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Stereoselective synthesis of a C1–C18 fragment of amphidinolides G and H

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1. Introduction

Marine microorganisms belonging to several phyla have attracted the attention of natural product chemists because of their role as the actual producers of many bioactive metabolites, initially found in, and deemed specific to, various marine microorganisms.¹ Amongst these metabolites, the amphidinolides are a family of macrolides isolated from marine dinoflagellates of the Amphidinium genus that are symbiotic to Amphiscolops flatworm species.² These macrolides have been found to display a range of pharmacological properties, most particularly cytotoxicity against several tumoral cell lines. Amphidinolides G 1 and H 2 (Fig. 1, now renamed G_1 and H_1) have been shown to be very potent in this aspect $(IC_{50} < 1 \text{ nm})$, a feature which renders these compounds promising for cancer chemotherapy. In the specific case of amphidinolide H, its pharmacological action has been related to its ability to covalently bind on actin subdomain 4 with subsequent stabilization of the actin filaments.³ In view of these pharmacological properties, it is not surprising that the amphidinolides have attracted considerable interest from the synthetic community. Indeed, many total syntheses of various amphidinolides have already been reported.⁴

ABSTRACT

A stereoselective synthesis of a C1–C18 segment of the structure of the cytotoxic macrolides amphidinolides G and H is reported. The target compound was retrosynthetically disconnected into three fragments. In the synthetic sense, connection of the fragments was made by means of a Stille coupling and a Julia–Kocienski olefination. Precursors from the chiral pool were used as the starting materials. © 2013 Elsevier Ltd. All rights reserved.



For the reasons stated above, we have been interested in performing a stereoselective synthesis of macrolides **1** and **2**. Hydrolytic lactone ring-opening of these two isomeric lactones would give the same open-chain hydroxy acid. Herein, we report a short









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synthesis of a $C_1{-}C_{18}$ fragment common to these two natural compounds. 5,6

Our retrosynthetic analysis for **1**, also valid for **2**, is shown in Fig. 2. Scission of the C_1 –O and C_{18} – C_{19} bonds via lactone ringopening and retroaldol cleavage gives rise to compounds **3** (fragment C_1 – C_{18}), our present target, and **4** (fragment C_{19} – C_{26}), which has previously been prepared by us.^{5e} Further bond scissions in **3** at C6=C7 via Julia–Kocienski olefination⁷ and at C_{13} – C_{14} via Stille coupling⁸ lead to the synthetic subtargets **5–8**.



Fig. 2. Retrosynthetic disconnection of amphidinolide G (1) (for acronyms and abbreviations, see below).

2. Results and discussion

The synthesis of tetrazolyl sulfone **6** was performed as depicted in Scheme 1. Conversion of 1,4-butanediol into the known primary allylic alcohol **9** was performed in 4 steps following literature procedures.⁹ Alcohol protection in **9** afforded **10**, which was desilylated to primary alcohol **11**. The latter was then converted into **6** by means of a standard procedure via sulfide **12**.

The known iodide **7** was prepared as shown in Scheme 2. The chiral and commercially available ester **13** was first converted into the known primary alcohol **14**.^{5f} Silylation of **14** gave **15**, which was then hydrogenolytically debenzylated to **16**.¹⁰ Swern oxidation¹¹ of the alcohol group in **16** followed by Corey–Fuchs homologation¹² of the intermediate aldehyde gave alkyne **17**,¹³ which was then converted into vinyl iodide **7** through the previously reported carbometallation–iodination sequence.^{6a,13}

For the preparation of alcohol **8**, we initially used alcohol **16** as the starting material (Scheme 3). Mesylation of **16** and treatment of the resulting mesylate with potassium cyanide in DMSO gave nitrile **18**, which was subsequently reduced to the corresponding



Scheme 1. Synthesis of sulfone 6. Acronyms and abbreviations: TBDPS, *tert*-butyldiphenylsilyl; MOM, methoxymethyl; DIPEA, *N*,*N*-diisopropyl ethylamine; TBAF, tetrabutylammonium fluoride; PT, 1-phenyl-1*H*-tetrazol-5-yl; DIAD, diisopropyl azodicarboxylate.



Scheme 2. Synthesis of iodide 7. Acronyms and abbreviations: Bn, benzyl.

aldehyde. Homologation of the latter to alkyne **19** was best performed in this case with the aid of the Ohira–Bestmann procedure.¹⁴ Desilylation of **19** gave **20**, projected to be the next member in the sequence leading to **8**.



Scheme 3. Synthesis of epoxy alcohol 8. Acronyms and abbreviations: Ms, methanesulfonyl; DME, 1,2-dimethoxyethane; DIBAL, diisobutylaluminum hydride; DMSO, dimethyl sulfoxide.

However, we found this reaction sequence too long (12 steps from the commercial ester **13**) and eventually replaced it by another more efficient one, also depicted in Scheme 3. The new sequence is based, with some modifications, on the one used by Cid and Pattenden¹⁵ in their route toward amphidinolide B, with methyl hydrogen (*R*)-3-methylglutarate **21** as the chiral starting material. Borane reduction of the carboxy group to primary alcohol, Swern oxidation of the latter to the aldehyde¹⁶ and Ohira–Bestmann homologation provided alkyne **22**.¹⁷ Reduction of the ester to primary alcohol afforded alcohol **20** in only four steps from the commercially available precursor **21**.

Conversion of **20** into epoxy alcohol **23** was performed in four steps.^{6a} Thus, the primary alcohol group of **20** was oxidized to the corresponding aldehyde, followed by Horner–Wadsworth– Emmons¹⁸ olefination of the aldehyde, DIBAL reduction of the resulting conjugated ester to an allylic alcohol and Sharpless epoxidation¹⁹ of the latter. Silylation of **23** gave **24**, which was then subjected to palladium-catalyzed silylstannation²⁰ to yield **25**. Treatment of the latter with TBAF caused both O- and C-desilylation and gave the desired **8**.

The next step was the Stille coupling⁸ of **7** and **8**, which was performed as shown in Scheme 4. Epoxide **8** was dissolved in dry NMP and treated with $Pd_2(dba)_3$ and Ph_3As ,²¹ followed by addition of iodide **7** and CuTC.²² This provided the desired diene **5** in 61% yield. Oxidation of the primary alcohol function in **5** to the corresponding aldehyde in **26** was best performed by means of IBX in DMSO.²³ Good conditions for the final coupling of **26** with sulfone **6** via Julia–Kocienski olefination were found only after extensive experimentation. The best results were found under the so-called Barbier conditions,⁷ which led to the desired compound **3** in 75% yield as a ca. 4:1 *E*/*Z* mixture. Separation of these configurational isomers proved not feasible at this stage. We hope that further advance in the projected synthesis will permit the separation of the two isomers in a later intermediate.



Scheme 4. Synthesis of compound **3**. Acronyms and abbreviations: NMP, *N*-methylpyrrolidone; IBX, iodoxybenzoic acid; KHMDS, potassium hexamethyldisilylazide; DMF, *N*,*N*-dimethylformamide; CuTC, copper(I) thiophene-2-carboxylate.

In summary, compound **3**, which constitutes a C_1-C_{18} fragment of the structures of amphidinolides G/H, has been prepared in a stereoselective way. Coupling of this fragment with a $C_{19}-C_{26}$ fragment previously reported by us^{5e} will hopefully lead to the preparation of the complete structure of these two strongly cytotoxic lactones.

3. Experimental

3.1. General experimental features

See Supplementary data.

3.1.1. (E)-6,13,13-Trimethyl-12,12-diphenyl-2,4,11-trioxa-12-silatetradec-6-ene (**10**). A solution of alcohol **9**⁹ (4.42 g, 12 mmol) in dry CH₂Cl₂ (60 mL) was treated at room temperature under N₂ with DIPEA (6.2 mL, 36 mmol) and MOMCI (1.82 mL, 24 mmol). The mixture was heated at reflux for 2 h. Work-up (extraction with CH₂Cl₂) followed by column chromatography on silica gel (hexanes/ EtOAc, 8:2) afforded **10** (4.85 g, 98%) as a yellowish oil: ¹H NMR δ 7.80–7.75 (4H, br m), 7.50–7.40 (6H, br m), 5.52 (1H, br t, $J \sim 6.8$ Hz), 4.68 (2H, s), 4.00 (2H, br s), 3.77 (2H, t, J=6.2 Hz), 3.44 (3H, s), 2.25 (2H, br q, $J \sim 7.5$ Hz), 1.75 (3H, br s), 1.75–1.70 (2H, m), 1.16 (9H, s); ¹³C NMR δ 134.0 (×2), 131.9, 19.2 (C), 135.5 (×4), 129.5 (×2), 128.1, 127.5 (×4) (CH), 95.2, 73.2, 63.3, 32.3, 24.0 (CH₂), 55.1, 26.8 (×3), 13.9 (CH₃); HR FABMS m/z 435.2344 (M+Na⁺), calcd for C₂₅H₃₆NaO₃Si, 435.2331.

3.1.2. (*E*)-6-(*Methoxymethoxy*)-5-*methylhex*-4-*en*-1-*ol* (**11**). A solution of compound **10** (4.54 g, 11 mmol) in dry THF (70 mL) was treated under N₂ with TBAF (3.45 g, 13.2 mmol). The mixture was stirred at room temperature for 2 h. Removal of all volatiles under reduced pressure was followed by column chromatography of the residue on silica gel (hexanes/Et₂O, 1:1) to yield alcohol **11** (1.88 g, 99%) as a colorless oil: IR ν_{max} 3420 (br, OH) cm⁻¹; ¹H NMR δ 5.46 (1H, br t, $J \sim 7.3$ Hz), 4.62 (2H, s), 3.93 (2H, br s), 3.65 (2H, t, J=6.4 Hz), 3.38 (3H, s), 2.14 (2H, br q, $J \sim 7.3$ Hz), 1.67 (3H, br s), 1.65 (2H, br quint, $J \sim 7$ Hz), 1.50 (1H, br s, OH); ¹³C NMR δ 132.4 (C), 127.8 (CH), 95.4, 73.3, 62.5, 32.4, 24.0 (CH₂), 55.2, 14.0 (CH₃); HR FABMS m/z 197.1165 (M+Na⁺), calcd for C₉H₁₈NaO₃, 197.1153.

3.1.3. (E)-5-[(6-Methoxymethoxy-5-methylhex-4-en-1-yl)thio]-1-phenyl-1H-tetrazole (**12**). A solution of alcohol **11** (1.74 g, 10 mmol) in dry THF (125 mL) was treated at 0 °C under N₂ with Bu₃P (5 mL, 20 mmol), 1-phenyl-1H-tetrazol-5-thiol (3.56 g, 20 mmol) and DIAD (4.92 mL, 25 mmol). The mixture was stirred at 0 °C for 1 h. Work-up (extraction with EtOAc) and column chromatography on silica gel (hexanes/EtOAc, 8:2) furnished sulfide **12** (2.64 g, 79%) as a yellowish oil: ¹H NMR δ 7.55–7.45 (5H, br m), 5.40 (1H, br t, $J \sim$ 7 Hz), 4.57 (2H, s), 3.89 (2H, br s), 3.35 (2H, t, J=7.2 Hz), 1.63 (3H, br s); ¹³C NMR δ 154.2, 133.6, 133.2 (C), 130.0, 129.6 (×2), 126.1, 123.7 (×2) (CH), 95.3, 72.9, 32.7, 28.7, 26.4 (CH₂), 55.1, 14.0 (CH₃); HR FABMS *m/z* 357.1369 (M+Na⁺), calcd for C₁₆H₂₂N₄NaO₂S, 357.1361.

3.1.4. (*E*)-5-[(6-(*Methoxymethoxy*)-5-*methylhex*-4-*en*-1-*yl*)*sulfonyl*]-1-*phenyl*-1*H*-*tetrazole* (**6**). A solution of sulfide **12** (1.67 g, 5 mmol) in EtOH (100 mL) was treated at 0 °C with (NH₄)₆Mo₇O₂₄·4H₂O (1.85 g, 1.5 mmol) and 30% H₂O₂ (5.6 mL, ~50 mmol). The mixture was stirred at room temperature for 16 h. The reaction was then quenched by addition of aqueous Na₂S₂O₃ (1.6 M, 100 mL). Work-up (extraction with EtOAc) and column chromatography on silica gel (hexanes/EtOAc, 8:2) afforded sulfone **6** (1.46 g, 80%) as a colorless oil: IR ν_{max} 1337, 1152 (SO₂) cm⁻¹; ¹H NMR δ 7.70–7.55 (5H, br m), 5.42 (1H, br t, *J*~7.3 Hz), 4.62 (2H, s), 3.93 (2H, br s), 3.72 (2H, br t, *J*~5.5 Hz), 3.37 (3H, s), 2.28 (2H, br q, *J*~7.3 Hz), 2.04 (2H, br quint, *J*~7.3 Hz), 1.67 (3H, br s); ¹³C NMR δ 153.4, 134.7, 133.0 (C), 131.4, 129.7 (×2), 125.1, 124.6 (×2) (CH), 95.5, 72.8, 55.4, 25.9, 21.9 (CH₂), 55.3, 14.1 (CH₃); HR FABMS *m/z* 389.1269 (M+Na⁺), calcd for C₁₆H₂₂N₄NaO₄S, 389.1259.

3.1.5. (S)-[4-(Benzyloxy)-3-methylbutoxy](tert-butyl) diphenylsilane (**15**). A solution of alcohol 14^{5f} (3.88 g, 20 mmol) in dry CH₂Cl₂

(100 mL) was treated at 0 °C under N₂ with Et₃N (4.2 mL, 30 mmol), TPSCl (6.24 mL, 24 mmol) and DMAP (24 mg, 0.2 mmol). The mixture was then stirred at room temperature for 3 h. Work-up (extraction with CH₂Cl₂) and column chromatography on silica gel (hexanes/EtOAc, 95:5) gave silyl ether **15** (8.13 g, 94%) as a colorless oil: $[\alpha]_D - 1.1$ (*c* 1.35, CHCl₃); ¹H NMR δ 7.70–7.65 (4H, m), 7.40–7.25 (11H, br m), 4.45 (2H, s), 3.74 (2H, t, *J*=6.5 Hz), 3.33 (1H, dd, *J*=9, 5.8 Hz), 3.24 (1H, dd, *J*=9, 6.5 Hz), 2.00 (1H, apparent sextuplet, *J*~6.5 Hz), 1.76 (1H, apparent sextuplet, *J*~6.5 Hz), 1.40 (1H, apparent sextuplet, *J*~6.5 Hz), 1.08 (9H, s), 0.93 (3H, d, *J*=6.7 Hz); ¹³C NMR δ 138.8, 134.1 (×2), 127.3, 30.3 (CH), 75.8, 72.9, 62.1, 36.6 (CH₂), 26.9 (×3), 17.3 (CH₃); HR FABMS *m/z* 455.2365 (M+Na⁺), calcd for C₂₈H₃₆NaO₂Si, 455.2382.

3.1.6. (S)-4-(*tert-Butyldiphenylsilyloxy*)-2-*methylbutan*-1-ol (**16**). Palladium hydroxide (20%, Degussa-type, 3 g) was suspended in EtOH (250 mL) under a H₂ atmosphere. After stirring at room temperature and ambient pressure for 15 min, a solution of compound **15** (6.49 g, 15 mmol) in EtOH (20 mL) was added via syringe. The mixture was then stirred for 90 min. When the starting compound was consumed (TLC monitoring), the reaction mixture was filtered through a Celite pad. The pad was then washed with EtOAc. The organic layers were evaporated to dryness and the residue was subjected to column chromatography on silica gel (hexanes/EtOAc, 3:1). This provided **16** (4.52 g, 88%) as a colorless oil having the reported physical and spectral properties.¹⁰

3.1.7. (*R*)-tert-Butyl(3-methylhex-5-ynyloxy) diphenylsilane (**19**). A solution of alcohol **16** (3.08 g, 9 mmol) in dry CH₂Cl₂ (40 mL) was treated at 0 °C under N₂ with Et₃N (2.5 mL, 18 mmol), MsCl (1.05 mL, 13.5 mmol), and DMAP (12 mg, 0.1 mmol). The mixture was then stirred at room temperature for 2 h. Work-up (extraction with CH₂Cl₂) gave a crude mesylate, which was dissolved in dry DMSO (45 mL) and treated under N₂ at room temperature with KCN (1.76 g, 27 mmol). The mixture was then stirred at 60 °C for 4 h. Work-up (extraction with Et₂O) afforded crude nitrile **18**, which was used as such in the next step: IR ν_{max} 2245 (C \equiv N) cm⁻¹; ¹H NMR δ 7.70–7.65 (4H, m), 7.45–7.35 (6H, m), 3.72 (2H, t, *J*=6 Hz), 2.38 (1H, dd, *J*=16.5, 5.4 Hz), 2.28 (1H, dd, *J*=16.5, 7 Hz), 2.15 (1H, m), 1.80–1.50 (2H, br m), 1.08 (3H, d, overlapped), 1.08 (9H, s).

A solution of 18, as obtained above, in dry hexane (50 mL) was treated under N₂ at -78 °C with DIBAL (1 M solution in hexane, 12 mL, 12 mmol). The mixture was then stirred at -78 °C for 15 min. Work-up (extraction with Et₂O) gave an oily residue, which was dissolved in MeOH (15 mL) and treated at room temperature under N₂ with K₂CO₃ (2.2 g, 16 mmol) and freshly prepared Ohira-Bestmann's reagent (1.85 g, 9.6 mmol). The mixture was stirred overnight at room temperature. Work-up (extraction with Et₂O) and column chromatography on silica gel (hexanes-Et₂O, 8:2) furnished alkyne 19 (2.42 g, 77% overall for the four steps from 16) as a yellowish oil: IR v_{max} 3300 (C=C) cm⁻¹; ¹H NMR δ 7.70–7.65 (4H, m), 7.45–7.35 (6H, m), 3.72 (2H, t, J=6.4 Hz), 2.19 (1H, ddd, J=16.5, 6, 2.5 Hz), 2.10 (1H, ddd, J=16.5, 6.5, 2.5 Hz), 2.00-1.85 (2H, m), 1.73 (1H, apparent sextuplet, $J \sim 6.5$ Hz), 1.49 (1H, apparent sextuplet, $J \sim 6.5$ Hz), 1.06 (9H, s), 1.00 (3H, d, J=6.6 Hz); ¹³C NMR δ 133.3 (×2), 83.2, 19.4 (C), 135.6 (×4), 129.6 (×2), 127.6 (×4), 69.2, 29.1 (CH), 62.0, 38.5, 25.7 (CH₂), 26.9 (×3), 19.3 (CH₃).

3.1.8. (*R*)-3-*Methylhex*-5-*yn*-1-*ol* (**20**). A solution of alkyne **19** (2.1 g, 6 mmol) in dry THF (40 mL) was treated under N₂ with TBAF (1.72 g, 6.6 mmol). The mixture was stirred at room temperature for 2 h. Removal of all volatiles under reduced pressure was followed by column chromatography of the residue on silica gel (hexanes/Et₂O, 1:1) to yield **20** (639 mg, 95%) as a yellowish oil: $[\alpha]_D$ +4.2 (*c* 0.8, CHCl₃); IR ν_{max} 3400 (br, OH), 3300 (C=C) cm⁻¹; ¹H NMR δ 3.67

(2H, m), 2.18 (1H, ddd, *J*=16.5, 6, 2.5 Hz), 2.13 (1H, ddd, *J*=16.5, 6.5, 2.5 Hz), 2.00 (1H, br s, OH), 1.96 (1H, t, *J*=2.5 Hz), 1.84 (1H, apparent sextuplet, $J \sim 6.5$ Hz), 1.69 (1H, apparent sextuplet, $J \sim 6.5$ Hz), 1.48 (1H, apparent sextuplet, $J \sim 6.5$ Hz), 1.00 (3H, d, *J*=6.8 Hz); ¹³C NMR δ 82.9 (C), 69.4, 29.1 (CH), 60.7, 38.6, 25.7 (CH₂), 19.4 (CH₃); HR EIMS *m/z* (rel int.) 97.0645 (M⁺–Me, 11), 91 (26), 55 (100), calcd for C₇H₁₂O–Me, 97.0653.

3.1.9. tert-Butyl (2S,3S)-3-[(R)-2-methylpent-4-ynyl] oxiran-2ylmethoxy diphenylsilane (24). A solution of alcohol 23^{6a} (617 mg, 4 mmol) in dry CH₂Cl₂ (40 mL) was treated at room temperature under N₂ with imidazole (408 mg, 6 mmol) and TPSCl (1.25 mL, 4.8 mmol). The mixture was then stirred at room temperature for 1 h. Work-up (extraction with CH_2Cl_2) and column chromatography on silica gel (hexanes/EtOAc, 95:5) gave silyl ether 24 (1.54 g, 98%) as a colorless oil: $[\alpha]_{D}$ –10.8 (c 0.65, CHCl₃); IR ν_{max} 3296 (C=C) cm $^{-1};\,^{1}$ H NMR δ 7.70–7.65 (4H, m), 7.45–7.35 (6H, m), 3.78 (2H, m), 2.91 (1H, td, J=4.5, 2 Hz), 2.84 (1H, td, J=6, 2 Hz), 2.24 (1H, ddd, J=16.5, 6, 2.5 Hz), 2.18 (1H, ddd, J=16.5, 6.5, 2.5 Hz), 1.99 (1H, t, *J*=2.5 Hz), 1.92 (1H, m), 1.69 (1H, dt, *J*=14, 6 Hz), 1.45 (1H, ddd, *J*=14, 8.3, 5.5 Hz), 1.08 (3H, d, overlapped), 1.07 (9H, s); $^{13}\mathrm{C}$ NMR δ 133.3 (×2), 82.7, 19.2 (C), 135.6 (×2), 135.5 (×2), 129.8 (×2), 127.7 (×4), 64.2, 58.6, 54.9, 30.6 (CH), 69.6, 37.9, 26.1 (CH₂), 26.8 (×3), 19.3 (CH₃); HR FABMS *m*/*z* 393.2269 (M+H⁺), calcd for C₂₅H₃₃O₂Si, 393.2249.

3.1.10. Silylstannane 25. A solution of 24 (1.18 g, 3 mmol) in dry, degassed DME (25 mL) was treated at room temperature under N₂ with PhMe₂Si–SnMe₃ (900 mg, \sim 3 mmol),^{20c} Ph₃P (140 mg, 0.54 mmol), and Pd(OAc)₂ (27 mg, 0.12 mmol). The mixture was then stirred at 35 °C for 7 d. After consumption of the starting material (TLC monitoring), the mixture was evaporated under reduced pressure and the residue was subjected to column chromatography on silica gel (hexanes/EtOAc, 98:2) to yield 25 (1.39 g, 67%) as a colorless oil: $[\alpha]_D$ –4.9 (*c* 0.8, CHCl₃); ¹H NMR δ 7.75–7.70 (4H, m), 7.56 (2H, m), 7.50-7.35 (9H, br m), 6.50 (1H, br s), 3.80 (2H, m), 2.90 (1H, td, J=4.5, 2 Hz), 2.85 (1H, td, J=6, 2 Hz), 2.50 (1H, dd, J=12.5, 6.3 Hz), 2.30 (1H, ddd, J=12.5, 7.5 Hz), 1.80-1.70 (2H, m), 1.36 (1H, ddd, *J*=14, 8.3, 5.5 Hz), 1.10 (9H, s), 0.98 (3H, d, *J*=6.5 Hz), 0.40 (6H, s), 0.09 (9H, s); ¹³C NMR δ 167.2, 139.6, 133.3 (×2), 19.2 (C), 143.4 (×2), 135.6 (×4), 134.1 (×2), 129.8 (×2), 128.9, 127.7 (×5), 59.0, 55.0, 30.5 (CH), 64.3, 55.8, 38.5 (CH₂), 26.8 (×3), 19.4, -0.06 $(\times 2)$, -6.8 $(\times 3)$ (CH₃).

3.1.11. (2S,3S)-3-[(S)-2-(Methyl-4-(trimethylstannyl) pent-4-enyl)oxiran-2-yl]methanol (8). A solution of compound 25 (1.38 g, 2 mmol) in dry DMSO (50 mL) was treated under N2 with TBAF (1 M solution in THF, 12 mL, 12 mmol). The mixture was stirred at 80 °C for 10 min and then at room temperature for 20 min. Work-up (extraction with Et₂O) and column chromatography on silica gel (hexanes/Et₂O, 1:1) furnished alcohol **8**¹⁵ (415 mg, 65%) as a colorless oil: $[\alpha]_D$ –21.8 (c 0.8, CHCl₃); IR ν_{max} 3400 (br, OH) cm⁻¹; ¹H NMR δ 5.63 (1H, br s), 5.20 (1H, br d, J=2.5 Hz), 3.91 (1H, dd, J=12.5, 2.4 Hz), 3.63 (1H, dd, J=12.5, 4.3 Hz), 2.98 (1H, td, J=6, 2.2 Hz), 2.89 (1H, m), 2.34 (1H, dd, J=13.5, 6.6 Hz), 2.18 (1H, ddd, J=13.5, 7.7 Hz), 2.00 (1H, br s, OH), 1.75 (1H, m), 1.65 (1H, m), 1.30 (1H, m), 0.94 (3H, d, *J*=6.5 Hz), 0.14 (9H, s); ¹³C NMR δ 154.3 (C), 58.9, 54.6, 30.7 (CH), 126.2, 61.6, 49.0, 38.4 (CH₂), 19.6, $-9.5 (\times 3)$ (CH₃); HR EIMS m/z (rel int.) 305.0575 (M⁺-Me, 22), 165 (100), calcd for C₁₂H₂₄O₂Sn-Me, 305.0558 (value calculated for ¹²⁰Sn, the most abundant isotope of tin).

3.1.12. (2S,3S)-(3-[(2R,7S,E)-9-tert-Butyldiphenylsilyl-oxy-2,6,7-trimethyl-4-methylenenon-5-enyl]oxiran-2-yl)methanol (**5**). A solution of epoxy alcohol**8**(383 mg, 1.2 mmol) in dry, degassed NMP (10 mL) was treated under N₂ at room temperature with Ph₃As (220 mg, 0.72 mmol) and Pd₂(dba)₃ (165 mg, 0.18 mmol). The

reaction mixture was stirred under N2 at room temperature for 15 min. Addition first of iodoalkene 7 (600 mg, 1.25 mmol) in dry, degassed NMP (15 mL) and then of CuTC (344 mg, 1.8 mmol) was followed by stirring under N2 at 35 °C for 40 min. Work-up (extraction with Et₂O) and column chromatography on silica gel (hexanes/EtOAc, 8:2) furnished compound 5 (371 mg, 61%) as a yellowish oil: $[\alpha]_D - 7.4$ (*c* 1.24, CHCl₃); IR ν_{max} 3440 (br, OH) cm⁻¹; ¹H NMR δ 7.70–7.65 (4H, m), 7.45–7.35 (6H, m), 5.57 (1H, br s), 4.96 (1H, br s), 4.79 (1H, br s), 3.90 (1H, br d, J~12 Hz), 3.70–3.60 (3H, m), 2.91 (1H, td, J=6, 2 Hz), 2.86 (1H, m), 2.40 (1H, apparent sextuplet, *J*~7 Hz), 2.10 (1H, dd, *J*=12.5, 6.5 Hz), 1.94 (2H, br dd, *J*~13.5, 8 Hz, overlapping OH signal), 1.80–1.55 (5H, br m), 1.66 (3H, s), 1.07 (9H, s), 1.02 (3H, d, J=7 Hz), 0.90 (3H, d, J=6.8 Hz); ¹³C NMR δ 144.3, 142.4, 134.1 (×2), 19.2 (C), 135.5 (×4), 129.5 (×2), 127.6 (×4), 125.2, 58.9, 54.7, 39.7, 29.6 (CH), 114.5, 62.4, 61.7, 45.8, 38.4, 37.7 (CH₂), $26.9 (\times 3)$, 19.7, 19.5, 14.1 (CH₃); HR EIMS m/z (rel int.) 506.3211 (M⁺, 3), 449 (14), 431 (19), 199 (100), calcd for C₃₂H₄₆O₃Si, 506.3216.

3.1.13. Compound **3**. A solution of alcohol **5** (355 mg, 0.7 mmol) in dry DMSO (10 mL) was treated under N₂ at room temperature with IBX (392 mg, 1.4 mmol, 2 equiv). The reaction mixture was stirred under N₂ at 50 °C for 1 h. During the work-up, extraction with Et₂O had to be repeated 8–10 times, due to the slow extraction of the product with this solvent. Column chromatography on silica gel (hexanes/EtOAc, 7:3) provided aldehyde **26**, pure enough for use in the next step.

The material from the previous step and sulfone 6 (385 mg. 1.05 mmol) were dissolved under N_2 in dry DMF (15 mL). The solution was then cooled to -78 °C and treated dropwise with KHMDS (0.5 M in toluene, 2 mL, 1 mmol). The reaction mixture was then stirred overnight under N₂ at -78 °C. Work-up (extraction with Et₂O) and column chromatography on silica gel (hexanes/ EtOAc, 8:2) afforded compound 3 (338 mg, 75% overall for the two steps) as a yellowish oil. NMR analysis revealed that the compound was an inseparable ~80:20 mixture of E/Z stereoisomers. A small sample could be partially concentrated in the E isomer for analytical purposes: oil; ¹H NMR (signals of the major *E* stereoisomer) δ 7.70–7.65 (4H, m; TPS aromatic), 7.45–7.35 (6H, m; TPS aromatic), 5.90 (1H, dt, *J*=15.5, 6.5 Hz; H-6), 5.56 (1H, br s; H-14), 5.46 (1H, m; H-3), 5.20 (1H, dd, J=15.5, 8 Hz; H-7), 4.95 (1H, br s; C=CH₂), 4.76 (1H, br s; C=CH₂), 4.63 (2H, s; CH₂OMe), 3.94 (2H, br s; H-1/1'), 3.62 (2H, m; H-18/18'), 3.38 (3H, s; OMe), 3.00 (1H, dd, J=8, 2 Hz; H-8), 2.76 (1H, td, J=5.8, 2 Hz; H-9), 2.38 (1H, m; H-16), 2.20-2.05 (5H, br m; H-4/4'/5/5'/12), 1.92 (1H, br dd, J=13.5, 7.7 Hz; H-12'), 1.67 (3H, s; MeC₂ or MeC₁₅), 1.64 (3H, s; MeC₁₅ or MeC₂), 1.80–1.50 (5H, br m; H-10/10'/11/17/17'), 1.06 (9H, s; ^tBu), 1.01 (3H, d, *J*=6.8 Hz; *MeC*₁₁ or *MeC*₁₆), 0.89 (3H, d, *J*=6.8 Hz; *MeC*₁₆ or *MeC*₁₁); ¹³C NMR (signals of the major *E* stereoisomer) δ 144.4, 142.3, 132.4 (×3), 19.2 (C), 135.5 (×4), 134.1, 129.5 (×2), 128.1, 127.6 (×4), 127.5, 125.3, 59.1 (×2), 39.7, 29.7 (CH), 114.4, 95.4, 73.2, 62.4, 45.9, 38.9, 37.7, 32.1, 27.2 (CH₂), 55.3, 29.6, 26.9 (×3), 19.7, 19.5, 14.1 (CH₃) (for atom numbering, see Fig. 2).

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Supplementary data

Supplementary data associated with this article (graphical NMR spectra) can be found in the online version. Supplementary data related to this article can be found at http://dx.doi.org/10.1016/j.tet.2013.02.062.

References and notes

- (a) König, G. M.; Kehraus, S.; Seibert, S. F.; Abdel-Lateff, A.; Müller, D. *Chem-BioChem* **2006**, *7*, 229–238; (b) Pandian, S. R. K.; ManiKanth, S. B.; Kalishwaralal, K.; Deepak, V.; Sangiliyandi, G. Adv. Environ. Res. **2011**, *9*, 153–179; (c) Blunt, J. W.; Copp, B. R.; Keyzers, R. A.; Munro, M. H. G.; Prinsep, M. R. Nat. Prod. Rep. **2012**, *29*, 144–222.
- For reviews partially or totally centered on the chemistry and/or biology of amphidinolides, see: (a) Kobayashi, J.; Ishibashi, M. Chem. Rev. 1993, 93, 1753–1769; (b) Maranda, L.; Shimizu, Y. J. Phycol. 1996, 32, 873–979; (c) Ishibashi, M.; Kobayashi, J. Heterocycles 1997, 44, 543–572; (d) Chakraborty, T. K.; Das, S. Curr. Med. Chem. 2001, 1, 131–149; (e) Kobayashi, J.; Shimbo, K.; Kubota, T.; Tsuda, M. Pure Appl. Chem. 2003, 75, 337–342; (f) Kobayashi, J.; Tsuda, M. Nat. Prod. Rep. 2004, 21, 77–93; (g) Kobayashi, J.; Tsuda, M. Phytochem. Rev. 2004, 3, 267–274; (h) Kang, E. J.; Lee, E. Chem. Rev. 2005, 105, 4348–4378; (i) Nicholas, G. M.; Phillips, A. J. Nat. Prod. Rep. 2006, 23, 79–99; (j) Kobayashi, J.; Kubota, T. J. Nat. Prod. 2007, 70, 451–460; (k) Kobayashi, J. J. Antibiot. 2008, 61, 271–284; (l) Morris, J. C.; Phillips, A. J. Nat. Prod. Rep. 2009, 26, 245–265.
- (a) Usui, T.; Kazami, S.; Dohmae, N.; Mashimo, Y.; Kondo, H.; Tsuda, M.; Terasaki, A. G.; Ohashi, K.; Kobayashi, J.; Osada, H. *Chem. Biol.* 2004, *11*, 1269–1277; (b) Saito, S.; Feng, J.; Kira, A.; Kobayashi, J.; Ohizumi, Y. *Biochem. Biophys. Res. Commun.* 2004, *320*, 961–965.
- 4. For a review, see: Fürstner, A. Isr. J. Chem. 2011, 51, 329-345.
- Fragments of the structures of several members of the amphidinolide G/H family have been prepared using various methodologies: (a) Chakraborty, T. K.; Suresh, V. R. Tetrahedron Lett. **1998**, 39, 9775–7778; (b) Chakraborty, T. K.; Suresh, V. R. Tetrahedron Lett. **1998**, 39, 9109–9112; (c) Liesener, F. P.; Kalesse, M. Synlett **2005**, 2236–2238; (d) Liesener, F. P.; Jannsen, U.; Kalesse, M. Synthesis **2006**, 2590–2602; (e) Formentín, P.; Murga, J.; Carda, M.; Marco, J. A. Tetrahedron: Asymmetry **2006**, 17, 2938–2942; (f) Deng, L.; Ma, Z.; Zhang, Y.; Zhao, G. Synlett **2007**, 87–90; (g) Petri, A. F.; Schneekloth, J. S., Jr.; Mandal, A. K.; Crews, C. M. Org. Lett. **2007**, 9, 3001–3004; (h) Hara, A.; Morimoto, R.; Ishikawa, Y.; Nishiyama, S. Org. Lett. **2011**, 13, 4036–4039.
- 6. For total syntheses of members of the amphidinolide G/H family, see: (a) Fürstner, A.; Bouchez, L. C.; Morency, L.; Funel, J. A.; Liepins, V.; Porée, F. H.; Gilmour, R.; Laurich, D.; Beaufils, F.; Tamiya, M. Chem.-Eur, J. 2009, 15, 3983–4010; (b) Hara, A.; Morimoto, R.; Iwasaki, Y.; Saitoh, T.; Ishikawa, Y.; Nishiyama, S. Angew. Chem., Int. Ed. 2012, 51, 9877–9880.
- For a review, see: Blakemore, P. R. J. Chem. Soc., Perkin Trans. 1 2002, 2563–2585.
 (a) Farina, V.; Krishnamurthy, V.; Scott, W. J. Org. React. 1997, 50, 1–652; (b)
- Fugami, K.; Kosugi, M. In *Cross-coupling Reactions*; Miyaura, N., Ed.; Springer: Berlin, 2002; pp 87–130; (c) Mitchell, T. N. In *Metal-catalyzed Cross-coupling Reactions*, 2nd ed.; de Meijere, A., Diederich, F., Eds.; Wiley-VCH: Weinheim, Germany, 2004; Vol. 1, pp 125–161.
- (a) Freeman, F.; Kim, D. S. H. L. J. Org. Chem. **1992**, 57, 1722–1727; (b) Kobayashi,
 J.; Hatakeyama, A.; Tsuda, M. Tetrahedron **1998**, 54, 697–704; (c) Poulin, J.;
 Grisé-Bard, C. M.; Barriault, L. Angew. Chem., Int. Ed. **2012**, 51, 2111–2114.
- 10. Uenishi, J.; Kawahama, R.; Yonemitsu, O. J. Org. Chem. 1997, 62, 1691-1701.
- 11. Tidwell, T. T. Org. React. 1990, 39, 297–572.
- 12. Corey, E. J.; Fuchs, P. L. Tetrahedron Lett. 1972, 13, 3769-3772.
- 13. Kurosawa, S.; Mori, K. Eur. J. Org. Chem. 2000, 955–962.
- 14. (a) Pietruszka, J.; Witt, A. Synthesis **2006**, 4266–4268; (b) Patil, U. D. Synlett **2009**, 2880–2881.
- 15. Cid, M. B.; Pattenden, G. Tetrahedron Lett. 2000, 41, 7373–7378.
- Hong, B.-C.; Chen, F.-L.; Chen, S.-H.; Liao, J.-H.; Lee, G.-H. Org. Lett. 2005, 7, 557–560.
- 17. Kobayashi, K.; Fujii, Y.; Hayakawa, I.; Kigoshi, H. Org. Lett. 2011, 13, 900-903.
- 18. Maryanoff, B. E.; Reitz, A. B. Chem. Rev. 1989, 89, 863-927.
- 19. Katsuki, T.; Martín, V. S. Org. React. 1996, 48, 1–300.
- (a) Chenard, B. L.; Van Zyl, C. M. J. Organomet. Chem. 1986, 51, 3561–3566; (b) Mitchell, T. N.; Wickenkamp, R.; Amamria, A.; Dicke, R.; Schneider, U. J. Org. Chem. 1987, 52, 4868–4874; (c) Ritter, K. Synthesis 1989, 218–221.
- 21. Faust, R.; Goebelt, B. J. Prakt. Chem. 1998, 340, 90-93.
- (a) The beneficial effects of the addition of Cu(l) salts, particularly CuTC, have been advocated by Liebeskind: Allred, G. D.; Liebeskind, L. S. J. Am. Chem. Soc. 1996, 118, 2748–2749; (b) Innitzer, A. Synlett 2005, 2405–2406.
- Gallen, M. J.; Goumont, R.; Clark, T.; Terrier, F.; Williams, C. M. Angew. Chem., Int. Ed. 2006, 45, 2929–2934 and references therein.