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PII: S0040-4020(17)30299-5

DOI: 10.1016/j.tet.2017.03.056

Reference: TET 28562

To appear in: Tetrahedron

Received Date: 24 January 2017

Revised Date: 2 March 2017

Accepted Date: 20 March 2017

Please cite this article as: Kavas E, Altug C, Direct access to 1,4-benzothiazine 4,4-dioxides and 4-oxides *via* a domino reaction, *Tetrahedron* (2017), doi: 10.1016/j.tet.2017.03.056.

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Direct access to 1,4-benzothiazine 4,4-dioxides and 4-oxides via a domino reaction

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ARTICLE INFO

ABSTRACT

 Article history:

 Received

 Received in revised form

 Accepted

 Available online

 Keywords:

 1,4-benzothiazine

 a-chloro oxime

 isoxazole

 one-pot

 cyclization

 domino reaction

1. Introduction

A domino reaction includes a consecutive series of organic reactions, which often proceed with highly reactive intermediates in a reaction vessel or in nature.^{1,2} Single acyclic organic starting materials can be converted into multifunctional cyclic or heterocyclic organic molecules in this way.³ It is well known that phenothiazine analogues of 1,4-benzothiazines have stimulated considerable interest in recent years due to their broad spectrum of pharmacological properties.⁴ Their biological features have been well documented over the years.^{5,6} Structure–activity analyses have revealed that the biological properties of 1,4-benzothiazines are strongly dependent on substitution, which makes them pharmacologically relevant and highly attractive.⁷ In particular, antimalarial activity,⁸ histamine H-receptor antagonist,⁹ central nervous system depressants,¹⁰ antifungal agents,¹¹ calcium ion channel antagonists,¹² potassium ion channel openers¹³ and anticancer properties¹⁴ make this class of compounds good candidates for biological studies.

In addition, isoxazole-fused heterocyclic systems have broad application in both synthetic organic chemistry and biological studies.^{15–18} The synthesis of new heterocyclic systems, the development of facile procedures for the synthesis of these compounds and the improvement of available methods have been, and still remain, important areas of research in synthetic organic chemistry.

Several synthetic methods have been developed for the synthesis of 1,4-benzothiazines.⁴ Reported procedures for the preparation of 1,4-benzothiazines include the reaction of 2-

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The domino reactions of 2-fluoro benzensulfonyl acetonitrile and α -chloro oximes in the presence of Cs₂CO₃ in aprotic high boiling point solvents have been achieved to provide isoxazole–fused 4*H*-1,4-benzothiazine-4,4-dioxides *via* an unprecented transition metal-free one-pot addition/cyclization process. The tunable synthesis of either isoxozolo-1,4-benzothiazin-4-oxides or their precursor 5-aminoisoxazoles can be controlled depending on the solvent selection. The observed products were characterized by means of (IR, ¹H, ¹³C NMR and HRMS) and physical methods.

> aminothiols with unsaturated C-C bonds,^{19,20} ring expansions or ring cleavages of various heterocyclic systems^{21–24} and metal catalyzed C-S bond formations^{25–27} but, to the best of our knowledge, there is no literature concerning the synthesis of heterocycles, which contain isoxazolo-1,4-benzothiazines from simple starting materials *via* domino reactions. In addition, reports of isoxazole-fused heterocyclic systems are very limited in the literature.^{28,29}

> In continuation of our isoxazole chemistry,^{30–34} herein, we have developed a robust, convenient and atom-economical method to obtain potentially bioactive isoxazolobenzothiazines by reacting *ortho*-fluoro substituted phenylsulfonyl acetonitrile and α -chloro oximes. The reaction conditions produce an enamino moiety which is used as a reactive intermediate for obtaining the title There are examples of the enamino compounds. intramolecular nucleophilic substitution reaction with orthosubstituted halides to obtain 1,4-benzothiazine 4,4-dioxides. Lautens' group prepared enamines by the Rh-catalyzed addition of arylboronic acids to (0 fluorophenylsulfonyl)acetonitrile. The subsequent intramolecular nucleophilic substitution of those enamines gave *N*-unsubstituted 2-aryl-1,4-benzothiazine S,Sdioxides.³⁵ Another method, employing 2-aminopyrroles bearing a 2-chlorophenylsulfone group at the 3-position to obtain 1,4-benzothiazines via Pd-catalyzed intramolecular cyclization with carboxylic acid auxiliary, has also been reported.³⁶ However, the aforementioned studies required expensive metal catalysts to obtain either the starting materials or cyclization products. Recently, Gu et al. developed a method to obtain benzothiazines via the K₂S initiated sulfur insertion into enaminones without using a transition-metal catalyst.³⁷ In our work, a one-pot metal-free synthesis of isoxazolo-1,4-benzothiazine-4 oxides and 4,4dioxides is reported for the first time. This method enables

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several advantages such as simple work-up procedure, high yields, broad applicability and practicability.

2-((2-Fluorophenyl)sulfonyl)acetonitrile (1) and 2-((2fluorophenyl)sulfinyl)acetonitrile (4) were obtained from the oxidation of 2-((2-fluorophenyl)thio)acetonitrile with H₂O₂ (30%)and *m*-CPBA respectively. Then. 2 - ((2 fluorophenyl)sulfonyl)acetonitrile (1) reacted with α chlorooximes to furnish isoxazolo[5,4-e][1,4]benzothiazine 4,4-dioxides 3a-k and reaction conditions were optimized with conventional and microwave heating methods, see Table 1. Reactions with microwave heating had slightly higher yields and shorter reaction times than conventional heating. The reaction yields were similar for both electronwithdrawing and electron-releasing substituents of achlorooximes.

Table 1 Reactions of 2-((2-fluorophenyl)sulfonyl) acetonitrile (1) with α -chloro oximes (2)

0 N=	° S F +	$\begin{array}{c} Ar \\ & \\ \\ CI \end{array} \xrightarrow{OH} \begin{array}{c} CS_2CO_3 \\ \hline \\ DMSO \end{array}$	-	
	1	2		3
Entry	Compound	Ar	Yields* (%)	Yields** (%)
1	3a	2,6-di-ClC ₆ H ₃	30	37
2	3b	2,4-di-ClC ₆ H ₃	53	67
3	3c	4-F-C ₆ H ₄	40	58
4	3d	$4-Br-C_6H_4$	54	58
5	3e	$4-Cl-C_6H_4$	59	67
6	3f	C_6H_5	46	51
7	3g	2-Cl-4,5-diMeO-C ₆ H ₃	58	67
8	3h	$2-O_2N-C_6H_4$	50	69
9	3i	$4-O_2N-C_6H_4$	54	68
10	3ј	5-Cl-2-thienyl	65	72
11	3k	5-Cl-2-furyl	52	64

Yields obtained by conventional heating method.

** Yields obtained by microwave heating method.

Having examined the scope of the sulfone functionality in the reaction, we next examined the synthesis of isoxazolo-1,4-benzothiazin-4-oxides. The reaction in a one-pot manner gave lower yields (for **6a-d**), but firstly isolation of 5aminoisoxazole **5e-h** intermediate and then, its heating in high boiling point aprotic solvents (DMSO or DMF) in the presence of Cs_2CO_3 afforded the corresponding isoxazolo-1,4-benzothiazine 4-oxides **6a-h** with better yields, see Table 2. Table 2 Synthesis of 1,4-benzothiazine-4-oxides



^aYields obtained through one-pot single step reaction. ^bYields obtained through isolation of compounds **5** e-h and then, reaction in the presence of Cs₂CO₃ in DMSO.

The structures of all compounds were elucidated according to IR, ¹H and ¹³C NMR spectroscopy and HRMS methods. The IR spectra of compounds 3a-k showed NH stretching between 3244 cm⁻¹ and 3214 cm⁻¹, SO₂ asymmetric stretching of compounds 3a-k was seen at 1342-1261 cm⁻¹ and symmetric stretching at 1173–1142 cm⁻¹. While NH₂ asymmetric and symmetric stretching frequencies for compounds **5e-h** appeared between 3383-3310 cm⁻¹ and 3317–3171 cm⁻¹, NH stretching frequencies of compounds **6a-h** were observed between 3445 and 3433 cm⁻¹. The compounds 5e-h and 6a-h exhibited a SO stretching frequency at 1007–1069 cm⁻¹ and 999–957 cm⁻¹. respectively. The ¹H NMR spectra of compounds 3a-k and 6a-h displayed a broad singlet for NH protons between 13.28-13.14 ppm and 13.21-12.97 ppm in DMSO respectively. When the ¹H NMR was run in DMSO, the amino protons resonated at 8.08-8.29 ppm due to interaction with water in DMSO, when same pattern of compound 5f was run in CDCl₃, the amino group protons resonated at 5.57 ppm. The ¹³C NMR spectra of compounds **3a-k** contained a signal in the range of 163.0-161.5, 158.7-147.9 and 95.1-91.3 ppm, which are in accordance with isoxazole carbons. There is no dramatic effect of the sulfoxide or sulfone groups on the carbon signals in isoxazole ring for the two series of compounds. In addition, the isoxazole ring showed similar signals for compounds 5e-h for the C5 carbon at 171.0-169.1 ppm, for the C4 carbon at 161.2–159.3 ppm, for the C3 carbon at 91.9–90.6 ppm and for compounds 6a-h for the C5 carbon at 161.8-160.7, for the C4 carbon at 160.2-157.6, for the C3 carbon at 95.0-93.0 ppm. Moreover, the proposed intermolecular nucleophilic substitution reaction can be easily understood by the disappearance of C-F couplings from starting materials (**5e–h**). When the ¹³C NMR of compound **6a** was analyzed the restricted rotation for the 2,6-dichloro substituents results in each carbon atom on the ring having different resonances. HRMS results were in agreement with the expected molecular weights of all novel compounds.

Regarding the reaction mechanism, it would be reasonable to assume that two reaction mechanisms may occur until the formation of 5-aminoisoxazoles. First proposed mechanism may work through deprotonation of the carbon atom next to the nitrile occurs. This would then be followed by nucleophilic addition to the iminic carbon atom of α -hydroxymoyl chlorides. Finally, the cyclisation occurs with firstly the attack of the OH onto the nitrile carbon atom to give 5-aminoisoxazoles. Another mechanism may involve, under basic conditions α -chloro oximes will form nitrile oxides. These will then engage in 1,3-dipolar cycloaddition with the deprotonated starting material to give 5-aminoisoxazoles. Then an S_NAr displacement reaction of fluoride with NH₂ group results in cyclization 1,4-benzothiazines.^{38,39}

2. Conclusion

In conclusion, we have developed a very efficient synthesis of the new tricyclic isoxazole–fused 1,4-benzothiazine 4,4-dioxides and 4-oxides *via* a domino reaction under transition-metal free conditions.

3. Experimental Section

Anhydrous solvents and all reagents were purchased from Sigma-Aldrich and Carlo Erba. NMR spectra were recorded in DMSO-d₆ or CDCl₃ at 400 or 300 MHz (for ¹H NMR) and at 125 or 100 MHz (for ¹³C NMR) with Jeol 400 or Bruker Avance 300 instruments at 25 °C. Chemical shifts are reported in parts per million (ppm) and the coupling constants (J) are expressed in Hertz (Hz). Splitting patterns are designated as follows: s (singlet); d (doublet); m (multiplet); broad s (broad singlet); dd (doublet of doublets); app s (apparent singlet); app d (apparent doublet); app t (apparent triplet). Fourier Transform Infrared (FT-IR) spectra were recorded on a SHIMADZU FT-IR-8400S. The mass analyses were performed on a Waters 2695 Alliance Micromass ZQ (LC-MS) and Waters SYNAPT (HRMS). Analytical thin-layer chromatography (TLC) was carried out on Merck silica gel F-254 plates. Flash chromatography purifications were performed on Merck Silica gel 60 (230-400 mesh ASTM) as the stationary phase and ethyl acetate/nhexane were used as eluents. Melting points (mp) were measured in open capillary tubes with a MELTEMP apparatus and are uncorrected.

General Synthesis of 3-Aryl-9H-benzo[b]isoxazolo[5,4e][1,4] thiazine 4,4-dioxide (Conventional Heating Method)

2-((2-Fluorophenyl)sulfonyl)acetonitrile (199 mg, 1.0 mmol) and α -chloro oxime (1.0 mmol) were dissolved in DMSO (5 mL). Subsequently, Cs₂CO₃ (652 mg, 2.0 mmol) was added to this solution and the mixture was stirred at reflux for 2 h. The progress of the reaction was monitored by TLC. The mixture was extracted with EtOAc (3 x 15 mL), subsequently washed with brine (15 mL) and dried (Na₂SO₄).

The solvent was evaporated under reduced pressure and the resulting solid was recrystallized from benzene.

General Synthesis of 3-Aryl-9H-benzo[b]isoxazolo[5,4e][1,4] thiazine 4,4-dioxide (Microwave Heating Method)

(2-Fluorobenzensulfonyl)acetonitrile (199 mg, 1.0 mmol) and α -chloro oxime (1.0 mmol) were dissolved in DMSO (5 mL). To this mixture Cs₂CO₃ (652 mg, 2.0 mmol) was added. The mixture was heated in microwave reactor at 200 W, 150 °C for 10 min. Progress of the reaction was monitored by TLC. After completion of the reaction, the mixture was washed with ice-cold water (10 mL) and extracted with EtOAc (3x15 mL). Then, the combined organic layers were washed with brine and dried (Na₂SO₄). The solvent was removed under reduced pressure and the remaining solid was recrystallized from benzene.

3-(2,6-Dichlorophenyl)-9H-benzo[b]isoxazolo[5,4e][1,4]thiazine 4,4-dioxide (3a)

White solid; 112 mg, 30% (conventional), 136 mg, 37% (microwave); mp 228–230 °C; $R_f = 0.45$ (EtOAc 1:1 Hex). IR (KBr): 3244 (NH), 3090 (CH), 1636 (C=N) and 1261–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 13.28$ (br s, 1H, NH), 7.92 (dd, J = 8.2, 1.3 Hz, 1H), 7.76 – 7.72 (m, 1H), 7.72 – 7.63 (m, 3H), 7.45 – 7.41 (m, 1H), 7.40 – 7.37 (m, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆): $\delta = 161.9$, 154.0, 134.6, 133.6, 133.5, 133.1, 128.5, 125.7, 124.5, 124.2, 122.9, 118.2, 94.4. MS (TOF ES⁺): m/z (%) = 367 (M+H⁺, 100%). HRMS: *m/z* [M+H]⁺ calc. for C₁₅H₉³⁷Cl₂N₂O₃S: 366.9711; found: 366.9706.

3-(2,4-Dichlorophenyl)-9H-benzo[b]isoxazolo[5,4e][1,4]thiazine 4,4-dioxide (3b)

White solid; 229 mg, 62% (conventional), 246 mg, 67% (microwave); mp 229–231 °C; $R_f = 0.46$ (EtOAc 1:1 Hex). IR (KBr): 3214 (NH), 3090 (CH), 1632 (C=N) and 1265–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 13.20$ (br s, 1H, NH), 7.93 (dd, J = 8.2, 1.1 Hz, 1H), 7.91 (d, J = 2.0 Hz, 1H), 7.78 (d, J = 8.3 Hz, 1H), 7.75 – 7.70 (m, 1H), 7.68 (dd, J = 8.3, 2.0 Hz, 1H), 7.43 – 7.36 (m, 2H). ¹³C NMR (100 MHz, DMSO-*d*₆): $\delta = 162.4$, 155.6, 137.0, 134.2, 134.0, 133.8, 133.4, 130.4, 128.4, 126.2, 124.9, 124.6, 123.5, 118.7, 95.1. MS (TOF ES⁺): m/z (%) = 367 (M+H⁺, 100%), 369 (65%), 329 (28%), 307 (18%), 143 (12%). HRMS: *m/z* [M+H]⁺ calc. for $C_{15}H_9{}^{37}Cl_2N_2O_3S$: 366.9711; found: 366.9703.

3-(4-Florophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4,4-dioxide (3c)

Light yellow solid; 126 mg, 40% (conventional), 184 mg, 58% (microwave); mp 239–240 °C; $R_f = 0.45$ (EtOAc 1:1 Hex). IR (KBr): 3244 (NH), 3090 (CH), 1632 (C=N) and 1269–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO- d_6) $\delta = 13.16$ (br s, 1H, NH), 8.11 (dd, J = 8.9, ⁴ J_{H-F} 5.4 Hz, 2H), 8.01 (dd, J = 8.1, 1.0 Hz, 1H), 7.75 – 7.70 (m, 1H), 7.47 (t, J = 8.9 Hz, 2H), 7.43 – 7.38 (m, 2H). ¹³C NMR (100 MHz, DMSO- d_6): $\delta = 164.3$ (d, ¹ $J_{C-F} = 249.6$ Hz), 162.5, 157.7, 134.2, 133.5, 130.9 (d, ³ $J_{C-F} = 9.0$ Hz), 125.8, 124.5, 123.8, 123.3 (d, ⁴ $J_{C-F} = 3.2$ Hz), 118.4, 116.9 (d, ² $J_{C-F} = 22.2$ Hz), 93.6. MS (TOF ES⁺): m/z (%) = 317 (M+H⁺, 100%), 281 (15%), 271 (17%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₀FN₂O₃S: 317.0396; found: 317.0407.

3-(4-Bromophenyl)-9H-benzo[b]isoxazolo[5,4e][1,4]thiazine 4,4-dioxide (3d)

White solid; 204 mg, 54% (conventional), 216 mg, 58% (microwave); mp 253–255 °C; $R_f = 0.45$ (EtOAc 1:1 Hex). IR (KBr): 3229 (NH), 3090 (CH), 1643 (C=N) and 1269–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ = 13.19 (br s, 1H, NH), 8.02 – 8.00 (m, 1H), 7.98 (d, *J* = 8.5 Hz, 2H), 7.83 (d, *J* = 8.5 Hz, 2H), 7.72 (t, *J* = 7.8 Hz, 1H), 7.41 (t, *J* = 7.8 Hz, 2H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ = 162.2, 157.3, 133.8, 133.0, 132.4, 129.9, 125.6, 125.3, 125.2, 124.0, 123.3, 118.0, 93.1. MS (TOF ES⁺): m/z (%) = 377 (M+H⁺, 88%), 379 (100%), 329 (20%), 307 (20%), 291 (17%). HRMS: *m*/z [M+H]⁺ calc. for C₁₅H₁₀⁷⁹BrN₂O₃S: 376.9596; found: 376.9585.

3-(4-Chlorophenyl)-9H-benzo[b]isoxazolo[5,4e][1,4]thiazine 4,4-dioxide (3e)

White solid; 196 mg, 59% (conventional), 222 mg, 67% (microwave); mp 241–242 °C; $R_f = 0.46$ (EtOAc 1:1 Hex). IR (KBr): 3244 (NH), 3090 (CH), 1632 (C=N) and 1273–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO- d_6): $\delta = 13.16$ (br s, 1H, NH), 8.06 (d, J = 8.7 Hz, 2H), 8.02 (dd, J = 8.1, 1.2 Hz, 1H), 7.74 (app. dd, J = 7.2, 1.4 Hz, 1H), 7.70 (d, J = 8.7 Hz, 2H), 7.44 – 7.38 (m, 2H). ¹³C NMR (101 MHz, DMSO- d_6): $\delta = 162.1$, 157.2, 136.3, 133.8, 133.0, 129.7, 129.4, 125.3, 125.2, 124.04, 123.3, 118.0, 93.2. MS (TOF ES⁺): m/z (%) = 333 (M+H⁺, 100%), 291 (35%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₀³⁷ClN₂O₃S: 333.0101; found: 333.0108.

3-Phenyl-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4,4dioxide (3f)

White solid; 136 mg, 46% (conventional), 152 mg, 51% (microwave); mp 238–239 °C; $R_f = 0.45$ (EtOAc 1:1 Hex). IR (KBr): 3217 (NH), 3075 (CH), 1640 (C=N) and 1261–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 13.14$ (br s, 1H, NH), 8.08 – 8.04 (m, 2H), 8.01 (d, *J* = 8.1 Hz, 1H), 7.73 (td, *J* = 7.7, 1.3 Hz, 1H), 7.67 – 7.56 (m, 3H), 7.41 (t, J = 7.8 Hz, 2H). ¹³C NMR (100 MHz, DMSO-*d*₆): $\delta = 162.5$, 158.7, 134.2, 133.5, 131.9, 129.7, 128.5, 126.8, 125.9, 124.5, 123.8, 118.4, 93.7. MS (TOF ES⁺): m/z (%) = 299 (M+H⁺, 100%). HRMS: *m*/*z* [M+H]⁺ calc. for C₁₅H₁₁N₂O₃S: 299.0490; found: 299.0489.

3-(2-Chloro-4,5-dimethoxyphenyl)-9H-benzo[b]isoxazolo [5,4-e][1,4]thiazine 4,4-dioxide (3g)

Light brown solid; 209 mg, 58% (conventional), 238 mg, 67% (microwave); mp 197–199 °C; $R_f = 0.33$ (EtOAc 1:1 Hex). IR (KBr): 3221 (NH), 3090 (CH), 1636 (C=N) and 1261–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 13.14$ (br s, 1H, NH), 7.96 (dd, J = 8.1, 1.2 Hz, 1H), 7.72 (ddd, J = 8.5, 7.2, 1.4 Hz, 1H), 7.41 (s, 1H), 7.41 – 7.36 (m, 2H), 7.24 (s, 1H), 3.87 (s, 3H), 3.79 (s, 3H). ¹³C NMR (100 MHz, DMSO-*d*₆): $\delta = 161.5, 155.9, 150.9, 147.5, 133.6, 133.1, 125.6, 123.9, 123.7, 123.1, 118.0, 116.4, 114.1, 113.5, 94.5, 56.1 55.8. MS (TOF ES⁺): m/z (%) = 393 (M+H⁺, 100%). HRMS: <math>m/z$ [M+H]⁺ calc. for $C_{17}H_{14}^{37}CIN_2O_5S$: 393.0312; found: 393.0315.

3-(2-Nitrophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4,4-dioxide (3h) Light brown solid, 170 mg, 50% (conventional), 236 mg, 69% (microwave); mp 253–256 °C; $R_f = 0.33$ (EtOAc 1:1 Hex). IR (KBr): 3221 (NH), 3082 (CH), 1640 (C=N) and 1273–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO- d_6): $\delta = 13.26$ (br s, 1H, NH), 8.28 (dd, J = 8.4, 1.2 Hz, 1H), 8.01 – 7.96 (m, 1H), 7.95 – 7.88 (m, 3H), 7.73 (ddd, J = 8.5, 7.2, 1.4 Hz, 1H), 7.44 – 7.41 (m, 1H), 7.39 (dd, J = 8.1, 1.0 Hz, 1H). ¹³C NMR (100 MHz, DMSO- d_6): $\delta = 161.7$, 155.6, 147.8, 134.2, 133.7, 133.2, 132.6, 131.8, 125.5, 125.2, 124.1, 122.9, 120.4, 118.2, 94.0. MS (TOF ES⁺): m/z (%) = 344 (M+H⁺, 100%), 274 (25%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₀N₃O₅S: 344.0341; found: 344.0344.

3-(4-Nitrophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4,4-dioxide (3i)

Light brown solid; 184 mg, 54% (conventional), 134 mg, 68% (microwave); mp 285–287 °C; $R_f = 0.34$ (EtOAc 1:1 Hex). IR (KBr): 3221 (NH), 3093 (CH), 1628 (C=N) and 1342–1142 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO- d_6): $\delta = 13.18$ (br s, 1H, NH), 8.44 (d, J = 8.9 Hz, 2H), 8.29 (d, J = 8.9 Hz, 2H), 8.02 (dd, J = 8.4, 1.2 Hz, 1H), 7.83 – 7.65 (m, 1H), 7.42 (t, J = 7.4 Hz, 2H). ¹³C NMR (100 MHz, DMSO- d_6): $\delta = 160.0$, 157.1, 149.6, 134.3, 133.5, 132.8, 129.8, 125.8, 124.9, 124.6, 123.7, 118.6, 93.9. MS (TOF ES⁺): m/z (%) = 344 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₀N₃O₅S: 344.0341; found: 344.0324.

3-(5-Chlorothiophen-2-yl)-9H-benzo[b]isoxazolo[5,4-e] [1,4]thiazine 4,4-dioxide (3j)

White solid; 197 mg, 65% (conventional), 243 mg, 72% (microwave); mp 215–217 °C; $R_f = 0.47$ (EtOAc 1:1 Hex). IR (KBr): 3221 (NH), 3082 (CH), 1640 (C=N) and 1265–1138 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 13.24$ (br s, 1H, NH), 8.04 (d, J = 8.2 Hz, 1H), 7.92 – 7.84 (m, 1H), 7.73 (t, J = 8.4 Hz, 1H), 7.43 (d, J = 7.3 Hz, 1H), 7.41 – 7.38 (m, 2H). ¹³C NMR (100 MHz, DMSO-*d*₆): $\delta = 162.5$, 152.5, 134.3, 133.4, 133.3, 132.2, 128.9, 126.0, 125.7, 124.6, 123.8, 118.6, 92.6. MS (TOF ES⁺): m/z (%) = 339 (M+H⁺, 100%), 329 (53%). HRMS: *m*/z [M+H]⁺ calc. for C₁₃H₈N₂O₃S₂³⁷Cl: 338.9665; found: 338.9674.

3-(5-Chlorofuran-2-yl)-9H-benzo[b]isoxazolo[5,4-e][1,4] thiazine 4,4-dioxide (3k)

Brown solid; 150 mg, 52% (conventional), 206 mg, 64% (microwave); mp 233 °C decomp; $R_f = 0.42$ (EtOAc 1:1 Hex). IR (KBr): 3225 (NH), 3078 (CH), 1636 (C=N) and 1282–1173 (SO₂) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 13.20$ (br s, 1H, NH), 8.03 (dd, J = 8.1, 1.3 Hz, 1H), 7.73 (ddd, J = 8.5, 7.3, 1.4 Hz, 1H), 7.44 – 7.41 (m, 2H), 7.41 – 7.38 (m, 1H), 6.86 (d, J = 3.7 Hz, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆): $\delta = 161.7$, 147.9, 140.0, 138.9, 133.7, 132.9, 125.3, 124.1, 123.2, 118.1, 118.0, 109.8, 91.3. MS (TOF ES⁺): m/z (%) = 323 (M+H⁺, 100%). HRMS: *m*/*z* [M+H]⁺ calc. for C₁₃H₈³⁷ClN₂O₃S₂: 322.9893; found: 322.9899.

General Synthesis of 3-Aryl-4-((2-fluorophenyl)sulfinyl) isoxasazol-5-amine

2-((2-Fluorophenyl)sulfinyl)acetonitrile (1.0 eq, 183.2 mg, 2.0 mmol) and any α -chloro oxime (1.0 eq, 2.0 mmol) were dissolved in MeOH. NaOH (3.0 eq, 120.0 mg, 6 mmol) was added and the reaction mixture was refluxed 2 h. Progress of the reaction was monitored by TLC. The solvent was

evaporated in *vacuo*. Column chromatography was applied for purification. (EtOAc 1:2 Hexane)

3-(4-Chlorophenyl)-4-((2-fluorophenyl)sulfinyl)isoxazol-5amine (5e)

White solid; 247 mg, 37%; mp 157–158 °C; $R_f = 0.40$ (EtOAc 1:1 Hex). IR (KBr): 3383–3178 (NH₂), 1643 (C=N) and 1069 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.15$ (s, 2H, NH₂), 7.66 – 7.55 (m, 3H), 7.46 – 7.34 (m, 3H), 7.19 – 7.10 (m, 2H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 171.0$, 161.0, 156.6 (d, ¹ $J_{CF} = 246.0$ Hz), 134.9, 132.6 (d, ³ $J_{C-F} = 7.9$ Hz), 130.2, 129.4, 128.95 (d, ² $J_{C-F} = 14.8$ Hz), 127.0, 126.4, 124.2 (d, ³ $J_{C-F} = 3.2$ Hz), 115.8 (d, ² $J_{C-F} = 19.9$ Hz), 90.7. MS (TOF ES⁺): m/z (%) = 337 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₁³⁷ClFN₂O₂S: 337.0214; found: 337.0215.

4-((2-Fluorophenyl)sulfinyl)-3-phenylisoxazol-5-amine (5f)

White solid; 232 mg, 38%; mp 164–166 °C; $R_f = 0.37$ (EtOAc 1:1 Hex). IR (KBr): 3383–3317 (NH₂), 1643 (C=N) and 1069 (SO) cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.77$ (ddd, J = 8.9, 7.4, 1.5 Hz, 1H), 7.67 – 7.61 (m, 2H), 7.41 – 7.32 (m, 4H), 7.23 (ddd, J = 8.7, 7.6, 1.1 Hz, 1H), 6.98 (ddd, J = 9.5, 8.2, 1.0 Hz, 1H), 5.57 (s, 2H, NH₂). ¹³C NMR (75 MHz, CDCl₃): $\delta = 169.1$, 161.2, 158.2 (d, ¹J_{C-F} = 250.4 Hz), 133.3 (d, ³J_{C-F} = 7.8 Hz), 130.4, 130.1 (d, ²J_{C-F} = 15.5 Hz), 128.7, 128.3, 127.6, 126.5 (d, ⁴J_{C-F} = 1.4 Hz), 124.7 (d, ³J_{C-F} = 3.5 Hz), 116.5 (d, ²J_{C-F} = 20.2 Hz), 90.6 MS (TOF ES⁺): m/z (%) = 303 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₂FN₂O₂S: 303.0604; found: 303.0618.

3-(2-Chloro-4,5-dimethoxyphenyl)-4-((2-fluorophenyl) sulfinyl)isoxazol-5-amine (5g)

Yellow solid; 401 mg, 50%; mp 118–120 °C; $R_f = 0.17$ (EtOAc 1:1 Hex). IR (KBr): 3314–3171 (NH₂), 1640 (C=N) and 1011 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.08$ (s, 2H, NH₂), 7.45 – 7.35 (m, 1H), 7.31 (td, J = 7.6, 1.7 Hz, 1H), 7.23 – 7.15 (m, 1H), 7.07 (td, J = 7.7, 1.0 Hz, 1H), 6.88 (s, 1H), 6.75 (s, 1H), 3.78 (s, 3H), 3.71 (s, 3H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 170.1, 159.5, 156.7$ (d, J = 246.0 Hz), 150.2, 148.1, 147.0, 132.3 (d, J = 7.8 Hz), 129.1 (d, J = 15.1 Hz), 125.6, 124.1 (d, J = 3.2 Hz), 123.7, 118.0, 115.8 (d, J = 19.9 Hz), 113.4 (d, J = 124.8 Hz), 91.9, 56.0, 55.6. MS (TOF ES⁺): m/z (%) = 397 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₇H₁₅³⁷CIFN₂O₄S: 397.0425; found: 397.0437.

4-((2-Fluorophenyl)sulfinyl)-3-(2-nitrophenyl)isoxazol-5amine (5h)

Yellow solid; 319 mg, 46%; mp 156–157 °C; $R_f = 0.17$ (EtOAc 1:1 Hex). IR (KBr): 3310–3178 (NH₂), 1647 (C=N) and 1007 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.29$ (s, 2H, NH₂), 8.00 – 7.94 (m, 1H), 7.68 – 7.63 (m, 2H), 7.38 – 7.27 (m, 2H), 7.18 – 7.10 (m, 1H), 7.02 (td, J = 7.6, 1.8 Hz, 1H), 6.88 (td, J = 7.7, 1.0 Hz, 1H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 170.2$, 159.3, 156.8 (d, ${}^{1}J_{C-F} = 246.5$ Hz), 147.4, 133.3, 132.3 (d, ${}^{3}J_{C-F} = 7.8$ Hz), 132.1, 131.3, 128.8 (d, ${}^{2'}J_{C-F} = 15.1$ Hz), 123.9 (d, ${}^{3'}J_{C-F} = 3.2$ Hz), 125.1, 124.3, 122.1, 115.8 (d, ${}^{2}J_{C-F} = 19.8$ Hz), 91.6. MS (TOF ES⁺): m/z (%) = 348 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₁FN₃O₄S: 348.0454; found: 348.0454.

General Synthesis of 3-Aryl-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4-oxide (Method A)

2-((2-fluorophenyl)sulfinyl)acetonitrile (1.0 eq., 183 mg, 1.0 mmol,) and α -chloro oxime (1.0 eq., 1.0 mmol) were dissolved in DMSO (5 mL) and Cs₂CO₃ (2.0 eq., 652 mg, 2.0 mmol) was added, and the mixture stirred at reflux for 2 h. The progress of the reaction was monitored by TLC. The mixture was extracted with EtOAc (3 x 15 mL), subsequently washed with brine (15 mL) and dried (Na₂SO₄). The solvent was removed under reduced pressure and the resulting solid was recrystallized from benzene.

General Synthesis of 3-Aryl-9H-benzo[b]isoxazolo[5,4e][1,4]thiazine 4-oxide (Method B)

3-Aryl-4-((2-fluorophenyl)sulfinyl)isoxazol-5-amine (1.0 mmol) was dissolved in DMSO (5 mL) and Cs_2CO_3 (326 mg, 1.0 mmol) was added, and the mixture stirred at reflux for 1 h. The progress of the reaction was monitored by TLC. Then, the mixture was extracted with EtOAc (3 x 15 mL), subsequently washed with brine and dried on Na₂SO₄. The solvent was evaporated under reduced pressure and the resultant solid was recrystallized from benzene.

3-(2,6-Dichlorophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4] thiazine 4-oxide (6a)

Brown solid; 60 mg, 17% (method A); mp 209 °C decomp; R_f = 0.31 (EtOAc 2:1 Hex). IR (KBr): 3433 (NH), 1636 (C=N) and 995 (SO) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ = 13.10 (br s, 1H, NH), 7.96 (dd, *J* = 7.9, 1.3 Hz, 1H), 7.74 – 7.73 (m, 1H), 7.72 – 7.65 (m, 3H), 7.47 (dd, *J* = 8.3, 0.9 Hz, 1H), 7.36 (t, *J* = 8.1 Hz, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ = 160.9, 157.6, 134.8, 134.4, 133.4, 133.3, 132.9, 132.3, 128.9, 128.6, 124.9, 124.3, 124.1, 118.2, 94.7. MS (TOF ES⁺): m/z (%) = 351 (M+H⁺, 100%). HRMS: *m/z* [M+H]⁺ calc. for C₁₅H₉³⁷Cl₂N₂O₂S: 350.9762; found: 350.9762.

3-(2,4-Dichlorophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4] thiazine 4-oxide (6b)

Brown solid; 67 mg, 19% (method A), 312 mg, 89% (method B); mp 139–141 °C; $R_f = 0.31$ (EtOAc 2:1 Hex). IR (KBr): 3433 (NH), 1632 (C=N) and 957 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 13.14$ (br s, 1H, NH), 8.01 (dd, J = 7.9, 1.3 Hz, 1H), 7.97 (d, J = 1.9 Hz, 1H), 7.79 (d, J = 8.3 Hz, 1H), 7.76 – 7.68 (m, 2H), 7.49 (d, J = 7.5 Hz, 1H), 7.39 (t, J = 7.5 Hz, 1H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 160.9$, 158.6, 136.5, 133.4, 133.2, 132.9, 132.8, 132.3, 130.2, 128.2, 124.8, 124.1, 124.0, 118.1, 95.0. MS (TOF ES⁺): m/z (%) = 351 (M+H⁺, 100%), 203 (22%), 143 (17%). HRMS: *m*/z [M+H]⁺ calc. for C₁₅H₉³⁷Cl₂N₂O₂S: 350.9762; found: 350.9760.

3-(4-Fluorophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4] thiazine 4-oxide (6c)

Brown solid; 55 mg, 18% (method A), mp 165–166 °C; $R_f = 0.32$ (EtOAc 2:1 Hex). IR (KBr): 3433 (NH), 1636 (C=N) and 995 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 13.02$ (br s, 1H, NH), 8.07 – 8.03 (m, 2H), 8.02 (d, J = 1.7 Hz, 1H), 7.71 (ddd, J = 8.5, 7.2, 1.4 Hz, 1H), 7.51 (t, J = 8.9 Hz, 2H), 7.46 (dd, J = 8.3, 0.9 Hz, 1H), 7.39 (t, J = 8.1 Hz, 1H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 163.7$ (d, ¹ $J_{C-F} = 249.0$ Hz), 162.0, 161.2, 159.4, 133.2, 132.9, 132.2, 130.1 (d,

 ${}^{3}J_{C-F} = 8.9$ Hz), 123.9, 123.7 (d, ${}^{4}J_{C-F} = 2.2$ Hz), 117.9, 116.7 (d, ${}^{2}J_{C-F} = 22.1$ Hz), 93.0. MS (TOF ES⁺): m/z (%) = 301 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₀FN₂O₂S: 301.0447; found: 301.0144.

3-(4-Bromophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4] thiazine 4-oxide (6d)

Light brown solid; 70 mg, 19% (method A); mp 175–176 °C; $R_f = 0.34$ (EtOAc 2:1 Hex). IR (KBr): 3445 (NH), 1628 (C=N) and 991 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆): $\delta = 13.05$ (br s, 1H, NH), 8.04 (dd, J = 7.9, 1.3 Hz, 1H), 7.93 (d, J = 8.7 Hz, 2H), 7.87 (d, J = 8.7 Hz, 2H), 7.71 (ddd, J =8.4, 7.4, 1.4 Hz, 1H), 7.46 (d, J = 7.5 Hz, 1H), 7.39 (t, J =7.5 Hz, 1H). ¹³C NMR (75 MHz, DMSO-*d*₆): $\delta = 161.3$, 159.4, 133.2, 133.0, 132.6, 132.2, 129.5, 126.3, 124.8, 124.0, 123.6, 118.0, 93.0. MS (TOF ES⁺): m/z (%) = 361 (M+H⁺, 98%), 363 (100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₀⁷⁹BrN₂O₂S: 360.9646; found: 360.9646.

3-(4-Chlorophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4] thiazine 4-oxide (6e)

Brown solid; 50 mg, 16% (method A), 257 mg 81% (method B), mp 165–167 °C; $R_f = 0.31$ (EtOAc 2:1 Hex). IR (KBr): 3433 (NH), 1632 (C=N) and 991 (SO) cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ = 12.97 (br s, 1H, NH), 8.01 (dd, *J* = 7.9, 1.1 Hz, 1H), 7.98 (d, *J* = 8.5 Hz, 2H), 7.71 (d, *J* = 8.5 Hz, 2H), 7.68 (dd, *J* = 8.4, 1.2 Hz, 1H), 7.44 (d, *J* = 7.9 Hz, 1H), 7.37 (t, *J* = 7.5 Hz, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ = 161.8, 159.8, 136.5, 133.7, 133.4, 132.7, 130.1, 129.8, 126.5, 124.4, 124.1, 118.5, 93.5.MS (TOF ES⁺): m/z (%) = 317 (M+H⁺, 100%), 271 (40%), 260 (32%). HRMS: *m*/z [M+H]⁺ calc. for C₁₅H₁₀³⁷ClN₂O₂S: 317.0152; found: 317.0156.

3-Phenyl-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4-oxide (6f)

White solid; 220 mg, 78% (method B); mp 144–145 °C; $R_f = 0.33$ (EtOAc 2:1 Hex). IR (KBr): 3445 (NH), 1632 (C=N) and 991 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 13.05$ (br s, 1H, NH), 8.08 (dd, J = 7.9, 1.3 Hz, 1H), 8.06 – 8.03 (m, 1H), 8.02 (d, J = 2.5 Hz, 1H), 7.75 (ddd, J = 8.5, 7.2, 1.5 Hz, 1H), 7.72 – 7.65 (m, 3H), 7.50 (dd, J = 8.3, 0.9 Hz, 1H), 7.43 (t, J = 8.1 Hz, 1H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 161.2$, 160.2, 133.3, 132.9, 132.2, 131.2, 129.5, 127.6, 127.2, 123.9, 123.7, 118.0, 93.1. MS (TOF ES⁺): m/z (%) = 283 (M+H⁺, 100%). HRMS: m/z [M+H]⁺ calc. for C₁₅H₁₁N₂O₂S: 283.0541; found: 283.0549.

3-(2-Chloro-4,5-dimethoxyphenyl)-9H-benzo[b]isoxazolo [5,4-e][1,4]thiazine 4-oxide (6g)

Light yellow solid; 323 mg, 81% (method B); mp 164–166 °C; $R_f = 0.16$ (EtOAc 2:1 Hex). IR (KBr): 3445 (NH), 1624 (C=N) and 984 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO- d_6): $\delta = 13.07$ (br s, 1H, NH), 8.06 (dd, J = 7.9, 1.3 Hz, 1H), 7.76 (ddd, J = 8.5, 7.3, 1.4 Hz, 1H), 7.52 (dd, J = 8.2, 0.8 Hz, 1H), 7.46 – 7.39 (m, 1H), 7.36 (s, 1H), 7.34 (s, 1H), 3.94 (s, 3H), 3.87 (s, 3H). ¹³C NMR (75 MHz, DMSO- d_6): $\delta = 160.7$, 159.3, 151.0, 147.9, 133.3, 132.9, 132.3, 124.0, 123.9, 123.7, 118.0, 117.1, 113.9, 113.6, 95.0, 56.2, 56.0. MS (TOF ES⁺): m/z (%) = 377 (M+H⁺, 100%), 290 (10%), 143 (13%). HRMS: m/z [M+H]⁺ calc. for $C_{17}H_{14}^{37}CIN_2O_4S$: 377.0363; found: 377.0367.

3-(2-Nitrophenyl)-9H-benzo[b]isoxazolo[5,4-e][1,4]thiazine 4-oxide (6h)

Yellow solid; 272 mg, 83% (method B); mp 191–193 °C; R_f = 0.14 (EtOAc 2:1 Hex). IR (KBr): 3445 (NH), 1632 (C=N) and 999 (SO) cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₀): δ = 13.21 (br s, 1H, NH), 8.32 (dd, *J* = 8.2, 1.1 Hz, 1H), 8.10 – 8.03 (m, 2H), 8.02 – 7.95 (m, 2H), 7.78 (ddd, *J* = 8.5, 7.3, 1.4 Hz, 1H), 7.54 (dd, *J* = 8.3, 0.8 Hz, 1H), 7.45 (t, *J* = 8.1 Hz, 1H). ¹³C NMR (75 MHz, DMSO-*d*₀): δ = 160.9, 158.5, 148.5, 134.1, 133.2, 133.1, 132.5, 132.4, 131.8, 125.3, 124.2, 123.9, 120.6, 118.2, 94.4. MS (TOF ES⁺): m/z (%) = 328 (M+H⁺, 100%). HRMS: *m*/z [M+H]⁺ calc. for C₁₅H₁₀N₃O₄S: 328.0392; found: 328.0399.

Acknowledgment

We are grateful to TUBITAK (Project No:114Z181) for financial support.

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Supplementary Material

Copies of ¹H and ¹³C NMR spectra of all new compounds.