ISSN 1070-4280, Russian Journal of Organic Chemistry, 2010, Vol. 46, No. 5, pp. 644–648. © Pleiades Publishing, Ltd., 2010. Original Russian Text © E.A. Dikusar, V.I. Potkin, N.A. Zhukovskaya, 2010, published in Zhurnal Organicheskoi Khimii, 2010, Vol. 46, No. 5, pp. 655–659.

Synthesis of Chiral Azomethines Starting from L-Valine Methyl Ester Hydrochloride and Substituted Benzaldehydes of Vanillin Series

E. A. Dikusar, V. I. Potkin, and N. A. Zhukovskaya

Institute of Pysical Organic Chemistry, National Academy of Sciences of Belarus, Minsk, 220072 Belarus e-mail: evgen 58@mail.ru

Received June 9, 2009

Abstract—A preparative method of synthesis of chiral azomethines was developed consisting in the reaction of L-valine methyl ester hydrochloride reaction with substituted benzaldehydes of vanillin series.

DOI: 10.1134/S1070428010050076

L-Valine methyl ester hydrochloride (I) may serve as a con-venient and available synthon for the synthesis of chiral nitrogen-containing organic compounds [1–3]. Many among the derivatives of natural amino acids derivatives, in particular, of N-[O-(C)-carboranyl]alanine, exhibit high antibacterial and antitumor action [4].

The condensation of L-valine methyl ester hydrochloride (I) with substituted benzaldehydes II in



III, R = R' = H(a); R = H, R' = MeO(b); R = MeO, R' = HO(c), MeO(d), MeC(O)O(e), EtC(O)O(f), PrC(O)O(g), $Me_2CHC(O)O(h)$, BuC(O)O(i), $Me_2CHCH_2C(O)O(j)$, $Me(CH_2)_6C(O)O(k)$, $Me(CH_2)_8C(O)O(l)$, $Me(CH_2)_{16}C(O)O(m)$, $H_2C=C(Me)C(O)O(n)$, $C_6H_5CH(Me)CH_2C(O)O(o)$, $C_6H_5C(O)O(p)$, 2,4- $Cl_2C_6H_3C(O)O(q)$, 4- $BrC_6H_4C(O)O(r)$, 3- $O_2NC_6H_4C(O)O(s)$, MeOC(O)O(t), EtOC(O)O(u), 1-AdC(O)O(v), o- $HCB_{10}H_{10}CC(O)O(w)$, m- $HCB_{10}H_{10}CC(O)O(d)$, $Re_2CHC(O)O(c)$, EtC(O)O(d), PrC(O)O(e), $Me_2CHC(O)O(f)$, BuC(O)O(g), $Me_2CHCH_2C(O)O(b)$, 4- $MeC_6H_4C(O)O(i)$, MeOC(O)O(g), $Me_2CHCH_2C(O)O(h)$, 4- $MeC_6H_4C(O)O(i)$, MeOC(O)O(h), MeOC(O)O(h), MeC(O)O(h), MeOC(O)O(h), MeC(O)O(h), MeOC(O)O(h), MeC(O)O(h), MeC(h), MeC(h)

the presence of sodium hydrogen carbonate (at the stoichiometric reagents ratio 1:1:1) afforded new chiral alkylaromatic azomethines (Schiff bases) containing ether and ester groups IIIa–IIIx, IVa–IVn, V, VIa, VIb. The condensation was carried out at boiling the mixture of initial reagents in anhydrous methanol for 30–45 min. Chiral alkylaromatic azomethines IIIa–IIIx, IVa–IVn, V, VIa, VIb, VIb, were obtained in preparative yields 78–89%.



The obtained azomethines IIIa–IIIx, IVa–IVn, V, VIa, VIb are colorless or weakly colored thick oily fluids or crystalline substances, they do not require additional purification and are free of impurities of the initial compounds. The composition and structure of azomethines

IIIa–IIIx, IVa–IVn, V, VIa, VIb were proved by IR, UV, ¹H NMR spectra, elemental analysis, and cryoscopic measurement of the molecular weight. According to ¹H NMR spectra the purity of compounds obtained reached $94 \pm 2\%$.

In the IR spectra of azomethines IIIa–IIIx, IVa– IVn, V, VIa, VIb the following characteristic absorption bands were observed, v, cm⁻¹: 3080– 3000, 870–620 (CH_{arom}), 2995–2825 (CH_{aliph}), 1770– 1739 (C=O_{ester}) (IIIe–IIIx, IVc–IVn, VIa, VIb); 1643–1637 (C=N), 1607–1417 (C=C_{arom}), 1270–1033 (C–O) cm⁻¹. In the IR spectrum of compound IIIs the presence of NO₂ group was confirmed by the characteristic absorption bands in the region 1531 and 1348 cm⁻¹. In the IR spectra of carborane-containing azomethines IIIw, IIIx, IVv, IVn the absorption bands of C–H_{carborane} appeared at 3070 (IIIw, IVm) and 3032 (IIIx, IVn); of B–H, at 2680–2660 cm⁻¹.

In the UV spectra of compounds IIIe–IIIx, IVc– IVn, VIa, VIb characteristic absorption maxima are observed at λ_{max} , nm (e): 208 (13000), 220 (13000), 254 (9000), 300 (400) due to the presence in these molecules of the fragments of methyl (*E*,*S*)-2-(4-acyloxy-3alkoxybenzylideneamino)-3-methybutyrate.

The ¹H NMR spectra of azomethines IIIa–IIIx, IVa– IVn, V, VIa, VIb contain the signals of protons from Me₂C group as two doublets in the region 0.80–1.10 ppm, signals of CH_{aliph} as two multiplets in the region 2.15-2.60 and 3.30–3.70 ppm, signals of CO_2Me as a singlet in the range 3.65–3.75 ppm. In the ¹H NMR spectra of azomethines IIIb–IIIx, IVb, VIa the signals of MeO group appear as a singlet in the range 3.74–3.80 ppm, in the spectra of compounds IVa-IVn, VIb the protons of EtO groups give rise to a triplet in the range 0.90–1.30 ppm (Me) and a quartet at 3.80-4.20 ppm (CH₂). The signals of the aromatic protons in the spectra of compounds IIIa-IIIx, IVa-IVn, V, VIa, VIb are observed in the region 6.80–7.50 ppm, the protons of the azomethine group HC=N appear as a singlet in the region 8.11–8.22 ppm characteristic of (*E*)- azomethines [5].

In the IR, UV, and ¹H NMR spectra of azomethines IIIa–IIIx, IVa–IVn, V, VIa, VIb the appropriate absorption bands and proton signals prove the presence of the structural fragments of ester groups [6].

EXPERIMENTAL

IR spectra of compounds were recorded on FTIR spectrophotometer Nicolet Protege-460 from thin films

or in KBr pellets, UV spectra, on spectrophotometer Specord UV Vis from 1×10^{-4} M solution of compounds in methanol, ¹H NMR spectra were registered on a spectrometer Tesla BS-587A (100 MHz) from 5% solutions in deuterochloroform or DMSO- d_6 , intertnal reference TMS. The specific rotation was measured on a CM-2 instrument from 5% solutions of substances in methanol. Elemental analyses were carried out on a C,H,N,O,Sanalyzer Elementar Vario EL-III, accuracy of the measurements 0.1%. The molecular weights were determined by cryoscopy in benzene.

L-Valine methyl ester hydrochloride (I) used in the research was of "pure for analysis" grade (purity 98%), mp 171–172°C, $[\alpha]_D^{20}$ +23.4°. Vanillin and vanillal esters II were obtained by procedures [7–12].

Chiral alkylaromatic azomethins containing ether and ester groups IIIa–IIIx, IVa–IVn. A mixture of 5 mmol of L-valine methyl ester hydrochloride (I), 5 mmol of substituted benzaldehyde II, and 5 mmol of sodium hydrogen carbonate in 39 ml of anhydrous methanol was boiled for 30–45 min. The hot solution was filtered through a folded paper filter, the solvent was removed in a vacuum. Azomethines IIIa–IIIx, IVa–IVn were dissolved in 10 ml of anhydrous ethyl ether, the solution was filtered through a folded paper filter, the solvent was removed in a vacuum. The final purification was performed by column chromatography on neutral aluminum oxide, 40–100 µm, II grade Brockmann activity, eluent dichloromethane.

Methyl (*E*,*S*)-2-benzylideneamino-3-methylbutyrate (IIIa). Yield 84%, d_{20}^{20} 1.2630, n_D^{20} 1.5212, $[\alpha]_D^{20}$ -41.1°. Found, %: C 71.54; H 7.88; N 6.12. *M* 210.4. C₁₃H₁₇NO₂. Calculated, %: C 71.21; H 7.81; N 6.39. *M* 219.3.

Methyl (*E,S*)-3-methyl-2-(4-methoxybenzylideneamino)-butyrate (IIIb). Yield 88%, d_{20}^{20} 1.2801, n_D^{20} 1.5296, $[\alpha]_D^{20}$ -50.4°. Found, %: C 67.59; H 7.83; N 5.36. *M* 235.2. C₁₄H₁₉NO₃. Calculated, %: C 67.45; H 7.68; N 5.62. *M* 249.3.

Methyl (E,S)-2-(4-hydroxy-3-methoxy-benzylideneamino)-3-methylbutyrate (IIIc). Yield 78%, d_{20}^{20} 1.2864, n_D^{20} 1.5568, $[\alpha]_D^{20}$ -38.5°. Found, %: C 63.52; H 7.29; N 4.93. *M* 252.7. C₁₄H₁₉NO₄. Calculated, %: C 63.38; H 7.22; N 5.28. *M* 265.3.

Methyl (*E*,*S*)-2-(3,4-dimethoxybenzylideneamino)-3-methylbutyrate (IIId). Yield 89%, d_{20}^{20} 1.2814, n_D^{20} 1.5338, $[\alpha]_D^{20}$ -36.8°. Found, %: C 64.80; H 7.64;

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 46 No. 5 2010

N 4.85. *M* 270.5. C₁₅H₂₁NO₄. Calculated, %: C 64.50; H 7.58; N 5.01. *M* 279.3.

Methyl (*E*,*S*)-2-(4-acetyloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIIe). Yield 85%, d_{20}^{20} 1.2410, n_D^{20} 1.5192, $[\alpha]_D^{20}$ -32.7°. Found, %: C 62.53; H 6.89; N 4.56. *M* 299.6. C₁₆H₂₁NO₅. Calculated, %: C 62.53; H 6.89; N 4.56. *M* 307.3.

Methyl (*E*,*S*)-3-methyl-2-(3-methoxy-4-propionyloxybenzylideneamino)butyrate (IIIf). Yield 88%, d_{20}^{20} 1.2365, n_D^{20} 1.5198, $[\alpha]_D^{20}$ -31.0°. Found, %: C 63.86; H 7.35; N 4.07. *M* 314.8. C₁₇H₂₃NO₅. Calculated, %: C 63.54; H 7.21; N 4.36. *M* 321.4.

Methyl (*E*,*S*)-2-(4-butyryloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIIg). Yield 88%, d_{20}^{20} 1.2280, n_D^{20} 1.5084, $[\alpha]_D^{20}$ -27.2°. Found, %: C 64.72; H 7.68; N 3.85. *M* 321.0. C₁₈H₂₅NO₅. Calculated, %: C 64.46; H 7.51; N 4.18. *M* 335.4.

Methyl (*E,S*)-2-(4-isobutyryloxy-3-methoxybenzylidene-amino)-3-methylbutyrate (IIIh). Yield $82\%, d_{20}^{20}$ 1.2212, n_D^{20} 1.4988, $[\alpha]_D^{20}$ -25.9°. Found, %: C 64.58; H 7.60; N 3.89. *M* 322.4. C₁₈H₂₅NO₅. Calculated, %: C 64.46; H 7.51; N 4.18. *M* 335.4.

Methyl (*E*,*S*)-2-(4-valeroyloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIIi). Yield 85%, d_{20}^{20} 1.1840, n_D^{20} 1.5125, $[\alpha]_D^{20}$ -24.6°. Found, %: C 65.70; H 7.85; N 3.80. *M* 334.3. C₁₉H₂₇NO₅. Calculated, %: C 65.31; H 7.79; N 4.01. *M* 349.4.

Methyl(E,S)-2-(4-isovaleroyloxy-3-
methoxybenz-ylideneamino)-3-methylbutyrate(IIIj). Yield 84%, d_{20}^{20} 1.1748, n_D^{20} 1.4926, $[\alpha]_D^{20}$ -25.2°.Found, %: C 65.76; H 7.92; N 3.82. M 335.1.C₁₉H₂₇NO₅. Calculated, %: C 65.31; H 7.79; N 4.01.
M 349.4.

Methyl (*E*,*S*)-2-(4-capryloyloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIIk). Yield 81%, d_{20}^{20} 1.1730, n_D^{20} 1.4944, $[\alpha]_D^{20}$ -22.3°. Found, %: C 67.12; H 8.34; N 3.46. *M* 368.2. C₂₁H₃₁NO₅. Calculated, %: C 66.82; H 8.28; N 3.71. *M* 377.5.

Methyl (*E*,*S*)-2-(4-decanoyloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIII). Yield 83%, d_{20}^{20} 1.1645, n_D^{20} 1.4972, $[\alpha]_D^{20}$ -21.7°. Found, %: C 68.37; H 8.82; N 3.18. *M* 389.5. C₂₃H₃₅NO₅. Calculated, %: C 68.12; H 8.70; N 3.45. *M* 405.5.

Methyl (*E,S*)-3-methyl-2-(3-methoxy-4stearoyloxy-benzylideneamino)butyrate (IIIm). Yield 80%, mp 37–38°C (methanol), $[α]_D^{20}$ –17.4°. Found, %: C 72.23; H 10.07; N 2.49. *M* 501.6. C₃₁H₅₁NO₅. Calculated, %: C 71.91; H 9.93; N 2.71. *M* 517.7.

Methyl (*E*,*S*)-2-(4-methacryloyloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIIn). Yield $84\%, d_{20}^{20}$ 1.210, n_D^{20} 1.5130, $[[\alpha]_D^{20}$ -30.3°. Found, %: C 65.14; H 7.08; N 4.13. *M* 324.8. C₁₈H₂₃NO₅. Calculated, %: C 64.85; H 6.95; N 4.20. *M* 333.4.

Methyl (*E*,*S*)-3-methyl-2-[3-methoxy-4-(3phenyl-butyryloxy)benzylideneamino]-butyrate (IIIo). Yield 84%, d_{20}^{20} 1.2303, n_D^{20} 1.5352, $[\alpha]_D^{20}$ -21.8°. Found, %: C 70.26; H 7.25; N 3.06. *M* 401.2. C₂₄H₂₉NO₅. Calculated, %: C 70.05; H 7.10; N 3.40. *M* 411.5.

Methyl (*E*,*S*)-2-(4-benzoyloxy-3-methoxybenzylideneamino)-3-methylbutyrate (IIIp). Yield 89%, d_{20}^{20} 1.2415, n_D^{20} 1.5512, $[\alpha]_D^{20}$ -25.3°. Found, %: C 68.47; H 6.38; N 3.37. *M* 355.7. C₂₁H₂₃NO₅. Calculated, %: C 68.28; H 6.28; N 3.79. *M* 369.4.

Methyl (*E*,*S*)-2-[4-(2,4-dichlorobenzoyloxy)-3methoxy-benzylideneamino]-3-methylbutyrate (IIIq). Yield 83%, d_{20}^{20} 1.3419, n_D^{20} 1.5578, $[\alpha]_D^{20}$ -20.6°. Found, %: C 57.84; H4.82; Cl 15.85; N 2.93. *M* 419.1. C₂₁H₂₁Cl₂NO₅. Calculated, %: C 57.55; H 4.83; Cl 16.18; N 3.20. *M* 438.3.

Methyl (*E*,*S*)-2-[4-(4-bromobenzoyloxy)-3methoxy-benzylideneamino]-3-methyl-butyrate (IIIr). Yield 85%, d_{20}^{20} 1.3503, n_D^{20} 1.5668, $[\alpha]_D^{20}$ -21.4°. Found, %: C 56.61; H 5.07; Br 17.36; N 2.90. *M* 425.8. C₂₁H₂₂BrNO₅. Calculated, %: C 56.26; H 4.95; Br 17.82; N 3.12. *M* 448.3.

Methyl (*E,S*)-3-methyl-2-[-3-methoxy-4-(3-nitrobenzoyl-oxy)benzylideneamino]butyrate (IIIs). Yield 80%, mp 73–74°C, $[\alpha]_D^{20}$ –19.7°. Found, %: C 61.11; H 5.38; N 6.43. *M* 402.6. C₂₁H₂₂N₂O₇. Calculated, %: C 60.86; H 5.35; N 6.76. *M* 414.4.

Methyl (*E,S*)-3-methyl-2-(4-methoxycarbonyloxy-3-methoxybenzylideneamino)butyrate (IIIt). Yield 79%, d_{20}^{20} 1.2246, n_D^{20} 1.5065, $[\alpha]_D^{20}$ -32.5°. Found, %: C 59.84; H 6.70; N 4.02. *M* 316.5. C₁₆H₂₁NO₆. Calculated, %: C 59.43; H 6.55; N 4.33; *M* 323.3.

Methyl (*E*,*S*)-3-methyl-2-(-3-methoxy-4-ethoxycarbonyloxybenzylideneamino)butyrate (IIIu). Yield 82%, d_{20}^{20} 1.2204, n_D^{20} 1.5048, $[\alpha]_D^{20}$ -30.4°. Found, %: C 60.88; H 6.94; N 3.81. *M* 322.0. C₁₇H₂₃NO₆. Calculated, %: C 60.52; H 6.87; N 4.15. *M* 337.4.

Methyl (E,S)-2-[4-(1-adamantanemethanoyloxy)-

3-methoxybenzylideneamino]-3-methylbutyrate (IIIv). Yield 88%, mp 60–61°C, $[\alpha]_D^{20}$ –26.6°. Found, %: C 70.56; H 7.99; N 2.97. *M* 412.5. C₂₅H₃₃NO₅. Calculated, %: C 70.23; H 7.78; N 3.28. *M* 427.5.

Methyl (*E*,*S*)-2-[4-(*O*-carborane-C-methanoyloxy)-3-methoxybenzylideneamino]-3-methylbutyrate (IIIw). Yield 83%, mp 76–77°C, $[\alpha]_D^{20}$ –21.7°. Found, %: C 48.25; H 4.52; B 25.12; N 2.93. *M* 405.6. C₁₇H₁₉B₁₀NO₅. Calculated, %: C 47.99; H 4.50; B 25.41; N 3.29. *M* 425.4.

Methyl (*E*,*S*)-2-[4-(μ -carborane-C-methanoyloxy)-3-methoxybenzylideneamino]-3-methylbutyrate (IIIx). Yield 80%, mp 85–86°C, [α]_D²⁰ –22.5 [α]_D²⁰. Found, %: C 48.34; H 4.59; B 25.02; N 2.98. *M* 410.2. C₁₇H₁₉B₁₀NO₅. Calculated, %: C47.99; H 4.50; B 25.41; N 3.29. *M* 425.4.

Methyl (*E*,*S*)-2-(4-hydroxy-3-ethoxy-benzylideneamino)-3-methylbutyrate (IVa). Yield 80%, d_{20}^{20} 1.2835, n_D^{20} 1.5362, $[\alpha]_D^{20}$ -36.5°. Found, %: C 64.57; H 7.66; N 4.82. *M* 265.4. C₁₅H₂₁NO₄. Calculated, %: C 64.50; H 7.58; N 5.01. *M* 279.3.

Methyl (*E,S*)-3-methyl-2-(4-methoxy-3-ethoxybenzylideneamino)butyrate (IVb). Yield 83%, d_{20}^{20} 1.2663, n_D^{20} 1.5255, $[\alpha]_D^{20}$ -38.1°. Found, %: C 65.70; H 8.00; N 4.34. *M* 281.4. C₁₆H₂₃NO₄. Calculated, %: C 65.51; H 7.90; N 4.77. *M* 293.4.

Methyl (*E,S*)-2-(4-acetyloxy-3-ethoxybenzylideneamino)-3-methylbutyrate (IVc). Yield 88%, d_{20}^{20} 1.2370, n_D^{20} 1.5208, $[\alpha]_D^{20}$ -37.2°. Found, %: C 63.80; H 7.32; N 4.02. *M* 313.2. C₁₇H₂₃NO₅. Calculated, %: C 63.54; H 7.21; N 4.36. *M* 321.4.

Methyl (*E*,*S*)-3-methyl-2-(4-propionyloxy-3-oxybenzylideneamino)butyrate (IVd). Yield 87%, d_{20}^{20} 1.2342, n_D^{20} 1.5036, $[\alpha]_D^{20}$ -36.8°. Found, %: C 64.67; H 7.68; N 3.95. *M* 362.3. C₁₈H₂₅NO₅. Calculated, %: C 64.46; H 7.51; N 4.18. *M* 335.4.

Methyl (*E,S*)-2-(4-butyryloxy-3-ethoxybenzylideneamino)-3-methylbutyrate (IVe). Yield 82%, d_{20}^{20} 1.2235, n_D^{20} 1.5070, $[\alpha]_D^{20}$ -34.0°. Found, %: C 65.67; H 7.86; N 3.83. *M* 335.3. C₁₉H₂₇NO₅. Calculated, %: C 65.31; H 7.79; N 4.01. *M* 349.4.

Methyl (*E*,*S*)-2-(4-isobutyryloxy-3-ethoxybenzylidene-amino)-3-methylbutyrate (IVf). Yield 86%, d_{20}^{20} 1.2206, n_D^{20} 1.5012, $[\alpha]_D^{20}$ -27.5°. Found, %: C 65.63; H 7.82; N 3.80. *M* 339.4. C₁₉H₂₇NO₅. Calculated, %: C 65.31; H 7.79; N 4.01. *M* 349.4. idene-amino)-3-methylbutyrate (IVg). Yield 86%, d_{20}^{20} 1.1817, n_D^{20} 1.5170, $[\alpha]_D^{20}$ -25.6°. Found, %: C 66.36; H 8.27; N 3.61. *M* 349.5. C₂₀H₂₉NO₅. Calculated, %: C 66.09; H 8.04; N 3.85. *M* 363.4.

Methyl (*E*,*S*)-2-(4-isovaleroyloxy-3-ethoxybenzylidene-amino)-3-methylbutyrate (IVh). Yield 85%, d_{20}^{20} 1.1714, n_D^{20} 1.5028, $[\alpha]_D^{20}$ -24.8°. Found, %: C 66.31; H 8.20; N 3.52. *M* 351.6. C₂₀H₂₉NO₅. Calculated, %: C 66.09; H 8.04; N 3.85. *M* 363.4.

Methyl (*E*,*S*)-3-methyl-2-[4-(4-methyl-benzoyloxy)-3-ethoxybenzylideneamino]butyrate (IVi). Yield 85%, d_{20}^{20} 1.2226, n_D^{20} 1.5490, $[\alpha]_D^{20}$ -20.3°. Found, %: C 69.64; H 6.97; N 3.19. *M* 385.8. C₂₃H₂₇NO₅. Calculated, %: C 69.50; H 6.85; N 3.52. *M* 397.5.

Methyl (*E*,*S*)-2-{[4-(methoxycarbonyl)oxy]-3ethoxybenzylideneamino}-3-methylbutyrate (IVj). Yield 80%, d_{20}^{20} 1.2208, n_D^{20} 1.5090, $[\alpha]_D^{20}$ -34.8°. Found, %: C 60.80; H 6.99; N 3.92. *M* 328.3. C₁₇H₂₃NO₆. Calculated, %: C 60.52; H 6.87; N 4.15. *M* 337.4.

Methyl (*E*,*S*)-3-methyl-2-(3-ethoxy-4-ethoxycarbonyl-oxybenzylideneamino)butyrate (IVk). Yield 82%, d_{20}^{20} 1.2168, n_D^{20} 1.5060, $[\alpha]_D^{20}$ -32.5°. Found, %: C 61.87; H 7.23; N 3.76. *M* 337.4. C₁₈H₂₅NO₆. Calculated, %: C 61.52; H 7.17; N 3.99. *M* 351.4.

Methyl (*E,S*)-2-[4-(1-adamantanemethanoyloxy)-3-ethoxybenzylideneamino]-3-methylbutyrate (IVI). Yield 86%, d_{20}^{20} 1.2658, n_D^{20} 1.5280, $[\alpha]_D^{20}$ -21.0°. Found, %: C 70.93; H 8.13; N 2.90. *M* 428.7. C₂₆H₃₅NO₅. Calculated, %: C 70.72; H 7.99; N 3.17. *M* 441.6.

Methyl (*E*,*S*)-2-[4-(*O*-carborane-C-methanoyloxy)-3-ethoxybenzylideneamino]-3-methylbutyrate (IVm). Yield 80%, mp 70–71°C, $[\alpha]_D^{20}$ –19.7°. Found, %: C 49.46; H 5.06; B 24.18; N 2.90. *M* 418.7. C₁₈H₂₁B₁₀NO₅. Calculated, %: C 49.19; H 4.82; B 24.60; N 3.19. *M* 439.5.

Methyl (*E*,*S*)-2-[4-(*m*-carborane-C-methanoyloxy)-3-ethoxybenzylideneamino]-3-methylbutyrate (IVn). Yield 81%, mp 79–80°C, $[\alpha]_D^{20}$ –19.1°. Found, %: C 49.50; H 5.03; B 24.11; N 2.92. *M* 422.3. C₁₈H₂₁B₁₀NO₅. Calculated, %: C 49.19; H 4.82; B 24.60; N 3.19. *M* 439.5.

Diazomethines V, VIa, VIb were obtained and isolated analogously to compounds **IIIa–IIIx, IVa–IVn**. A mixture of 10 mmol of compound I, 5 of dialdehyde, and 10 mmol of sodium hydrogen carbonate in 50 ml of anhydrous methanol was boiled for 45 min.

Methyl (E,S)-2-(4-valeroyloxy-3-ethoxybenzyl-

Dimethyl (S,S)-2,2'-{biphenyl-4,4'-diylbis[(E)-

RUSSIAN JOURNAL OF ORGANIC CHEMISTRY Vol. 46 No. 5 2010

methylidenenitrilo]}-3,3'-dimethyldibutanoate (**V**). Yield 84%, mp 124–125°C, $[\alpha]_D^{20}$ –62.6°. Found, %: C 71.69; H 7.52; N 6.14. *M* 415.9. C₂₆H₃₂N₂O₄. Calculated, %: C 71.53; H 7.39; N 6.42. *M* 436.5.

Di[2-methoxy-4-(*E*,*S***)-(2-methyl-1-methoxy-carbonyl-propyliminomethyl)phenyl]succinate (VIa).** Yield 87%, mp 67–68°C, $[\alpha]_D^{20}$ –40.3°. Found, %: C 62.95; H 6.59; N 4.18. *M* 598.2. C₃₂H₄₀N₂O₁₀. Calculated, %: C 62.73; H 6.58; N 4.57. *M* 612.7.

Di[4-(*E*,*S*)-(2-methyl-1-methoxycarbonylpropyliminomethyl)-2-ethoxyphenyl]succinate (VIb). Yield 83%, mp 62–63°C, $[\alpha]_D^{20}$ –39.1°. Found, %: C 64.16; H 7.08; N 4.05. *M* 618.5. C₃₄H₄₄N₂O₁₀. Calculated, %: C 63.74; H 6.92; N 4.37. *M* 640.7.

REFERENCES

- 1. Meierhenrich, U.J., Amino Acids and the Asymmetry of Life, Berlin: Springer-Verlag, 2008.
- Barrett, G.C., Amino Acid Derivatives: A Practical Approach (Practical Approach Series), Oxford University Press, 2000.
- 3. Paquette, L.A., *Handbook of Reagents for Organic Synthesis Chiral Reagents for Asymmetric Synthesis*, New York: Wiley, 2003, p. 582.

- Oros, G., Ujvary, I., and Nachman, R.J., *Amino Acids*, 1999, vol. 17, p. 357.
- Daier, D.R., Prilozheniya absorbtsionnoi spektroskopii organicheskikh soedinenii (Applications of Absorption Spectroscopy of Organic Copmounds), Moscow: Khimiya, 1970.
- Dikusar, E.A., Kozlov, N.G., Tlegenov, R.T., and Uteniyazov, K., *Azometiny na osnove vanilina i vanilalya* (Azomethines on the Basis of Vaniline and Valile), Nukus: "Karakalpakstan," 2007, 207.
- Dikusar, E.A., Vyglazov, O.G., Moiseichuk, K.L., Zhukovskaya, N.A., and Kozlov, N.G., *Zh. Prikl. Khim.*, 2005, vol. 78, p. 122.
- Dikusar, E.A.and Kozlov, N.G., *Khim. Polim. Soedin.*, 2005, p. 74.
- Dikusar, E.A., Kozlov, N.G., *Zh. Org. Khim.*, 2005, vol. 41, p. 1015.
- 10. Dikusar, E.A., Zh. Prikl. Khim., 2006, vol. 79, p. 1043.
- Dikusar, E.A., Kozlov, N.G., Potkin, V.I., Zvereva, T.D., Yuvchenko, A.P., Bei, M.P., and Kovganko, N.V., *Khim. Polim. Soedin.*, 2006, p. 434.
- Dikusar, E.A., Potkin, V.I., Kozlov, N.G., Yuvchenko, A.P., Bei, M.P., and Kovganko, N.V., *Zh. Org. Khim.*, 2008, vol. 44, p. 1321.