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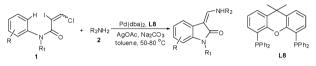
Synthesis of 3-(aminomethylene)-2-oxoindolines by palladium-catalyzed annulation of 3-chloro-2-iodo-*N*-arylacrylamides with amides or amines[†]

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A new palladium-catalyzed C-H bond activation-annulationamination tandem method was presented for selectively synthesizing 3-(aminomethylene)-2-oxoindolines. In the presence of Pd(dba)₂, xantphos (L8), AgOAc and Na₂CO₃, a variety of 3-chloro-2-iodo-*N*-arylacrylamides underwent the reaction with amides or amines to afford the corresponding 3-(aminomethylene)-2-oxoindolines in moderate to good yields.

3-Methyleneindolin-2-ones,¹⁻⁷ particularly 3-(aminomethylene)-2-oxoindolines,² are pharmacologically important components of numerous biologically active natural products and medicinally significant compounds.¹ The common classical methods used for the construction of the 3-methyleneindolin-2-one framework are the intermolecular condensation of oxindoles with diarvl ketones, and often they are limited due to nonavailability of suitably substituted starting precursors and unsatisfactory selectivity.3 Therefore, considerable effort has been made on the development of efficient metal-catalyzed methods for selectively synthesizing 3-methyleneindolin-2-ones.⁴⁻⁶ Among these methods, palladium-catalyzed reactions that involve the arene C-H bond functionalization are particularly attractive^{4,5} because they can avoid the use of expensive 2-haloanilide or 2-(alkynyl)phenylisocyanate reagents.⁶ Zhu and co-workers have disclosed Pd(0)-catalyzed arene C-H bond activation domino reactions of N-arylpropiolamides with aryl halide electrophilic reagents leading to the corresponding 3-(diarylmethylenyl)oxindoles under basic conditions.⁴ Our group also found that N-arylpropiolamides could react with a wide range of nucleophilic reagents, such as phthalimide, acids, ArI(OAc)₂ and alcohols, through a Pd(II)catalyzed oxidative C-H bond functionalization process with the aid of oxidants.⁵ However, only an amide, phthalimide, was employed to react with N-arylpropiolamides leading to (E)-(2-oxindolin-3-ylidene)phthalimides.^{5a} In view of the



Scheme 1 A new route to 3-(aminomethylene)-2-oxoindolines.

importance of 3-(aminomethylene)-2-oxoindolines,² the development of new methods involving the C–H bond activation strategy for their preparation is desirable.^{5*a*,6*j*} Here, we report a novel route to (*Z*)-3-(aminomethylene)-2-oxoindolines by palladium-catalyzed C–H bond activation–annulation–amination of 3-chloro-2-iodo-*N*-arylacrylamides with amides or amines (Scheme 1).^{8,9}

Our investigation began with the reaction of 3-chloro-2iodo-N-methyl-N-phenylacrylamide (1a) with benzamide (2a) to determine the optimal reaction conditions (Table 1).¹⁰ Initially, a series of ligands L1-L10 were examined, and xantphos (L8) gave the best results (entries 1-12). While the reaction of substrate 1a with 2a, Pd(OAc)₂, AgOAc and Cs₂CO₃ afforded a trace of the desired product 3 without ligands (entry 1), the yield of 3 was increased to 20% in the presence of PPh₃ (L1) (entry 2). Gratifyingly, the yield was enhanced to 45% using xantphos (L8) (entry 10). Notably, Na_2CO_3 was more effective than Cs_2CO_3 (entries 8 and 9). Subsequently, a number of other Pd catalysts, such as PdCl₂, $Pd(PPh_3)_2Cl_2$, $Pd(PPh_3)_4$ and $Pd(dba)_2$, were tested, and they were more efficient than $Pd(OAc)_2$ (entries 13–16). The results showed that the Ag effect affected the reaction (entries 16-22). The activities of other Ag salts, such as CF₃CO₂Ag, Ag₂O or Ag₂CO₃, were lowered slightly in terms of yields (entries 16–19). We found that the amount of AgOAc has a fundamental influence on the reaction (entries 20-22): the best results were obtained at 2 equiv. AgOAc. From the reaction temperature examination, it turned out that the reaction at 50 °C gave the highest yield after prolonging the reaction time (entries 23 and 24). However, the yield of 3 was reduced to 58% in the presence of 5 mol% Pd(dba)₂ (entry 25).

As shown in Table 2, the scope of both 3-chloro-2-iodo-Narylacrylamides 1 with amides 2 was explored under the optimal conditions. The results demonstrated that substrate 1b with a N-Bn group was reacted with amide 2a, Pd(dba)₂, L8, AgOAc and Na₂CO₃ smoothly to afford the target product 4 in 60% yield (entry 1), which was less active than

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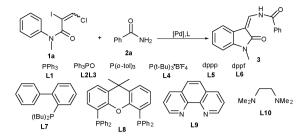
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 Table 1
 Screening optimal conditions^a

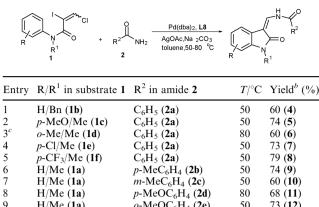


Entry	[Pd] (mol%)	Ligand	[Ag] (equiv.)	Base	$\operatorname{Yield}^{b}(\%)$
1	Pd(OAc) ₂	_	AgOAc (1)	Cs ₂ CO ₃	Trace
2	$Pd(OAc)_2$	L1	AgOAc (1)	Cs_2CO_3	20
3	$Pd(OAc)_2$	L2	AgOAc (1)	Cs_2CO_3	Trace
4	$Pd(OAc)_2$	L3	AgOAc (1)	Cs_2CO_3	30
5	$Pd(OAc)_2$	L4	AgOAc (1)	Cs_2CO_3	21
6	$Pd(OAc)_2$	L5	AgOAc (1)	Cs_2CO_3	Trace
7	$Pd(OAc)_2$	L6	AgOAc (1)	Cs_2CO_3	Trace
8	$Pd(OAc)_2$	L7	AgOAc (1)	Cs_2CO_3	30
9	$Pd(OAc)_2$	L7	AgOAc (1)	Na ₂ CO ₃	35
10	$Pd(OAc)_2$	L8	AgOAc (1)	Na ₂ CO ₃	45
11	$Pd(OAc)_2$	L9	AgOAc (1)	Na ₂ CO ₃	25
12	$Pd(OAc)_2$	L10	AgOAc (1)	Na ₂ CO ₃	Trace
13	PdCl ₂	L8	AgOAc (1)	Na ₂ CO ₃	67
14	Pd(PPh ₃) ₂ Cl ₂	L8	AgOAc (1)	Na ₂ CO ₃	65
15	$Pd(PPh_3)_4$	L8	AgOAc (1)	Na ₂ CO ₃	66
16	$Pd(dba)_2$	L8	AgOAc (1)	Na ₂ CO ₃	68
17	$Pd(dba)_2$	L8	$CF_3CO_2Ag(1)$	Na_2CO_3	46
18	$Pd(dba)_2$	L8	$Ag_2O(1)$	Na ₂ CO ₃	56
19	$Pd(dba)_2$	L8	$Ag_2CO_3(1)$	Na_2CO_3	50
20	$Pd(dba)_2$	L8	AgOAc (1.5)	Na ₂ CO ₃	79
21	$Pd(dba)_2$	L8	AgOAc (2)	Na ₂ CO ₃	85
22	$Pd(dba)_2$	L8	AgOAc (3)	Na ₂ CO ₃	80
23^c	$Pd(dba)_2$	L8	AgOAc (1)	Na ₂ CO ₃	86
24^d	$Pd(dba)_2$	L8	AgOAc (1)	Na ₂ CO ₃	72
$25^{c,e}$	$Pd(dba)_2$	L8	AgOAc (1)	Na ₂ CO ₃	58

^{*a*} Reaction conditions: **1a** (0.3 mmol), **2a** (0.45 mmol), [Pd] (10 mol%), ligand (10 mol%), [Ag] (1 equiv.) and base (2 equiv.) in the solvent (2 mL) at 80 °C for 8 h. ^{*b*} Isolated yield. ^{*c*} At 50 °C for 13 h. ^{*d*} At room temperature for 96 h. ^{*e*} 5 mol% of Pd(dba)₂ was added.

N-Me-substituted substrate 1a (entry 23 in Table 1). Consequently, a number of N-Me-substituted substrates 1c-1f were tested (entries 2-5). To our delight, substrates 1c-1f, bearing electron-donating or electron-deficient groups on the N-aryl ring, were compatible under the optimal conditions. Substrate 1d with an o-Me group, for instance, underwent the reaction with amide 2a in 60% yield (entry 3). Good yield was still achieved from substrate 1f with a p-CF₃ group (entry 5). Subsequently, a variety of amides 2b-2l were evaluated under the standard conditions (entries 6-16). Screening disclosed that the annulation had a wide range of amides compatibility, including arylamides, vinylamides, alkylamides and carbamates. Moreover, several functional groups, such as Me, MeO, Cl or Br, on the aryl ring were tolerated well (entries 6-11). It was noted that amide 2g with an o-Br group was successfully annulated with substrate 1a, Pd(dba)₂, L8, AgOAc and Na₂CO₃, furnishing the desired Br-containing product 14 in 78% yield (entry 11), which provides a route to introduction of a new group at the Br-substituted position. Gratifyingly, heteroarylamide 2h was also suitable for the reaction (entry 12). In the presence of Pd(dba)₂, L8, AgOAc and Na_2CO_3 , acrylamide (2i) or acetamide (2j) could react

Table 2 Palladium-catalyzed tandem reactions of 3-chloro-2-iodo-N-arylacrylamides (1) with amides (2)^{*a*}



p-CF ₃ /Me (II)	$C_6H_5(2a)$	50	/9 (8)				
H/Me (1a)	$p-MeC_6H_4$ (2b)	50	74 (9)				
H/Me (1a)	m-MeC ₆ H ₄ (2c)	50	60 (10)				
H/Me (1a)	p-MeOC ₆ H ₄ (2d)	80	68 (11)				
H/Me (1a)	o-MeOC ₆ H ₄ (2e)	50	73 (12)				
H/Me (1a)	$p-\text{ClC}_6\text{H}_4$ (2f)	50	93 (13)				
H/Me (1a)	o-BrC ₆ H ₄ (2g)	50	78 (14)				
H/Me (1a)	5-Mepyridin-2-yl (2h)	50	61 (15)				
H/Me (1a)	Vinyl (2i)	80	51 (16)				
H/Me (1a)	Me (2j)	50	65 (17)				
H/Me (1a)	OMe $(2k)$	50	75 (18)				
H/Me (1a)	OBn (2l)	50	80 (19)				
^{<i>a</i>} Reaction conditions: 1 (0.3 mmol), 2 (0.45 mmol), Pd(dba) ₂							
(10 mol%), L8 (10 mol%), AgOAc (2 equiv.) and Na ₂ CO ₃ (2 equiv.)							
	H/Me (1a) H/Me (1a)	$ \begin{array}{llllllllllllllllllllllllllllllllllll$	$ \begin{array}{llllllllllllllllllllllllllllllllllll$				

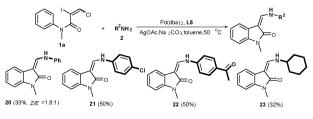
with substrate **1a** leading to the corresponding products **16** and **17** in moderate yields (entries 13 and 14). It is noteworthy that annulation of substrate **1a** with carbamates **2k** or **2l** is

successful under the same conditions (entries 15 and 16).

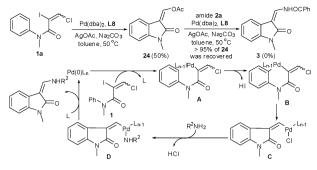
in toluene (2 mL) for 13 h. ^b Isolated yield. ^c For 24 h.

In light of the above results, a variety of amines were employed for the annulation reaction with 3-chloro-2-iodo-*N*-methyl-*N*-arylacrylamide (**1a**) under the optimal conditions (Scheme 2). However, the reaction between substrate **1a** and aniline gave a mixture of Z/E-isomers **20** in a low total yield. To our delight, amines with a *para*-substituent on the aryl ring were selectively treated with substrate **1a**, providing the corresponding products **21** and **22** in moderate yields. Cyclohexanamine reacted with substrate **1a**, Pd(dba)₂, **L8**, AgOAc and Na₂CO₃ in 32% yield.

To our surprise, an acetoxypalladation product 24, not (*Z*)-3-(chloromethylene)-1-methylindolin-2-one, was obtained exclusively without amides 2. However, product 24 could not be converted into product 3 under the present optimal conditions (Scheme 3). Therefore, a possible mechanism as outlined in Scheme 3 was proposed.^{2e,4,5,8,9} Oxidative addition of Pd(0) to the C–X bond of substrate 1 yields intermediate **A**, followed by arene C–H bond activation affording intermediate **B**.^{4,5,8} Sequential reductive elimination and oxidative addition of intermediate **B** results in intermediate **C**, which is supported



Scheme 2 Annulation of substrate (1a) with amines (2).



Scheme 3 Possible mechanism.

by the formation of acetoxypalladation product **24**. Insertion of intermediate **C** with an amine forms intermediate **D**.⁹ Finally, intermediate **D** undergoes reductive elimination to furnish the desired product and regenerate the active Pd(0) species. The role of Ag salts is to activate the arene C–H bond.⁸

In summary, we have described a new palladium-catalyzed tandem protocol for the synthesis of (Z)-3-(aminomethylene)-2-oxoindolines. This method allows two new bonds, a C–C bond and a C–N bond, formation in one-step by a sequential C–H bond activation/annulation/amination tandem process. Importantly, this method displays high substrates compatibility, which makes it more attractive for organic synthesis and industry.

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- 10 See the detailed data in ESI.† The stereochemistry of the products was assigned as the (*Z*)-isomer on the basis of the chemical shift of the olefinic proton according to the authoritative ¹H NMR spectra (see ref. 1*j*, 2*a* and 2*b*) and the X-ray single-crystal diffraction analysis of product **3**. (The CCDC number is 822997.).