

Article

Pyrrolizidine Alkaloid-Protein Adducts - Potential Non-Invasive Biomarkers of Pyrrolizidine Alkaloid-Induced Liver Toxicity and Exposure

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Chem. Res. Toxicol., **Just Accepted Manuscript** • DOI: 10.1021/acs.chemrestox.6b00120 • Publication Date (Web): 07 Jul 2016

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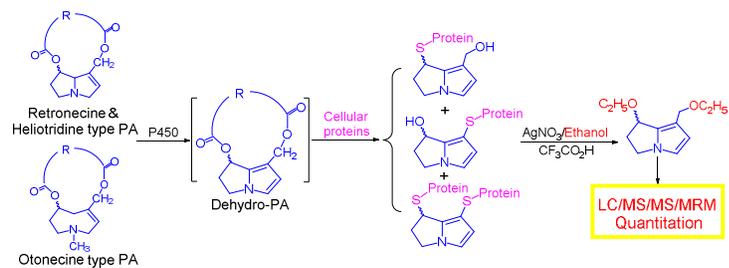
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24 Running Title: Blood Protein Adducts from Pyrrolizidine Alkaloids as a Biomarker
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ABSTRACT:

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Pyrrolizidine alkaloids (PAs) are phytochemicals present in hundreds of plant species from different families widely distributed in many geographical regions around the world. PA-containing plants are probably the most common type of poisonous plants affecting livestock, wildlife, and humans. There have been many large-scale human poisonings caused by the consumption of food contaminated with toxic PAs. PAs require metabolic activation to generate pyrrolic metabolites to exert their toxicity. In this study, we developed a novel method to quantify pyrrole-protein adducts present in the blood. This method involves the use of AgNO₃ in acidic ethanol to cleave the thiol linkage of pyrrole-protein (DHP-protein) adducts, and the resulting 7,9-di-C₂H₅O-DHP is quantified by HPLC-ES-MS/MS multiple reaction monitoring analysis in the presence of a known quantity of isotopically-labeled 7,9-di-C₂D₅O-DHP internal standard. Using this method, we determined that diester-type PAs administered to rats produced higher levels of DHP-protein adducts than other types of PAs. The results suggest that DHP-protein adducts can potentially serve as a minimally invasive biomarker of PA exposure.

Keywords: Pyrrolizidine alkaloid; DHP; DHP-protein adducts; HPLC-ES-MS/MS.

■ INTRODUCTION

Pyrrolizidine alkaloids (PAs) are common secondary phytochemical metabolites of hundreds of plant species from different families that are widely distributed in many geographical regions around the world.¹⁻¹¹ To date, more than 660 structurally different PAs and PA *N*-oxides have been identified in over 6,000 plants worldwide and about half of them are hepatotoxic.^{2, 8-10} About 3 - 5% of the flowering plants in the world contain toxic PAs.^{11,12} PA-containing plants are the most common type of poisonous plants affecting livestock, wildlife, and humans.^{2, 5, 8, 13-16} Humans are exposed to toxic PAs through the consumption of contaminated food staples, herbal medicines, herbal dietary supplements, teas, herbal teas, honey, and milk.^{1, 6, 8, 18-21} There have been many large scale human poisonings in many countries caused by the intake of food contaminated with toxic PAs.^{7, 8, 17, 20,22}

PAs and PA *N*-oxides that contain a double bond at the C1 and C2 positions of the necine base (Figure 1) are toxic. These PAs undergo metabolic activation to generate

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3 dehydropyrrolizidine alkaloids (dehydro-PAs) that react with cellular proteins and DNA, leading
4 to hepatotoxicity, genotoxicity, and tumorigenicity.^{5, 7, 8, 15, 21, 23-29} Acute poisoning causes
5 massive hepatotoxicity, including haemorrhagic necrosis, hepatomegaly, ascites, and endothelial
6 proliferation.^{7, 8, 19} Further liver damage can lead to occlusion of hepatic veins, resulting in veno-
7 occlusion disease, also called hepatic sinusoidal obstruction syndrome, which is a characteristic
8 histological sign of PA-induced poisoning.^{1, 7, 8, 19} PA-induced DNA damage includes DNA
9 strand breakage, unscheduled DNA synthesis, DNA-DNA cross-linking, DNA-protein cross-
10 linking, and DNA adduct formation.^{2, 5, 7, 8, 15, 24-26, 28, 30-42}

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12 We recently reported that riddelliine, a representative tumorigenic PA, induces liver tumors
13 through a genotoxic mechanism mediated by covalent binding of the pyrrolic metabolite,
14 dehydroriddelliine, to cellular DNA to form a set of 6,7-dihydro-7-hydroxy-1-hydroxymethyl-
15 5H-pyrrolizine (DHP)-derived DNA adducts.^{2, 15, 16} The DHP-DNA adduct levels correlated
16 closely with the riddelliine-induced tumorigenic potencies in the rats fed with riddelliine; also,
17 the metabolic pattern and DNA adduct profiles are highly relevant to humans.^{2, 5, 39} Our further
18 mechanistic studies demonstrated that different types of hepatotumorigenic PAs generated the
19 same set of DHP-derived DNA adducts (designated as DHP-dG-3, DHP-dG-4, DHP-dA-3, and
20 DHP-dA-4), but these DNA adducts were not formed from non-hepatotumorigenic PAs (Figure
21 1).¹⁵ These results indicate that this set of DNA adducts can serve as a common biological
22 biomarker of PA tumorigenicity and exposure.

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24 Pyrrole-protein (DHP-protein) adducts have long been considered to be responsible for PA-
25 induced hepatotoxicity, and can be used as biomarkers of toxic PAs exposure.^{1, 8} Thus, an
26 important goal has been to develop accurate and sensitive analytical methods to detect and
27 quantify DHP-protein adducts *in vivo*. In the early 1990's, Mattocks and co-workers developed
28 an analytical method to detect DHP-protein adducts in the blood and liver of rodents, as well as
29 in livestock.⁴³⁻⁴⁶ Briefly, blood or liver samples were reacted with silver nitrate in acidic ethanol
30 to cleave the thiol linkage of DHP-protein adducts at the C7 and C9 positions, leading to the
31 protein moiety being replaced by ethoxyl groups. The resulting 7,9-di-C₂H₅O-DHP was then
32 identified by GC/MS measurements or by a colorimetric test by using Ehrlich's reagent, 4-
33 dimethylaminobenzaldehyde (DABA), to form a 7,9-di-C₂H₅O-DHP-DABA product that has an
34 intense yellow color.⁴⁷ In 2011, Lin and co-workers used a similar approach, coupled with
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3 LC/MS/MS analysis, to detect DHP-protein adducts in patients who suffered from hepatic
4 sinusoidal obstruction syndrome due to the consumption of the PA-containing herbal plant
5 *Gynura segetum*.⁴⁸ This represented the first time that DHP-protein adducts had been detected in
6 human blood samples. This method was further modified, by using 7,9-diglutathionyl-DHP (7,9-
7 di-GS-DHP) to construct a calibration curve to quantify the 7,9-di-C₂H₅O-DHP-DABA derived
8 from DHP-protein adducts in the blood of the 15 liver-injured patients who ingested *Gynura*
9 *segetum*.⁴⁹

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12 In the present study, we report the development of a method to quantify DHP-protein adducts
13 formed *in vivo*. This method involves: (i) following the experimental procedure of Mattocks and
14 Jukes,⁴⁵ treatment of blood samples with silver nitrate in acidic ethanol to convert DHP-protein
15 adducts into 7,9-di-C₂H₅O-DHP; and (ii) utilizing a synthetically prepared isotope-labeled 7,9-
16 di-C₂D₅O-DHP as an internal standard, with quantification being conducted by LC/MS/MS
17 multiple reaction monitoring (MRM) analysis. The method was then applied to quantify the
18 DHP-protein adducts formed *in vivo* in two separate animal experiments: (i) female rats orally
19 gavaged with different concentrations of riddelliine for 5 or 30 consecutive days; and (ii) a series
20 of PAs (Figure 1)¹⁵ orally gavaged to female rats for 3 consecutive days. This quantitation study
21 indicates that diester-type PAs (Figure 1) produced higher levels of DHP-protein adducts than
22 other types of PAs. These results indicate that the developed HPLC-ES-MS/MS method is
23 accurate and precise for the detection and quantitation of DHP-protein adducts in the blood and
24 suggest that DHP-protein adducts can potentially serve as minimally invasive biomarkers of PA
25 exposure.

26 ■ EXPERIMENTAL PROCEDURES

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28 **Caution:** *Riddelliine, retrorsine, monocrotaline, lycopsamine, retronecine, lasiocarpine,*
29 *clivorine, senkirkine, heliotrine, and riddelliine N-oxide are carcinogenic in laboratory animals.*
30 *They should be handled with extreme care, using proper personal protective equipment and a*
31 *well-ventilated hood.*

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34 **Chemicals.** Monocrotaline, silver nitrate, trifluoroacetic acid, formic acid, sodium
35 bicarbonate, diethyl ether, heparin, deuterated ethanol (99.5%), Protein Quantification Kit, and

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3 Hemoglobin Assay Kit were purchased from the Sigma-Aldrich (St. Louis, MO). Absolute
4 ethanol (200 proof) was purchased from AAPER Alcohol and Chemical Co. (Shelbyville, KY).
5 Riddelliine was obtained from Dr. Po-Chan, National Toxicology Program, National Institute of
6 Environmental Health Sciences (Research Triangle Park, NC). Dehydromonocrotaline was
7 synthesized from the reduction of monocrotaline by *o*-chloroanil in chloroform as described
8 previously.⁵⁰ All chemicals used for the animal studies were analyzed by HPLC and found to be
9 >97% pure. All solvents used were HPLC grade.

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18 **Synthesis of 7,9-Di-C₂H₅O-DHP.** 7,9-Di-C₂H₅O-DHP was synthesized following the
19 method of Mattocks and Jukes⁴³ with modification. Briefly, absolute C₂H₅OH (20 mL) was
20 added to dehydromonocrotaline (200 mg, 0.62 mmol), and the reaction mixture was stirred for
21 24 hr at ambient temperature. To the reaction mixture was added 10 mL 8% sodium bicarbonate
22 solution, the ethanol was removed under reduced pressure, and the remaining material was
23 extracted with diethyl ether (20 mL) three times. The ether extracts were combined and the
24 solvent was removed under reduced pressure, resulting in practically pure 7,9-di-C₂H₅O-DHP
25 (85 mg; 65.7% yield).

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32 The 7,9-di-C₂H₅O-DHP was further purified by reversed phase HPLC (a Waters HPLC
33 system consisting of a Model 600 controller, a Model 996 photodiode array detector, and a 600
34 pump) (Milford, MA), using a Phenomenex Luna C18 (2) column (250 x 4.6 mm) (Phenomenex,
35 Inc. CA) by eluting with acetonitrile at 800 μL/min and monitoring at 220 nm (Figure S1 in the
36 Supporting Information). The product, which eluted at 4.3 min, was collected and identified as
37 7,9-di-C₂H₅O-DHP by MS and ¹H NMR spectral analyses.

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43 The ¹H NMR spectral assignments of 7,9-di-C₂H₅O-DHP in acetone-d₆ are: δ 1.15 (3H, t,
44 CH₃, J_{CH₂-CH₃} = 7 Hz), 1.16 (3H, t, CH₃, J_{CH₂-CH₃} = 7 Hz), 2.46 (1H, m, H6b), 2.71 (1H, m, H6a),
45 3.45 (4H, m, CH₂), 3.62 (4H, m, CH₂), 3.93 (1H, m, H5b), 4.07 (1H, m, H5a), 4.35 (1H, d, H9b,
46 J_{9a,9b} = 11.4 Hz), 4.41 (1H, d, H9a, J_{9a,9b} = 11.4 Hz), 4.83 (1H, dd, H7, J_{6,7} = 6.1 and J_{5,7} = 1.6
47 Hz), 6.12 (H, d, H2, J_{2,3} = 2.7 Hz), and 6.63 (1H, d, H3, J_{2,3} = 2.7 Hz).

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52 The 7,9-di-C₂H₅O-DHP was stable in petroleum ether, diethyl ether, acetone, acetonitrile,
53 and ethanol, but unstable in dichloromethane and chloroform.

Synthesis of Isotopically Labeled 7,9-Di-C₂D₅O-DHP Adduct. Isotopically labeled 7,9-di-C₂D₅O-DHP was similarly synthesized by the reaction of dehydromonocrotaline (20 mg) with 2 mL of C₂D₅OH. After the reaction, the 7,9-di-C₂D₅O-DHP was purified by reversed phase HPLC using the conditions described above. Its structure was determined based on analysis of its HPLC retention time, UV-visible spectrum, mass, and ¹H NMR spectra. Its ¹H NMR spectrum is shown in Figure S2 in the Supporting Information.

The ¹H NMR spectral assignments of 7,9-di-C₂D₅O-DHP in acetone-d₆ are: δ 2.46 (1H, m, H6b), 2.71 (1H, m, H6a), 3.93 (1H, m, H5b), 4.07 (1H, m, H5a), 4.35 (1H, d, H9b, J_{9a,9b} = 11.4 Hz), 4.41 (1H, d, H9a, J_{9a,9b} = 11.4 Hz), 4.83 (1H, dd, H7, J_{6,7} = 6.1 and J_{5,7} = 1.6 Hz), 6.12 (H, d, H2, J_{2,3} = 2.7 Hz), and 6.63 (1H, d, H3, J_{2,3} = 2.7 Hz).

Treatment of Rats with Pyrrolizidine Alkaloids. Procedures involving care and handling of rats were reviewed and approved by the National Center for Toxicological Research (NCTR) Laboratory Animal Care and Use Committee. Female F344 rats (8-10 weeks of age) were obtained from the NCTR breeding colony and maintained on a 12-hr light-dark cycle. There were two separate sets of animal studies.

1. Treatment of rats with different doses of riddelliine - Female rats (4 per group) were orally gavaged for 5 or 30 consecutive days at daily doses of 0, 0.1, 1, 2, or 5 mg riddelliine/kg body weight (bw) in 2.5 mL of 0.1 M sodium phosphate (pH 7.4). Twenty-four hr after the last dose, the rats were sacrificed by exposure to CO₂, followed by exsanguination. The livers were removed, rinsed with cold saline, and stored at -70 °C before isolation of DNA for DNA adducts analyses. Blood samples were collected by cardiac puncture into heparin treated tubes. After centrifugation, each sample was separated into two fractions, the supernatant (plasma fraction) and precipitated blood cells, the latter of which contained mainly erythrocytes and was designated the hemoglobin fraction. All blood samples were stored at -70 °C before protein adducts analyses.
2. Treatment of rats with eleven individual pyrrolizidine alkaloids and derivatives – As reported previously,¹⁵ riddelliine, retrorsine, monocrotaline, lycopsamine, retronecine, riddelliine *N*-oxide, lasiocarpine, heliotrine, senkirkine, clivorine, and platyphylline (a mixture of 70% platyphylline and 30% neoplatyphylline) were orally gavaged to female rats

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3 (4 per group) for 3 consecutive days, at daily doses of 24 (high dose) and 4.5 (low dose)
4 $\mu\text{mol/kg}$ bw in 500 μL of 10% DMSO in water. Rats for vehicle control group were treated
5 with 500 μL of 10% DMSO in water. Twenty-four hr after the last dose, the rats were
6 sacrificed by exposure to CO_2 followed by exsanguination.¹⁵ Liver and blood samples were
7 collected and stored as described above. The DNA adducts in the liver of these rats were
8 identified and quantified by LC/MS/MS analysis; these results were published previously.¹⁵
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16 **Reaction of DHP-Protein Adducts in Blood of Rats with Silver Nitrate in**

17 **Alcohol.** The reaction of silver nitrate in alcohol with DHP-protein adducts contained in the
18 blood samples was assessed by using many different experimental conditions, including different
19 pHs, solvents (ethanol and methanol), with or without an acid (trifluoroacetic acid,
20 trichloroacetic acid, formic acid), reaction time, and isolation methods. Plasma and hemoglobin
21 fractions of blood samples were tested and the results were compared. The optimal solvent
22 system for the reaction was determined to be ethanol in the presence of trifluoroacetic acid. The
23 experimental procedure that provided the optimal reaction yield is described below.
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30 Each blood sample (100 μL of plasma or hemoglobin) was mixed with 500 μL acetone,
31 vortexed, and centrifuged at 1000 g for 5 min. The supernatant was removed and the residue
32 was washed with 500 μL of absolute ethanol and then 100 μL of a freshly prepared 2% silver
33 nitrate/ethanol solution containing 5% of trifluoroacetic acid was added. After the mixture was
34 shaken for 1 hr at room temperature, the mixture was adjusted to $\sim\text{pH}$ 7 with 60 μL of 25%
35 K_2CO_3 , and then 100 μL of the supernatant was transferred for LC-MS/MS analysis. To each
36 100 μL sample was added 10 μL internal quantitation standard (200 ng/mL 7,9-di- $\text{C}_2\text{D}_5\text{O}$ -DHP).
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43 For comparison, the yields of 7,9-di- $\text{C}_2\text{H}_5\text{O}$ -DHP produced from several other experimental
44 conditions are shown below:
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- 46 (i) When methanol was used in place of absolute ethanol, the yield of 7,9-di- CH_3O -DHP
47 was about 90% of 7,9-di- $\text{C}_2\text{H}_5\text{O}$ -DHP obtained with absolute ethanol (both reactions
48 were conducted in the presence of trifluoroacetic acid).
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50 (ii) When 5% trifluoroacetic acid was replaced by formic acid or trichloroacetic acid, the
51 yields of 7,9-di- $\text{C}_2\text{H}_5\text{O}$ -DHP were about 95 and 80%, respectively, of that obtained with
52 trifluoroacetic acid.
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3 (iii) When an acid was not added to the silver nitrate/ethanol solution, the yield of 7,9-di-
4 C₂H₅O-DHP was about 10% of that obtained in the presence of acid, accompanied by the
5 production of 7-C₂H₅O-DHP as the major product. These results are consistent with the
6 findings previously described by Mattocks et al.⁵¹ and by Lame et al.⁵²
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11 **Detection and Quantitation of 7,9-Di-C₂H₅O-DHP Adducts in Rat Blood**

12 **Samples by LC-MS/MS.** Bioanalytical methods were developed to determine the
13 concentration of 7,9-di-C₂H₅O-DHP in the blood samples using atmospheric pressure ionization
14 LC-MS/MS in positive electrospray ionization mode. The liquid chromatography system
15 consisted of a Shimadzu Prominence HPLC system, including a CBM-20A controller, two LC-
16 20AD pumps, a SIL-20AC HT autosampler, a SPD-20A UV/VIS detector (Shimadzu Scientific
17 Instrument, Columbia, MD), and an automated switching valve (TPMV, Rheodyne, Cotati, CA).
18 The switching valve was used to divert the column effluent to either a waste collector or to the
19 MS instrument. A Shimadzu Prominence HPLC system was used for sample injection and
20 separation. Each sample (10 μL) was loaded onto a reverse phase column (ACE AQ 5 μm, C18,
21 50 x 2.0 mm, MAC-MOD Analytical, Chadds Ford, PA) with acidic aqueous buffer/acetonitrile
22 gradient at 500 μL/min, and the sample components were eluted into the mass spectrometer. The
23 column chamber temperature was set to 40 °C and the autosampler temperature was set at 5 °C.
24 The mobile phases were 2 mM ammonium acetate in water containing 0.1% formic acid and
25 acetonitrile. The initial gradient consisted of 15% acetonitrile for 0.3 min followed by a linear
26 gradient up to 95% acetonitrile over 2 min. After holding 95% acetonitrile for 1 min, the
27 instrument was reset to the initial conditions in 0.1 min. The analytical column was equilibrated
28 with the mobile phase for 1.2 min. The total run time for a sample analysis was 4.6 min.
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46 The HPLC eluate was coupled to an AB Sciex 4000 QTrap LC/MS/MS system (AB Sciex,
47 Foster City, CA), equipped with a Turbo V™ ion source and a desolvation temperature of
48 450 °C. Nitrogen was used as curtain gas, nebulizer gas, heater gas, and collision gas. The
49 samples were acquired in positive electrospray ionization mode using multiple reaction
50 monitoring methods (MRM). The MRM method for detection and quantitation of 7,9-di-C₂H₅O-
51 DHP adducts in the samples used a combination of two ion transitions: (i) m/z 227 → m/z 164,
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$[M+NH_4]^+ \rightarrow [M+H-NH_3-C_2H_5OH]^+$; and (ii) m/z 210 \rightarrow m/z 164, $[M+H]^+ \rightarrow [M+H-C_2H_5OH]^+$, at collision energies of 27 and 15 eV, respectively (Figure 2). The internal isotope-labeled standard was 7,9-di- C_2D_5O -DHP, the ion transition of the MRM method was m/z 237 \rightarrow m/z 169 ($[M+NH_4]^+ \rightarrow [M+H-NH_3-C_2D_5OH]^+$), and the collision energy was 12 eV. The declustering potential was 25 V.

7,9-Di- C_2H_5O -DHP and the internal standard, 7,9-di- C_2D_5O -DHP, were quantified by using a reference standard calibration curve. One set of standard curve and quality control samples was run before the blood and plasma sample analysis; quality control samples were run each day again at the end of sample analysis period. The lower limit of quantification (LLOQ) was 1 ng/mL for 7,9-di- C_2H_5O -DHP in the plasma and hemoglobin fractions. A linear calibration curve ranging 1-2000 ng/mL for 7,9-di- C_2H_5O -DHP was fitted to 1/x weighted linear regression model (Figure S3 in the Supporting Information). Accuracy and precision of the standard curve and quality control samples were all in the acceptable range for bioanalytical analysis. The coefficient (r^2) of determination for the standard curve of 7,9-di- C_2D_5O -DHP was > 0.99 .

Validation of the LC-MS/MS Quantitation Method. To ensure the reliability of the LC-MS/MS method, the LC-MS/MS method for quantification of 7,9-di- C_2H_5O -DHP was partially validated. The selectivity of the method, calibration standard curve (including inter-day calibration curve) linearity, concentration dynamic range, accuracy, and precision were evaluated. No matrix interference of the analyte peak was observed at LLOQ. The LLOQ was 1 ng/mL. A linear calibration curve ranging from 1 ng/mL to 2000 ng/mL was fitted to 1/x weighted linear regression model. Inter-day standard calibration curves of 7,9-di- C_2H_5O -DHP are presented in Figure S4. The accuracy, precision, and the standard calibration curve parameters of the LC-MS/MS methods for 7,9-di- C_2H_5O -DHP are shown in Table S1. Accuracy and precision of the standards used for standard curve generation are summarized in Table S2. The coefficient (r^2) of determination for all standard curves was ≥ 0.995 (Table S1). The results of quality control samples of 7,9-di- C_2H_5O -DHP for the standard calibration curves are shown in Table S3. Quality control (QC) samples (low, mid, and high concentrations at 2.5, 50, and 1600 ng/mL)

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3 and the accuracy of the individual QC samples ranged from 85-115% of the nominal
4 concentration. The precisions were within a CV% value of $\leq 8.1\%$.
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7 Incurred rat blood samples were also analyzed during the LC-MS/MS method validation.
8 Four aliquots of the pooled rat blood after reaction with silver nitrate in alcohol were quantified
9 for 7,9-di-C₂H₅O-DHP with each standard calibration curve. The summary result of the 7,9-di-
10 C₂H₅O-DHP concentrations in the pooled blood samples is showed in Table S4, with the relative
11 standard derivation (RSD) of 6.89, 7.59, and 14.2%, respectively, indicating the method with
12 good reproducibility.
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19 **Statistical Analyses.** Statistical analyses were performed by analysis of variance, with
20 comparisons between groups being conducted by the Holm-Sidak method. When necessary, the
21 data were log transformed to maintain an equal variance or normal data distribution. Dose-
22 related trends were assessed linear regression analysis. Correlations between DHP-protein levels
23 and DHP-DNA adduct levels were assessed by Pearson product moment analysis.
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30 RESULTS

31 **Synthesis of 7,9-Di-C₂H₅O-DHP and Isotopically Labeled 7,9-Di-C₂D₅O-DHP.**

32 In our developed quantitation method, the reaction of silver nitrate in acidic ethanol with DHP-
33 protein adducts in the plasma and hemoglobin fractions of blood from rats produced 7,9-di-
34 C₂H₅O-DHP, which was identified and quantified by LC/MS/MS MRM analysis (Figure 3). For
35 this purpose, 7,9-di-C₂H₅O-DHP was synthesized for structural identification of the product and
36 to construct a calibration curve. Isotopically labeled 7,9-di-C₂D₅O-DHP was synthesized for use
37 as a quantitation standard in the LC/MS/MS MRM analysis.
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46 7,9-Di-C₂H₅O-DHP was previously synthesized by Mattocks and Jukes,⁴³ but the reaction
47 yield and NMR spectral data were not provided. In our study, 7,9-di-C₂H₅O-DHP was
48 synthesized in a 65.7% yield and its structure was fully characterized by mass and ¹H NMR
49 spectral analysis. Isotopically labeled 7,9-di-C₂D₅O-DHP was similarly synthesized by reaction
50 of dehydromonocrotaline with deuterated ethanol. As described in the Experimental Procedures
51 section, its structure was determined by mass spectral data and ¹H NMR spectroscopic analysis.
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LC-ES-MS/MS Analysis of DHP-Protein Adducts Formed in the Blood of Rats Dosed Riddelliine and Other Pyrrolizidine Alkaloids.

Blood Samples of Rats Dosed Riddelliine. Female rats (4 per group) were dosed by oral gavage at 0, 0.1, 1.0, 2.0, or 5.0 mg riddelliine/kg bw for 5 or 30 days. Blood samples were collected and separated into plasma and hemoglobin fractions. The levels of DHP-protein adducts, identified as 7,9-di-C₂H₅O-DHP in the plasma and hemoglobin fractions, were quantified by LC-ES-MS/MS MRM analysis.

LC/MS/MS MRM quantitation of 7,9-di-C₂H₅O-DHP formed from DHP-protein adducts contained in the hemoglobin fraction of rats dosed with 5 mg riddelliine/kg bw for 30 consecutive days is shown in Figure 4. Panel A is MRM total ions of the 7,9-di-C₂H₅O-DHP. Panels B and C are the MRM ion transitions for m/z 210 \rightarrow 164 and m/z 227 \rightarrow 164, respectively. Panel D is the MRM ion transition for m/z 237 \rightarrow 169 of the isotopically-labeled standard 7,9-di-C₂D₅O-DHP.

LC/MS/MS MRM chromatograms of 7,9-di-C₂H₅O-DHP formed from DHP-protein adducts contained in the hemoglobin fractions of rats dosed with: (A) 0 mg; (B) 0.1 mg; (C) 1.0 mg; (D) 2.0 mg; and (E) 5 mg riddelliine/kg bw for 30 consecutive days are shown in Figure 5. The left panels represent the level of DHP-protein adducts (measured as 7,9-di-C₂H₅O-DHP) from control rats or riddelliine dosed rats. The right panels are the signals for 50 fmol 7,9-di-C₂D₅O-DHP internal standard added to each sample. In the MRM method for quantitation of 7,9-di-C₂H₅O-DHP in the samples, the two ion transitions were monitored: (i) m/z 227 \rightarrow 164 ($[M+H]^+ \rightarrow [M+H-C_2H_5OH]^+$) and (ii) m/z 210 \rightarrow 164 ($[M+NH_4]^+ \rightarrow [M+H-NH_3-C_2H_5OH]^+$) at collision energy 27 and 15 eV, respectively. The combined peak area of the selected ion current peaks of these two ion transitions was used in the quantitation.

There was a significant ($P < 0.05$) dose- and time-dependent increase in the levels of DHP-protein adducts with both plasma and hemoglobin, with the extent of protein binding being significant at all doses of riddelliine (Figure 6 and Table S5 in the Supporting Information). A comparison of the plasma and hemoglobin binding indicated a more dynamic response with

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3 hemoglobin due, in part, to the lower background levels obtained with the control hemoglobin
4 samples. Based upon the more dynamic response observed with hemoglobin, subsequent work
5 with a series of PAs focused on DHP-hemoglobin adducts.
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10 **Blood Samples of Rats Dosed Different PAs.** Female rats (4 per group) were dosed by oral
11 gavage daily with riddelliine, retrorsine, monocrotaline, lycopsamine, retronecine, riddelliine *N*-
12 oxide, lasiocarpine, heliotrine, senkirkine, clivorine, or platyphylliine (a mixture of 70%
13 platyphylliine and 30% neoplatyphylliine) at 4.5 or 24 $\mu\text{mol/kg}$ bw in 0.5 mL of 10% DMSO in
14 water or vehicle for three consecutive days. One day after the last treatment, blood samples were
15 collected and separated into plasma and hemoglobin fractions. Using the analytical method
16 described previously, each hemoglobin sample was converted into 7,9-di- $\text{C}_2\text{H}_5\text{O}$ -DHP adducts
17 that were identified and quantified by LC-MS/MS MRM analysis.
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25 The results of the 7,9-di- $\text{C}_2\text{H}_5\text{O}$ -DHP adducts formed from the hemoglobin fractions,
26 expressed as ng 7,9-di- $\text{C}_2\text{H}_5\text{O}$ -DHP/g hemoglobin/mg PA/kg bw, are summarized in Figure 7
27 and Table S6 in the Supporting Information. The lowest level of binding was detected
28 platyphylliine, a non-tumorigenic PA that lacks a double bond in the nicotine ring. This was
29 followed by retronecine, heliotrine, and lycopsamine, PAs that contain a hydroxyl substituent at
30 C7. The next highest hemoglobin binding was observed with clivorine and senkirkine, PAs that
31 contain an *N*-methyl substituted nicotine ring. The greatest extent of DHP hemoglobin adduct
32 formation was obtained with riddelliine *N*-oxide, retrorsine, lasiocarpine, riddelliine, and
33 monocrotaline, PAs that are cyclic diesters.
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42 **Pearson Product Moment Correction Coefficients and P Values.** With both the
43 PA DHP-DNA adducts reported previously¹⁵ and the PA DHP-hemoglobin adducts available, we
44 determined the correlation between (i) the PA DHP-DNA adducts and (ii) the PA DHP-DNA
45 adducts and PA DHP-hemoglobin adducts. As summarized in Table 1, there is an excellent
46 correlation between any two of the PA DHP-DNA adducts (*i.e.*, among the DHP-dG-3, DHP-
47 dG-4, DHP-dA-3, and DHP-dA-4 adduct), with correlation coefficients of 0.951 - 0.998 and P
48 values of 1×10^{-5} - 9×10^{-12} (Table 1). A representative correlation plot is shown in Figure 8. A
49 more modest correlation was observed between PA DHP-hemoglobin adducts and PA DHP-
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3 DNA adducts, with correlation coefficients of 0.573 - 0.702 and P values of 0.07 - 0.02 (Table 1
4 and Figure 8).
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10 DISCUSSION

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12 Mattocks and co-workers previously developed a method to detect pyrrole-protein (DHP-
13 protein) adducts present in the blood and liver of rodents administered toxic PAs.^{44, 45} This
14 method specifically detects pyrrole-protein adducts that are formed through binding of the
15 proteins' thiol group at the C7 and/or C9-position of the necine base by using the AgNO₃ in
16 acidic ethanol to cleave the thioether group (at the C7 and/or C9-position). The resulting 7,9-di-
17 C₂H₅O-DHP product was then separated by TLC and identified by GC/MS; however, the 7,9-di-
18 C₂H₅O-DHP product was not quantified. In our study, based on this method, we developed a
19 novel approach to quantify the DHP-protein adducts formed in the blood of rats administered
20 PAs. In the developed method, blood samples were treated with AgNO₃ in acidic ethanol, and
21 the resulting 7,9-di-C₂H₅O-DHP was identified and quantified by HPLC-ES-MS/MS MRM
22 analysis in the presence of a known quantity of an isotopically-labeled 7,9-di-C₂D₅O-DHP
23 internal standard. This HPLC-ES-MS/MS method was accurate and precise, with the lower limit
24 of quantification (LLOQ) at 1 ng/mL for 7,9-di-C₂H₅O-DHP. The coefficient (r²) of
25 determination for the standard curve of 7,9-di-C₂H₅O-DHP ranging 1 - 2000 ng/mL was over
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39 We determined that when the reaction of DHP-protein adducts in blood of rats with silver
40 nitrate in alcohol was conducted without the addition of an acid, the yield of 7,9-di-C₂H₅O-DHP
41 was about 10% of that obtained in the presence of acid and this was accompanied by the
42 production of 7-C₂H₅O-DHP as the major product. This result indicates that under experimental
43 conditions, the DHP-protein adducts isolated as 7-C₂H₅O-DHP are the predominant DHP-protein
44 adducts, while DHP-di-protein adducts (protein-protein cross-link) are formed no more than
45 10%. PA-induced protein-protein cross-linking and DNA-protein cross-linking have been
46 suggested to cause animal toxicity and may lead to tumor formation.^{25,26,31,32} Further
47 investigations are clearly necessary to determine the role of protein-protein cross links, DNA-
48 protein cross links, and DHP-protein adducts in PA-induced toxicity and tumorigenicity.
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With this developed method, we were able to detect and quantify pyrrole-protein (DHP-protein) adducts formed in the blood of rats dosed by gavage with riddelliine of different concentrations for 5 and 30 consecutive days (Figure 6 and Table S1 in the Supporting Information) and with a series of different PAs for 3 consecutive days (Figure 7 and Table S2 in the Supporting Information). The results indicate that the levels of DHP-protein adducts formed in the hemoglobin and plasma fractions are time dependent (Figure 6). Their formation was also dose dependent; with the DHP-protein adducts formation in the hemoglobin fraction having a more dynamic dose response than that in the plasma fraction. This suggests that for the quantitation of the DHP-protein adduct formation, the use of DHP-hemoglobin fraction is a better choice.

Drug-induced liver injury (DILI) is an important disease in humans. In the United States, it is one of the leading causes of acute liver failure.⁵³ PA-induced liver injury has been recognized for a long time and it may be one of the most common causes for drug-induced liver injury.^{48, 54, 55} In a clinical study, Ruan et al.⁵⁶ tested 31 DILI patients with unknown etiology and found DHP-protein adducts in 12 (39%) of the patients. It is noteworthy that the LC/MS/MS method reported in our study can potentially be utilized as a mechanism-based clinical bioassay to quantify DHP-protein adducts present in human blood samples, and consequently DHP-protein adducts can potentially serve as minimally invasive biomarkers of PA exposure in humans.

In the present study we determined that: (i) the most extensive DHP-hemoglobin adduct formation occurred with cyclic diester PAs, including retrorsine, lasiocarpine, riddelliine, monocrotaline, and riddelliine *N*-oxide; (ii) the next most extensive DHP-hemoglobin adduct formation occurred with clivorine and senkirkine, otonecine-type PAs containing an *N*-methyl substituted nicotine ring; (iii) this was followed by retronecine, heliotrine, and lycopsamine, PAs that contain a hydroxyl group at C7 of the necine base; and (iv) the lowest extent of binding was with platyphylline, a non-tumorigenic PA that lacks a double bond in the nicotine ring (Figure 7). We have previously used the livers of these rats to quantify the DHP-DNA adduct formation and found that the levels of DHP-DNA adducts (DHP-dG-3, DHP-dG-4, DHP-dA-3, and DHP-dA-4) are in the order: retrorsine > lasiocarpine > riddelliine ~ monocrotaline > riddelliine *N*-oxide > senkirkine > heliotrine ≥ clivorine >>> lycopsamine ~ retronecine ~ platyphyllin ~ vehicle control.¹⁵ Although the Pearson product moment correlation coefficients between DHP-protein adducts and DHP-DNA adducts were moderate (Table 1 and Figure 8), when the comparison is

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3 based on the structural features of the parent PAs, the same general relationship holds for both
4 DHP-DNA adducts in the liver and DHP-hemoglobin adducts in the blood with cyclic diester
5 PAs > otonicine-type PAs > monoester PAs (and retronecine) > PAs lacking a double bond in
6 the nicotine ring.
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10 The Edman rearrangement reaction is a commonly used method for quantifying metabolite-
11 bound protein adducts *in vivo*.^{57, 58} This method uses the phenyl isothiocyanate (PITC) or its
12 derivatives to bind to the adducted terminal valine group of hemoglobin (in our case, e.g. DHP-
13 hemoglobin adducts), resulting in the replacement of the hemoglobin moiety by the
14 phenylthiohydantoin group followed by structural identification and quantitation. We have
15 attempted to use this method to quantify DHP-protein adducts, by measuring the resulting
16 phenylthiohydantoin-DHP adducts by LC/MS/MS.²⁹ We found that reaction of PITC with
17 DHP-hemoglobin adducts that were prepared from the reaction of rat hemoglobin with
18 dehydromonocrotaline generated four isomeric phenylthiohydantoin (DHP-valine-PITC)
19 products.²⁹ Quantitation of four products by LC/MS/MS would be less favorable (and probably
20 less accurate) than the quantitation of a single product (7,9-di-C₂H₅O-DHP). In addition, we
21 were unable to find a practical method to prepare synthetically the four deuterium-labeled DHP-
22 valine-PITC adducts for use as quantitation standards for LC/MS/MS MRM analysis. Mattocks
23 and Jukes^{44, 45} have reported that reactive pyrrolic metabolites, dehydro-PAs, preferentially bind
24 to proteins at thiols rather than nitrogens, which suggests that DHP-protein adducts results from
25 reactions with thiols should be formed in a higher yield than those from reactions with nitrogens.
26 Consequently, we conclude that PA-protein binding measurements based upon the detection of
27 7,9-di-C₂H₅O-DHP have distinct advantages over the detection of DHP-valine adducts.
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42 Our study focused on characterizing DHP-hemoglobin adducts in the blood of rats treated
43 with PAs. Rat hemoglobin has ten free thiol groups (from the cysteinyl residues), six from the
44 two α -chains and four from the two β -chains.⁵⁹ In contrast, human hemoglobin has six free thiol
45 groups, one from each of the two α -chains and two from each of the two β -chains.⁵⁹ This
46 suggests that at equal levels of exposure, more DHP-hemoglobin adducts may be detected from
47 rats as compared to humans. Furthermore, the binding capability and the stereochemical activity
48 of the thiol groups may differ between human and rat hemoglobin,⁵⁹⁻⁶² which could also affect
49 the extent of adduct formation from PAs between rats and humans. Additional experiments are
50 clearly necessary to demonstrate the utility of our LC/MS/MS with human hemoglobin samples;
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3 nonetheless, as noted above, DHP-protein adducts have been detected in the blood of patients
4 consuming the herbal plant *Gynura segetum*.⁴⁸
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8 9 **SUPPORTING INFORMATION**

10 Tables S1 and S2. LC-MS/MS analysis calibration curve parameters, curve linearity, accuracy,
11 and precision data of 7, 9 di-C₂H₅O-DHP. Table S3. LC-MS/MS analysis quality control results
12 for 7, 9 di-C₂H₅O-DHP standard calibration curve from inter-day. Table S4. Summary of 7, 9 di-
13 C₂H₅O-DHP concentration in pooled blood sample containing pyrrolizidine alkaloids. Tables S5
14 and S6. The levels of DHP-protein adducts (ng/g protein) identified as 7,9-di-C₂H₅O-DHP in
15 hemoglobin and plasma fractions of female rats. Figure S1. HPLC profile of 7,9-di-C₂H₅O-DHP
16 prepared from reaction of dehydromonocrotaline with ethanol. Figure S2. ¹H NMR spectra of
17 7,9-di-C₂H₅O-DHP and 7,9-di-C₂D₅O-DHP. Figures S3 and S4. Standard calibration curve of
18 7,9-di-C₂H₅O-DHP. This material is available free of charge via the Internet at
19 <http://pubs.acs.org>.
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31 **Funding**

32 This research was supported, in part, by the appointment (Y.Z.) to the Postgraduate Research
33 Program at the NCTR administered by the Oak Ridge Institute for Science and Education
34 through an interagency agreement between the U.S. Department of Energy and the FDA.
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38 **Notes**

39 This article is not an official U.S. Food and Drug Administration (FDA) guidance or policy
40 statement. No official support or endorsement by the U.S. FDA is intended or should be
41 inferred. The authors declare no competing financial interest.
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47 **■ ACKNOWLEDGMENTS**

48 We thank Ms. Linda S. Von Tungeln, Ms. Michelle Vanlandingham, and Mr. Gene White for
49 assistance with the animals, Dr. Po-Chuen Chan of NTP for supplying riddelliine, Dr. John A.
50 Edgar, CSIRO Livestock Industries, Australia for supplying lasiocarpine, and Dr. H.S. Chen, the
51 Secondary Military Medical University, Shanghai, China for supplying platyphylline.
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■ ABBREVIATIONS

PAs, pyrrolizidine alkaloids; dehydro-PAs, dehydropyrrolizidine alkaloids; DHR, dehydroretronecine or (-)-R-6,7-dihydro-7-hydroxy-1-hydroxymethyl-5H-pyrrolizine); DHP, (±)-6,7-dihydro-7-hydroxy-1-hydroxymethyl-5H-pyrrolizine; HPLC-ES-MS/MS, high-performance liquid chromatography-electrospray ionization tandem mass spectrometry; NCTR, National Center for Toxicological Research; MRM, multiple reaction monitoring; LOD, limit of detection

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Table 1. Pearson Product Moment Correction Coefficients and P Values between (i) PA DHP Hemoglobin Adducts and PA DHP-DNA Adducts and (ii) PA DHP-DNA Adducts.

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PA hemoglobin or DNA adduct	ng DHP/g hemoglobin/mg PA/kg BW	DHP-dG-3/mg PA/kg BW	DHP-dG-4/mg PA/kg BW	DHP-dA-3/mg PA/kg BW	DHP-dA-4/mg PA/kg BW
ng DHP/g hemoglobin/mg PA/kg BW					
DHP-dG-3/mg PA/kg BW	0.631 ^a 0.04 ^b				
DHP-dG-4/mg PA/kg BW	0.702 0.02	0.947 1 x 10 ⁻⁵			
DHP-dA-3/mg PA/kg BW	0.573 0.07	0.956 4 x 10 ⁻⁶	0.951 7 x 10 ⁻⁶		
DHP dA-4/mg PA/kg BW	0.633 0.04	0.955 5 x 10 ⁻⁶	0.951 7 x 10 ⁻⁶	0.998 9 x 10 ⁻¹²	

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^aCorrelation coefficient

^bP-value

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Figure legends

Figure 1. Structures of pyrrolizidine alkaloids and derivatives used in this study.

Figure 2. The ion transitions of the MRM method for specific detection of 7,9-diethoxyl-DHP adducts in the samples were obtained through a combination of two routes: (i) $[M+H]^+ m/z 227 \rightarrow [M+H-C_2H_5OH]^+ m/z 164$ and (ii) $[M+NH_4]^+ m/z 210 \rightarrow m/z 164$, and collision energy 27 and 15 eV, respectively.

Figure 3. The quantitation of protein-DHP adducts in blood samples by converting the DHP-protein adducts into 7,9-di- C_2H_5O -DHP adduct by reaction of the protein adducts with silver nitrate in acidic ethanol.

Figure 4. LC/MS/MS MRM chromatograms of 7,9-di- C_2H_5O -DHP formed from the DHP-protein adducts contained in the hemoglobin fraction of rats dosed with 5 mg/kg body weight of riddelline for 30 consecutive days: (A) MRM total ions; (B) the MRM $m/z 210 - 164$; (C) the MRM $m/z 227 - 164$; and (D) the MRM $m/z 237 - 169$.

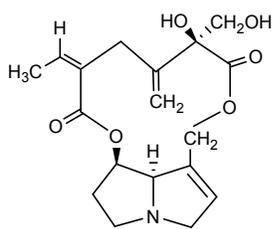
Figure 5. LC/MS/MS MRM chromatograms of 7,9-di- C_2H_5O -DHP formed from the DHP-protein adducts contained in the hemoglobin fraction of rats dosed with: (A) 0 mg; (B) 0.1 mg; (C) 1.0 mg; (D) 2.0 mg; and (E) 5 mg/kg body weight of riddelline for 30 consecutive days.

Figure 6. The levels of DHP-protein adducts (ng/g protein) identified as 7,9-di- C_2H_5O -DHP in hemoglobin and plasma fractions of female rats dosed with 0, 0.1, 1.0, 2.0, or 5.0 mg riddelline/kg bw for 5 (circle) or 30 (triangle) days (4 rats per group).

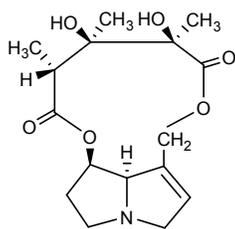
Figure 7. The mean levels of DHP-protein adducts identified as 7,9-di- C_2H_5O -DHP in the hemoglobin fraction of female rats dosed with PAs at 4.5 and 24 $\mu\text{mol/kg bw/day}$ for three consecutive days. The data are normalized to ng 7,9-di- C_2H_5O -DHP/g hemoglobin/mg PA/kg bw.

Figure 8. Pearson Product Moment Correction coefficient and P value; (A) between DHP-hemoglobin adducts and DHP-dA-4; and (B) between DHP-dA-3 and DHP-dA-4 of female rats dosed with PAs at 4.5 or 24 $\mu\text{mol/kg bw/day}$ for three consecutive days.

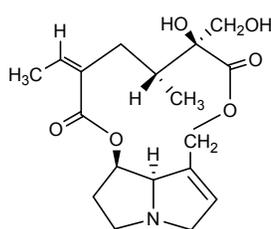
Retronecine-type tumorigenic PA



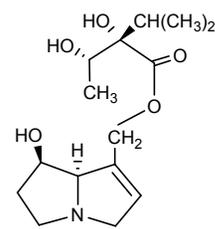
Riddelliine



Monocrotaline

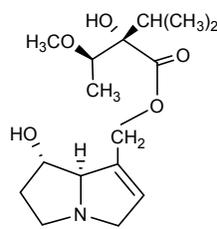


Retrorsine

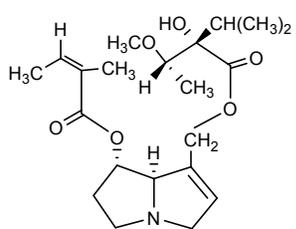


Lycopsamine

Heliotridine-type tumorigenic PA

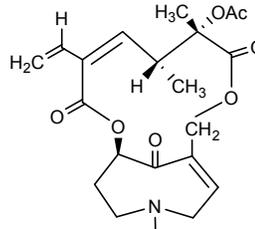


Heliotrine

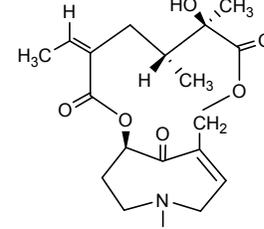


Lasiocarpine

Otonecine-type tumorigenic PA

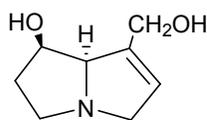


Clivorine

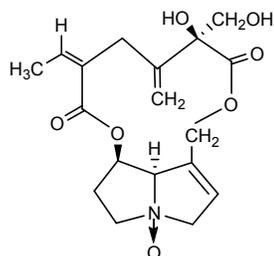


Senkirkine

Tumorigenic PA N-oxide

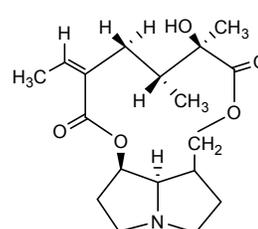


Retronecine



Riddelliine N-oxide

Non-tumorigenic PA



Platyphylline

Figure 1.

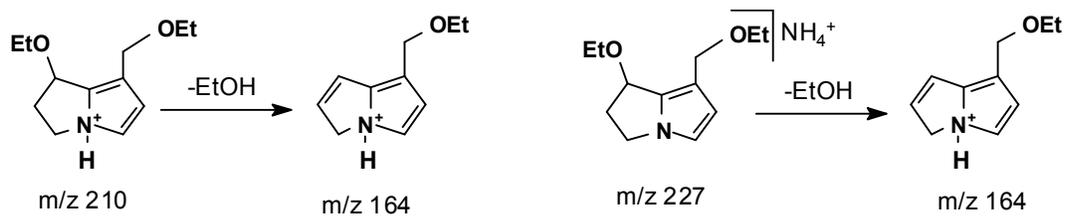


Figure 2.

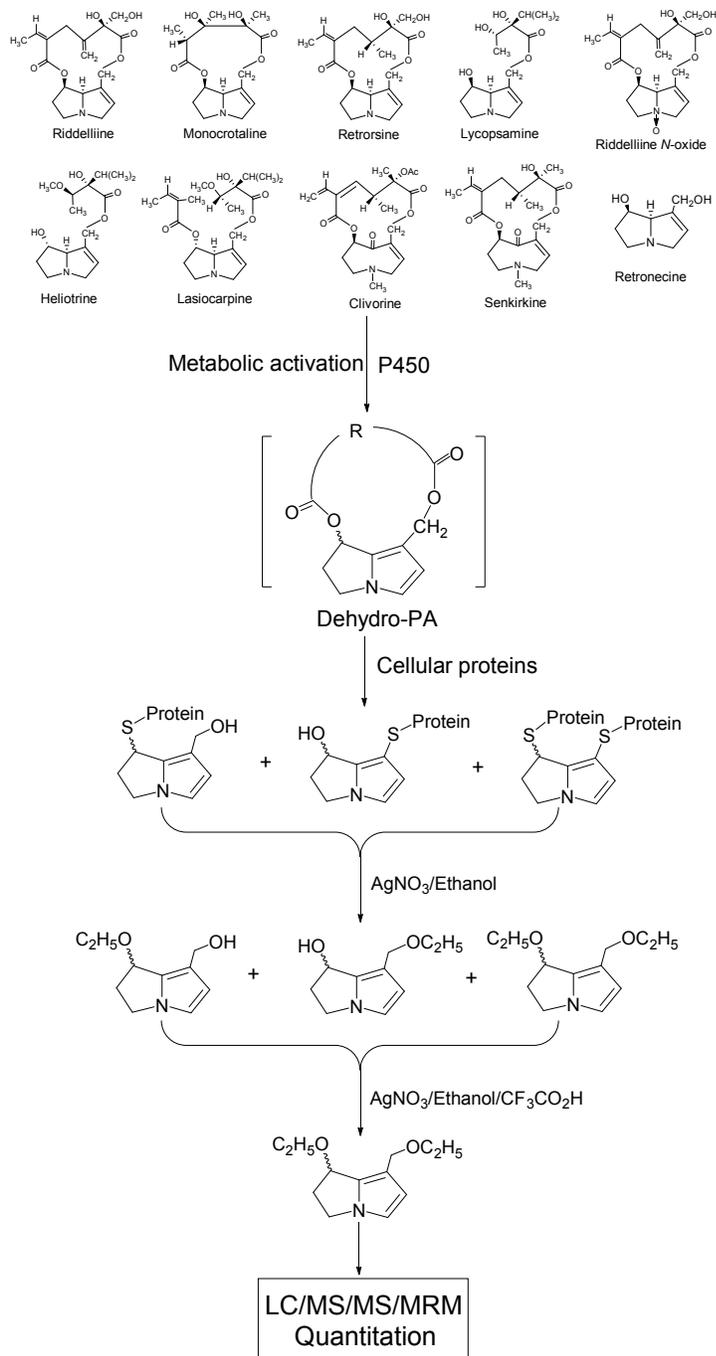


Figure 3.

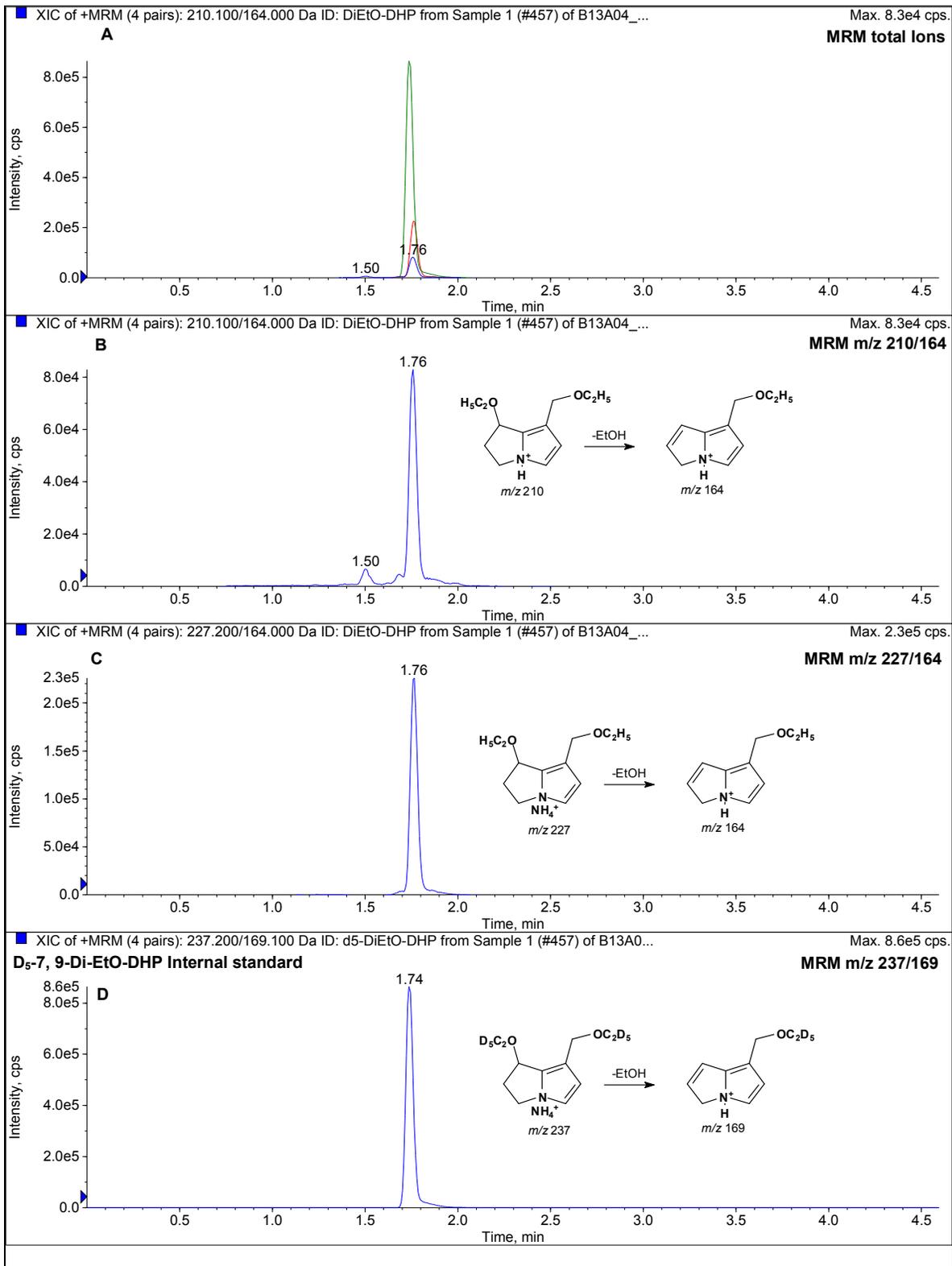


Figure 4.

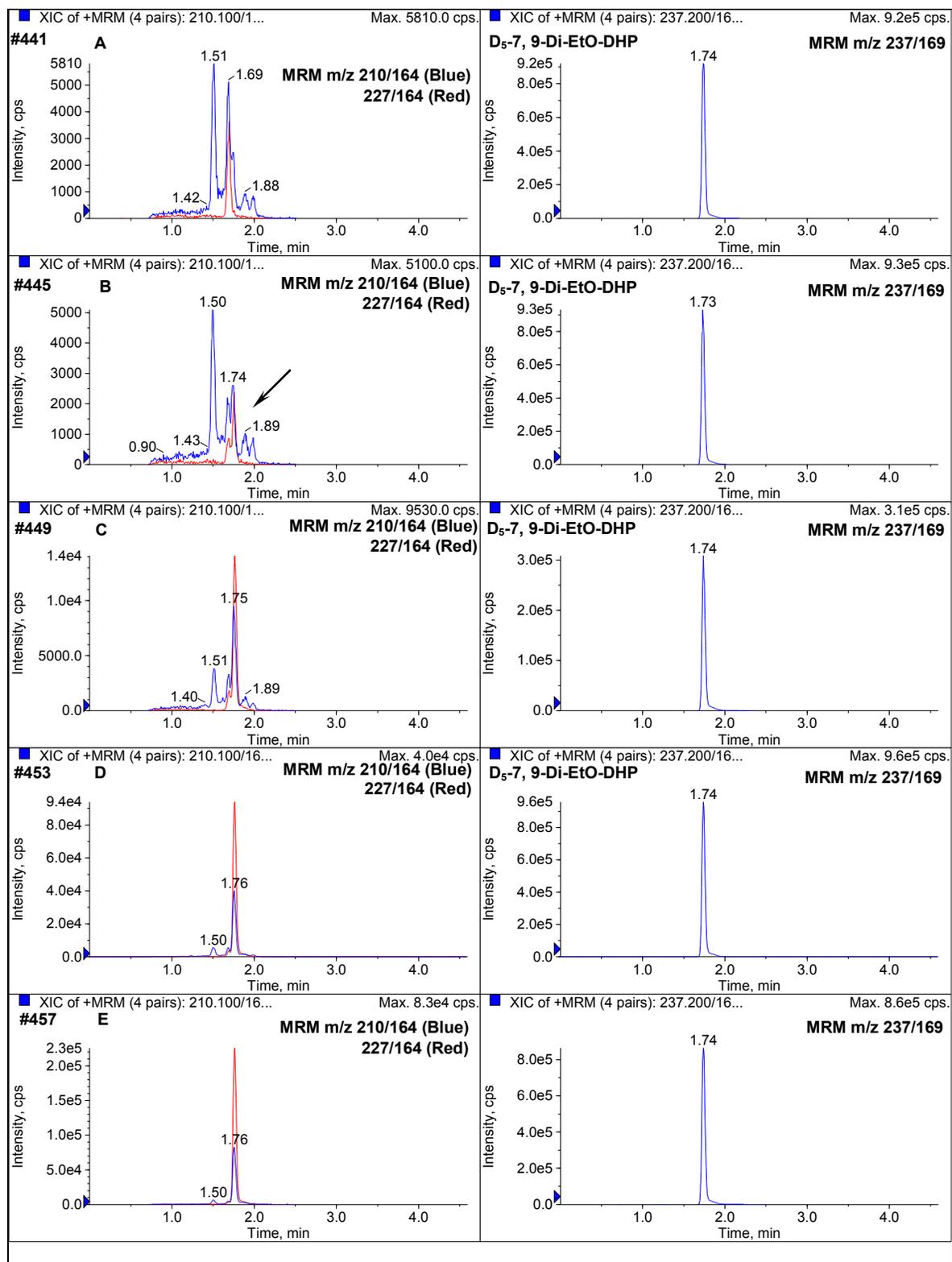


Figure 5.

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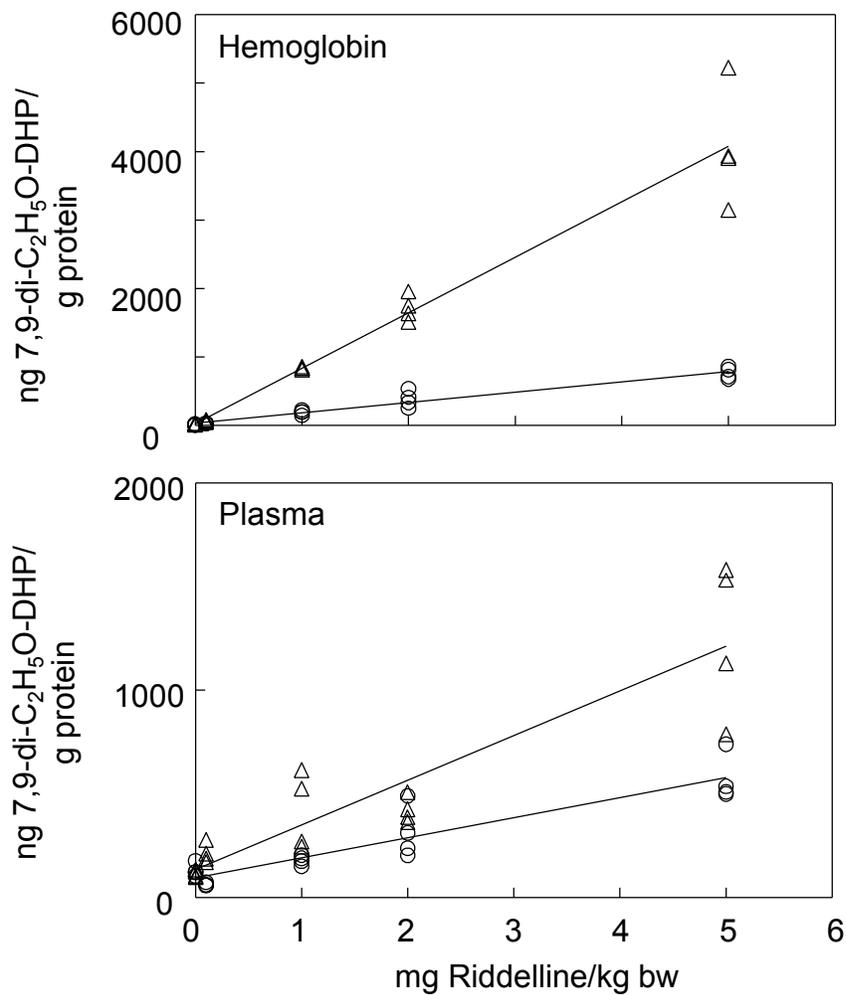


Figure 6.

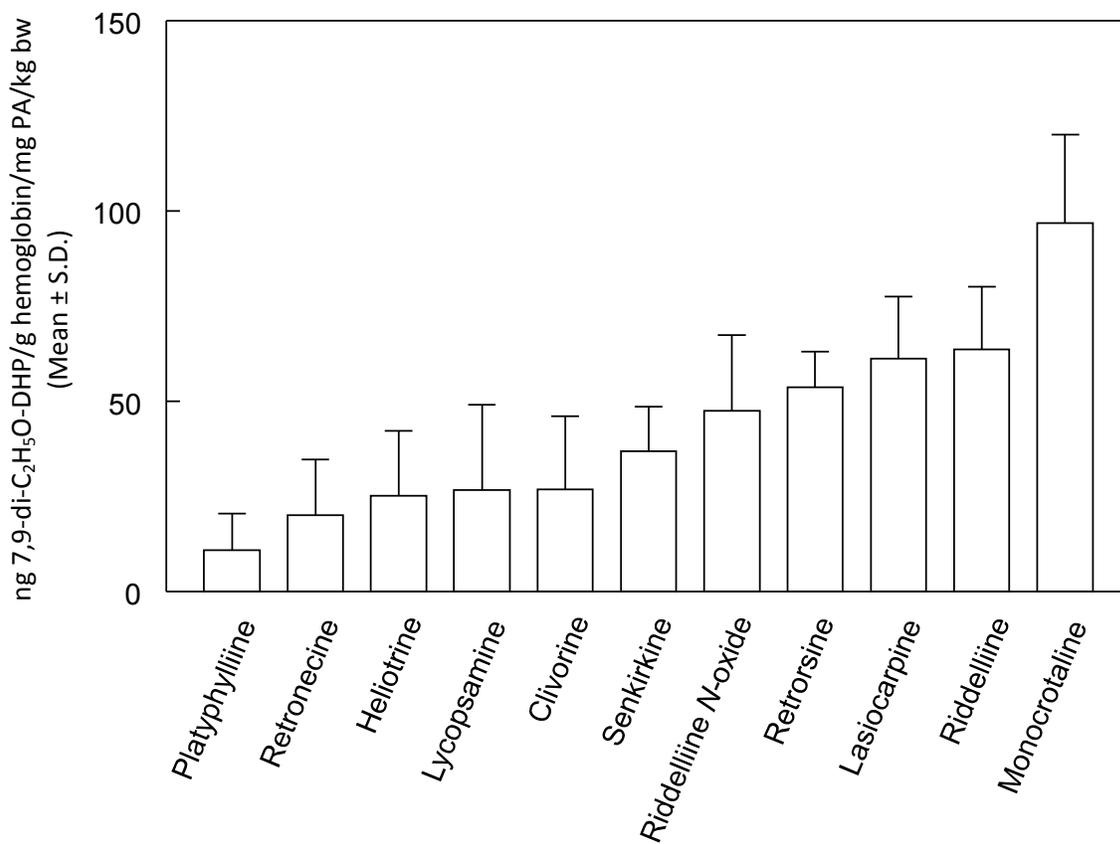


Figure 7.

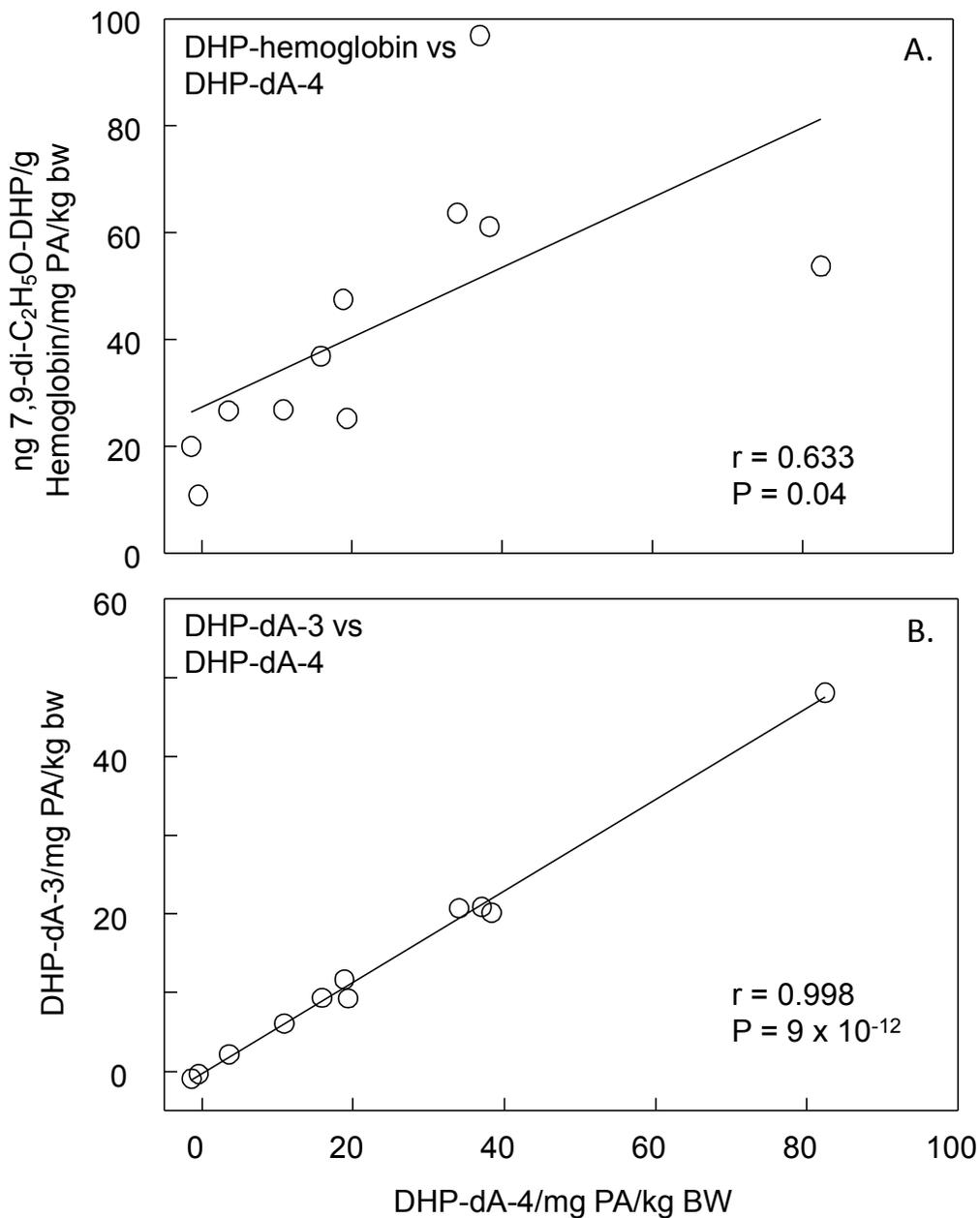


Figure 8.