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Synthesis of Pyrazolo[4',3':5,6]pyrano[2,3-*d*]pyrimidine Derivatives for Antiviral Evaluation

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6-Amino-3-methyl-4-(4-nitrophenyl)-2,4-dihydropyrano[2,3-c]pyrazole-5-carbonitrile (1) was used as a precursor for preparation of some novel 3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidine derivatives 3-6, and some of their corresponding N^2 - and C^5 -S-acyclic nucleosides 7 and 8. Furthermore, the preparation of 5-amino-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-1H-pyrazole derivatives 10-16were described. Some of the prepared products were selected and tested for antiviral activity against Herpes Simplex Virus type-1 (HSV-1).

Keywords: Acyclic nucleosides / Antiviral activities / Pyranopyrazoles / Pyrazolopyranopyrimidines

Received: January 11, 2007; accepted: February 14, 2007

DOI 10.1002/ardp.200700005

Introduction

The chemistry of pyrazole derivatives, especially 3methyl-1-phenyl-2-pyrazolin-5-one, has received great interest, since this nucleus is the parent skeleton of pyrine drugs and other medicinal compounds [1-2]. Pyrano[2,3-c]pyrazole derivatives showed biological activities as bactericides [3] and virucides [4]. Moreover, a large number of fused pyrimidine derivatives exhibited antimycobacterial [5], antitumor [6], and antiviral activities [7]. In continuation of our previous work on pyrazoles [8–10], pyrano[2,3-c]pyrazoles [11], pyrimidines [12], and fused pyrimidines [9, 13–14], we intended to synthesize a pyrano[2,3-c]pyrazole ring system fused to a pyrimidine moiety to obtain new compounds which are expected to possess notable chemical and biological activities.

Results and discussion

Chemistry

This work deals with synthesis and study of the reactions of fused pyrano[2,3-c]pyrazoles from their corresponding

β-enaminonitrile precursor: 6-amino-2,4-dihydro-3methyl-4-(4-nitrophenyl)pyrano[2,3-c]pyrazole-5-carbonitrile 1 [11, 15] as key compound for this study and for further syntheses of other fused heterocyclic compounds. Thus, when compound 1 was stirred at room temperature with acetic anhydride it gave 6-acetamido-3-methyl-4-(4-nitrophenyl)-2,4-dihydropyrano[2,3-c]pyrazole-5-carbonitrile 2. The structure of the latter compound was confirmed on the basis of its elemental analysis and spectral data (see Experimental and Scheme 1). The IR spectrum of compound 2 showed absorption bands assignable to the NH groups, CN-, and CO group. Its ¹H-NMR spectrum showed signals at 1.77 (s, 3H, C^3 -CH₃), 2.24 (s, 3H, O=C-CH₃), 11.91, and 12.63 for $2 \times NH$ (D₂O exchangeable) and MS gave the molecular ion peak at m/z (%): 339 $[M^+]$ (42.72). While carrying out the same reaction under reflux, acetylation took place to give 2-acetyl-3,7dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5(6H)-one **3**.

The structure of the aforementioned compound was confirmed by spectral data (see Experimental and Scheme 1). The ¹H-NMR spectrum showed a signal at 2.26 (s, 3H, C³-CH₃). Actually, by surveying the alkylation and acylation reactions of pyrazole derivatives, we can conclude that N^2 -acetylation which took place for compound **3**, is in accordance with results which judged that the ¹H-NMR chemical shifts of C³-methyl group adjacent to N^2 -

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Scheme 1. Synthesis route of compounds 1-8.

alkyl group are more deshielded and appear invariably at a lower field ($\sim \delta 2.25-2.30$ ppm), when compared with that of the C³-methyl protons of N¹-alkyl derivatives ($\sim \delta$ 1.9–2.0 ppm) [16, 17]. Furthermore, the ¹H-NMR data given by many authors for alkylation, acylation [11, 13, 16, 18–20] and nucleosidation reactions [21], the single crystal X-ray analysis proved no doubt that acetylation took place at the N²-nitrogen of the pyrazole ring (Fig. 1, Table 1). It was reported by Arakawa *et al.* [18] and Kashima [19] that formation of acyl pyrazoles is governed by the steric interaction between the ring substitution groups and the acyl moiety which isomerizes slowly from the N^2 -position into the more thermodynamically stable N^1 -acetyl product. However, when we attempted isomerization of product **3** into its corresponding N^1 -acetyl product under very mild conditions, only deacetylation started to take place with formation of 3,7-dimethyl-4-(4-nitrophenyl)-2,4-

Crystal system	Monoclinic	Torsion angles (°)	
Space group	P-1	C15-N4-N9-C22	-0.3 (7)
a (Å)	24.6358 (14)	N9-N4-C15-C24	-0.3 (6)
b (Å)	7.2915 (4)	N9-N4-C15-C29	-177.5 (11)
c (Å)	21.7147 (14))	C18-N4-N9-C22	176.2 (9)
α (°)	90.00°	N9-N4-C18-O23	-171.9 (11)
β(°)	$93.600(3)^{\circ}$	N9-N4-C18-C28	7.3 (7)
γ (°)	90.00°	C15-N4-C18-O23	3.9 (8)
V (Å ³)	3893.0 (4)Å ³	C18-N4-C15-C24	-176.3 (11)
Z	8	C15-N4-C18-C28	-176.9 (12)
D _x	1.452 Mg m^{-3}	C18-N4-C15-C29	6.4(7)
R	0.043	C24-C15-N4-N9	-0.3 (6)
wR	0.076	C24-C15-N4-C18	-176.3 (11)
Bond length (Å)		C29-C15-N4-C18	6.4(7)
N4-N9	1.386 (2)	O23-C18-N4-N9	-171.9(11)
N4-C15	1.385 (2)	O23-C18-N4-C15	3.9 (8)
N4-C18	1.417(2)	C28-C18-N4-N9	7.3 (7)
Bond angles (°)		C28-C18-N4-C15	-176.9 (12)
N9-N4-C15	112.42 (12)		
N9-N4-C18	118.6 (2)		
C15-N4-C18	128.9 (2)		
N4-N9-C22	100.69 (12)		
N4-C18-O23	119.6 (2)		
N4-C18-C28	115.7 (2)		
O23-C18-C28	124.7 (2)		

Table 1. Crystal data, bond lengths, bond angles, and torsion angles of compound 3.



Figure 1. Single crystal X-ray structure of compound **3** and dioxane solvent.

dihydropyrazolo[4',3':5,6]pyrano[2,3-*d*]pyrimidin-5(6*H*)one **4** (see Experimental and Scheme 1). Compound **4** was converted into its corresponding 5-chloro derivative **5** (Scheme 1) by refluxing with phosphorus oxychloride. The mass spectra of compound **5** gave fragments showing the isotopic pattern due to the presence of chlorine atom (see Experimental). Compound **5** was converted into its corresponding 5-thiono derivative **6** (Scheme 1) following the procedure of Schiba *et al.* [22] and its ¹³C-NMR spectrum showed a signal at d 183 ppm accountable for the C=S group (see Experimental).

Since the synthesis of acyclovir, as one of the potent antiherpetic drug by Schaffer et al. [23], many attempts have been directed by nucleoside chemists to prepare a lot of related compounds with various side chains and glycons [24, 25]. However in this connection, there is only one recent report in the literature [21] on the N¹-pyrazole acyclic nucleosides of pyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidine derivatives which were tested for their antiinflammatory activity. The N2-acyclic nucleosides and Sacyclic nucleosides of pyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidine derivatives are still not known (to the best of our knowledge) in the literature. Thus, in continuation of our previous work [9, 26] in preparing various cyclic and acyclic nucleosides of different heterocyclic compounds, in this report, we describe the synthesis of some N²- and Sacyclic nucleosides related to pyrazolo[4',3':5,6]-pyrano[2,3-d]pyrimidine derivatives by treating the sodium salt of compounds 4 and 6 (generated in situ, see Experimental) with 2-chloroethyl methyl ether; they afforded the corresponding N^2 - and C^5 -S-acyclic nucleosides: 3,7dimethyl-2-(2-methoxyethyl)-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-ol 7 and 5-(2-methoxy-ethylsulfanyl)-3,7-dimethyl-4-(4-nitrophenyl)-1,4-dihydropyrazolo[4',3':5,6]pyrano[3,2-d]pyrimidine 8, respectively (Scheme 1). The structural assignments and sites of nucleosidation of compounds 7 and 8 were assigned on the basis of their elemental and spectral data (see Experimental and Scheme 1). The ¹H-NMR spectrum



of compound **7** revealed the absence of signals at δ 12.18, 12.40 ppm of the NH of the pyrazole and the pyrimidine ring, however a signal at δ 6.79 ppm assignable to the OH of the pyrimidine moiety due to the tautomerization. The assignment of the N²- nucleosidation was judged by the appearance of the C³-CH₃ protons at δ 2.25 ppm, and this is in accordance with reported results for N²-alkylation of related compounds [16]. In compound **8**, the pre-

sence of NH at δ 11.30 ppm (exchangeable with D₂O) in ¹H-NMR spectrum and the absence of the C=S group in its ¹³C-NMR spectrum proved that nucleosidation took place at the C⁵-S-pyrimidine. Deprotection of nucleosides **7** and **8** (using ammonium hydroxide solution/ethanol) afforded again compounds **4** and **6**, respectively.

When compound **5** was heated with hydrazine hydrate, it gave a compound assigned to the structure of

3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo-

[4',3':5,6]pyrano[2,3-d]pyrimidin-5-ylhydrazine 9 (Experimental, Scheme 2). When the compound 9 was refluxed with ethoxymethylenemalononitrile, methylethoxymethylenemalononitril, bis-(methylthio)methylenemalononitrile, tetracyanoethylene or ethyl(ethoxy-methylene)cyano-acetate, it afforded the corresponding substiderivatives 10-14, tuted pyrazole respectively (Scheme 2). The structures of the latter compounds were confirmed on the basis of their elemental analysis and spectral data (Experimental). The IR spectra of compounds 10-13 showed absorption bands characteristic for NH₂ and CN groups, while compound 14 revealed absorption bands characteristic for NH₂ and C=O. Also, ¹H-NMR spectra showed signals at δ = 6.79, 6.75, 6.85, 6.55, and 6.53 ppm due to NH_2 (exchangeable with D_2O) for compounds 10-14, respectively. The MS gave the molecular ion peaks at m/z (%) = 429 (23.49), 443 (20.85), 475 (3.25), 454 (18.30), and 476 (41.88) for compounds 10-14, respectively.

Similarly, when a mixture of compound **9** and acetylacetone in ethanol was refluxed, it gave 5-(3,5-dimethyl-pyrazol-1-yl)-3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydro-pyrazolo[4',3':5,6]pyrano[2,3-*d*]pyrimidine **15** (Experimental and Scheme 2). The ¹H-NMR spectrum of the latter compound showed signals at $\delta = 1.75$ ppm and 2.25 ppm for (2 × CH₃). The MS, gave the molecular ion peak as a base peak at *m*/*z* (%) = 417 (100).

On the other hand, when a mixture of compound 9 and excess of ethyl acetoacetate was refluxed, it lead to an unexpected product assigned to the structure of 1-[3,7dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-3,4-dimethyl-3a,7a-dihydro-1H-pyrano[2,3-c]pyrazol-6-one 16 (Experimental, Scheme 2). The formation of compound 16 might have taken place via the formation of the pyrazolone intermediate which reacted further with another mole of ethyl acetoacetate. The ¹H-NMR spectrum of compound **16** showed signals at 2.35 ppm, 2.51 ppm for $(2 \times CH_3)$, and 5.69 ppm for (pyranyl-H), and its IR spectrum revealed the presence of a C=O group. The MS gave the molecular ion peak as a base peak at m/z (%) = 485 (100).

Antiviral Screening

The plaque infectivity assay was carried out to test compounds **3**, **4**, **6**, **8**, **10**, **11**, **12**, **13**, and **16** for antiviral activity. The test was performed to include the three possibilities for antiviral activity – viricidal effect, virus adsorption, and effect on virus replication for HSV-1. The results are summarized in Fig. 2. Compound **12** revealed a toxic effect on viral growth. On the other hand, compounds **3**, **4**, **6**, and **8** revealed the highest anti-HSV-1 activity in com-



Figure 2. Effect of novel derivatives on HSV-1.

parison with acyclovir as control. At both concentration, all tested compounds revealed no higher antiviral activity than acyclovir. In general, compound **4** showed the highest effect on HSV-1 than the other tested compounds, where its antiviral activity increased from 63% at concentration of $20 \ \mu g/10^5$ cells to 95% at concentration of $40 \ \mu g/10^5$ cells.

Conclusion

Some novel pyrazolo[4',3':5,6]pyrano[2,3-*d*]pyrimidine derivatives 3-6, and some of their corresponding N^2 - and C^5 -S-acyclic nucleosides 7 and 8 were prepared. Furthermore, some pyrazole derivatives 10-16 were described. Structure-activity correlation of the obtained results revealed that the pyrazolopyranopyrimidine ring, in compounds 3, 4, 6, and 8 showed promising antiviral activity. When the acetyl group of compound 3 was removed, it was noticed that the new pyrazolopyranopyrimidine derivative 4 was found to be more active as antiviral agent for HSV-1 (due to the two free NH groups). While replacement of oxygen (in compound 4) with sulfur (in compound 6), decreased the activity. Addition of another ring e.g., a pyrazole ring in compounds 10, 11, 12, 13, and 16 decreased the activity% of HSV-1 reduction.

We are thankful to Dr. M. A. Ali, Virology Laboratory, National Research Centre for antiviral evaluation.

Experimental

Chemistry

All melting points are measured using Electro-thermal IA 9100 apparatus, (Shimadzu, Japan). IR spectra were recorded as potassium bromide pellets on a Perkin-Elmer 1650 spectrophotometer (Perkin Elmer, USA), National Research Centre, Cairo, Egypt. ¹H-NMR and ¹³C-NMR spectra were determined on a Jeol-Ex-300 NMR spectrometer (JEOL, Tokyo, Japan) and chemical shifts were expressed as part per million; ppm (d values) against TMS as internal reference (Faculty of Science, Cairo University, Cairo, Egypt). Mass spectra were recorded on EI + Q1 MSLMR UPLR, National Research Centre, Cairo, Egypt. Microanalyses were operated using Mario El Mentar apparatus, Organic Microanalysis Unit, National Research Centre, Cairo, Egypt and the results were within the accepted range (± 0.40) of the calculated values. Column chromatography was performed on Silica gel 60 (particle size 0.06 – 0.20 mm; Merck, Darmstadt, Germany).

Compound **1** was prepared according to a reported method [11].

Synthesis of compounds 2, 3, 4, 5, and 6

6-Acetamido-3-methyl-4-(4-nitrophenyl)-2,4dihydropyrano[2,3-c]pyrazole-5-carbonitrile **2**

A solution of compound **1** (2.97 g, 1 mmol) in acetic anhydride (10 mL) was stirred at room temperature for 10 h. The formed precipitate was filtered off, dried, and recrystallized from ethanol to give compound **2**. Yield 89%, m.p. $240 - 243^{\circ}$ C. IR (KBr, cm⁻¹) 3315 (NH), 3201 (NH), 2200 (CN) and 1735 (CO). ¹H-NMR (DMSO-d₆) δ 1.77 (s, 3H, C³-CH₃), 2.24 (s, 3H, CO-CH₃), 4.98 (s, 1H, pyran), 7.58 (d, *J* = 12 Hz, 2H, Ar-H), 8.23 (d, *J* = 13.08 Hz, 2H, Ar-H), 11.91 (s, 1H, NH, D₂O exchangeable), and 12.63 (s, 1H, NH, D₂O exchangeable). MS *m/z* (%): 339 [M⁺] (42.72). Anal. calcd. for C₁₆H₁₃N₅O₄: C 56.64, H 3.86, N 20.64. Found: C 56.53, H 3.90, N 20.71.

2-Acetyl-3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5(6H)-one **3**

A solution of compound **1** (2.97 g, 1 mmol) in acetic anhydride (10 mL) was refluxed for 3 h. The reaction mixture was evaporated till dryness and the remaining solid was recrystallized from dioxane to give compound **3**. Yield 50%, m.p. $356-358^{\circ}$ C. IR (KBr, cm⁻¹) 3440 (NH), 1735 (CO) and 1650 (CO). ¹H-NMR (DMSO-d₆) δ 2.26 (s, 3H, C³-CH₃), 2.31 (s, 3H, C⁷-CH₃), 2.56 (s, 3H, COCH₃), 5.28 (s, 1H, pyran), 7.56 (d, *J* = 8.7 Hz, 2H, Ar-H), 8.14 (d, *J* = 8.7 Hz, 2H, Ar-H) and 12.65 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ 13.0 (C3-CH₃), 21.0 (C7-CH₃), 23.1 (CH₃-acetyl), 33.3 (C-4), 99 (C-4a), 106.6 (C-3a), 123, 129, 140, 146 (Ar-C), 150 (C-3), 155 (C-9a), 159.6 (C-8a), 161 (C-7), 162.4 (C-5), 171 (CO). MS *m/z* (%): 381 [M⁺] (25.41). Anal. calcd. for C₁₈H₁₅N₅O₅: C 56.69, H 3.96, N 18.36. Found: C 56.73, H 4.01, N 18.27.

3,7-Dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo-[4',3':5,6]pyrano[2,3-d]pyrimidin-5(6H)-one **4**

A mixture of compound **3** (3.81 g, 1 mmol) and ammonium hydroxide solution (1 mL, 25%) in ethanol (30 mL) was stirred at room temperature for 3 h. The formed precipitate was filtered off, washed with water several times, and recrystallized from dioxane to give compound **4**. Yield 89%, m.p. $390 - 392^{\circ}$ C IR (KBr, cm⁻¹) 3164 (NH), 3200 (NH) and 1653 (CO). ¹H-NMR (DMSO-d₆) δ 1.90 (s, 3H, C³-CH₃), 2.28 (s, 3H, C⁷-CH₃), 5.14 (s, 1H, pyran), 7.46 (d, *J* = 9.0 Hz, 2H, Ar-H), 8.10 (d, *J* = 8.7 Hz, 2H, Ar-H), 12.18 (s, 1H, NH, D₂O exchangeable) and 12.40 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ 9.8 (C3-CH₃), 20.9 (C7-CH₃), 34 (C-4), 98 (C-4a), 99.5 (C-3a), 123, 129, 136, 145 (Ar-C), 152 (C-3), 155 (C-9a), 158 (C-8a), 162 (C-7), 162.8 (C5). MS *m/z* (%): 339 [M⁺] (18.10). Anal. calcd. for C₁₆H₁₃N₅O₄: C 56.64, H 3.86, N 20.64. Found: C 56.51, H 3.90, N 20.73.

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5-Chloro-3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidine **5**

A mixture of compound **4** (0.339 g, 1 mmol) in phosphorus oxychloride (10 mL) was heated for 2 h. The solution was cooled and poured into ice-water, and the formed precipitate was filtered off, dried, and recrystallized from dioxane to give compound **5**. Yield 85%, m.p. $340-342^{\circ}$ C. ¹H-NMR (DMSO-d₆) δ 2.00 (s, 3H, C³-CH₃), 2.49 (s, 3H, C⁷-CH₃), 5.59 (s, 1H, pyran), 7.52 (d, *J* = 8.7 Hz, 2H, Ar-H), 8.13 (d, *J* = 8.7 Hz, 2H, Ar-H) and 12.29 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 359 [M⁺] Cl³⁷, (3.66), 357 [M⁺] Cl³⁵, (9.81). Anal. calcd. for C₁₆H₁₂ClN₅O₃: C 53.72, H 3.38, Cl 9.91, N 19.58. Found: C 53.85, H 3.31, Cl 10.02, N 19.41.

3,7-Dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo-[4',3':5,6]pyrano[2,3-d]pyrimidine-5(6H)-thione **6**

A solution of compound **5** (3.57 g, 1 mmol) and thiourea (1.52 g, 2 mmol) in ethanol (20 mL) was heated at reflux for 4 h. The formed precipitate was filtered off, dried, and recrystallized from dimethylformamide to give compound **6**. Yield 77%, m.p. $380-382^{\circ}$ C. IR (KBr, cm⁻¹) 3250 (NH), and 3225 (NH). ¹H-NMR (DMSO-d₆) δ 2.01 (s, 3H, C³-CH₃), 2.43 (s, 3H, C⁷-CH₃), 5.47 (s, 1H, pyran), 7.47 (d, *J* = 8.7 Hz, 2H, Ar-H), 8.07 (d, *J* = 8.7 Hz, 2H, Ar-H), 12.16 (s, 1H, NH, D₂O exchangeable) and 13.97 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ 9.8 (C3-CH₃), 20 (C7-CH₃), 37 (C-4), 98 (C-4a), 112 (C-3a), 123, 129, 136, 145 (Ar-C), 151 (C-3), 154 (C-9a), 158 (C-8a), 161 (C-7), 183 (C-5). MS *m/z* (%): 355 [M⁺] (78.38). Anal. calcd. for C₁₆H₁₃N₅O₃S: C 54.08, H 3.69, N 19.71, S 9.02. Found: C 54.00, H 3.75, N 19.63, S 9.12.

Synthesis of compounds 7 and 8; general procedure

A mixture of compound **4** (0.34 g, 0.1 mmol) or **6** (0.36 g, 0.1 mmol), and 50% oil-immersed sodium hydride (0.05 g, 0.2 mmol) in dry dimethylformamide (30 mL) was stirred at 70°C for 1 h. The solution was cooled, and then 2-chloroethylmethyl ether (0.1 mmol) was added and stirred at 90°C for 8 h. The reaction mixtures were evaporated under reduced pressure and chromatographed on silica gel column using chloroform/ methanol mixture (9 : 1) as an eluent to give compounds **7** or **8**, respectively.

3,7-Dimethyl-2-(2-methoxyethyl)-4-(4-nitrophenyl)-2,4dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-ol **7**

Yield 49%. Oil. IR (KBr, cm⁻¹) 3420–3300 (broad, OH). ¹H-NMR (DMSO-d₆) δ 2.25 (s, 3H, C³-CH₃), 2.29 (s, 3H, C⁷-CH₃), 3.0 (s, 3H, OCH₃), 3.52–3.60 (m, 4H, CH₂CH₂), 5.49 (s, 1H, pyran-H), 6.79 (brs, 1H, OH, D₂O exchangeable), 7.12 (d, *J* = 9.0 Hz, 2H, Ar-H), and 7.65 (d, *J* = 8.4 Hz, 2H, Ar-H). Anal. calcd. for C₁₉H₁₉N₅O₅: C 57.43, H 4.82, N 17.62. Found: C 57.51, H 4.77, N 17.59.

5-(2-Methoxyethylsulfanyl)-3,7-dimethyl-4-(4nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[3,2-d]pyrimidine **8**

Yield 72%, m.p. 237 – 240°C. ¹H-NMR (DMSO-d₆) δ 1.80 (s, 3H, C³-CH₃), 2.25 (s, 3H, C⁷-CH₃), 3.00 (s, 3H, OCH₃), 3.21-3.29 (m, 4H, CH₂CH₂), 4.85 (s, 1H, pyran-H), 7.05 (d, *J* = 9.0 Hz, 2H, Ar-H), 7.80 (d, *J* = 8.4 Hz, 2H, Ar-H) and 11.30 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ 9.9 (C3-CH₃), 25 (C7-CH₃), 29 (C-4), 35.7 (S-CH₂), 57.8 (O-CH₃), 70 (CH₂-O), 97 (C-4a), 108 (C-3a), 123, 129, 136, 146 (Ar-C), 150 (C-3), 154 (C-9a), 162 (C-8a), 165 (C-7), 170 (C-5).

Anal. calcd. for $C_{19}H_{19}N_5O_4S$: C 55.20, H 4.63, N 16.94, S 7.76. Found: C 55.25, H 4.68, N 17.00, S 7.60.

Attempted deprotection of compounds 7 and 8

To a solution of compounds **7** (0.19 g, 0.5 mmol) or **8** (0.20 g. 0.5 mmol) in ethanol (25 mL), ammonium hydroxide solution (20 mL, 25%) was added with stirring at room temperature for 1 h. The isolated solids were filtered off, and dried to give compounds identical in all aspects with compounds **4** and **6**, respectively.

3,7-Dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo-[4',3':5,6]pyrano[2,3-d]pyrimidin-5-ylhydrazine **9**

To a solution of compound **5** (3.59 g, 1 mmol) in dry dioxane (30 mL), hydrazine hydrate (3 mL, 99%) was added and the reaction mixture was heated on a water bath for 1 h. After cooling the precipitated material was filtered off, washed several times with water, dried and recrystallized from ethanol to give compound **9**. Yield 99%, m.p. 273 – 274°C. IR (KBr, cm⁻¹) 3420 – 3238 (NH₂, NH). ¹H-NMR (DMSO-d₆) δ 1.97 (s, 3H, C³-CH₃), 2.37 (s, 3H, C⁷-CH₃), 4.34 (s, 2H, NH₂, D₂O exchangeable), 5.37 (s, 1H, pyran), 7.99 (d, *J* = 9.0 Hz, 2H, Ar-H), 7.81(s, 1H, NH, D₂O exchangeable), 8.13 (d, *J* = 9.0 Hz, 2H, Ar-H), and 12.06 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 353 [M⁺] (20.61). Anal. calcd. for C₁₆H₁₅N₇O₃: C 54.39, H 4.28, N 27.75. Found: C 54.22, H 4.32, N 27.88.

5-Amino-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-3,4disubstituted-1H-pyrazoles **10–14**; general procedure

To a solution of compound **7** (3.53 g, 1 mmol) in ethanol (30 mL), ethoxymethylenemalononitrile, methylethoxymethylenemalononitril, bis(methylthio)methylene-malononitrile, tetracyanoethylene, or ethyl(ethoxymethylene)cyanoacetate (1 mmol) was added, respectively. The reaction mixture was heated for 2 – 4 h. Then the formed precipitate was filtered off and recrystallized from appropriate solvent to give compounds **10–14**.

5-Amino-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-1Hpyrazole-4-carbonitrile **10**

Yield after 2 h: (96%, dioxane); m.p. $322-325^{\circ}$ C. IR (KBr, cm⁻¹) 3430, 3300 (NH₂), 3220 (NH), 2222 (CN). ¹H-NMR (DMSO-d₆) δ 1.89 (s, 3H, C³⁻CH₃), 2.59 (s, 3H, C⁷⁻CH₃), 5.94 (s, 1H, pyran), 6.79 (s, 2H, NH₂, D₂O exchangeable), 7.15 (d, *J* = 8.7 Hz, 2H, Ar-H), 7.82 (s, 1H, pyrazole-H), 8.01 (d, *J* = 8.7 Hz, 2H, Ar-H), and 12.30 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 429 [M⁺] (23.49). Anal. calcd. for C₂₀H₁₅N₉O₃: C 55.94, H 3.52, N 29.36. Found: C 56.06, H 3.44, N, 29.32.

5-Amino-3-methyl-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-1Hpyrazole-4-carbonitrile **11**

Yield after 4 h: (93%, dioxane); m.p. $342-344^{\circ}$ C. IR (KBr, cm⁻¹) 3450, 3350 (NH₂), 3235 (NH), and 2220 (CN). ¹H-NMR (DMSO-d₆) δ 1.90 (s, 3H, C³⁻CH₃), 1.99 (s, 3H, C³⁻CH₃), 2.58 (s, 3H, C⁷⁻CH₃), 6.04 (s, 1H, pyran), 6.75 (s, 2H, NH₂, D₂O exchangeable), 7.15 (d, *J* = 9.0 Hz, 2H, Ar-H), 7.88 (d, *J* = 8.7 Hz, 2H, Ar-H) and 12.42 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 443 [M⁺] (20.85). Anal. calcd. for C₂₁H₁₇N₉O₃: C 56.88, H 3.86, N 28.43. Found: C 56.80, H 3.92, N 28.45.

5-Amino-3-methylthio-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5yl]-1H-pyrazole-4-carbonitrile **12**

Yield after 3 h: (84%, dioxane); m.p. $209 - 211^{\circ}$ C. IR (KBr, cm⁻¹) 3313, 3299 (NH₂), 3200 (NH) and 2222 (CN) cm⁻¹. ¹H-NMR (DMSO-d₆) δ 1.90 (s, 3H, C³-CH₃), 2.60 (s, 3H, C⁷-CH₃), 3.76 (s, 3H, SCH₃), 5.99 (s, 1H, pyran), 6.85 (s, 2H, NH₂, D₂O exchangeable), 7.16 (d, *J* = 8.7 Hz, 2H, Ar-H), 7.95 (d, *J* = 8.7 Hz, 2H, Ar-H), and 12.35 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 475 [M⁺] (3.25). Anal. calcd. for C₂₁H₁₇N₉O₃S: C 53.05, H 3.60, N 26.51, S 6.74. Found: C 53.16, H 3.64, N 26.38, S 6.72.

5-Amino-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-1Hpyrazole-3,4-dicarbonitrile **13**

Yield after 4 h: (93%, ethyl acetate); m.p. $277-279^{\circ}$ C. IR (KBr, cm⁻¹) 3436, 3350 (NH₂), 3161 (NH), and 2229 (CN). ¹H-NMR (DMSO-d₆) δ 1.89 (s, 3H, C^{3'}-CH₃), 2.59 (s, 3H, C^{7'}-CH₃), 6.12 (s, 1H, pyran), 6.55 (s, 2H, NH₂, D₂O exchangeable), 7.15 (d, *J* = 9.0 Hz, 2H, Ar-H), 8.01 (d, *J* = 9.0 Hz, 2H, Ar-H) and 12.39 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 454 [M⁺] (18.30). Anal. calcd. for C₂₁H₁₄N₁₀O₃: C 55.51, H 3.11, N 30.82. Found: C 55.64, H 3.04, N 30.76.

Ethyl-5-amino-1-[3,7-dimethyl-4-(4-nitrophenyl)-2,4dihydropyrazolo[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-1Hpyrazole-4-carboxylate **14**

Yield after 4 h: (95%, dioxane); m.p. $272-274^{\circ}$ C. IR (KBr, cm⁻¹) 3436, 3350 (NH₂), 3150 (NH) and 1683 (CO). ¹H-NMR (DMSO-d₆) δ 1.23 (t, *J* = 7.2 Hz, 3H, CH₂CH₃), 1.90 (s, 3H, C³⁻CH₃), 2.60 (s, 3H, C⁷⁻CH₃), 4.15 (q, *J* = 6.9 Hz, 2H, CH₂CH₃), 6.12 (s, 1H, pyran), 6.53 (s, 2H, NH₂, D₂O exchangeable), 7.17 (d, *J* = 8.7 Hz, 2H, Ar-H), 7.75 (s, 1H, pyrazole-H), 7.96 (d, *J* = 8.7 Hz, 2H, Ar-H), and 12.32 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 476 [M⁺] (41.88). Anal. calcd. for C₂₂H₂₀N₈O₅: C 55.46, H 4.23, N 23.52. Found: C 55.52, H 4.20, N 23.54.

5-(3,5-Dimethylpyrazol-1-yl)-3,7-dimethyl-4-(4-nitrophenyl)-2,4-dihydro-pyrazolo[4',3':5,6]pyrano[2,3d]pyrimidine **15**

A mixture of compound **9** (3.53 g, 1 mmol), acetylacetone (1 mmol) in anhydrous ethanol (30 mL) was refluxed for 5 h. The reaction mixture was allowed to cool and the formed precipitate was filtered off, dried, and recrystallized from ethanol to give compound **15**. Yield 65%, m.p. $337 - 340^{\circ}$ C. ¹H-NMR (DMSO-d₆) δ 1.75 (s, 3H, C³⁻CH₃), 1.86 (s, 3H, C³⁻CH₃), 2.25 (s, 3H, C^{5'}-CH₃), 2.58 (s, 3H, C⁷⁻CH₃), 5.80 (s, 1H, pyran-H), 5.97 (s, 1H, pyrazole-H) 7.06 (d, *J* = 9.0 Hz, 2H, Ar-H), 7.95 (d, *J* = 9.0 Hz, 2H, Ar-H), and 12.27 (s, 1H, NH, D₂O exchangeable). MS *m*/*z* (%): 417 [M⁺] (100). Anal. calcd. for C₂₁H₁₉N₇O₃: C 60.43, H 4.59, N 23.49. Found: C 60.33, H 4.62, N 23.56.

1-[3,7-Dimethyl-4-(4-nitrophenyl)-2,4-dihydropyrazolo-[4',3':5,6]pyrano[2,3-d]pyrimidin-5-yl]-3,4-dimethyl-3a,7adihydro-1H-pyrano[2,3-c]pyrazol-6-one **16**

A solution of compound **9** (3.53 g, 1 mmol) in ethyl acetoacetate (30 mL) was refluxed for 5 h. The mixture was evaporated under reduced pressure and the obtained oil washed several time with diethyl ether and recrystallized from ethanol to give compound

16. Yield 66%, m.p. 307-310°C. ¹H-NMR (DMSO-d₆) δ 1.82 (s, 3H, C3'-CH3), 2.35 (s, 3H, C3-CH3), 2.51 (s, 3H, C4-CH3), 2.63 (s, 3H, C7-CH₃), 5.69 (s, 1H, pyran-H), 5.71 (s, 1H, pyran-H) 7.15 (d, J = 9.0 Hz, 2H, Ar-H), 7.89 (d, J = 9.0 Hz, 2H, Ar-H) and 12.35 (s, 1H, NH, D₂O exchangeable). MS m/z (%): 485 [M⁺] (100). Anal. calcd. for C₂₄H₁₉N₇O₅: C 59.38, H 3.94, N 20.20. Found: C 59.51, H 4.01, N 20.00.

Antiviral screening

Preparation of synthetic compounds for bioassay

Tested compounds were dissolved as 10 mg each in 1 mL of 10% DMSO in water. The final concentration was 10 µg/µL (stock solution). The dissolved stock solutions were decontaminated by addition of 10 mL antibiotic-antimycotic mixture (10 000 U penicillin G sodium, 10000 µg streptomycin sulfates and 250 µg amphotericin B, PAA Laboratories GmbH, Austria).

Cell culture

African green monkey kidney-derived cells (Vero) were used. Cells were propagated in Dulbeccos' Minimal Essential Medium (DMEM) supplemented with 10% fetal bovine serum, 1% antibiotic-antimycotic mixture. The pH was adjusted at 7.2-7.4 by 7.5% sodium bicarbonate solution. The mixture was sterilized by filtration through 0.2 µm pore size nitrocellulose membrane. Viruses: Herpes Simplex Virus type-1 was obtained from Environmental Virology Lab., Department of Water Pollution Research, National Research Centre, Cairo, Egypt

Cytotoxicity assay

Cytotoxicity was assayed for both dimethylsulfoxide (DMSO) and the tested compounds. Serial dilutions were prepared and inoculated on Vero cells grown in 96-well tissue culture plates. The maximum tolerated concentration (MTC) for each compound was determined by both cell morphology and cell viability by staining with trypan blue dye.

Plague reduction infectivity assay

A 6-well plate was cultivated with cell culture (10⁵ cell/mL) and incubated for 2 days at 37°C. HSV-1 was diluted to give 10⁴ PFU/ mL and mixed with the tested compound at 10 and 20 μg concentrations and incubated overnight at 4°C. Growth medium was removed from the multiwell plate and virus-compound mixture was inoculated (100 µL/well). After 1 h contact time, the inoculum's was aspirated and 3 mL of MEM with 1% agarose was overlaid the cell sheets. The plates were left to solidify and incubated at 37°C until the development of virus plaques. Cell sheets were fixed in 10% formalin solution for 2 h and stained with crystal violet stain. Control virus and cells were treated identically without chemical compounds. A standard acyclovir (Sigma-Aldrich) was used at the same concentrations as the tested compounds. Virus plaques were counted and the percentage of reduction was calculated [27].

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