

Article

Divergent Asymmetric Total Synthesis of All Four Pestalotin Diastereomers from (*R*)-Glycidol

Mizuki Moriyama, Kohei Nakata, Tetsuya Fujiwara and Yoo Tanabe * 

Department of Chemistry, School of Science and Technology, Kwansai Gakuin University, 2-1 Gakuen, Sanda, Hyogo 669-1337, Japan; dbe04644@kwansai.ac.jp (M.M.); knakata01@okuno.co.jp (K.N.); tt_fujiwara@jp.daicel.com (T.F.)

* Correspondence: tanabe@kwansai.ac.jp; Tel.: +81-79-565-8394

Academic Editor: Rafael Chinchilla

Received: 28 December 2019; Accepted: 13 January 2020; Published: 17 January 2020



Abstract: All four chiral pestalotin diastereomers were synthesized in a straightforward and divergent manner from common (*R*)-glycidol. Catalytic asymmetric Mukaiyama aldol reactions of readily-available bis(TMSO)diene (Chan's diene) with (*S*)-2-benzyloxyhexanal derived from (*R*)-glycidol produced a *syn*-aldol adduct with high diastereoselectivity and enantioselectivity using a Ti(*i*OPr)₄/*(S)*-BINOL/LiCl catalyst. Diastereoselective Mukaiyama aldol reactions mediated by catalytic achiral Lewis acids directly produced not only a (1'*S*,6*S*)-pyrone precursor via the *syn*-aldol adduct using TiCl₄, but also (1'*S*,6*R*)-pyrone precursor via the anti-aldol adduct using ZrCl₄, in a stereocomplementary manner. A Hetero-Diels-Alder reaction of similarly available mono(TMSO)diene (Brassard's diene) with (*S*)-2-benzyloxyhexanal produced the (1'*S*,6*S*)-pyrone precursor promoted by Eu(*fod*)₃ and the (1'*S*,6*R*)-pyrone precursor Et₂AlCl. Debonylation of the (1'*S*,6*S*)-precursor and the (1'*S*,6*R*)-precursor furnished natural (–)-pestalotin (99% ee, 7 steps) and unnatural (+)-epipestalotin (99% ee, 7 steps), respectively. Mitsunobu inversions of the obtained (–)-pestalotin and (+)-epipestalotin successfully produced the unnatural (+)-pestalotin (99% ee, 9 steps) and (–)-epipestalotin (99% ee, 9 steps), respectively, in a divergent manner. All four of the obtained chiral pestalotin diastereomers possessed high chemical and optical purities (optical rotations, ¹H-NMR, ¹³C-NMR, and HPLC measurements).

Keywords: asymmetric total synthesis; divergent synthesis; pyran-2-one; pestalotin; epipestalotin; asymmetric Mukaiyama aldol reaction; hetero Diels-Alder reaction; Mitsunobu inversion; Chan's diene; Brassard's diene

1. Introduction

Products possessing the 4-methoxy-5,6-dihydroxy-pyran-2-one structure are distributed in nature [1], including the (i) kavalactone series, such as kavain, methylsitan, dihydrokavain, dihydromethylsitan, etc. [2], and (ii) (–)-pestalotin [3], with the three unnatural diastereomers of (–)-epipestalotin, (+)-pestalotin, and (+)-epipestalotin (Figure 1). (–)-Pestalotin was isolated from *Pesalotia cryptomeriaecola* Sawada by Kimura and Tamura's group; it possesses distinctive bioactivity as a gibberellin synergist [3–5]. Independently, the same compound was isolated from unidentified penicillium species as a minor component (code number: LLP-880α) by Ellestad's group [6].

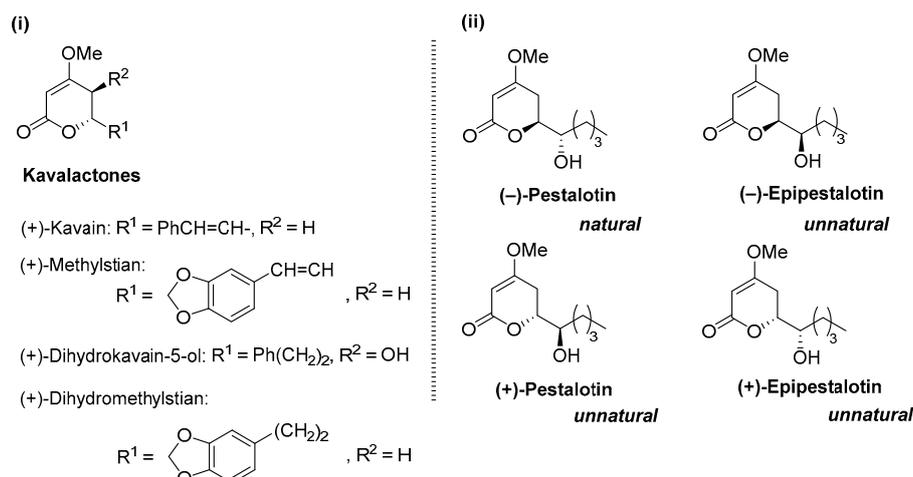


Figure 1. Natural and the related unnatural products of 4-methoxy-5,6-dihydroxy-pyran-2-one.

(-)-Pestalotin has received considerable attention as a synthetic target due to its characteristic structure, which includes two consecutive stereogenic centers. Several asymmetric total syntheses of (-)-pestalotin have therefore been performed to date, and the features are described in chronologic order of their development: (i) Dianion addition using ethyl acetoacetate with aldehyde containing a 1,3-dithian group, and successive asymmetric reduction using a chiral lithium hydro aluminate derived from chiral diamino tartrate, but with ca. 10% ee (Seebach's group) [7]; (ii) Sharpless asymmetric kinetic resolution of allyl alcohol producing (-)-pestalotin and diastereomeric (-)-epipestalotin, and chiral pool synthesis starting from glycel aldehyde acetonide derived from D-mannitol to produce antipodal (+)-pestalotin and (+)-epipestalotin (Mori's group) [8,9]; (iii) Derivatization of chiral diethyl tartarate and the incorporation of a tosyl group as a latent scaffold (Masaki's group) [10]; (iv) Asymmetric reduction using (*S*)-alpine-borane reagent of ethynyl ketone intermediate and successive hetero-Diels-Alder reaction with Brassard's siloxydiene [11] (Midland and Graham) [12]; (v) Chiral pool synthesis using unnatural (*S*)-norleucine, associated with successive syn-diastereoselective Mukaiyama aldol additions using Chan's 1,3-disiloxydiene [13] (Hagiwara's group) [14,15]; (vi) Cycloaddition strategy for chiral 1,2-diol with chiral induction utilizing Oppolzer's camphor sultum (Curran and Zhang) [16]; (vii) Sharpless asymmetric dihydroxylation of ester including a non-conjugated ene-yne precursor (Wang and Shen) [17]; and (viii) Sharpless asymmetric dihydroxylation of ethyl heptenoate and successive β -ketoester formation via Birch reduction of the *m*-methoxyphenyl ring (Rao's group) [18].

A review of these fruitful works revealed that the synthesis of all four pestalotin diastereomers is limited to the report by Mori's group [9]. The syntheses are somewhat lengthy [(-)-pestalotin: 8 steps, 4% overall yield; (-)-epipestalotin: 6 steps, 9% overall yield; (+)-pestalotin: 10 steps, 1% overall yield; (+)-epipestalotin: 10 steps, 3% overall yield], and commence with two quite different starting compounds. Nonetheless, this work contributed significantly to clarifying the stereostructure-activity relationship of these families; 1'*S* configuration in the side chain was critical for the synergistic mode of action for gibberellin [6,9].

On the other hand, there are three natural 3-acyl-4-hydroxy-5,6-dihydroxy-pyran-2-one products relevant to 4-methoxy-5,6-dihydroxy-pyran-2-ones: (*R*)-podoblastins [19], (*R*)-lachnelluloic acid [20], and alternaric acid [21] (Figure 2). We previously reported asymmetric total syntheses of all these natural products utilizing a catalytic asymmetric Mukaiyama aldol reaction and an asymmetric Ti-Claisen condensation as the crucial steps [22,23].

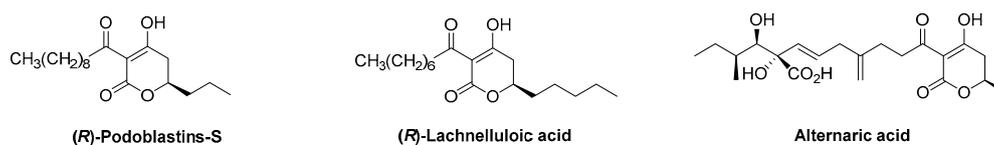


Figure 2. All three 3-acyl-5,6-dihydro-2*H*-pyran-2-one natural products.

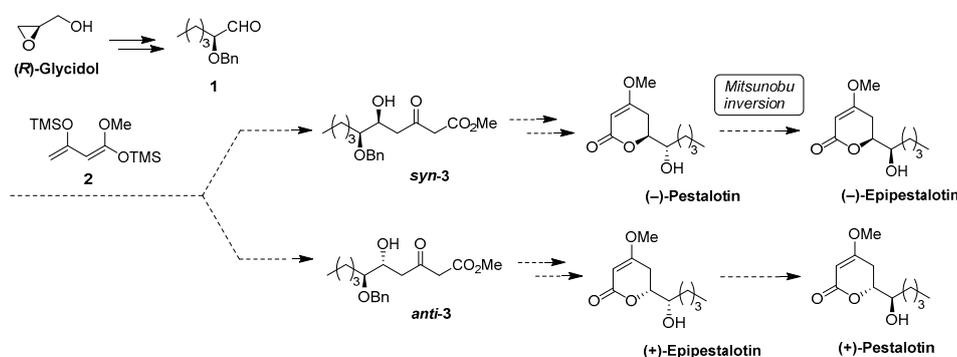
Consistent with our expeditious total syntheses of all these compounds, we envisaged a divergent synthetic access to all four chiral pestalotin diastereomers starting from a common and readily-available chiral building block, i.e., (*R*)-glycidol.

2. Results and Discussion

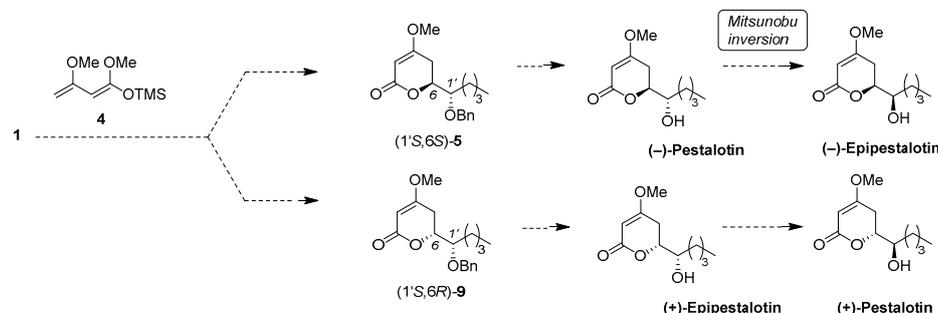
2.1. General Strategy for the Total Syntheses of All Four Pestalotin Diastereomers

A couple of the present divergent strategies involve a catalytic asymmetric and a diastereoselective Mukaiyama aldol addition, and a diastereoselective hetero-Diels-Alder reaction, followed by a Mitsunobu inversion as the crucial steps (Scheme 1). (*R*)-Glycidol is transformed to a common starting (*S*)-2-benzyloxyhexanal (**1**) by the epoxide opening with a Grignard reagent. Syn- and anti-selective Mukaiyama aldol additions of readily-available bis(TMSO)diene (so-called Chan's diene) **2** [13] with (*S*)-aldehyde **1** produce stereocomplementary chiral aldol adducts *syn*-**3** and *anti*-**3**, respectively. Alternatively, *syn*- and *anti*-selective hetero-Diels-Alder reactions of similarly available mono(TMSO)diene (so-called Brassard's diene) **4** [11,24] with **1** produce diastereomeric chiral pyrone-adducts *syn*-**5** and *anti*-**5**, respectively. Following a conventional synthetic procedure [15], *syn*-**3** and *anti*-**3** are transformed to (−)-pestalotin and (+)-epipestalotin, respectively. Mitsunobu inversions of (−)-pestalotin and (+)-epipestalotin produce (−)-epipestalotin and (+)-pestalotin, respectively.

< Catalytic Asymmetric and Diastereoselective Mukaiyama Aldol Reactions >



< Catalytic Hetero-Diels Alder Reaction >

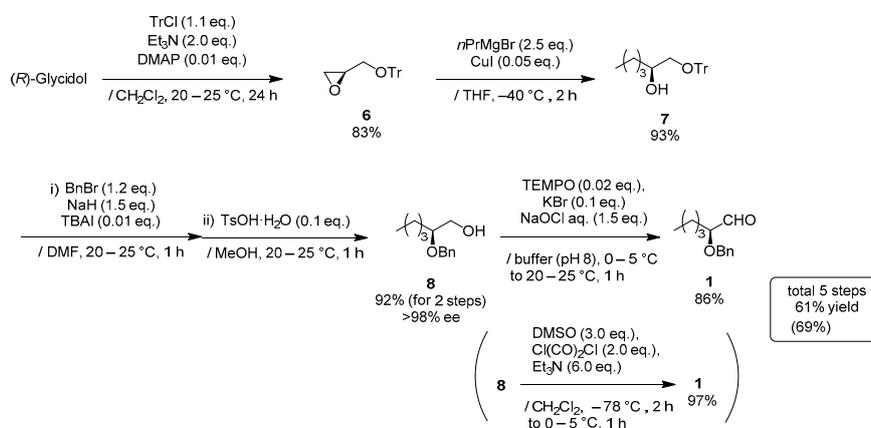


Scheme 1. Strategies for asymmetric total syntheses of pestalotin diastereomers.

2.2. Total Syntheses of All Four Pestalotin Diastereomers

Synthesis of (*S*)-2-benzyloxyhexanal (**1**)

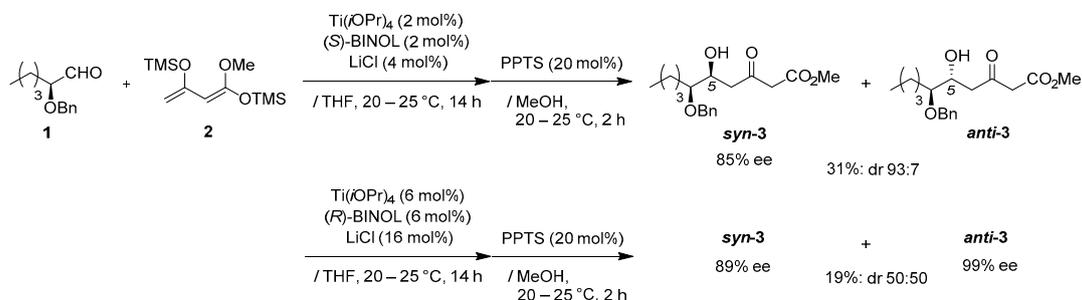
(*S*)-2-Benzyloxyhexanal (**1**) was synthesized from (*R*)-glycidol as shown in Scheme 2. (*R*)-Glycidol was converted to trityl ether **6** (or commercially available) as a crude solid, which was purified by recrystallization (83% yield). CuI-catalyzed Grignard reaction of *n*-PrMgBr with epoxide **6** [25] gave secondary alcohol **7** in 93% yield. After the benzyl group protection of **7**, the trityl group was removed using a PTS•H₂O catalyst to afford primary alcohol **8** in 92% yield (2 steps). Finally, TEMPO (or Swern) oxidation of **8** produced (*S*)-2-benzyloxyhexanal **1** in 86% (or 97%) yield. Because of its easier recrystallization purification procedure, trityl protection method was selected instead of an alternative *p*-methoxybenzyl protective method. The present sequence (four steps and 61% overall yield) is superior regarding steps and overall yield compared with the relevant reported route starting from (*S*)-norleucine (five steps and 27% overall yield) [14].



Scheme 2. Synthesis of common (*S*)- α -benzyloxy aldehyde **1**.

2.3. Catalytic Asymmetric and Diastereoselective Mukaiyama Aldol Reactions

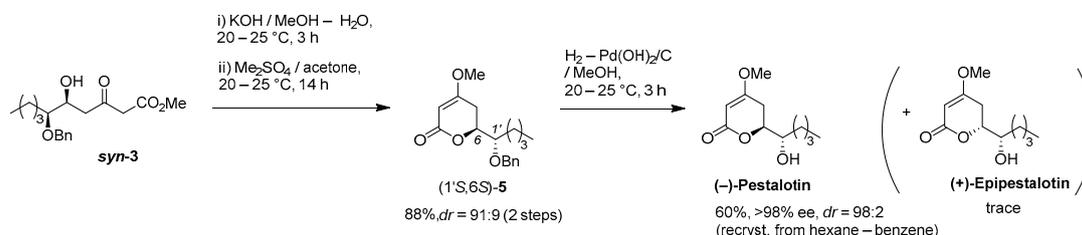
With (*S*)-aldehyde **1** in hand, we next investigated a catalytic asymmetric Mukaiyama aldol reaction using readily-available Chan's diene **2** [13] with **1** (Scheme 3). For this purpose, we employed the procedure applied for the asymmetric syntheses of (*R*)-podoblastin-S and (*R*)-lachnelluloic acid [22], as well as that described in Organic Syntheses, recently [26]. The reaction by using catalysis of Ti(*i*OPr)₄ (2 mol%)/(*S*)-BINOL (2 mol%)/LiCl (4 mol%) and subsequent treatment with PPTS/MeOH afforded the desired aldol adduct *syn*-**3** in 31% yield with high diastereoselectivity and enantioselectivity [*syn*/*anti* = 93:7, 85% ee (C-5 position) by HPLC analysis].



Scheme 3. Catalytic asymmetric Mukaiyama aldol reaction.

Instead of (*S*)-BINOL, antipodal (*R*)-BINOL (6 mol %) was examined under identical conditions. Expectedly, the results differed with regard to the yield and diastereoselectivity [*syn*/*anti* = 50:50, 89% ee (*syn*), and 99% ee (*anti*) by HPLC analysis] (mismatching).

Pyrone formation and successive O-methylation using *syn*-3 according to the reported method [15] produced 4-methoxy-5,6-dihydro-2*H*-pyran-2-one precursor (1'*S*,6*S*)-5 in 88% yield (*dr* = 91:9) in two steps (Scheme 4). Finally, Pd/C-catalyzed debenzoylation of (1'*S*,6*S*)-5 furnished (–)-pestalotin in 60% yield and 99% ee (C-6 position) by HPLC analysis after recrystallization, together with a trace amount of (+)-epipestalotin.



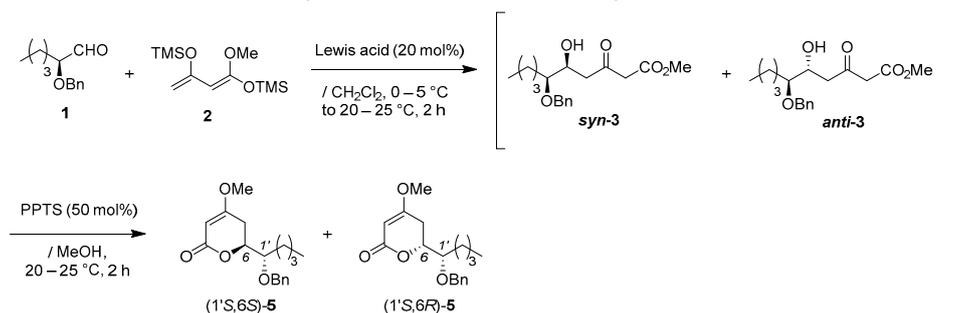
Scheme 4. Synthesis of (–)-pestalotin.

2.4. Diastereoselective Mukaiyama Aldol Reactions Promoted by Achiral Lewis Acids

Several simpler achiral Lewis acids were screened for diastereoselective Mukaiyama aldol reactions (Table 1). Hagiwara's pioneering work addressed Lewis acid-mediated crossed-aldol reactions between 1 and 2 to afford *syn*-3 adducts [15]; TiCl_4 (100 mol %) produced excellent *syn*-3 diastereoselectivity, but the anti-3 selectivity was insufficient when using several other Lewis acids ($\text{BF}_3 \cdot \text{OEt}_2$, Et_2AlCl , ZnCl_2). Taking this information into account, we reinvestigated this procedure with the aim of enhancing stereocomplementary anti-3 selectivity. The salient features are as follows: (i) The amount of TiCl_4 (100 mol %) could be decreased to a catalytic amount (20 mol %), by which aldehyde 1 was sufficiently consumed (entries 1–3). (ii) Notably, the aldol-step reaction mixture was directly treated with PPTS/MeOH solution following the procedure mentioned described in Section 2.2 to furnish the desired 4-methoxy-5,6-dihydro-2*H*-pyran-2-one precursor (1'*S*,6*S*)-5 smoothly with good *syn*-/anti-selectivity and excellent enantioselectivity at the C6-position (entry 2). This one-pot furan formation is the first finding among previously reported total syntheses. (iii) The use of other strong Lewis acids such as AlCl_3 , SnCl_4 , and $\text{BF}_3 \cdot \text{OEt}_2$, did not afford fruitful results (entries 4–6). (iv) Fortunately, the reaction using ZrCl_4 switched the selectivity from *syn*- to anti- to afford (1'*S*,6*S*)-5 as a major product with moderate diastereoselectivity but with excellent enantioselectivity (entry 8). (v) The use of mild metal triflate reagents such as $\text{M}(\text{OTf})_n$ ($\text{M} = \text{Sc, La, Cu}$) were examined next. In contrast to TiCl_4 and ZrCl_4 , $\text{Cu}(\text{OTf})_2$ produced a satisfactory yield with excellent *syn*-3 selectivity and enantioselectivity (entry 11).

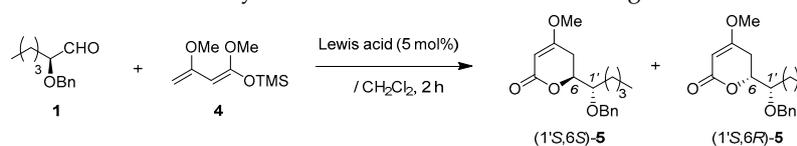
2.5. Catalytic Diastereoselective Hetero-Diels-Alder Reaction

Next, our attention was focused on a hetero-Diels-Alder reaction between aldehyde 1 and Brassard's siloxydiene (4) [11] to construct pyrone precursors (1'*S*,6*S*)-5 and (1'*S*,6*R*)-5 in a straightforward manner, basically according to Midland's protocol [12] (Scheme 2). The salient features are as follows: (i) Several Lewis acid catalysts (TiCl_4 , AlCl_3 , SnCl_4 , $\text{BF}_3 \cdot \text{OEt}_2$, ZnCl_2 , and MgCl_2) were screened (Table 2). The reaction profile apparently differed from the result listed in Table 1; i.e., both the yield and stereoselectivities were moderate to low (entries 1–5). (ii) Among metal triflate catalysts $\text{M}(\text{OTf})_n$ ($\text{M} = \text{Sc, La, Cu}$), only $\text{Sc}(\text{OTf})_3$ afforded moderate result (entry 6), and, in contrast to our expectation $\text{Cu}(\text{OTf})_2$ afforded a disappointing result (entry 8). (iii) A reinvestigation of Midland's best conditions using "chiral" $\text{Eu}(\text{hfc})_3$ revealed good selectivity for (1'*S*,6*S*)-5 (entry 9). (iv) Notably, the use of more inexpensive and accessible "achiral" $\text{Eu}(\text{fod})_3$ produced superior diastereoselectivity and enantioselectivity (entry 10).

Table 1. Catalytic diastereoselective Mukaiyama aldol reaction.

Entry	Lewis Acid	Temp./°C	Yield/% 3–5	Syn-3/ Anti-3 ^a	(1'S,6S)-5/(1'S,6R)-5 ^a	ee/% ^a		ee/% ^a		
						syn-3	anti-3	(1'S,6S)-5	(1'S,6R)-5	
1	TiCl ₄	20–25	7	33	-	60:40	-	-	95	ND ^b
2		0–5	0	49	-	87:13	-	-	91	90
3		-20	23	31	55:45	62:38	ND ^b	98	>98	ND ^b
4	AlCl ₃	0–5	trace	trace	-	-	-	-	-	-
5	SnCl ₄	0–5	33	0	60:40	-	70	87	-	-
6	BF ₃ •OEt ₂	0–5	trace	trace	-	-	-	-	-	-
7 ^c	ZrCl ₄	0–5	21	43	33:67	33:67	ND ^b	98	>98	ND ^b
8 ^c		0–5	trace	41	-	35:65	-	-	-	>98
9 ^c	Sc(OTf) ₃		32	5	56:44	-	ND ^b	-	-	-
10 ^c	La(OTf) ₃		trace	trace	-	-	-	-	-	-
11 ^c	Cu(OTf) ₂		53	5	93:7	-	98	-	-	-

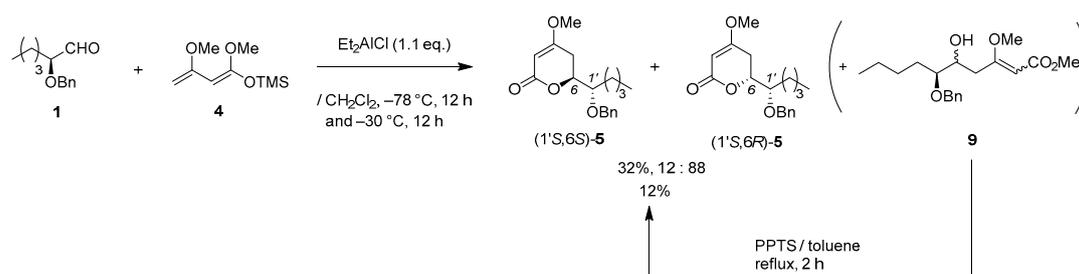
a) Concerning the C6-position. Determined by HPLC analysis (DAICEL Chiralcel AD-3). b) Not determined due to the HPLC peak overlap of **3** and **5**. c) PPTS/MeOH step: 40–45 °C.

Table 2. Stereoselective catalytic hetero-Diels-Alder reaction using Brassard's siloxydiene.

Entry	Lewis Acid	Temp./°C	Yield/% ^a	(1'S,6S)-5/(1'S,6R)-5 ^a
1	TiCl ₄		19	55:45
2	SnCl ₄		22	
3	BF ₃ •OEt ₂		complex mixtures	
4	ZnCl ₂		33	58:42
5	MgCl ₂		31	64:36
6	Sc(OTf) ₃	-78 to 20–25	37	79:21
7	La(OTf) ₃		complex mixtures	
8	Cu(OTf) ₂		complex mixtures	
9	Eu(hfc) ₃	0–5 to 20–25	41	88 (95% ee) ^b :12
10	Eu(fod) ₃	0–5	67 ^c	98 (>98% ee) ^b :2

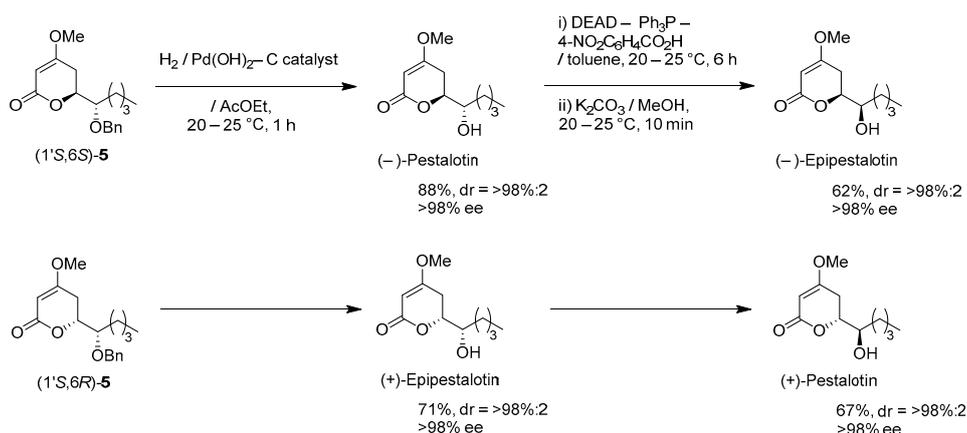
a) Determined by ¹H-NMR analysis. b) Concerning the C6-position. c) Isolated.

According to Midland's report, stereocomplementary (1'S,6R)-diastereoselective reaction using Et₂AlCl catalyst was examined to obtain pyrone (1'S,6R)-**5** in our hands (Scheme 5). Due to the subtle reported conditions, the reaction was hardly reproducible, and our best result was addressed; the obtained crude product contained considerable amounts of aldol-type compound **9** with the desirable product (1'S,6R)-**5**. Compound **9** was converted to (1'S,6R)-**5** by PPTS/toluene under reflux conditions, albeit in poor yield (12%).



Scheme 5. (1'S,6R)-Diastereoselective hetero-Diels-Alder reaction using Brassard's siloxydiene.

Finally, debenylation of (1'S,6S)-5 and (1'S,6R)-5 using the $H_2/Pd(OH)_2-C$ catalyst produced (–)-pestalotin and (+)-epipestalotin, respectively, in good yield and with excellent optical purities (Scheme 6). Gratifyingly, Mitsunobu inversions of (–)-pestalotin and (+)-epipestalotin smoothly proceeded to furnish (+)-epipestalotin and (–)-pestalotin, respectively (Scheme 6). The present inversion step increases the value of the whole synthesis by a convergent process. Physical and spectral data (mp, optical rotation, 1H -NMR) of all four pestalotin diastereomers matched completely with Mori's reported data [9]. Additional ^{13}C -NMR spectral data and HPLC measurements are described in the experimental and in the ESI, respectively. The present divergent methodology is superior compared with Mori's approach to the only reported total synthesis of all four pestalotin families [9] in the following respects: (i) common (*R*)-glycidol starting compound, (ii) short syntheses (7 and 9 steps), and (iii) higher total yield.



Scheme 6. Final stage of the total synthesis all four pestalotin diastereomers.

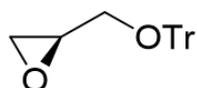
3. Materials and Methods

All reactions were carried out in oven-dried glassware under an argon atmosphere. Flash column chromatography was performed with silica gel 60 (230–400 mesh ASTM, Merck, Darmstadt, Germany). TLC analysis was performed on Merck 0.25 mm Silicagel 60 F₂₅₄ plates. Melting points were determined on a hot stage microscope apparatus (ATM-01, AS ONE, Osaka, Japan) and were uncorrected. NMR spectra were recorded on a JEOLRESONANCE EXC-400 or ECX-500 spectrometer (JEOL, Akishima, Japan) operating at 400 MHz or 500 MHz for 1H -NMR, and 100 MHz and 125 MHz for ^{13}C NMR. Chemical shifts (δ ppm) in $CDCl_3$ were reported downfield from TMS (=0) for 1H -NMR. For ^{13}C -NMR, chemical shifts were reported in the scale relative to $CDCl_3$ (77.00 ppm) as an internal reference. Mass spectra were measured on a JMS-T100LC spectrometer (JEOL, Akishima, Japan). HPLC data were obtained on a SHIMADZU (Kyoto, Japan) HPLC system (consisting of the following: LC-20AT, CMB20A, CTO-20AC, and detector SPD-20A measured at 254 nm) using Chiracel AD-H or Ad-3 column (Daicel, Himeji, Japan, 25 cm) at 25 °C. Optical rotations were measured on a JASCO DIP-370 (Na lamp, 589 nm).

(R)-2-((trityloxy)methyl)oxirane (6)

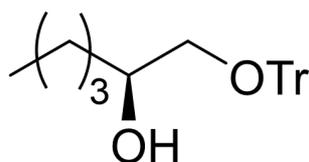
TrCl (15.3 g, 55 mmol) in CH₂Cl₂ (35 mL) was added to a stirred solution of (*R*)-(+)-glycidol (3.70 g, 50 mmol) and Et₃N (13.9 mL, 100 mmol) and DMAP (61 mg, 0.5 mmol) in CH₂Cl₂ (15 mL) at 0–5 °C under an Ar atmosphere, followed by stirring at 20–25 °C for 24 h. The mixture was quenched with sat. NH₄Cl aq., which was extracted three times with Et₂O. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. The obtained crude solid was purified by recrystallization from MeOH (100 mL) to give the desired product **6** (13.1 g, 83%).

Colorless crystals, mp 99–100 °C [lit. [25], 100 °C (EtOH)]; ¹H-NMR (400 MHz, CDCl₃): δ = 2.63 (dd, *J* = 2.3 Hz, 5.0 Hz, 1H), 2.78 (dd, *J* = 4.6, 1H), 3.09–3.18 (m, 2H), 3.32 (dd, *J* = 2.3 Hz, 10.0 Hz, 1H), 7.20–7.35 (m, 10H), 7.42–7.50 (m, 5H); ¹³C-NMR (100 MHz, CDCl₃): δ = 44.6, 51.0, 64.7, 86.6, 127.0 (3C), 127.8 (6C), 128.6 (6C), 143.8.

**(S)-1-(Trityloxy)hexan-2-ol (7)**

1-Bromopropane (8.60 mL, 95 mmol) was gradually added to a stirred Mg granular (2.31 g, 95 mmol) and a small amounts of I₂ in THF (60 mL) at 20–25 °C under an Ar atmosphere, and the mixture was stirred for 0.5 h at 20–25 °C. CuI (143 mg, 0.80 mmol) was added, the mixture was cooled down to –40 °C and (*S*)-oxirane **6** (12.1 g, 38 mmol) in THF (100 mL) was added to the mixture at the same temperature, followed by stirring for 2 h. The mixture was quenched with sat. NH₄Cl aq., which was extracted three times with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. The obtained crude product was purified by SiO₂-column chromatography (hexane/AcOEt = 15/1) to give the desired alcohol **7** (12.7 g, 93%).

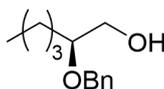
Pale yellow oil; ¹H-NMR (400 MHz, CDCl₃): δ = 0.86 (t, *J* = 6.9 Hz, 3H), 1.16–1.46 (m, 6H), 2.30 (d, *J* = 3.7 Hz, 1H), 3.02 (dd, *J* = 7.8 Hz, 9.2 Hz, 1H), 3.18 (dd, *J* = 3.2 Hz, 9.2 Hz, 1H), 3.72–3.80 (m, 1H), 7.19–7.35 (m, 10H), 7.40–7.47 (m, 5H); ¹³C-NMR (100 MHz, CDCl₃): δ = 13.9, 22.6, 27.6, 33.0, 67.7, 70.9, 86.6, 127.0 (3C), 127.8 (6C), 128.6 (6C), 143.8.

**(S)-2-(Benzyloxy)hexan-1-ol (8) [15]**

A mixture of benzyl bromide (4.85 mL, 41 mmol) and (*S*)-alcohol **7** (12.4 g, 34 mmol) in DMF (25 mL) were added to a stirred suspension of NaH (60%; 2.04 mg, 51 mmol) in DMF (10 mL) at 0–5 °C under an Ar atmosphere. TBAI (126 mg, 0.3 mmol) was added to the mixture and the mixture was allowed to warm up to 20–25 °C, followed by stirring for 1 h. The mixture was quenched with MeOH and K₂CO₃, which was extracted three times with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. The obtained crude oil (15.6 g) was used for the next step without purification.

TsOH·H₂O (647 mg, 3.4 mmol) was added to a solution of the oil (15.6 g) in MeOH (70 mL) at 20–25 °C under an Ar atmosphere, and the mixture was stirred for 1 h at the same temperature. The mixture was quenched with sat. NaHCO₃ aq. and concentrated, which was extracted three times with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. The obtained crude product was purified by SiO₂-column chromatography (hexane/AcOEt = 15:1–3:1) to give **8** (6.52 g, 92% for 2 steps, >98% ee).

Yellow oil; $[\alpha]_{\text{D}}^{24} +21.4$ (*c* 1.16, CHCl_3) [lit. [15], $[\alpha]_{\text{D}}^{\text{unknown}} +22.3$ (*c* 1.13, CHCl_3)]; $^1\text{H-NMR}$ (400 MHz, CDCl_3): $\delta = 0.90$ (t, *J* = 6.9 Hz, 3H), 1.23–1.40 (m, 4H), 1.44–1.71 (m, 2H), 1.93 (brs, 1H), 3.47–3.58 (m, 2H), 3.65–3.75 (m, 1H), 4.54 (d, *J* = 11.5 Hz, 1H), 4.63 (d, *J* = 11.5 Hz, 1H), 7.27–7.39 (m, 5H); $^{13}\text{C NMR}$ (500 MHz, CDCl_3): $\delta = 13.8, 22.6, 27.3, 30.3, 63.9, 71.3, 79.7, 127.4, 127.6$ (2C), 128.2 (2C), 138.3. HPLC analysis (AD-H, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 30/1) t_{R} (racemic) = 9.33 min and 10.27 min. t_{R} [(*S*)-form] = 8.95 min.



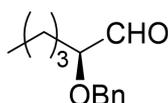
(*S*)-2-(Benzyloxy)hexanal (1) [15]

TEMPO (106 mg, 0.68 mmol) and KBr (407 mg, 3.4 mmol) was added to a stirred solution of alcohol **8** (7.08 g, 34 mmol) in CH_2Cl_2 (34 mL) at 0–5 °C under an Ar atmosphere. A mixture of NaOCl aq. (1.5 M, 34 mL, 51 mmol), NaHCO_3 (6.7 g, 80 mmol), and Na_2CO_3 (318 mg, 3 mmol) in water (220 mL), was added to the solution at same temperature. The mixture was allowed to warm to 20–25 °C, followed by stirring at the same temperature for 1 h. The mixture was quenched with water, which was extracted twice with CH_2Cl_2 . The combined organic phase was washed with water, brine, dried (Na_2SO_4), and concentrated. The obtained crude oil was purified by Florisil® column chromatography (hexane/AcOEt = 5:1) to give the desired product **1** (6.04 g, 86%).

Yellow oil; $[\alpha]_{\text{D}}^{24} -81.2$ (*c* 1.08, CHCl_3) [lit. [15], $[\alpha]_{\text{D}}^{\text{unknown}} -86.1$ (*c* 0.98, CHCl_3)]; $^1\text{H-NMR}$ (500 MHz, CDCl_3): $\delta = 0.90$ (t, *J* = 7.5 Hz, 3H), 1.24–1.49 (m, 4H), 1.69 (q, *J* = 6.9 Hz, 13.8 Hz, 2H), 3.76 (t, *J* = 6.3 Hz, 1H), 4.54 (d, *J* = 11.5 Hz, 1H), 4.68 (d, *J* = 11.5 Hz, 1H), 7.27–7.41 (m, 5H), 9.66 (s, 1H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): $\delta = 13.7, 22.3, 26.7, 29.6, 72.3, 83.3, 127.8, 127.9, 128.4, 137.3, 203.6$.

An alternative method is following:

DMSO (4.26 mL, 60 mmol) in CH_2Cl_2 (20 mL) was added slowly to a stirred solution of oxalyl dichloride (3.43 mL, 40 mmol) in CH_2Cl_2 (60 mL) at –78 °C under an Ar atmosphere. After the mixture was stirred for 5 min, **8** (4.22 g, 20 mmol) in CH_2Cl_2 (20 mL) was added and the mixture was stirred for 0.5 h at the same temperature. Et_3N (16.6 mL, 120 mmol) was added to the mixture and the mixture was allowed to warm up to 0–5 °C over a period of 1 h, followed by stirring for 1 h at 0–5 °C. The mixture was quenched with water, which was extracted three times with Et_2O . The combined organic phase was washed with a large amounts of water, brine, dried (Na_2SO_4), and concentrated. The obtained crude product was purified by SiO_2 -column chromatography (hexane/AcOEt = 25/1) to give the desired product **1** (3.99 g, 97%).

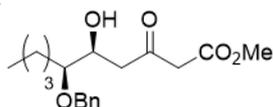


Methyl (5*S*,6*S*)-6-(benzyloxy)-5-hydroxy-3-oxodecanoate (*syn*-3) [15]

Preparation for Ti-BINOL solution: A suspension of $\text{Ti}(\text{O}i\text{Pr})_4$ (2.9 mg, 10 μmol), and (*S*)-BINOL (2.8 mg, 10 μmol) in THF (0.4 mL) was stirred at 20–25 °C under an Ar atmosphere for 1 h.

Asymmetric Mukaiyama aldol reaction: Ti-BINOL solution was added to a stirred suspension of aldehyde **1** (103 mg, 0.50 mmol) and LiCl (0.85 mg, 20 μmol) in THF (0.5 mL) at 20–25 °C under an Ar atmosphere, followed by stirring at the same temperature for 0.5 h. Chan's diene **2** (260 mg, 1.0 mmol) in THF (0.3 mL) was added slowly to the mixture, which was stirred for 14 h. PPTS (25 mg, 0.10 mmol) in MeOH (1.0 mL) was added to the mixture, followed by stirring at the same temperature for 2 h. The resulting mixture was quenched with sat. NaHCO_3 aq., which was extracted three times with Et_2O . The combined organic phase was washed with water, brine, dried (Na_2SO_4), and concentrated. The obtained crude oil was purified by SiO_2 -column chromatography (hexane/AcOEt = 8/1) to give the desired product *syn*-3 (85% ee, dr 93:7, 51 mg, 31%).

Pale yellow oil; $[\alpha]_D^{25} +1.0$ (c 1.0, CHCl_3) [lit. [15], $[\alpha]_D^{\text{unknown}} +1.2$ (c 1.00, CHCl_3)]; 85% ee; HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 30:1) $t_R(\text{racemic}) = 13.51$ min, 14.13 min, 18.89 min and 19.82 min. $t_R[(5S,6S)\text{-form}] = 18.69$ min. $^1\text{H-NMR}$ (500 MHz, CDCl_3): $\delta = 0.91$ (t, $J = 6.9$ Hz, 3H), 1.24–1.70 (m, 6H), 2.62–2.64 (m, 1H), 2.71–2.74 (m, 1H), 3.34–3.37 (m, 1H), 3.477 (s, 1H), 3.480 (s, 1H), 3.73 (s, 3H), 4.13–4.18 (m, 1H), 4.49 (d, $J = 11.5$ Hz, 1H), 4.63 (d, $J = 11.5$ Hz, 1H), 7.28–7.37 (m, 5H); $^{13}\text{C-NMR}$ (125 MHz, CDCl_3): $\delta = 14.0, 22.8, 27.6, 29.3, 46.0, 49.6, 52.3, 68.3, 72.2, 80.8, 127.8, 127.9, 128.4, 138.2, 167.4, 202.7$.

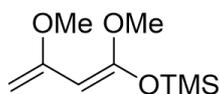


(*E*)-((1,3-dimethoxybuta-1,3-dien-1-yl)oxy)trimethylsilane (**4**) (Brassard's diene)

Concentrated H_2SO_4 (0.27 mL, 5.0 mmol) was added to a stirred mixture of methyl acetoacetate (11.6g, 100 mmol) and trimethyl orthoformate (26.5 g, 250 mmol) at 0–5 °C under an Ar atmosphere, followed by stirring at 20–25 °C for 24 h. K_2CO_3 (5.0 g) was added to the mixture, which was filtered through a glass filter. The filtrate was concentrated under reduced pressure. The obtained crude oil was purified by distillation (bp 72–75 °C/3.2 kPa) to give the desired (*E*)-methyl-3-methoxybut-2-enoate (9.08 g, 70%).

*n*BuLi (1.63 M in hexane, 13.6 mL, 22 mmol) was added to stirred solution of *i*Pr₂NH (3.11 mL, 22 mmol) in THF (10 mL) at 0–5 °C under an Ar atmosphere, followed by stirring for 10 min. The mixture was cooled down to –78 °C and (*E*)-methyl-3-methoxybut-2-enoate (2.22 g, 17 mmol) in THF (4.0 mL) was added to the mixture, followed by stirring at the same temperature for 0.5 h. TMSCl (2.58 mL, 20 mmol) in THF (3.0 mL) was added to the mixture at the same temperature and the mixture was allowed to warm up to 0–5 °C over a period of 1 h. The mixture was concentrated and filtered through Celite® (No.503) using a glass filter, and washing with hexane (10 mL × 3). The filtrate was concentrated under reduced pressure and the obtained crude oil was purified by distillation to give the desired product **4** (2.62 g, 76%).

Colorless oil; bp 40–43 °C/50 Pa; $^1\text{H-NMR}$ (400 MHz, CDCl_3): $\delta = 0.26$ (s, 9H), 3.56 (s, 3H), 3.57 (s, 3H), 3.99 (t, $J = 1.4$ Hz, 1H), 4.03 (d, $J = 1.4$ Hz, 1H), 4.34 (d, $J = 1.8$, 1H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): $\delta = 0.3, 54.0, 55.0, 75.5, 78.6, 158.7$



(*S*)-6-[(*S*)-1-(Benzyloxy)pentyl]-4-methoxy-5,6-dihydro-2*H*-pyran-2-one [(1'*S*,6*S*)-**5**] [15]

(1) 1M-KOH aq. (0.37 mL) was added to a stirred solution of (5*S*,6*S*)-aldol adduct *syn*-**3** (108 mg, 0.33 mmol) in MeOH (0.37 mL) at 20–25 °C under an Ar atmosphere, followed by stirring at the same temperature for 3 h. The mixture was quenched with 1M-HCl aq., which was extracted twice with AcOEt. The combined organic phase was washed with brine, dried (Na_2SO_4), and concentrated. The obtained crude oil was purified by SiO₂-gel column chromatography (hexane/AcOEt = 5/1–2/1) to give the desired 4-hydroxy-5,6-dihydro-2*H*-pyran-2-one precursor (dr 93:7, 94 mg, 98%).

$^1\text{H-NMR}$ (500 MHz, CDCl_3): $\delta = 0.92$ (t, $J = 6.9$ Hz, 3H), 1.28–1.40 (m, 4H), 1.71–1.77 (m, 2H), 2.58 (dd, $J = 5.2$ Hz, 17.2 Hz, 1H), 2.76 (dd, $J = 5.2$ Hz, 17.2 Hz, 1H), 3.30 (d, $J = 20.1$ Hz, 1H), 3.41 (d, $J = 20.1$ Hz, 1H), 3.42–3.45 (m, 1H), 4.44 (d, $J = 10.9$ Hz, 1H), 4.59 (d, $J = 10.9$ Hz, 1H), 4.71–4.74 (m, 1H), 7.26–7.37 (m, 1H); $^{13}\text{C-NMR}$ (125 MHz, CDCl_3): $\delta = 13.9, 22.7, 27.5, 29.3, 40.6, 46.2, 72.3, 75.9, 80.0, 128.1, 128.2, 128.5, 136.9, 167.7, 199.4$

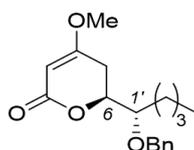
K_2CO_3 (80 mg, 0.58 mmol) was added to a stirred suspension of the precursor (85 mg, 0.29 mmol) and Me_2SO_4 (55 mg, 0.44 mmol) in acetone (1.5 mL) at 20–25 °C under an Ar atmosphere, followed by

stirring at the same temperature for 14 h. The mixture was quenched with water, which was extracted three times with Et₂O. The combined organic phase was washed with brine, dried (Na₂SO₄), and concentrated. The obtained crude oil was purified by SiO₂-gel column chromatography (hexane/AcOEt = 6/1–4/1) to give the desired product (1'*S*,6*S*)-5 (dr 91:9, 79 mg, 89%).

Pale yellow oil; $[\alpha]_{\text{D}}^{25} -93.7$ (*c* 0.72, CHCl₃). [lit. [15], $[\alpha]_{\text{D}}^{\text{unknown}} -99.1$ (*c* 0.93, CHCl₃)].

(2) TiCl₄ (0.02 mL, 0.2 mmol) was added to a solution of aldehyde **1** (206 mg, 1.0 mmol) in CH₂Cl₂ (3.0 mL) at 0–5 °C under an Ar atmosphere, followed by stirring at the same temperature for 10 min. Chan's diene (61 % purity, 520 mg, 1.2 mmol) was added to the mixture, which was stirred at 0–5 °C for 5 min and at 20–25 °C for 1 h. MeOH (2 mL) and PPTS (125 mg, 0.5 mmol) was successively added to the mixture, followed by stirring at the same temperature for 2 h. The mixture was quenched with sat. NaHCO₃ aq., which was filtered through Celite®. The filtrate was extracted twice with AcOEt, and the combined organic phase was washed with water, brine dried (Na₂SO₄), and concentrated. The obtained crude oil was purified by SiO₂-column chromatography (hexane/AcOEt = 4:1) to give the desired product (1'*S*,6*S*)-5 [165 mg, 49%, 91% ee, dr = 87:13].

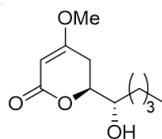
(3) Aldehyde **1** (413 mg, 2.0 mmol) in CH₂Cl₂ (1.0 mL) was added to a stirred suspension of Eu(fod)₃ (104 mg, 0.1 mmol) in CH₂Cl₂ (1.0 mL) at 0–5 °C under an Ar atmosphere, followed by stirring at the same temperature for 5 min. Brassard's diene **4** (607 mg, 3.0 mmol) in CH₂Cl₂ (2.0 mL) was added to the mixture at the same temperature, followed by stirring for 2 h. The mixture was quenched with water, which was extracted three times with AcOEt. The combined organic phase was washed with water, brine, dried (Na₂SO₄), and concentrated. The obtained crude product was purified by SiO₂-column chromatography (hexane/AcOEt = 3/1) to give the desired product [(1'*S*,6*S*)-5] (370 mg, 67%, >98% ee, dr = 98:2). HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 30:1) *t*_R(racemic) = 23.25 min and 24.77 min. *t*_R[(1*S*,6*S*)-form] = 25.53 min.; ¹H-NMR (500 MHz, CDCl₃): δ = 0.89 (t, *J* = 6.9 Hz, 3H), 1.25–1.72 (m, 6H), 2.26 (dd, *J* = 4.0 Hz, 17.2 Hz, 1H), 2.70 (ddd, *J* = 1.7 Hz, 13.2 Hz, 17.2 Hz, 1H), 3.58–3.61 (m, 1H), 3.74 (s, 3H), 4.52 (dt, *J* = 4.0 Hz, 13.2 Hz, 1H), 4.62 (d, *J* = 11.5 Hz, 1H), 4.66 (d, *J* = 11.5 Hz, 1H), 5.13 (d, *J* = 1.7 Hz, 1H), 7.27–7.36 (m, 5H); ¹³C NMR (125 MHz, CDCl₃): δ = 14.0, 22.7, 27.9, 28.4, 29.3, 56.1, 72.9, 76.3, 79.0, 90.2, 127.8, 127.9, 128.4, 138.1, 167.0, 173.3.



(–)-Pestalotin; (S)-6-[(S)-1-Hydroxypentyl]-4-methoxy-5,6-dihydro-2*H*-pyran-2-one [9]

A suspension of benzyl ether [(1*S*,6*S*)-5] (448 mg, 1.5 mmol) and 20% Pd(OH)₂/C (53 mg, 0.08 mmol) in AcOEt (15 mL), equipped with a H₂ balloon, was stirred at 20–25 °C for 1 h. The mixture was filtered through Celite® (No.503) using glass filter and the filtrate was concentrated under reduced pressure. The obtained crude solid (384 mg) was purified by SiO₂-column chromatography (hexane/AcOEt = 3:2) to give the desired (–)-pestalotin (283 mg, 88%, >98% ee, dr = >98:2).

Colorless crystals; mp 84–86 °C (lit. [9], 85.8–86.0 °C); $[\alpha]_{\text{D}}^{25} -91.9$ (*c* 0.44, MeOH) [lit. [9], $[\alpha]_{\text{D}}^{21} -90.2$ (*c* 1.17, MeOH)]; HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 30:1) *t*_R(racemic) = 45.23 min and 48.13 min. *t*_R[(1*S*,6*S*)-form] = 45.85 min.; ¹H-NMR (500 MHz, CDCl₃): δ = 0.92 (t, *J* = 6.9 Hz, 3H), 1.30–1.67 (m, 6H), 2.07 (brs, 1H), 2.25 (dd, *J* = 4.0 Hz, 17.2 Hz, 1H), 2.80 (ddd, *J* = 1.7 Hz, 12.6 Hz, 17.2 Hz, 1H), 3.61–3.64 (m, 1H), 3.76 (s, 3H), 4.30 (dt, *J* = 4.0 Hz, 12.6 Hz, 1H), 5.15 (d, *J* = 1.7 Hz, 1H); ¹³C-NMR (125 MHz, CDCl₃): δ = 13.9, 22.6, 27.6, 29.6, 32.4, 56.1, 72.4, 78.4, 90.0, 166.7, 173.1.

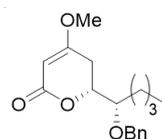


(R)-6-[(S)-1-(Benzyloxy)pentyl]-4-methoxy-5,6-dihydro-2H-pyran-2-one [(1'S,6R)-5]

(1) Aldehyde **1** (206 mg, 1.0 mmol) was added to a stirred suspension of $ZrCl_4$ (47 mg, 0.2 mmol) in CH_2Cl_2 (0.9 mL) at 0–5 °C under an Ar atmosphere. After 10 min, Chan's diene (ca. 60% purity; 520 mg, 1.2 mmol) was added to the mixture, which was allowed to warm up to 20–25 °C, followed by stirring for 1 h. MeOH (2.0 mL) and PPTS (125 mg 0.5 mmol) was successively added to the solution, followed by stirring at 40–45 °C for 14 h. Sat. $NaHCO_3$ aq. solution was added to the mixture, which was filtered through Cerite[®]. The filtrate was extracted twice with AcOEt, and the combined organic phase was washed with water, brine dried (Na_2SO_4), and concentrated. The obtained crude oil was purified by SiO_2 -column chromatography (hexane/AcOEt = 4:1) to give the desired product (1'S,6R)-5 (126 mg, 41%, >98% ee, dr = 35:65).

Colorless oil. HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 30:1) t_R (racemic) = 17.72 min and 19.60 min. t_R [(1S,6R)-form] = 18.10 min.; 1H -NMR (500 MHz, $CDCl_3$): δ = 0.89 (t, J = 6.9 Hz, 3H), 1.29–1.64 (m, 6H), 2.35 (dd, J = 4.0 Hz, 17.2 Hz, 1H), 2.81 (ddd, J = 1.7 Hz, 12.6 Hz, 17.2 Hz, 1H), 3.74 (s, 3H), 3.73–3.78 (m, 1H), 4.39 (dt, J = 4.0, 12.6, 1H), 4.63 (d, J = 11.5, 1H), 4.74 (d, J = 11.5, 1H), 5.14 (d, J = 1.7, 1H), 7.28–7.35 (m, 5H); ^{13}C -NMR (125 MHz, $CDCl_3$): δ = 13.9, 22.6, 27.4, 27.9, 30.7, 56.0, 73.3, 78.4, 79.1, 90.0, 127.6, 127.8, 128.3, 138.3, 167.0, 173.4.

(2) Et_2AlCl (1.0 M, 0.6 mL, 0.6 mmol) was added to a stirred solution of aldehyde **1** (103 mg, 0.5 mmol) in CH_2Cl_2 (0.5 mL) at –78 °C under an Ar atmosphere. After 5 min, diene **4** (202 mg, 0.6 mmol) in CH_2Cl_2 (0.5 mL) was added to the mixture, which was stirred for 14 h at the same temperature. The mixture was allowed to warm up to –30 °C, followed by stirring for 14 h. The mixture was quenched by MeOH, which was extracted three times with AcOEt. The combined organic phase was washed with water, brine, dried (Na_2SO_4), and concentrated. The obtained crude product was purified by SiO_2 -column chromatography (hexane/AcOEt = 5/1) to give a mixture of aldol adduct **9** and (1'S,6R)-5 (45:55, 48 mg, 30%). The mixture (48 mg) and PPTS (2 mg, 0.007 mmol) in toluene (1.4 mL), was added at 80–85 °C for 1 h under an Ar atmosphere. After cooling to room temperature, water was added to the mixture, which was extracted twice with AcOEt. The combined organic phase was washed with water and brine, dried (Na_2SO_4), and concentrated. The obtained crude solid purified by SiO_2 -column chromatography (hexane/AcOEt = 5/1) to give the desired product (1'S,6R)-5 (23 mg, 2 steps 15%, ca. 30% of (1'S,6S)-5 was contained).

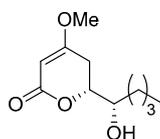


(+)-Epipestalotin; (R)-6-[(S)-1-Hydroxypentyl]-4-methoxy-5,6-dihydro-2H-pyran-2-one [9]

A suspension of benzyl ether [(1S,6R)-5] (365 mg, 1.2 mmol) and 20% $Pd(OH)_2/C$ (42 mg, 0.06 mmol) in AcOEt (12 mL), equipped with a H_2 balloon, was stirred at 20–25 °C for 1 h. The mixture was filtered through Celite[®] (No.503) using glass filter and the filtrate was concentrated under reduced pressure. The obtained crude solid was purified SiO_2 -column chromatography (hexane/AcOEt = 3/2) to give the desired (+)-epipestalotin (187 mg, 71%, >98% ee, dr = 98:2).

Colorless crystals; mp 92–94 °C (lit. [9], 93.0–94.0 °C); $[\alpha]_D^{20}$ + 75.3 (c 0.39, MeOH) [lit. [9], $[\alpha]_D^{17}$ + 75.9 (c 0.39, MeOH)]; >99% ee; HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 25:1) t_R (racemic) = 33.00 min and 35.46 min. t_R [(1S,6R)-form] = 34.81 min.; 1H NMR (500 MHz, $CDCl_3$): δ = 0.92 (t, J = 6.9 Hz, 3H), 1.30–1.56 (m, 6H), 2.04 (brs, 1H), 2.24 (dd, J = 4.0 Hz, 17.2 Hz, 1H),

2.84 (ddd, $J = 1.7$ Hz, 12.6 Hz, 17.2 Hz, 1H), 3.76 (s, 3H), 3.94–3.97 (m, 1H), 4.34 (dt, $J = 3.4$ Hz, 12.6 Hz, 1H), 5.14 (d, 1.7 Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3): $\delta = 13.8, 22.4, 26.8, 27.7, 31.4, 56.0, 71.3, 78.7, 89.7, 167.1, 173.5$.

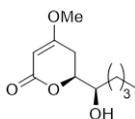


(-)-Epipestalotin; (S)-6-[(R)-1-Hydroxypentyl]-4-methoxy-5,6-dihydro-2H-pyran-2-one [9]

DEAD (40% in toluene, 0.91 mL, 2.0 mmol) was added slowly to a stirred mixture of (-)-pestalotin (214 mg, 1.0 mmol) and 4-nitrobenzoic acid (334 mg, 2.0 mmol) and PPh_3 (525 mg, 2.0 mmol) in toluene (10 mL) at 0–5 °C under an Ar atmosphere, followed by stirring at 20–25 °C for 6 h. The mixture was quenched with water, which was extracted three times with AcOEt. The combined organic phase was washed with sat. NaHCO_3 aq., brine, dried (Na_2SO_4), and concentrated. The obtained crude product was purified by SiO_2 -column chromatography (hexane/AcOEt = 3:1) to give a mixture of the desired (-)-Epipestalotin and diethyl hydrazodicarboxylate, which was used in the next step without further purification.

A suspension of the mixture and K_2CO_3 (138 mg, 1.0 mmol) in MeOH (10 mL) was stirred at 20–25 °C under an Ar atmosphere for 10 min. The mixture was filtered through Celite® (No.503) using a glass filter washing with AcOEt (5 mL \times 3). The filtrate was concentrated under reduced pressure and the obtained crude oil, which was purified by SiO_2 -column chromatography (hexane/AcOEt = 2:1) to give the desired (-)-epipestalotin (133 mg, 62% for 2 steps, >98% ee, dr = >98:2).

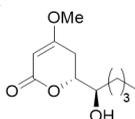
Colorless crystals; mp 89–91 °C (lit. [9], 90.7–91.2 °C); $[\alpha]_{\text{D}}^{20} -75.8$ (c 0.58, MeOH) [lit. [9], $[\alpha]_{\text{D}}^{17} -75.6$ (c 0.58, MeOH)]; HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 25:1) t_{R} (racemic) = 33.00 min and 35.46 min. t_{R} [(1R,6S)-form] = 32.31 min.; ^1H -NMR (500 MHz, CDCl_3): $\delta = 0.92$ (t, $J = 6.9$ Hz, 3H), 1.30–1.55 (m, 6H), 2.04 (brs, 1H), 2.24 (dd, $J = 4.0$ Hz, 17.2 Hz, 1H), 2.84 (ddd, $J = 1.7$ Hz, 12.6 Hz, 17.2 Hz, 1H), 3.76 (s, 3H), 3.94–3.97 (m, 1H), 4.34 (dt, $J = 3.4$ Hz, 12.6 Hz, 1H), 5.14 (d, $J = 1.7$ Hz, 1H); ^{13}C -NMR (125 MHz, CDCl_3): $\delta = 13.8, 22.4, 26.8, 27.7, 31.4, 56.0, 71.3, 78.7, 89.7, 167.1, 173.5$.



(+)-Pestalotin; (R)-6-[(R)-1-Hydroxypentyl]-4-methoxy-5,6-dihydro-2H-pyran-2-one [9]

Following the procedure for the preparation of (-)-epipestalotin, the reaction of (+)-epipestalotin (107 mg, 0.5 mmol) using DEAD (40% in toluene, 0.45 mL, 1.0 mmol), 4-nitrobenzoic acid (167 mg, 1.0 mmol), PPh_3 (262 mg, 1.0 mmol), and K_2CO_3 (69 mg, 0.5 mmol) give the desired (+)-pestalotin (72 mg, 67% for 2 steps, >98% ee, dr > 98:2).

Colorless crystals; mp 82–84 °C (lit. [9], 83.0–84.5 °C); $[\alpha]_{\text{D}}^{20} +97.5$ (c 0.65, MeOH) [lit. [9], $[\alpha]_{\text{D}}^{17} +88.7$ (c 0.65, MeOH)]; HPLC analysis (AD-3, flow rate 1.00 mL/min, solvent: hexane/2-propanol = 30:1) t_{R} (racemic) = 45.23 min and 48.13 min. t_{R} [(1R,6R)-form] = 49.57 min.; ^1H -NMR (500 MHz, CDCl_3): $\delta = 0.92$ (t, $J = 6.9$ Hz, 3H), 1.30–1.67 (m, 6H), 2.07 (brs, 1H), 2.25 (dd, $J = 4.0$ Hz, 17.2 Hz, 1H), 2.80 (ddd, $J = 1.7$ Hz, 12.6 Hz, 17.2 Hz, 1H), 3.61–3.64 (m, 1H), 3.76 (s, 3H), 4.30 (dt, $J = 4.0$ Hz, 12.6 Hz, 1H), 5.15 (d, $J = 1.7$ Hz, 1H); ^{13}C -NMR (125 MHz, CDCl_3): $\delta = 13.9, 22.6, 27.6, 29.6, 32.4, 56.1, 72.4, 78.4, 90.0, 166.7, 173.1$.



4. Conclusions

We achieved an asymmetric total synthesis of all four chiral pestalotin diastereomers using common and commercially-available (*R*)-glycidol as the starting compound. The present synthesis involves a couple of divergent strategies, including *syn*- and anti-selective Mukaiyama aldol additions and hetero-Diels-Alder reactions.

Catalytic asymmetric Mukaiyama aldol reactions of readily-available *bis*(TMSO)diene (Chan's diene) with (*S*)-2-benzyloxyhexanal derived from (*R*)-glycidol afforded a *syn*-aldol adduct with high diastereoselectivity and enantioselectivity. Diastereoselective Mukaiyama aldol reactions mediated by catalytic achiral Lewis acids directly produced not only a (1'*S*,6*S*)-pyrone precursor via the *syn*-aldol adduct using TiCl₄, but also (1'*S*,6*R*)-pyrone precursor derived from an anti-aldol adduct using ZrCl₄ in a stereocomplementary manner.

A hetero-Diels-Alder reaction of similarly available mono(TMSO)diene (Brassard's diene) with (*S*)-2-benzyloxyhexanal produced the (1'*S*,6*S*)-pyrone precursor promoted by Eu(fod)₃ and the (1'*S*,6*R*)-pyrone precursor EtAlCl₂.

Debenzylation of (1'*S*,6*S*)- and (1'*S*,6*R*)-precursors furnished natural (–)(–)-pestalotin and unnatural (+)-epipestalotin, respectively. The unnatural (+)-pestalotin and (–)-epipestalotin were successfully synthesized by Mitsunobu inversion of (–)-pestalotin and (+)-epipestalotin, respectively, in a divergent manner. All four chiral pestalotin diastereomers obtained possessed high chemical and optical purities (optical rotations, ¹H-NMR, ¹³C-NMR, and HPLC measurements).

The present divergent method affords concise access to asymmetric syntheses directed for these types of compounds with consecutive chiral dihydroxy groups, and is useful for accessible asymmetric Mukaiyama aldol reactions and relevant hetero-Diels-Alder reactions.

Copies of the ¹H, ¹³C-NMR spectra for compounds *syn*-3, (1'*S*,6*S*)-5, (–)-pestalotin, (1'*S*,6*R*)-5, (+)-epipestalotin are available in the Supplementary Information. Copies of the HPLC chromatogram of (±)-8, (*S*)-8, (±)-3, *syn*-3, (±)-5, (1'*S*,6*S*)-5, (1'*S*,6*R*)-5, (±)-pestalotin, (+)-pestalotin, (–)-pestalotin, (±)-epipestalotin, (+)-epipestalotin, and (–)-epipestalotin are available in the Supplementary Information.

Supplementary Materials: The following are available online. Supplementary 1: ¹H, ¹³C-NMR spectra for compounds *syn*-3, (1'*S*,6*S*)-5, (–)-pestalotin, (1'*S*,6*R*)-5, (+)-epipestalotin, Supplementary 2: HPLC chromatogram of (±)-8, (*S*)-8, (±)-3, *syn*-3, (±)-5, (1'*S*,6*S*)-5, (1'*S*,6*R*)-5, (±)-pestalotin, (+)-pestalotin, (–)-pestalotin, (±)-epipestalotin, (+)-epipestalotin, and (–)-epipestalotin.

Author Contributions: M.M., K.N., and T.F. contributed the majority of experiments. Y.T. conceived and designed the project, and prepared the whole manuscript. All authors have read and agreed to the published version of the manuscript.

Funding: This research was partially supported by Grant-in-Aids for Scientific Research on Basic Area (B) "18350056", Basic Areas (C) 15K05508, and Priority Areas (A) "17035087" and "18037068", and Exploratory Research "17655045" from the Ministry of Education, Culture, Sports, Science and Technology (MEXT).

Acknowledgments: We thank Momoyo Kawamoto and Daiki Ueura in our laboratory for their help of some experiments.

Conflicts of Interest: The authors declare no conflict of interest.

Dedication: This article is dedicated to the late professor Teruaki Mukaiyama who deceased in 2018 and the late professor Kenji Mori who deceased in 2019. One of the authors (Y.T) offer his warmest congratulations to Professor Ben L. Feringa (University of Groningen, The Netherlands) on being awarded the 2016 Nobel Prize in Chemistry.

References

1. Davies-Coleman, M.T.; Rivett, D.E.A. Naturally Occurring 6-Substituted 5,6-Dihydro- α -pyrones. In *Progress in the Chemistry of Organic Natural Products*; Herz, W., Grisebach, H., Kirby, G.W., Tamm, C., Eds.; Springer: New York, NY, USA, 1989; Volume 55, pp. 1–35. ISBN 978-3-7091-9004-3.
2. Sotheeswaran, S. Kawa and the Australian aborigine. *Chem. Aust.* **1987**, 377–378.

3. Kimura, Y.; Katagiri, K.; Tamura, S. Structure of pestalotin, a new metabolite from *Pestalotia cryptomeriaecola*. *Tetrahedron Lett.* **1971**, *12*, 3137–3140. [[CrossRef](#)]
4. Kimura, Y.; Tamura, S. Isolation and structure of pestalotin, a gibberellin synergist from *Pestalotia cryptomeriaecola*. *Agric. Biol. Chem.* **1972**, 1925–1930. [[CrossRef](#)]
5. Kimura, Y.; Suzuki, A.; Tamura, S.; Mori, K.; Oda, M.; Matsui, M. Biological activity of pestalotins on the elongation growth of rice seedlings. *Plant Cell Physiol.* **1977**, *18*, 1177–1179. [[CrossRef](#)]
6. Ellestad, G.A.; McGahren, W.J.; Kunstmann, M.P. Structure of a new fungal lactone, LL-P880 α , from an unidentified *Penicillium* species. *J. Org. Chem.* **1972**, *37*, 2045–2047. [[CrossRef](#)]
7. Seebach, D.; Meyer, H. Synthesis of (\pm)-Pestalotin and (\pm)-Epipestalotin and of Optically Pure (–)-Pestalotin by Asymmetric Synthesis. *Angew. Chem. Int. Ed.* **1974**, *13*, 77. [[CrossRef](#)]
8. Mori, K.; Oda, M.; Matsui, M. Synthesis of (+)-(6R:1'R)-pestalotin and (+)-(R1'S)-epipestalotin. *Tetrahedron Lett.* **1976**, *17*, 3173–3174. [[CrossRef](#)]
9. Mori, K.; Otsuka, T.; Oda, M. Synthetic microbial chemistry V. Synthesis of all of the four possible stereoisomers of pestalotin, a gibberellin synergist isolated from *Pestalotia cryptomeriaecola* Sawada. *Tetrahedron* **1984**, *40*, 2929–2934. [[CrossRef](#)]
10. Masaki, Y.; Nagata, K.; Serizawa, Y.; Kaji, K. Facile and rapid entry to functionalized and optically active pyrans from tartaric acid by way of 6,8-dioxabicyclo[3.2.1]Octanes. Application to the synthesis of (–)-(6S,1'S)-pestalotin. *Tetrahedron Lett.* **1984**, *25*, 95–96. [[CrossRef](#)]
11. Brassard, P.; Savard, J. Regiospecific syntheses of quinones using vinylketene acetals derived from unsaturated esters. *Tetrahedron Lett.* **1979**, 4911–4914. [[CrossRef](#)]
12. Midland, M.M.; Graham, R.S. High threo diastereoselectivity via europium(III)-catalyzed cyclocondensation of a silyloxy diene with α -alkoxy aldehydes. Synthesis of (–)pestalotin. *J. Am. Chem. Soc.* **1984**, *106*, 4294–4296. [[CrossRef](#)]
13. Soriente, A.; De Rosa, M.; Stanzione, M.; Villano, R.; Scettri, A. An efficient asymmetric aldol reaction of Chan's diene promoted by chiral Ti(IV)–BINOL complex. *Tetrahedron Asymmetry* **2001**, *12*, 959. [[CrossRef](#)]
14. Hagiwara, H.; Kimura, K.; Uda, H. Highly diastereoselective titanium tetrachloride-mediated aldol condensation of the bistrimethylsilyl enol ether of acetoacetic ester with 2-benzyloxyhexanal. A synthesis of (–)pestalotin. *J. Chem. Soc. Chem. Commun.* **1986**, 860–861. [[CrossRef](#)]
15. Hagiwara, H.; Kimura, K.; Uda, H. High diastereoselection in the aldol reaction of the bis(trimethylsilyl enol ether) of methyl acetoacetate with 2-(benzyloxy)hexanal: synthesis of (–)pestalotin. *J. Chem. Soc. Perkin Trans. 1* **1992**, 693–700. [[CrossRef](#)]
16. Zhang, J.; Curran, D.P. Stereoselective synthesis of 1,2-diols by the cycloadditive strategy: Total synthesis of (\pm)-exo-brevicomine and (\pm)- and (–)-pestalotin. *J. Chem. Soc. Perkin Trans. 1* **1991**, 2627–2631. [[CrossRef](#)]
17. Wang, Z.M.; Shen, M. An efficient synthesis of (–)-pestalotin and its enantiomer using Sharpless asymmetric dihydroxylation. *Tetrahedron Asymmetry* **1997**, *8*, 3393–3396. [[CrossRef](#)]
18. Kumar, A.S.; Bhaket, P.; Rao, B.V. Stereoselective synthesis of (–)-pestalotin. *Arkiboc* **2005**, 74–82. [[CrossRef](#)]
19. Miyakado, M.; Inoue, S.; Tanabe, Y.; Watanabe, K.; Ohno, N.; Yoshioka, H.; Mabry, T. Podblastin A, B and C. New antifungal 3-acyl-4-hydroxy-5,6-dihydro-2-pyrones obtained from *Podophyllum peltatum* L. *Chem. Lett.* **1982**, 1539. [[CrossRef](#)]
20. Ayer, W.A.; Villar, J.D.F. Metabolites of *Lachnellulafuscosanguinea* (Rehm). Part 1. The isolation, structure determination, and synthesis of lachnelluloic acid. *Can. J. Chem.* **1985**, *63*, 1161. [[CrossRef](#)]
21. Brian, P.W.; Curtis, P.J.; Hemming, H.G.; Unwin, C.H.; Wright, J.M. Alternaric Acid, a Biologically Active Metabolic Product of the Fungus *Alternaria solani*. *Nature* **1949**, *164*, 534. [[CrossRef](#)]
22. Fujiwara, T.; Takeshi, T.; Nakata, K.; Nakatsuji, H.; Tanabe, Y. Asymmetric total syntheses of two 3-acyl-5,6-dihydro-2H-pyrones: (R)-podblastin-S and (R)-lachnelluloic acid with its verification of the absolute configuration. *Molecules* **2017**, *22*, 69. [[CrossRef](#)] [[PubMed](#)]
23. Nagase, R.; Oguni, Y.; Ureshino, S.; Mura, H.; Misaki, T.; Tanabe, Y. Asymmetric Ti-crossed Claisen condensation: application to concise asymmetric total synthesis of alternaric acid. *Chem. Commun.* **2013**, 49, 7001. [[CrossRef](#)] [[PubMed](#)]
24. Midland, M.M.; Koops, R.W. Asymmetric hetero Diels-Alder reaction of α -alkoxy aldehydes with activated dienes. The scope of Lewis acid chelation-controlled cycloadditions. *J. Org. Chem.* **1990**, *55*, 5058–5065. [[CrossRef](#)]

25. Faul, M.M.; Winneroski, L.L.; Krumrich, C.A.; Sullivan, K.A.; Gillig, J.R.; Neel, D.A.; Rito, C.J.; Jirousek, M.R. Macrocyclic Bisindolylmaleimides: Synthesis by Inter- and Intramolecular Cyclization. *J. Org. Chem.* **1998**, *63*, 1961–1973. [[CrossRef](#)]
26. Tsutsumi, T.; Moriyama, M.; Tanabe, Y. Catalytic Asymmetric Mukaiyama Aldol Addition using 1,3-Bis(siloxy)diene Promoted by a Ti(OiPr)₄/(S)-BINOL Catalyst. *Org. Synth.* **2020**, accepted.

Sample Availability: Not available.



© 2020 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>).