.75 H <sub>2</sub> O	H <sub>2</sub> O	208.5-212	43	3

<sup>a</sup>Yield from 2-methylthio-4-pyrimidones (Table II, Scheme 2) or 2-nitroamino-4-pyrimidones (Table III, Scheme 3). <sup>b</sup>Cl<sup>-</sup>: theory 22.8%; found 22.0%. <sup>c</sup>Cl<sup>-</sup>: theory 22.8%; found 22.0%. N: theory 18.0%; found 17.4%. <sup>d</sup>Cl<sup>-</sup>: theory 22.3%; found 21.7%. <sup>e</sup>Cl<sup>-</sup>: theory 20.5%; found 19.8%. <sup>f</sup>Br<sup>-</sup>: theory 35.5%; found 35.0%.

All other micro-analytical data for C, H, N, S and halogen to within  $\pm 0.4\%$  of theory.

led to appreciable ring reduction, so Raney nickel was used as catalyst in the presence of ammonium hydroxide to prevent ring reduction [6] (Scheme 1c). The ethyl 3-(heterocyclyl) propionates were then converted into the required isocytosines by one of two routes (Schemes 2 and 3).

In the first route (Scheme 2) the substituted propionate esters were subjected to a formylation reaction using ethyl formate and sodium wire in dry ether as base, to give 2formyl-3-(heterocyclyl) propionate esters. These were reacted *in situ* with thiourea to give 2-thiouracils (Table I), which in turn were alkylated with methyl iodide, to give the 2methylthio-4-pyrimidones (Table II). Finally, the 2-methylthio-4-pyrimidones were treated at high temperature, under fusion conditions, in a nucleophilic displacement with the primary amines to give the target isocytosines (Table IV: Route shown in Table).

In the second route (Scheme 3), the substituted propionate esters were formylated with ethyl formate in 1,2-dimethoxyethane in the presence of sodium hydride as base. (This formylation procedure is more convenient than that described in Scheme 2 above and gives more consistent yields). The resulting 2-formyl-3-(heterocyclyl) propionate esters, as their sodium enolates, were treated with nitroguanidine in methanol to give 2-nitroamino-4-pyrimidones (Table III), which were refluxed in a suitable solvent in the presence of the selected primary amine to give isocytosines (Table IV: Route shown in Table).

Although it was shown that in Scheme 2 modest yields of isocytosines could be obtained from the 2-methylthio compounds by replacing the fusion conditions with refluxing pyridine, Scheme 3 proved much more convenient [1] and during the course of this work replaced Scheme 2 as the method of choice for the preparation of isocytosine histamine antagonists.

A brief general description of typical preparations viaSchemes 1–3 is given under Experimental protocols.

# Pharmacology

Compounds were assayed *in vitro* for  $H_2$ -receptor histamine antagonist activity by measurement of the antagonism of histamine-stimulated tachycardia in the guinea pig isolated right atrial preparation [7] and activities were expressed as  $pA_2$  values when possible. Some of the compounds were slow to reach equilibrium (see Table V) and were difficult to wash out of the tissue.

Anti-secretory activities of the compounds were obtained by measurement of the inhibition of histamine-stimulated gastric acid secretion in the lumen-perfused stomach of the anaesthetised rat after rapid intravenous injection and, for selected compounds, in the conscious Heidenhain—pouch dog by intravenous injection or oral administration [7].

Compounds were assayed *in vitro* for  $H_1$ -receptor histamine antagonist activity by measurement of the inhibition of histamine-induced contractions of the guinea pig isolated ileum [7] and activities were expressed as  $pA_2$  values.

# **Results and Discussion**

The *in vitro* and *in vivo* H<sub>2</sub>-antagonist and *in vitro* H<sub>1</sub>antagonist activities of the compounds under discussion are shown in Table V. In addition, because of the high acute toxicity observed in some examples from the previously described [1] series of isocytosine H<sub>2</sub>-antagonists, the  $LD_{50}$ 's (mouse) of selected compounds in the present series were measured (Table V).

The series of compounds containing a 2-(5-methyl-4imidazolylmethylthio)ethylamino side chain 6a-h are direct analogues of oxmetidine 2 and the activity profile of this compound is included in Table V for comparison. Replacement of the 3,4-methylenedioxyphenyl group of 2 by pyridyl gave the three isomers 6a-c. The 2- and 3-pyridyl isomers are very potent H<sub>2</sub>-receptor antagonists, with the latter **6b** being slightly more active than oxmetidine *in vivo*, but both compounds have relatively high acute toxicity. The 4-pyridyl isomer **6a** has lower H<sub>2</sub>-antagonist activity but is less acutely toxic. This pattern is similar to that seen in the oxmetidine series, when the introduction of a substituent into the phenyl ring of the 5-benzyl group para to the -CH<sub>2</sub>— linkage gave compounds with reduced toxicity [1]. In this series, the pyridyl ring nitrogen of **6a** appears to be having the same effect. The introduction into 6b of a methyl group *para* to the  $-CH_2$  – linkage gave **6d**, a potent  $H_2$ -receptor antagonist with reduced toxicity. The quinolyl analogue of 6b, compound 6e however, is appreciably less potent at the H2-receptor and shows slightly greater H<sub>1</sub>-antagonist activity than other compounds in this series. Compounds 6f-h, although potent  $H_2$ -receptor antagonists, lack a strategically-placed substituent, are thus toxic or potentially toxic and were not studied further.

Replacement of the 5-methyl-4-imidazolyl ring of cimetidine 1 with alternative nitrogen-containing heterocycles has been described [3], such compounds being the first non-imidazole H2-receptor antagonists. When similar modifications on selected compounds were carried out in the present series, interesting combined H<sub>1</sub>- and H<sub>2</sub>-antagonist activities were observed. Thus the 2-thiazolyl compounds 7a, b, d, e, have  $H_2$ -antagonist activities very similar to the corresponding 5-methyl-4-imidazolyl analogues described above, but  $H_1$ -antagonist activity is increased. In the 3bromo-2-pyridyl series 8a-d H<sub>2</sub>-antagonist activity is, in most cases, marginally lower for a given 5-substituent, but H<sub>1</sub>-antagonist activity is further increased and for some compounds (e.g., 8d) the  $pA_2$  at the H<sub>1</sub>-receptor (guinea pig ileum) is greater than that at the H<sub>2</sub>-receptor (guinea pig atrium). It should be noted, however, that the H<sub>1</sub>receptor antagonist activity of these compounds is still substantially lower than for the classical 'anti-histamines' (e.g., mepyramine  $pA_2$  (g.p. ileum) = 9.36 [8]). Moving from 4-methyl-5-imidazolyl through 2-thiazolyl to 3-bromo-2-pyridyl rings increases the lipophilicity within a given series of compounds (estimated increase in octanol/water partition,  $\log P$ , + 0.5 and + 0.6, respectively, (R.C. Mitchell, unpublished results, measured in [3])) and there is a corresponding increase in H<sub>1</sub>-antagonist activity (e.g.,  $6d \rightarrow 7d \rightarrow 8d$ ). For compounds with a given isocytosine 5-substituent, the order of acute toxicity was 5-methyl-4-



Compour	nd	H <sub>2</sub> -Antag	gonist	H <sub>l</sub> -Antagonist	Acute Toxicity	
10.	ĸ	pA <sub>2</sub> (Slope) <sup>a</sup> (G.P. Atrium)	ED <sub>50</sub> (rat) <sup>b</sup> (µmol./kg)	pA <sub>2</sub> (Slope) <sup>C</sup> (G.P. Ileum)	LD(Mouse) <sup>d</sup> (µmol/kg-i.v.)	
2 (oxmet <sup>:</sup>	C <sub>6</sub> H <sub>3</sub> ,3,4-OCH <sub>2</sub> O- idine)	*6.9 (30',0.65)	0.09	5.44 (1.11) [ <b>2</b> ]	260	
ба	4-Pyridy]	6.46 (0.85)	1.15	4.20 (0.78)	160	
6b	3-Pyridyl	6.98 (1.01)	0.05	*5.23 (1.39)	24	
6c	2-Pyridyl	7.13 (60', 0.94)	0.23	5.02 (1.08)	20	
6d	6-Methyl-3- pyridyl	7.05 (30', 1.15)	0.09	5.49 (0.88)	120	
6e	3-Quinoliny1	6.12 (30', 1.1)	0.34	5.93 (1.0)	130	
6f	2-Thieny]	*6.4 (2.2)	0.1	*5.85 (1.41)	1.5	
6g	2-Thiazolyl	7.38 (0.9)	0.07	5.05 (1.33)	17	
6h	2-Furyl	7.08 (60', 0.9)	0.27	5.17 (1.08)	-	



<sup>a</sup>Antagonism of histamine-stimulated tachycardia in guinea pig right atrium *in vitro*. For some compounds, preliminary experiments showed that antagonism had not reached equilibrium within the normal 8 min period but it had been achieved within the time shown. Where indicated, this equibration time was therefore used in the determination of  $pA_2$  values.

For those compounds marked \* the  $pA_2$  values must only be considered approximate figures, since the slope of the Schild plot was significantly different from unity.

<sup>b</sup>Antagonism of histamine-stimulated gastric acid secretion in the lumen-perfused stomach of the anaesthetised rat. Compounds were given by rapid intravenous injection during a near maximal plateau of acid secretion. The  $ED_{50}$  is the dose required to produce 50% inhibition and was estimated from the linear regression of log [J/100-I] on log dose, where I is the percentage inhibition.

<sup>c</sup>Antagonism of histamine-induced contractions of the isolated guinea pig ileum *in vitro*. Equilibration was achieved within 2 min and  $pA_2$  values determined. For those compounds marked \* the  $pA_2$  values must only be considered approximate figures, since the slope of the Schild plot was significantly different from unity.

<sup>a</sup>Groups of 5 or 10 mice dosed by i.v. injection *via* the tail vein using a suitable range of doses. Mortalities were recorded for up to 7 days. Medium lethal dose was estimated using the method of Litchfield and Wilcoxon [25]. All tests were completed some years ago, before such tests were dicontinued at SK&F.



8a	4-Pyridyl	6.37 (0.92)	1.27	6.58 (0.87)	240
8ъ	3-Pyridyl	6.89 (30', 0.84)	0.36	7.03 (0.76)	56
8c	2-Pyridyl	*6.8 (0.66)	2.72	6.25 (1.02)	80
8d	6-Methyl-3- pyridyl	6.85 (30', 0.88)	0.18	7.15 (0.87)	120



imidazolyl > 3-bromo-2-pyridyl > 2-thiazolyl, as indicated by increasing  $LD_{50}$  values.

Using the 5-(3-pyridylmethyl) isocytosine moiety as a model system, further modifications to the amino side chain have been made. Thus, replacement of the sulphur atom in 6b with a  $-CH_2$  group to give 9 reduces  $H_2$ -antagonist activity. A similar reduction in activity in other series of H<sub>2</sub>-receptor antagonists has been rationalised [9] by consideration of the tautomerism of the imidazole ring. Extending the linking chain of 6b by an extra --CH<sub>2</sub>-- group to give the methylthiopropyl analogue 10 leads to a further reduction in H<sub>2</sub>-antagonist activity. Replacement of the (5-methyl-4imidazolylmethylthio)ethyl side chain of 6b with 3-(4imidazolyl)propyl led to a compound 11 which has the properties of a weak partial agonist. This is of interest, since the 3-(4-imidazolyl)propyl moiety is a feature of the potent, selective H<sub>2</sub>-agonist, impromidine [10]. The H<sub>1</sub>antagonist activities of 9-11 are similar to that of 6b.

Compound 6d has the best combination of low acute toxicity and good selectivity for the  $H_2$ -receptor and was further evaluated in the Heidenhain—pouch dog by both intravenous and oral routes of administration and compared to oxmetidine 2 and cimetidine 1 (Table VI). When given

Table VI. Anti-secretory activity in the conscious Heidenhein—pouch dog measured as the % inhibition of maximally stimulated acid secretion at a given dose<sup>a</sup>.

Compd.	i.v. dose (µmol/kg)	Inhibition (%)	Oral dose (µmol/kg)	Inhibition (%)
6d	0.5 0.25	87 50	5.0	41
Oxmetidine 2	0.5	61 [2]	5.0	53 [2]
Cimetidine 1	4.0	68 [26]	2.0	82 [26]

<sup>a</sup>Acid secretion stimulated by a continuous intravenous infusion of histamine at a rate of 20  $\mu$ mol/h.

by the i.v. route, **6d** is somewhat more potent than oxmetidine in inhibiting histamine-stimulated gastric acid secretion and has a similar duration of action. Oral activity was somewhat lower. Compound **6d** underwent preliminary studies as a parenterally administered  $H_2$ -antagonist but was not evaluated clinically, being replaced by the more potent, longer-acting agent donetidine (SK&F 93574) [11].

However, the 5-(6-methyl-3-pyridylmethyl)isocytosine moiety of 6d became the basis, on furthur modification of the group in the 2-position, for a series of clinically effective agents with differing pharmacological profiles. Thus, icotidine (SK&F 93319), an antagonist of histamine at both  $H_1$ - and  $H_2$ -receptors [12], temelastine (SK&F 93944), a potent, selective, non-sedative  $H_1$ -antagonist [13] and lupitidine (SK&F 93479), a potent, selective, long-acting  $H_2$ -antagonist [14], all contain this isocytosine group and will be the focus of future medicinal chemistry publications.

#### **Experimental** protocols

#### Chemistry

All new compounds quoted in the Tables were fully characterized spectroscopically. NMR spectra were recorded on a Jeol PFT 100P or a Varian AC60 spectrometer using (CH<sub>3</sub>)<sub>4</sub>Si for reference. Infra-red spectra were recorded on a Perkin—Elmer 577 or a Perkin—Elmer 580B spectrophotometer and samples were presented as a nujol mull or a KBr disc.

Thin—layer chromatograms (TLC) were run on pre-coated silica gel 60  $F_{254}$  plates with ethyl acetate:methanol:880 NH<sub>3</sub> (5:1:1), as the mobile phase. Micro-analytical data are within  $\pm$  0.4% of theoretical values unless quoted otherwise.

Melting points were recorded in open capillaries on an electrothermal apparatus and are uncorrected.

#### Starting materials used in Scheme 1

The aldehydes used as starting materials in Scheme 1 were all obtained commercially except 6-methylnicotinaldehyde and 2-thiazolealdehyde which were prepared by published procedures from commercially available 2-methyl-5-vinylpyridine [15] and from 2-hydroxymethyl-thiazole [16], respectively. 2-Hydroxymethylthiazole was prepared from commercially available 2-bromothiazole [16].

#### Primary amines

The side chain primary amines used in the final stage of Schemes 2 and 3 to prepare the isocytosines in Table IV were prepared according to literature procedures as indicated below: 4-[[(2-aminoethyl)thio]-methyl]-5-methylimidazole [17]; 2-[[(2-aminoethyl)thio]methyl]-5-methylimidazole [17]; 2-[[(2-aminoethyl)thio]methyl]-3-bromopyridine [3]; 4-(3-amino-propyl)midazole [18]; 4-(4-aminobutyl)-5-methylimidazole [17]; 4-[[(3-aminopropyl)thio]methyl]-5-methylimidazole [3].

#### Procedures for the preparation of ethyl 3-(heterocyclyl) propionates [19]

#### $\beta$ -(Heterocyclyl) acrylic acids

The appropriate aldehydes were converted to  $\beta$ -substituted acrylic acids by the general method of Marvel *et al.* [20], using malonic acid in pyridine containing a small quantity of piperidine (yields 70–90%).

#### Ethyl 3-(heterocyclyl) propionates

(Where heterocyclic ring is 2,3 or 4-pyridyl, 6-methyl-3-pyridyl, 3quinolinyl and 2-thiazolyl).

The  $\beta$ -(heterocyclyl) acrylic acids were esterified in ethanol in the presence of concentrated sulphuric acid [21] to give ethyl  $\beta$ -(heteroethyl 3-(heterocyclyl) propionates. The reduction step could be accelerated by warming the reaction mixture to ~ 45°C and this was partiethyl 3-(heterocyclyl) propionates. The reduction step could be accelerated by warming the reaction mixture to ~ 45°C and this was particularly useful in the case of the 2-thiazolyl analogue when reduction at room temperature was very slow.

# Ethyl 3-(2-thienyl) propionate

 $\beta$ -(2-Thienyl) acrylic acid was reduced in methanol (H<sub>2</sub>/10% Pd/C/ 50°C/50 psi) to give 3-(2-thienyl) propionic acid in 84% yield. (mp: 49—50°C from ether/40—60 petroleum ether; literature mp: 48—49°C [4]). 3-(2-Thienyl) propionic acid was esterified in the usual way with ethanol in concentrated sulphuric acid to give ethyl 3-(2-thienyl) propionate in 74% yield.

# Ethyl 3-(2-furyl) propionate [6]

To a solution of  $\hat{\beta}$ -(2-furyl) acrylic acid (20 g, 0.144 mol) in acetone (200 ml) was added, dropwise and simultaneously, diethyl sulphate (22.2 g, 0.144 mol) and tri-*n*-propylamine (21.5 g, 0.150 mol). The reaction mixture was then heated at reflux temperature for 2 h, cooled, poured onto ice (300 g) and extracted with diethyl ether (4 × 100 ml). The ether extracts were washed (aq. K<sub>2</sub>CO<sub>3</sub> and then H<sub>2</sub>O), dried (MgSO<sub>4</sub>) and evaporated to give ethyl  $\beta$ -(2-furyl) acrylate (24 g, ~ 100%) yield) as a brown oil.

A solution of ethyl  $\beta$ -(2-furyl) acrylate in ethanol and conc. NH<sub>4</sub>OH was reduced by the method described in [6] (H<sub>2</sub>/Raney nickel) to give ethyl 3-(2-furyl) propionate in 80% yield.

All intermediates in this section were checked for purity/identity by TLC and NMR and/or IR.

#### Isocytosines [19] [23] (Table IV)

#### 2-Thiouracils [19] (Table I)

The ethyl 3-(heterocyclyl) propionates (1 eq) and ethyl formate (1 eq) were added over a period of 6 h to a stirred mixture of sodium wire (1 eq) in dry ether (60 ml/0.1 mol of ester) cooled in an ice-salt bath. The mixture was stirred for 18 h at room temperature and evaporated to dryness to give the ethyl 2-formyl-3-(heterocyclyl) propionates as their sodium enolates. The residue was refluxed for 7 h with thiourea (1 eq based on starting ester) and ethanol (70 ml/0.1 mol of thiourea). The mixture was evaporated to dryness and the residue was dissolved in water. Acetic acid was added until the mixture attained pH = 4. The white precipitate was collected by filtration, washed well with water and crystallised from a suitable solvent (see Table I).

#### 2-Methylthio-4-pyrimidones [19] (Table II)

A solution of substituted 2-thiouracil (1 eq), methyl iodide (1 eq) and sodium hydroxide (1 eq) in water (100 ml/0.1 mol of 2-thiouracil) and ethanol (200 ml/0.1 mol of 2-thiouracil) was stirred at 60°C. A white solid began to precipitate almost immediately. Stirring was continued for 30 min and the mixture allowed to cool. The solid was collected, washed well with water, dried and crystallised from an appropriate solvent (see Table II).

Isocytosines (compounds 6a-c, 6e-g, 7d-e, 8a-d) Scheme 2 An intimate mixture of substituted 2-methylthio-4-pyrimidone (1 eq) and primary amine base (1 eq, see Table IV) was heated at 145--150°C for 4-5 h. After cooling the reaction mixture was triturated with water to give the required isocytosine as the free base.

This was converted into a suitable crystalline salt by either: 1) dissolution in ethanolic hydrogen chloride, evaporation and crystallisation, or 2) dissolution in a suitable solvent (methanol or ethanol), treatment with 2 N HBr, evaporation to dryness and crystallisation.

#### 2-Nitroamino-4-pyrimidones [23, 24] (Table III)

To a stirring suspension of sodium hydride (50% in oil, 1.25 eq) in 1,2-dimethoxyethane (25 ml/0.1 mol of NaH) were added dropwise at 0°C over a 2h period the ethyl  $\beta$ -(heterocyclyl) propionate (1 eq) and ethyl formate (1.5 eq). Frothing occurred and the mixture was allowed to stand at room temperature for 20 h, poured onto ice and the resulting solution extracted with ether  $(\times 3)$ . The aqueous solution was acidified to pH = 6 with 2 N H<sub>2</sub>SO<sub>4</sub> to precipitate the ethyl 2formyl-3-(heterocyclyl) propionate as a white solid which was collected, washed with water, dried and used immediately without further purifi-cation (yields 50-70%).

Nitroguanidine (1 eq) was added to sodium methoxide (from 1.1 eq of sodium) in methanol (100 ml/0.1 mol of nitroguanidine) and the mixture was heated under reflux for 45 min. The ethyl 2-formyl-3-(heterocyclyl) propionate (1 eq) was added over 1 h, the mixture heated under reflux for 18 h and then evaporated to dryness. The residue was dissolved in water and the solution was extracted with chloroform. The residual aqueous phase was adjusted to pH = 5 with glacial acetic acid when the title compounds precipitated as white solids and were crystallised, if necessary, from a suitable solvent (see Table III).

# Isocytosines (compounds 6d, 6h, 7a-b, 9-11) Scheme 3

A mixture of substituted 2-nitroamino-4-pyrimidone (1 eq) and primary amine base (1 eq, see Table IV) was dissolved in absolute ethanol (200 ml/0.1 mol) and heated under reflux for 24 h. The ethanol was evaporated and the residue washed with water. Alternatively, refluxing dry pyridine (200 ml/0.1 mol) could be used as solvent (reaction time 12 h) when greater care was needed at the water-washing stage to remove all residual pyridine.

Purification of the isocytosines was achieved by crystallisation of the base (e.g., compound 6h) or conversion to a suitable salt as described previously.

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