

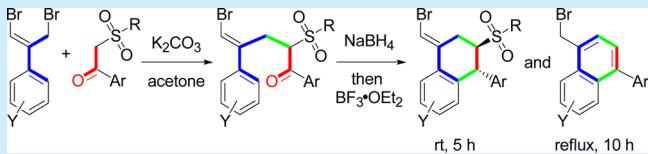
Synthesis of 1-Aryltetralins and 1-Arylnaphthalenes via (4 + 2) Annulation of β -Ketosulfones with Styryl Bromides

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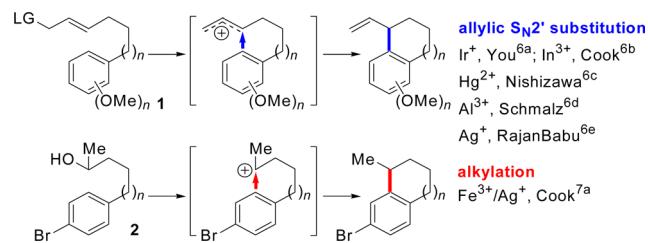
Supporting Information

ABSTRACT: A novel route has been developed for the synthesis of various substituted 1-aryltetralins **6** and 1-arylnaphthalenes **8** via (1) K_2CO_3 -mediated α -styrylation of β -ketosulfones **3** with bromostyryl bromides **4** and (2) stereocontrolled $NaBH_4$ -promoted reduction of the resulting γ -alkenones **5**, followed by $BF_3\cdot OEt_2$ -catalyzed intramolecular annulation of the corresponding γ -alkenols **7** under rt/5 h and reflux/10 h conditions, respectively. The key structures of **6** and **8** were confirmed by X-ray crystallographic analysis. A plausible mechanism has been proposed.



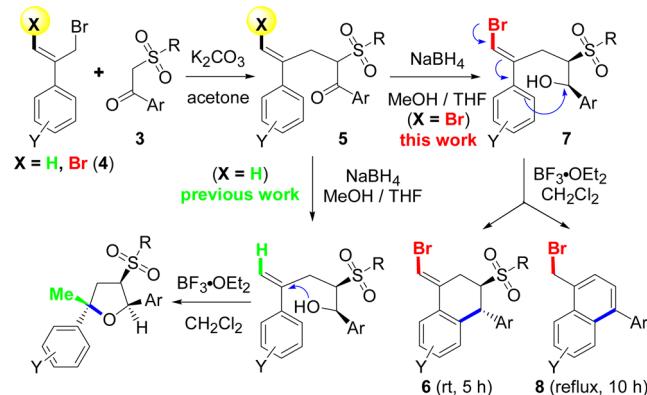
Tetralin is a highly privileged core structure found in many valuable biologically active molecules, pharmaceuticals, and natural products.^{1–4} With regard to the construction of this key building block, the 90-year-old Darzens tetralin synthesis (H_2SO_4 -mediated intramolecular ring-closure of a 1-phenyl-4-pentene)⁵ still stands out as one of the most direct and convenient methods. It uses arenes with an allylic chain⁶ **1** or an unactivated alcohol⁷ **2** as the starting materials; it is especially effective when combined with the recent remarkable advances in metal-catalyzed annulation. Various catalysts have been employed to accelerate the formation of a carbon–carbon bond via a carbocation-like intermediate (Scheme 1).⁸

Scheme 1. Intramolecular Friedel–Crafts Substitution Reaction



As part of our recent studies of $BF_3\cdot OEt_2$ -promoted tandem cyclization reactions,⁹ we devised an efficient stereocontrolled synthesis of sulfonyl 2,5-diaryltetrahydrofurans in good yield via the Lewis acid promoted intramolecular hydroalkoxylation of sulfonyl γ -alkenols ($X = H$),¹⁰ as shown in Scheme 2. In this paper, we report the $BF_3\cdot OEt_2$ -mediated intramolecular Friedel–Crafts electrophilic substitution of sulfonyl γ -alkenols ($X = Br$) by tethering a β -bromovinyl group to the aryl rings. In contrast to a hydrogen atom, the weakly deactivating bromo group initiates the stereospecific progression of the delocalized electron pair (in an *exo* → *endo* → *exo* direction), leading to the formation of a six-membered ring under metal-free catalyzed conditions. This stereospecific annulation reaction provides facile access to

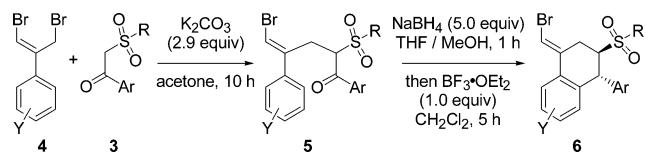
Scheme 2. Synthetic Route of 2,5-Diaryltetrahydrofurans, 1-Aryltetralins, and 1-Arylnaphthalenes



functionalized 1-aryltetralins. However, to the best of our knowledge, the bromo group-initiated intramolecular Friedel–Crafts alkylation procedure for the preparation of 1-aryltetralins has not been reported,¹¹ nor has the intramolecular annulation of bromostyryl β -ketosulfones been explored. Herein, we report our studies^{12,13} on these two processes for aryltetralin synthesis.

K_2CO_3 -mediated α -styrylation of β -ketosulfones **3a–r** with bromostyryl bromides **4a–d** (prepared by *p*-TsOH-catalyzed double bromination of styrene with 2.5 equiv of NBS in CH_2Cl_2 at reflux) in acetone at reflux provided **5a–x** in a yield range of 78–89%, as shown in Table 1. The yields of **5a–x** did not change much as a function of the structure of **3a–r** under these reaction conditions. The bromo group promoted intramolecular ring closure of **5** was examined via a two-step route: stereoselective $NaBH_4$ -mediated reduction followed by $BF_3\cdot OEt_2$ -promoted Friedel–Crafts annulation of the resulting sulfonyl 4-alkenols. For different R substituents on **5**, a diversity of electron-

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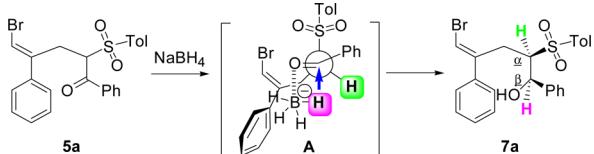
Table 1. Synthesis of 5 and 6^{a,b}

entry	3, Ar = , R = ; 4, Y-Ar =	yield ^c (%)	
		5	6
1	3a, Ph, Me; 4a, Ph	5a, 80	6a, 82
2	3b, Ph, Ph; 4a, Ph	5b, 84	6b, 80
3	3c, Ph, Tol; 4a, Ph	5c, 89	6c, 76
4	3d, 4-FC ₆ H ₄ ; Tol; 4a, Ph	5d, 83	6d, 78
5	3e, 4-MeOC ₆ H ₄ ; Tol; 4a, Ph	5e, 86	6e, 74
6	3f, 4-MeC ₆ H ₄ ; Tol; 4a, Ph	5f, 83	6f, 70
7	3g, 4-CF ₃ C ₆ H ₄ ; Tol; 4a, Ph	5g, 80	6g, - ^d
8	3h, 4-NO ₂ C ₆ H ₄ ; Tol; 4a, Ph	5h, 80	6h, - ^d
9	3i, 3-NO ₂ C ₆ H ₄ ; Tol; 4a, Ph	5i, 82	6i, - ^d
10	3j, 4-PhC ₆ H ₄ ; Tol; 4a, Ph	5j, 84	6j, 74
11	3k, 2-naphthalene; Tol; 4a, Ph	5k, 88	6k, 70
12	3l, Ph, 4-FC ₆ H ₄ ; 4a, Ph	5l, 82	6l, 78
13	3m, Ph, 4-MeOC ₆ H ₄ ; 4a, Ph	5m, 80	6m, 81
14	3n, Ph, 3-MeC ₆ H ₄ ; 4a, Ph	5n, 84	6n, 82
15	3o, Ph, 4-EtC ₆ H ₄ ; 4a, Ph	5o, 82	6o, 78
16	3p, Ph, 4-nBuC ₆ H ₄ ; 4a, Ph	5p, 83	6p, 74
17	3b, Ph, Ph; 4b, 4-FC ₆ H ₄	5q, 84	6q, 80
18	3c, Ph, Tol; 4b, 4-FC ₆ H ₄	5r, 83	6r, 80
19	3b, Ph, Ph; 4c, 4-PhC ₆ H ₄	5s, 84	6s, 78
20	3c, Ph, Tol; 4c, 4-PhC ₆ H ₄	5t, 85	6t, 76
21	3c, Ph, Tol; 4d, 2-naphthalene	5u, 82	6u, 80
22	3k, 2-naphthalene, Tol; 4d, 2-naphthalene	5v, 79	6v, 80
23	3q, 3,4-Cl ₂ C ₆ H ₃ ; Tol; 4a, Ph	5w, 78	6w, 81
24	3r, 2-thiophene, Tol; 4a, Ph	5x, 80	6x, 82

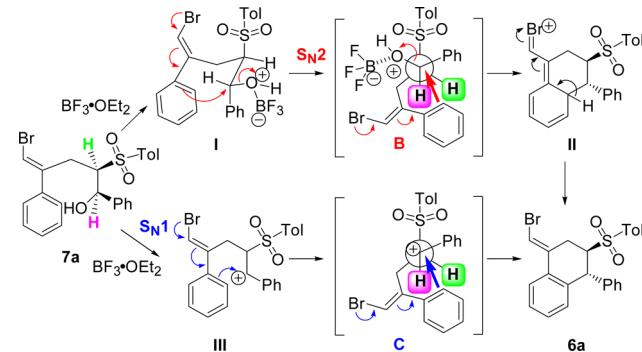
^aAlkylation reaction: 3 (1.0 mmol), K₂CO₃ (2.9 mmol), 4 (1.05 mmol), acetone (15 mL), reflux, 10 h. ^bAnnulation reaction: 5 (0.2 mmol), NaBH₄ (1.0 mmol), THF (5 mL)/MeOH (5 mL), 0 °C, 1 h, then BF₃·OEt₂ (0.2 mmol), CH₂Cl₂ (5 mL), 25 °C, 5 h. ^cIsolated yields. ^dNo reaction; 7g (89%), 7h (92%), and 7i (90%) were recovered.

withdrawing and electron-donating groups was well-tolerated. However, when the Ar group was strongly electron-withdrawing, none of the desired products 6g–i were formed, and only 7g–i were observed. Thus, in the presence of 3- or 4-nitrophenyl and 4-(trifluoromethyl)phenyl groups, no intramolecular annulation was observed. Moreover, for entries 7–9, different substituents on 5 did not affect the annulation, and the isolated yield of 6 was maintained. The structures of 5e, 6c, 6f, 6l, and 6u were all determined by single-crystal X-ray crystallography.¹⁴

In accord with our previous experience, a single skeleton derived from the sulfonyl 4-alkenol was obtained. On the basis of the Felkin–Anh model A,¹⁵ the stereochemical centers of two protons (green H–C_α and pink H–C_β in 7a) are well established. As shown in Scheme 3, a plausible mechanism

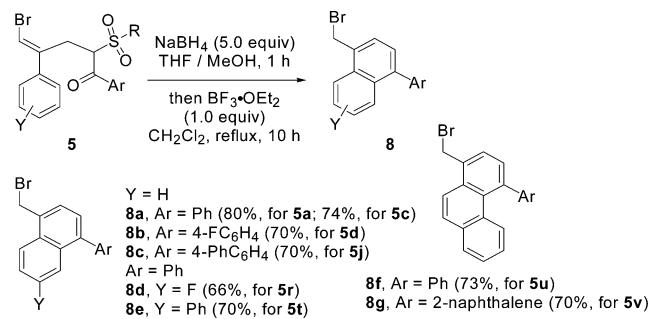
Scheme 3. Proposed Reduction Mechanism

suggests that NaBH₄-mediated carbonyl reduction of 5a is affected by the steric hindrance of the sulfonyl substituent such that the hydride can only attack the less hindered carbonyl face, yielding a single stereoisomer.^{12b} Subsequently, complexation of the resulting hydroxyl group by BF₃·OEt₂ gives I (Scheme 4).

Scheme 4. Proposed Annulation Mechanism

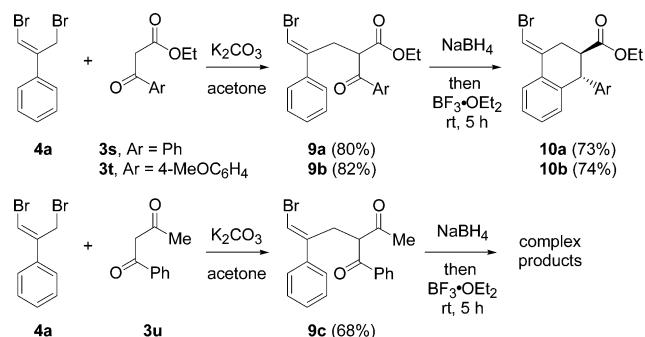
Following model B, a stereocontrolled bromo group-initiated intramolecular S_N2 annulation leads to II via a six-membered chairlike conformation having two adjacent equatorial protons. An S_N1 process can also be envisioned with III (model C), deprotonation of II affording 6a via a sequential aromatization process.

The BF₃·OEt₂-mediated intramolecular annulation of 5 was investigated at elevated temperatures and longer reaction times, as shown in Scheme 5. Specifically, changing the reaction

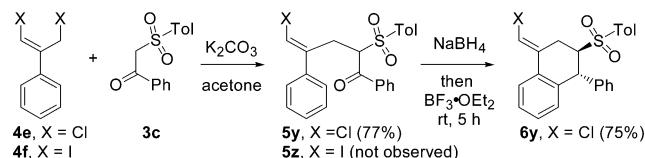
Scheme 5. Synthesis of 8

conditions from 25 °C and 5 h to CH₂Cl₂ at reflux for 10 h generated substituted 1-bromomethyl naphthalenes 8a–e (66–80%) and 1-bromomethyl phenanthrenes 8f,g (73% and 70%) via BF₃·OEt₂-promoted cascade desulfonyative aromatization of 5a,c,d,j,r,t–v. 1-Arylnaphthalene has been shown to exhibit a wide range of potential biological activities.¹⁶ The present protocol provides a novel synthetic route for the preparation of substituted naphthalenes.¹⁷ The structure of 8d was determined by single-crystal X-ray crystallography.¹⁴

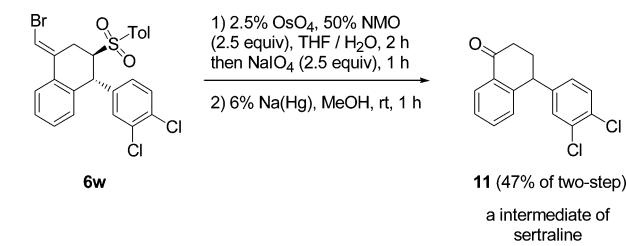
By changing the α-substituent from a sulfonyl group to an ethyl ester group (3s, Ar = Ph; 3t, Ar = 4-MeOC₆H₄), 9a and 9b were isolated in 80% and 82% yields, respectively, as shown in Scheme 6. Using the above synthetic protocol, 10a,b were generated in 73% and 74% yields. Comparing the 1,3-dicarbonyl synths, both β-ketosulfones and β-ketoesters provided similar yields of 1-aryl tetralins. However, 9c containing a β-diketone group (prepared by alkylation of 3u with 4a) failed to afford the tetralin skeleton under the standard reaction conditions.

Scheme 6. Reaction of 4a with 3s–u

In further work, the bromo group was replaced with a chloro or iodo group, affording facile access to 1-aryltetralins. As shown in Scheme 7, α -styrylation of 3c with chlorostyryl chloride 4e

Scheme 7. Reaction of 3c with 4e,f

afforded 5y in a 77% yield. The formation of 6y (75%) was achieved via intramolecular annulation of 5y using the above protocol. The structure of 5y was determined by single-crystal X-ray crystallography.¹⁴ However, when 3c was allowed to react with 4f, a complex mixture was formed and no 5z was observed. To examine the applicability of this synthetic route, a formal synthesis of sertraline (Zoloft)¹⁸ was developed (Scheme 8).

Scheme 8. Formal Synthesis of Sertraline

Sertraline is a potent selective serotonin reuptake inhibitor (SSRI), which has become a popular synthetic target. For the formal synthesis of sertraline, 6w was chosen as the starting material. Oxidative cleavage of 6w, which contains a bromovinyl group, provided the 1-tetralone skeleton via a one-pot reaction in the presence of OsO₄/NMO/NaIO₄.¹⁹ Without further purification, removal of the β -sulfonyl group using freshly prepared 6% Na(Hg) (prepared from sodium and mercury in toluene at reflux) in MeOH at 25 °C²⁰ afforded the known intermediate 11.^{18a} The overall yield of the two-step process was 47%.

In summary, we have developed a novel synthesis of various substituted 1-aryltetralins 6 and 1-arylnaphthalenes 8 via (1) K₂CO₃-mediated α -styrylation of substituted β -ketosulfones 3 with bromostyryl bromides 4 and (2) stereocontrolled NaBH₄-promoted reduction of the resulting γ -alkenones 5, followed by a BF₃·OEt₂-catalyzed intramolecular annulation of the corresponding γ -alkenols. A plausible mechanism has been proposed

for these cyclization reactions. The structures of the key products 6 and 8 have been confirmed by X-ray crystallography. Further investigations regarding the asymmetric synthesis of chiral 6 will be studied via the desymmetrical annulation of 5.

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: [10.1021/acs.orglett.6b00603](https://doi.org/10.1021/acs.orglett.6b00603).

X-ray analysis data of 5e ([CIF](#))

X-ray analysis data of 5y ([CIF](#))

X-ray analysis data of 6c ([CIF](#))

X-ray analysis data of 6f ([CIF](#))

X-ray analysis data of 6l ([CIF](#))

X-ray analysis data of 6u ([CIF](#))

X-ray analysis data of 8d ([CIF](#))

Detailed experimental procedures and spectroscopic data for all new compounds ([PDF](#))

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Notes

The authors declare no competing financial interest.

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- (14) CCDC 1445559 (**5e**), 14455598 (**5y**), 1445564 (**6c**), 14455597 (**6f**), 1445560 (**6l**), 1445561 (**6u**) and 1446957 (**8d**) contain the supplementary crystallographic data for this paper. This data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.html (or from the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK; fax: 44-1223-336033; e-mail: deposit@ccdc.cam.ac.uk).
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