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Highly Enantioselective Iridium-Catalyzed Coupling Reaction of Vinyl Azides and Racemic Allylic Carbonates

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ABSTRACT: The Iridium-catalyzed enantioselective coupling reaction of vinyl azides and allylic electrophiles is presented and provides access to β -chiral carbonyl derivatives. Vinyl azides are used as acetamide enolate or acetonitrile carbanion surrogates, leading to γ , δ -unsaturated β -substituted amides as well as nitriles with excellent enantiomeric excess. These products are readily transformed into chiral *N*-containing building blocks and pharmaceuticals. A mechanism was proposed to rationalize the chemoselectivity of this coupling reaction.

INTRODUCTION

Transition metal-catalyzed asymmetric allylic alkylation (AAA) reaction represents a powerful and appealing tool for construction of nonracemic compounds.¹ The Ir-catalyzed asymmetric allylation of carbonyl compounds is one of the most important methods for generating a stereocenter, particularly, in the β position of carbonyl derivatives. (Scheme 1a) Pioneered by Takeuchi^{2a} and Helmchen,^{2b} a variety of "soft" or stabilized carbanions (e.g. malonic esters, malononitrile, sulfonvlacetic esters and disulfones) have been used in this reaction.³ However, simple carbonyl compounds with weak α C-H acidity are challenging nucleophiles. The difficulty is attributed to the formed "hard" unstabilized enolates, which are fairly basic and not compatible with the allylic electrophiles or adducts. A desirable solution is to seek appropriate enolates surrogates (Scheme 1b).⁴ As the seminal work, Hartwig demonstrated the highly efficient allylation reaction of ketones^{4a and 4e}, α , β -unsaturated ketones,^{4b-d} and aliphatic esters^{4f} by using their silyl enol ethers. Meanwhile, the elegant Ir-catalyzed AAA of β -ketocarboxylates.⁵ ketene acetal^{6a} and ketene aminal^{6b} were reported, respectively, by You and Carreira, providing access to a variety of enantioenriched carbonyl compounds.

Amides and nitriles are extremely useful not only in the organic chemistry but also in the bio-chemistry and material science.⁷ Nevertheless, few allylation reactions appeared regarding the use of amide enolates or α -carbanion of nitriles as nucleophiles,⁸ probably because of side reactions caused by the extremely "hard" property. For example, in the studies on Pd-catalyzed AAA, Hou and coworkers found that the α -carbanion of acetonitrile (pKa \approx 31 in DMSO)⁹ couldn't react with allylic electrophiles but attacked another acetonitrile to form enamine (Thorpe reaction).¹⁰ Thus, the highly general and enantioselective alkylation of acetonitrile is still elusive.

Vinyl azides, molecules that contain conjugate alkene and azide moieties, have been known for around a century.

Owing to the recent development of more convenient synthetic approaches,^{11c} the vinyl azide chemistry has received increasing interests of synthetic chemists.¹¹ Since the seminal work by Chiba,12b Lewis/Brønsted acid-promoted coupling reactions of vinyl azides and C-electrophiles (i.e. aldehydes, imines, propargylic alcohols, *p*-quinone methides) have been developed and become a straightforward approach towards amides and nitriles.¹² We envisioned that the cooperative Lewis acid and chiral transition metal catalysis might facilitate the asymmetric coupling of vinyl azides and C-electrophiles. Herein, the Lewis acid-promoted and Ir-catalysed AAA¹³ with vinyl azides was demonstrated, wherein vinyl azides were used as the acetamide enolate as well as acetonitrile carbanion surrogates (Scheme 1c). The method represents the first catalytic enantioselective approach towards synthesis of enantioenriched amides and nitriles with the use of vinyl azides.

Scheme 1. Ir-Catalyzed Enantioselective Allylations of Unstabilized Enolates

(a) The state of art: the Ir-catalyzed AAA of carbonyl compounds



(c) This work: vinyl azides as amide enolate and nitrile carbanion surrogates



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RESULTS AND DISCUSSION

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In initial studies, the enantioselective coupling of (1-azidovinyl)benzene (2a) and branched allylic alcohols or carbonates was tested with [Ir(cod)Cl]₂ (2.5 mol%), (R)-L1 (10 mol%) and $BF_3 \cdot OEt_2$ (10 mol%) in dichloromethane. The tert-butyl (1-phenylallyl) carbonate (1a) was found to be the optimal allylic electrophile, producing the desired product **3aa** in 40% yield and 93% ee at 15 °C. (See the supporting information for full optimization details) Through systematic optimization of the Lewis acid, solvent and temperature, a promising result was achieved. (entry 1, Table 1) We then turned our attention to the study of salt effects, in an attempt to accelerate isomerization of intermediary π allyliridium complexes.¹⁴ Indeed, the use of halide salts as additives improved the catalytic efficiency. Inorganic salts (e.g. LiCl, LiBr, ZnBr₂) improved the yield of 3aa, albeit with erosion of ee value (entries 2-5). Pleasingly, Bu₄NCl or Bu₄NBr as the additive significantly improved the dynamic kinetic asymmetric transformation (DYKAT) of 1a to afford 3aa in moderate yields (68% and 65%) with high enantio enrichment (entries 6 and 7). By increasing the loading of BF_3 .Et₂O to 50 mol% and lowering the temperature to 10 °C, 3aa was obtained in good yield (74%) and excellent enantiopurity (entry 9). The dosage of **2a** could be further reduced to 1.2 equiv. without loss of yield or ee (entry 10), indicating that vinyl azides were relatively stable and robust amide enolate surrogates. When changing the chiral ligand or Ir catalyst precursor, the reaction gave worse results. (entries 11-13) Control experiments confirm that no product was observed in the absence of [Ir(cod)Cl]₂, the ligand or Lewis acid, (entries 14 and 15) indicating the necessity of each catalvst.

30 With optimal conditions in hand, various (hetero)aryl 31 substituted allylic alcohol carbonates were tested with 2a 32 to give the corresponding products with good to high yields 33 and nearly all in absolute stereochemical control. (Table 2) 34 The stereo- and electronic nature of the aryl groups on the 35 allylic substrates had little effect on the yields and enanti-36 oselectivities. Aryl units bearing alkyl (Me and ⁱPr), phenyl, 37 CF₃ and halo (Cl and Br) substituents were successfully in-38 corporated into β -aryl γ , δ -unsaturated amides (**3ab-aj**). The 39 (R)-configuration of **3aj** was confirmed by the X-ray diffraction of single crystal. The allylic carbonates with 3,4- or 2,6-40 dichloro substituted aryl groups were also reactive to pro-41 duce **3ak** and **3al** in moderate yields (69% and 63%) with 42 excellent ees (>99% and 97%). Fused ring- (α - or β -naph-43 thyl) and hetero-aryl (3-thienyl or 3-indolyl) substituted al-44 lylic carbonates were readily transformed into enantiopure 45 amides (**3am-ap**). And the asymmetric transformation of 46 alkyl-substituted allylic electrophiles was demonstrated by 47 synthesis of 3aq in 37% yield with 99% ee. However, 2-phe-48 nylbut-3-en-2-ol or 2-methyl-1-phenylprop-2-en-1-ol de-49 rived carbonates had no reactivity, probably due to the ste-50 ric hindrance. In order to get insights into the electronic ef-51 fects of vinyl azides on the allylic substitution as well as 52 Schmidt rearrangement process, the scope of vinyl azides 53 was evaluated. The catalysis was tolerant of alkyl, aryl, 54 alkoxyl, halo, carbonyl, nitro, ester, CF3 and cyano groups (for **3ba-3ma**) installed on the *p*- or *m*-position of α -aryl 55 unit of vinyl azides, but not tolerant of o-substituents on aryl 56

groups. On the other hand, α -alkyl or alkenyl-substituted vinyl azides were also tested, but had no reactivity.

Table 1. Development of the Vinyl Azide-Allyl Coupling Reaction.^{*a*}



entry	changes of reaction conditions	yield	ee
		(%) ^b	(%) ^c
1	none	63	90
2	LiCl (10 mol%) as the additive	74	88
3	LiBr (10 mol%) as the additive	73	84
4	ZnCl2 (10 mol%) as the additive	63	56
5	ZnBr2 (10 mol%) as the additive	77	50
6	Bu4NCl (10 mol%) as the	68	98
	additive		
7	Bu4NBr (10 mol%) as the	65	99
	additive		
8	Bu ₄ NI (10 mol%) as the	34	99
	additive		
9 ^d	Bu4NBr (10 mol%) as the	74	99
	additive		
10 ^{d,e}	Bu4NBr (10 mol%) as the	76	99
	additive		
11 ^{d,e}	(<i>R</i>)- L2 as the ligand	59	99
12 ^{d,e}	(<i>R</i>)- L3 as the ligand	trace	/
13 ^{d,e}	C1 as the Ir catalyst	trace	/
14 ^{d,e}	without BF3.Et20	0	/
15 ^{d,e}	without Ir or (<i>R</i>)- L1	0	/

^{*a*} The reaction was performed with **1a** (0.2 mmol), **2a** (0.3 mmol), [Ir(COD)Cl]₂ (2.5 mol%), (*R*)-**L1** (10 mol%), BF₃.Et₂O (30 mol%) in toluene (2.0 mL). ^{*b*} The isolated yield of **3aa**. ^{*c*} Determined by chiral HPLC. ^{*d*} The reaction was performed at 10 °C with 50 mol% of BF₃.Et₂O. ^{*e*} 1.2 equiv. of **2a** and 4.0 mL of toluene were used.



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Table 2. Synthesis of Enantioenriched β -Substituted γ δ -Unsaturated Amides^a



^a The reaction was performed with 1 (0.20 mmol), 2 (0.24 mmol), [Ir(COD)Cl]₂ (2.5 mol%), (R)-L1 (10 mol%), BF₃.Et₂O (50 mol%) and Bu₄NBr (10 mol%) in toluene (4.0 mL). ^b Isolated yield. ^c Determined by chiral HPLC. ^d The absolute configuration of **3aj** was confirmed by the X-ray diffraction (CCDC 1950457). ^e The reaction was performed at 35 oC.^f The dosage of vinyl azide 2 was 0.30 mmol.

Next, the method was applied to synthesize chiral nitriles with α -hydroxypropan-2-yl substituted vinyl azide (4)¹⁵ as acetonitrile carbanion surrogate. Initially, the coupling of 1a and 4 failed with the standard conditions of Table 2. None of product was formed, while both substrates were decomposed. (entry 1, Table 3) The effect of Lewis acid was then evaluated and zinc salts were found to promote the reaction. (entries 3, 6 and 9) The strong Lewis acids would decompose the vinyl azide to erode the yield (entries 2 and 4), while the weak Lewis acids couldn't facilitate the catalytic process (entries 5 and 8). Finally, the reaction, promoted both by $ZnBr_2$ (15 mol%) and Ir/(R)-L1 (5 mol%), gave the enantiopure (*R*)-3-phenylpent-4-enenitrile (5a) in

excellent yield (entry 10), the absolute configuration determined by the comparison of its specific rotation with that in the literature.¹⁶ Under the optimized reaction conditions (ZnCl₂ or ZnBr₂ as the Lewis acid), an array of racemic arylsubstituted allylic carbonates were efficiently transformed into enantiopure β -(hetero)aryl pent-4-enenitriles. As depicted in Table 4, a wide range of functional groups, including alkyl, alkynyl, aryl, halo, carbonyl, carbonate, on the aryl units of allylic carbonates were tolerated and provided the enantioenriched homoallylic nitriles (5b-5p) in moderate to high yields with excellent ees. Meanwhile, fused ring- (α or β -naphthyl) and hetero-aryl (3-thienyl or 3-indolyl) substituted allylic carbonates were readily transformed into

nearly enantiopure nitriles. (**5q-5t**) Although the alkoxyl(MeO or BnO)-aryl derived allylic carbonate was unstable and unable to provide the desired product, the corresponding allylic alcohol was appropriate to participate in the catalytic asymmetric coupling (the transformation of **13** into **14** in the Scheme 2). To the best of our knowledge, this Table 2. The Enontice clearly for a size of a the section of the transformation of the t

is the first catalytic method for asymmetric synthesis of β substituted pent-4-enenitriles, which are difficult to prepare through other methods, for example unavailable by enantioselective allylation of acetonitrile or asymmetric conjugated addition of α , β -unsaturated nitriles.¹⁷

Table 3. The Enantioselective Coupling Reaction of *α*-Hydroxypropan-2-yl Substituted Vinyl Azide (4)^{*a*}

	OBc Ph ^{,,,,,} (+/-) 1a	+ N ₃ + 4 (1.2 eq.)	[Ir(COD)CI] ₂ /(<i>R</i> Lewis acid toluene, argo 25 °C, 12 h)- L1 n Ph ^{vivi} (<i>R</i>)-	CN [α] _D = -9.8, 999 ^{lit} [α] _D = -13.7, 9	% ee (<i>R</i>) 55% ee (<i>R</i>)	
Entry	Lewis acid	yield (%) b	ee (%) ^c	Entry	Lewis acid	yield (%) ^b	ee (%) ^c
1	BF ₃ .Et ₂ O	trace	nd	6	$ZnCl_2$	86	99
2	Sc(OTf)₃	trace	nd	7	$InCl_3$	69	99
3	Zn(OTf)2	45	99	8	MgCl ₂	n.r.	
4	In(OTf)₃	38	99	9	ZnBr ₂	93	98
5	Mg(OTf) ₂	8		10^{d}	$ZnBr_2$	95	99

^{*a*} The reaction was performed with **1** (0.20 mmol), **4** (0.24 mmol), [Ir(COD)Cl]₂ (2.5 mol%), (*R*)-**L1** (10 mol%) and the Lewis acid (10 mol%) in toluene (2 mL). ^{*b*} The isolated yield. ^{*c*} Determined by chiral HPLC. ^{*d*} 15 mol% of ZnBr₂ was used.

Table 4. Synthesis of Enantioenriched β-Substituted γ,δ-Unsaturated Nitriles ^a



^{*a*} The reaction was performed with **1** (0.20 mmol), **4** (0.24 mmol), [Ir(COD)Cl]₂ (2.5 mol%), (*R*)-**L1** (10 mol%) and the Lewis acid (15 mol%) in toluene (2 mL). ^{*b*} The isolated yield. ^{*c*} Determined by chiral HPLC. ^{*d*} ZnCl₂ as the Lewis acid. ^{*e*} ZnBr₂ as the Lewis acid.

To demonstrate the utility of this catalytic approach, synthetic transformations of products were carried out. (Scheme 2) The reduction of enantiopure **3aa** by LiAlH₄ enabled access to *N*-phenyl pent-4-enylamine (**6**) (68% yield, 98% ee), which underwent an intramolecular alkene-carboamination to obtain 2,3-bissubstituted pyrrolidine (**7**) in moderate yield with excellent stereoselectivity. The sequential *N*-allylation and ring-closing metathesis was successful to transform **3aa** into a seven membered lactam **8** in 57% yield and 99% ee. In addition, a Pd-catalyzed intramolecular cyclization of **3aj** via the C-N bond formation was demonstrated to obtain 3,4-dihydroquinolinone (**9**) in 95% yield with no detriment to the enantiomeric excess. Enantioenriched homoallylic nitrile products were readily transformed into important N-containing building blocks. For example, the enantiopure **5a** underwent the hydrolysis, in the presence of hydrogen peroxide and base,⁸ to afford free amide **10** in high yield and conservation of enantiomeric excess. The selective reduction of **5a** by LiAlH₄ gave **11** without loss of ee value. Meanwhile, the Ni-mediated 1

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Scheme 2. Synthetic Transformations of Products



Moreover, the asymmetric method was successfully applied to prepare a potent GluN2B inhibitor (BMS-986169), which is used to treat major depressive disorder.¹⁸ The previous synthetic method for this chiral molecule relies on chiral HPLC resolution of the racemic piperidine intermediate.¹⁹ Herein, the first asymmetric and formal synthetic protocol was realized. The synthetic route began with the (R)-L1/Ir-catalyzed allylic coupling with 1-(4-(benzyloxy)phenyl)prop-2-en-1-ol (13) as the allylic electrophile, due to the incompetence of corresponding allylic carbonate. Fortunately, the catalytic reaction proceeded smoothly with InCl₃ as the Lewis acid and afforded the enantiopure homoallylic nitrile (14) in 76% yield. And the N-Boc homoallylic amine (15) was obtained through a LiAlH₄ reduction and *N*-Boc protection sequence in 80% yield. Followed by the asymmetric Sharpless dihydroxylation²⁰ and then the selective sulfonylation of primary hydroxyl group,²¹ both **16a** and **16a'** were given with moderate ratio (4/1). The mixture could be separated by column chromatography to afford the

desired isomer (**16a**) in 64% yield. Finally, a sequential three-step transformation of **16a** successfully gave the enantiopure **17** in high yield.²² The absolute configuration of (*S*,*S*)-**17** was confirmed by the X-ray diffraction (CCDC 2012123) and found consistent with that of the target.¹⁹

Mechanistic Considerations. The mechanism was proposed to explain chemo- and stereoselectivity of the catalytic coupling reaction. (Scheme 3) In the presence of Lewis acid (i.e. BF₃·OEt₂ or Zn(II) salt), the chiral Ir(I) complex activates allylic carbonate (1a) through oxidative addition to form an Ir(III) intermediate (I) and [LA-O^tBu]⁻. An enantioselective C-C bond coupling of I and vinyl azide follows to generate enantioenriched iminodiazonium ion (II). When bearing a 2-hydroxypropan-2-yl group at the α position (i.e., from 4), the intermediate II undergoes a fragmentation process to give **5a**. When R is a phenyl group, the 1,2-phenyl migration (Schmidt rearrangement) leads to the highly reactive nitrilium ion (III), the latter immediately trapped by [LA-O^tBu]⁻ to form *tert*-butyl imidate (IV)²³ or directly hydrolyzed to the amide product. High resolution mass spectrum (HRMS) study on the reaction system indicated a major species matching the imidate intermediate IV with a proton. (For the detail, see SI) And the control experiment demonstrated that trace of water would lead to the ketone product (3a') formation (Chart 1), probably via direct hydrolysis of the iminodiazonium intermediate II. These results indicate that, in the coupling reaction of α -aromaticsubstituted vinyl azides and allyl tert-butyl carbonates, the desired amide products are obtained through the Schmidt rearrangement/alcoholysis pathway.

Chart 1. Water Effect on the Coupling Reaction







CONCLUSIONS

In summary, vinyl azides were explored in the transition metal-catalyzed coupling reaction, for the first time, as the equivalent of acetamide enolate and acetonitrile carbanion. The amide/nitrile moiety and the chiral center were generated simultaneously in a single step. The utility of products was demonstrated through transformations into several pharmaceutical cores, including chiral pyrrolidine and piperidine (the key intermediate of **BMS-986169**). Preliminary evidence suggests that the nitrilium ion (**IV**) is trapped by an alkoxyl group (i.e. -O'Bu) en route to the desired amide products. The method complements significantly the catalytic asymmetric alkylation reactions with vinyl azides.

ASSOCIATED CONTENT

Supporting Information. Experimental procedures and characterization data are available free of charge via the Internet at http://pubs.acs.org.

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