Synthesis of Hyrtiosin B, a Bis-indole Alkaloid from the Okinawan Marine Sponge *Hyrtios erecta*

Jan Bergman,* Tomasz Janosik, Ann-Louise Johnsson

Department of Organic Chemistry, KI, Novum Research Park, SE-141 57 Huddinge, Sweden

Fax +46(8)6081501; E-mail: jabe@cnt.ki.se

Received 16 July 1998; revised 23 November 1998

Abstract: The bis-indole alkaloid hyrtiosin B (1) has been prepared in good yield from 5-benzyloxyindole (4a) or 5-methoxyindole (4b), involving a coupling reaction of the zinc metallated indoles with the corresponding glyoxylyl chlorides **5a–b**, and finally debenzylation or demethylation using boron tribromide to yield the title compound 1.

Key words: hyrtiosin B, bis-indole alkaloids, transmetallation, *O*-dealkylation

Several classes of compounds have been isolated from marine organisms belonging to the genus *Hyrtios*, such as terpenoids¹ and sterols.² In 1990, the first isolation of alkaloids from this genus was reported;³ hyrtiosin B (1), a bis-indole alkaloid, was found in the Okinawan marine sponge *Hyrtios erecta* together with the related compounds hyrtiosin A (2) and the previously reported 5-hydroxyindole-3-carboxaldehyde (3),⁴ and was demonstrated to possess cytotoxic activity against human epidermoid carcinoma KB cells in vitro.

yields (36–65%) (Scheme). Acylation of zincated indoles has previously been shown to give yields superior to those resulting from acylation of indole magnesium halides.⁶ The final O-dealkylation was accomplished by treatment of **6a–b** with boron tribromide in dichloromethane to give hyrtiosin B (1) in good yields (79-81%) after purification by column chromatography. The benzyl derivative 6a was debenzylated easily with BBr₃ (4 equiv), whereas demethylation of **6b** required a larger excess of BBr₃ (12 equiv) and longer reaction times. Attempted catalytic debenzylations of **6a** in ethanol or methanol failed, as did attempted dealkylations using Prey's method⁷ (reflux in pyridine hydrochloride). The spectral data of synthetic hyrtiosin B were identical in all respects with those of the natural product. The possibility to prepare a wide diversity of unsymmetrical derivatives was demonstrated by preparation of compound 6c by treatment of 1-indolylmagnesium bromide with 5a as outlined in the Scheme.





We now describe a short and efficient synthesis of **1** in three steps from 5-benzyloxyindole (**4a**) or 5-methoxyindole (**4b**). The indoles **4a** or **4b** were transformed in excellent yields into the corresponding indole-3-glyoxylyl chlorides **5a–b** according to the procedure of Speeter and Anthony.⁵ The key step in our approach is a coupling reaction involving transmetallation of the indole magnesium halides based on the indoles **4a** or **4b** with anhydrous zinc chloride,⁶ and subsequent treatment of the resulting zincated indoles with acid chlorides **5a–b**, which afforded the symmetrical coupled products **6a–b** in moderate

Scheme

The monobenzyloxy substituted compound 6c was similarly debenzylated using boron tribromide to yield the hydroxy derivative 7, a compound closely related to hyrtiosin B (1). A compound structurally related to 7 (Cl instead of OH) has previously been prepared by Van

Vranken et al.⁸ using the technique previously developed by Bergman and Venemalm.⁶

5-Benzyloxyindole and 5-methoxyindole were purchased from Lancaster and Aldrich respectively. $ZnCl_2$ (1.0 M solution in Et_2O) was purchased from Aldrich. BBr₃ (1.0 M solution in CH_2Cl_2) was purchased from Lancaster. All solvents were purified by distillation, Et_2O was stored over sodium wire. The equipment for the anhydrous reactions was dried in an oven and was assembled hot. NMR spectra were recorded on a Bruker DPX 300 spectrometer (300 MHz, TMS) and the IR spectra were recorded using a Perkin Elmer 1600 FT-IR instrument. Mass spectra were obtained using a Micromass Platform II spectrometer. Elemental analyses were performed by H. Kolbe Mikroanalytisches Laboratorium, Mülheim an der Ruhr, Germany. Melting points were taken on a Büchi Melting Point B-545 apparatus and are uncorrected. Chromatography was performed on Merck silica gel 60, TLC analyses were run on Merck silica gel F_{254} plates.

(5-Benzyloxyindol-3-yl)glyoxylyl Chloride (5a)

A solution of 5-benzyloxyindole (**4a**; 1.12 g, 5.04 mmol) in anhyd Et_2O (20 mL) was cooled to 0°C. Oxalyl chloride (0.45 mL, 5.15 mmol) was added dropwise, producing an orange precipitate. The mixture was stirred at 0°C for 1 h, thereafter **5a** was collected as an orange solid, washed with anhyd Et_2O and dried; yield: 1.38 g (87%). The product was used immediately in the next step, or was stored in a dessicator; mp 149–151°C (dec) [Lit.⁵ mp 146–150°C (dec)].

IR(KBr): $\nu = 3189,\,1784,\,1630,\,1474,\,1430,\,1266,\,988,\,775,\,716$ $cm^{-1}.$

(5-Methoxyindol-3-yl)glyoxylyl Chloride (5b)

The same procedure as above was used, with 5-methoxyindole (**4b**; 735 mg, 5.00 mmol) and oxalyl chloride (0.45 mL, 5.15 mmol); yield: 1.05 g (88%); orange solid; mp 137–138°C (dec) (Lit.⁹ mp 134°C).

IR(KBr): v = 3195, 1781, 1624, 1476, 1427, 1267, 991, 774, 706 cm⁻¹.

1,2-Di(5-benzyloxyindol-3-yl)ethane-1,2-dione (6a)

To a solution of EtMgBr prepared from Mg (78 mg, 3.21 mmol) and bromoethane (0.24 mL, 3.22 mmol) in Et₂O (10 mL) was added a solution of **4a** (670 mg, 3.01 mmol) in Et₂O (15 mL) at r.t. The mixture was allowed to stir for 15 min, then a 1.0 M Et₂O solution of ZnCl₂ (3.1 mL, 3.1 mmol) was added, and the stirring was continued for 30 min, followed by addition of **5a** (941 mg, 3.00 mmol) in one portion. After stirring at r.t. for 3.5 h, the mixture was quenched with sat. aq NH₄Cl solution (50 mL), and extracted with EtOAc (100 mL). The organic phase was washed with brine (30 mL), dried (MgSO₄) and evaporated to dryness. Trituration of the residue with EtOH (15 mL) afforded **6a** as a yellow solid (710 mg). A second crop (260 mg) was collected after concentration of the mother liquor; total yield: 970 mg (65%); mp 233°C.

IR(KBr): v = 3386, 3319, 1596, 1506, 1425, 1252, 1188, 1113, 784 cm⁻¹.

¹H NMR (DMSO-*d*₆): δ = 5.20 (s, 4 H), 7.00 (dd, *J* = 2.5, 8.8 Hz, 2 H), 7.33–7.52 (m, 12 H), 7.89 (d, *J* = 2.2 Hz, 2 H), 8.16 (d, *J* = 2.9 Hz, 2 H), 12.15 (d, *J* = 2.2 Hz, 2 H).

¹³C NMR (DMSO- d_6): $\delta = 69.7$ (t), 104.6 (d), 112.4 (s), 113.4 (d), 113.7 (d), 126.5 (s), 127.6 (d), 127.7 (d), 128.4 (d), 131.7 (s), 137.4 (d), 154.9 (s), 188.7 (s).

Anal. Calcd for $C_{32}H_{24}N_2O_4$: C, 76.79; H, 4.83; N, 5.60. Found: C, 76.88, H, 4.81; N, 5.62.

1,2-Di(5-methoxyindol-3-yl)ethane-1,2-dione (6b)

To a solution of EtMgBr prepared from Mg (102 mg, 4.20 mmol) and bromoethane (0.315 mL, 4.22 mmol) in Et₂O (10 mL) was added a solution of **4b** (588 mg, 4.00 mmol) in Et₂O (15 mL) at r.t. After 15 min a 1.0 M Et₂O solution of ZnCl₂ (4.1 mL, 4.1 mmol) was added and the mixture was stirred for 30 min. Compound **5b** (954 mg, 4.01 mmol) was then added in one portion, and the stirring was continued for 3.5 h. Workup with sat. aq NH₄Cl solution (50 mL) and extraction with EtOAc (100 mL) [in this case addition of MeOH (2 mL) was necessary in order to obtain a clear two-phase mixture], washing of the organic extracts with brine, drying (MgSO₄) and subsequent evaporation of the solvent gave a residue that afforded **6b** (455 mg) as a pink solid after trituration with EtOH (15 mL). A second crop (50 mg) was collected after concentration of the mother liquor; yield: 505 mg (36%); mp 309–310°C.

IR (KBr): $v = 3190, 1606, 1477, 1426, 1265, 1211, 783, 710 \text{ cm}^{-1}$.

¹H NMR (DMSO- d_6): $\delta = 3.83$ (s, 6 H), 6.92 (dd, J = 2.5, 8.8 Hz, 2 H), 7.44 (d, J = 8.8 Hz, 2 H), 7.79 (d, J = 2.4 Hz, 2 H), 8.17 (s, 2 H), 12.13 (br s, 2H).

¹³C NMR (DMSO- d_6): $\delta = 55.3$ (q), 103.1 (d), 112.4 (s), 113.2 (d), 113.4 (d), 126.6 (s), 131.5 (s), 137.4 (d), 155.9 (s), 188.7 (s).

MS (EI, 70 eV): m/z = 348 (M⁺, 5), 175 (11), 174 (100), 159 (18), 146 (17), 131 (23), 119 (11), 103 (17), 76 (14).

HRMS (EI): m/z Calcd for $C_{20}H_{16}N_2O_4$: 348.1110. Found: 348.1094.

Hyrtiosin B (1) by Debenzylation of 6a

A 1.0 M solution of BBr₃ in CH₂Cl₂ (4.0 mL, 4.0 mmol) was added dropwise to a stirred suspension of **6a** (500 mg, 1.00 mmol) in CH₂Cl₂ (20 mL) at -78° C under N₂ atmosphere. The mixture was allowed to reach r.t. over 18 h and H₂O (30 mL) was added. Extraction of the mixture with three portions of EtOAc (total 120 mL), washing of the combined extracts with brine, drying (MgSO₄) and finally evaporation of the solvent gave a brown residue, which was subjected to column chromatography using EtOAc as eluent to yield pure hyrtiosin B (1) (254 mg, 79%) as a pale green solid; mp >400°C (Lit.³ mp >310°C).

IR(KBr): v = 3259 (br), 1595, 1514, 1466, 1421, 1202, 786, 723 cm⁻¹.

¹H NMR (DMSO-*d*₆): δ = 6.76 (dd, *J* = 2.3, 8.7 Hz, 2 H), 7.32 (d, *J* = 8.7 Hz, 2 H), 7.66 (d, *J* = 2.0 Hz, 2 H), 8.03 (d, *J* = 3.3 Hz, 2 H), 9.13 (br s, 2 H), 11.97 (d, *J* = 2.7 Hz, 2 H).

¹³C NMR (DMSO-*d*₆): δ = 105.8 (d), 112.1 (s), 113.0 (d), 113.1 (d), 126.8 (s), 130.8 (s), 136.9 (d), 153.6 (s), 188.8 (s).

Anal. Calcd for $C_{18}H_{12}N_2O_4$: C, 67.50; H, 3.78; N, 8.75. Found: C, 67.38; H, 3.84; N, 8.62.

Hyrtiosin B (1) by Demethylation of 6b

A 1.0 M solution of BBr₃ in CH₂Cl₂ (6.0 mL, 6.0 mmol) was added to a suspension of **6b** (174 mg, 0.50 mmol) in CH₂Cl₂ (15 mL) at -78° C under a N₂ atmosphere. The mixture was allowed to reach r.t. over 72 h. The same workup procedure as above yielded hyrtiosin B (1) (130 mg, 81%) as a pale green solid, in all respects identical with the sample prepared by debenzylation of **6a**.

1-(5-Benzyloxyindol-3-yl)-2-(indol-3-yl)-ethane-1,2-dione (6c)

1-Indolylmagnesium bromide in Et₂O (15 mL), prepared from Mg (102 mg, 4.20 mmol), bromoethane (0.315 mL, 4.22 mmol) and indole (468 mg, 4.00 mmol), was transmetallated by treating with a 1.0 M solution of ZnCl₂ in Et₂O (4.1 mL, 4.1 mmol) at r.t. The resulting mixture was stirred for 30 min. Compound **5a** (1.25 g, 3.92 mmol) was added in one portion, and stirring was continued for 16 h. The mixture was quenched with sat. aq NH₄Cl solution (50 mL) and extracted with EtOAc (100 mL). The organic extract was washed with brine (50 mL) and dried (MgSO₄). Evaporation of the solvent gave an orange residue, which was triturated with Et₂O (20 mL) containing EtOH (1 mL) to give **6c** (810 mg, 52%) as a cream coloured solid; mp 210.5–211.5°C.

IR(KBr): $\nu = 3232$ (br), 1598, 1511, 1436, 1241, 1122, 782, 742 $cm^{-1}.$

¹H NMR (DMSO- d_6): $\delta = 5.17$ (s, 2 H), 7.01 (dd, J = 2.5, 8.8 Hz, 1 H), 7.28–7.57 (m, 9 H), 7.90 (d, J = 2.4 Hz, 1 H), 8.17 (d, J = 3.2 Hz, 1 H), 8.23 (d, J = 3.1 Hz, 1 H), 8.27–8.30 (m, 1 H), 12.16 (d, J = 2.5 Hz, 1 H), 12.25 (d, J = 2.4 Hz, 1 H).

¹³C NMR (DMSO- d_6): $\delta = 69.7$ (t), 104.7 (d), 112.4 (s), 112.5 (s), 112.6 (d), 113.4 (d), 113.7 (d), 121.3 (d), 122.4 (d), 123.4 (d), 125.6 (s), 126.5 (s), 127.6 (d), 127.7 (d), 128.4 (d), 131.7 (s), 136.7 (s), 137.3 (d), 137.4 (d), 137.4 (s), 154.9 (s), 188.6 (s), 188.8 (s).

MS (EI, 40 eV): *m*/*z* = 394 (M⁺, 2), 251 (11), 250 (67), 160 (14), 159 (30), 144 (76), 131 (29), 116 (21), 103 (11), 91 (100), 89 (14).

HRMS (EI): m/z Calcd for $C_{25}H_{18}N_2O_3$: 394.1317. Found: 394.1320.

1-(5-Hydroxyindol-3-yl)-2-(indol-3-yl)-ethane-1,2-dione (7)

To a suspension of **6c** (197 mg, 0.50 mmol) in CH_2Cl_2 (15 mL) at $-78^{\circ}C$ under N₂ atmosphere was added a 1.0 M solution of BBr₃ in CH_2Cl_2 (3.0 mL, 3.0 mmol). The mixture was allowed to reach r.t. over 24 h, H₂O (30 mL) was added, followed by extraction with EtOAc (75 mL). The aqueous phase was extracted once more with EtOAc (25 mL). Washing of the combined organic extracts with brine, drying (MgSO₄), evaporation of the solvent and finally column chromatography (elution with EtOAc) gave **7** as a pale green solid (125 mg, 82%); mp 271–272°C.

IR(KBr): $v = 3408, 3291, 3244, 1597, 1508, 1423, 1240, 1112, 780, 729 \text{ cm}^{-1}$.

¹H NMR (DMSO- d_6): $\delta = 6.77$ (dd, J = 2.3, 8.7 Hz, 1 H), 7.25–7.29 (m, 2 H), 7.33 (d, J = 8.7 Hz, 1 H), 7.52-7.55 (m, 1 H), 7.68 (d, J =

2.2 Hz, 1 H), 8.06 (s, 1 H), 8.19 (s, 1 H), 8.24-8.27 (m, 1 H), 9.16 (br s, 1 H), 12.08 (br s, 2 H).

¹³C NMR (DMSO- δ_6): $\delta = 105.6$ (d), 112.1 (s), 112.5 (d), 112.6 (s), 113.0 (d), 113.2 (d), 121.3 (d), 122.4 (d), 123.4 (d), 125.6 (s), 126.8 (s), 130.8 (s), 136.7 (s), 137.0 (d), 137.2 (d), 153.6 (s), 188.5 (s), 189.1 (s).

MS (EI, 70 eV): *m*/*z* = 304 (M⁺, 11), 160 (90), 144 (100), 132 (42), 116 (53), 105 (33), 89 (73), 77 (27).

HRMS (EI): m/z Calcd for $C_{18}H_{12}N_2O_3$: 304.0848. Found: 304.0844.

References

- (1) (a) Amade, P.; Chevolot, L.; Perzanowski, H. P.; Scheuer P. J. *Helv. Chim. Acta* **1983**, *66*, 1672.
 (b) Crews, P.; Bescansa, P. *J. Nat. Prod.* **1986**, *49*, 1041.
 (c) Crews, P.; Bescansa, P.; Bakus, G. J. *Experientia* **1985**, *41*, 690.
- (2) Koch, P.; Djerassi, C.; Lakshmi, V.; Schmitz, F. J. *Helv. Chim. Acta* **1983**, *66*, 2431.
- (3) Kobayashi, J.; Murayama, T.; Ishibashi, M.; Kosuge, S.; Takamatsu, M.; Ohizumi, Y.;Kobayashi, H.; Ohta, T.; Nozoe, S.; Sasaki, T. *Tetrahedron* **1990**, *46*, 7699.
- (4) (a) Kveder, S.; Iskric, S. *Biochem. J.* **1965**, *94*, 509.
 (b) Iskric, S.; Stancic, L.; Kveder, S. *Clin. Chim. Acta* **1969**, 25, 435.
- (5) Speeter, M. E.; Anthony, W.C. J. Am. Chem. Soc. 1954, 76, 6208.
- (6) Bergman, J.; Venemalm, L. *Tetrahedron* **1990**, *46*, 6061.
- (7) (a) Prey, V. Ber. Dtsch. Chem. Ges. 1941, 74, 1219.
 (b) Prey, V. Ber. Dtsch. Chem. Ges. 1942, 75, 350.
- (8) Gilbert, E. J.; Ziller, J. W.; Van Vranken, D. L. *Tetrahedron* 1997, 53, 16553.
- (9) Takatori, K.; Takashima, M. Yakugaku Zasshi 1963, 83, 795.