

Highly Selective Fluorescent Probe Based on Hydroxylation of Phenylboronic Acid Pinacol Ester for Detection of Tyrosinase in Cells

Huihui Li, Wei Liu, Fengyuan Zhang, Xinyue Zhu, Liqui Huang, and Haixia Zhang

Anal. Chem., **Just Accepted Manuscript** • DOI: 10.1021/acs.analchem.7b03681 • Publication Date (Web): 04 Dec 2017

Downloaded from <http://pubs.acs.org> on December 4, 2017

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.



1
2
3
4 **Highly Selective Fluorescent Probe Based on Hydroxylation of Phenylboronic**
5
6 **Acid Pinacol Ester for Detection of Tyrosinase in Cells**
7

8 Huihui Li^a, Wei Liu^a, Fengyuan Zhang^a, Xinyue Zhu^a, Liqiu Huang^a, Haixia Zhang^{a,*}
9

10 ^a State Key Laboratory of Applied Organic Chemistry and College of Chemistry and Chemical
11

12
13 Engineering, Lanzhou University, Lanzhou 730000, China
14

15
16 * Corresponding author Tel.: +86 931 8912510; fax: +86 931 8912582;
17

18 E-mail address: zhanghx@lzu.edu.cn
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

ABSTRACT

The detection of tyrosinase, a biomarker for melanoma, is of great significance. Herein, a fluorescent tyrosinase probe, with resorufin as the fluorophore and *m*-tolylboronic acid pinacol ester as the receptor, is proposed. The response relies on tyrosinase-catalyzed hydroxylation of phenylboronic acid pinacol ester at adjacent position followed by 1,6-rearrangement-elimination to release resorufin. This probe well quantifies tyrosinase in the range from 1 to 100 U mL⁻¹ with detection limit of 0.5 U mL⁻¹. Importantly, the probe exhibits high selectivity for tyrosinase over other biological substances including reactive oxygen species. In addition, it is successfully applied to imaging of tyrosinase in cells. This probe provides a novel platform for selective detection of tyrosinase in biosystem.

KEYWORDS: Fluorescent Probe; Tyrosinase; Hydroxylation; Phenylboronic Acid Pinacol Ester

INTRODUCTION

In mammals, melanocytes as melanin-producing cells are predominantly localized in skin, hair and eyes.^{1,2} The malignant mutation of melanocytes results in a kind of skin cancer called melanoma. Tyrosinase (EC 1.14.18.1), a key enzyme in melanogenesis, catalyzes the hydroxylation of monophenols to *o*-diphenols and the subsequent oxidation of *o*-diphenols to *o*-quinones.^{3,4} Tyrosinase is a biomarker for melanoma, because its expression is mainly limited to melanocytes and its level is up-regulated during malignant mutation.⁵⁻⁸ Therefore, the detection of tyrosinase in biosystem is of great importance.

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
Fluorescent probes have drawn much attention to monitor tyrosinase in biosystem due to their high sensitivity, simplicity and noninvasion. Phenols, natural substrate of tyrosinase, are usually designed as the receptors of fluorescent tyrosinase probes.⁹⁻¹⁶ Most fluorescent tyrosinase probes are based on tyrosinase-catalyzed oxidation of phenols to quinones,⁹⁻¹⁴ but their selectivity is not perfect and show response to reactive oxygen species (ROS), because phenols are also oxidized to quinones by ROS.¹⁷⁻²⁰ Considering that ROS do not have the function of hydroxylation, fluorescent tyrosinase probes based on tyrosinase-catalyzed hydroxylation of 3-methylphenol at adjacent position have been reported recently,^{15,16} and they exhibit high selectivity and show no response to ROS. Given the above, highly selective fluorescent probes for tyrosinase are rare, and efforts are still needed.

30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
In this work, fluorescent tyrosinase probe 1, synthesized via linking *m*-tolylboronic acid pinacol ester to resorufin, is developed for the detection of tyrosinase in cells. Probe 1 exhibits high selectivity and shows no response to ROS. As shown in Figure 1, phenylboronic acid pinacol ester in probe 1 is hydroxylated at adjacent position by tyrosinase, and then resorufin is released via 1,6-rearrangement-elimination. Resorufin is chosen as the fluorophore owing to its long wavelengths ($\lambda_{\text{ex/em}} = 550/583 \text{ nm}$), high fluorescence quantum yield ($\Phi = 0.75$) and efficient fluorescence quenching via the alkylation of 7-hydroxy.²¹⁻²³

50 51 52 53 54 55 56 57 58 59 60 **EXPERIMENTAL SECTION**

Reagents. 3-(Bromomethyl)phenylboronic acid, phenylboronic acid pinacol ester, 2-hydroxyphenylboronic acid pinacol ester, pinacol and resorufin were

1
2
3
4 purchased from Energy Chemical (Shanghai, China). Tyrosinase from mushroom,
5
6 lipase from *Candida rugosa* and trypsin from bovine pancreas were obtained from
7
8 Sigma-Aldrich (Shanghai, China). Catalase from bovine liver, glucose oxidase from
9
10 *Aspergillus niger*, kojic acid, hydrogen peroxide, sodium hypochlorite, *tert*-butyl
11
12 hydroperoxide (TBHP), potassium dioxide, glucose, vitamin C, vitamin B6, glycine,
13
14 glutamic acid, cysteine, glutathione, tyrosine and creatinine were purchased from
15
16 Aladdin Chemistry Co. Ltd (Shanghai, China). The water used throughout the
17
18 experiments was supplied by Milli-Q system (Millipore, USA). Cell lines (B16 and
19
20 HepG2) were obtained from Cell Bank of Chinese Academy of Sciences (Shanghai,
21
22 China). Other reagents were from qualified reagent suppliers. ROS including H₂O₂,
23
24 TBHP, ClO⁻, *tert*-butoxy radical (TBO[·]), HO[·], ¹O₂, O₂⁻, ONOO⁻ were prepared
25
26 according to reported methods.^{24,25} The stock solution of probe 1 (5.0 mM) was
27
28 prepared in DMSO and stored at 4 °C in the dark.
29
30
31
32
33
34

35 **Instrumentation.** NMR spectra were measured with a JNM-ECS 400M
36
37 Instrument (Japan). HRMS were measured using a Bruker Daltonics APEX II 47e
38
39 FT-ICR Spectrometer (America). Fluorescence spectra were recorded on a RF-5301pc
40
41 Fluorescence Spectrometer (Japan). Absorption spectra were measured on a TU-1810
42
43 UV-VIS Spectrophotometer (China). Fluorescence imaging was conducted on a
44
45 DMI4000B Inverted Fluorescence Microscope (Germany).
46
47
48
49

50 **Synthesis of Compound 1.** 3-(Bromomethyl)phenylboronic acid (43 mg, 0.20
51
52 mmol) and pinacol (28 mg, 0.24 mmol) were dissolved in diethyl ether (15 mL). The
53
54 solution was stirred overnight at room temperature, and then concentrated under
55
56
57
58
59
60

1
2
3 vacuum. The crude product was purified by silica gel column chromatography using
4 petroleum ether/ethyl acetate (2/1, v/v) as eluent. Compound 1 was obtained as white
5 solid in 64% yield (38 mg). ^1H NMR (400 MHz, CDCl_3): δ 7.82 (s, 1H), 7.74 (d, 1H),
6 7.50 (d, 1H), 7.36 (t, 1H), 4.51 (s, 2H), 1.35 (s, 12H). ^{13}C NMR (400 MHz, CDCl_3): δ
7 137.18, 135.27, 134.87, 132.03, 128.33, 84.01, 33.55, 24.94. HRMS (ESI, m/z) Calcd.
8 for $[\text{C}_{13}\text{H}_{18}\text{BBrO}_2 + \text{H}^+]$: 297.0656, found: 297.0655.
9
10
11
12
13
14
15
16
17

18 **Synthesis of Probe 1.** To a suspension of resorufin (26 mg, 0.12 mmol) in
19 anhydrous DMF (10 mL), K_2CO_3 (25 mg, 0.18 mmol) and compound 1 (24 mg, 0.08
20 mmol) were added. After stirred overnight at room temperature, the mixture was
21 diluted with ethyl acetate (50 mL) and then washed with saturated NaCl. The organic
22 phase was collected and concentrated under vacuum. The crude product was purified
23 by silica gel column chromatography using petroleum ether/ethyl acetate (4/1, v/v) as
24 eluent. Probe 1 was obtained as orange solid in 47% yield (16 mg). ^1H NMR (400
25 MHz, CDCl_3): δ 7.88 (s, 1H), 7.82 (d, 1H), 7.71 (d, 1H), 7.55 (d, 1H), 7.45–7.41 (m,
26 2H), 7.02 (dd, 1H), 6.89 (d, 1H), 6.84 (dd, 1H), 6.33 (d, 1H), 5.17 (s, 2H), 1.36 (s,
27 12H). ^{13}C NMR (400 MHz, CDCl_3): δ 186.41, 162.81, 149.91, 145.73, 145.70,
28 135.12, 134.79, 134.33, 134.07, 131.69, 130.61, 128.57, 128.37, 114.38, 106.84,
29 101.13, 84.13, 71.01, 24.98. HRMS (ESI, m/z) Calcd. for $[\text{C}_{25}\text{H}_{24}\text{BNO}_5 + \text{H}^+]$:
30 430.1820, found: 430.1819.
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49

50 **General Procedure for Spectra Measurement.** PBS buffer (930 μL , 10 mM,
51 pH 7.4), probe 1 (20 μL , 250 μM) and tyrosinase (50 μL) were placed into a 2.0 mL
52 centrifugal tube. The solution was incubated at 37 $^\circ\text{C}$ for 5 h, and then its fluorescence
53
54
55
56
57
58
59
60

1
2
3
4 spectrum was recorded.

5
6 **Cell Culture and Fluorescence Imaging.** The two cell lines (B16 and HepG2)
7
8 were cultured in RPMI 1640 medium supplemented with fetal bovine serum (10%),
9
10 penicillin (100 U mL^{-1}) and streptomycin (100 U mL^{-1}) at $37 \text{ }^\circ\text{C}$ under 5% CO_2
11
12 environment. For fluorescence imaging, the adherent cells in a 6-well plate were
13
14 incubated with probe 1 ($2.5 \text{ } \mu\text{M}$) for 5 h and then washed with PBS three times. In
15
16 inhibitor experiment, cells were pretreated with kojic acid ($200 \text{ } \mu\text{M}$) for 2 h, then
17
18 washed with PBS three times, and finally incubated with probe 1 ($2.5 \text{ } \mu\text{M}$) for 5 h.
19
20
21

22
23 **Cytotoxicity Assay.** Cytotoxic effect of probe 1 was measured with colorimetric
24
25 methyl thiazolyl tetrazolium (MTT) assay. B16 cells (1×10^4 cells/well) were seeded
26
27 in a 96-well plate, then incubated for 24 h in the presence of probe 1 at different
28
29 concentrations (0, 1.25, 2.5, 5, 7.5 μM), followed by adding MTT solution (0.5
30
31 mg/mL) to each well. The culture medium was removed after 4 h and 150 μL of
32
33 DMSO was added to each well to dissolve the formazan crystals. Then the absorbance
34
35 at 490 nm was measured with a microplate reader.
36
37
38

39 40 **RESULTS AND DISCUSSION**

41
42 **Response of Probe 1 to Tyrosinase.** Absorption and fluorescence spectra of
43
44 probe 1 before and after adding tyrosinase were investigated. As shown in Figure 2A,
45
46 probe 1 exhibits an absorption band at 475 nm, and after addition of tyrosinase a new
47
48 absorption band at 570 nm appears concomitant with a color change from colorless to
49
50 pink. As shown in Figure 2B, probe 1 shows almost no emission, due to the protection
51
52 of 7-hydroxy in resorufin,^{22,23} and fluorescence intensity at 583 nm enhances
53
54
55
56

1
2
3 significantly after addition of tyrosinase. As shown in Figure S1, resorufin exhibits an
4 absorption band at 570 nm and an emission band at 583 nm, demonstrating that the
5 response originates from the generation of resorufin. Kojic acid is an inhibitor of
6 tyrosinase.²⁶ The presence of kojic acid leads to a weaker response (Figure S2A), and
7 kojic acid has no effect on the fluorescence intensity of resorufin and probe 1 (Figure
8 S2B), indicating that the response is ascribed to tyrosinase-catalyzed reaction of probe
9
10
11
12
13
14
15
16
17
18 1.

19
20 The experimental conditions were then optimized, including pH, temperature and
21 time. As shown in Figure S3A, the change in pH, especially in pH 7–8, affects the
22 response of probe 1 to tyrosinase seriously, and the optimal response is obtained at pH
23 8.0. As shown in Figure S3B, temperature also has some effect on the response, and
24 the optimal response is achieved at temperature 37–42 °C. Probe 1 is intended for
25 application in cells, and thus physiological conditions (pH 7.4 and temperature 37 °C)
26 are used in this study. As shown in Figure S4A, time course reveals that
27 tyrosinase-catalyzed reaction of probe 1 is time-dependent, and 5 h is employed in
28 subsequent experiments with overall consideration of time and selectivity.
29 Additionally, kinetic parameters for tyrosinase-catalyzed reaction of probe 1 were
30 determined via Lineweaver-Burk analysis (Figure S4B). Michaelis constant (K_m) and
31 maximum initial reaction rate (V_{max}) are calculated to be 6.5 μM and 0.9 nM min^{-1} ,
32 respectively.
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51

52 Under the optimized conditions, the response of probe 1 to different
53 concentrations of tyrosinase was investigated. As expected (Figure 3A), a gradual
54
55
56

1
2
3 increase of fluorescence intensity is observed with the increase of tyrosinase
4 concentration. Besides, probe 1 exhibits a good linear response to tyrosinase in the
5
6 range of 1–100 U mL⁻¹ with detection limit of 0.5 U mL⁻¹ (S/N = 3) (Figure 3B).
7
8
9

10 **Response Mechanism.** It has been reported that some boronic esters are prone to
11 hydrolysis.^{27,28} Therefore, the hydrolytic property of probe 1 was evaluated. As shown
12 in Figure S5, after incubation at physiological conditions (pH 7.4 and temperature
13 37 °C) for 5 h, the chromatographic peak intensity of probe 1 is not changed and no
14 new chromatographic peak appears, indicating the hydrolytic stability of probe 1
15 under experimental conditions. This occurs because the hydrolysis of boronic esters is
16 structure-dependent.^{27,28}
17
18
19
20
21
22
23
24
25
26
27

28 The catalytic function of tyrosinase is hydroxylation at adjacent position, and the
29 two copper ions in the active site of tyrosinase capture substrate via coordination.^{3,4}
30 Phenylboronic acid pinacol ester can coordinate with copper ions,²⁹ and thus we
31 speculate that phenylboronic acid pinacol ester is a substrate of tyrosinase. As a result,
32 phenylboronic acid pinacol ester in probe 1 is hydroxylated at adjacent position by
33 tyrosinase followed by 1,6-rearrangement-elimination to produce resorufin and
34 compound 2 (Figure 1).
35
36
37
38
39
40
41
42
43
44

45 To verify the above speculation, the reaction products of probe 1 with tyrosinase
46 were analyzed with HPLC and HRMS. As shown in Figure S6, after addition of
47 tyrosinase, the chromatographic peak of probe 1 decreases concomitant with the
48 emergence of the chromatographic peak of resorufin. Additionally, HRMS of probe 1
49 after addition of tyrosinase displays the signals of resorufin (M + H⁺, *m/z* 214.0496)
50
51
52
53
54
55
56
57
58
59
60

1
2
3 and compound 2 ($M + H^+$, m/z 233.1355) (Figure S7). These results demonstrate that
4
5
6 the reaction of probe 1 with tyrosinase produces resorufin and compound 2,
7
8 confirming the above speculation.
9

10
11 To further verify the above speculation, the reaction of phenylboronic acid
12
13 pinacol ester with tyrosinase was investigated. As shown in Figure S8, after addition
14
15 of tyrosinase, the chromatographic peak of phenylboronic acid pinacol ester decreases,
16
17 confirming that phenylboronic acid pinacol ester is a substrate of tyrosinase, but the
18
19 chromatographic peak of 2-hydroxyphenylboronic acid pinacol ester does not appear,
20
21 because 2-hydroxyphenylboronic acid pinacol ester can further react with tyrosinase
22
23 (Figure S9). 2-Hydroxyphenylboronic acid pinacol ester is not only a phenylboronic
24
25 acid pinacol ester derivative but also a phenol derivative, and thus it can react with
26
27 tyrosinase. These results further confirm the above speculation.
28
29
30
31

32
33 **Selectivity.** The response of probe 1 to other biological substances was
34
35 investigated, including inorganic salts, glucose, vitamins, amino acids, creatine, urea,
36
37 enzymes and ROS. As shown in Figure 4, it is observed that these biological
38
39 substances including ROS have little effect on the fluorescence intensity of probe 1.
40
41 The above result suggests the high selectivity of probe 1 toward tyrosinase, and this is
42
43 attributed to that only tyrosinase has the function of hydroxylation.
44
45
46

47
48 Although H_2O_2 is not responded by probe 1, it has two-side influence on the
49
50 response. On one hand, H_2O_2 can oxidize phenylboronic acid pinacol ester to
51
52 phenol.^{30,31} Phenol, the natural substrate of tyrosinase, consequentially can be
53
54 hydroxylated at adjacent position by tyrosinase with a faster rate.¹⁵ On the other hand,
55
56
57

1
2
3 excess H_2O_2 can further oxidize phenol to quinone.¹⁷ As expected (Figure S10), after
4 adding H_2O_2 (50 μM) the response increases, but with adding more H_2O_2 (100 μM)
5
6 the response further becomes weaker. H_2O_2 and tyrosinase are mainly produced in
7
8 different organelles^{32,33}, but they coexist in cells because of molecular diffusion.
9
10
11
12
13 Thereby, H_2O_2 can affect the imaging of tyrosinase in cells.
14

15
16 Table S1 lists the comparison between probe 1 and reported fluorescent
17
18 tyrosinase probes⁹⁻¹⁶. It is found that probe 1 exhibits high selectivity, and it is also
19
20 one of the excellent probes for tyrosinase.
21

22
23 **Fluorescence Imaging of Tyrosinase in Cells.** To begin with, the cytotoxicity of
24
25 probe 1 was evaluated using a conventional MTT assay. As shown in Figure S11,
26
27 when B16 cells are treated with probe 1 at a concentration of no more than 2.5 μM for
28
29 24 h, no significant cytotoxic response (cell viability $\geq 85\%$) is observed. Thereby,
30
31 2.5 μM of probe 1 is used for cell imaging.
32
33
34

35
36 Then fluorescence imaging of tyrosinase was conducted in B16 cells, a
37
38 melanoma cell line that expresses tyrosinase. As shown in the image a of Figure 5,
39
40 B16 cells display almost no fluorescence, which benefits from the long wavelengths
41
42 of probe 1. As shown in the images b and c of Figure 5, B16 cells treated with probe 1
43
44 exhibit strong fluorescence, but B16 cells treated with kojic acid and probe 1 show
45
46 weak fluorescence, demonstrating that probe 1 can be used to monitor intracellular
47
48 tyrosinase activity. Fluorescence imaging of tyrosinase in HepG2 cells
49
50 (non-melanoma cell line) was also carried out. As shown in the image d of Figure 5,
51
52 HepG2 cells treated with probe 1 exhibit almost no fluorescence, and this is because
53
54
55
56
57
58
59
60

1
2
3 the expression of tyrosinase is mainly limited to melanocytes.^{7,8}
4
5

6 CONCLUSIONS

7
8 In summary, a fluorescent probe based on tyrosinase-catalyzed hydroxylation of
9
10 phenylboronic acid pinacol ester is developed for the detection of tyrosinase in cells.
11
12 The probe exhibits high selectivity for tyrosinase over other biological substances
13
14 including ROS, but it does not avoid the interference of H₂O₂. This probe provides an
15
16 alternative method for selective detection of tyrosinase in biosystem.
17
18
19

20 ACKNOWLEDGEMENTS

21
22
23 This work is financially supported by National Natural Science Foundation of
24
25 China (No. 21375052 and 21575055).
26
27

28 ASSOCIATED CONTENT

29
30 **Supporting Information.** Additional information as noted in the text (Figures
31
32 S1–S11, Table S1); ¹H NMR, ¹³C NMR, HRMS of compound 1 and probe 1.
33
34

35 REFERENCES

- 36
37 (1) Millet, A.; Martin, A. R.; Ronco, C.; Rocchi, S.; Benhida, R. *Med. Res. Rev.* **2017**, *37*, 98–148.
38
39 (2) Han, R.; Baden, H. P.; Brissette, J. L.; Weiner, L. *Pigm. Cell Res.* **2002**, *15*, 290–297.
40
41 (3) Rolff, M.; Schottenheim, J.; Decker, H.; Tuzcek, F. *Chem. Soc. Rev.* **2011**, *40*, 4077–4098.
42
43 (4) Solem, E.; Tuzcek, F.; Decker, H. *Angew. Chem. Int. Ed.* **2016**, *55*, 2884–2888.
44
45 (5) Vargas, A. J.; Sittadjody, S.; Thangasamy, T.; Mendoza, E.; Limesand, K. H.; Burd, R. *Integr.*
46
47 *Cancer Ther.* **2011**, *10*, 328–340.
48
49 (6) Jawaid, S.; Khan, T. H.; Osborn, H. M. I.; Williams, N. A. O. *Anticancer Agents Med. Chem.*
50
51 **2009**, *9*, 717–727.
52
53
54
55
56
57
58
59
60

- 1
2
3
4 (7) Hearing, V. J.; Jimenez, M. *Int. J. Biochem.* **1987**, *19*, 1141–1147.
5
6 (8) Jordan, A. M.; Khan, T. H.; Malkin, H.; Osborn, H. M. I. *Bioorg. Med. Chem.* **2002**, *10*,
7
8 2625–2633.
9
10 (9) Kim, T. I.; Park, J.; Park, S.; Choi, Y.; Kim, Y. *Chem. Commun.* **2011**, *47*, 12640–12642.
11
12 (10) Li, Z. P.; Wang, Y. F.; Zhang, X.; Zeng, C. C.; Hu, L. M.; Liang, X. J. *Sens. and Actuators B*
13
14 **2017**, *242*, 189–194.
15
16 (11) Yan, S. Y.; Huang, R.; Wang, C. C.; Zhou, Y. M.; Wang, J. Q.; Fu, B. S.; Weng, X. C.; Zhou,
17
18 X. *Chem. Asian J.* **2012**, *7*, 2782–2785.
19
20 (12) Zhou, J.; Shi, W.; Li, L. H.; Gong, Q. Y.; Wu, X. F.; Li, X. H.; Ma, H. M. *Anal. Chem.* **2016**,
21
22 88, 4557–4564.
23
24 (13) Li, X. H.; Shi, W.; Chen, S. M.; Jia, J.; Ma, H. M.; Wolfbeis, O. S. *Chem. Commun.* **2010**, *46*,
25
26 2560–2562.
27
28 (14) Wang, C. C.; Yan, S. Y.; Huang, R.; Feng, S.; Fu, B. S.; Weng, X. C.; Zhou, X. *Analyst* **2013**,
29
30 *138*, 2825–2828.
31
32 (15) Wu, X. F.; Li, X. H.; Li, H. Y.; Shi, W.; Ma, H. M. *Chem. Commun.* **2017**, *53*, 2443–2446.
33
34 (16) Wu, X. F.; Li, L. H.; Shi, W.; Gong, Q. Y.; Ma, H. M. *Angew. Chem. Int. Ed.* **2016**, *55*,
35
36 14728–14732.
37
38 (17) Yu, F. B.; Li, P.; Song, P.; Wang, B. S.; Zhao, J. Z.; Han, K. L. *Chem. Commun.* **2012**, *48*,
39
40 4980–4982.
41
42 (18) Kim, J.; Kim, Y. *Analyst* **2014**, *139*, 2986–2989.
43
44 (19) Peng, T.; Wong, N. K.; Chen, X. M.; Chan, Y. K.; Ho, D. H. H.; Sun, Z. N.; Hu, J. J.; Shen, J.
45
46 G.; EI-Nezami, H.; Yang, D. *J. Am. Chem. Soc.* **2014**, *136*, 11728–11734.
47
48
49
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3
4 (20) Zhang, W.; Li, P.; Yang, F.; Hu, X. F.; Sun, C. Z.; Zhang, W.; Chen, D. Z.; Tang, B. *J. Am.*
5
6 *Chem. Soc.* **2013**, *135*, 14956–14959.
7
8 (21) Bueno, C.; Villegas, M. L.; Bertolotti, S. G.; Previtali, C. M.; Neumann, M. G.; Encinas, M. V.
9
10 *Photochem. Photobiol.* **2002**, *76*, 385–390.
11
12 (22) Zhang, H. M.; Xu, C. L.; Liu, J.; Li, X. H.; Guo, L.; Li, X. M. *Chem. Commun.* **2015**, *51*,
13
14 7031–7034.
15
16 (23) Li, Z.; Li, X. H.; Gao, X. H.; Zhang, Y. Y.; Shi, W.; Ma, H. M. *Anal. Chem.* **2013**, *85*,
17
18 3926–3932.
19
20 (24) Peng, T.; Yang, D. *Org. Lett.* **2010**, *12*, 4932–4935.
21
22 (25) Zhang, J. J.; Li, C. W.; Zhang, R.; Zhang, F. Y.; Liu, W.; Liu, X. Y.; Lee, S. M. Y.; Zhang, H.
23
24 *X. Chem. Commun.* **2016**, *52*, 2679–2682.
25
26 (26) Chang, T. S. *Int. J. Mol. Sci.* **2009**, *10*, 2440–2475.
27
28 (27) Achilli, C.; Ciana, A.; Fagnoni, M.; Balduini, C.; Minetti, G. *Cent. Eur. J. Chem.* **2013**, *11*,
29
30 137–139.
31
32 (28) Bernardini, R.; Oliva, A.; Paganelli, A.; Menta, E.; Grugni, M.; De Munari, S.; Goldoni, L.
33
34 *Chem. Lett.* **2009**, *38*, 750–751.
35
36 (29) Li, M.; Ge, H. B.; Arrowsmith, R. L.; Mirabello, V.; Botchway, S. W.; Zhu, W. H.; Pascu, S. I.;
37
38 James, T. D. *Chem. Commun.* **2014**, *50*, 11806–11809.
39
40 (30) Xu, J.; Zhang, Y.; Yu, H.; Gao, X. D.; Shao, S. J. *Anal. Chem.* **2016**, *88*, 1455–1461.
41
42 (31) Zhang, W.; Liu, W.; Li, P.; Huang, F.; Wang, H.; Tang, B. *Anal. Chem.* **2015**, *87*, 9825–9828.
43
44 (32) Theos, A. C.; Tenza, D.; Martina, J. A.; Huibain, I.; Peden, A. A.; Sviderskava, E. V.; Stewart,
45
46 A.; Robinson, M. S.; Bennett, D. C.; Cutler, D. F.; Bonifacino, J. S.; Marks, M. S.; Raposo, G. A.
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4 *Mol. Biol. Cell* **2005**, *16*, 5356–5372.

5
6 (33) Marcu, R.; Rapino, S.; Trinei, M.; Valenti, G.; Marcaccio, M.; Pelicci, P. G.; Paolucci, F.;

7
8 Giorgio, M. *Bioelectrochemistry* **2012**, *85*, 21–28.
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

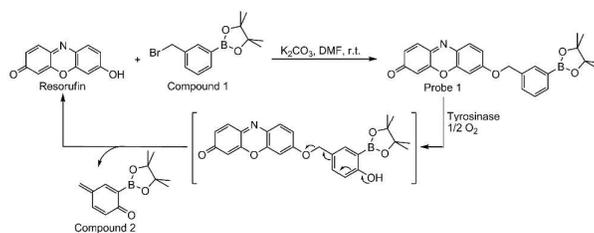


Figure 1. Synthetic route of probe 1 and tyrosinase-catalyzed reaction of probe 1.

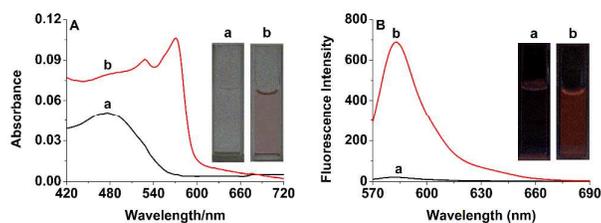


Figure 2. Absorption (A) and fluorescence (B) spectra of probe 1 ($5 \mu\text{M}$) before (a) and after (b) adding tyrosinase (150 U mL^{-1}) at 37°C for 10 h in DMSO/PBS buffer (1/49 v/v, 10 mM, pH 8.0).

$\lambda_{\text{ex}} = 550 \text{ nm}$, slit widths = 3.0/3.0 nm.

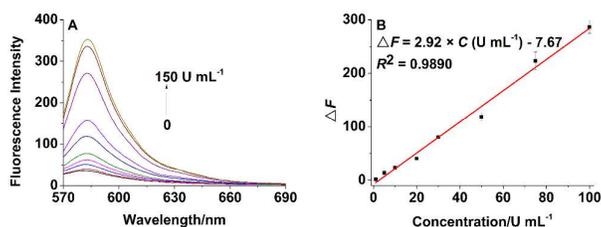


Figure 3. (A) Fluorescence spectra of probe 1 ($5 \mu\text{M}$) after adding different concentrations of tyrosinase (0, 1, 5, 10, 20, 30, 50, 75, 100, 150 U mL^{-1}). (B) Linear correlation between fluorescence enhancement and tyrosinase concentration ($1\text{--}100 \text{ U mL}^{-1}$). $\lambda_{\text{ex/em}} = 550/583 \text{ nm}$, slit widths = 3.0/5.0 nm.

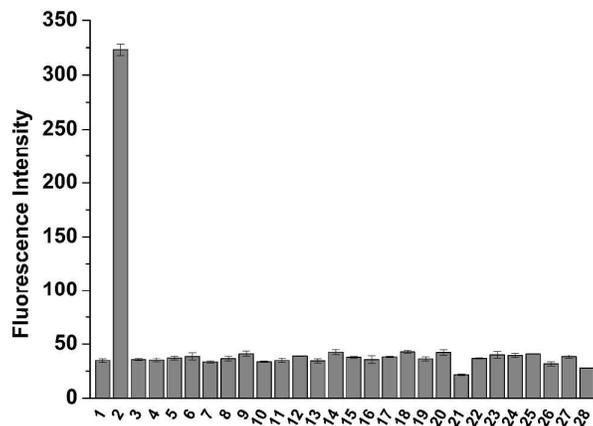


Figure 4. Fluorescence intensity of probe 1 (5 μM) after adding various substances. (1) none, (2) tyrosinase (150 U mL⁻¹), (3) KCl (150 mM), (4) MgCl₂ (2.5 mM), (5) CaCl₂ (2.5 mM), (6) FeCl₃ (100 μM), (7) glucose (10 mM), (8) vitamin C (1 mM), (9) vitamin B6 (1 mM), (10) glycine (1 mM), (11) glutamic acid (1 mM), (12) cysteine (100 μM), (13) glutathione (100 μM), (14) tyrosine (100 μM), (15) creatinine (10 mM), (16) urea (20 mM), (17) catalase (150 U mL⁻¹), (18) lipase (150 U mL⁻¹), (19) trypsin (15 mg L⁻¹), (20) glucose oxidase (150 U mL⁻¹), (21) H₂O₂ (50 μM), (22) TBHP (50 μM), (23) ClO⁻ (50 μM), (24) TBO[•] (50 μM), (25) HO[•] (50 μM), (26) ¹O₂ (50 μM), (27) O₂⁻ (50 μM), (28) ONOO⁻ (50 μM). $\lambda_{\text{ex/em}} = 550/583$ nm, slit widths = 3.0/5.0 nm.

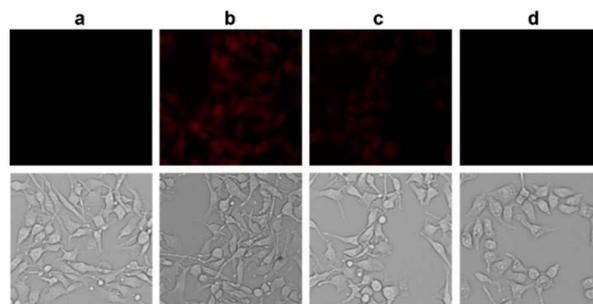


Figure 5. Fluorescence images. (a) B16 cells, (b) B16 cells treated with probe 1 (2.5 μM) for 5 h, (c) B16 cells pretreated with kojic acid (200 μM) for 2 h and then treated with probe 1 (2.5 μM) for 5 h, (d) HepG2 cells treated with probe 1 (2.5 μM) for 5 h. The second row shows the corresponding brightfield image.

Table of Contents

