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Synthesis and evaluation of new fluorine-18 labeled verapamil analogs to investigate the function of P-glycoprotein in the blood-brain barrier

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Abstract

P-glycoprotein is an efflux transporter located in the blood-brain barrier. (*R*)-[¹¹C]verapamil is widely used as a PET tracer to investigate its function in patients with epilepsy, Alzheimer's disease and other neurodegenerative diseases. Currently it is not possible to use this successful tracer in clinics without a cyclotron, because of the short half-life of carbon-11. We developed two new fluorine-18 labeled (*R*)-verapamil analogs, with the benefit of a longer half-life. The synthesis of (*R*)-*N*-[¹⁸F]fluoroethylverapamil ([¹⁸F]**1**) and (*R*)-*O*-[¹⁸F]fluoroethylnorverapamil ([¹⁸F]**2**), has been described. [¹⁸F]**1** was obtained in reaction of (*R*)-norverapamil with the volatile [¹⁸F]fluoroethyltriflate acquired from bromoethyltosylate and a silvertriflate column with a radiochemical yield of $2.7 \pm 1.2\%$. [¹⁸F]**2** was radiolabeled by direct fluorination of precursor **13** and required final Boc-deprotection with TFA resulting in a radiochemical yield of $17.2 \pm 9.9\%$. Both tracers [¹⁸F]**1** and [¹⁸F]**2** were administered to Wistar rats, and blood plasma and brain samples were analyzed for metabolic stability. Using [¹⁸F]**1** and [¹⁸F]**2**, PET scans were performed in Wistar rats at baseline and after blocking with tariquidar, showing a 3.6 and 2.4-fold increase in brain uptake in the blocked rats, respectively. In addition, for both [¹⁸F]**1** and [¹⁸F]**2**, PET scans in *Mdr1a/b*^(-/-), *Bcrp1*^(-/-) and WT mice were acquired, in which [¹⁸F]**2** showed a more specific brain uptake in *Mdr1a/b*^(-/-) mice and no increased signal in *Bcrp1*^(-/-) mice. [¹⁸F]**2** was selected as the best performing tracer and should be evaluated further in clinical studies.

Keywords

P-gp, BCRP, positron emission tomography, blood-brain barrier, verapamil, 18F, ABC transporters

Introduction

The blood-brain barrier (BBB) is a diffusion barrier between the central nervous system and the circulation that protects the brain from entrance of neurotoxic substances. Endothelial cells with tight junctions prevent paracellular passage. Transcellular passage is possible by selective transport of, for example, amino acids, vitamins, and sugars via their specific transporters.¹ ATP-binding cassette (ABC) transporters constitute an important transporter family at the BBB with the most investigated ones being P-glycoprotein (P-gp/ABCB1) and breast cancer resistance protein (BCRP/ABCG2). Both are ATP dependent efflux transporters, mediating the transport of structurally diverse compounds from brain to blood, and share some substrates. Research supports that P-gp function is diminished in Alzheimer's disease, causing accumulation of β -amyloid, in the brain.² On the other hand, several studies have shown increased P-gp function in epilepsy patients, causing drug resistance.³ A number of studies have used positron emission tomography (PET) to investigate the function of P-gp with radiolabeled substrates. The most widely used PET tracer for P-gp is [¹¹C]verapamil, originally a calcium channel blocker, but also a substrate of P-gp.⁴ In early studies it was used as a racemic mixture, but later it was shown that the (*R*)-enantiomer had better *in vivo* stability than the (*S*)-isomer and the use of a single isomer is essential for quantification of P-gp function.⁵ In a clinical PET study using (*R*)-[¹¹C]verapamil, (mild) AD patients showed 50% reduced P-gp function in brain regions related to AD compared with healthy, age-matching controls.⁶

Although (*R*)-[¹¹C]verapamil has shown its usefulness in clinical research, it has several drawbacks. Metabolite studies have shown that N-dealkylation of the parent compound results in the radiolabeled metabolite [¹¹C]D617, which turned out to be a P-gp substrate as well. Metabolite formation could disturb the PET image when the radiolabeled metabolites interact with P-gp in a different matter than the parent compound and makes quantification challenging. To circumvent this problem, [¹¹C]D617 itself was investigated as a P-gp tracer.⁷ *In vivo* studies, however, showed lower affinity of [¹¹C]D617 for P-gp than (*R*)-[¹¹C]verapamil and therefore it was not be a satisfactory substitute. The aspect of (*R*)-[¹¹C]verapamil we want to tackle is the use of the isotope carbon-11, with the short half-life of 20 minutes. Although, the rapidly decaying isotope gives the opportunity for multiple PET scans in one day with the same object, it is only possible to perform studies with the PET tracer at a facility in close proximity of a cyclotron. A fluorine-18 labeled PET tracer would give the opportunity to investigate P-gp in almost all facilities in possession of a PET scanner. Besides transport, a longer half-life gives the

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3 opportunity for longer scan time and multiple patient batches out of one tracer production. The energy
4 of the positron emission of fluorine-18 is lower compared with carbon-11 and results in a lower range
5 and better resolution of the PET image.
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9 Recently, some new fluorine-18 labeled P-gp PET tracers have been developed. [¹⁸F]Gefitinib, originally
10 an epidermal growth factor receptor inhibitor and a substrate for P-gp and BCRP,⁸ was used in a study to
11 examine drug-drug interactions at the BBB. However, again this is not a specific P-gp tracer, since it is
12 also transported by BCRP.⁹ In another comprehensive study of Sander et al., fluorine-18 labeled tracers
13 were used in a new approach to metabolically activate a dual P-gp/BCRP substrate. Although the proof-
14 of-concept was successful, slow rates of enzymatic conversion towards the prodrug tracer did not give
15 the anticipated results.¹⁰ Another study proposed three new fluorine-18 labeled P-gp tracers. Only one
16 of the three showed P-gp substrate functionality *in vivo* although all tracers were evaluated as
17 substrates *in vitro*.^{11, 12}
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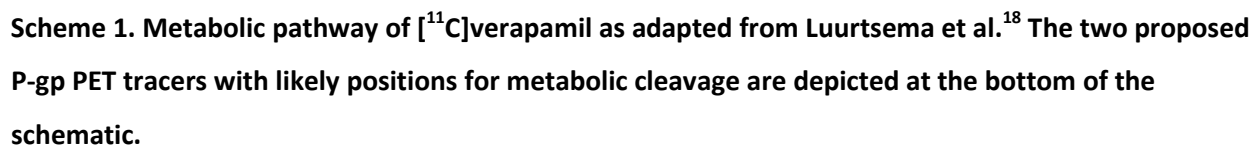
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21 Whilst these studies are all attempts to label a substrate of P-gp, to image the expression levels of P-gp
22 in relevant brain regions, an inhibitor of binding ligand is needed. Two fluorine-18 labeled P-gp tracers
23 were based on analogs of the P-gp inhibitors tariquidar and elacridar. 1-[¹⁸F]fluoroelacridar showed
24 excessive defluorination *in vivo* and was not further developed.¹³ [¹⁸F]fluoroethylelacridar and
25 [¹⁸F]fluoroethyltariquidar showed P-gp substrate behavior *in vivo*, with a small increase in brain uptake
26 for *Mdr1a/b*^(-/-) mice, but much higher for *Mdr1a/b*^(-/-)*Bcrp1*^(-/-) mice, indicating that they were
27 substrates for both P-gp and BCRP.¹⁴ Therefore, the tracers could not be used for the intended purpose
28 of quantification of P-gp function.
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32 To date, no fluorine-18 labeled P-gp tracer is available for clinical use. Since (*R*)-[¹¹C]verapamil has
33 shown its value in multiple clinical studies, the purpose of the present study was to investigate two
34 novel fluorine-18 analogs of verapamil, (*R*)-*N*-[¹⁸F]fluoroethylverapamil ([¹⁸F]**1**) and (*R*)-*O*-
35 [¹⁸F]fluoroethylnorverapamil ([¹⁸F]**2**), as tracers of P-gp function.
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Results & Discussion

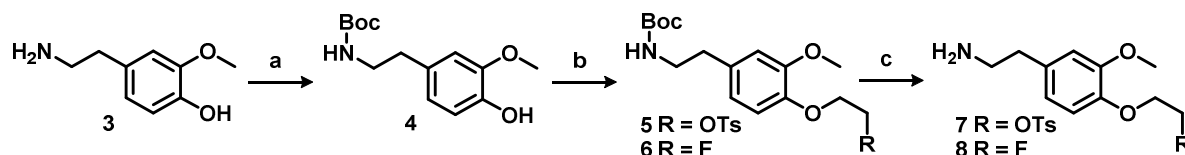
Chemistry

The aim of this study was to develop a new fluorine-18 labeled P-gp substrate, therefore two verapamil analogs containing a fluoroethyl group were investigated. The two different positions of the fluoroethyl group were chosen in order to investigate the effect on pharmacokinetics and metabolism. For [^{18}F]**1**, the fluoroethyl group was placed on the original position of the [^{11}C]methyl group on the amine. To make [^{18}F]**2** less prone to metabolism, one of the methoxy groups was replaced with the fluoroethyl group, keeping in mind that the average bond enthalpy between a C-O bond is higher and therefore stronger than a C-N bond.¹⁵ The metabolic pathway of [^{11}C]verapamil has been studied thoroughly¹⁶ and its main metabolite is norverapamil, with the [^{11}C]methyl group cleaved off resulting in polar radioactive metabolites (Scheme 1). In addition, the formed metabolite [^{11}C]D617 of [^{11}C]verapamil, was proven to be a substrate of P-gp,⁷ and it is assumed that the ^{18}F -analog of D617 would also act as a substrate of P-gp. The formation of radioactive metabolites could provide an (unknown) background signal to the PET images and, therefore, make quantitative interpretation of PET data difficult if not impossible. To circumvent these issues, the [^{18}F]fluoroethyl group was placed on the phenol that is not part of D617, at the same time removing the methyl group on the amine to avoid the first metabolic step towards norverapamil, since this is a substrate of P-gp as well.¹⁷

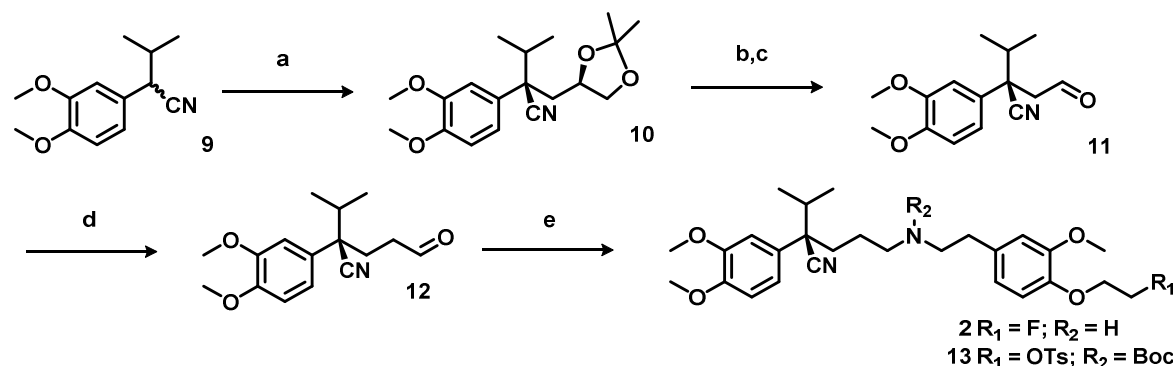


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reductive amination, without reducing the volume of solvent. With a yield of 44% over the reductive amination and Boc protection, this is a viable method to synthesize (*R*)-precursor **13**. The reference compound **2** was synthesized from aldehyde **12** and amine **8** containing the fluoroethoxy group.



Scheme 2. (a) Boc_2O , Et_3N , MeOH, r.t., 2h (b) synthesis of **5**: Ethylene di(*p*-toluenesulfonate), K_2CO_3 , KI, DMF, r.t., 3h; synthesis of **6**: 1-bromo-2-fluoroethane, K_2CO_3 , KI, DMF, 75°C , 18h (c) TFA, DCM, r.t., 1h



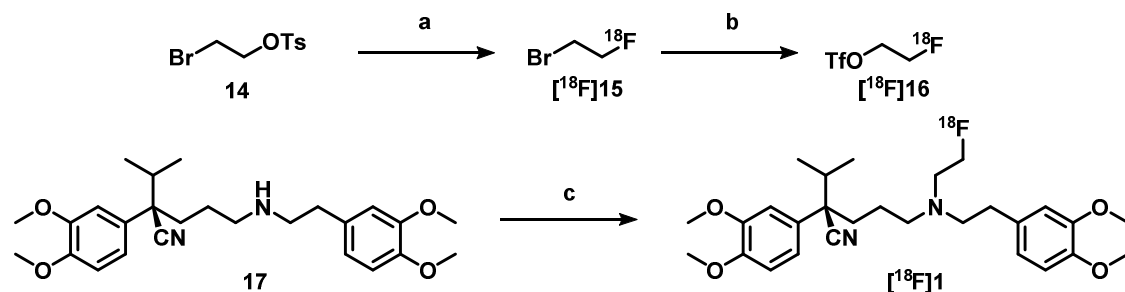
Scheme 3. (a) NaH, DMF, 65°C , 3h, (*R*)-2,2-dimethyl-1,3-dioxolan-4-ylmethyl 4-methylbenzenesulfonate, 65°C , 18h, flash column chromatography; (b) AcOH, H_2O , r.t., 18h; (c) NaIO_4 , NaHCO_3 , CH_2Cl_2 , r.t., 4h; (d) $\text{MeOCH}=\text{PPh}_3$, THF, -80°C to r.t., 4h; subsequently: *p*-TSA, *i*-PrOH, H_2O , 80°C , 3h; (e) synthesis of **2**: **8**, $\text{NaBH}(\text{OAc})_3$, Na_2SO_4 , MeOH, r.t., 18h; synthesis of **10**: (i) **7**, $\text{NaBH}(\text{OAc})_3$, Na_2SO_4 , DCE, r.t., 18h; (ii) Boc_2O , Et_3N , EtOAc, r.t., 1.5h;

The precursor of [^{18}F]**1**, (*R*)-norverapamil, was kindly donated by Abbott Laboratories and therefore the synthesis of the reference compound of [^{18}F]**1** was straightforward. (*R*)-norverapamil was alkylated with 1,2-bromofluoroethane to obtain reference compound **1**.

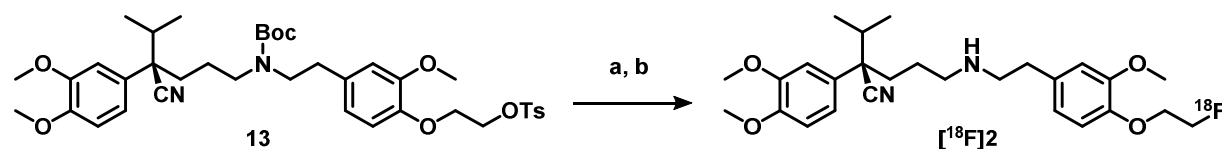
Radiochemistry

Although its precursor was readily available, the radiochemical synthesis of [^{18}F]**1** was more challenging. Radiolabeling of (*R*)-norverapamil to obtain [^{18}F]**1** was first pursued by fluorinating bromoethyltosylate to give bromo[^{18}F]fluoroethane ([^{18}F]**15**), which was distilled into a second vial containing a solution with

3 mg (*R*)-norverapamil **17** and 1.6 mg K_2CO_3 that was heated to 120 °C in MeCN. However, for the second step, only 0.5% conversion was observed on HPLC. Therefore, a silvertriflate oven was introduced, to conduct volatile bromo ^{18}F fluoroethane conversion into ^{18}F fluoroethyltriflate (^{18}F **16**), substituting the bromine for a stronger leaving group, and this intermediate was bubbled through the precursor solution (Scheme 4). This resulted in higher conversion and when the solution was also stirred to improve solubility of K_2CO_3 , the final reaction conditions were achieved, resulting in an overall yield of $2.4 \pm 1.4\%$, a specific activity of 143 ± 88 GBq/ μ mol and a radiochemical purity of >99%. The yield of ^{18}F **1** was still lower than expected, which is mainly caused by poor trapping of activity in the second reaction vial after distillation. Nevertheless, sufficient activity was obtained for preclinical studies. For ^{18}F **2**, the precursor synthesis was designed to prevent this problem by addition of a tosyl group for direct fluorination, followed by Boc-deprotection (Scheme 5). No optimization was required for animal experimentation and the overall yield was $14.3 \pm 6.8\%$, with a specific activity of 151 ± 74 GBq/ μ mol and a radiochemical purity of >99%.



Scheme 4. (a) $^{18}F/K_{2.2.2}/K^+$, DMF, 90°C, 15 min; (b) AgOTf, 200°C, 15 min; (c) ^{18}F fluoroethyltriflate, K_2CO_3 , MeCN, 120°C, 15 min



Scheme 5. (a) $^{18}F/K_{2.2.2}/K^+$, MeCN, 90°C, 5 min; (b) TFA, 20°C, 10 min

Biodistribution

The biodistribution of ^{18}F **1** in Wistar rats shows a relatively slow washout from the blood pool and other related organs and low brain uptake, presumably caused by the efflux transport of P-gp (Figure 1). Since bone uptake was low and did not increase over time, defluorination of the tracer appears to be

absent. Different behavior in rats was observed for [¹⁸F]**2**, with lower brain uptake (compared with [¹⁸F]**1**) and high initial uptake in kidney and lung, which decreases over time. This is similar to the biodistribution of [¹¹C]verapamil.²¹ In some cases, high lung uptake can be explained by perfusion, which is related to the activity in plasma. However, the blood values for [¹⁸F]**2** are even lower than [¹⁸F]**1**, therefore this could not be concluded. [¹⁸F]**2** is not prone to defluorination based on its low bone uptake.

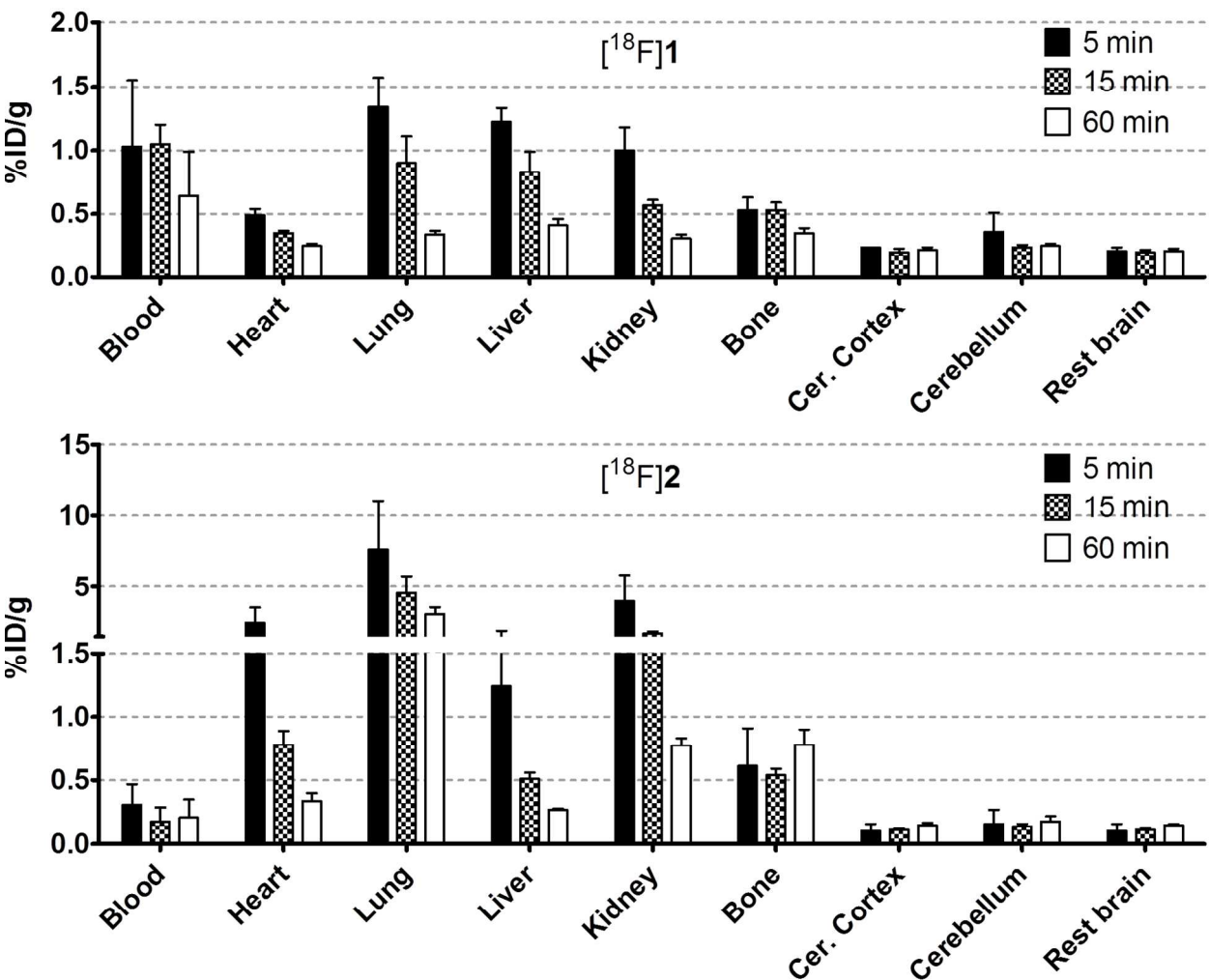


Figure 1. Biodistribution of [¹⁸F]**1** and [¹⁸F]**2** in selected organs. Data are expressed as percentage injected dose per gram tissue (%ID/g).

Metabolite analysis

Metabolite analyses of both tracers were performed in healthy Wistar rats, 5, 15 and 60 min after tracer injection, and the results are shown in Table 1. For both tracers, the parent tracer was found in the non-polar fraction and was identified by HPLC. The metabolite analysis showed a rapid rate of metabolism in rats for both PET tracers. The main metabolite fraction was the polar metabolites. Since bone uptake is not observed in the PET scans (*vide infra*) and biodistributions, the polar metabolites are not products of defluorination. Therefore, the high polar fraction consists of metabolites most likely formed via cleavage of the fluoroethyl group.

For [^{18}F]**1**, the metabolic pathway is expected to be similar to [^{11}C]verapamil, which gives nor-verapamil as the main metabolite, next to polar mono-carbon labeled ones like [^{11}C]formaldehyde, [^{11}C]formic acid and [^{11}C]carbon dioxide (Scheme 1). The high polar fraction observed in the metabolite analysis of [^{18}F]**1** indicates that the [^{18}F]fluoroethyl group is even more prone to metabolism than the [^{11}C]methyl group, resulting in [^{18}F]fluoroacetaldehyde, and other small oxidized fractions. Furthermore, a peak, likely corresponding to [^{18}F]fluoroethyl labeled D617, was observed in the non-polar fraction on HPLC. This presumably can act as a substrate of P-gp, similar to [^{11}C]D617, which could interfere with the PET signal when it interacts with P-gp in a different matter than the parent tracer.

With this in mind, it was expected that [^{18}F]**2** would show a lower rate of metabolism, since this first metabolic step is avoided by eliminating the amine bound methyl group in the designed structure. However, this effect was not observed and high polar fractions were measured. Possible enzyme cleavage sites are the [^{18}F]fluoroethyl group and the amine bound alkyl groups, with the latter being less sterically hindered in the secondary amine compared with a tertiary amine (Scheme 1). Labeled metabolites will not be analogs of D617, since that part of the molecule was not labeled with fluorine-18. Therefore, it is unlikely that labeled metabolites are substrates of P-gp.

It was shown that metabolism of [^{11}C]verapamil was less rapid in humans compared with rats. While only $28.1 \pm 2.7\%$ of the total activity in the rat plasma was parent tracer at 60 min,¹⁸ in humans it was $45 \pm 9\%$.²² Therefore we anticipate that while tracers [^{18}F]**1** and [^{18}F]**2** are metabolized rapidly in rats, in humans this will likely be slower, but must be monitored during the translation of these tracers to humans.

Table 1. [¹⁸F]**1** and [¹⁸F]**2** and its radiolabeled metabolites in plasma and brain tissue (percentage of total radioactivity, mean ± SD) in Wistar rats.

	min	[¹⁸ F] 1		[¹⁸ F] 2	
		plasma	brain	plasma	brain
parent tracer	5	46 ± 14%	41 ± 10%	20 ± 3%	26 ± 6%
	15	19 ± 2%	14 ± 2%	8 ± 3%	17 ± 7%
	60	3 ± 1%	2 ± 0.3%	4 ± 1%	6 ± 1%
non-polar metabolites	5	5 ± 2%		5 ± 3%	
	15	9 ± 3%		5 ± 1%	
	60	5 ± 1%		3 ± 1%	
polar metabolites	5	49 ± 11%		75 ± 3%	
	15	71 ± 2%		87 ± 1%	
	60	92 ± 1%		93 ± 2%	
Brain metabolites	5		59 ± 10%		74 ± 6%
	15		86 ± 2%		83 ± 7%
	60		98 ± 0.3%		94 ± 1%

P-gp blocking study with tariquidar

Rats were treated with P-gp inhibitor tariquidar 30 min before injection of [¹⁸F]**1** or [¹⁸F]**2**, a method in line with previous reports studying P-gp in rats.²³ Brain uptake of [¹⁸F]**1** was 3.6-fold higher than in the baseline scans (Figure 2a). The highest brain uptake was at 5 min. For [¹⁸F]**2**, a 2.4-fold higher brain uptake was observed in tariquidar treated animals compared with baseline brain uptake (Figure 2b). However, a difference was observed in pharmacokinetics between the two tracers. Whilst behavior of [¹⁸F]**1** was similar to that of verapamil, with an injection peak followed by washout over time, [¹⁸F]**2** showed no injection peak, while both tracers were injected via a comparable bolus injection. Instead, a more steady time activity curve of [¹⁸F]**2** resulted in higher activity levels at the end of the scan. It is likely that this difference is caused by the higher polarity of [¹⁸F]**2**, due to the secondary amine, as compared with the tertiary amine of [¹⁸F]**1**. This could lead to more difficult, thus slower, passive diffusion of [¹⁸F]**2** through the BBB and therefore delayed interaction with P-gp. This is supported by the measured log *D* values of 2.08 and 1.61, of [¹⁸F]**1** and [¹⁸F]**2**, respectively.

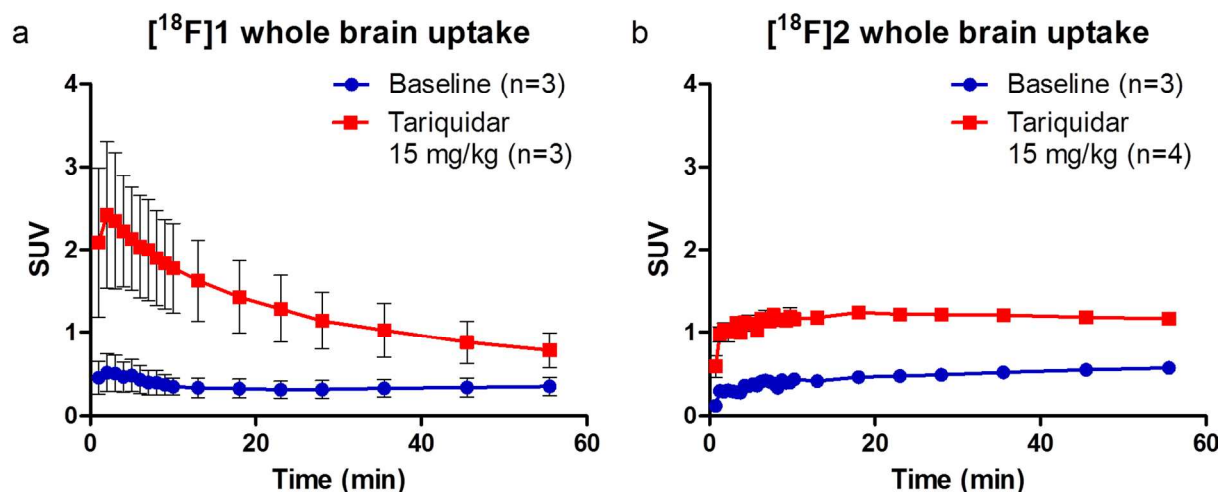


Figure 2. Whole brain time-activity curves of $[^{18}\text{F}]\mathbf{1}$ and $[^{18}\text{F}]\mathbf{2}$ in Wistar rats at ● baseline and ■ after treatment with tariquidar (15 mg/kg).

PET imaging in $Mdr1a/b^{-/-}$ and $Bcrp1^{-/-}$ mice

Given that, at a dose of 15 mg/kg, tariquidar is known to be an inhibitor of both P-gp and BCRP,^{24, 25} another PET study using knockout animals was performed to assess the specificity of the two tracers (Figure 3). To enable a direct comparison, a similar study was performed using (*R*)- $[^{11}\text{C}]\text{verapamil}$ (Figure 4). Unfortunately, as the HRRT scanner had just been decommissioned, the $[^{18}\text{F}]\mathbf{2}$ study in $Mdr1a/b^{-/-}$ mice vs. wild-type (WT) mice was performed using new PET/CT and PET/MR small animal scanners. This led to slightly different results for the wild-type animals, compared with the $Bcrp1^{-/-}$ vs. wild-type mice study, probably due to differences in spatial resolution. Therefore, TACs are presented separately in Figures 3a-d. Figures 3e and 3f represent the ratio of brain uptake in $Mdr1a/b^{-/-}$ /WT and $Bcrp1^{-/-}$ /WT for both $[^{18}\text{F}]\mathbf{1}$ and $[^{18}\text{F}]\mathbf{2}$, respectively, to normalize for the scanner differences.

For both tracers, no significant differences in brain uptake were observed between wild-type and $Bcrp1^{-/-}$ mice, indicating that neither are BCRP substrates (Figures 3c,d). Both tracers showed increased brain uptake in $Mdr1a/b^{-/-}$ mice (Figures 3a,b). A smaller difference in brain uptake was noticed for $[^{18}\text{F}]\mathbf{1}$, with only a 1.4-fold higher brain uptake during the first 10 minutes compared with wild-type mice. This is mainly due to high uptake in the wild-type animals, which suggests that $[^{18}\text{F}]\mathbf{1}$ is a poorer tracer of P-gp (Figure 3e) than $[^{18}\text{F}]\mathbf{2}$. For $[^{18}\text{F}]\mathbf{2}$, uptake in wild-type mice was similar to that of $[^{11}\text{C}]\text{verapamil}$. Although metabolism of $[^{18}\text{F}]\mathbf{2}$ was even more rapid than that of $[^{18}\text{F}]\mathbf{1}$, $[^{18}\text{F}]\mathbf{2}$ in $Mdr1a/b^{-/-}$ mice showed steady brain uptake up to a 6.4-fold higher level than that in wild-type mice, reaching a SUV plateau of about 0.9 at the end of the scan. The slower increase in brain uptake could be

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due to the lower log *D* value and therefore difficulty in crossing the BBB. The irreversible kinetics of [¹⁸F]**2** are in general not ideal for PET analysis. In this study, no blood samples were collected from mice, therefore kinetic modelling was not possible. To obtain more insight in the relationship between the activity concentration in the blood and total brain uptake, blood data was collected from the scans.

As shown in Table 2, both [¹⁸F]**1** and [¹⁸F]**2** showed a higher brain-to-blood AUC ratio in *Mdr1a/b*^(-/-) mice, compared with WT animals. However, [¹⁸F]**1** also showed an increased brain-to-blood ratio in *Bcrp1*^(-/-) mice. Since both compounds are close analogs of (*R*)-[¹¹C]verapamil, which is a P-gp specific PET tracer, the dual P-gp/Bcrp PET tracer behavior of [¹⁸F]**1** is an unexpected observation.

Table 2. Brain-to-blood AUC ratios

	groups	brain-to-blood AUC ratios
[¹⁸ F] 1	WT	0.52 ± 0.05
	<i>Mdr1a/b</i> ^(-/-)	0.98 ± 0.07
	<i>Bcrp1</i> ^(-/-)	0.88 ± 0.06
[¹⁸ F] 2	WT	0.18 ± 0.04
	<i>Mdr1a/b</i> ^(-/-)	0.75 ± 0.14
	<i>Bcrp1</i> ^(-/-)	0.20 ± 0.02

In case of [¹⁸F]**2**, brain-to-blood AUC ratios were statistically significant (*P* < 0.05, Student's *t*-test for paired data) for *Mdr1a/b*^(-/-) vs. WT and *Mdr1a/b*^(-/-) vs. *Bcrp*^(-/-) animals. This was not the case for [¹⁸F]**1** where the difference between *Mdr1a/b*^(-/-) vs. WT mice and *Bcrp*^(-/-) vs. WT mice were significant, but not that between *Mdr1a/b*^(-/-) vs. *Bcrp*^(-/-) animals.

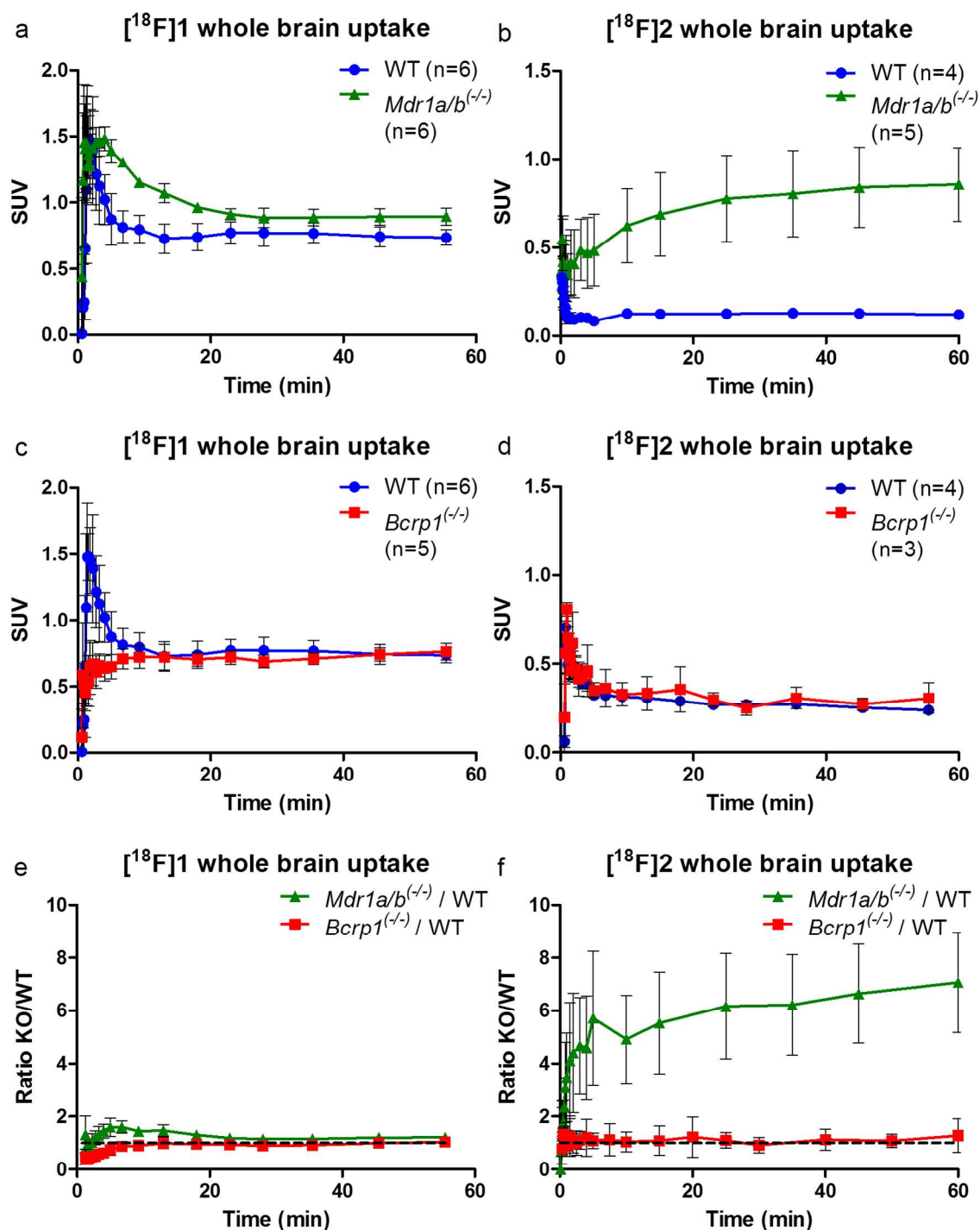


Figure 3. Time-activity curves of whole brain uptake of $[^{18}\text{F}]1$ and $[^{18}\text{F}]2$ in \bullet wild-type (WT) mice \blacksquare $Bcrp1^{-/-}$ mice or \blacktriangle $Mdr1a/b^{-/-}$ mice. $[^{18}\text{F}]1$ in a) WT vs. $Mdr1a/b^{-/-}$ c) WT vs. $Bcrp1^{-/-}$ e) Ratio of both $Mdr1a/b^{-/-}$ and $Bcrp1^{-/-}$ over WT. $[^{18}\text{F}]2$ in b) WT vs. $Mdr1a/b^{-/-}$ d) WT vs. $Bcrp1^{-/-}$ f) Ratio of both $Mdr1a/b^{-/-}$ and $Bcrp1^{-/-}$ over WT.

The high $Mdr1a/b^{-/-}$ /WT ratio of $[^{18}\text{F}]2$ over 60 min (Figure 3f) shows a specific P-gp substrate at the same ratio of (R)- $[^{11}\text{C}]$ verapamil, which is confirmed by the significant differences of the brain-to-blood

AUC ratios, despite the rapid rate of metabolism from the previous analysis (Table 1). Possible explanations are differences in expression of P-gp between species and the rate of metabolism in rats vs. mice.^{18, 26, 27} Another explanation of the stable TAC pattern of [¹⁸F]**2**, is found in the similarity with [¹¹C]dLop, another P-gp PET tracer, which was shown to be caused by acidic lysosomal trapping in the brain. This is possible for more basic radiotracers, like [¹⁸F]**2** with the secondary amine, that have the ability to diffuse into lysosome and subsequently be protonated in the acid lysosomal interior.²⁸ A third possibility is the effect of anesthesia on the different species and the different time frames (repeated anesthesia in rats vs. one time anesthesia in mice), which was not examined. However, in a paper of Wanek et al. mice were tested with (*R*)-[¹¹C]verapamil under long and short period of time (160 and 5 min, respectively) under anesthesia, and no significant effect was observed.²⁷ Whilst this behavior is not fully explained for [¹⁸F]**2**, it does show specific brain uptake in *Mdr1a/b*^(-/-) mice supporting its potential use as P-gp PET tracer.

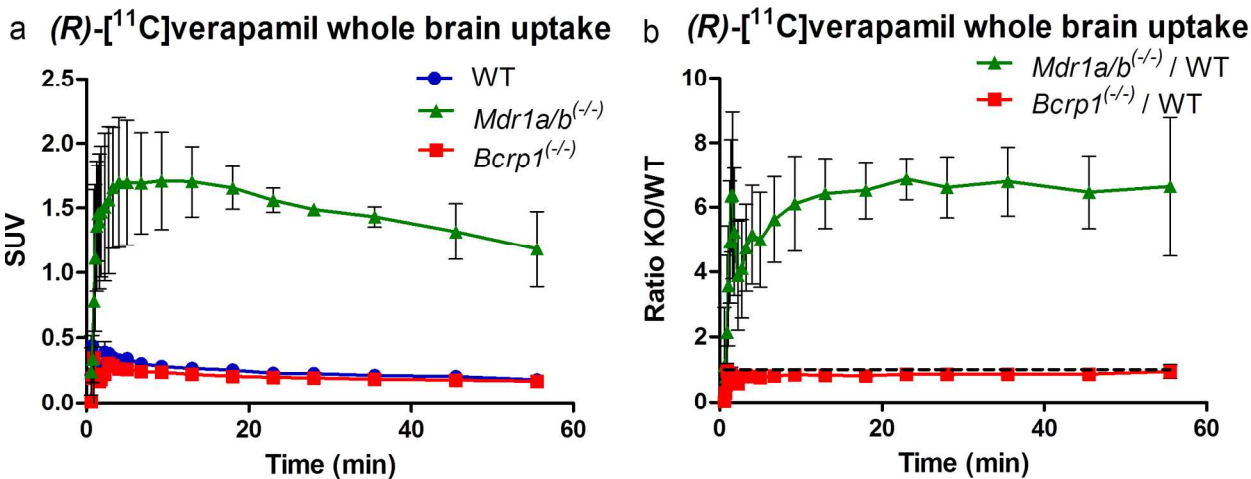


Figure 4. a) Time-activity curve of whole brain uptake of [¹¹C]verapamil in ● wild-type (WT) mice (n=2) ■ *Bcrp1*^(-/-) mice (n=3) or ▲ *Mdr1a/b*^(-/-) mice (n=2). b) Ratio of both *Mdr1a/b*^(-/-) and *Bcrp1*^(-/-) over WT.

Conclusion

Both PET tracers showed substrate behavior in tariquidar treated rats with different patterns over time. Whilst [¹⁸F]**1** showed higher initial uptake followed by faster washout, [¹⁸F]**2** showed slower brain uptake. As tariquidar is an inhibitor of both BCRP and P-gp, the PET study in knockout mice showed that [¹⁸F]**2** was more specific for P-gp, despite its fast rate of metabolism. As the rate of metabolism is

species dependent further studies in humans are needed to assess the potential of [^{18}F]**2** as a clinical PET tracer.

Methods

General

Chemicals and solvents were purchased from commercial sources, Sigma-Aldrich (Zwijndrecht, the Netherlands), Fluorochem (Hadfield Derbyshire, UK), ABCr GmbH (Karlsruhe, Germany) and Biosolve (Valkenswaard, the Netherlands), without further purification unless stated otherwise. (*R*)-desmethyl-verapamil was kindly donated by Abbott Laboratories (IL, USA). Dichloromethane (DCM), dichloroethane (DCE), methanol (MeOH) and dimethylformamide (DMF) were dried over 3 Å molecular sieves for at least 24 h prior to use. Tetrahydrofuran (THF) was first distilled from LiAlH_4 and then dried over 3 Å molecular sieves. Thin layer chromatography (TLC) was performed on Merck (Darmstadt, Germany) precoated silica gel 60 F254 plates. Spots were visualized by UV quenching or ninhydrin. Column chromatography was carried out either manually by using silica gel 60 Å (Sigma-Aldrich) or on a Buchi (Flawil, Switzerland) sepacore system (comprising of a C-620 control unit, a C-660 fraction collector, 2 C601 gradient pumps and a C640 UV detector) equipped with Buchi sepacore preppacked flash columns. ^1H and ^{13}C nuclear magnetic resonance (NMR) spectra were recorded on a Bruker (Billerica, USA) Avance 500 (500.23 MHz and 125.78 MHz, respectively) with chemical shifts (δ) reported in ppm relative to the solvent. Electrospray ionization mass spectrometry (ESI-MS) was carried out using a Bruker microTOF-Q instrument in positive ion mode (capillary potential of 4500 V).

QC analysis was performed using an HPLC system of Jasco (Easton, MD, USA) containing a PU-2089 pump station equipped with a Varian Kromasil C18 column (10 μm , 250 \times 4.6 mm, EKA Chemicals AB or AkzoNobel, Sweden) using $\text{H}_2\text{O}/\text{MeCN}/\text{DIPA}$ (40:60:0.2, v/v/v, method: A) as eluent, or Grace Alltima C18 column (5 μm , 250 mm \times 4.6 mm, Grace, Columbia, USA) using $\text{H}_2\text{O}/\text{MeCN}/\text{DIPA}$ (40:60:0.2, v/v/v, method: B) or $\text{H}_2\text{O}/\text{MeCN}/\text{TFA}$ (60:40:0.2, v/v/v, method: C) or $\text{H}_2\text{O}/\text{MeCN}/\text{TFA}$ (30:70:0.2, v/v/v, method: D (intermediate)) as eluent at a flow rate of 1 $\text{mL}\cdot\text{min}^{-1}$, with a Jasco UV-2075 UV detector ($\lambda=232$ nm) and a NaI radioactivity detector (Raytest, Straubenhardt, Germany). Chromatograms were acquired using Raytest GINA Star software (version 5.01). Semi-preparative HPLC was performed on a Jasco PU-2089 pump station equipped with Luna C18(2) column (10 μm , 250 mm \times 10 mm, Phenomenex, California, USA) using 5 mM $\text{K}_3\text{PO}_4/\text{MeCN}$ (28:72, v/v, pH=10.0 method: E) or

H₂O/MeCN/TFA (60:40:0.2, v/v/v, method: F) as eluent at a flow rate of 4 mL·min⁻¹, a Jasco UV-1575 Plus UV detector (λ =254 nm), a custom made radioactivity detector and Jasco ChromNAV CFR software (version 1.14.01) for data acquisition.

Metabolite analysis was performed on Dionex (Sunnyvale, CA, USA) UltiMate 3000 HPLC equipment with Chromeleon software (version 6.8). A LUNA C8 (5 μ m, 250 mm x 10 mm, Phenomenex) column was used (method F) using 5 mM NH₄OAc/MeCN (1:1, v/v, pH=4.2) as eluent at a flow rate of 3.5 mL/min.

Chemistry

(R)-2-(3,4-dimethoxyphenyl)-2-(((R)-2,2-dimethyl-1,3-dioxolan-4-yl)methyl)-3-methylbutanenitrile (10)

To a stirred solution of 2-(3,4-dimethoxyphenyl)-3-methylbutanenitrile (**9**) (3.83 g, 17.5 mmol) in 60 mL dry DCM was added sodium hydride (60% dispersion in mineral oil; 1.40 g, 34.9 mmol) and heated to 65°C. After 3 h the reaction mixture was cooled to room temperature, (R)-(2,2-dimethyl-1,3-dioxolan-4-yl)methyl 4-methylbenzenesulfonate (5.00 g, 17.5 mmol) dissolved in 5 mL dry THF was added and the reaction mixture was stirred at 65°C overnight. The dark brown mixture was quenched with water, extracted into diethyl ether, organic layers were washed with water and brine, dried over Na₂SO₄ and the solvent was evaporated in vacuo. The crude product was purified by flash column chromatography (10-30% EtOAc/hexane) to yield the R,R-diastereomer **10** (3.96 g, 11.9 mmol, 68.1% yield) as a light brown oil. ¹H NMR (CDCl₃) δ 6.95-6.83 [3H, m, CH_{AR}], 4.03-3.92 [1H, m, CHO], 3.89 [3H, s, CH₃O], 3.90 [3H, s, CH₃O], 3.19 [1H, dd, *J* = 5.6 and 8.3 Hz, CH₂O], 2.98 [1H, t, *J* = 7.9 Hz, CH₂O], 2.65 [1H, dd, *J* = 4.6 and 13.8 Hz, CH₂CHO], 2.15 [1H, sept, *J* = 6.7 Hz, CH(CH₃)₂], 1.90 [1H, dd, *J* = 7.7 and 13.6 Hz, CH₂CHO], 1.36 and 1.26 [3H each, s, C(CH₃)₂], 1.23 and 0.79 [3H each, d, *J* = 6.7 Hz, CH(CH₃)₂]; ¹³C NMR (CDCl₃) δ 149.12, 148.71, 129.9, 120.47, 118.6, 111.19, 109.53, 108.14, 73.7, 69.34, 56.06, 55.92, 51.1, 42.38, 37.92, 26.78, 25.72, 18.6, 18.52; ESI-HRMS: calculated for C₁₉H₂₇NO₄: 333.1940; found 334.2058 [M + H]⁺, 356.1898 [M+Na]⁺.

(R)-2-(3,4-dimethoxyphenyl)-2-isopropyl-4-oxobutanenitrile (11)

(R)-2-(3,4-dimethoxyphenyl)-2-(((R)-2,2-dimethyl-1,3-dioxolan-4-yl)methyl)-3-methylbutanenitrile **10** (847 mg, 2.54 mmol) was dissolved in water (4 ml) and acetic acid (12 ml) and stirred at room temperature overnight. The volume was reduced by evaporation, giving the crude diol as a colourless

oil. The crude reaction mixture was dissolved in DCM (35 mL), and 1M NaHCO₃ (16.9 mL, 16.9 mmol) was added. To the stirred biphasic system was added a solution of sodium periodate (2.20 g, 10.3 mmol) in water (11 mL) dropwise over 2 hours. The white suspension was stirred for another 2 hours. The suspension was diluted with DCM and the organic phase was separated, washed with water and dried over Na₂SO₄ and the solvent was evaporated in vacuo. The resulting brown oil was purified by flash column chromatography (25% EtOAc/Hex) to obtain (*R*)-2-(3,4-dimethoxyphenyl)-2-isopropyl-4-oxobutanenitrile **11** (434 mg, 1.66 mmol, 65.4% yield) as a colorless oil. ¹H NMR (CDCl₃) δ 6.93-6.84 [3H, m, CH_{AR}], 3.90 [3H, s, OCH₃], 3.88 [3H, s, OCH₃], 3.14-2.87 [2H, m, CH₂CHO], 2.14 [1H, sept, *J* = 6.7 Hz, CH(CH₃)₂], 1.18 and 0.89 [3H each, d, *J* = 6.7 Hz, CH(CH₃)₂]; ¹³C NMR (CDCl₃) δ 149.35, 148.93, 128.75, 120.45, 118.81, 111.35, 109.70, 56.09, 55.95, 49.80, 48.66, 38.19, 18.43, 18.22 ESI-HRMS: calculated for C₁₅H₁₉NO₃: 261.1365; 262.1458 [M + H]⁺ and 284.1285 [M + Na]⁺ found.

(*R*)-2-(3,4-dimethoxyphenyl)-2-isopropyl-5-oxopentanenitrile (12)

(Methoxymethyl)triphenylphosphonium chloride (3.33 g, 9.71 mmol) was suspended in THF (50 mL), brought to -50 °C, and *n*-butyllithium (1.6 M in hexanes; 12.1 mL, 19.4 mmol) was added dropwise via a dropping funnel over 30 min, stirring for 1 hour at -80 °C. To the resulting dark red solution, (*R*)-2-(3,4-dimethoxyphenyl)-2-isopropyl-4-oxobutanenitrile **11** (705 mg, 2.70 mmol) dissolved in 4 mL THF was added dropwise over 15 min and stirred for 2 hours at -80 °C and was slowly brought to r.t. The reaction was quenched with water and extracted with Et₂O. The organic layers were washed with water and brine, dried over Na₂SO₄ and the solvent was evaporated in vacuo. The crude oil was purified by flash column chromatography (5-20% EtOAc/Hex) to obtain a mixture of the *E/Z*-isomers of the desired intermediate (356 mg, 1.23 mmol, 45.6% yield). To a solution of (*R*)-2-(3,4-dimethoxyphenyl)-2-isopropyl-5-methoxypent-4-enenitrile (356 mg, 1.23 mmol) in 2-propanol (3 mL) and water (3 mL), 4-methylbenzenesulfonic acid hydrate (14.0 mg, 0.074 mmol) was added and stirred to reflux for 3 hours. The reaction was quenched with water and extracted with Et₂O. The organic layers were washed with 1M NaHCO₃ and brine, dried over Na₂SO₄ and the solvent was evaporated in vacuo. The crude oil was purified with flash column chromatography (10% EA in hexane) obtaining the preferred product **12** (95 mg, 0.35 mmol, 28% yield) ¹H NMR (CDCl₃) δ 6.94-6.82 [3H, m, CH_{AR}], 3.89 [6H, m, (OCH₃)₂], 2.70-2.40 [2H, m, CH₂CH₂CHO], 2.24-2.06 [3H, m, CH(CH₃)₂ and CH₂CHO], 1.22 and 0.81 [3H each, d, *J* = 6.7 Hz, CH(CH₃)₂]; ¹³C NMR (CDCl₃) δ 200.44, 149.22, 148.58, 129.58, 120.80, 118.69, 111.19, 109.21, 56.03, 18

55.93, 52.67, 40.60, 38.00, 29.92, 19.06, 18.60 ESI-HRMS: calculated for $C_{16}H_{21}NO_3$: 275.1521: impossible to ionize.

***tert*-butyl 4-hydroxy-3-methoxyphenethylcarbamate (**4**)**

Triethylamine (0.42 mL, 3.0 mmol) and di-*tert*-butyl dicarbonate (0.69 mL, 3.0 mmol) dissolved in 3 mL methanol were added to a solution of 4-(2-aminoethyl)-2-methoxyphenol (**3**) (500 mg, 2.99 mmol) in 10 mL methanol and stirred for 2 hours at 50 °C. The solvent was evaporated in vacuo and the crude product was extracted with EtOAc, washed with water and brine, dried over Na_2SO_4 , and the solvent was evaporated in vacuo. The white crystallized solid obtained was purified using flash column chromatography (1-5% MeOH/DCM) resulting in the desired product as a clear oil (460 mg, 1.72 mmol, 57.5% yield) 1H NMR ($CDCl_3$) δ 6.87-6.66 [3H, m, CH_{AR}], 4.54 [1H, bs, OH], 3.88 [3H, s, OCH_3], 3.34 [2H, q, J = 6.4 Hz, $NHCH_2$], 2.72 [2H, t, J = 7.1 Hz, $NHCH_2CH_2$], 1.44 [9H, s, $C(CH_3)_3$]; ^{13}C NMR ($CDCl_3$) δ (ppm) 158.49, 148.9, 145.96, 132.15, 122.32, 116.21, 113.47, 80.02, 56.39, 43.4, 36.89, 28.92; ESI-HRMS: calculated for $C_{14}H_{21}NO_4$: 267.1471; 290.1328 $[M + Na]^+$ found.

***tert*-butyl 4-(2-fluoroethoxy)-3-methoxyphenethylcarbamate (**6**)**

tert-butyl 4-hydroxy-3-methoxyphenethylcarbamate (**4**) (309 mg, 1.16 mmol), 1-bromo-2-fluoroethane (0.345 mL, 4.62 mmol), K_2CO_3 (639 mg, 4.62 mmol) and potassium iodide (768 mg, 4.62 mmol) were stirred overnight at 75 °C in 8 mL dry DMF. The reaction was quenched with water, extracted twice with ethylacetate, washed with brine, dried over Na_2SO_4 , and the solvent was evaporated in vacuo. The crude yellow oil was purified using flash column chromatography (15-25% EtOAc/hex) and the desired product was obtained (170 mg, 0.542 mmol, 46.9 % yield) as a yellow oil. 1H NMR ($CDCl_3$) δ 6.87-6.69 [3H, m, CH_{AR}], 4.76 [2H, dt, J = 47 and 4.1 Hz, (OCH_2CH_2F)], 4.20 [2H, dt, J = 27 and 4.3 Hz, (OCH_2CH_2F)], 3.86 [3H, s, OCH_3], 3.35 [2H, q, J = 6.4 Hz, $NHCH_2$], 2.74 [2H, t, J = 7.1 Hz, $NHCH_2CH_2$], 1.43 [9H, s, $C(CH_3)_3$]; ^{13}C NMR ($CDCl_3$) δ 155.84, 149.78, 146.4, 132.88, 120.71, 114.61, 112.65, 81.32, 68.7, 68.54, 55.91, 41.81, 35.77, 28.39; ESI-HRMS: calculated for $C_{16}H_{24}FNO_4$: 313.1689; 314.1678 $[M + H]^+$, 336.1504 $[M + Na]^+$ found

2-(4-(2-fluoroethoxy)-3-methoxyphenyl)ethanamine (8)

tert-butyl 4-(2-fluoroethoxy)-3-methoxyphenethylcarbamate (**6**) (200 mg, 0.638 mmol) was dissolved in 4 mL DCM and 4 mL TFA and stirred at room temperature for 3 h. The solvent was evaporated in vacuo and the crude product was purified using flash column chromatography (15-20% MeOH/DCM) to obtain the desired product (125 mg, 0.586 mmol, 92.0%) as white crystals. ^1H NMR ($\text{DMSO}-d_6$) δ 6.93-6.76 [3H, m, CH_{AR}], 4.71 [2H, dt, J = 48 and 3.8 Hz, CH_2F], 4.17 [2H, dt, J = 30 and 3.9 Hz, OCH_2], 3.77 [3H, s, CH_3], 3.03 [2H, t, J = 6.7 Hz, H_2NCH_2], 2.77 [2H, t, J = 6.7 Hz, $\text{H}_2\text{NCH}_2\text{CH}_2$]; ^{13}C NMR (CDCl_3) δ 149.82, 146.84, 129.87, 120.63, 114.61, 112.47, 82.47, 81.11, 55.69, 40.62, 32.90; ESI-HRMS: calculated for $\text{C}_{11}\text{H}_{16}\text{FNO}_2$: 213.1165; 214.1238 $[\text{M} + \text{H}]^+$ found.

2-(4-(2-((tert-butoxycarbonyl)amino)ethyl)-2-methoxyphenoxy)ethyl 4-methylbenzenesulfonate (5)

tert-butyl 4-hydroxy-3-methoxyphenethylcarbamate (**4**) (613 mg, 2.29 mmol) was dissolved in 15 mL DMF, and ethane-1,2-diyl bis(4-methylbenzenesulfonate) (849 mg, 2.29 mmol) and cesium carbonate (149 mg, 4.59 mmol) were added. The reaction mixture was stirred at room temperature for 3 h and quenched with water, extracted with EtOAc, washed with brine and dried over Na_2SO_4 . The solvent was removed in vacuo and crude product was purified using flash column chromatography (15-35% EtOAc/hex) to obtain the desired product as a white solid (440 mg, 0.945 mmol, 41% yield). ^1H NMR (CDCl_3) δ 7.82 [2H, d, J = 8.2 Hz, OTs], 7.33 [2H, d, J = 8.2 Hz, OTs], 6.76-6.68 [3H, m, CH_{AR}], 4.35 [2H, t, J = 5.0 Hz, ($\text{OCH}_2\text{CH}_2\text{OTs}$)], 4.19 [2H, t, J = 5.0 Hz, ($\text{OCH}_2\text{CH}_2\text{OTs}$)], 3.81 [3H, s, OCH_3], 3.33 [2H, q, J = 6.6 Hz, NHCH_2], 2.72 [2H, t, J = 6.8 Hz, NHCH_2CH_2], 2.44 [3H, s, $\text{SC}_6\text{H}_4\text{CH}_3$], 1.43 [9H, s, $\text{C}(\text{CH}_3)_3$]; ^{13}C NMR (CDCl_3) δ 155.83, 149.82, 145.97, 144.85, 133.26, 132.78, 129.81, 128.01, 120.75, 115.21, 112.71, 68.1, 67.12, 60.39, 55.86, 41.81, 35.77, 28.39, 21.66; ESI-HRMS: calculated for $\text{C}_{23}\text{H}_{31}\text{NO}_7\text{S}$: 465.1821; 488.1719 $[\text{M} + \text{Na}]^+$ found.

2-(4-(2-aminoethyl)-2-methoxyphenoxy)ethyl 4-methylbenzenesulfonate (7)

2-(4-(2-((tert-butoxycarbonyl)amino)ethyl)-2-methoxyphenoxy)ethyl 4-methylbenzenesulfonate (440 mg, 0.945 mmol) was dissolved in 5 mL DCM and 5 mL TFA and stirred for 30 min at room temperature. The solvent was evaporated and the crude product was purified on flash column chromatography (5-20

10% MeOH/DCM) to obtain the purified product 2-(4-(2-aminoethyl)-2-methoxyphenoxy)ethyl 4-methylbenzenesulfonate (350 mg, 0.958 mmol, quantitative yield) as a white powder. ^1H NMR (MeOD) δ 7.81 [2H, d, J = 8.4 Hz, OTs], 7.42 [2H, d, J = 8.4 Hz, OTs], 6.88-6.74 [3H, m, CH_{AR}], 4.32 [2H, m, ($\text{OCH}_2\text{CH}_2\text{OTs}$)], 4.15 [2H, m, ($\text{OCH}_2\text{CH}_2\text{OTs}$)], 3.82 [3H, s, OCH_3], 3.15 [2H, q, J = 6.6 Hz, NHCH_2], 2.88 [2H, t, J = 7.1 Hz, NHCH_2CH_2], 2.45 [3H, s, $\text{SC}_6\text{H}_4\text{CH}_3$]; ^{13}C NMR (MeOD) δ 151.78, 148.41, 146.64, 134.46, 132.07, 131.19, 129.21, 122.25, 116.79, 114.32, 70.33, 68.66, 56.66, 42.12, 34.32, 21.74; ESI-HRMS: calculated for $\text{C}_{18}\text{H}_{23}\text{NO}_5\text{S}$: 365.1297; 366.1435 $[\text{M} + \text{H}]^+$ found.

(R)-2-(4-(2-((tert-butoxycarbonyl)(4-cyano-4-(3,4-dimethoxyphenyl)-5-methylhexyl)amino)ethyl)-2-methoxyphenoxy)ethyl 4-methylbenzenesulfonate (13)

Na_2SO_4 and (R)-2-(3,4-dimethoxyphenyl)-2-isopropyl-5-oxopentanenitrile (**12**) (151 mg, 0.547 mmol) in 2.5 mL DCE were added to a solution of 2-(4-(2-aminoethyl)-2-methoxyphenoxy)ethyl 4-methylbenzenesulfonate (**7**) (300 mg, 0.821 mmol) in 2.5 mL DCE. The reaction mixture was stirred at room temperature overnight under argon. Sodium triacetoxhydroborate (174 mg, 0.821 mmol) was added to the mixture and stirred for 1.5 hour at room temperature. The reaction was quenched with 1M NaHCO_3 , extracted with EtOAc (15 mL), washed with water (2x) and brine, and organic layers were dried over Na_2SO_4 and used as such in the next step. Di-tert-butyl dicarbonate (239 mg, 1.095 mmol) and triethylamine (152 μL , 1.09 mmol) were added to the diluted reaction mixture and stirred at room temperature for 1.5 h. The reaction mixture was diluted with EtOAc, washed with water and brine, dried over Na_2SO_4 , and solvent was removed in vacuo. The crude mixture was purified by flash column chromatography (30-50% EtOAc/hexane) to obtain the purified product as colorless oil (174 mg, 0.240 mmol, 43.9 % yield). ^1H NMR (CDCl_3) δ 7.82 (2H, d, J = 8.0 Hz, OTs), 7.33 (2H, d, J = 8.0 Hz, OTs), 6.89-6.61 (6H, m, CH_{AR}), 4.34 [2H, t, J = 4.9 Hz, ($\text{OCH}_2\text{CH}_2\text{OTs}$)], 4.18 [2H, t, J = 5.0 Hz, ($\text{OCH}_2\text{CH}_2\text{OTs}$)], 3.88 (3H, s, OCH_3), 3.86 (3H, s, OCH_3), 3.79 (3H, s, OCH_3), 3.25-2.99 (4H, m, CH_2NCH_2), 2.67 (2H, t, J = 6.5 Hz, $\text{NCH}_2\text{CH}_2\text{Ar}$), 2.44 (3H, s, TsCH_3), 2.05 (2H, m, $\text{CH}_2\text{CH}_2\text{CH}_2$), 2.04-1.54 (3H, m, $\text{CH}(\text{CH}_3)_2$ and CCH_2), 1.42 (9H, m, Boc), 1.17 and 0.78 (3H each, d, J = 6.7 Hz, $\text{CH}(\text{CH}_3)_2$); ^{13}C NMR (CDCl_3) δ (ppm) 149.69, 148.93, 148.2, 145.85, 145.78, 144.82, 133.42, 132.71, 130.34, 129.78, 127.95, 121.31, 120.73, 118.63, 115.18, 112.75, 110.95, 109.24, 79.35, 68.09, 67.07, 55.89, 55.81, 55.79, 49.06, 48.66, 47.27, 37.84, 35.13, 34.77, 28.34, 24.58, 21.61, 18.89, 18.50; ESI-HRMS: calculated for $\text{C}_{39}\text{H}_{52}\text{N}_2\text{O}_9\text{S}$: 724.3394; 725.3477 $[\text{M} + \text{H}]^+$ and 747.3318 $[\text{M} + \text{Na}]^+$ found.

(*R*)-2-(3,4-dimethoxyphenyl)-5-((4-(2-fluoroethoxy)-3-methoxyphenethyl)amino)-2-isopropylpentanenitrile (2)

To a solution of (*R*)-2-(3,4-dimethoxyphenyl)-2-isopropyl-5-oxopentanenitrile (**12**) (108 mg, 0.392 mmol) in 5 mL dry MeOH were added Na₂SO₄ (167 mg, 1.18 mmol) and 2-(4-(2-fluoroethoxy)-3-methoxyphenyl)ethanamine (**8**) (125 mg, 0.588 mmol) under argon and stirred overnight at room temperature. Sodium triacetoxhydroborate (125 mg, 0.588 mmol) was added and stirred for 2 hours at room temperature under argon. The reaction was quenched with saturated NaHCO₃, diluted with ether, and the crude mixture was filtered. The filtrate was washed with brine, dried over Na₂SO₄ and the solvent was removed in vacuo. The crude product was purified by flash column chromatography (1-10% MeOH/DCM) to obtain the desired product as an oil. (20 mg, 0.042 mmol, 10 % yield) ¹H NMR (CDCl₃) δ 6.93-6.66 [6H, m, CH_{AR}], 4.74 [2H, m, CH₂CH₂F], 4.22 [2H, m, CH₂CH₂F], 3.87, 3.85 and 3.82 [9H, 3xs, OCH₃], 2.80 [4H, m, CH₂NHCH₂CH₂], 2.67 [2H, m, NHCH₂CH₂], 2.06 [1H, m, CH(CH₃)₂], 2.16 and 1.88 [1H each, dt, J = 4.6 and 13 Hz, CCH₂], 1.61 and 1.23 [1H each, m, CH₂CH₂CH₂], 1.15 and 0.76 [3H each, d, J = 7 Hz, CH(CH₃)₂]; ¹³C NMR (CDCl₃) δ (ppm) 149.74, 148.96, 148.23, 146.41, 132.52, 130.13, 121.2, 120.52, 118.59, 114.51, 112.53, 110.97, 109.31, 82.61, 81.25, 68.61, 68.44, 55.94, 55.88, 55.79, 49.90, 48.35, 37.85, 35.21, 34.42, 24.58, 18.86, 18.53; ESI-HRMS: calculated for C₂₇H₃₇FN₂O₄: 472.2737; [M + H]⁺ 473.2745 found.

(*R*)-5-((3,4-dimethoxyphenethyl)(2-fluoroethyl)amino)-2-(3,4-dimethoxyphenyl)-2-isopropylpentanenitrile (1)

To a solution of (*R*)-5-((3,4-dimethoxyphenethyl)amino)-2-(3,4-dimethoxyphenyl)-2-isopropylpentanenitrile (74 mg, 0.17 mmol) in 2 mL DMF, 1-bromo-2-fluoroethane (500 μL, 6.72 mmol) and potassium carbonate (255 mg, 1.85 mmol) were added and stirred at 65°C for 4 h. The reaction mixture was extracted with EtOAc, washed with water and brine, dried over Na₂SO₄, and the solvent was removed in vacuo. The crude product was purified by flash column chromatography (0-1% MeOH/DCM) to obtain the desired product as an oil. (40 mg, 0.082 mmol, 49 % yield) ¹H NMR (CDCl₃) δ 6.91-6.66 [6H, m, CH_{AR}], 4.46 [2H, m, CH₂CH₂F], 3.89, 3.88, 3.86 and 3.85 [12H, 4xs, OCH₃], 2.81-2.62 [6H, m, CH₂N(CH₂)CH₂CH₂], 2.54 [2H, m, NCH₂CH₂], 2.05 [1H, m, CH(CH₃)₂], 2.12 and 1.85 [1H each, dt, J = 4.3 and 13 Hz, CCH₂], 1.52 and 1.15 [1H each, m, CH₂CH₂CH₂], 1.18 and 0.79 [3H each, d, J = 6.7 Hz, CH(CH₃)₂]; ¹³C

NMR (CDCl₃) δ (ppm) 148.96, 148.78, 148.23, 147.28, 132.8, 130.6, 121.47, 120.5, 118.64, 112.04, 111.17, 111.02, 109.57, 83.25, 81.92, 56.36, 55.95, 55.88, 55.84, 55.80, 53.92, 53.70, 53.3, 37.93, 35.41, 32.98, 23.35, 18.91, 18.57; ESI-HRMS: calculated for C₂₈H₃₉FN₂O₄: 486.2894; 487.2975 [M + H]⁺ found.

Radiochemistry:

(R)-5-((3,4-dimethoxyphenethyl)(2-[¹⁸F]fluoroethyl)amino)-2-(3,4-dimethoxyphenyl)-2-isopropylpentanenitrile ([¹⁸F]1)

[¹⁸F]F⁻ was produced by the ¹⁸O(p,n)¹⁸F nuclear reaction using an IBA (Louvain-la-Neuve, Belgium) Cyclone 18/9 cyclotron. Radioactivity levels were measured using a Veenstra (Joure, The Netherlands) VDC-405 dose calibrator. Radiochemistry was carried out in homemade, remotely controlled synthesis units.²⁹ After irradiation, [¹⁸F]fluoride was trapped on a PS-HCO₃ column and eluted with 1 mL MeCN/H₂O (9:1, v/v) containing 13 mg (35 μ mol) Kryptofix 2.2.2 and 2 mg (14 μ mol) K₂CO₃ into a screw cap reaction vial. The [¹⁸F]K₂₂₂/KF/K₂CO₃ complex was dried at 90 °C under a Helium flow of 50 mL·min⁻¹ and reduced pressure for 6 minutes. 0.5 mL MeCN was added and the complex was dried for 3 minutes resulting in a white tarnish at the bottom of the vial. 10 mg (36 μ mol) 2-bromoethyltosylate was dissolved in 0.5 mL DMF and added to the vial containing the dried complex, and this reaction mixture was heated to 90 °C. After 10 min, the formed volatile intermediate 1-bromo-2-[¹⁸F]fluoroethane was distilled at 100 °C through a preheated silvertriflate column at 200 °C resulting in [¹⁸F]fluoroethyltriflate, which was bubbled to the second reaction vial containing a reaction mixture with 1.5 mg (3.4 μ mol) (R)-desmethyl-norverapamil, 1.5 mg (11 μ mol) K₂CO₃ and a stirring bar in 0.5 mL ACN at 0 °C (Scheme 4). After distillation, the reaction was stirred for 15 min at 120 °C, quenched with 1 mL of water, and purified by semi-preparative HPLC (method E). The product eluted at 8 min, was collected for 1.5 min and the fraction was diluted with 40 mL of water. The mixture was passed through a Sep-Pak Plus tC18 cartridge and subsequently rinsed with 20 mL water. The product was eluted from the Sep-Pak Plus tC18 cartridge with 1 mL ethanol (96%). The final product containing [¹⁸F]1 was diluted with a solution of 7.11 mM NaH₂PO₄ in 0.9% NaCl (w/v in water), pH 5.2, resulting in a final solution with 5% ethanol. The purity was > 95% and the specific activity was 143 \pm 88 GBq/ μ mol, which was derived from a calibration curve of reference compound 1 using HPLC (methods A and B). In the end, 400-1600 MBq was isolated and the overall radiochemical yield was 2.7 \pm 1.2 % DC, starting from 20-50 GBq [¹⁸F]F⁻ (n=7).

(R)-2-(3,4-dimethoxyphenyl)-5-((4-(2-[¹⁸F]fluoroethoxy)-3-methoxyphenethyl)amino)-2-isopropylpentanenitrile ([¹⁸F]2)

The [¹⁸F]K₂₂₂/KF/K₂CO₃ complex was dried as described above, 1.0 mg of precursor **13** in 0.5 mL MeCN was added to the reaction vial, and heated at 90 °C for 15 min. The reaction mixture was cooled down to room temperature and 0.2 mL TFA was added. After 10 min, the reaction was quenched with 0.9 mL of 2.5 M NaOH and purified by semi-preparative HPLC (method F). The product eluted at 15 min, was collected for 1.5 min and fraction was diluted with 40 mL of water. The mixture was passed through the Sep-Pak Plus tC18 cartridge and subsequently rinsed with 20 mL water. The product [¹⁸F]2 was eluted with 1 mL ethanol (96%) and diluted with a solution of 7.11 mM NaH₂PO₄ in 0.9% NaCl (w/v in water), pH 5.2, resulting in a final solution with 5% ethanol, with a purity of > 95%. The specific activity was 151 ± 74 GBq/μmol, which was derived from a calibration curve of reference compound **2** using HPLC (method C). At the end of the synthesis 300-8000 MBq was isolated and the overall radiochemical yield was 17.2 ± 9.9 % DC, starting from 15-50 GBq [¹⁸F]F⁻ (n=7).

(R)-[¹¹C]verapamil

Radiolabeling of (R)-[¹¹C]verapamil was performed as described previously.³⁰ The product was purified by semi-preparative HPLC (Method E) with a retention time of 10 min. The collected HPLC fraction was diluted with 40 ml of water and the mixture was passed through a Sep-Pak Plus tC18 cartridge and subsequently rinsed with 20 mL water. The product was eluted with 1 mL ethanol (96%) and diluted with a solution of 7.11 mM NaH₂PO₄ in 0.9% NaCl (w/v in water), pH 5.2, resulting in a final solution with 5% ethanol, with a purity of > 95% and a specific activity of 118 GBq/μmol (n=1) as determined on HPLC (Method B). At the end of the synthesis, 5.2 GBq was isolated, overall radiochemical yield was 28.2 % DC, starting from 60 GBq [¹¹C]CO₂ (n=1).

Log D_{7.4}

The distribution of [¹⁸F]**1** and [¹⁸F]**2** between equal volumes of 0.2 M phosphate buffer (pH 7.4) and 1-octanol was measured in triplicate at room temperature. 1 mL of a 1-5 MBq/mL solution of [¹⁸F]**1** or [¹⁸F]**2** in 0.2 M phosphate buffer (pH 7.4) was vigorously mixed with 1 mL of 1-octanol for 1 min at room temperature using a vortex. After 30 min, five samples of 100 μL were taken from both layers, avoiding cross-contamination. To determine recovery, 5 samples of 100 μL were taken from the 1-5 MBq/mL solution. All samples were counted for radioactivity. The log D_{oct,7.4} value was calculated according to log

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$D_{\text{Oct},7.4} = {}^{10}\log(A_{\text{Oct}}/A_{\text{buffer}})$, where A_{Oct} and A_{buffer} represent average radioactivity of 5 1-octanol and 5 buffer samples, respectively.³¹

Animals

Healthy male Wistar rats were obtained from Harlan Netherlands B.V. (Horst, the Netherlands) and male wild-type mice, *Mdr1a/b*^(-/-) mice, and *Bcrp1*^(-/-) mice developed from the FVB line were purchased from Taconic (Hudson, USA). All animals were housed in groups of four to six per cage until treatment and were kept at room temperature of 20-24 °C with a relative humidity of 50-70% and at a 12 h light/dark cycle. Animals had unrestricted access to food (Teklad Global 16% Protein Rodent Diet, Harlan, Madison, WI, USA) and tap water. All animal experiments were performed in compliance with Dutch laws on animal experimentation and after approval by the local animal ethics committee.

Biodistribution

Healthy Wistar rats (218-244 g) were injected with 50 ± 2 MBq of either [¹⁸F]**1** or [¹⁸F]**2** in the tail vein under isoflurane anesthesia (2% in O₂ at 1 L·min⁻¹). Rats were conscious for the allowed time between injection and sacrificing, except for the animals of the 5 minute time point, which were left unconscious for the whole time. Animals were sacrificed under isoflurane anesthesia at time points 5, 15 or 60 minutes (n=3). Blood was collected via a heart puncture from each rat. Heart, lungs, liver, kidneys, bone, cerebral cortex, cerebellum and the rest of the brain were collected, weighed and counted for radioactivity in a Wallac Universal Gamma Counter 1282 (PerkinElmer, Waltham, MA, USA). Biodistribution data were expressed as percentage injected dose per gram tissue (%ID/g).

Metabolite analysis

Healthy Wistar rats (230-291 g) were injected with 28 ± 4 or 43 ± 24 MBq of [¹⁸F]**1** or [¹⁸F]**2**, respectively, in the tail vein under isoflurane anesthesia. After injection, rats were conscious for the allowed time and sacrificed under isoflurane anesthesia at time point 5, 15 or 60 minutes (n=3). Blood samples were collected via heart puncture, and the brain was removed from the skull and cut in half. Blood was collected in a heparin tube and centrifuged for 5 min at 4000 rpm (Hettich universal 16, Depex B.V., the Netherlands). Plasma was separated from blood cells, 1 mL was loaded onto an Sep-Pak tC18 cartridge (Waters, Etten-Leur, the Netherlands), and the cartridge was washed with 20 mL of water. This eluate was defined as the polar radiolabeled metabolite fraction. Next, the Sep-Pak cartridge was eluted with

1.5 mL of methanol. This eluate was defined as the non-polar fraction and contains the parent tracer. This eluate was analyzed using HPLC (method G). The recovery from the Sep-pak procedure was >85% and rest activity was not taken into account. One half of the brain was counted for activity and the other half was homogenized with an IKA T18 ULTRA-TURRAX Basic Homogenizer (IKA, Germany) in cold H₂O/MeCN (1:1, v/v), under ice cooling, and subsequently centrifuged at 4000 rpm for 5 min. Separated supernatants were analyzed using HPLC.

PET imaging and data analysis

For all experiments, animals were anesthetized via a nose mask initially using 4% isoflurane in oxygen at a rate of 1 L/min. One hour prior to each study, a jugular vein was cannulated for administration of the radiotracer. Rats were positioned in pairs and mice were scanned in groups of 4-6 using a double LSO/LYSO layer High Resolution Research Tomograph (HRRT; Siemens/CTI, Knoxville, TN, USA).³² During scanning anesthesia was maintained using 2% isoflurane in oxygen. For each scanning session, first a transmission scan was acquired using a 740 MBq 2-dimensional (2D) fan-collimated ¹³⁷Cs (662 keV) moving point source.³³ This scan was used to correct subsequent emission scans for attenuation and scatter. As the HRRT was decommissioned, the final study (*Mdr1a/b*^(-/-) vs. WT mice of [¹⁸F]**2**) had to be performed on nanoPET/CT and nanoPET/MR scanners (Mediso Ltd., Budapest, Hungary)³⁴ with identical PET components. In this study the CT scan was used for attenuation correction and the MR scan for co-registration purposes.

For the emission scans, Wistar rats (341-450 g) were injected with 9.9 ± 3.2 MBq of [¹⁸F]**1** (n=4) or 9.9 ± 0.6 MBq of [¹⁸F]**2** (n=6) and scanned for 1 hour. Next, a second emission scan (30 minutes) following injection of 9.0 ± 1.7 MBq of [¹⁸F]NaF was performed to delineate bone. The following day, the same rats received an intravenous bolus injection of 15 mg/kg (3.5 mg/mL) tariquidar, dissolved in a vehicle consisting of 5% glucose in saline. 30-40 minutes after tariquidar administration, rats were injected with 9.8 ± 1.4 MBq of [¹⁸F]**1** or 8.7 ± 0.5 MBq of [¹⁸F]**2**. Again, this was followed by an [¹⁸F]NaF (7.9 ± 3.5 MBq) scan. [¹⁸F]**1** and [¹⁸F]**2** scans were acquired in list mode and rebinned into the following frame sequence: 10x60, 4x300 and 3x600 s. The [¹⁸F]NaF scans were processed as a single static scan.

Mice (28.3-39.7 g) were injected with 3.6 ± 0.6 MBq of [¹⁸F]**1** or 3.8 ± 0.4 MBq of [¹⁸F]**2** and scanned for 60 minutes. Emission scans were acquired in list mode and rebinned into the following frame sequence:

7x10, 1x20, 3x30, 2x60, 2x150, 4x300 and 3x600 s for all HRRT scans. For the nanoPET scans, the following frame sequence was used: 4x5, 4x10, 2x30, 3x60, 2x300, 3x600 and 1x 900 s.

Following corrections for decay, dead time, scatter and randoms, HRRT emission scans were reconstructed using an iterative 3D-ordered subsets weighted least-squares algorithm (3D-OSWLS).³³ The point source resolution varied across the field of view from approximately 2.3 to 3.2 mm full width at half maximum in the transaxial direction and from 2.5 to 3.4 mm in the axial direction.

Reconstruction of the nanoPET emission scans was performed using an iterative 3D Poisson ordered-subsets expectation-maximization algorithm (Tera-Tomo; Mediso Ltd.³⁴) with 4 iterations and 6 subsets, resulting in an isotropic 0.4 mm voxel dimension.

PET images were analyzed using the freely available AMIDE software (version 0.9.2).³⁵ An MR based rat brain atlas was used to define a whole brain region of interest in rats. For each rat, this validated MR atlas was aligned visually with the [¹⁸F]NaF image using a procedure described previously to obtain a corresponding whole brain region of interest (ROI).³⁶ Next, these ROIs were projected onto the dynamic [¹⁸F]**1** and [¹⁸F]**2** image sequence, generating whole brain time-activity curves (TACs). For mice, ellipsoidal shaped ROIs were drawn manually over the brain. Again, these ROIs were projected onto the dynamic image sequences, generating whole brain TACs. All TACs were expressed as standardized uptake values (SUV), i.e. mean ROI radioactivity concentration normalized to injected dose and body weight. In addition, an ellipsoid was drawn over the complete animal (excluding the cannula) to obtain the image derived percentage injected dose per cc (%ID/cc).

Finally, ellipsoid shaped ROIs manually drawn over the left ventricle were used to obtain a blood curve. The area under the curve (AUC) of the blood and brain curves were determined by Graphpad PRISM (v 5.02, Graphpad Software Inc) and brain-to-blood AUC ratios were derived from these values.

Author Contributions

R.M.R. was involved with all aspects of the studies and performed (radio)synthesis of compounds. E.J.M.K., R.C.S. assisted with the metabolite and PET studies. R.M.R., G.L., P.H.E., A.A.L., A.D.W. designed the studies, performed analysis of the data, and wrote the manuscript.

Conflict of Interest

None of the authors had any conflict of interest.

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