

PII: S0040-4020(97)00161-0

Stereoselectivity in the Amination of Chiral Cyclohex-3-en-1-one Ketals

Stefania Fioravanti, Giuseppe Luna, Lucio Pellacani, Paolo A. Tardella

Dipartimento di Chimica, Università "La Sapienza", P.le Aldo Moro 2, I-00185 Roma, Italy

Abstract: Optically active cyclohex-3-en-1-one ketals by photolysis of N_3CO_2Et or $N_3C(OEt)NMs$ or by CaO induced α -elimination of NsONHCO₂Et give diastereomeric aziridines (up to 72% yields, up to 60% d.e.). A simple HPLC separation allows to obtain practically pure aziridines. The conversion of the main product to the ketal of (*R*)-*N*-(ethoxycarbonyl)- β -aminocyclohexanone is also reported and a further oxidation directly gives a derivative of (*R*)-2-aminoadipic acid. © 1997 Elsevier Science Ltd.

We have recently proposed an improved amination methodology, based on the use of solid bases to deprotonate sulphonyloxycarbamates,^{1.3} and we have shown that homoallylic cyclic ketals were promising substrates.^{1.4}

We now report on results obtained with optically active cyclohex-3-en-1-one ketals **1a-e** upon reaction with ethyl {[(4-nitrobenzene)sulphonyl]oxy}carbamate (NsONHCO₂Et)⁵ and in the photolysis of ethyl azidoformate (N₃CO₂Et)⁶ and of a related imidoyl azide, ethyl *N*-(methanesulphonyl)azidoformimidate [N₃-C(OEt)NMs].^{7,8} All reactions were performed at room temperature in dichloromethane.



Scheme 1

Isolated yields and diastereomeric excesses of the aziridination products are collected in the Table. Reactions run with NsONHCO₂Et are simpler and faster (≤ 1 h) and the reagent is easier to prepare and to handle with respect to the azides.⁹

1 / 2 / 2'	\mathbf{R}_1	R_2	NsONHCO ₂ Et		N ₃ CO ₂ Et		N ₃ C(OEt)NMs	
			% Yield	% d. e.	% Yield	% d. e.	% Yield	% d. e.
a	н	Н	62 ^{<i>a</i>}		40		75	
b	CH ₂ OMe	Н	54	8	22	4	23	8
с	Me	Н	33	16	67	17	72	22
d	b	Η	43	25				
e	Н	Ph	13	51	30	57	50	60

Table. Aziridines 2 and 2' from Cyclohex-3-en-1-one Ketals 1.

^a See reference 1. ^b $R_1 = -(CH_2)_4$ -

The yields of **2'** are usually higher than those of **2**. The reactions performed with NsONHCO₂Et or N₃C(OEt)NMs appear nearly quantitative, by inspection of ¹³C NMR spectra of the crude reaction mixtures. However the isolated yields result lower than those expected, very likely due to chromatographic instability of aziridines, as observed in other occasions. As the diastereomeric excesses are concerned the three methods appear equivalent: in all cases the best selectivity has been achieved when phenyl groups are the substituents in the stereogenic centres. Nevertheless the diastereomers **2c-e** and **2'c** were easily and completely separated by HPLC.

In order to chemically correlate the different aziridines, a mixture of diastereomeric $2^{\circ}e$ was converted into diastereomeric 2e, by subsequent methanolysis⁸ and chloroformylation.



The comparison by GC-MS and HPLC analyses has shown that the three aminating agents lead to the same major diastereomer. In all three procedures here reported the attack of the reagent always happened mainly from the same face of the chiral homoallylic ketals.

The optically active aziridines so obtained are versatile synthetic intermediates. By reductive ring cleavage¹⁰ we were able to cleanly transform a single diastereomer 2c into 4c by using L-Selectride^{Φ} with

complete regioselectivity,¹¹ while both LiAlH₄ and NaBH₄ failed to react. We obtained the optically pure N-(ethoxycarbonyl) β -amino ketal 4c with a 69% yield, after a HPLC purification.



This clean ring scission after the aziridination step would provide a useful access to β -amino carbonyl compounds,¹² that are interesting building blocks for synthesis.

Next, aiming to know the absolute configuration of the carbon atom carrying the carbamate function, compound **4c** has been submitted to oxidation. Thus, the treatment¹³ with KMnO₄ and Na₂CO₃ in water at 60°C for 24 h, followed by MeOH/H₂SO₄ esterification, allowed the formation of dimethyl 2-[(ethoxycarbonyl)amino]adipate **5** in 44% yield, in addition to a 9% of 3-isomer **6**. The two isomers were easily separated by HPLC.



By comparison with the optical properties of the enantiomer prepared from commercially available (S)-2aminoadipic acid, we were able to assign the (R) configuration to 4c and consequently the (S,R) configuration to the newly created chiral carbon atoms of the major aziridines 2 and 2'.

We would like to emphasise the synthetic potential of the above one-step oxidative cleavage.

EXPERIMENTAL SECTION

GC analyses were performed on a HP 5890 Series II gas chromatograph with a capillary column (methyl silicone, 12.5 m x 0.2 mm). GC-MS were done on a HP G1800A GCD System with a capillary column (phenyl methyl silicone, 30 m x 0.25 mm). ¹H NMR and ¹³C NMR spectra were obtained in CDCl₃ on a Varian XL-300 spectrometer, with CHCl₃ as an internal standard. IR spectra in CCl₄ were done with a Perkin-Elmer 1600 Series FTIR spectrometer. The separations by HPLC were done with a Varian 9001 instrument equipped with a Varian RI-4 differential refractrometer. Solvents were HPLC-grade. Optical rotations were recorded at the Sodium D line with a Perkin-Elmer 457 polarimeter (1-cm cell).

Synthesis of ketals 1a-e. A solution of cyclohex-2-en-1-one (12 mmol), the appropriate commercial diol (10 mmol), benzene (toluene for 1d) and 0.1 g of TsOH was refluxed under azeotropic conditions. Progress of the reaction was monitored by GC-MS (3-6 h). When the reaction was complete, the mixture was concentrated *in vacuo* and the residue diluted with CH₂Cl₂, washed with an 8% solution of NaHCO₃, dried over Na₂SO₄, filtered and concentrated. The crude product was purified by flash chromatography on silica gel (hexane/ethyl acetate, 9:1).

1b: $[\alpha]_D$ +13.7 (CH₂Cl₂); IR 1687, 1114 cm⁻¹; ¹H NMR δ 1.7 (*t*, 2 H, CH₂), 2.16-2.32 (*m*, 4 H, CH₂), 3.36 (*s*, 6 H, CH₃), 3.50 (*d*, 2 H, OCH₂), 3.51 (*d*, 2 H, OCH₂), 3.91-3.97 (*dt*, 1 H, OCH), 3.98-4.05 (*dt*, 1 H OCH), 5.50-5.68 (*m*, 2 H, HC=CH); ¹³C NMR δ 24.37, 23.13, 36.83 (CH₂), 59.33, 59.37 (CH₃), 73.26, 73.36 (OCH₂), 77.00, 77.05 (OCH), 108.95 (C), 124.12, 126.37 (HC=CH); MS *m*/*z* 228 (M⁺, 27), 174 (50), 115 (56), 112 (15), 102 (15), 100 (12), 96 (12), 87 (18), 85 (39), 84 (68), 81 (21), 79 (15), 77 (10), 71 (16), 70 (30), 69 (16), 68 (24), 67 (20), 59 (28), 58 (16), 55 (32), 54 (29), 53 (17), 45 (100), 43 (29), 42 (10), 41 (33).

1c: $[\alpha]_D$ -3.1 (CH₂Cl₂); IR 1652, 1110 cm⁻¹; ¹H NMR δ 1.24 (*d*, 3 H, CH₃), 126 (*d*, 3 H, CH₃), 1.75 (*t*, 2 H, CH₂), 2.20-2.27 (*m*, 4 H, CH₂), 3.60-3.69 (*m*, 2 H, OCH), 5.56-5.72 (*m*, 2 H, HC=CH); ¹³C NMR δ 16.82, 17.00 (CH₃), 24.45, 32.56, 37.18 (CH₂), 77.97, 78.15 (OCH), 106.90 (C), 124.36, 126.54 (HC=CH); MS *m*/z 168 (M⁺, 25), 114 (100), 96 (26), 81 (22), 79 (20), 72 (18), 68 (11), 67 (12), 57 (11), 55 (24), 54 (11), 43 (51), 42 (16), 41 (16).

1d: $[\alpha]_D$ -5.1 (CH₂Cl₂); IR 1444, 1118 cm⁻¹; ¹H NMR δ 1.16-1.50 (*m*, 4 H, CH₂), 1.68-1.88 (*m*, 4 H, CH₂), 2.05-2.34 (*m*, 6 H, CH₂), 3.21-3.38 (*m*, 2 H, OCH), 5.53-5.76 (*m*, 2 H, HC=CH); ¹³C NMR δ 23.62, 24.31, 28.87, 32.22, 36.83 (CH₂), 79.93, 80.11 (CHO), 107.93 (C), 124.52, 126.84 (HC=CH); MS *m*/*z* 194 (M⁺, 48), 140 (100), 112 (13), 99 (12), 98 (41), 97 (22), 96 (56), 81 (62), 80 (29), 79 (28), 77 (18), 71 (12), 69 (39), 68 (76), 67 (41), 57 (12), 55 (27), 54 (78), 53 (21), 43 (24), 42 (18), 41 (66).

1e: $[\alpha]_D$ -67.7 (CH₂Cl₂); IR 3033, 1496, 1135 cm⁻¹; ¹H NMR δ 1.93-2.09 (*m*, 2 H, CH₂), 2.29-2.63 (*m*, 4 H, CH₂), 4.72 (*s*, 2 H, CH), 5.59-5.72 (*m*, 2 H, HC=CH), 7.13-7.28 (*m*, 10 H, aromatic H); ¹³C NMR δ 24.54, 32.59, 37.11 (CH₂), 85.25 (CHPh), 108.67 (OCO), 124.20, 129.79 (HC=CH), 126.70, 126.81, 128.20, 128.29, 128.35, 128.41 (aromatic CH), 136.63, 136.88 (aromatic C); MS *m*/*z* 292 (M⁺, 2), 187 (15), 186 (100), 180 (19), 179 (20), 178 (14), 168 (15), 167 (50), 165 (18), 105 (27), 96 (20), 91 (90), 90 (15), 89 (19), 81 (13), 80 (77), 79 (44), 77 (33), 68 (13), 54 (13).

Reaction of 1 with NsONHCO₂Et. To a stirred solution of the substrate (0.6 mmol) in 6 ml of CH₂Cl₂ at room temperature, CaO and NsONHCO₂Et were added portionwise, in the molar ratios

substrate:NsONHCO₂Et:CaO=1:5:5 for 1b and 1d, 1:4:4 for 1c and 1:7:7 for 1e. After 1 h of stirring, 20 ml of CH₂Cl₂ and 200 ml of petroleum ether (bp 30-50 °C) were added. After filtration, the liquid phase was concentrated *in vacuo*. The aziridines 2b were separated, as a diastereomeric mixture, by flash chromatography on silica gel (hexane/ethyl acetate, 1:1), while 2c, 2d and 2e were obtained as pure diastereomers by HPLC (hexane/ethyl acetate, 8:2 for 2c and 2e; 7:3 for 2d).

2b: IR 1720, 1098 cm⁻¹; ¹H NMR δ 1.23 (*t*, 3 H, CH₂CH₃), 1.32 (*t*, 2 H, CH₂), 2.00-2.11 (*m*, 4 H, CH₂) 2.58-2.65 (*m*, 2 H, NCH), 3.36 (*s*, 6 H, OCH₃), 3.43-3.50 (*m*, 4 H, CH₂CH), 3.85-3.92 (*m*, 2 H, OCH), 4.16 (*q*, 2 H, CH₂CH₃); ¹³C NMR δ 14.28 (CH₃), 21.89, 28.40, 28.90, 34.63, 35.26 (CH₂), 35.58, 35.61, 37.14, 37.22 (NCH), 59.37 (OCH₃), 62.30 (CH₂CH₃), 72.99, 73.20 (CH₂CH), 77.01, 77.16 (OCH), 107.78, 107.89 (C), 163.86 (CO); MS *m*/*z* 315 (M⁺, 0.4), 270 (12), 242 (10), 174 (24), 115 (73), 112 (17), 110 (17), 102 (15), 100 (13), 85 (39), 84 (72), 83 (11), 82 (18), 71 (17), 70 (33), 69 (20), 68 (25), 67 (13), 62 (11), 59 (14), 58 (13), 56 (13), 55 (34), 54 (20), 45 (100), 43 (25), 42 (12), 41 (30).

2c. Major diastereomer: $[\alpha]_D$ + 3.7 (CH₂Cl₂); IR 1721, 1214 cm⁻¹; ¹H NMR δ 1.19 (2d, 6 H, CHC*H*₃), 1.24 (*t*, 3 H, *C*H₂*CH*₃), 1.55-1.69 (*m*, 2 H, CH₂), 1.96-2.19 (*m*, 4 H, CH₂), 2.58-2.70 (*m*, 2 H, NCH), 3.45-3.60 (*m*, 2 H, OCH), 4.10 (*q*, 2 H, OCH₂); ¹³C NMR δ 14.30 (CH₂CH₃), 16.57 (CHCH₃), 22.00, 29.33, 34.85 (CH₂), 35.77, 37.25 (NCH), 62.32 (OCH₂), 77.69, 77.78 (OCH), 105.83 (C), 164.00 (CO); MS *m/z* 255 (M⁺, 0.4) 184 (12), 182 (13), 114 (100), 110 (13), 72 (11), 68 (11), 56 (16), 55 (23), 43 (22), 41 (13). Minor diastereomer: $[\alpha]_D$ -3.0 (CH₂Cl₂); ¹³C NMR δ 14.27 (CH₂CH₃), 16.89 (CHCH₃), 22.11, 28.73, 35.62 (CH₂), 35.90, 37.11 (NCH), 62.32 (OCH₂), 78.14, 78.55 (OCH), 105.63 (C), 164.00 (CO).

2d. Major diastereomer: $[\alpha]_D$ + 6.1 (CH₂Cl₂); IR 1723, 1097 cm⁻¹; ¹H NMR δ 1.07 (*t*, 3 H, CH₂CH₃), 1.15-2.20 (*m*, 14 H, CH₂), 2.59-2.68 (*m*, 2 H, NCH), 3.10-3.30 (*m*, 2 H, OCH), 4.10 (*q*, 2 H, CH₂CH₃); ¹³C NMR δ 14.31 (CH₂CH₃), 21.91, 23.73, 28.77, 29.03, 34.59 (CH₂), 35.85, 37.41 (NCH), 62.39 (OCH₂), 79.61, 80.56 (OCH), 106.57 (C), 169.78 (CO); MS *m*/*z* 281 (M⁺, 0.7), 184 (40), 141 (16), 149 (100), 138 (19), 112 (14), 110 (31), 99 (17), 98 (48), 97 (20), 96 (20), 95 (14), 94 (11), 83 (11), 82 (21), 81 (43), 80 (32), 71 (13), 70 (11), 69 (48), 68 (28), 67 (27), 57 (13), 56 (27), 55 (41), 54 (26), 43 (27), 42 (20), 41 (46). Minor diastereomer: $[\alpha]_D$ -2.7 (CH₂Cl₂); ¹³C NMR δ 14.35 (CH₂CH₃), 22.21, 23.67, 28.95, 29.16, 35.59 (CH₂), 35.80, 37.17 (NCH), 62.39 (OCH₂), 79.43, 80.18 (OCH), 106.73 (C), 169.78 (CO).

2e. Major diastereomer: $[\alpha]_D - 10.3$ (CH₂Cl₂); IR 1725 cm⁻¹; ¹H NMR δ 1.28 (*t*, 3 H, CH₂CH₃), 1.50-2.37 (*m*, 6 H, CH₂), 2.68-2.74 (*m*, 2 H, NCH), 4.14 (*q*, 2 H, CH₂CH₃), 4.66 (*s*, 2 H, OCH), 7.13-7.32 (*m*, 10 H, aromatic CH); ¹³C NMR δ 14.37 (CH₂CH₃), 22.10, 29.46, 34.91 (CH₂), 35.77, 37.29 (NCH), 62.44 (OCH₂), 84.88, 84.95 (OCH), 107.38 (OCO), 126.41, 126.58, 126.82, 126.93, 128.21, 128.34, 128.41, 128.48, 128.56 (aromatic CH), 136.13, 136.25 (aromatic C), 163.91 (CO); MS *m/z* 273 (M⁺-106, 35), 207 (29), 196 (15), 184 (44), 183 (10), 181 (17), 180 (100), 179 (39), 178 (23), 168 (18), 167 (78), 166 (10), 165 (26), 152 (10), 138 (19), 118 (11), 110 (12), 107 (15), 105 (43), 96 (18), 95 (12), 94 (27), 92 (12), 91 (93), 90 (29), 89 (19), 82 (15), 80 (11), 79 (20), 78 (25), 77 (35), 69 (11), 68 (16), 67 (33), 65 (13), 62 (15), 56 (14), 55 (23), 54 (26), 44 (16), 41 (27). Minor diastereomer: $[\alpha]_D$ +12.3 (CH₂Cl₂); ¹³C NMR δ 14.34 (CH₂CH₃), 22.18, 28.72, 34.91 (CH₂), 35.64, 37.10 (NCH), 62.44 (OCH₂), 85.36, 85.56 (OCH), 107.76 (OCO), 126.41, 126.58, 126.82, 126.93, 128.21, 128.34, 128.41, 128.48, 128.56 (aromatic CH), 136.43, 137.00 (aromatic C), 163.91 (CO).

Photolysis of N_3CO_2Et or $N_3C(OEt)NMs$ with 1. The azide (12 mmol) and the ketal (10 mmol) in 1 ml of CH₂Cl₂ were photolysed in a quarz vessel under an atmosphere of nitrogen at room temperature, using a

medium pressure Hanovia PCR lamp (100 W). When the azide band disappeared in the IR spectrum [2-35 h for N_3CO_2Et , 4-30 h for $N_3C(OEt)NMs$] the solvent was evaporated *in vacuo*. The aziridines **2b** were separated, as a diastereomeric mixture, by flash chromatography on silica gel (hexane/ethyl acetate, 1:1), while **2c** and **2e** were obtained as pure diastereomers by HPLC. The aziridines **2'a**, **2'b** and **2'e** were separated, as diastereomeric mixtures, by flash chromatography on silica gel (hexane/ethyl acetate, 1:1), **2'c** were obtained as pure diastereomers by HPLC. The aziridines **2'a**, **2'b** and **2'e** were separated, as diastereomeric mixtures, by flash chromatography on silica gel (hexane/ethyl acetate, 1:1), **2'c** were obtained as pure diastereomers by HPLC (hexane/ethyl acetate, 8:2).

2'a: IR 1320, 1110 cm⁻¹; ¹H NMR δ 1.26 (*t*, 3 H, CH₂CH₃), 1.40-1.50 (*m*, 2 H, CH₂), 1.60-2.20 (*m*, 4 H, CH₂), 2.40-2.60 (*m*, 2 H, NCH), 3.12 (*s*, 3 H, SCH₃), 3.78 (*m*, 4 H, CH₂O), 4.15 (*q*, 2 H, CH₂CH₃); ¹³C NMR δ 13.65 (CH₂CH₃), 21.70, 27.43, 33.13 (CH₂), 38.91, 40.62 (NCH), 42.06 (SCH₃), 63.88, 64.34 (CH₂O), 65.50 (CH₂CH₃), 106.72 (C), 164.71 (C=N); MS *m*/z 304 (M⁺, 0.2), 225 (38), 218 (37), 179 (10), 154 (51), 140 (12), 139 (88), 138 (61), 137 (15), 125 (21), 121 (37), 112 (15), 111 (82), 110 (20), 100 (19), 99 (100), 95 (19), 94 (12), 93 (16), 87 (39), 86 (91), 82 (27), 80 (11), 79 (40), 69 (15), 68 (36), 67 (31), 66 (12), 65 (12), 56 (11), 55 (20), 54 (13), 43 (18), 42 (25), 41 (22), 29 (24), 28 (14), 27 (10).

2'b: IR 1590, 1110 cm⁻¹; ¹H NMR δ 1.25 (*t*, 3 H, CH₂CH₃), 1.41-1.83 (*m*, 4 H, CH₂), 2.00-2.42 (*m*, 2 H, NCH), 2.98 (*s*, 3 H, SCH₃), 3.34 (*s*, 3 H, CH₃O), 3.36 (*s*, 3 H, CH₃O), 3.43 (*d*, 2 H, CH₂O), 3.46 (*d*, 2 H, CH₂O), 3.86-3.91 (*dt*, 2 H, OCH), 4.14 (*q*, 4 H, CH₂CH₃); ¹³C NMR δ 13.61 (CH₂CH₃), 21.50, 28.45, 28.85, 34.11, 34.73 (CH₂), 39.01, 39.13, 40.72, 40.88 (NCH), 41.91 (SCH₃), 59.32 (OCH₃), 65.46 (CH₂CH₃), 72.96, 73.15 (CH₂), 77.09, 77.19 (OCH), 107.77, 107.83 (C), 164.75 (C=N); MS *m*/z 392 (M⁺, 0.3), 313 (14), 226 (58), 218 (11), 213 (13), 187 (23), 175 (26), 174 (13), 167 (21), 139 (21), 121 (15), 115 (95), 112 (13), 111 (18), 110 (16), 100 (10), 95 (10), 94 (12), 93 (11), 85 (37), 84 (53), 82 (15), 79 (27), 71 (13), 70 (24), 69 (18), 68 (25), 67 (32), 66 (11), 58 (11), 55 (24), 45 (100), 43 (30), 41 (26).

2'c. Major diastereomer: $[\alpha]_D$ +1.0 (CH₂Cl₂); IR 1310, 1110 cm⁻¹; ¹H NMR δ 1.15 (*m*, 6 H, CH₃), 1.23 (*t*, 3 H, CH₂CH₃), 1.40-1.82 (*m*, 6 H, CH₂), 2.09-2.41 (*m*, 2 H, NCH), 2.97 (*s*, 3 H, SCH₃), 3.45-3.65 (*m*, 2 H, CH), 4.13 (*q*, 2 H, CH₂CH₃); ¹³C NMR δ 13.91 (CH₂CH₃), 16.30, 16.41 (CH₃CH), 21.50, 29.22, 34.29 (CH₂), 39.18, 40.74 (NCH), 41.90 (SCH₃), 65.47 (CH₂CH₃), 78.14, 78.51 (OCH), 105.60 (C), 164.83 (C=N); MS *m*/z 332 (M⁺, 3), 253 (13), 182 (20), 166 (26), 153 (10), 140 (15), 139 (13), 128 (15), 127 (69), 122 (22), 115 (18), 114 (100), 111 (17), 110 (23), 95 (11), 94 (11), 82 (13), 79 (26), 68 (14), 67 (22), 56 (18), 55 (36), 54 (13), 43 (25), 42 (11), 41 (17). Minor diastereomer: $[\alpha]_D$ -1.5 (CH₂Cl₂); ¹³C NMR δ 13.62 (CH₂CH₃), 16.70, 16.93 (CH₃CH), 21.60, 28.70, 35.32 (CH₂), 39.10, 40.60 (NCH), 41.76 (CH₃S), 65.47 (CH₂CH₃), 78.14, 78.51 (OCH), 105.78 (C), 164.83 (C=N).

2'e. Major diastereomer: IR 1575, 1310, 1120 cm⁻¹; ¹H NMR δ 1.32 (*t*, 3 H, CH₂CH₃), 1.78-2.43 (m, 6 H, CH₂), 2.41-2.85 (*m*, 2 H, NCH), 3.06 (*s*, 3 H, SCH₃), 4.22 (*q*, 2 H, CH₂CH₃), 4.67 (*2d*, 2 H, OCH), 7.16-7.23 (*m*, 5 H, aromatic CH), 7.26-7.33 (*m*, 5 H, aromatic CH); ¹³C NMR δ 13.90 (CH₂CH₃), 21.80, 29.47, 34.46 (CH₂), 39.45, 40.66 (NCH), 42.28 (SCH₃), 65.60 (CH₂CH₃), 84.86, 85.04 (OCH), 107.30 (OCO) 126.60, 128.53 (aromatic CH), 136.00, 136.18 (aromatic C), 164.05 (C=N). Minor diastereomer: ¹³C NMR δ 13.90 (CH₂CH₃), 21.80, 28.84, 35.35 (CH₂), 39.19, 40.76 (NCH), 42.28 (SCH₃), 65.60 (CH₂CH₃), 85.33, 85.51 (OCH), 107.66 (OCO), 126.60, 128.53 (aromatic CH), 136.00, 138.18 (aromatic C), 164.05 (C=N).

Conversion of 2'e into 2e. The methanolysis of 2'e was performed according to a reported procedure,⁸ giving 3e: IR 3300, 3035, 1100 cm⁻¹; ¹H NMR δ 1.30-1.50 (m, 2 H, CH₂), 1.70-1.90 (m, 2 H, NCH), 2.10-2.40 (m, 4 H, CH₂), 2.40 (br, 1 H, NH), 4.63-4.81 (m, 2 H, OCH), 7.10-7.25 (m, 5 H, aromatic CH), 7.27-7.43 (m,

5 H, aromatic CH); ¹³C NMR δ 15.06, 22.43, 29.10, 29.34, 35.74 (CH₂), 84.92, 85.11 (OCH), 108.44 (OCO), 126.62, 126.80, 126.92, 126.99, 127.03, 128.33, 128.42, 128.51, 128.62, 128.71 (aromatic CH), 136.61 (aromatic C); MS *m*/z 307 (M⁺, 0.01), 202 (15), 201 (100), 181 (11), 180 (72), 179 (50), 178 (24), 167 (45), 165 (31), 152 (12), 146 (21), 132 (66), 110 (13), 108 (11), 107 (41), 106 (22), 105 (35), 96 (14), 95 (27), 94 (60), 92 (11), 91 (32), 90 (13), 89 (21), 82 (19), 80 (22), 79 (20), 77 (37), 69 (14), 68 (32), 67 (27), 65 (12), 56 (49), 55 (24), 54 (22), 51 (12), 43 (20), 42 (13), 41 (28). The obtained aziridine **3e** underwent chloroformylation and **2e** was isolated and characterised.

Reduction of 2c. To a stirred solution of the major diastereomer of 2c (3 mmol) in 10 ml of anhydrous THF, 6.5 ml (6.2 mmol) of 1 M L-Selectride^{*} in THF were added dropwise at room temperature. After 15 min, H₂O was added and the crude mixture was extracted with ethyl ether, dried over Na₂SO₄, filtered and concentrated. The product 4c was purified by HPLC (hexane/ethyl acetate, 8:2): $[\alpha]_D$ +11.9 (CH₂Cl₂); IR 3400, 1750, 1100 cm⁻¹; ¹H NMR δ 1.19 (*m*, 9 H, CH₃), 1.70-2.00 (*m*, 8 H, CH₂), 3.54-3.64 (*m*, 2 H, CHCH₃), 3.90 (*m*, 1 H, NCH), 4.07 (*q*, 2 H, CH₂CH₃), 5.40 (*br* 1 H, NH); ¹³C NMR δ 14.48 (CH₂CH₃), 16.60, 16.74 (CH₃), 19.52, 30.70, 36.08, 41.24 (CH₂), 47.32 (CHN), 60.30 (CH₂CH₃), 77.90, 78.06 (CHCH₃), 107.91 (C), 156.02 (CO); MS *m*/z 257 (M⁺, 1), 215 (11), 169 (12), 142 (17), 141 (30), 129 (14), 128 (24), 127 (57), 116 (10), 115 (15), 113 (66), 112 (15), 97 (17), 96 (19), 73 (13), 70 (12), 69 (19), 56 (20), 55 (48), 43 (36), 42 (15), 41 (22).

Synthesis of (R)-5 and (R)-6. To a stirred solution of 69.0 mg (0.27 mmol) of 4c, 0.31 ml of H₂O and 53 mg (0.5 mmol) of Na₂CO₃, 84.16 mg (0.5 mmol) of KMnO₄ in 2.7 ml of H₂O were added during 24 h at 60 °C. After 30 min, the crude mixture was filtered, acidified with concentrated HCl and concentrated. Then, 10 ml of MeOH and 0.08 ml of concentrated H₂SO₄ were added and the mixture was refluxed for 24 h. After cooling at room temperature, a saturated solution of NaOH was added. After concentration, the residue was diluted with H₂O, extracted with CH₂Cl₂, dried over Na₂SO₄ and the products were separated by HPLC (hexane/ethyl acetate, 7:3).

(*R*)-5: $[\alpha]_D$ -10.6 (CH₂Cl₂); IR 3438, 1749, 1700, 1511 cm⁻¹, ¹H NMR δ 1.18 (*t*, 3 H, CH₂CH₃), 1.58-1.86 (*m*, 4 H, CH₂), 2.28 (*t*, 2 H, CH₂), 3.61 (*s*, 3 H, CH₃), 3.68 (*s*, 3 H, CH₃), 4.06 (*q*, 2 H, CH₂CH₃), 4.31 (*m*, 1 H, CH), 5.30 (*br*, 1 H, NH); ¹³C NMR δ 14.22 (CH₂CH₃), 20.37, 31.69, 33.06 (CH₂), 51.40, 52.19, 53.27, 61.01 (CH₂CH₃), 156.30 (NCO), 173.01, 173.62 (CO); MS *m*/*z* 261 (M⁺, 0.02), 230 (11), 203 (11), 202 (100), 170 (21), 156 (50), 128 (49), 124 (20), 115 (10), 98 (68), 88 (11), 84 (10), 59 (12), 56 (30), 55 (27), 43 (11).

(*R*)-6: $[\alpha]_D$ +14.0 (CH₂Cl₂); IR 3438, 1749, 1700, 1511 cm⁻¹; ¹H NMR δ 1.21 (*t*, 3 H, CH₂CH₃), 1.85 (*q*, 2 H, CH₂), 2.38 (*t*, 2 H, CH₂), 2.54 (*d*, 2 H, CH₂), 3.65 (*s*, 1 H, CH₃), 3.66 (*s*, 1 H, CH₃), 3.95 (*m*, 1 H, CH), 4.07 (*q*, 2 H, CH₂CH₃), 5.22 (*br*, 1 H, NH); MS *m*/*z* 261 (M⁺, 0.01), 230 (13), 188 (100), 174 (47), 156 (43), 142 (32), 128 (30), 102 (24), 84 (48), 71 (14), 70 (33), 60 (35), 59 (23), 56 (46), 55 (14), 43 (29), 42 (16), 41 (21).

Acknowledgements. We thank Dr. Roberto Marianeschi for experimental assistance and the Italian Ministero dell'Università e della Ricerca Scientifica e Tecnologica (MURST) and Consiglio Nazionale delle Ricerche (CNR) for financial support.

REFERENCES AND NOTES

- 1. Fioravanti, S.; Loreto, M. A.; Pellacani, L.; Tardella, P. A. Tetrahedron Lett. 1993, 34, 4353-4354.
- 2. Barani, M.; Fioravanti, S.; Loreto, M. A.; Pellacani, L.; Tardella, P. A. Tetrahedron 1994, 50, 3829-3834.
- 3. Barani, M.; Fioravanti, S.; Pellacani, L.; Tardella, P. A. Tetrahedron 1994, 50, 11235-11238.
- The cyclopropanation of allylic and homoallylic ketals has been thoroughly investigated: Mash, E. A.; Nelson, K. A. Tetrahedron 1987, 43, 679-692; Mash, E. A.; Hemperly, S. B.; Nelson, K. A.; Heidt, P. C.; Van Deusen, S. J. Org. Chem. 1990, 55, 2045-2055; Mash, E. A.; Hemperly, S. B. J. Org. Chem. 1990, 55, 2055-2060.
- 5. Lwowski, W.; Maricich, T. M. J. Am. Chem. Soc. 1965, 87, 3630-3637.
- 6. Lwowski, W.; Mattingly, T. W. Jr. J. Am. Chem. Soc. 1965, 87, 1947-1958.
- 7. Lwowski, W.; Subba Rao, O. Tetrahedron Lett. 1980, 21, 727-730.
- 8. Subbaraj, A.; Subba Rao, O.; Lwowski, W. J. Org. Chem. 1989, 54, 3945-3952.
- 9. *CAUTION*: like all the azides, N₃CO₂Et and N₃C(OEt)NMs might explode and they must be manipulated carefully. Their vapours are also toxic, as well as those of HN₃, involved in the preparation of the imidoyl azide.
- 10. For recent examples of aziridine ring fission see: Molander, G. A.; Stengel, P. J. J. Org. Chem. 1995, 60, 6660-6661 and references therein.
- L-Selectride[®] is reported to stereoselctively reduce aziridino ketones, without aziridine ring opening: Lee, W. K.; Kim, B. C. *Tetrahedron* 1996, 37, 12117-12124.
- 12. Kleinman, E. F. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I.; Heathcock, C. H. Eds; Pergamon: New York, 1991; Vol. 2, p. 893-951 and references therein.
- Rosenlew, E. Chem. Ber. 1906, 39, 2202; Wolfe, S.; Ingold, C. F. J. Am. Chem. Soc. 1983, 105, 7755-7757; Yang, D. T. C.; Cao, Y. H.; Kabalka, G. W. Synth. Commun. 1995, 25, 3695-3699; Rothenberg, G.; Sasson, Y. Tetrahedron 1996, 52, 13641-13648.

(Received in UK 20 December 1996; accepted 13 February 1997)