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Tandem photoaffinity labeling of a target protein using a linker with biotin, alkyne and benzophenone groups and a bioactive small molecule with an azide group

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A novel linker containing biotin, alkyne and benzophenone groups (1) was synthesized to identify target proteins using a small molecule probe. This small molecule probe contains an azide group (azide probe) that reacts with an alkyne in 1 via an azidealkyne Huisgen cycloaddition. Cross-linking of benzophenone to the target protein formed a covalently bound complex consisting of the azide probe and the target protein via 1. The biotin was utilized via biotin-avidin binding to identify the cross-linked complex. To evaluate the effectiveness of 1, it was applied in a model system using an allene oxide synthase (AOS) from the model moss Physcomitrella patens (PpAOS1) and an AOS inhibitor that contained azide group (3). The cross-linked complex consisting of PpAOS1, 1 and 3 was resolved via SDS-PAGE and visualized using a chemiluminescent system. The method that was developed in this study enables the effective identification of target proteins.

Key words: Huisgen cycloaddition; linker; photoaffinity labeling; target protein; Western blotting

Photoaffinity probes have been utilized to identify target proteins in combination with small bioactive molecules with photophores and tags for detection.¹⁾ Aryl azide, benzophenone and 3-aryldiazirine are well-known photophores that are photoactivated by UV irradiation and used to cross-link small molecules and target proteins.^{1–3)} Tags such as biotin or fluorescent groups are used to detect proteins that are cross-linked with photoaffinity probes.^{4–6)} Photoaffinity probes are useful for identifying proteins bound to small bioactive molecules. However, the introduction of photophores and tags into small molecules often decreases both their affinity for the target protein and the bioactivity of the

small molecule, limiting the use of photoaffinity probes for protein identification. In contrast, Budin et al. used non-denaturing protein electrophoresis to determine that a complex containing a non-cross-linked target protein and an azide probe could be created through the Huisgen [3 + 2] cycloaddition known as click chemistry, which is an important bioorthogonal reaction.⁷⁻¹⁰⁾ However, separation of proteins by non-denaturing gel electrophoresis is less effective than separation by SDS-PAGE. To overcome the limitations of photoaffinity probes and the use of protein gel electrophoresis under non-denaturing conditions, this study provides a novel method to identify target proteins using an azide probe, which is a small bioactive molecule; this probe is created by introducing an azide group into a small molecule, which minimizes the loss of both affinity for the protein and the bioactivity of the original molecule. A linker containing alkyne, biotin, and photophore groups connects the azide probe with the target protein. The linker thus performs the function of binding the azide probe to the target protein and labeling the protein for detection, which are roles that conventional photoaffinity probes have fulfilled until recently. A summary of the method developed in this study is shown in Fig. 1: (1) the azide probe is incubated with the target protein; (2) the linker is added to a mixture of the azide probe and the target protein, and a conjugate between the linker and the azide probe is formed via Huisgen cycloaddition; (3) UV irradiation causes the complex to bind both the target protein and the conjugate of the azide probe through the linker's benzophenone group; and (4) the ternary complex that is formed between the target protein, the azide probe and the linker is identified via Western blotting. Therefore, the azide probe does not require a photophore and a tag that could prevent access to the binding site of the target protein. Thus, the method proposed in this study can be used to effectively identify target proteins.

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Abbreviations: AOS, allene oxide synthase; AOC, allene oxide cyclase; 12,13-EOT, (12,13*S*)-epoxy-(9*Z*,11*E*,15*Z*)-octadecatrienoic acid; 13-HPOT, 13(*S*)-hydroperoxylinolenic acid; IPTG, isopropyl-β-D-thiogalactopyranoside; OPDA, 12-oxo-phytodienoic acid.



Fig. 1. Illustration of the method for identifying a target protein in this study.



Fig. 2. Structures of linker 1, AOS inhibitor 2, and azide probe 3.

Linker 1 that was used in this study is shown in Fig. 2. Biotin is commonly used as a tag for protein detection via Western blotting. Benzophenone was introduced into the linker. Polyether is commonly used as a hydrophilic spacer, and an alkyne is necessary for the Huisgen cycloaddition. To examine whether this strategy is effective for identifying proteins targeted with small molecules, an allene oxide synthase (AOS) was selected as the target protein, and AOS inhibitor 2 and azide probe 3 were chosen as the small molecules for use in this study (Fig. 2).

AOS, which is an enzyme involved in the biosynthetic pathway of jasmonic acid (Fig. 3), a plant hormone, converts 13(S)-hydroperoxylinolenic acid



Fig. 3. Biosynthesis of jasmonic acid.

(13-HPOT) into (12,13*S*)-epoxy-(9*Z*,11*E*,15*Z*)-octadecatrienoic acid (12,13-EOT),¹¹⁾ which is the first specific reaction in the biosynthesis of jasmonic acid (Fig. 3). AOS is classified as cytochrome P450 CYP74A and contains a heme iron in its active center. The inhibitory mechanism of cytochrome P450s has been studied.¹²⁾ Imidazole derivatives inhibit cytochrome P450s, apparently due to the affinity of the electron pair on the nitrogen in heterocyclic molecules for the heme iron in the active center.^{13,14)} Oh and Murofushi synthesized imidazole derivatives with alkylcarboxylate moieties as AOS inhibitors and introduced the alkylcarboxylate moieties into the AOS inhibitors to mimic the partial structure of 13-HPOT.¹⁵⁾

This study shows that the method described above enables the identification of target proteins using a model system of PpAOS1 and an AOS inhibitor introducing only an azide group (azide probe).

Experimental

General. ¹H NMR spectra were recorded on a JEOL EX-270 FT NMR (270 MHz) spectrometer (JEOL Ltd., Tokyo, Japan), and CDCl₃ (7.24 ppm) or CD₃OD (3.30 ppm) was used as an internal standard. Mass spectra were collected on a JEOL JMS-T100GCV mass spectrometer (JEOL Ltd., Tokyo, Japan). Specific rotation values were measured on a JASCO DIP-310 polarimeter (JASCO Corporation, Tokyo, Japan). Chemical reagents were purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan), unless otherwise noted.

tert-Butyl {2-[2-(2-hydroxyethoxy)ethoxy]ethyl}carbamate (5). 2-[2-(2-Chloroethoxy)ethoxy]ethanol (4, 6.3 g, 37 mmol), NaI (1.8 g, 12 mmol), and NaN₃ (3.9

g, 60 mmol) were dissolved in 30 mL of deionized water and then stirred at 60 °C for 16 h. The reaction mixture was filtered to remove insoluble matter. The filtered solution was extracted with 60 mL of EtOAc. The organic layer was dried over Na₂SO₄ and evaporated to yield the azide (6.5 g). The azide (6.5 g) was dissolved in MeOH containing 0.5 g of 5% Pd-C and was stirred under a hydrogen atmosphere at room temperature for 12 h. The reaction mixture was filtered using Celite and was evaporated to yield the corresponding amine (6.2 g). The amine was dissolved in 150 mL of CH₃CN. This was followed by the addition of 30 mL of deionized water, 30 mL of 1 M NaOH, and (Boc)₂O (8.1 g, 37 mmol) at room temperature for 16 h. The reaction mixture was evaporated and the resultant residue was added to 100 mL of water and then extracted with an equal volume of EtOAc. The organic layer was dried over Na₂SO₄. After evaporation of the organic layer, the residue that was obtained was purified by silica gel column chromatography (200 g, MeOH-CHCl₃ 1/50) to yield 5 (6.2 g, 25 mmol) as a colorless oil. The yield from the three steps was 68%. ¹H NMR (270 MHz, CDCl₃): 5.21 (1H, br. s), 3.68 (2H, t, J = 7.1 Hz), 3.61-3.52 (6H, m), 3.50 (2H, t, t)J = 7.1 Hz), 3.24 (2H, t, J = 6.9 Hz), 2.88 (1H, br. s), 1.37 (9H, s); ¹³C NMR (67.8 MHz, CDCl₃): 155.9, 79.1, 72.5, 72.4, 70.2, 70.1, 61.5, 40.3, 28.2; FD-MS m/z (rel.int.): 250 (100, $[M + H]^+$), 249 (13.0, $[M]^+$), 57.0 (25.1), 31.0 (16.1); FD-HR-MS m/z: calcd. for C₁₁H₂₄NO₅, 250.16545, found 250.16572.

tert-Butyl {2-[2-(2-bromoethoxy)ethoxy]ethyl}carba-Compound 5 (713 mg, 2.85 mmol), CBr₄ mate (6). (1.23 g, 3.70 mmol), and K₂CO₃ (510 mg, 3.70 mmol) were dissolved in 5 mL of CH₂Cl₂. Ten mL of CH₂Cl₂ containing PPh₃ (1.2 g, 4.56 mmol) was added by syringe to the solution and stirred at room temperature for 24 h. After evaporation of the reaction mixture, the resultant residue was added to n-hexane, and the insoluble material was filtered. The resultant material, which was obtained by evaporation of the filtrate, was purified by silica gel column chromatography (40 g, n-hexane-EtOAc, 1/1) to yield 6 (745 mg, 2.40 mmol, 84%) as a colorless oil. ¹H NMR (270 MHz, CDCl₃): 4.98 (1H, br.s), 3.76 (2H, t, J = 7.0 Hz), 3.63–3.55 (4H, m), 3.50 (2H, t, J = 7.1 Hz), 3.43 (2H, t, J = 7.3 Hz), 3.25 (2H, t)t, J = 7.1 Hz), 1.39 (9H, s); ¹³C NMR (67.8 MHz, CDCl₃): 155.8, 79.1, 71.0, 70.3, 70.1, 70.0, 40.2, 30.1, 28.3; FI-MS m/z (rel.int.): 314 (52.5), 313 (100), 312 $(58.3, [M + H]^{+}), 311 (94.8, [M]^{+}), 57.0 (84.6); FI-$ HR-MS m/z: calcd. for C₁₁H₂₂BrNO₄, 311.07322, found 311.07016.

tert-Butyl {2-[2-(2-prop-2-ylethoxy)ethoxy]ethyl}carbamate (7). 10 mL of dry THF was added to a flask containing 2-propyn-1-ol (560 mg, 10 mmol) and NaH (480 mg, 20 mmol) and was stirred on ice for 20 min. After the addition of **6** (488 mg, 1.6 mmol) to the solution, it was stirred for an additional 2 h. An aqueous solution of NH_4Cl was added to the reaction mixture to terminate the reaction, and the solution was then extracted with 60 mL of EtOAc. The organic layer was dried over Na₂SO₄. The resultant matter was purified by silica gel column chromatography (40 g, *n*-hexane-EtOAc 3/2) to yield 7 (90 mg, 0.31 mmol, 20%) as a yellow oil. ¹H NMR (270 MHz, CDCl₃): 5.00 (1H, br), 4.18 (2H, d, J = 3.3 Hz), 3.70–3.66 (4H, m), 3.64–3.56 (4H, m), 3.50 (2H, t, J = 7.3 Hz), 3.28 (2H, t, J = 7.2 Hz), 2.40 (1H, t, J = 3.3 Hz), 1.41 (s, 9H); ¹³C NMR (67.8 MHz, CDCl₃): 155.9, 79.5, 79.1, 74.5, 70.4, 70.3, 70.19, 70.17, 69.0, 58.3, 40.3, 28.3; FD-MS m/z (rel.int.): 288 (100, [M + H]⁺), 287 (19.0, [M]⁺), 69.0 (95.7), 57.0 (91.3); FD-HR-MS m/z: calcd. for C₁₄H₂₆NO₅, 288.18110, found 288.18038.

2-[2-(2-Prop-2-ynyloxyethoxy)ethoxy]ethylamine (8). Compound 7 (90 mg, 0.31 mmol) was dissolved in 50% TFA-CH₂Cl₂ (2 ml) and stirred at 0 °C for 2 h. The reaction mixture was evaporated to yield 8 (58 mg, 0.31 mmol, 99%) as a yellow oil. ¹H NMR (270 MHz, CD₃OD): 4.80 (2H, s), 4.20 (2H, d, J = 3.1 Hz), 3.68– 3.56 (8H, m), 3.30 (2H, t, J = 7.2 Hz), 2.95 (2H, t, J = 7.4 Hz), 2.85 (1H, t, J = 3.1 Hz); ¹³C NMR (67.8 MHz, CD₃OD): 80.5, 76.0, 71.4, 71.25, 71.23, 70.5, 70.0, 59.0, 41.3; FD-MS *m*/*z* (rel.int.): 188 (100, [M + H]⁺); FD-HR-MS *m*/*z*: calcd. for C₉H₁₈NO₃, 188.12867, found 188.12836.

(+)-5-(2-Oxohexahydrothieno[3,4-d]imidazol-4-yl)pentanoic acid (2-(4-benzoylphenyl)-1-{2-[2-(2-prop-2ynyloxyethoxy)ethoxy]ethylcarbamoyl}ethyl)amide (1). (+)-Biotin (9, 24.4 mg, 0.10 mmol), HOBT (13.5 mg, 0.10 mmol), and EDC (19.1 mg, 0.10 mmol) were dissolved in DMF (5 mL) and then stirred at room temperature for 1 h. 2-Amino-3-(4-benzoylphenyl)propanoic acid (10, 20.0 mg, 0.090 mmol) was added to the reaction mixture, and the mixture was stirred for 16 h. Then, EDC (19.1 mg, 0.10 mmol) and 8 (15.0 mg, 0.08 mmol) were added to the reaction mixture, which was then stirred for an additional 16 h. The reaction solution was added to EtOAc (20 ml) and was then washed with 30 mL of 1 M HCl, followed by deionized water. The organic layer was dried over Na₂SO₄ and was concentrated in vacuo. The resultant residue was purified by silica gel column chromatography (20 g, CHCl3-MeOH 9/ 1) to give 1 (11 mg, 0.017 mmol, 21%) as a yellow oil. ¹H NMR (270 MHz, CDCl₃): 7.83 (2H, d, J = 8.0 Hz), 7.72 (2H, m), 7.60 (1H, m), 7.44 (2H, m), 7.21 (2H, m), 4.35 (1H, m), 4.15 (2H, m), 3.99 (2H, m), 3.63-3.46 (10H, m), 3.42 (2H, t, J = 7.3 Hz), 3.16 (1H, m), 2.87 (2H, m), 2.68 (2H, m), 2.14 (1H, m), 2.05 (2H, m), 1.72 (2H, m), 1.50 (2H, m), 1.16 (2H, t, J = 7.2 Hz); ¹³C NMR (67.8 MHz, CDCl₃): 196.4, 179.9, 171.4, 142.2, 138.3, 133.8, 132.6, 130.8, 130.3, 130.1, 129.9, 129.6, 128.6, 128.3, 127.9, 81.8, 77.2, 76.2, 72.7, 72.1, 71.4, 70.7, 70.3, 64.8, 57.9, 57.5, 56.8, 55.2, 51.5, 44.6, 42.7, 33.6, 28.3, 27.9, 24.7; FD-MS m/z (rel.int.): 665 (100, $[M+H]^+$), 664 (52.1, $[M]^+$); FD-HR-MS *m/z*: calcd. for C35H45N4O7S, 665.30089, 665.29875; found $[\alpha]_D^{25} + 21.0^\circ (c \ 0.40, \text{MeOH})$

Synthesis of 2 and 3. The synthesis of 2 and 3 is described in the Supplementary data.

AOS inhibitory assay. Recombinant PpAOS1 was obtained according to the method of Bandara (2009).¹⁷⁾ An AOS inhibitory assay was performed according to the method of Oh and Murofushi with minor modifications.15) The reaction mixture containing PpAOS1 (0.5 µM), 13-HPOT (10 µM), and each concentration of AOS inhibitor in 100 µL of 50 mM phosphate buffer at pH 7.0 was incubated at 25 °C for 30 min. A portion of the reaction solution (10 µL) was loaded onto an HPLC (column: Capcelpak C18 UG120, 4.6 mm i.d. × 150 mm, Shiseido, Japan; solvent: 50% CH₃CN aq. with 0.1% phosphoric acid; flow: 1.0 ml/ min; detection: UV, 235 nm). AOS inhibitory activity was calculated from the decrease in the peak area of 13-HPOT as a substrate.

Detection of PpAOS1 by Western blotting. Azide probe **3** (30 μ M) and recombinant PpAOS1 (10 μ M) were pre-incubated in 60 μ L of 50 mM sodium phosphate buffer at pH 7.8 for 1 h on ice prior to bioorthogonal coupling. Azide alkyne Huisgen cycloaddition was initiated by the addition of linker **1** (30 μ M) and CuI (0.5 μ M) and was incubated for 1 h. After the cycloaddition, the mixture was irradiated with UV (365 nm) on ice for an additional 1 h. For control experiments, PpAOC2 (10 μ M)¹⁸⁾ was added to the reaction mixture instead of PpAOS1.

The sample solution was centrifuged to remove insoluble CuI from the reaction mixture and was then analyzed by SDS-PAGE. The gel was equilibrated with a blotting buffer (25 mM Tris-HCl, pH 8.3, 192 mM glycine, 10% MeOH) for 5 min. A PVDF membrane (Millipore, USA), which was treated with MeOH and water, was also equilibrated with the blotting buffer. The protein was transferred from the gel onto the PVDF membrane for 1 h using a Trans-blot semi dry SD cell (Bio-Rad, USA). The electric current was set at 0.8 mA/cm². For blocking, the membrane was incubated in 5% non-fat dry milk in PBS-T buffer (80 mM Na₂HPO₄, 20 mM NaH₂PO₄, 100 mM NaCl and 0.1% Tween 20) overnight at 4 °C, and the membrane was then washed twice using PBS-T buffer. The membrane after blocking was incubated in PBS-T buffer containing streptavidin horseradish peroxidase conjugate (50 ng/mL, Pierce, USA). After washing the membrane, it was soaked in Amersham ECL Western blotting detection reagents (GE Healthcare, USA) at room temperature for 5 min. The chemiluminescence derived from the ternary complex was detected using a Lumi-VisionPro 400EX fluorescence imager (AISIN, Japan).

To examine the inefficiency of the bioorthogonal coupling of linker 1 and azide probe 3 before incubation of 1 with PpAOS1 for PpAOS1 detection, linker 1 (1 mM), azide probe 3 (1 mM) and CuI (5 μ M) were mixed in 5 mL of MeOH, and the solution was incubated at room temperature for 1 h. Two microliters of the reaction mixture containing the Huisgen cycloaddition and 56 μ L of PpAOS1 (10 μ M) were incubated in 50 mM sodium phosphate buffer, pH 7.8, on ice for 1 h. Then, UV irradiation of the reaction mixture was performed as described above.

For competition assays, AOS inhibitor 2 (300 μ M) was added to 60 μ L of 50 mM sodium phosphate buffer, pH 7.8, containing azide probe 3 (30 μ M) and the recombinant PpAOS1 (10 μ M). Other procedures were performed as described above.

Evaluation of the dose-dependent PpAOS1 signal increase by PpAOS1 and 3. To evaluate the concentration dependence of PpAOS1 on chemiluminescence, the PpAOS1 concentration was varied from 1.25 to 10 μ M. Similarly, variable amounts of azide probe 3 from 3 to 3000 nM were added to the reaction mixture. The other procedure for PpAOS1 detection was performed as described above.

Detection of PpAOS1 in lysates of E. coli overexpressing PpAOS1. A single colony of E. coli transformed with pET23a containing PpAOS1 was incubated in 5 mL of LB medium containing 100 µg/ mL of ampicillin at 37 °C. After the culture had grown to a cell density of $OD_{600} = 0.6$, PpAOS1 synthesis was induced by the addition of IPTG at 0.2 mM. After further incubation at 37 °C for 3 h, the cells were collected by centrifugation at 5000 g for 5 min and then resuspended with 50 mM sodium phosphate buffer at pH 7.8 and disrupted by ultrasonication. The cell debris was removed by centrifugation at 15,000 g for 30 min. 56 µL of the resultant supernatant (protein concentration; 500 µg/mL) was used for Western blotting. For control experiment, the lysate of the E. coli strain was prepared as described above except for IPTG treatment.

Results and discussion

Synthesis of 1, 2, and 3

To synthesize linker 1, alcohol 5 was prepared from 2-[2-(2-chloroethoxy)ethoxy]ethanol 4 via three reactions: introduction of the azide group, reduction of the azide to an amine, and protection of the amino group using a *tert*-butoxycarbonyl (Boc) group. Alcohol 5 was then converted to bromide 6. The reaction of bromide 6 with 2-propyn-1-ol produced alkyne 7, and deprotection of the Boc group formed polyether amine 8 (Scheme 1). The conjugation of biotin 9 and 2-amino-2-[4-(phenylcarbonyl)phenyl]acetic acid 10¹⁶) followed by amidation with biotin polyether amine 8 yielded the desired linker containing biotin, alkyne, and benzophenone group 1 (Scheme 2).

The AOS inhibitor **2** and azide probe **3**, which was the azide group introduced into the structure of AOS inhibitor **2**, were synthesized as model compounds according to the method of Oh and Murofushi with some modifications (Supplementary data).¹⁵⁾ The AOS inhibitory activities of **2** and **3** were examined. The AOS reaction was conducted by incubating purified recombinant PpAOS1¹⁷⁾ from the model moss *Physcomitrella patens* with 13-HPOT as a substrate and various concentrations of **2** or **3** at 25 °C for 30 min. The activities of **2** and **3** were estimated by measuring the decreased peak areas of 13-HPOT by HPLC analy-



Scheme 1. (a) (1) NaN₃, NaI, H₂O, 60 °C, 16 h. (2) Pd/C, H₂, CH₃OH, r.t., 12 h. (3) (Boc)₂O, NaOH, CH₃CN, r.t., 12 h, 68% (3 steps); (b) CBr₄, K₂CO₃, PPh₃, CH₂Cl₂, r.t., 12 h, 84%; (c) 2-propyn-1-ol, NaH, THF, 0 °C, 2 h, 20%; (d) 50% TEA-CH₂Cl₂, 0 °C, 2 h, 99%.



Scheme 2. (a) (1) EDC, HOBT, DMF, r.t., 18 h; (2) 8, EDC, DMF, r.t., 16 h, 21% (2 steps).

sis,¹⁵⁾ showing that the AOS inhibitory activity of an azide probe **3** was higher than that of AOS inhibitor **2** (Fig. S1). The azide group, which was attached to the terminal alkyl group in **3**, increased AOS inhibitory activity. The data indicated that azide probe **3** also interacts with AOS and provides access to the AOS for chemical labeling. Therefore, azide probe **3** was used in the next step.

Detection of a recombinant PpAOS1 as a model system

Detection of PpAOS1 was attempted using azide probe **3** and linker **1**. Azide probe **3** was incubated with PpAOS1 in 50 mM phosphate buffer at pH 7.8. Linker **1** was conjugated with **3** via the click reaction using Cu (I) as a catalyst (Fig. 4). The PpAOS1 and the conjugate **11** (Fig. 4) were cross-linked by UV irradiation to a benzophenone group. The ternary complex



Fig. 4. Structure of conjugate of linker 1 and azide probe 3 (11).



Fig. 5. Identification of the PpAOS1–azide probe–linker complex by Western blotting.

Notes: Azide probe **3** was incubated with PpAOS1 and was then bound to linker **1** via a Huisgen cycloaddition. The conjugate **11** was covalently bound to PpAOS1 by UV irradiation (365 nm). The ternary complex was subjected to SDS–PAGE and was detected by Western blotting using streptavidin conjugated with horseradish peroxidase. For the competitive inhibition assay, PpAOS1 was incubated with azide probe **3** and a 10-fold excess of AOS inhibitor **2** (competitor) prior to Huisgen cycloaddition with linker **1**. To examine the specificity of this method, PpAOS1 was replaced by PpAOC2 and chemiluminescent detection was attempted as described above.

containing PpAOS1, 1 and 3 was analyzed by SDS– PAGE. To detect the biotinylated components, the ternary complex was electrophoretically transferred onto a membrane surface and the blot was visualized via chemiluminescence using streptavidin and horseradish peroxidase. Chemiluminescence arising from the ternary complex was observed (Fig. 5).

Azide probe 3 was a derivative of 2 with the azide group bound to the terminal alkyl group. To determine if 3 binds to the same PpAOS1 site as 2, a 10-fold excess of AOS inhibitor 2 as a competitor was incubated with 3 in a PpAOS1 solution. The results indicated that the addition of 2 caused the PpAOS1 chemiluminescent signal to disappear (Fig. 5). A competition assay was performed with an excess of 2 to confirm that labeling was the result of a specific inter-



Fig. 6. Dose-dependent increase in chemiluminescence with both PpAOS1 and azide probe **3.**

Notes: (A) PpAOS1 (1.25–10 μ M), linker **1** (30 μ M), and azide probe **3** (30 μ M) in 50 mM phosphate buffer, pH 7.8. (B) Azide probe **3** (3–3000 nM), PpAOS1 (10 μ M), and linker **1** (30 μ M) in 50 mM phosphate buffer, pH 7.8.



Fig. 7. Examination of the effectiveness of azide probe **3** for detecting the chemiluminescence of the PpAOS1–azide probe–linker complex.

Notes: Lane 1: PpAOS1 incubated with azide probe **3** prior to Huisgen cycloaddition of linker **1**; lane 2: PpAOS1 incubated with the conjugate **11**.

action. Accordingly, AOS inhibitor **2** and azide probe **3** bound to the same binding site of PpAOS1, as expected.

To examine whether azide probe **3** specifically binds PpAOS1, PpAOS1 was replaced with PpAOC2, which is another JA biosynthetic enzyme.¹⁸⁾ As a result, no signal corresponding to PpAOC2 was detected, confirming the specific binding of **3** with PpAOS1. Therefore, the method could specifically detect a target protein bound to an azide probe.

Dose dependence of azide probe 3 and PpAOS1 for PpAOS1 chemiluminescent signal intensity

To investigate whether the intensity of the chemiluminescent signal varied with the PpAOS1 concentration, the concentration of PpAOS1 was varied from 1.25 to 10 μ M. The results indicated that the intensity of the chemiluminescent signal increased dose dependently from 1.25 to 10 μ M PpAOS1 (Fig. 6(A)). The concentration dependence of the chemiluminescent signal intensity on azide probe **3** was also examined. The concentration of **3** ranged from 3 to 3000 nM, and a dose-dependent increase in the chemiluminescent signal of azide probe **3** was also observed (Fig. 6(B)). These data confirm that the developed method for identifying target proteins bound to small molecules yielded dosedependent increases in the chemiluminescent signal intensity with the target protein and the azide probe.



Fig. 8. Detection of PpAOS1 in lysates of *E. coli* overexpressing *PpAOS1*.

Notes: Lysates of a strain of *E. coli* with an expression vector containing *PpAOS1* with or without IPTG induction were used for the detection of PpAOS1. After each lysate was incubated with **3**, bioorthogonal coupling of **1** with **3** was performed; then, the conjugate **11** with PpAOS1 was irradiated by UV (365 nm). The lysates were analyzed by SDS–PAGE and then visualized by Western blotting and CBB staining.

Efficacy of the developed method using linker 1 and azide probe 3

In the previously described experiments to detect PpAOS1 as a target protein, a Huisgen cycloaddition was performed after incubating 3 with PpAOS1. Azide probe 3 detected PpAOS1 without requiring a photophore in the structure of 3. To test the efficacy of the azide probe for detecting a target protein, the conjugate 11 was formed via Huisgen cycloaddition and was then incubated with PpAOS1. As a result, no chemiluminescent signal was detected via Western blotting (Fig. 7). Although the detailed structure of PpAOS1 and 3 as a complex has not been examined, previous studies on cytochrome P450 inhibitors indicated that the benzene and imidazole moieties are located near the active site.^{13,14}) The azide group in 3 appear to reside on the surface of PpAOS1. Linker 1 could provide access to the azide group of 3 on the surface of PpAOS1, and the resultant conjugate 11 still provides a stable interaction with PpAOS1. On the other hand, the bulky moiety in conjugate 11, which is preliminarily formed before incubating **3** with PpAOS1, should inhibit its approach to the binding site of PpAOS1. Therefore, these data showed the efficacy of the method for identifying the binding protein using an azide probe that was minimally modified by an azide group.

Detection of PpAOS1 in a crude protein extract

As described above, the developed method to identify a target protein with small bioactive compounds was shown to work in a model system using recombinant PpAOS1 and azide probe 3. However, it is possible that the method in this study detects non-specific protein signals in a crude protein solution, which can obscure a signal derived from a target protein. To examine whether the method in this study can effectively detect the target protein of a small bioactive compound in a protein mixture, linker 1 and azide probe 3 were incubated in a lysate of E. coli transformed with an overexpressing vector containing *PpAOS1*; then, detection of PpAOS1 was attempted by Western blotting (Fig. 8). A chemiluminescent signal derived from PpAOS1 in a lysate of the E. coli strain, whose protein expression was induced by isopropyl-β-D-thiogalactopyranoside (IPTG), was clearly observed. Conversely, a PpAOS1derived signal was not detected in a lysate of the E. coli strain without IPTG treatment. These data show that the method developed in this study can be applied to the detection of a target protein of a bioactive small molecule in a crude protein extract.

Conclusions

This study demonstrated that a minimally modified azide probe, produced by the introduction of an azide group, can effectively detect target proteins. Linker **1** effectively binds the azide probe to the protein, allowing for chemiluminescent detection. These data indicate that the method developed here demonstrates the superiority of small functional groups for detecting target proteins and that the detection was specific for the target protein. The method in this study was developed using a model system and may be applicable for the detection of target proteins in a mixture. This strategy facilitates the detection of proteins bound to small bioactive molecules.

Authors contributions

Study concept and design: Kosaku Takahashi. Acquisition of data: Tomoaki Anabuki, Miu Tsukahara, and Kosaku Takahashi. Analysis and interpretation of data: Tomoaki Anabuki, Miu Tsukahara, Hideyuki Matsuura, Kosaku Takahashi. Drafting of the manuscript: Hideyuki Matsuura and Kosaku Takahashi. All authors reviewed and approved the final manuscript.

Disclosure statement

No potential conflict of interest was reported by the authors.

Supplemental material

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