# Synthesis and Antitumor Activity of Novel Mitomycin Derivatives Containing Functional Groups at the C-6-Methyl Position

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A series of C-6-substituted methyl mitomycins was synthesized and evaluated for anticellular and antitumor activities. These novel compounds were prepared by Michael addition of various alcohols or thiols to 6-demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-methylidenemitosanes followed by treatment with NH<sub>3</sub> or MeOH/K<sub>2</sub>CO<sub>3</sub>. Most compounds were potent against HeLa S<sub>3</sub>, and some of them showed superior activity to that of mitomycin C (MMC) against P388 leukemia and sarcoma 180 in mice. In addition, some compounds exhibited remarkable activity against MMCresistant P388 in mice. FAB-MS spectra of these mitomycin derivatives showed the elimination of the C-6-methyl substituents from the mitomycin skeletons to form quinonemethides. Interestingly, treatment of 6-demethyl-6-[[(2-pyrimidinyl)thio]methyl]mitomycin C (12v) with diethylamine afforded 6-demethyl-6-[(diethylamino)methyl]mitomycin C (31) in good yield. These results suggested that the C-6-substituted methyl mitomycins would have different biological character from that of MMC.

### Introduction

Mitomycins are well known to be potent antitumor antibiotics, produced by various Streptomyces cultures.1 Among these compounds, mitomycin C (MMC, 1) has been extensively used in cancer chemotherapy against a variety of solid tumors, but its use is limited by detrimental side effects, such as myelosuppression and gastrointestinal damage. Consequently, about a thousand derivatives intended to have less toxicity and more efficacy have been synthesized by modification mainly at the C-7, N-1a, and C-10 positions.<sup>2</sup> Some of the C-7-substituted mitomycins have been reported to possess superior activity to that of MMC against experimental tumors and are now under clinical investigation.<sup>3</sup> Considering the synthesis of derivatives which have quite a new concept, modification of mitomycins at another position, especially the C-6methyl position, may be a useful and fascinating strategy to serve desired derivatives for the reasons cited below. The C-6-methyl position is suitable to install additional functions because the methyl group does not play an important role in an activation process of mitomycins.<sup>1</sup> Consequently, by the introduction of functional groups into the C-6-methyl position, increasing (decreasing) the lipophilicity<sup>4</sup> or installation of an additional alkylation site<sup>5</sup> should be achieved without affecting the main action of mitomycins. However, these modifications have scarcely been accomplished because of the instability of the mitomycin skeleton under many reaction conditions.6 During the course of our study of mitomycin chemistry, we reported the novel replacement of hydrogen at the C-6methyl position by deuterium using 6-demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-methylidenemitosane (7).7 Taking the reactivity of 7 as a Michael acceptor into account, nucleophilic introduction of functional groups

Figure 1. Structure of natural and C-6-substituted methyl mitomycins and 6-methylidene intermediate 7.

into the C-6-methyl position should become feasible, and subsequent conversion to the mitomycin skeletons would afford quite novel mitomycin derivatives containing a functional group at the C-6-methyl position. Herein, we describe the synthesis of these C-6-substituted methyl mitomycin derivatives and their antitumor activities.

#### Chemistry

We have already reported the synthesis of 6-demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-methylidenemitosane (7), a reactive Michael acceptor, by the seleno oxidation of 8 prepared from mitomycin A (MMA, 2) in three steps. We therefore first tried to react the 6-methylidene intermediate 7 with MeOH for the introduction of a methoxy group into the C-6-methyl position. As mentioned previously,7 since 7 could not be isolated pure from the reaction mixture

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#### Scheme 1 a

a (a) Ac<sub>2</sub>O, pyridine, CHCl<sub>3</sub>; (b) KOH (catalytic), ethylene glycol, THF; (c) PhSeBr, Et<sub>3</sub>N, THF; (d) mCPBA, K<sub>2</sub>CO<sub>3</sub>; (e) ROH or RSH; (f) NH<sub>3</sub>, MeOH; (g) MeOH, KOH.

due to the instability on silica gel, the reaction mixture containing 7 was used for further reactions without any treatment (sequential method). 6-Demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-(methoxymethyl)mitosane (9a) was obtained easily by adding MeOH to the reaction mixture containing 7 at room temperature. A solution of NH<sub>3</sub> in MeOH (method A) was added to the resulting mixture,7 and after purification, 6-demethyl-6-(methoxymethyl)mitomycin C (11a) was obtained in 28% yield based on 8 in three steps. On the other hand, 6-demethyl6-(methoxymethyl)mitomycin A (13) was also prepared from 9a on treatment with KOH in MeOH (10% yield based on 8) (Scheme 1).7

Similar 6-demethyl-6-(methoxymethyl)mitomycins having various mitomycin skeletons were also prepared to evaluate the antitumor activity. 7,7-(Ethylenedioxy)-6,7dihydro-6-(phenylseleno)mitosanes 14 and 18 having mitomycin F (MMF, 4) and mitomycin B (MMB, 6) skeletons were prepared by a similar procedure as that described in the synthesis of 8, i.e., (1) formation of the ethylidene

a (a) mCPBA,  $K_2CO_3$ ; (b) isolation; (c) ROH, Triton B (NEt<sub>3</sub> when ROH = n-PrOH), CHCl<sub>3</sub>; (d) NH<sub>3</sub>, MeOH; (e) DDQ, H<sub>2</sub>O-CHCl<sub>3</sub>. acetal at the C-7 position upon treatment with ethylene glycol in the presence of catalytic KOH and (2) introduction of the phenylseleno group into the C-6 position upon treatment with PhSeBr in the presence of NEt<sub>3</sub>. The selenoxide fragmentation of 14 and 18 by seleno oxidation using mCPBA in the presence of  $K_2CO_3$  followed by treatment with MeOH afforded mixtures containing 16 and 20, respectively. They were treated with a solution of NH<sub>3</sub> in MeOH and afforded 17 and 21 in 63% and 20% yields based on 14 and 18, respectively (sequential method).

For the purpose of introducing other alkoxy groups at the C-6-methyl position, the reaction of the 6-methylidene intermediate 7 with several alcohols was next attempted. The Michael additions of other primary alcohols to 7 were also successful with the same method and afforded the corresponding 6-demethyl-6-(alkoxymethyl)mitomycin C 11 after treatment with NH<sub>3</sub> in MeOH. However, in the case of using secondary alcohols as nucleophiles, the method mentioned above was unsuccessful. In order to obtain the adducts with secondary alcohols, the use of isolated 78 was attempted (improved method). As shown in Scheme 2, isolated 7 was reacted with i-PrOH in the presence of catalytic amounts of Triton B, and this reaction successfully afforded the adduct 9c in 52% yield. This was treated with NH3 in MeOH to afford 6-demethyl-6-(isopropoxymethyl)mitomycin C (11c) in 60% yield. For the introduction of the hydroxy group at the C-6-methyl position, adduct 9g prepared from 7 and 3,4-dimethoxybenzyl alcohol was oxidized by DDQ9 to form 22, which was converted into 6-demethyl-6-(hydroxymethyl)mitomycin C (11g).10 Compound 11g is the key compound for transformations to the C-6-methyl position intended to Scheme 3

control lipophilicity or elimination efficiency since the hydroxy group of 11g can be modified (e.g., by acylation) rather easily.

These results are summarized in Table 1.

Furthermore, we tried to introduce various thiols into the C-6-methyl position (Schemes 1 and 3). The adducts 10 were obtained generally in moderate or good yields by the reaction of 7 with slight excesses of thiols without additional bases (for the sequential method in Scheme 1) or in the presence of NEt<sub>3</sub> (for the improved method in Scheme 3). On the contrary, a subsequent conversion step to 12 having the MMC skeleton on treatment with NH<sub>3</sub> in MeOH (method A) often did not afford the desired product 12 in acceptable yield. In the case of the adduct 10 having the thio groups with aromatic heterocycles, the reaction caused the decomposition of substrates and did not afford 12 at all. So, we examined the condition of the C-7-amination/N-1a-deacetylation process using 6-demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-[(phenylthio)methyl]mitosane (10i) as a model compound. The complete decomposition of 10 having the thio groups with aromatic heterocycles at the C-6-methyl position suggested that  $\beta$ -elimination had occurred during the basic reaction sequence due to the high elimination efficiency of thiols. To prevent  $\beta$ -elimination, NH<sub>4</sub>OAc was used instead of NH<sub>3</sub> to suppress the basicity of the reaction system (method B). By this method, the yield of 6-demethyl-6-[(phenylthio)methyl]mitomycin C (12i) was slightly improved (from 29% to 36%). Further improvement was achieved by using anhydrous THF and anhydrous NH<sub>3</sub> gas that were dried over sodium, respectively (method C). Though the reaction rate of this method was very slow because of the poor solubility of NH<sub>3</sub> in anhydrous THF, the product was obtained in good yield (63%). In addition, the formation of the 1a-acetyl derivative of 12i11 was observed at the early stage of the reaction, which indicated the rate-determining step of this conversion was the N-1adeacetylation.

Using the methods described above, various mitomycin derivatives having the thio groups at the C-6-methyl position were prepared. As shown in Table 2, most reactions afforded the corresponding products in reasonable yields.

In the case of the reaction with 2-aminoethanethiol or 2-aminothiophenol, compound 7 directly afforded unique cyclization compounds 23 or 25. N-1a-Deacetylation of 23 and 25 was performed easily by ammonolysis and gave 24 and 26, respectively.

Table 1. Preparation and Anticellular Activity of 11, 13, 17, and 21

			yield (%)				
compd	MM skeleton	w	9, <sup>a</sup> 16, and 20	$product^b$	empirical formula <sup>c</sup>	<sup>1</sup> H NMR (δ, ppm) <sup>d</sup> C-6 subst	HeLa S <sub>3</sub> ¢ IC <sub>50</sub> (μΜ)
11a	MMC	MeO	f, g	28	C <sub>16</sub> H <sub>20</sub> N <sub>4</sub> O <sub>6</sub> ·0.3H <sub>2</sub> O	3.31 (s, 3 H), 4.37 (d, J = 12.6 Hz, 1 H), 4.44 (d, J = 12.6 Hz, 1 H)	2.6
13	MMA	MeO	f, g	10	$C_{17}H_{21}N_3O_7$	3.37 (s, 3 H), 4.21 (d, $J = 10.1$ Hz, 1 H), 4.26 (d, $J = 10.1$ Hz, 1 H)	0.12
17	PFM	MeO	f, g	59	$C_{17}H_{22}N_4O_6$	3.17 (s, 3 H), 4.57 (d, $J = 11.5$ Hz, 1 H), 4.61 (d, $J = 11.5$ Hz, 1 H)	>100
21	MMD	MeO	f, g	20	$C_{16}H_{20}N_4O_6$	3.16 (s, 3 H), 4.50 (d, $J = 13.0$ Hz, 1 H), 4.54 (d, $J = 13.0$ Hz, 1 H)	>82
11b	MMC	n-PrO	34 <sup>h</sup>	18	C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O <sub>6</sub> -0.8H <sub>2</sub> O	$0.80  ext{ (t, } J = 7.4  ext{ Hz, } 3  ext{ H), } 1.51  ext{ (m, } 2  ext{ H), } 3.34  ext{ (t, } J = 6.6  ext{ Hz, } 2  ext{ H), } 4.60  ext{ (d, } J = 11.6  ext{ Hz, } 1  ext{ H), } 4.67  ext{ (d, } J = 11.6  ext{ Hz, } 1  ext{ H)}$	5.3
11c	MMC	i-PrO	52 <sup>h</sup>	60	C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O <sub>6</sub> <sup>i</sup>	1.17 (d, $J$ = 6.1 Hz, 3 H), 1,18 (d, $J$ = 6.1 Hz, 3 H), 3.59–3.65 (m, 1 H), 4.40 (d, $J$ = 12.5 Hz, 1 H), 4.49 (d, $J$ = 12.5 Hz, 1 H)	2.5
11 <b>d</b>	MMC	n-BuO	f, g	23	$\mathrm{C}_{19}\mathrm{H}_{26}\mathrm{N}_4\mathrm{O}_6{}^i$	0.91 (t, $J = 7.6$ Hz, 3 H), 1.30–1.58 (m, 4 H), 3.41 (t, $J = 6.6$ Hz, 2 H), 4.40 (d, $J = 12.6$ Hz, 1 H), 4.47 (d, $J = 12.6$ Hz, 1 H)	19
lle	MMC	PhCH <sub>2</sub> O	f, g	27	$C_{22}H_{24}N_4O_6$	4.46 (s, 2 H), 4.49 (d, <i>J</i> = 12.6 Hz, 1 H), 4.55 (d, <i>J</i> = 12.6 Hz, 1 H), 7.29–7.37 (m, 5 H)	5.5
11 <b>f</b>	MMC	CH <sub>2</sub> —CHCH <sub>2</sub> O	42 <sup>h</sup>	41	$C_{18}H_{22}N_4O_6$	3.97 (m, 2 H), 4.64 (d, $J = 12.2$ Hz, 1 H), 4.71 (d, $J = 12.2$ Hz, 1 H), 5.09 (ddd, J = 1.7, 3.5, 10.4 Hz, 1 H), 5.30 (ddd, J = 1.7, 3.5, 17.4 Hz, 1 H), 5.95 (m, 1 H)	5.5
11 <b>g</b> <sup>h</sup>	MMC	НО	j	j	$C_{18}H_{24}N_4O_7{}^i$	5.00  (d,  J = 12.3  Hz,  1  H),  5.05  (d,  J = 12.4  Hz,  1  H)	>29
	nycin C nycin A						0.59-1.1 0.0024

<sup>a</sup> Yield based on 7. <sup>b</sup> Prepared by method A. When 9, 16, and 20 were isolated, the yields were calculated based on them. In other cases, the yields were based on 8, 14, and 18, respectively. Determined by elemental analysis unless otherwise noted. Analytical results were within  $\pm 0.40\%$  of theoretical values for C, H, and N. d The solvent was CDCl<sub>3</sub> (13 and 11a,c-e) or pyridine- $d_5$  (17, 21, and 11b,f,g). In vitro anticellular activity against HeLa S3 cells. The cells were cultured in 96-well plates on day 0 and treated with drugs for 1 h on day 1. The anticellular activity was determined according to the method described previously (see ref 3c). Not isolated. Prepared by the sequential method. Prepared by the improved method. Determined by FAB-HRMS. Analytical results were within ±5 mmu of the theoretical value. See Scheme 2.

# Scheme 4

In addition, as previously reported,8 the Michael adduct of amines was not isolated since the retro-Michael or retro-Mannich reaction had occurred preferentially in the reaction or the purification steps to form 7 or 6-demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-(phenylseleno)mito-

The sulfide moiety in the C-6 substituent of 12e was cleanly oxidized with mCPBA to afford sulfoxides 27 and 28 and sulfone 29. Since sulfoxides 27 and 28 were obtained as a diastereomeric mixture bearing the asymmetric center at sulfur, each diastereomer was separated by preparative HPLC.

## **Biological Activity and Discussion**

All C-6-substituted methyl mitomycins were studied in vitro for their anticellular activity against HeLa S<sub>3</sub> cells (Tables 1 and 2). In several C-6-methoxymethyl compounds having various mitomycin skeletons, compounds of the MMC and MMA type (11a and 13 in Table 1) showed stronger activity than those of the PFM and MMD type (17 and 21 in Table 1). None of compounds 11 having various alkoxy groups at the C-6-methyl position showed outstanding anticellular activity. Substituents on the C-6methyl of 12 also affect largely the anticellular activity. Many compounds having polar functional groups (e.g., OH, NH) in the C-6-methyl substituent did not show sufficient activity (12c,f,n,x,ak in Table 2). However, protection of the hydroxy or amino groups in the C-6methyl substituent resulted in an increase of activity (12p vs 12n, 12y vs 12x, and 12aj vs 12ak in Table 2). Interestingly, compounds 120,u have relatively higher potency, which indicates the importance of the position of the phenyl substituent. A similar relationship was also observed for the activity of 12L vs 12k and 12q vs 12p. Sulfoxides 27 and 28 and sulfone 29 did not show antitumor activity. Other compounds not mentioned above showed generally significant activity, especially 12b,r,ab,ac,ae,ag and 26.

The in vivo activities against several murine tumors of selected compounds are given in Table 3. Most compounds listed in the table were effective in suppressing tumor volume (T/C) against sarcoma 180 murine solid tumor. Among those compounds, 120 and 26 were shown to have

Table 2. Preparations and Anticellular Activity of 12

		7 or 8 to 10	10 to 12				
compd	RS	10 yield (%) <sup>a</sup>	meth-	12 yield (%)	empirical formula <sup>c</sup>	¹Η NMR (δ, ppm) <sup>d</sup> C-6 subst	HeLa S <sub>3</sub> e IC <sub>50</sub> (µM)
12a	EtS	f, g	A	29	$C_{17}H_{22}N_4O_5S^h$	1.27 (t, J = 7.4 Hz, 3 H), 2.50 (q, J = 7.4 Hz, 1 H), 3.50 (d, J = 13.6 Hz, 1 H), 3.66 (d, J = 13.6 Hz, 1 H)	2.2
1 <b>2b</b>	$MeO_2C(CH_2)_2S\\$	58#	A	42	C <sub>19</sub> H <sub>24</sub> N <sub>4</sub> O <sub>7</sub> S∙ 0.2H <sub>2</sub> O	2.66-2.76 (m, 4 H), $3.54$ (d, $J = 13.3$ Hz, 1 H), $3.65$ (d, $J = 13.3$ Hz, 1 H), $3.70$ (s, 3 H)	0.97
12c	$HO(CH_2)_2S$	60s	A	55	$C_{17}H_{22}N_4O_6S^h$	2.87-2.92 (m, 2 H), $3.95$ (d, $J = 12.9$ Hz, 1 H), $4.08$ (d, $J = 12.9$ Hz, 1 H), $4.12$ (t, $J = 6.2$ Hz, 2 H), $6.67$ (s, 1 H)	>100
12 <b>d</b>	$\mathrm{CH_3}(\mathrm{CH_2})_{11}\mathrm{S}$	628	A	20	$C_{27}H_{42}N_4O_5S^h$	0.88 (t, J = 6.9 Hz, 3 H), 1.25-1.37 (m, 18 H), 1.59 (m, 2 H), 2.47 (m, 2 H), 3.48 (d, J = 13.4 Hz, 1 H), 3.63 (d, J = 13.4 Hz, 1 H)	5.5
12e	i-PrS	85 <sup>i</sup>	A	34	$C_{18}H_{24}N_4O_5S$	1.25 (d, $J = 6.8$ Hz, 3 H), 1.28 (d, $J = 6.8$ Hz, 3 H), 2.93-3.00 (m, 1 H), 3.82 (d, $J = 13.3$ Hz, 1 H), 3.98 (d, $J = 13.3$ Hz, 1 H)	5.4
12 <b>f</b>	HOCH <sub>2</sub> CH(OH)- CH <sub>2</sub> S	548	A	45	$C_{18}H_{24}N_4O_7S^h$	3.01-3.14 (m, 2 H), 4.04-4.16 (m, 5 H)	>100
12g	c-PrCH <sub>2</sub> S	56 <sup>i</sup>	С	44	C <sub>19</sub> H <sub>24</sub> N <sub>4</sub> O <sub>5</sub> S· 0.4H <sub>2</sub> O	0.18 (m, 2 H), 0.43 (m, 2 H), 1.10 (m, 1 H), 2.52 (dd, J = 7.1, 13.0 Hz, 1 H), 2.60 (dd, J = 7.1, 13.0 Hz, 1 H), 4.04 (d, J = 12.9 Hz, 1 H), 4.57 (d, J = 12.9 Hz, 1 H)	1.1
12h	CH₂ <del>=</del> CHCH₂S	61 <sup>i</sup>	C	42	C <sub>18</sub> H <sub>22</sub> N <sub>4</sub> O <sub>5</sub> S- 0.1CHCl <sub>3</sub>	3.22 (br d, J = 7.3 Hz, 2 H), 3.79 (d, J = 12.9 Hz, 1 H), 3.91 (d, J = 12.9 Hz, 1 H), 5.02 (dd, J = 1.7, 9.4 Hz, 1 H), 5.23 (dd, J = 1.7, 11.1 Hz, 1 H), 5.85-6.05 (m, 1 H)	2.3
12 <b>i</b>	PhS	53 <sup>g</sup>	Α	29	$C_{21}H_{22}N_4O_5S$ 0.1CHCl <sub>3</sub>	3.89 (d, J = 12.5 Hz, 1 H), 4.00 (d, J = 12.5 Hz, 1 H), 7.18-7.40 (m, 5 H)	1.0
		$71^i$	B C	36 63	0.1011013	112), 1120 1120 (22)	
12j	PhCH <sub>2</sub> S	488	Ā	44	C <sub>22</sub> H <sub>24</sub> N <sub>4</sub> O <sub>5</sub> S· 0.5H <sub>2</sub> O	3.47  (d,  J = 13.3  Hz,  1  H),  3.52  (d,  J = 13.3  Hz,  1  H),  3.68  (d,  J = 13.5  Hz,  1  H),  3.71  (d,  J = 13.5  Hz,  1  H),  7.21-7.33  (m,  5  H)	2.2
12 <b>k</b>	ci 🔾 s	628	D	35	$C_{21}H_{21}ClN_4O_5S^h$	4.22 (br s, 2 H), 7.0–7.6 (m, 4 H)	53
12 <b>L</b>	€ ci	43 <sup>i</sup>	С	35	$C_{21}H_{21}ClN_4O_5S^h$	4.26 (d, $J = 11.4$ Hz, 1 H), 4.34 (d, $J = 11.4$ Hz, 1 H), 6.97 (dt, $J = 1.6$ , 7.7 Hz, 1 H), 7.05 (dt, $J = 1.6$ , 7.7 Hz, 1 H), 7.27 (dt, $J = 1.6$ , 7.7 Hz, 1 H), 7.31 (dt, $J = 1.6$ , 7.7 Hz, 1 H)	7.3
12 <b>m</b>	S Br	57 <sup>i</sup>	С	40	C <sub>21</sub> H <sub>21</sub> BrN <sub>4</sub> O <sub>5</sub> S· 0.5CHCl <sub>3</sub>	4.25 (d, $J = 11.0$ Hz, 1 H), 4.33 (d, $J = 11.0$ Hz, 1 H), 6.88 (dt, $J = 1.3$ , 7.7 Hz, 1 H), 7.08 (dt, $J = 1.3$ , 7.7 Hz, 1 H), 7.1–7.3 (m, 1 H), 7.50 (dd, $J = 1.3$ , 7.7 Hz, 1 H)	48
12n	HO	68	В	29	$\mathrm{C}_{21}\mathrm{H}_{22}\mathrm{N}_4\mathrm{O}_6\mathrm{S}^h$	4.25 (d, J = 12.1 Hz, 1 H), 4.35 (d, J = 12.1 Hz, 1 H), 7.0-7.1 (m, 2 H), 7.5-7.6 (m, 2 H)	116
120	© s	49 <sup>i</sup>	С	32	$^{\mathrm{C_{21}H_{22}N_4O_6S}}_{0.2\mathrm{H_2O}}$	4.31 (d, J = 11.6 Hz, 1 H), 4.40 (d, J = 11.6 Hz, 1 H), 6.7-6.8 (m, 1 H), 7.1-7.3 (m, 2 H), 7.5-7.7 (m, 1 H), 12.1 (br s, 1 H)	2.2
12 <b>p</b>	CH <sub>3</sub> O	63 <sup>i</sup>	A	37	$C_{22}H_{24}N_4O_6S^h$	3.61 (s, 3 H), 4.26 (d, <i>J</i> = 12.0 Hz, 1 H), 4.32 (d, <i>J</i> = 12.0 Hz, 1 H), 6.8–6.9 (m, 2 H), 7.5–7.6 (m, 2 H)	14
12 <b>q</b>	©CH₃ s	44 <sup>i</sup>	С	15	$C_{22}H_{24}N_4O_6S^h$	3.73 (s, 3 H), 4.23 (d, <i>J</i> = 11.5 Hz, 1 H), 4.32 (d, <i>J</i> = 11.5 Hz, 1 H), 6.8-6.9 (m, 2 H), 7.1-7.2 (m, 1 H), 7.4-7.5 (m, 1 H)	2.3
12 <b>r</b>	O <sub>z</sub> N S	54 <sup>i</sup>	С	4.0	$C_{21}H_{21}N_5O_7S^h$	4.34  (d,  J = 11.5  Hz, 1  H), 4.43  (d,  J = 11.5  Hz, 1  H), 7.33  (dd,  J = 1.9, 7.1  Hz, 2  H), 8.03  (dd,  J = 1.9, 7.1  Hz, 2  H)	0.78
12s	S CH(CH <sub>3</sub> ) <sub>2</sub>	55 <sup>i</sup>	С	40	C <sub>24</sub> H <sub>28</sub> N <sub>4</sub> O <sub>5</sub> S- 0.2CHCl <sub>3</sub>	1.18 (d, $J = 6.4$ Hz, 6 H), 3.50–3.63 (m, 1 H), 4.23 (d, $J = 11.7$ Hz, 1 H), 4.29 (d, $J = 11.7$ Hz, 1 H), 7.0–7.4 (m, 4 H)	27
12 <b>t</b>	€ <sub>N</sub> L <sub>s</sub>	61 <sup>i</sup>	В	16	$C_{20}H_{21}N_5O_5S^h$	4.50 (s, 2 H), 6.91 (m, 1 H), 7.20–7.25 (m, 1 H), 7.37 (dt, J = 1.9, 7.8 Hz, 1 H), 8.41 (br d, J = 4.2 Hz, 1 H)	1.8
1 <b>2u</b>	N S S S S S S S S S S S S S S S S S S S	47 <sup>i</sup>	С	30	$C_{20}H_{21}N_5O_6S^h$	5.66 (d, $J = 14.5$ Hz, 1 H), 5.80 (d, $J = 14.5$ Hz, 1 H), 6.53 (dd, $J = 6.7$ , 7.7 Hz, 1 H), 7.00 (dd, $J = 1.3$ , 7.7 Hz, 1 H), 7.91 (dd, $J = 1.1$ , 6.7 Hz, 1 H), 9.33 (br s, 1 H)	0.19
12v	N	68 <sup>i</sup>	C	40	$C_{19}H_{20}N_6O_5S^h$	4.53 (s, 2 H), 6.85 (t, <i>J</i> = 4.9 Hz, 1 H), 8.47 (d, <i>J</i> = 4.9 Hz, 2 H)	0.95
12 <b>w</b>	CH <sub>3</sub>	69 <sup>i</sup>	C	62	C <sub>21</sub> H <sub>24</sub> N <sub>6</sub> O <sub>5</sub> S- 0.9H <sub>2</sub> O	2.26 (s, 6 H), 4.49 (s, 2 H), 6.52 (s, 1 H)	1.1

Table 2 (Continued)

		7 or 8 to 10	10 to 12				
compd	RS	10 yield (%) <sup>a</sup>	$_{\mathrm{od}^{b}}^{\mathrm{meth-}}$	12 yield (%)	empirical formula <sup>c</sup>	$^1\mathrm{H}\ \mathrm{NMR}\ (\delta,\mathrm{ppm})^d\ \mathrm{C-6}\ \mathrm{subst}$	HeLa S <sub>3</sub> ' IC <sub>50</sub> (μΜ
12 <b>x</b>	√ <sub>N</sub> × <sub>S</sub>	28 <sup>i</sup>	С	70	$C_{18}H_{20}N_6O_5S^h$	5.18 (d, $J = 14.4$ Hz, 1 H), 5.30 (d, $J = 14.4$ Hz, 1 H), 6.70 (d, $J = 2.0$ Hz, 1 H), 7.14 (d, $J = 2.0$ Hz, 1 H), 14.0 (br s, 1 H)	92
12y	N S	58 <sup>i</sup>	С	51	C <sub>19</sub> H <sub>22</sub> N <sub>6</sub> O <sub>5</sub> S- 0.2CHCl <sub>3</sub>	3.50 (s, 3 H), 5.10 (d, $J = 14.4$ Hz, 1 H), 5.21 (d, $J = 14.4$ Hz, 1 H), 6.81 (d, $J = 2.2$ Hz, 1 H), 7.04 (d, $J = 2.2$ Hz, 1 H)	0.76
12 <b>z</b>	N-N N S	$15^i$	С	28	$C_{17}H_{19}N_7O_5S^h$	5.09 (d, J = 14.5 Hz, 1 H), 5.17 (d, J = 14.5 Hz, 1 H), 8.51 (s, 1 H), 8.8 (br s, 1 H)	6.8
12aa	N-N CH <sub>3</sub>	37 <sup>i</sup>	С	70	$C_{18}H_{21}N_7O_5S^h$	3.48 (s, 3 H), 5.46 (s, 2 H), 8.35 (s, 1 H)	2.5
12ab	и-и и. <sub>N</sub> ≻s	55 <sup>i</sup>	c	25	$C_{17}H_{20}N_8O_5S^h$	3.73 (s, 3 H), 5.40 (d, J = 14.8 Hz, 1 H), 5.46 (d, J = 14.8 Hz, 1 H)	0.12
12ac	×× s	53 <sup>i</sup>	С	5.0	$C_{22}H_{22}N_8O_5S^h$	5.54 (s, 2 H), 7.3–7.5 (m, 5 H)	0.27
12ad	$\sqrt[n]{s}$	$34^i$	С	32	C <sub>19</sub> H <sub>20</sub> N <sub>4</sub> O <sub>5</sub> S <sub>2</sub> . 0.2CHCl <sub>3</sub>	4.25 (s, 2 H), 6.91 (dd, $J = 3.7, 5.4$ Hz, 1 H), 7.15–7.25 (m, 1 H), 7.38 (dd, $J = 1.2$ , 5.4 Hz, 1 H)	4.9
12ae	$\langle x \rangle_s$	$63^i$	С	67	$C_{18}H_{21}N_5O_5S_2^h$	3.04 (t, $J = 8.1$ Hz, 2 H), $3.96$ (t, $J = 8.1$ Hz, 2 H), $4.81$ (d, $J = 14.3$ Hz, 1 H), $5.13$ (d, $J = ca$ . $14$ Hz, 1 H)	0.82
12af	[ L	63 <sup>i</sup>	C	63	$C_{18}H_{19}N_5O_5S_2$ 0.1CHCl <sub>3</sub>	5.18 (d, $J = 14.4$ Hz, 1 H), 5.30 (d, $J = 14.4$ Hz, 1 H), 6.78 (d, $J = 4.6$ Hz, 1 H), 7.51 (d, $J = 4.6$ Hz, 1 H)	1.4
12ag		5 <b>4</b> <sup>i</sup>	C	36	$C_{22}H_{21}N_5O_5S_2^h$	5.59 (d, $J = 15.0 \text{ Hz}, 1 \text{ H})$ , 5.71 (d, $J = 15.0 \text{ Hz}, 1 \text{ H})$ , 7.15–7.25 (m, 1 H), 7.5–7.6 (m, 2 H), 8.02 (m, 1 H)	0.70
12ah	H₃C	79 <sup>į</sup> s	C	49	$C_{18}H_{20}N_6O_5S_2^h$	2.12 (s, 3 H), 5.53 (s, 2 H)	1.1
12 <b>a</b> i		68 <sup>i</sup>	С	53	$C_{22}H_{21}N_5O_6S^h$	5.28 (br s, 2 H), 7.06–7.17 (m, 2 H), 7.29–7.32 (m, 1 H), 7.81–7.85 (m, 1 H)	5.1
1 <b>2a</b> j	Aco OA	ac 45 <sup>i</sup> _s .c	С	60	C <sub>29</sub> H <sub>36</sub> N <sub>4</sub> O <sub>14</sub> S- 1.0H <sub>2</sub> O	1.99 (s, 3 H), 1.99 (s, 3 H), 2.01 (s, 3 H), 2.01 (s, 3 H), 4.02 (d, $J = 13.6$ Hz, 1 H), 4.13 (ddd, $J = 2.5$ , 4.9, 9.8 Hz, 1 H), 4.22 (d, $J = 13.6$ Hz, 1 H), 4.40 (dd, $J = 2.5$ , 12.5 Hz, 1 H), 4.58 (dd, $J = 4.9$ , 12.5 Hz, 1 H), 5.23 (d, $J = 10.1$ Hz, 1 H), 5.52 (t, $J = 9.5$ Hz, 1 H), 5.58 (t, $J = 9.5$ Hz, 1 H), 5.74 (t, $J = 9.2$ Hz, 1 H)	1.1
12ak	но он	s j	$\mathbf{A}^{j}$	79 <sup>j</sup>	C <sub>21</sub> H <sub>28</sub> N <sub>4</sub> O <sub>10</sub> S- 2.3H <sub>2</sub> O	3.9-4.2 (m, 6 H), $4.21$ (d, $J = 13.4$ Hz, 1 H), $4.29$ (dd, $J = 6.8$ , $11.5$ Hz, 1 H), $4.60$ (br d, $J = 10.1$ Hz, 1 H), $5.11$ (d, $J = 9.2$ Hz, 1 H), $6.60$ (br s, 1 H), $7.11$ (br s, 1 H)	82
24 <sup>k</sup>					$C_{17}H_{20}N_4O_5S$	3.08 (br t, $J = 5.4$ Hz, 2 H), 3.82 (d, $J = 16.2$ Hz, 1 H), 3.87 (d, $J = 16.2$ Hz, 1 H), 3.91 (br t, $J = 5.4$ Hz, 2 H)	1.3
26 <sup>k</sup>					$C_{21}H_{20}N_4O_5S^h$	3.84 (d, $J = 15.5$ Hz, 1 H), 4.19 (d; $J = 15.5$ Hz, 1 H), 7.03 (dt, $J = 1.3$ , 7.4 Hz, 1 H), 7.26 (dt, $J = 1.5$ , 7.2 Hz, 1 H), 7.36 (dd, $J = 1.3$ , 7.1 Hz, 1 H), 7.48 (dd, $J = 1.5$ , 7.7 Hz, 1 H)	0.31
<b>27</b> <sup>l</sup>	i-PrSO				C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O <sub>6</sub> S· 0.2CHCl <sub>3</sub>	1.14 (d, J = 6.8 Hz, 3 H), 1.28 (d, J = 6.8 Hz, 3 H),	>10
28 <sup>l</sup> (diaste	i-PrSO ereomer of 27)				0.2CHCl <sub>3</sub> C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O <sub>6</sub> S· 0.2CHCl <sub>3</sub>	1.14 (d, $J = 6.8$ Hz, 3 H), 1.28 (d, $J = 6.8$ Hz, 3 H), 2.83 (septet, $J = 6.8$ Hz, 1 H), 3.99 (d, $J = 13.9$ Hz, 1 H), 4.20 (d, $J = 13.9$ Hz, 1 H)	>10
<b>29</b> <sup>l</sup>	i-PrSO <sub>2</sub>				$C_{18}H_{24}N_4O_7S^h$	1.38 (d, $J = 6.8$ Hz, 3 H), 1.42 (d, $J = 6.8$ Hz, 3 H), 3.32 (septet, $J = 6.8$ Hz, 1 H), 4.43 (d, $J = 14.4$ Hz, 1 H), 4.58 (d, $J = 14.4$ Hz, 1 H)	>10
mi	tomycin C					,	0.86-2

<sup>&</sup>lt;sup>a</sup> Sequential method: yield based on 8. Improved method: yield based on 7. <sup>b</sup> Method A: NH<sub>3</sub>, MeOH, room temperature. Method B: NH<sub>4</sub>OAc, MeOH, room temperature. Method C: dry NH<sub>3</sub>, dry THF, room temperature. Method D: (1) TBSCl, imidazole, DMF, 0 °C, and then (2) NH<sub>3</sub>-MeOH, room temperature. <sup>c</sup> Determined by elemental analysis. Analytical results were within ±0.40% of theoretical values for C, H, and N. <sup>d</sup> The solvent was pyridine-d<sub>5</sub> except for 12a,b,d,i,j and 24 (CDCl<sub>3</sub>). <sup>e</sup> In vitro anticellular activity against HeLa S<sub>3</sub> cells. The cells were cultured in 96-well plates on day 0 and treated with drugs for 1 h on day 1. The anticellular activity was determined according to the method described previously (see ref 3c). <sup>f</sup> Not isolated. <sup>g</sup> Prepared by the sequential method. <sup>h</sup> Determined by FAB-HRMS. Analytical results were within ±5 mmu of the theoretical value. <sup>i</sup> Prepared by the improved method. <sup>j</sup> Prepared from 12aj. <sup>k</sup> See Scheme 4. <sup>l</sup> Prepared by an oxidation of 12e with mCPBA. See the Experimental Section.

Table 3. Antitumor and Anticellular Activities of Mitomycin Derivatives

		sarcoma 180 (sc-iv) <sup>b</sup>			P388 (ip-ip)c			
	HeLa Saa	$\frac{\mathrm{ED}_{50}{}^{d}}{(\mathrm{mg/kg})}$	ODe (mg/kg)	$T/C^f$ min.	ILS max %g (OD,e mg/kg)			
compd	IC <sub>50</sub> (μM)				sensitiv	ve to MMC	resistan	t to MMC
11c	2.5	15	50	0.38	55	(25)	>97	(25)
12b	0.97		3.1	0.63	69	(3.1)	$\mathrm{nt}^h$	, ,
12e	5.4	8.4	25	0.27	58	(50)	30	(25)
12 <b>g</b>	1.1	13	25	0.29	43	(13)	>115	(1.6)
12ĥ	2.3		14	0.77	nt	, ,	33	(6.0)
1 <b>2</b> i	1.0	6.1	22	0.27	45	(50)	nt	
12o	2.2	22	46	0.14	nt	, ,	>67	(46)
12u	0.19	3.7	6.3	0.39	49	(3.1)	26	(3.1)
12v	0.95	8.9	13	0.41	58	(0.78)	>92	(1.6)
12w	1.1		6.3	0.58	50	(3.1)	nt	()
12aa	2.5	39	46	0.44	nt	` ,	>71	(9.0)
12ad	4.9	29	46	0.35	nt		>56	(46)
12ae	0.82	5.9	6.3	0.45	62	(3.1)	64	(3.1)
12af	1.4	8.9	13	0.40	49	(3.1)	26	(3.1)
12ag	0.70	36	50	0.44	40	(13)	57	(50)
12aj	. 1.1	24	25	0.45	58	(13)	>143	(13)
26	0.31	35	100	0.16	36	(13)	48	(25)
nitomycin C	0.82 - 1.4	1.9-3.7	4.0-6.0	0.28-0.46	38-84	(1.0-6.0)	-19-23	(1.0-4.0

 $^a$  In vitro anticellular activity against HeLa S<sub>3</sub> cells. The cells were cultured in 96-well plates on day 0 and treated with compounds for 1 h on day 1. The anticellular activity was determined according to the method described previously (see ref 3c).  $^b$  In vivo antitumor activity against sarcoma 180. Sarcoma 180 cells were inoculated sc into the axillary region of ddY mice on day 0. Compounds were administered iv on day 1.  $^c$  In vivo antitumor activity against lymphocytic leukemia P388 in mice. P388 cells were inoculated ip into CDF<sub>1</sub> mice on day 0. Compounds were administered ip on day 1.  $^d$  Dose that gave 50% inhibition of tumor growth calculated from the dose–response curve.  $^e$  Optimal dose.  $^f$  Treated versus control value of tumor volume. Tumor volume was calculated according to the method described previously (see ref 3c).  $^e$  Maximal increase in life span, calculated (T/C-1) 100, where T and C are mean survival days of treated and control mice, respectively.  $^h$  Not tested.

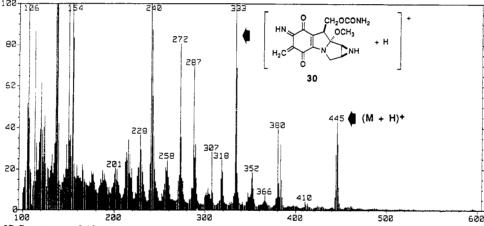


Figure 2. FAB-MS Spectrum of 12v.

superior activity to that of MMC. However, the optimal dose of those compounds is generally higher than that of MMC. Antitumor activity against P388 murine leukemia is also nearly equal to that of MMC in prolongation of life span (ILS), but the optimal dose is rather variable depending on the C-6-methyl substituent of the derivatives. In general, the compounds having high in vivo activity are potent against in vitro anticellular activity. On the other hand, those having high in vitro activity are not always effective in vivo. The most significant result of these assays is the excellent effectiveness of some derivatives against MMC-resistant P388 murine leukemia (P388/MMC) in ILS. And, surprisingly, even an evidently higher activity against P388/MMC than against P388 (sensitive to MMC) was observed in 11c and 12g,v,aj. It is concluded from these results that the C-6-substituted methyl mitomycin derivatives have some different character in antitumor activity from that of MMC. The difference may arise from the nature of the C-6-methyl substituent.

To conjecture the difference of character between such derivatives and MMC, we examined the derivatives from

a structural viewpoint. First, from the FAB-MS spectra of C-6-substituted methyl derivatives, the C-6-methyl substituents were found to be eliminated easily from the mitomycin skeleton to form quinonemethide<sup>12</sup> 30 in an ionization process. The fragment peak [m/z 333] (M - $RSH) + H]^+$  or 334  $[(M - RSH) + 2H]^+]$  was observed in all compounds except 24 and 26, and the intensity of most of those was greater than each parent peak. Figure 2 shows the FAB-MS spectrum of the representative derivative 12v. While the above ionization condition is drastic, the lability compared to the other functional groups, i.e., C-10-carbamoyloxy and C-9a-methoxy which are eliminated in the reductive activation process, is noteworthy. Secondly, we confirmed the ability of the alkylation by nucleophiles at the C-6-methyl position of derivatives using the representative compound 12v. Diethylamine was used as a nucleophile for its high nucleophilicity and easy determination. As a result, by contact of 12v with diethylamine at room temperature for 48 h, 6-demethyl-6-[(diethylamino)methyl]mitomycin C (31) was obtained in 77% yield, and this demonstrates the effective alkylation at the C-6-methyl position. It is worth

emphasizing that this nucleophilic alkylation occurred without the reductive activation. These findings suggest that the C-6-methyl position of derivatives acts as an alkylation site in addition to the C-1 and C-10 positions, the conventional alkylation sites to DNA. The role of "the third alkylation site" is unclear. However, there is a possibility that the derivatives bind covalently with both DNA and protein through the conventional DNA-alkylation sites and the C-6-methyl position which is positioned away sufficiently from DNA, and such complexes may contribute to characteristic antitumor activity.<sup>13</sup> In view of these results, although limiting, the biological activities characteristic of C-6-substituted methyl mitomycins may result from the newly created alkylation site at the C-6methyl position.

# Conclusions

A series of C-6-substituted methyl mitomycins was prepared, and some of them were effective against several tumors in mice including MMC-resistant P388 leukemia. It was shown that the substituents at the C-6-methyl position play a significant role. Further detailed studies on antitumor spectra and toxicity of some of these derivatives are in progress.

### **Experimental Section**

Unless otherwise noted, materials were obtained from commercial suppliers except for mitomycins and were used without purification. THF was distilled from sodium/benzophenone immediately prior to use. NH3 gas was also distilled from sodium immediately prior to use. Proton nuclear magnetic resonance (1H NMR) spectra were recorded on Bruker AM 400, JEOL JNM-GX270, and JEOL JNM-EX270 instruments. Mass spectral (MS) data were obtained from Hitachi M-80B and JEOL JMS-D300 mass spectrometers. Infrared spectra (IR) were recorded on a Nihon Bunko IR-810 instrument. Elemental analyses were performed by a Perkin-Elmer 2400 C, H, N analyzer. The purity of the samples was checked by chromatographic methods (HPLC and TLC) and careful analysis of NMR spectra. The representative analytical data are listed in Tables 1 and 2.

6-Demethyl-6-(methoxymethyl)mitomycin C (11a). Preparation via the Sequential Method. To a solution of 87 (297 mg, 0.516 mmol) in CHCl<sub>3</sub> (10 mL) and pyridine (1.0 mL) was added mCPBA (about 70% purity, 200 mg, 0.81 mmol) in one portion at 0 °C. After the solution was stirred for 10 min at room temperature, MeOH (5.0 mL) was added and the mixture was stirred for an additional 5 h. The resulting reaction mixture was poured into a saturated NaHCO<sub>3</sub> aqueous solution and extracted with CHCl3. The organic layer was washed with brine and dried over Na<sub>2</sub>SO<sub>4</sub>. The solvent was removed on a rotary evaporator, and the resulting residue was purified by chromatography (silica gel, 97:3 CHCl<sub>3</sub>/MeOH as an eluent) to afford the adduct 9a as a yellow paste. To a solution of 9a in MeOH (5 mL) was added NH<sub>3</sub> in MeOH (6.8 M, 0.5 mL), and the mixture was allowed to stand at room temperature. After 4 h, the resulting mixture was concentrated on a rotary evaporator. The residue was subjected to chromatography (silica gel, 95:5 CHCl<sub>3</sub>/MeOH as an eluent) followed by drying under vacuum to afford 11a (53 mg, 28% based on 8) as a purple paste.

6-Demethyl-6-(methoxymethyl)mitomycin A (13). Preparation via the Sequential Method. To a solution of 8 (312 mg, 0.542 mmol) in CHCl<sub>3</sub> (4 mL) and pyridine (0.5 mL) was added mCPBA (about 70% purity, 274 mg, 1.1 mmol) in one portion at 0 °C. After 1 h, MeOH (2 mL) was added and the reaction mixture was stirred for an additional 3 h. A saturated NaHCO<sub>3</sub> aqueous solution was added, and the reaction mixture was extracted with CHCl3. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated on a rotary evaporator. The resulting residue was subjected to chromatography (silica gel, 30:1 CHCl<sub>3</sub>/MeOH as an eluent) to afford a crude adduct, 9a. To a solution of 9a in MeOH (4 mL) was added KOH (1 mg), and the reaction mixture was stirred at room temperature. After 12 h. dry ice and brine were added and the mixture was extracted with CHCl<sub>3</sub>. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated on a rotary evaporator. The residue was purified by column chromatography (silica gel, 20:1 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum to afford 13 (21 mg, 10% based on 8) as a reddish purple powder.

6-Demethyl-6-(methoxymethyl)porfiromycin (17). Preparation via the Sequential Method. To a stirred solution of 14 (70 mg, 0.13 mmol) in CHCl<sub>3</sub> (2.5 mL) and pyridine (0.20 mL) was added mCPBA (about 70% purity, 40 mg, 0.16 mmol) in one portion at 0 °C. After 5 min at 0 °C, MeOH (0.50 mL) was added and stirring of the reaction mixture was continued at room temperature. After 8 h, NH<sub>3</sub> in MeOH (6.8 M, 0.50 mL) was added and the mixture was stirred for an additional 14 h. The volatiles were removed on a rotary evaporator, and the resulting residue was purified by column chromatography (silica gel, 96:4 CHCl<sub>3</sub>/MeOH as an eluent) to afford the crude product, which was further purified by preparative TLC (silica gel, 9:1 CHCl<sub>3</sub>/ MeOH as a developing solvent) followed by trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum to afford 17 (29 mg, 59%) as a gray powder.

6-Demethyl-6-(methoxymethyl)mitomycin D (21). Preparation via the Sequential Method. To a stirred solution of 18 (368 mg, 0.689 mmol) in CHCl<sub>3</sub> (20 mL) and pyridine (1.5 mL) was added mCPBA (about 70% purity, 220 mg, 0.89 mmol) in one portion at room temperature. After 30 min, MeOH (5.0 mL) was added to the reaction mixture. After an additional 15 h, NH<sub>3</sub> in MeOH (6.8 M, 0.50 mL) was added and stirring of the mixture was continued for an additional 3 h. The resulting reaction mixture was poured into a saturated NaHCO<sub>3</sub> aqueous solution. The layers were separated, and the aqueous layer was subjected to column chromatography (HP-20, 1:1 water/MeOH as an eluent) to afford a purple fraction. The solvent was removed on a rotary evaporator, and the residue was further purified by column chromatography (silica gel, 95:5 CHCl<sub>3</sub>/MeOH as an eluent). After drying under vacuum, the desired product 21 (50 mg, 20%) was obtained.

6-Demethyl-6-(isopropoxymethyl)mitomycin C (11c). Preparation via the Improved Method. To a solution of 7 (90 mg, 0.21 mmol) in CHCl<sub>3</sub> (3 mL) and i-PrOH (4 mL) was added Triton B (1.1% of water solution, 50  $\mu$ L), and the mixture was stirred at room temperature for 2 days. The reaction was quenched by adding phosphate buffer (pH 4), and the mixture was extracted with CHCl<sub>3</sub>. The obtained organic extracts were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated on a rotary evaporator. After purification by column chromatography (silica gel, 97:3 CHCl<sub>3</sub>/MeOH as an eluent), an adduct, 9c (53 mg, 52% based on 7), was obtained as a yellow paste. The obtained 9c (53 mg) was dissolved in MeOH (3 mL) and NH3 in MeOH (6.8 M, 0.30 mL), and the mixture was stirred at 0 °C. After 16 h, the volatiles were removed on a rotary evaporator and the residue was subjected to chromatography (silica gel, 95:5 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>n-hexane and drying under vacuum to afford 11c (26 mg, 60% based on 9c) which was obtained as a gray powder.

6-Demethyl-6-(hydroxymethyl)mitomycin C (11g). As described in the synthesis of 9c, compound 7 (207 mg, 0.494 mmol) was treated with 3,4-dimethoxybenzyl alcohol (2.0 mL) and Triton B (1.1% in water, 50  $\mu$ L) in CHCl<sub>3</sub> (15 mL) to afford 9g (110 mg, 38%). The adduct obtained (34 mg, 0.058 mmol) was dissolved in CHCl<sub>3</sub> (10 mL). To the solution were added water (0.3 mL) and DDQ (16 mg, 0.070 mmol), and the resulting mixture was stirred at 0 °C for 6 h. The insoluble materials were filtered off and washed with CHCl<sub>3</sub>, and the combined organic layer was concentrated on a rotary evaporator. The residue was purified by preparative TLC (silica gel, 95:5 CHCl<sub>3</sub>/MeOH as a developing solvent) to afford 22 (10 mg, 40%). A similar procedure to that described for the synthesis of 11c was employed to convert 22 (2.5 mg, 5.7  $\mu$ mol) into 11g (1.2 mg, 60%)

6-Demethyl-6-[(isopropylthio)methyl]mitomycin C (12e). Preparation via Method A. To a stirred solution of 7 (230 mg, 0.55 mmol) in CHCl<sub>3</sub> (10 mL) was added 2-propanethiol (50 mg, 0.66 mmol). After 1 hat room temperature, the reaction mixture was poured into a saturated NaHCO3 aqueous solution and extracted with CHCl3. The organic layer was washed with brine, dried over  $Na_2SO_4$ , and concentrated on a rotary evaporator. The resulting residue was purified by column chromatography (silica gel, 97:3 CHCl<sub>3</sub>/MeOH as an eluent) to afford an adduct, 10e (231 mg, 85%). A similar procedure to that described for the synthesis of 11c was employed to convert 10e (231 mg, 0.467 mmol) into 12e (65 mg, 34%).

1a-Acetyl-7-demethoxy-6-demethyl-7,7-(ethylenedioxy)-6,7-dihydro-6-[(phenylthio)methyl]mitomycin A (10i). (1) Sequential Method. To a suspension of 8 (292 mg, 0.507 mmol) and powdered K<sub>2</sub>CO<sub>3</sub> (149 mg, 1.08 mmol) in CHCl<sub>3</sub> (15 mL) was added mCPBA (about 70% purity, 1.49 g, 6.00 mmol) in one portion at 0 °C. The reaction mixture was stirred at room temperature. After 3.5 h, phosphate buffer (0.1 M, pH 7) and aqueous  $Na_2S_2O_3$  (0.1 M) were added. The layers were separated. and the aqueous layer was extracted with CHCl<sub>3</sub>. The combined organic layers were washed with brine and dried over Na2SO4 to afford crude 7 as a yellow solution. To this solution (100 mL) were added NEt<sub>3</sub> (100  $\mu$ L) and thiophenol (60  $\mu$ L, 0.54 mmol), and the mixture was stirred at room temperature for 3.5 h. The reaction mixture was washed successively with a saturated NaHCO<sub>3</sub> aqueous solution and brine, and the volatiles were removed on a rotary evaporator. The resulting residue was purified by column chromatography (silicagel, 97:3 CHCl3/MeOH as an eluent) to afford a brown paste. This paste was triturated with  $CHCl_3$ -n-hexane followed by drying under vacuum to afford 10i (142 mg, 53%) as a brown powder.

(2) Improved Method. To a stirred solution of 7 (1.00 g, 2.39 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) were added NEt<sub>3</sub> (0.50 mL) and thiophenol (0.30 mL, 2.7 mmol). After 1.5 h at room temperature, the reaction mixture was washed successively with phosphate buffer (pH 4) and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated on a rotary evaporator. The obtained residue was purified by column chromatography (silica gel, 20:1 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum to afford the adduct 10i (901 mg, 71%) as a brown powder. Compound 10i was obtained as a mixture of two diastereomers (approximately 1:1) at C-6 in CDCl<sub>3</sub>: <sup>1</sup>H NMR δ (270 MHz, CDCl<sub>3</sub>) 2.05 (s, 3/2 H, 1a-acetyl), 2.10 (s, 3/2 H, 1aacetyl), 3.20 (s, 3/2 H, 9a-OCH<sub>3</sub>), 3.22 (s, 3/2 H, 9a-OCH<sub>3</sub>), 3.15-3.55 (m, 3 H, 6-CH<sub>2</sub> + 6-H), 3.20-3.25 (m, 1 H, 2-H), 3.41 (dd,  $J = 1.7, 13.1 \text{ Hz}, 1/2 \text{ H}, 3\alpha\text{-H}), 3.48 \text{ (dd}, J = 2.0, 13 \text{ Hz}, 1/2 \text{ H},$  $3\alpha$ -H), 3.50 (d, J = 4.5 Hz, 1/2 H, 1-H), 3.51 (d, J = 4.5 Hz, 1/2H, 1-H), 3.71 (dd, J = 5.0, 8.9 Hz, 1/2 H, 9-H), 3.75 (dd, J = 5.0,  $9.4 \, \text{Hz}, \frac{1}{2} \, \text{H}, 9 - \text{H}, 4.0 - 4.5 \, (\text{m}, 5 \, \text{H}, \text{ethylenedioxy} + 10 - \text{H}_a), 4.01$  $(d, J = 13.4 \text{ Hz}, 1/2 \text{ H}, 3\beta\text{-H}), 4.28 (d, J = 12.9 \text{ Hz}, 1/2 \text{ H}, 3\beta\text{-H}),$ 4.88 (br s, 2 H, 10-OCONH<sub>2</sub>), 4.93 (dd, J = 5.0, 11.1 Hz, 1/2 H,  $10-H_b$ ), 4.95 (dd, J = 5.0, 10.9 Hz, 1/2 H,  $10-H_b$ ), 7.11-7.47 (m, 5 H, phenyl). Anal.  $(C_{25}H_{27}N_3O_8S\cdot 1.4H_2O)$  C, H, N.

6-Demethyl-6-[(phenylthio)methyl]mitomycin C (12i). (1) Method A: Conversion with NH<sub>3</sub> in MeOH. To a solution of 10i (110 mg, 0.208 mmol) in MeOH (25 mL) was added NH<sub>3</sub> in MeOH (6.8 M, 0.3 mL), and the mixture was allowed to stand at room temperature. After 11 h, the resulting mixture was concentrated on a rotary evaporator and the residue was subjected to chromatography (silica gel, 9:1 CHCl<sub>3</sub>/MeOH as an eluent) to afford a purple paste. This paste was triturated with CHCl<sub>3</sub>-n-hexane followed by drying under vacuum to afford 12i (27 mg, 29%) as a purple powder.

(2) Method B: Conversion with NH<sub>4</sub>OAc in MeOH. To a stirred solution of 10i (49 mg, 0.093 mmol) in MeOH (10 mL) was added NH<sub>4</sub>OAc (150 mg). After 14 h at room temperature, brine was added to the reaction mixture. The mixture was extracted with CHCl<sub>3</sub>, and the organic layer was dried over Na<sub>2</sub>-SO<sub>4</sub>. The volatiles were removed, and the resulting residue was treated using the foregoing procedure to afford 12i (15 mg, 36%) as a purple powder.

(3) Method C: Conversion with NH<sub>3</sub> in THF. A solution of 10i (256 mg, 0.484 mmol) in THF (30 mL) was allowed to stand at room temperature in an atmosphere of dry NH<sub>3</sub> for 1 week. The resulting mixture was concentrated on a rotary evaporator and treated using the foregoing procedure to afford 12i (136 mg, 63%) as a purple powder: <sup>1</sup>H NMR  $\delta$  (400 MHz, CDCl<sub>3</sub>) 0.64 (br s, 1 H, 1a-H), 2.81 (br s, 1 H, 2-H), 2.89 (br s, 1 H, 1-H), 3.19 (s, 3 H, 9a-OCH<sub>3</sub>), 3.50 (br d, J = 12.9 Hz, 1 H, 3 $\alpha$ -H), 3.60 (dd, J = 4.4, 10.6 Hz, 1 H, 9-H), 3.89 (d, J = 12.5 Hz, 1 H, 6-CH<sub>2</sub>), 4.00 (d, J = 12.5 Hz, 1 H, 6-CH<sub>2</sub>), 4.21 (d, J = 12.9 Hz, 1 H, 3 $\beta$ -H),

4.51 (br t, 1 H, 10-H<sub>a</sub>), 4.68 (dd, J=4.4, 10.6 Hz, 1 H, 10-H<sub>b</sub>), 4.79 (br s, 2 H, 10-OCONH<sub>2</sub>), 5.85 (br s, 2 H, 7-NH<sub>2</sub>), 7.18-7.40 (m, 5 H, phenyl); secondary ion-MS m/z 443 (M<sup>+</sup> + 1); IR (KBr) 3425, 3320, 2886, 1708, 1650, 1600, 1553, 1480, 1439, 1334, 1221, 1165 cm<sup>-1</sup>. Anal. (C<sub>21</sub>H<sub>22</sub>N<sub>4</sub>O<sub>5</sub>S-0.1CHCl<sub>3</sub>) C, H, N.

6-[[(4-Chlorophenyl)thio]methyl]-6-demethylmitomycin C (12k). Preparation via Method D. A crude solution of 7 in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) prepared from 8 (300 mg, 0.523 mmol) was treated in a similar procedure to that described for the synthesis of 10i (sequential method) with 4-chlorothiophenol (153 mg, 1.06 mmol) to afford 10k (184 mg, 62%). To the solution of 10k (114 mg, 0.201 mmol) in DMF (2.5 mL) were added imidazole (50 mg, 0.73 mmol) and t-BuMe<sub>2</sub>SiCl (82 mg, 0.54 mmol). After being stirred for 105 min at 0 °C, the reaction mixture was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated on a rotary evaporator. The residue was purified by column chromatography (silica gel, from 3:1 to 2:1 CHCl<sub>3</sub>/MeCN as eluents) followed by trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum to afford 32 (98 mg, 72%) as a reddish purple powder. The desired product 12k (34 mg, 49%) was prepared by a similar procedure to that described in the synthesis of 12i (method A) from 32 on a 0.145-mmol scale.

#### Scheme 5

6-[[(2-Chlorophenyl)thio]methyl]-6-demethylmitomycin C (12L). Preparation via Method C. To a stirred solution of 7 (629 mg, 1.50 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) were added 2-chlorothiophenol (170  $\mu$ L, 1.50 mmol) and NEt<sub>3</sub> (150  $\mu$ L). After 1 h at room temperature, the resulting reaction mixture was washed successively with phosphate buffer (pH 4) and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated on a rotary evaporator. The residue was purified by column chromatography (silica gel, 30:1 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>-n-hexane to afford an adduct, 10L (364 mg, 43%), as a red powder. The product 10L (349 mg, 0.618 mmol) was dissolved in THF (30 mL), and the mixture was allowed to stand under an atmosphere of dry NH<sub>3</sub> at room temperature. After 136 h, the volatiles were removed on a rotary evaporator and the residue was purified by column chromatography (silica gel, 30:1 CHCl<sub>3</sub>/ MeOH as an eluent). After trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum, the desired 12L (103 mg, 35%) was obtained as a purple powder.

6-Demethyl-6-[[(3-hydroxypyridin-2-yl)thio]methyl]mitomycin C (12u). Preparation via Method C. To a stirred solution of 7 (313 mg, 0.747 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 mL) were added 3-hydroxy-2-mercaptopyridine (95 mg, 0.747 mmol) and NEt<sub>3</sub>  $(100 \,\mu\text{L})$ . After 30 min at room temperature, the resulting reaction mixture was washed successively with phosphate buffer (pH 4) and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The residue was purified by column chromatography (silica gel, 20:1 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum to afford 10u (190 mg, 47%) as a yellow powder. The obtained 10u (190 mg, 0.348 mmol) was dissolved in THF (20 mL), and the mixture was allowed to stand under an atmosphere of dry NH<sub>3</sub> at room temperature. After 49 h, the volatiles were removed on a rotary evaporator and the residue was purified by column chromatography (silica gel, from 20:1 to 10:1 CHCl<sub>3</sub>/MeOH as eluents) followed by trituration with CHCl3-n-hexane and drying under vacuum to afford 12u (48 mg, 30%) as a purple powder.

6-Demethyl-6-[[(2-pyrimidinyl)thio]methyl]mitomycin C (12v). Preparation via Method C. As described in the synthesis

of 12u, compound 7 (420 mg, 1.00 mmol) was treated with 2-mercaptopyrimidine (111 mg, 0.991 mmol) and NEt<sub>3</sub> (50  $\mu$ L) in CH<sub>2</sub>Cl<sub>2</sub> (40 mL) to afford an adduct, 10v (362 mg, 68%). A similar procedure to that described for the synthesis of 12u was employed to convert 10v (362 mg, 0.681 mmol) into 12v (121 mg, 40%).

6-Demethyl-6-[[(2,3,4,6-tetra-O-acetyl- $\beta$ -D-glucopyranosyl)thio]methyl]mitomycin C (12aj). Preparation via Method C. As described in the synthesis of 12u, compound 7 (215 mg, 0.513 mmol) was treated with 1-thio-β-D-glucose tetraacetate (195 mg, 0.536 mmol) and NEt<sub>3</sub> (50 µL) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) to afford an adduct, 10aj (182 mg, 45%). A similar procedure to that described for the synthesis of 12u was employed to convert 10aj (182 mg, 0.233 mmol) into 12aj (98 mg, 60%).

6-Demethyl-6-[(β-D-glucopyranosylthio)methyl]mitomycin C (12ak). Compound 12aj (25 mg, 0.036 mmol) was dissolved in NH<sub>3</sub> in MeOH (6.1 M,  $3.0 \,\mathrm{mL}$ ), and the mixture was allowed to stand at room temperature for 10.5 h. After the volatiles were removed on a rotary evaporator, the obtained residue was purified by a short column chromatography (Bond Elute C-18, from 100:0 to 60:40 water/MeCN as eluents) to obtain a purple aqueous solution. This solution was freeze-dried to afford 12ak (15 mg, 79%) as a purple amorphous solid.

2-Aminoethanethiol Adduct 24. As described in the synthesis of 10i, a CHCl<sub>3</sub> solution of crude 7 prepared from 8 (320 mg, 0.556 mmol) was treated with 2-aminoethanethiol hydrochloride (127 mg, 1.12 mmol) and pyridine (excess) in THF (10 mL) to afford an adduct, 23 (55 mg, 23%). The adduct 23 (55 mg, 0.13 mmol) was dissolved in MeOH (10 mL) and NH<sub>3</sub> in MeOH (6.8 M, 0.50 mL), and the solution was stirred at 0 °C for 4 h. The volatiles were removed on a rotary evaporator, and the residue was purified by column chromatography (silica gel, 9:1 CHCl<sub>3</sub>/MeOH as an eluent). After drying under vacuum, the desired product 24 (33 mg, 66%) was obtained.

2-Aminothiophenol Adduct 26. As described in the synthesis of 12L, compound 7 (299 mg, 0.714 mmol) was treated with 2-aminothiophenol (90 mg, 0.71 mmol) and NEt<sub>3</sub> (100 µL) in CH<sub>2</sub>Cl<sub>2</sub> (30 mL) to afford the adduct 25 (146 mg, 42%). The adduct 25 (123 mg, 0.255 mmol) was treated with NH<sub>3</sub> in MeOH as described above and afforded a green paste. This paste was purified by column chromatography (silicagel, 30:1 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum to afford 26 (76 mg, 57%) as a green powder.

6-Demethyl-6-[(isopropylsulfinyl)methyl]mitomycin C (27 and 28). To a slurry of 12e (110 mg, 0.270 mmol) and  $K_2CO_3$ (128 mg, 0.926 mmol) in CHCl<sub>3</sub> (50 mL) was added a solution of mCPBA (about 80% purity, 130 mg, 0.60 mmol) in CHCl<sub>3</sub> (10 mL) over a period of 13 min at 0 °C, and the mixture was stirred for 2 min. The reaction was quenched by adding an aqueous solution of Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (0.4 M), and the mixture was extracted with CHCl<sub>3</sub>. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated on a rotary evaporator to afford a purple paste. This paste was purified by column chromatography (silica gel, 20:1 CHCl<sub>3</sub>/MeOH as an eluent) followed by trituration with CHCl<sub>3</sub>*n*-hexane to afford the product (73 mg, 64%) as a purple powder. Since the material was found to be a mixture of diastereomers arising from the sulfinyl group at the C-6-methyl position, each diastereomer was isolated by preparative HPLC (ODS, 60:40 water/MeCN as an eluent), affording 27 (22 mg) and 28 (22 mg).

6-Demethyl-6-[(isopropylsulfonyl)methyl]mitomycin C (29). To a slurry of 12e (75 mg, 0.19 mmol) and  $K_2CO_3$  (102 mg, 0.738 mmol) in CHCl<sub>3</sub> (30 mL) was added a solution of mCPBA (about 80% purity, 128 mg, 0.59 mmol) in CHCl<sub>3</sub> (10 mL) over a period of 23 min at 0 °C, and the mixture was stirred for 3 min. The reaction was quenched by adding a Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> aqueous solution (0.4 M), and the mixture was extracted with CHCl<sub>3</sub>. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated on a rotary evaporator. The paste obtained was purified by column chromatography (silica gel, 20:1 CHCl<sub>3</sub>/MeOH as an eluent) and further purified by preparative HPLC (ODS, 60:40 water/MeOH as an eluent). MeOH was removed on a rotary evaporator, and

the aqueous solution was freeze-dried to afford 29 (16 mg, 20%) as a purple amorphous solid.

6-Demethyl-6-[(diethylamino)methyl]mitomycin C (31). To a solution of 12v (25 mg, 0.056 mmol) in  $CH_2Cl_2$  (10 mL) was added diethylamine (0.10 mL), and the reaction mixture was allowed to stand at room temperature for 48 h. The volatiles were removed on a rotary evaporator, and the obtained residue was purified by preparative TLC (silica gel, 4:1 CHCl<sub>3</sub>/MeOH as a developing solvent). After trituration with CHCl<sub>3</sub>-n-hexane and drying under vacuum, the product (17 mg, 77%) was obtained as a reddish purple powder: <sup>1</sup>H NMR (270 MHz, pyridine-d<sub>5</sub>)  $\delta$  1.25 (t, J = 7.2 Hz, 6 H, ethyl), 2.23 (s, 1 H, 1a-H), 2.80 (dd, J = 1.8, 4.4 Hz, 1 H, 2 -H, 3.0 - 3.2 (m, 5 H, ethyl + 1-H), 3.23 (s,3 H, 9a-OCH<sub>3</sub>), 3.63 (dd, J = 1.8, 12.8 Hz, 1 H,  $3\alpha$ -H), 4.01 (dd,  $J = 4.3, 11.1 \text{ Hz}, 1 \text{ H}, 9 \text{-H}, 4.17 \text{ (s, } 2 \text{ H, } 6 \text{-CH}_2), 4.52 \text{ (d, } J = 12.8$ Hz, 1 H,  $3\beta$ -H), 5.01 (br t, J = 11 Hz, 1 H, 10-H<sub>a</sub>), 5.34 (dd, J =4.3, 10.4 Hz, 1 H, 10-H<sub>b</sub>), 7.4-7.7 (br, 2 H, 10-OCONH<sub>2</sub>), 8.8-9.1 (br, 1 H, 7-NH<sub>2</sub>), 9.6-10.4 (br, 1 H, 7-NH<sub>2</sub>); FAB-MS m/z 406  $(M^+ + 1)$ ; FAB-HRMS calcd for  $C_{19}H_{28}N_5O_5$   $(M^+ + H)$  m/z406.2091, found 406.2093; IR (KBr) 3430, 3350, 3280, 2970, 2930, 2850, 1720, 1710, 1600, 1570, 1560, 1450, 1340, 1070 cm<sup>-1</sup>.

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Supplementary Material Available: Complete analytical data (<sup>1</sup>H NMR, IR, mass spectrum, elemental analysis) for new compounds except 10i, 12i, and 31 and <sup>1</sup>H NMR spectra (65 pages). Ordering information is given on any current masthead page.

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- (11) In the case of the conversion of 10r,ac to 12r,ac, respectively, it is difficult to suppress the decomposition of the substrate even by use of method C. It seems to be due to the increase of the elimination efficiency by the electron-withdrawing effect (for 10r) or steric hindrance (for 12ac).
- (12) Molecular formulas of the representative fragment peaks of 12v were confirmed by FAB-HRMS. m/z 445 [(M + H)+]: calculated for  $C_{19}H_{21}N_6O_6S$  m/z 445.1270, found 445.1281. m/z 383 [[(M HOCONH<sub>2</sub>-H)+H]+]: calculated for  $C_{19}H_{17}N_6O_3S$  m/z 383.1087, found 383.1069. m/z 352 [[(M HOCONH<sub>2</sub> CH<sub>3</sub>OH) + H]+]: calculated for  $C_{17}H_{14}N_8O_2S$  m/z 352.0897, found 352.0882. m/z 333 [[(M RSH) + H]+]: calculated for  $C_{15}H_{17}N_4O_5$  m/z 333.1243, found 333.1221. m/z 272 [[(M RSH HOCONH<sub>2</sub>) + H]+]: calculated for  $C_{14}H_{14}N_3O_3$  m/z 272.0973, found 272.1004. m/z 240 [[(M RSH HOCONH<sub>2</sub> CH<sub>3</sub>OH) + H]+]: calculated for  $C_{13}H_{10}N_3O_2$  m/z 240.0757, found 240.0765.
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