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Synthesis, X-ray molecular structure, biological evaluation and molecular docking studies of some N⁴-benzyl substituted 5-nitroisatin-3-thiosemicarbazones

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Abstract

A series of fifteen N⁴-benzyl substituted 5-nitroisatin-3-thiosemicarbazones **5a-o** was synthesized and evaluated for urease inhibitory, phytotoxic and cytotoxic influences. All the compounds proved to be highly potent inhibitors of the enzyme, showing inhibitory activity ($IC_{50} = 0.87 \pm 0.25 - 8.09 \pm$ 0.23 µM) much better than the reference inhibitor, thiourea ($IC_{50} = 22.3 \pm 1.12$ µM) and may thus act as persuasive leads for further studies. In phytotoxicity assay, twelve out of fifteen thiosemicarbazones tested i.e. **5a-e**, **5g**, **5i** and **5k-o** appeared to be active, exhibiting weak or nonsignificant (5-35%) growth inhibition at the highest tested concentrations (1000 or 500 µg/mL). In contrast, only one compound i.e. **5i** was active in the brine shrimp (*Artemia salina*) lethality bioassay, demonstrating cytotoxic activity with LD_{50} value 2.55×10^{-5} M. Molecular docking studies of compounds **5a-o** were also performed to identify their probable binding modes in the active site of the enzyme.

Keywords: Cytotoxicity, 5-Nitroisatin, Phytotoxicity, Thiosemicarbazones, Urease inhibition, Urease inhibitors

1. Introduction

Isatin and its derivatives are reported to exhibit a broad spectrum of chemotherapeutic properties.¹⁻²⁵ Among isatin derivatives, isatin-thiosemicarbazones have been found to demonstrate diverse biological activities such as analgesic and antiinflammatory, anticancer, antimicrobial, antitubercular, antiulcer, antiviral and enzymatic inhibition.^{1-5,7,11,12,17,21-25} In view of these facts and as a part of our synthetic work on potential biologically active isatin derivatives, we have recently synthesized a number of N⁴-aryl substituted isatin-3-thiosemicabazones as antibacterial, antifungal, antileishmanial, cytotoxic, phytotoxic and more importantly antiurease compounds.²⁶⁻³⁵ Investigation of the structure-activity relationships (SAR) revealed that in some cases, the nature and position of the substituents about the phenyl ring attached to N^4 of the thiosemicarbazone moiety and/or the presence of inductively electron-withdrawing/lipophilic groups (nitro, trifluoromethoxy, fluoro, chloro) at position-5 of the isatin scaffold played an important role in the inducement and/or increment of certain activities. Furthermore, during a preliminary screening of some isatinthiosemicarbazones prepared in our laboratory, oneviz.N-benzyl-2-(5-nitro-2-oxo-2,3-dihydro-1Hindol-3-ylidene)-1-hydrazinecarbothioamidedrew our attention, as it showed about four fold more urease inhibitory activity than the reference inhibitor, thiourea, used in the assay. In view of this, it was envisaged that the attachment of variously substituted benzyl groups (having one or two inductively electron-attracting /-donating substituents about the phenyl ring) to N⁴ of the thiosemicarbazone moiety would cause an increase in the urease inhibitory potential of the molecules. Thus, the present work to synthesize a series of fifteen title thiosemicarbazones and screen them mainly for their urease inhibitory as well as phytotoxicity and cytotoxicity potential was carried out.

This paper is one of the first to report on N^4 -benzyl substituted isatin-thiosemicarbazones with urease inhibitory potential. In it, we confine our discussion to those compounds, which are having attached concurrently strong inductively electron-attracting/lipophilic nitro group at position-5 of the isatin scaffold and the benzyl substituents (modified by substituting one inductively electron-donating and

one or two electron-attracting substituents about the phenyl ring) to N^4 of the thiosemicarbazone moiety.

2. Experimental

2.1. General

Melting points (uncorrected) were determined on a Fisher-Johns melting point apparatus. Elemental analyses were performed by using a Leco CHNS-9320 (USA) elemental analyzer. FT-IR spectra (KBr disks) were run on Shimadzu Prestige-21 FT-IR spectrophotometer. ¹H-NMR spectra were recorded in DMSO-_{d6} on Bruker Avance 300 and Bruker Spectrospin 400 spectrometers, operating at 300 and 400 MHz, respectively, and using TMS as an internal standard. The chemical shifts are reported in δ / ppm and coupling constants in Hz.¹³C-NMR spectra were recorded in DMSO-_{d6} on Bruker Avance 300, operating at 75 MHz with the same internal standard. Electron impact (EI) mass spectra were obtained on MAT-312 and JEOL MSRoute mass spectrometers. X-ray crystallographic data were taken on Bruker Kappa APEXII CCD. All the chemicals and solvents used in the present study were procured from Merck-Schuchatdt, Fluka and Sigma-Aldrich.

2.2. Synthesis of N-substituted thiosemicarbazides 3

The *N*-substituted thiosemicarbazides used in this study were synthesized according to the reported synthetic route.³⁶

2.3. Synthesis of isatin-thiosemicarbazones 5a-o

An appropriate thiosemicarbazide 3 (5 mmol) dissolved in ethanol (10 mL) was added to a hot solution of 5-nitroisatin 4 (5 mmol) in 50% aqueous ethanol (15 mL) under stirring and the reaction mixture was then heated to reflux for 2 h. The crystalline solid formed during heating in each case was collected by suction filtration and washed with hot aqueous ethanol to afford the desired compounds **5a-o**.

2.3.1. N-Benzyl-2-(5-nitro-2-oxo-2,3-dihydro-1*H*-indol-3-ylidene)-1-hydrazinecarbothioamide (5a)

Yellow crystals (82%): mp 262-264 °C; IR (KBr, cm⁻¹): 3344, 3118 (NH), 1693 (C=O), 1604 (C=N), 1157 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 4.91 (d, J = 6.0 Hz, 2H, benzyl CH₂), 7.12 (d, J = 8.8 Hz, 1H, indole C₇-H), 7.25-7.39 (m, 5H, benzyl C₂-H, C₃-H, C₄-H, C₅-H, C₆-H), 8.26 (dd, J = 8.8, 2.4 Hz, 1H, indole C₆-H), 8.55 (d, J = 2.0 Hz, 1H, indole C₄-H), 10.13 (t, J = 7.5 Hz, 1H, CSNH), 11.83 (s, 1H, indole NH), 12.44 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 47.24 (CH₂), 111.19 (CH), 116.20 (CH), 120.94, 126.89 (CH), 127.03 (CH), 127.19 (CH), 128.29 (CH), 130.10, 138.14, 142.67, 147.39, 162.95, 177.70; EIMS (70eV) m/z (%): 355(M⁺, 2), 327 (2), 296 (1), 279 (3), 220 (4), 192 (20), 164 (9), 131 (20), 106 (44), 91 (100), 77 (17), 65 (27), 51 (15). Anal. Calcd for C₁₆H₁₃N₅O₃S (355): C, 54.08; H, 3.66; N, 19.72. Found: C, 53.89; H, 3.62; N, 19.66.

2.3.2. *N*-(2-Methylbenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1hydrazinecarbothioamide (5b)

Yellow fluffy crystals (82%): mp 240-242 °C; IR (KBr, cm⁻¹): 3319, 3246 (NH), 1701 (C=O), 1601 (C=N), 1163 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 2.35 (s, 3H, CH₃), 4.87 (d, J = 6.0 Hz, 2H, benzyl CH₂), 7.12 (d, J = 9.0 Hz, 1H, indole C₇-H), 7.19 (s, 4H, benzyl C₃-H, C₄-H, C₅-H, C₆-H), 8.26 (dd, J = 9.0, 3.0 Hz, 1H, indole C₆-H), 8.56 (d, J = 3.0 Hz, 1H, indole C₄-H), 10.05 (t, J = 7.5 Hz, 1H, CSNH), 11.83 (s, 1H, indole NH), 12.47 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 55.27 (CH₂), 112.95 (CH), 113.59 (CH), 114.13 (CH), 122.24, 123.65 (CH), 127.20 (CH), 130.62, 131.06, 133.23 (CH), 141.38, 144.07 (CH), 157.43, 162.26, 176.45; EIMS (70eV) m/z (%): 369 (M⁺, 2), 341(2), 293(1), 220 (3), 192 (23), 178 (17), 163 (12), 145 (25), 120 (38), 105 (100), 91 (38), 79 (52), 77 (65), 65 (20). Anal.Calcd for C₁₇H₁₅N₅O₃S (369): C, 55.28; H, 4.07; N, 18.97. Found: C, 55.12; H, 4.04; N, 18.93.

2.3.3. *N*-(3-Methylbenzyl))-2-[5-nitro-2-oxo-2,3-dihydro-1*H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5c)

Yellow crystals (72%): mp 210-212 °C; IR (KBr, cm⁻¹): 3323, 3069 (NH), 1699 (C=O), 1610 (C=N), 1159 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 2.56 (DMSO, CH₃), 3.35 (DMSO, benzyl CH₂), 6.96 (d, J = 7.1 Hz, 1H, indole C₇-H), 7.35-7.33 (m, 1H, benzyl C₄-H), 7.39-7.50 (m, 4H,

benzyl C₂-H, C₅-H, C₆-H, indole C₆-H), 7.80 (d, J = 2.0 Hz, 1H, indole C₄-H), 10.82 (s, 1H, CSNH), 11.36 (s, 1H, indole NH), 12.71 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 112.64 (CH), 116.96 (CH), 120.83 (CH), 121.67, 126.55, 130.80 (CH), 131.75, 141.33, 162.36, 177.89; EIMS (70eV) m/z(%): 369 (M⁺, 3), 341 (2), 293 (2), 248 (2), 220 (8), 206 (6), 192 (46), 178 (20), 163 (17), 145 (46), 144 (11), 121 (65), 105 (100), 90 (18), 91 (55), 79 (19), 77 (73), 65 (34). Anal.Calcd for C₁₇H₁₅N₅O₃S (369): C, 55.28; H, 4.07; N, 18.97. Found: C, 54.99; H, 4.02; N, 18.94.

2.3.4. *N*-(4-Methylbenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5d)

Yellow crystals (89%): mp 208-210 °C; IR (KBr, cm⁻¹): 3316, 3163 (NH), 1678 (C=O), 1600 (C=N), 1159 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 2.29 (s, 3H, CH₃), 4.86 (d, J = 5.6 Hz, 2H, benzyl CH₂), 6.84 (d, J = 9.0 Hz, 1H, indole C₇-H), 7.07-7.34 (m, 4H, benzyl C₂-H, C₃-H, C₅-H, C₆-H), 8.27 (d, J = 9.0 Hz, 1H, indole C₆-H), 8.55 (s, 1H, indole C₄-H), 10.08 (t, J = 7.5 Hz, 1H, CSNH), 11.83 (s, 1H, indole NH), 12.43 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 20.67 (CH₃), 47.02 (CH₂), 111.19 (CH), 113.44 (CH), 116.20 (CH), 120.95, 126.88 (CH), 127.21 (CH), 128.51 (CH), 128.83 (CH), 130.11, 135.08, 142.66, 147.37, 162.96, 177.54; EIMS (70eV) m/z (%): 369 (M⁺, 19), 341 (3), 293 (7), 220 (13), 206 (76), 192 (44), 178 (47), 163 (34), 145 (44), 136 (27), 121 (96), 105 (100), 91 (22), 79 (17), 77 (28), 65 (9). Anal.Calcd for C₁₇H₁₅N₅O₃S (369): C, 55.28; H, 4.07; N, 18.97. Found: C, 54.91; H, 4.05; N, 18.91.

2.3.5. *N*-(2-Methoxybenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5e)

Yellow crystals (89%): mp 290-292 °C; IR (KBr, cm⁻¹): 3356, 3093 (NH), 1737 (C=O), 1606 (C=N), 1137 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 3.85 (s, 3H, CH₃), 4.41 (d, J = 5.0 Hz, 2H, benzyl CH₂), 6.91-6.99 (m, 2H, benzyl C₃-H, C₅-H), 7.11 (d, J = 9.0 Hz, 1H, indole C₇-H), 7.21-7.29 (m, 2H, benzyl C₄-H, C₆-H), 8.24 (d, J = 9.0 Hz, 1H, indole C₆-H), 8.42-8.49 (m, 2H, indole C₄-H, NH), 11.73 (s, 2H, CSNH, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 37.75 (CH₃), 55.34 (CH₂), 110.30 (CH), 110.97 (CH), 115.43, 115.58 (CH), 120.17 (CH), 120.69 (CH), 126.87 (CH), 127.75 (CH),

142.59, 146.60, 148.38, 153.75, 156.37, 162.92, 165.04; EIMS (70eV) *m/z* (%): 369 (3), 339 (3), 206 (100), 178 (80), 163 (93), 149 (20), 136 (51), 121 (97), 104 (26), 91 (91), 77 (26), 65 (17). Anal. Calcd for C₁₇H₁₅N₅O₄S (385): C, 52.99; H, 3.90; N, 18.18. Found: C, 52.74; H, 3.85; N, 18.12.

2.3.6. *N*-(3-Methoxybenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5f)

Yellow crystals (81%): mp 260-262 °C; IR (KBr,cm⁻¹): 3302, 3221 (NH), 1714 (C=O), 1597 (C=N), 1147 (C=S); ¹H-NMR (DMSO- d_6 , 300 MHz, δ , ppm): 3.74 (s, 3H, CH₃), 4.87 (d, J = 6.3 Hz, 2H, benzyl CH₂), 6.84 (dd, J = 8.4, 2.4 Hz, 1H, benzyl C₄-H), 6.92-6.95 (m, 2H, benzyl C₂-H, C₆-H), 7.11 (d, J = 8.7 Hz, 1H, indole C₇-H), 7.27 (t, J = 8.1Hz, 1H, benzyl C₅-H), 8.26 (dd, J = 8.7, 2.4 Hz, 1H, indole C₆-H), 8.54 (d, J = 2.4 Hz, 1H, indole C₄-H), 10.11 (t, J = 6.3 Hz, 1H, CSNH), 11.83 (s, 1H, indole NH), 12.44 (s,1H, NNH); ¹³C- NMR (DMSO- d_6 , δ , ppm): 113.03 (CH), 114.17, 120.74, 122.05, 123.75 (CH), 124.44 (CH), 127.86 (CH), 128.79 (CH), 130.24 (CH), 131.27, 133.50 (CH), 139.82, 141.58, 162.27, 176.19; EIMS (70eV) *m/z* (%): 385 (M⁺, 5), 309 (2), 220 (12), 193 (23), 161 (29), 136 (49), 121 (100), 109 (31), 104 (19), 91 (92), 77 (84), 65 (65). Anal.Calcd for C₁₇H₁₅N₅O₄S (385): C, 52.99; H, 3.90; N, 18,18. Found: C, 52.85; H, 3.88; N, 18.10.

2.3.7. *N*-(4-Methoxybenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5g)

Orange crystals (87%): mp 274-276 °C; IR (KBr, cm⁻¹): 3369, 3267 (NH), 1687 (C=O), 1620 (C=N), 1164 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 3.30 (DMSO, CH₃, benzylCH₂), 6.90 (d, J = 8.3 Hz, indole C₇-H), 7.11 (t, J = 9.0 Hz, benzyl C₅-H), 7.49-7.50 (m, 4H, benzyl C₂-H, C₃-H, C₆-H, indole C₆-H), 7.99 (d, J = 2.4 Hz, 1H, indole C₄-H), 10.92 (s, 1H, CSNH), 11.37 (s, 1H, indole NH), 12.67 (s, 1H, NNH); ¹³C- NMR (DMSO- d_6 , δ , ppm): 112.15 (CH), 112.48 (CH), 112.65 (CH), 113.03 (CH), 114.17, 121.34 (CH), 122.06, 123.78 (CH), 129.87 (CH), 129.99 (CH), 131.23, 133.49 (CH), 139.99, 141.57, 159.96, 162.28, 163.18, 176.14; EIMS (70eV) *m*/*z* (%): 385 (M⁺, 2), 309 (2), 248 (2), 206 (13), 194 (22), 161(57), 136 (34), 121 (100), 109 (22), 104 (11), 91 (38), 77 (74), 65

(28). Anal.Calcd for C₁₇H₁₅N₅O₄S (385): C, 52.99; H, 3.90; N, 18.18. Found: C, 52.88; H, 3.87; N, 18.14.

2.3.8. *N*-(2-Fluorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5h)

Yellow crystals (82%): mp 264-266 °C; IR (KBr, cm⁻¹): 3352, 3052 (NH), 1701 (C=O), 1607 (C=N), 1172 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 3.85 (s, 2H, benzyl CH₂), 6.90 (d, J = 8.3 Hz, 1H, indole C₇-H), 7.01 (t, J = 7.5 Hz, 1H, benzyl C₄-H), 7.14 (d, J = 8.0 Hz, 1H, benzyl C₆-H), 7.29 (td, J = 8.0, 1.2 Hz, 1H, benzyl C₅-H), 7.51-7.54 (m, 2H, benzyl C₃-H, indole C₆-H), 7.93 (d, J = 2.4 Hz, 1H, indole C₄-H), 10.56 (s, 1H, CSNH), 11.34 (s, 1H, indole NH), 12.59 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 55.65 (CH₂), 111.89 (CH), 112.98 (CH), 114.13, 120.06 (CH), 122.21, 123.52 (CH), 126.88, 127.97 (CH), 128.10 (CH), 130.73, 133.28 (CH), 141.46, 153.85, 162.24, 176.90; EIMS (70eV) m/z (%): 373 (M⁺, 4), 345 (2), 297 (2), 220 (4), 192 (26), 149 (47), 124 (53), 109 (100), 90 (12), 83 (55), 77 (21). Anal.Caled forC₁₆H₁₂FN₅O₃S (373): C, 51.47; H, 3.22; N, 18.77. Found: C, 51.39; H, 3.19; N, 18.69.

2.3.9. *N*-(3-Fluorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5i)

Yellow crystals (86%): mp 256-258 °C; IR (KBr, cm⁻¹): 3359, 3201 (NH), 1697 (C=O), 1606 (C=N), 1161 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 4.92 (d, J = 5.0 Hz, 2H, benzyl CH₂), 7.07-7.23 (m, 4H, benzyl C₂-H, C₅-H, C₆-H, indole C₇- H), 7.37-7.46 (m, 1H, benzyl C₄-H), 8.27 (d, J = 10 Hz, 1H, indole C₆-H), 8.54 (d, J = 2.5 Hz, 1H, indole C₄-H), 10.13 (t, J = 7.5Hz, 1H, CSNH), 11.84 (s, 1H, indole NH), 12.47 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 46.71 (CH₂), 111.21 (CH), 113.68 (CH), 116.20 (CH), 120.90, 123.16 (CH), 126.96 (CH), 130.24, 130.29, 130.37 (CH), 141.07, 142.67, 147.42, 162.95, 177.89; EIMS (70eV) m/z (%): 373 (M⁺, 4), 345 (2), 297 (2), 191 (24), 149 (20), 125 (52), 111 (74), 97 (90), 83 (73), 71 (85), 57 (100). Anal.Calcd forC₁₆H₁₂FN₅O₃S (373): C, 51.47; H, 3.22; N, 18.77. Found: C, 51.38; H, 3.20; N, 18.75.

2.3.10. *N*-(4-Fluorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5j)

Yellow fluffy crystals (89%): mp 276-278 °C; IR (KBr, cm⁻¹): 3342, 3130 (NH), 1722 (C=O), 1605 (C=N), 1161 (C=S)); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 3.30 (DMSO, benzyl CH₂), 6.91 (d, J = 8.0 Hz, 1H, indole C₇-H), 7.39-7.72 (m, 5H, benzyl C₂-H, C₃-H, C₅-H, C₆-H, indole C₆-H), 7.83 (d, J = 2.5 Hz, 1H, indole C₄-H), 10.90 (s, 1H, CSNH), 11.35 (s, 1H, indole NH), 12.64 (s, 1H, NNH); ¹³C- NMR (DMSO- d_6 , δ , ppm): 113.04 (CH), 114.15, 122.16, 123.65 (CH), 127.60 (CH), 129.02 (CH), 129.61 (CH), 130.61, 131.19, 131.57, 133.41 (CH), 135.94, 141.41, 162.22, 177.68; EIMS (70eV) m/z (%): 373 (M⁺, 2), 345 (2), 297 (2), 192 (24), 182 (9), 149 (25), 124 (40), 109 (100), 83 (70), 57 (100). Anal.Calcd forC₁₆H₁₂FN₅O₃S (373): C, 51.47; H, 3.22; N, 18.77. Found: C, 51.39; H. 3.19; N, 18.68.

2.3.11. *N*-(2-Chlorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5k)

Yellow crystals (91%): mp 268-270 °C; IR (KBr, cm⁻¹): 3348, 3157 (NH), 1692 (C=O), 1616 (C=N), 1188 (C=S); ¹H-NMR (DMSO- d_6 , 400 MHz, δ , ppm): 4.96 (d, J = 5.0 Hz, 2H, benzyl CH₂), 7.13 (d, J = 9.0 Hz, 1H, indole C₇-H), 7.27-7.37 (m, 3H, benzyl C₄-H, C₅-H, C₆-H), 7.49 (d, J = 7.5 Hz, 1H, benzyl C₃-H), 8.28 (d, J = 9.0 Hz, 1H, indole C₆-H), 8.55 (s, 1H, indole C₄-H), 10.12 (t, J = 6.0 Hz, 1H, CSNH), 11.85 (s, 1H, indole NH), 12.50 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 45.05 (CH₂), 111.24 (CH), 116.24 (CH), 120.89, 126.99 (CH), 127.01 (CH), 127.79, 128.65, 129.00 (CH), 130.00, 131.00, 135.01, 142.68, 147.45, 162.95, 178.19; EIMS (70eV) m/z (%): 354 (71), 325 (5), 268 (7), 206 (62), 192 (18), 183 (18), 178 (12), 163 (13), 140 (25), 125 (100), 106 (17), 89 (14), 77 (10), 63 (7). Anal.Calcd for C₁₆H₁₂ClN₅O₃S (389.5): C, 49.29; H, 3.08, N, 17.97. Found: C, 49.38; H, 3.07; N, 17.95.

2.3.12. *N*-(3-Chlorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5l)

Orange crystals (72%): mp 258-260 °C; IR (KBr,cm⁻¹): 3348, 3064 (NH), 1701 (C=O), 1605 (C=N), 1163 (C=S); ¹H-NMR (DMSO- d_6 , 300 MHz, δ , ppm): 4.90 (d, J = 6.0 Hz, 2H, benzyl CH₂), 7.12 (d, J = 9.0 Hz, 1H, indole C₇-H), 7.32-7.43 (m, 4H, benzyl C₂-H, C₄-H, C₅-H, C₆-H), 8.28 (dd, J = 8.7,

2.4 Hz, 1H, indole C₆-H), 8.54 (d, J = 2.4 Hz, 1H, indole C₄-H), 10.14 (t, J = 6.0 Hz, 1H, CSNH), 11.84 (s, 1H, indole NH), 12.47 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 47.14 (CH₂), 111.74 (CH), 116.70 (CH), 121.41, 126.43 (CH), 127.43 (CH), 127.50 (CH), 130.76 (CH), 130.91, 133.44, 141.25, 143.16, 147.97, 163.48, 178.34; EIMS (70eV) m/z (%): 389 (M⁺, 3), 313 (2), 220 (6), 192 (50), 178 (4), 165 (16), 140 (49), 125 (100), 89 (80), 77 (58), 64 (25). Anal.Calcd for C₁₆H₁₂ClN₅O₃S (389.5): C, 49.29; H, 3.08, N, 17.97. Found: C, 49.46; H, 3.09; N, 17.99.

2.3.13. *N*-(4-Chlorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5m)

Orange crystals (89%): mp 238-240 °C; IR (KBr,cm⁻¹): 3336, 3188 (NH), 1697 (C=O), 1604 (C=N), 1166 (C=S); ¹H-NMR (DMSO-*d*₆, 400 MHz, δ, ppm): 4.89 (d, *J* = 9.0 Hz, 2H, benzyl CH₂), 7.13 (d, *J* = 7.5 Hz, 1H, indole, C₇-H), 7.37-7.48 (m, 4H, benzyl C₂-H, C₃-H, C₅-H, C₆-H), 8.27 (d, *J* = 8.0 Hz, 1H, indole C₆-H), 8.54 (s, 1H, indole C₄-H), 10.14 (t, *J* = 9.0 Hz, 1H, CSNH), 11.84 (s, 1H, indole NH), 12.45 (s, 1H, NNH); ¹³C-NMR ((DMSO-*d*₆, δ, ppm): 46.56 (CH₂), 111.22 (CH), 116.07 (CH), 120.83, 126.77 (CH), 127.92 (CH), 129.09 (CH), 129.95, 131.52, 137.02, 142.30, 147.48, 162.35, 177.75; EIMS (70eV) *m/z* (%): 391/389 (M⁺, 2/7), 361/359 (3/2), 220 (4), 206 (46), 192 (25), 178 (7), 163 (7), 140 (25), 125 (100), 89 (13), 77 (8), 65 (16). Anal.Calcd for C₁₆H₁₂ClN₅O₃S (389.5): C, 49.29; H, 3.08, N, 17.97. Found: C, 49.36; H, 3.05; N, 17.93.

2.3.14. *N*-(2,4-Dichlorobenzyl)-2-[5-nitro-2-oxo-2,3-dihydro-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (5n)

Orange fluffy crystals (80%): mp 260-262 °C; IR (KBr,cm⁻¹): 3336, 3130 (NH), 1681 (C=O), 1606 (C=N), 1159 (C=S); ¹H-NMR (DMSO- d_6 , 300 MHz, δ , ppm): 4.91 (d, J = 6.0 Hz, 2H, benzyl CH₂), 7.13 (d, J = 8.4 Hz, 1H, indole C₇-H), 7.28 (d, J = 9.0 Hz, 1H, benzyl C₆-H), 7.46 (dd, J = 8.1, 2.1 Hz, 1H, benzyl C₅-H), 7.66 (d, J = 2.1 Hz, 1H, benzyl C₃-H), 8.28 (dd, J = 8.7, 2.4 Hz, 1H, indole C₆-H), 8.55 (d, J = 2.4 Hz, 1H, indole C₄-H), 10.13 (t, J = 6.0 Hz, 1H, CSNH), 11.86 (s, 1H, indole NH), 12.51 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm): 45.18 (CH₂), 111.77 (CH), 116.73 (CH), 121.34 (CH), 127.55 (CH), 127.85 (CH), 129.15 (CH), 129.74, 130.96, 132.76, 132.95, 134.79,

143.17, 147.99, 163.45, 178.70;EIMS (70eV) *m/z* (%): 390/388 (M⁺-HCl, 10/13), 208/206 (8/2), 192 (35), 176 (22), 159 (100), 140 (20), 123 (28), 103 (22), 89 (40), 75 (37), 63 (48), 51 (22). Anal.Calcd for C₁₆H₁₁Cl₂N₅O₃S (424): C, 45.28; H, 2.59; N, 16.51. Found: C, 45.22; H, 2.58; N, 16.49.

2.3.15. *N*-(**3,4-Dichlorobenzyl**)-2-[**5-nitro-2-oxo-2,3-dihydro**-*1H*-indol-3-ylidene]-1-hydrazinecarbothioamide (**5**0)

Orange crystals (79%): mp 270-272 °C; IR (KBr,cm⁻¹): 3336, 3228 (NH), 1693 (C=O), 1612 (C=N), 1166 (C=S); ¹H-NMR (DMSO- d_6 , 300 MHz, δ , ppm): 4.87 (d, J = 6.0 Hz, 2H, benzyl CH₂), 7.12 (d, J = 8.7 Hz, 1H, indole C₇-H), 7.35 (dd, J = 8.4, 2.1 Hz, 1H, benzyl C₆-H), 7.60-7.64 (m, 2H, benzyl C₂-H, C₅-H), 8.26 (dd, J = 8.7, 2.4 Hz, 1H, indole C₆-H), 8.52 (d, J = 2.1 Hz, 1H, indole C₄-H), 10.12 (t, J = 6.0 Hz, 1H, CSNH), 11.84 (s, 1H, indole NH), 12.47 (s, 1H, NNH); ¹³C-NMR (DMSO- d_6 , δ , ppm):): 46.64 (CH₂), 111.69 (CH), 116.64 (CH), 121.30 (CH), 127.46 (CH), 128.10 (CH), 129.65 (CH), 130.06, 130.86, 131.01, 131.36, 139.89, 143.13, 147.91, 163.41, 178.75; EIMS (70eV) m/z (%): 425/423 (M⁺, 2/2), 395 (4), 347 (3), 234/232 (4/5), 219/217 (17/21), 206 (84), 192 (29), 176/174 (27/36), 161/159 (94/100). 149/147 (91/61), 140 (39), 125/123 (15/7), 105/103 (15/7), 91/89 (10/10). Anal.Calcd for C₁₆H₁₁Cl₂N₅O₃S (424): C, 45.28; H, 2.59; N, 16.51. Found: C, 45.25; H, 2.54; N, 16.47.

2.4. X-ray structure determination

2.4.1. Crystallographic data collection and structural refinement

Specimen of **5g** of good quality and size was mounted on a thin glass fiber at room temperature and the reflection data were collected on a Bruker Kappa APE XII CCD diffractometer equipped with graphite mono-chromated Mo K α radiation ($\lambda = 0.71073$ Å). The data were corrected to Lorentz and polarization effect. The structure was solved using SHELXS-97.³⁷ Final refinement on F² was carried out by full-matrix least-squares techniques using SHELXL-97³⁷

2.5. Biological assays

In vitro urease inhibitory, phytotoxicity and cytotoxicity potential was determined in accordance with the literature protocols given in our published research articles.^{38,34,31}

2.6. Molecular docking studies

Molecular docking studies as well as HYDE assessment were carried out according to the literature method provided in our very recently published research paper.³⁹ In order to perform docking studies, the crystallographic structure of Jack bean urease (PDB ID: 3LA4) was obtained from the RCSB PDB database.⁴⁰

3. Results and discussion

This study describes the synthesis and *in vitro* evaluation of antiurease, phytotoxic and cytotoxic potential of fourteen presently and one previously reported⁴¹ N⁴-benzyl substituted 5-nitroisatin-thiosemicarbazones **5b-o** and **5a**, respectively.

3.1. Chemistry of compounds 5a-o

Appropriate *N*-substituted thiosemicarbazides **3** were treated with 5-nitroisatin **4** in 50% ethanol (aq.) to afford the respective isatin-3-thiosemicarbazones **5a-o** (Scheme 1) in good to excellent yields (72-91%).

The structures of the synthesized isatin-thiosemicarbazones **5a-o** were deduced by analytical and spectral (IR, ¹H-NMR, ¹³C-NMR, EIMS) as well as single crystal X-ray diffraction analyses. Satisfactory elemental (CHN) data were obtained in all the cases. Also, the spectral data were in tune with the respective literature.^{22,24,42,43} The IR spectra of **5a-o** showed NH stretchings of indole and thioamide functions in the 3359-3302 and 3250-3052 cm⁻¹ regions. The lactam C = O, azomethine C = N and thioamide C = S stretchings were observed in the 1722-1678, 1616-1597 and 1188-1147 cm⁻¹ regions, respectively.^{22,42,43} The ¹H-NMR spectra exhibited two separate singlets at δ 11.34-11.86 and 12.43-12.71 for indole NH and thiosemicarbazone N²-H, respectively, in all the cases except **5e**, wherein the indole NH resonance was observed as a multiplet at δ 8.42-8.49 due to overlapping of indole C₄-Hsignal, while the thiosemicarbazone N²-H resonated as a singlet at δ 11.73 together with thiosemicarbazone N⁴-H. Similarly, the thiosemicarbazone N⁴-H appeared as a triplet at δ 10.05-10.14 in all the cases^{22,24} except **5c**, **5g**, **5h** and **5j**, wherein it resonated as a singlet at δ 10.82, 10.92,

10.56 and 10.90, respectively. Furthermore, the benzyl CH₂ protons were observed as a doublet at δ 4.41-4.96 in all the cases^{22,24} except **5c**, **5g**, **5h** and **5j**; in the case of **5h**, they appeared as a singlet at δ 3.85, whereas in **5c**, **5g** and **5j**, they resonated together with the CH₃ group protons and/or the residual protons of DMSO. The ¹³C-NMR spectra of compounds **5a-o** supported the IR and ¹H-NMR findings. Also, the electron impact (EI) mass spectra of all the compounds except **5e**, **5k** and **5n** demonstrated molecular ions of varied intensity, which validated their molecular weights. The structures of **5e**, **5k** and **5n** were, however, made certain by the existence of the fragments related to thiosemicarbazone moieties, formed by the breakage of N-N and NH-CS bonds. The major fragmentation pathway involved the breakage of endocyclic NH-CO and exocyclic N-N, NH-CS bonds.⁴³ The proposed fragmentation pattern of **5c** is depicted in Figure S1 (see the SI). X-ray structure of a representative compound **5g** was determined to substantiate the assigned structures of the synthesized thiosemicarbazones **5a-o**.



Scheme 1. Reagents and conditions: (i) Et_3N , MeOH, CS_2 , stir, 30 °C (internal), 75 min., r.t., 1 h (ii) CH_3I , MeOH, stir, -10 °C, 20 min., r.t., 2 h (iii) $NH_2NH_2.H_2O$, EtOH, reflux, 2 h (iv) 50% EtOH (aq.), reflux, 2 h.

3.1.1. X-ray analysis of compound 5g

For the X-ray structure determinations, X-ray quality single crystals of 5g (Table 1) were obtained from its DMSO solution by slow evaporation at room temperature. The X-ray analysis demonstrates that it crystallizes in triclinic crystal lattice with the P^{-1} space group. The molecular structure (ORTEP diagram) of 5g along with crystallographic numbering scheme is depicted in Figure 1. The central N-iminothiourea (thiosemicarbazone) moiety is almost planar adopting preferably the S-cis/Strans conformation, most probably due to the formation of intramolecular hydrogen bond [N(5)-H(4)...N(3) 2.259 Å].⁴⁴⁻⁴⁶ The dihedral angles between S(1)-C(9)-N(4)-H(3), S(1)-C(9)-N(5)-H(4) and C(9)-N(4)-N(3)-C(7) are -1.61°, 176.07° and 173.08°, respectively. This planarity around the central N-iminothiourea moiety can be ascribed to extensive delocalization of N lone pairs onto the thiocarbonyl (C=S) function, which is clearly shown by the shorter N-C bond lengths [N(4)-C(9)]1.376(3) Å and N(5)-C(9) 1.325(3) Å], suggestive of the partial double bond character of N-C bonds. The slightly longer bond length in the case of N-C bond, directly connected to the imino (-C=N-) function of the N-iminothiourea moiety, indicates less delocalization of the N lone pair onto the thiocarbonyl (C=S) group, presumably due to the presence of inductively electron-attracting sp²hybridized N atom in its vicinity. Notably, it is the planarity and the mentioned intramolecular hydrogen bonding due to which the central N-iminothiourea moiety in such molecules adopts Scis/S-trans conformation, thus exposing the thioamide function to form centrosymmetric supramolecular dimers in their solid state packing.⁴⁴⁻⁴⁶ However, compound 5g has an additional amide moiety, offering a competing environment for the appearance of otherwise two reliable centrosymmetric dimer synthons $[R_2^2(8)\{\dots H-N-C=S\}_2$ and $R_2^2(8)\{\dots H-N-C=O\}_2]$ in crystal engineering.44-47

Crystal data5gCCDC1005939Chemical formulaC19H21N5O5S2

Table 1. Crystallographic data of 5g

$M_{ m r}$	463.53
Crystal system, space	group Triclinic, P^-1
Temperature (K	296
<i>a</i> , <i>b</i> , <i>c</i> (Å)	8.6754 (4), 11.9082 (6), 12.2545 (10)
α, β, γ (°)	112.750 (3), 98.066 (3), 106.130 (2)
$V(\text{\AA}^3)$	1076.79 (12)
Ζ	2
Radiation type	Μο Κα
μ (mm ⁻¹)	0.29
Crystal size (mn	a) $0.38 \times 0.23 \times 0.20$
	Data collection
Diffractometer	Bruker Kappa APEXII CCD
Absorption correct	tion Multi-scan (<i>SADABS</i> ; Bruker, 2005)
T_{\min}, T_{\max}	0.898, 0.944
No. of measured independent and observed [$I > 2\sigma($ reflections	l, 17062, 5051, 3780 l <i>I</i>]
$R_{\rm int}$	0.022
$(\sin \theta/\lambda)_{max}$ (Å ⁻¹) 0.658
0	Refinement
$R[F^2 > 2\sigma(F^2)], wR(F^2)$	F^2), S 0.045, 0.150, 1.07
No. of reflection	as 5051
No. of parameter	rs 283
H-atom treatmen	H-atom parameters constrained
$\Delta \rangle_{\rm max}, \Delta \rangle_{\rm min}$ (e Å	³) 0.85, -0.41



Figure 1. The molecular structure (ORTEP diagram) of 5g.

3.2. Biology of compounds 5a-o

3.2.1. Urease inhibition (*in vitro*)

The synthesized N⁴-benzyl substituted 5-nitroisatin-thiosemicarbazones **5a-o** were tested for their urease inhibitory potential against Jack bean urease. Thiourea served as a reference inhibitor in this assay. All the compounds proved to be highly potent inhibitors of the enzyme, demonstrating inhibitory activity (IC₅₀ = $0.87 \pm 0.25 - 8.09 \pm 0.23 \mu$ M) much better than the reference inhibitor, thiourea (IC₅₀ = $22.3 \pm 1.12 \mu$ M) (Table 2) and may thus act as convincing leads for further studies. The structure-activity relationship (SAR) studies in the isatin-thiosemicarbazones **5a-o** revealed that in comparison to compound **5a** having no substituent about the phenyl ring of the benzyl group attached to N⁴ of the thiosemicarbazone moiety, all the other compounds except **5k** and **5m**, irrespective of the type, number and place of the functional groups present on the phenyl ring, showed increased enzymatic activity (IC₅₀ values $0.87 \pm 0.25 - 2.54 \pm 0.18 \mu$ M vs. $5.92 \pm 0.25 \mu$ M). Compound **5g** possessing methoxy substituent at position-4 of the phenyl ring was found to be the

most potent urease inhibitor of the series in the present assay, exhibiting about seven and twenty six fold more activity than compound **5a** and the reference inhibitor, thiourea (IC₅₀ = $0.87 \pm 0.25 \mu$ M vs. 5.92 ± 0.25 and $22.3 \pm 1.12 \mu$ M, respectively). Next most potent urease inhibitor was compound 5c with methyl substituent at position-3 of the phenyl ring. This compound displayed somewhat less inhibitory activity than the most potent inhibitor 5g but much more than the reference inhibitor, thiourea (IC₅₀ value $1.27 \pm 0.19 \mu$ M vs. 0.87 ± 0.25 and $22.3 \pm 1.12 \mu$ M, respectively). The third most potent urease inhibitor was compound 5j having a fluoro substituent at position-4 of the phenyl ring. This compound, regardless of the nature and position of the substituents about the phenyl ring, exhibited almost the same activity as that of compound 5c (IC₅₀ value 1.29 \pm 0.27 vs.1.27 \pm 0.19 μ M). However, like compound 5c, it was found to be somewhat less active than the most potent urease inhibitor of the series i.e. compound 5g (IC₅₀ value 1.29 ± 0.27 vs. $0.87 \pm 0.25 \mu$ M) but about five and seventeen times more active than compound 5a and the reference inhibitor, thiourea (IC₅₀) value 1.29 ± 0.27 vs. 5.92 ± 0.25 and $22.3 \pm 1.12 \mu$ M, respectively). The remaining relatively more potent urease inhibitors were the mono- and dichloro-substituted derivatives 51 and 5n. Among these, 51 having the substituent at position-3 of the phenyl ring was found to be a little bit less active than the corresponding methyl-substituted derivative 5c but about four and fourteen times more active than compound **5a** and the reference inhibitor, thiourea (IC₅₀ value $1.63 \pm 0.18 \mu$ M vs. 1.27 ± 0.19 , 5.92 ± 0.25 and $22.3 \pm 1.106 \mu$ M, respectively). In contrast, the dichloro-substituted derivative **5n** with the substituents at positions-2,4 (ortho, para) of the phenyl ring was four times more active than the monochloro-substituted derivative 5k bearing the substituent at position-2 of the phenyl ring (IC₅₀ value 1.94 ± 0.13 vs. 8.09 ± 0.23 µM). This indicated that compound **5n** with an additional inductively electron-withdrawing chloro function at position-4 of the phenyl ring meddled with the enzyme in a different manner, giving rise to an extensive increase in its inhibitory activity.

The above results showed that the attachment of variously substituted benzyl groups (having one or two inductively electron-donating or –withdrawing functionalities about the phenyl ring) to N^4 of the

thiosemicarbazone moieties caused the molecules to interfere with the enzyme differently and in some instances much more competently.

Table 2. Inhibition of urease by compounds 5a-0							
	O_2N $NNHCSNHCH_2R$ O_2N O_2N						
Compound	R	$IC_{50} \pm SEM^*$					
	CII	<u>(µM)</u>					
5a	C_6H_5	5.92 ± 0.25					
5b	$2-CH_3C_6H_4$	1.39 ± 0.13					
5c	$3-CH_3C_6H_4$	1.27 ± 0.19					
5d	$4-CH_3C_6H_4$	1.52 ± 0.11					
5e	$2-CH_3OC_6H_4$	1.33 ± 0.12					
5 f	$3-CH_3OC_6H_4$	1.34 ± 0.17					
5g	$4-CH_3OC_6H_4$	0.87 ± 0.25					
5h	$2-FC_6H_4$	1.58 ± 0.16					
5 i	$3-FC_6H_4$	2.54 ± 0.18					
5j	$4-FC_6H_4$	1.29 ± 0.27					
5k	$2-ClC_6H_4$	8.09 ± 0.23					
51	$3-ClC_6H_4$	1.63 ± 0.18					
5m	$4-ClC_6H_4$	6.34 ± 0.18					
5n	$2,4-(Cl)_2C_6H_3$	1.94 ± 0.13					
50	$3,4-(Cl)_2C_6H_3$	2.11 ± 0.07					
Thiourea**		22.3 ± 1.06					

Table 2. Inhibition	n of urease	by compounds	5a-o
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Standard error of the mean; ** standard inhibitor for urease.

3.2.2. Phytotoxicity (in vitro)

The synthesized 5-nitroisatin-thiosemicarbazones 5a-o were further tested for their phytotoxicity potential at 1000, 100, 10 or 500, 50, 5 µg / mL concentrations. All the compounds of this series except 5f, 5h and 5j appeared to be active in the present assay, displaying weak or non-significant (5-35%) plant growth inhibition at the highest tested concentrations (1000 or 500 µg/mL) (Table 3).

Table 3. Growth inhibition of Lemna aequinocitalis by compounds 5a-o* at different concentrations

		O ₂ N	NNHCSI	NHCH ₂ R			
		~	Ĥ				
		1000	100	10	500	50	5
Compound	R	µg/mL	µg/mL	µg/mL	µg/mL	µg/mL	µg/mL
		(% G.I)	(% G.I)	(% G.I)	(% G.I)	(% G.I)	(% G.I)
5a	C_6H_5	_	—	—	10	00	00
5b	$2-CH_3C_6H_4$	15	05	00	_		_
5c	$3-CH_3C_6H_4$	20	05	00	-	-	_
5d	$4-CH_3C_6H_4$	05	00	00	-	_	_
5e	$2-CH_3OC_6H_4$	35	15	00	-	—	_
5 f	$3-CH_3OC_6H_4$	00	00	00		_	_
5g	$4-CH_3OC_6H_4$	10	06	00	_	_	_
5h	$2-FC_6H_4$	00	00	00	_	_	_
5i	$3-FC_6H_4$	10	00	00	_	_	_
5ј	$4-FC_6H_4$	00	00	-00	_	_	_
5k	$2-ClC_6H_4$	05	00	00	_	_	_
51	$3-ClC_6H_4$	05	00	00	_	_	_
5m	$4-ClC_6H_4$	05	00	00	_	_	_
5n	$2,4-(Cl)_2C_6H_3$	_	-	_	30	00	00
50	$3,4-(Cl)_2C_6H_3$	_		_	10	00	00

*The standard drug, paraquat, shows 100% growth inhibition at a concentration of 0.015 μ g/mL; G.I: growth inhibition.

3.2.3. Cytotoxicity (in vitro)

The synthesized thiosemicarbazones **5a-o** were also evaluated for their cytotoxicity potential by a brine shrimp lethality bioassay, using etoposide (a standard anticancer drug) as a reference point. From the data presented in Table 4, it is evident that all the compounds of this series except **5i** gave LD_{50} values >2.36 x $10^{-4} - 2.82 x 10^{-4}$ M against *Artemia salina* in the present assay and, therefore, can be considered to be almost inactive. Compound **5i** bearing 3-fluorobenzyl function at N⁴ of the thiosemicarbazone moiety, however, proved to be active, exhibiting cytotoxic activity with LD_{50} value 2.55×10^{-5} M.

Table 4. Brine shrimp (Artemia salina) lethality bioassay for compounds 5a-o



-	Compound	R	LD ₅₀ (M)	0
-	5a	C_6H_5	$>2.82 \times 10^{-4}$	
	5b	$2-CH_3C_6H_4$	>2.71 x 10 ⁻⁴	2-
	5c	$3-CH_3C_6H_4$	>2.71 x 10 ⁻⁴	
	5d	$4-CH_3C_6H_4$	>2.71 x 10 ⁻⁴	
	5e	$2-CH_3OC_6H_4$	>2.60 x 10 ⁻⁴	
	5f	3-CH ₃ OC ₆ H ₄	>2.60 x 10 ⁻⁴	
22 M.L. L	5g	4-CH ₃ OC ₆ H ₄	>2.60 x 10 ⁻⁴	1
3.3. Molecular	5h	$2-FC_6H_4$	>2.68 x 10 ⁻⁴	docking
3.3.1. Molecular of compounds	5i	$3-FC_6H_4$	2.55 x 10 ⁻⁵	docking studies 5a-o
To rationalize	5j	$4-FC_6H_4$	>2.68 x 10 ⁻⁴	the obtained in
vitro biological	5k	$2-ClC_6H_4$	>2.57 x 10 ⁻⁴	results,
molecular	51	3-ClC ₆ H ₄	>2.57 x 10 ⁻⁴	docking studies
of all the	5m	$4-ClC_6H_4$	>2.57 x 10 ⁻⁴	inhibitors 5a-o
were performed	5n	2,4-(Cl) ₂ C ₆ H ₃	$>2.36 \times 10^{-4}$	against Jack
bean urease PDB	50	3,4-(Cl) ₂ C ₆ H ₃	>2.36 x 10 ⁻⁴	ID: 3LA4. It was
rationalized that	*Tested a	at 100, 10, 1 μg/mL co	ncentrations.	most of the
compounds were				found to bind in

the same region of the active site. Before docking the most active inhibitor **5g** in the active site of the urease enzyme, molecular docking studies of the co-crystallized ligand KCX (Lysine NZ-carboxylic acid) was carried out. The ligand was reproduced successfully with an RMSD value of 1.5 Å. Molecular docking studies of all the test compounds **5a-o** were carried out to investigate the

interactions inside the active site. The most potent inhibitor 5g was successfully oriented inside the active site of the target enzyme. The putative binding mode of 5g is shown in Figure 2. After reproducing the co-crystallized reference ligand into the active site of the receptor, all the derivatives were docked inside the active site. The docking studies supported the experimental results that most of the inhibitors inhibit the catalytic site of the enzyme. As described previously,^{38,39} the most important interactions i.e., hydrogen bonding were seen between the oxygen and hydrogen atoms of the inhibitors and the amino acid residues Arg609, Asp494, Ala440 and His593. Similarly, as reported earlier by us,⁴⁸ stable hydrogen bonding interactions within the active pocket were noticed with the amino acid residues His492, His593 and His594. The 4-methoxy benzyl substituent was oriented towards bottom of the active site, while the isatin scaffold located towards opening of the active site and the mid-gorge area. Moreover, the thiosemicarbazone part was found to interact with Arg439 and Ala636 in the middle of the active site. Similarly, weak interactions were shown by all the compounds with Ni842 in the catalytic site of the urease enzyme. Furthermore, the benzyl substituent also showed π - π interaction with the imidazole ring of His594 and π -charge interactions were identified by the thiosemicarbazone moieties of the inhibitors. The putative binding pose of compound 5g was selected after HYDE assessment and visualization of top 30 poses and is represented in Figure 2.

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Figure 2. A possible binding mode of compound 5g is shown. Carbon atoms of 5g are colored blue, while that of protein light green; oxygen, sulfur and nitrogen atoms are colored red, yellow and blue, respectively. The two nickel ions in the active site are represented as small dark brown spheres.

3.3.2. HYDE assessment of compounds 5a-o

The HYDE affinity assessment was done for the first 20 top ranking docked conformations. The binding free energy ΔG , FlexX docking score and the most favorable poses for all the derivatives **5a**-**o** are given in Table 5. Most of the compounds bind to the receptor with a very high binding affinity.

Compound	FlexX score of the top	Poser rank	Binding free energy ΔG
	ranking pose		(kJ mol ⁻¹)
5a	-20.82	10	-12
5b	-20.78	8	-5
5c	-20.45	17	-5
5d	-20.75	12	-14
5e	-18.60	15	-5

Table 5. Docking scores and their corresponding ranks by HYDE affinity assessment

5f	-22.62	2	-2
5g	-19.62	19	-13
5h	-20.88	9	-5
5 i	-22.25	8	-2
5ј	-20.78	12	-13
5k	-20.30	16	-11
51	-21.10	8	-4
5m	-20.97	7	-12
5n	-20.33	9	-12
50	-19.94	15	-7

4. Conclusions

In this study, a series of fifteen N⁴-benzyl substituted 5-nitroisatin-thiosemicarbazones **5a-o** have been synthesized and evaluated for *in vitro* urease inhibitory, phytotoxic and cytotoxic activities. All the synthesized thiosemicarbazones proved to be highly potent inhibitors of the enzyme, exhibiting inhibitory activity even better than the reference inhibitor (i.e. thiourea) used in the assay. These compounds, by demonstrating no or weak/non-significant phytotoxicity, attract much attention to their utility as valuable soil ureases inhibitors because they could be mixed with fertilizers in small quantities to enhance the overall effectiveness of nitrogen utilization. Also, being non toxic, they could prove to be credible candidates for orally effective remedial agents used for the treatment of certain clinical conditions induced by microbial ureases.

This study presents the first case of display of urease inhibition by N^4 -benzyl substituted 5nitroisatin-thiosemicarbazones and provides a tangible basis for further studies on such compounds to develop more potent and safe urease inhibitors of agricultural/ medicinal interest.

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Graphical abstract

Synthesis, X-ray molecular structure, biological evaluation and molecular docking studies of some N⁴-benzyl substituted 5-nitroisatin-3-thiosemicarbazones

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A possible binding mode of compound **5g** bound to active pocket of Jack bean urease