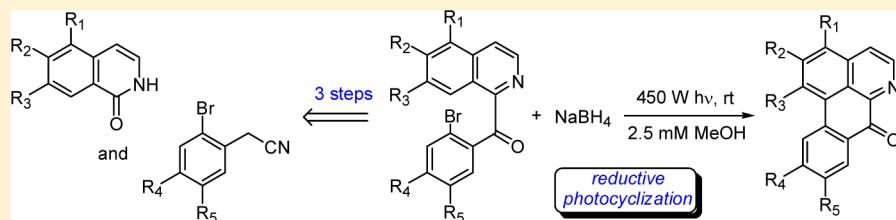


Direct Conversion of 1-(2-Bromobenzoyl)isoquinolines to Dibenzo[*d,e,g*]quinolin-7-ones via Reductive Photocyclization

Ta-Hsien Chuang,* Chien-Fu Li, Hong-Zin Lee, and Yu-Chia Wen

School of Pharmacy, China Medical University, No. 91, Hsueh-Shih Road, Taichung, 40402 Taiwan, Republic of China

Supporting Information

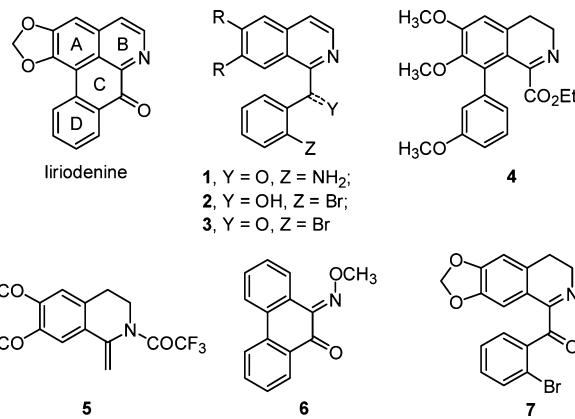


ABSTRACT: A series of A/D-ring substituted dibenzo[*d,e,g*]quinolin-7-ones was produced from the corresponding isoquinolinones and (2-bromophenyl)acetonitriles in four steps. This represents a convenient approach toward the synthesis of tetracyclic alkaloids. A direct conversion of 1-(2-bromobenzoyl)isoquinolines to dibenzo[*d,e,g*]quinolin-7-ones is the key step in the total synthesis. The yield of the reductive photocyclization depends on the position of the substituents at the isoquinolyl ring and the phenyl group. The mechanism of the reductive photocyclization is also discussed.

INTRODUCTION

Since the first isolation of liriodenine from *Atherosperma moschatum* Labill. in 1956,¹ dibenzo[*d,e,g*]quinolin-7-one alkaloids have been the focus of study in medicinal and synthetic chemistry. Some of the dibenzo[*d,e,g*]quinolin-7-one alkaloids have shown extensive pharmacological activities, such as antileishmanial,² antimicrobial,^{3,4} antivirus,^{5–7} antioxidative,⁸ vasorelaxing,^{8,9} antiplatelet aggregation,^{10–12} antiflammatory,¹³ and cytotoxic^{13–19} activities. Moreover, previous structure–activity relationship (SAR) studies have indicated that removal of the oxo function sharply reduces the cytotoxic activities against human lung carcinoma A-549 and colon HCT-8 tumor cells.²⁰ This finding suggests that the oxo function may contribute to anticancer activity by extending the conjugation of the aporphine ring system.^{20,21} However, dibenzo[*d,e,g*]quinolin-7-one alkaloids are present in minor amounts in plants, even though they may be widely distributed in many plant families (e.g., Annonaceae, Magnoliaceae).^{22–26} These tetracyclic alkaloids are most likely derived in the plants through the oxidation of the corresponding aporphines.^{27,28}

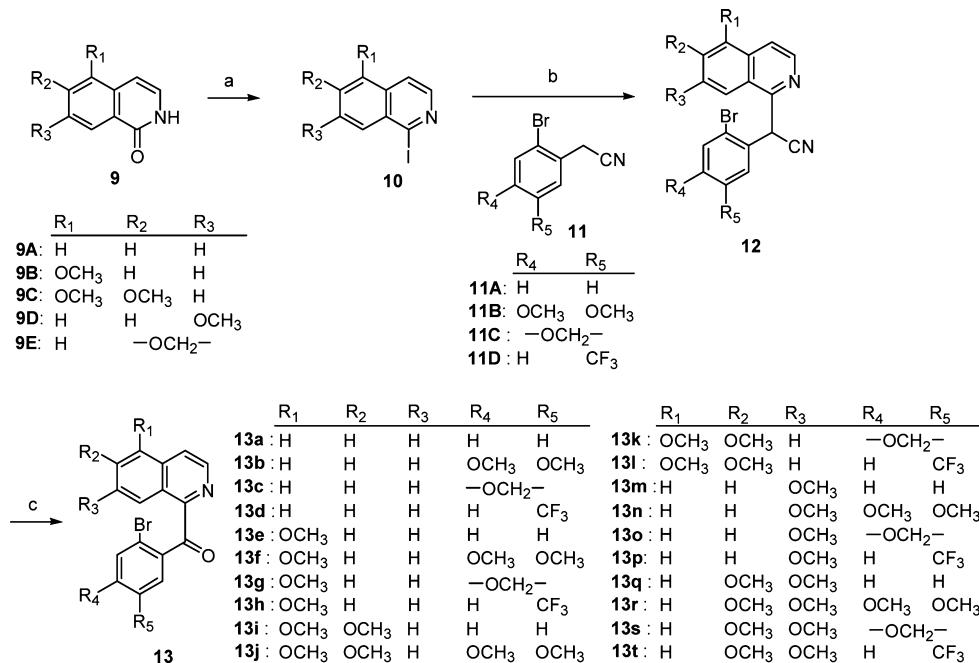
The wide range of pharmacological properties exhibited by dibenzo[*d,e,g*]quinolin-7-one alkaloids has attracted the attention of chemists. Over the past several decades, the following methodologies have allowed access to dibenzo[*d,e,g*]quinolin-7-one alkaloids: (a) Coupling of rings A/D: using the Pschorr cyclization of 1-(2-amionbenzoyl)isoquinoline 1, the first total synthesis of dibenzo[*d,e,g*]quinolin-7-ones was disclosed by Taylor in 1961,²⁹ and other methods are photocyclization³⁰ or Bu₃SnH-induced aryl radical cyclization³¹ of the bromides 2 or 3. (b) Ring closure between rings B and D: intramolecular Friedel–Crafts acylation of 8-phenyl-3,4-dihydroisoquinolines 4 in polyphosphoric acid has been utilized.³² (c) A convergent



method involving the intermolecular [4 + 2] cycloaddition of protected 1-methylene-isoquinoline 5 with benzene³³ or 10-(methoxyimino)phenanthren-9-one 6 with dimethyl acetylenedicarboxylate (DMAD)³⁴ to provide the tetracyclic skeleton. However, there are some inherent limitations in the above methodologies. For example, the symmetrical species benzene and DMAD were used. Moreover, the precursors of the Pschorr and radical cyclizations, obtained via alkylation of Reissert compounds, resulted in 1-cyanoisoquinoline byproducts that were difficult to separate from the products.³⁵ Furthermore, an attempt to prepare the aporphine skeleton by photocyclization of the bromo ketone 7, as reported by Kessar et al.,³⁶ was unsuccessful. In this paper, we disclose a convenient one-pot methodology to produce a variety of A/D-ring substituted dibenzo[*d,e,g*]quinolin-7-ones 8 via photocyclization of 1-(2-

Received: March 27, 2013

Published: April 23, 2013

Scheme 1. Synthesis of 1-(2-Bromobenzoyl)isoquinolines 13^a

^aReagents and conditions: (a) (i) Tf_2O , py, CH_3CN , rt, 2 h; (ii) NaI , TfOH , rt, overnight; (b) NaNH_2 , THF, reflux, overnight; (c) NaH , O_2 , THF, rt, 3 h.

Table 1. Formation of 12 and 13 from 1-Iodoisoquinolines 10 and (2-Bromophenyl)acetonitriles 11

entry	1-iodoisooquinoline (10), R =	(2-bromophenyl)acetonitrile (11), R =	products (%) ^a	
			12	13
1	10A, R = H	11A, R = H	12a (97)	13a (95)
2	10A, R = H	11B, R = 4,5-dimethoxy	12b (72)	13b (96)
3	10A, R = H	11C, R = 4,5-methylenedioxy	12c (76)	13c (81)
4	10A, R = H	11D, R = 5-trifluoromethyl	12d (89)	13d (59) ^b
5	10B, R = 5-methoxy	11A, R = H	12e (76)	13e (94)
6	10B, R = 5-methoxy	11B, R = 4,5-dimethoxy	12f (85)	13f (84)
7	10B, R = 5-methoxy	11C, R = 4,5-methylenedioxy	12g (95)	13g (67)
8	10B, R = 5-methoxy	11D, R = 5-trifluoromethyl	12h (75)	13h (55) ^b
9	10C, R = 5,6-dimethoxy	11A, R = H	12i (91)	13i (89)
10	10C, R = 5,6-dimethoxy	11B, R = 4,5-dimethoxy	12j (100)	13j (91)
11	10C, R = 5,6-dimethoxy	11C, R = 4,5-methylenedioxy	12k (97)	13k (94)
12	10C, R = 5,6-dimethoxy	11D, R = 5-trifluoromethyl	12l (93)	13l (65) ^b
13	10D, R = 7-methoxy	11A, R = H	12m (74)	13m (88)
14	10D, R = 7-methoxy	11B, R = 4,5-dimethoxy	12n (100)	13n (96)
15	10D, R = 7-methoxy	11C, R = 4,5-methylenedioxy	12o (84)	13o (88)
16	10D, R = 7-methoxy	11D, R = 5-trifluoromethyl	12p (80)	13p (74) ^b
17	10E, R = 6,7-methylenedioxy	11A, R = H	12q (92)	13q (92)
18	10E, R = 6,7-methylenedioxy	11B, R = 4,5-dimethoxy	12r (87)	13r (79)
19	10E, R = 6,7-methylenedioxy	11C, R = 4,5-methylenedioxy	12s (97)	13s (94)
20	10E, R = 6,7-methylenedioxy	11D, R = 5-trifluoromethyl	12t (91)	13t (53) ^b

^aIsolated yields. ^bRecovered starting materials 12d, 12h, 12l, 12p, and 12t in 24%, 25%, 30%, 12%, and 30% yields, respectively.

bromobenzoyl)isoquinolines 13 in the presence of sodium borohydride (NaBH_4). In addition, a systematic investigation of substituent effects on the reductive photocyclization is also discussed.

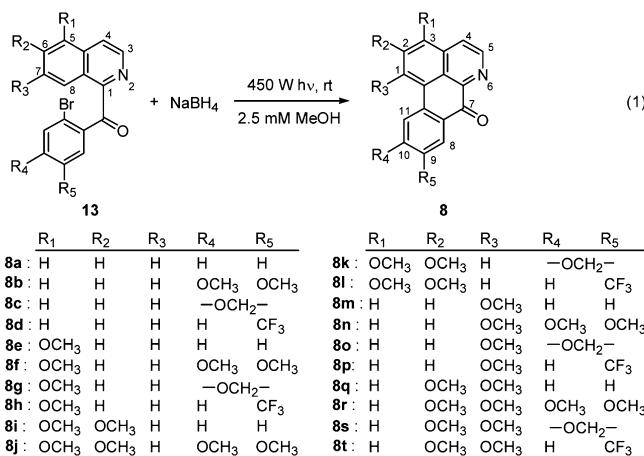
RESULTS AND DISCUSSION

The strategy for the preparation of 1-(2-bromobenzoyl)isoquinolines 13, the photocyclization precursors, is shown in Scheme 1. First, a series of isoquinolinones 9 was obtained via

thermal cyclization of the corresponding styryl isocyanates, as previously reported.³⁷ Subsequently, the reaction of compounds 9 with trifluoromethanesulfonic anhydride (Tf_2O) in the presence of pyridine led to the isoquinolyl triflates. Treatment of the intermediate triflates with sodium iodide and TfOH at room temperature in acetonitrile produced the desired iodides 10. The one-pot iodination of isoquinolinones 9 gave the 1-iodoisooquinolines 10 in high yields.

We originally planned to synthesize the photocyclization precursor **13a** via the Cu(I)-catalyzed acylation of 1-isoquinolylzinc iodide generated from the direct insertion of zinc into iodoisoquinoline **10a** in the presence of LiCl using 2-bromobenzoyl chloride.³⁸ Unfortunately, only the hydrolysis product isoquinoline was obtained despite the use of flame-dried glass vessels and standard Schlenk techniques. Therefore, we developed a two-step approach to synthesize the intermediates **13** via (2-bromophenyl)isoquinolin-1-ylacetonitriles **12**. The 1-iodoisoquinolines **10** reacted with commercially available (2-bromophenyl)acetonitriles **11** to give the adducts **12** in yields ranging from 72 to 100% (Table 1).³⁹ Then, the key photocyclization precursors **13** were readily produced in moderate to excellent yields (53 to 96%, Table 1) through oxidative decyanation of the corresponding nitriles **12** with oxygen in the presence of sodium hydride.⁴⁰ It should be mentioned here that some of the starting materials (**12d**, **12h**, **12l**, **12p**, and **12t**) were recovered from the oxidative decyanation reaction (Table 1, entries 4, 8, 12, 16, and 20). However, the conversion could not be completed by extending the reaction time to 6 h, most likely because the α -carbanions of the acetonitriles **12** were stabilized by the inductive effect of the electron-withdrawing substituent $-\text{CF}_3$.

With the series of 1-(2-bromobenzoyl)isoquinolines **13** in hand, the key cyclization reaction was investigated. The unsubstituted dibenzo[*d*₂*g*]quinolin-7-one **8a** was chosen as our initial target because its simplicity would allow us to test the feasibility of the cyclization. First, the bromo ketone **13a** was subjected to Bu₃SnH-induced aryl radical cyclization according to the method of Orito et al.³¹ and a somewhat modified version of this method to give the cyclization adduct **8a** in a low yield. Additionally, we attempted to synthesize dibenzo[*d*₂*g*]quinolin-7-one **8a** by photocyclization of bromo ketone **13a** in MeOH, which resulted in a complicated reaction mixture. This result is similar to Kessar's observation that the photoinduced cyclization of bromo ketone **7** in dilute aqueous HCl solution was unsuccessful.³⁶ Intriguingly, we found that dibenzo[*d*₂*g*]quinolin-7-one **8a** could be obtained in 55% yield by irradiation of a solution of the bromo ketone **13a** and 1.5 equiv of NaBH₄ in MeOH (eq 1). Thus, the direct conversion of 1-(2-bromobenzoyl)isoquinolines to dibenzo[*d*₂*g*]quinolin-7-ones using one-pot reductive photocyclization was accomplished.



Based on the results of the model study, we anticipated that a series of dibenzo[*d*₂*g*]quinolin-7-ones **8b–t** would be readily produced using the reductive photocyclization method (eq 1). The substituent effects on the reaction time and the yields from

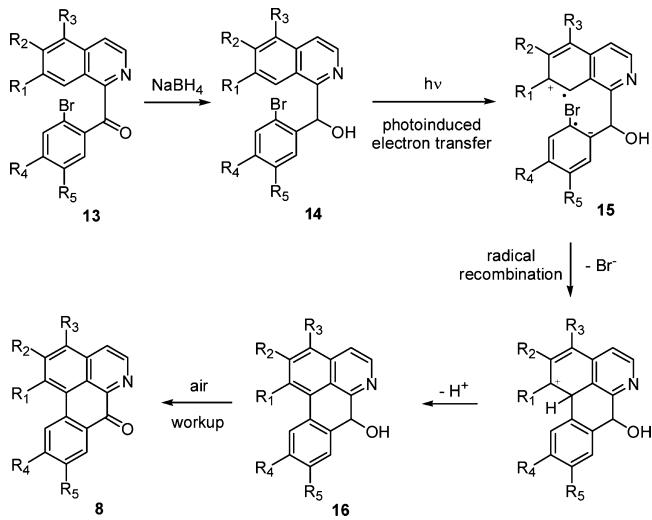
the reductive photocyclization of 1-(2-bromobenzoyl)-isoquinolines **13a–t** are shown in Table 2. First, with the bromo ketones **13b** and **13c**, which have vicinal dimethoxy or methylenedioxy groups at the 4,5-positions of the benzoyl group, the yields of dibenzo[*d*₂*g*]quinolin-7-ones **8b** and **8c** slightly increased (compare entry 1 with entries 2 and 3). However, the photocyclization of bromo ketone **13d**, which has a strong electron-withdrawing group ($-\text{CF}_3$) at the 5-position of the phenyl group, was complete after a short period of time (9 h) and led to the production of the corresponding dibenzo[*d*₂*g*]quinolin-7-one **8d** in 33% yield along with several complicated inseparable products (entry 4). We hypothesized that a shorter reaction time was needed because the photoinduced electron transfer from isoquinolyl ring to phenyl ring could be enhanced by the electron-withdrawing substituent $-\text{CF}_3$ on the phenyl ring. Second, bromo ketones **13e**, **13f**, **13g**, and **13h** bearing the electron-donating group 5-OCH₃ on the isoquinolyl ring produced the corresponding dibenzo[*d*₂*g*]quinolin-7-ones **8e**, **8f**, **8g**, and **8h** in higher yields (86%, 95%, 80%, and 99%, respectively). Similarly, when bromo ketones **13m**, **13n**, **13o**, and **13p**, which have a methoxy group at the 7-positions of the isoquinolyl ring, were utilized the photocyclizations provided higher yields of dibenzo[*d*₂*g*]quinolin-7-ones **8m**, **8n**, **8o**, and **8p** (63%, 95%, 81%, and 80%, respectively). The results revealed that the reactivity of the photocyclization of 1-(2-bromobenzoyl)isoquinolines could be increased by the presence of 5-OCH₃ and 7-OCH₃ substituents (compare entry 1 with entries 5 and 13, entry 2 with entries 6 and 14, entry 3 with entries 7 and 15, and entry 4 with entries 8 and 16). Third, when the bromo ketones **13i**, **13j**, **13k**, and **13l**, which have vicinal 5,6-dimethoxy groups on the isoquinolyl ring, were subjected to the same conditions, formation of the dibenzo[*d*₂*g*]quinolin-7-ones was disfavored (compare entry 5 with entry 9, entry 6 with entry 10, entry 7 with entry 11, and entry 8 with entry 12). The lower yields were also observed when the bromo ketones **13q**, **13r**, **13s**, and **13t**, which have a 6,7-methylenedioxy group on the isoquinolyl ring, were used (compare entry 13 with entry 17, entry 14 with entry 18, entry 15 with entry 19, and entry 16 with entry 20). We hypothesized that the yields of dibenzo[*d*₂*g*]quinolin-7-ones decreased because the photoinduced electron transfer from isoquinolyl ring to phenyl ring could be retarded by the inductive effect of the oxygen atom at the 6-position of the isoquinolyl ring. Clearly, the reductive photocyclization was promoted by methoxy groups at the 5- or 7-positions of the isoquinolyl ring or the trifluoromethyl group at the 5-position of the phenyl group. In contrast, the reactivity of the reductive photocyclization was decreased by a methoxy group on the 6-position of the isoquinolyl ring.

Mechanistic study. On the basis of the above results, a complete mechanistic route for the formation of dibenzo[*d*₂*g*]quinolin-7-ones **8** through a reductive photocyclization of 1-(2-bromobenzoyl)isoquinolines **13** with NaBH₄ is proposed and depicted in Scheme 2. First, NaBH₄ apparently plays a role in reducing bromo ketones **13** into the corresponding α -hydroxy-(2-bromobenzyl)isoquinolines **14** prior to photocyclization because irradiation of bromo ketone **13a** in MeOH in the absence of NaBH₄ led to a complicated reaction mixture. Subsequently, the photoreaction of intermediates **14** might be initiated by photoinduced electron transfer, and gave diradicals **15** with a radical-cation on the isoquinolyl ring and a radical-anion on the phenyl ring.⁴¹ The carbocation of the intermediates **15** could be stabilized by the resonance effect of

Table 2. Reductive Photocyclization of Bromo Ketone Precursors **13a–t** with NaBH₄ (1.5 equiv) in MeOH (2.5 mM) To Provide Dibenzo[*d,e,g*]quinolin-7-ones **8a–t**

entry	reactant	time (h)	substituent					product 8 (%)
			R ₁	R ₂	R ₃	R ₄	R ₅	
1	13a	15	H	H	H	H	H	8a (55)
2	13b	24	H	H	H	OCH ₃	OCH ₃	8b (77)
3	13c	15	H	H	H	—OCH ₂ O—		8c (69)
4	13d	9	H	H	H	H	CF ₃	8d (33)
5	13e	15	OCH ₃	H	H	H	H	8e (86)
6	13f	20	OCH ₃	H	H	OCH ₃	OCH ₃	8f (95)
7	13g	15	OCH ₃	H	H	—OCH ₂ O—		8g (80)
8	13h	10	OCH ₃	H	H	H	CF ₃	8h (99)
9	13i	48	OCH ₃	OCH ₃	H	H	H	8i (61)
10	13j	24	OCH ₃	OCH ₃	H	OCH ₃	OCH ₃	8j (70)
11	13k	24	OCH ₃	OCH ₃	H	—OCH ₂ O—		8k (78)
12	13l	24	OCH ₃	OCH ₃	H	H	CF ₃	8l (80)
13	13m	15	H	H	OCH ₃	H	H	8m (63)
14	13n	20	H	H	OCH ₃	OCH ₃	OCH ₃	8n (95)
15	13o	15	H	H	OCH ₃	—OCH ₂ O—		8o (81)
16	13p	10	H	H	OCH ₃	H	CF ₃	8p (80)
17	13q	13	H	—OCH ₂ O—		H	H	8q (45)
18	13r	18	H	—OCH ₂ O—		OCH ₃	OCH ₃	8r (81)
19	13s	18	H	—OCH ₂ O—		—OCH ₂ O—		8s (53)
20	13t	5	H	—OCH ₂ O—		H	CF ₃	8t (72)

Scheme 2. Mechanism of the Reductive Photocyclization of 1-(2-Bromobenzoyl)isoquinolines **13** with NaBH₄



the methoxy groups at the 5- or 7-positions of the isoquinolyl ring, and the carbanion of the intermediates **15** could also be stabilized by the inductive effect of the trifluoromethyl group on the phenyl ring. Hence, the reductive photocyclization of (2-bromo-5-trifluoromethylphenyl)-(5-methoxyisoquinolin-1-yl)-methanone (**13h**) was obtained in the highest yield. In contrast, the reductive photocyclization of bromo ketones with a methoxy group at the 6-position of the isoquinolyl ring was retarded because the inductive effect of oxygen made the carbocation on the isoquinolyl ring less stable. The facts supported the existence of the diradical-ions **15**. After elimination of the bromide as well as radical recombination, a loss of proton from the isoquinolyl ring would lead to the formation of 7-hydroxyaporphines **16**. No further oxidation occurred before workup due to the presence of NaBH₄, and this is supported by the obvious color change (from orange to

yellow) observed during the workup. The auto-oxidation of intermediates **16** with air resulted in the formation of the desired dibenzo[*d,e,g*]quinolin-7-ones **8** after thorough removal of the reducing reagent during the workup procedure.⁴²

CONCLUSION

To our knowledge, this paper is the first to describe the reductive photocyclization of 1-(2-bromobenzoyl)isoquinolines **13** to give dibenzo[*d,e,g*]quinolin-7-ones **8** using NaBH₄ as a reducing agent. A series of substituted dibenzo[*d,e,g*]quinolin-7-ones **8** was synthesized from the corresponding bromo ketones **13** in a one-pot reaction. The substituent effects on the subsequent reductive photocyclization are shown. When —OCH₃ is at the 5- or 7-positions of the isoquinolyl ring or —CF₃ at the 5-position of the phenyl group, the photocyclization can be enhanced. Hence, 9-trifluoromethyl-3-methoxy-dibenzo[*d,e,g*]quinolin-7-one (**8h**) could be obtained in the highest yield.

EXPERIMENTAL SECTION

General Methods. All reagents were purchased and used without further purification. Column chromatography was carried out using a 230–400 mesh silica gel. Nuclear magnetic resonance spectra were recorded on 500 and 600 MHz FT-NMR spectrometers; all chemical shifts were reported in ppm from tetramethylsilane as an internal standard. Melting points are uncorrected. FT-IR, MS, and elemental analysis were measured with conventional spectrometers. High-resolution electron ionization mass spectra (HREI-MS) were recorded on a high-resolution E/B mass spectrometer using a double-focusing magnetic-sector mass analyzer and orbitrap mass analyzer used for HRESI-MS measurements.

General Procedure for the Preparation of 1-Iodoisoquinoline **10.** Trifluoromethanesulfonic anhydride (11.0 mmol) was slowly added to a mixture of the isoquinolinones **9**³⁷ (10.0 mmol) and pyridine (11.5 mmol) in acetonitrile (20 mL) cooled in an ice bath. The reaction mixture was stirred for 2 h at room temperature under N₂. Without purification, sodium iodide (50.0 mmol) was added in one portion at room temperature, and trifluoromethanesulfonic acid (11.0 mmol) was then added dropwise via syringe. The reaction

mixture was stirred at room temperature overnight. After the reaction was complete, the resulting solution was diluted with water (35 mL) and toluene (45 mL). Then, the mixture was quenched with 10 M sodium hydroxide to reach a pH of ca. 10 while being cooled in an ice bath. The organic layer was separated and washed with 5% sodium thiosulfate (2×10 mL), 1 M sodium hydroxide (10 mL), and water (3×10 mL). The organic extract was dried with anhydrous $MgSO_4$, filtered, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane–EtOAc (10:1) to give the pure 1-iodoisooquinolines **10**. The full spectral data of **10A–E** are described as follows.

1-Iodoisooquinoline (10A): yield 90% (2.30 g, 9.0 mmol); white crystal; mp 70.5–71 °C (hexane–EtOAc) (lit.⁴³ mp 72–74 °C); ¹H NMR (500 MHz, $CDCl_3$) δ 7.58 (1H, d, $J = 5.6$ Hz), 7.67–7.76 (3H, m), 8.12 (1H, d, $J = 8.7$ Hz), 8.26 (1H, d, $J = 5.6$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 121.3, 127.3, 127.5, 129.0, 131.1, 132.0, 132.9, 136.2, 143.1; IR (KBr) 1612, 1578, 1541 cm^{-1} ; EIMS m/z (rel int) 128 (100), 255 (64, M^+). Anal. Calcd for C_9H_6IN : C, 42.38; H, 2.37; N, 5.49. Found: C, 42.65; H, 2.37; N, 5.34.

1-Iodo-5-methoxyisooquinoline (10B): yield 88% (2.51 g, 8.8 mmol); white needle; mp 115.5–116 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 4.02 (3H, s), 7.05 (1H, d, $J = 8.2$ Hz), 7.57 (1H, t, $J = 8.2$ Hz), 7.68 (1H, d, $J = 8.2$ Hz), 7.98 (1H, d, $J = 5.6$ Hz), 8.24 (1H, d, $J = 5.6$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 55.9, 108.2, 110.0, 115.7, 124.5, 126.6, 128.7, 132.6, 142.6, 154.5; IR (KBr) 1578, 1483 cm^{-1} ; EIMS m/z (rel int) 285 (100, M^+). Anal. Calcd for $C_{10}H_8INO$: C, 42.13; H, 2.83; N, 4.91. Found: C, 41.77; H, 3.05; N, 4.53.

1-Iodo-5,6-dimethoxyisooquinoline (10C): yield 87% (2.74 g, 8.7 mmol); white needle; mp 91–91.5 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 3.98 (3H, s), 4.04 (3H, s), 7.42 (1H, d, $J = 9.3$ Hz), 7.82 (1H, d, $J = 5.8$ Hz), 7.88 (1H, d, $J = 9.3$ Hz), 8.15 (1H, d, $J = 5.8$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 56.5, 61.3, 115.0, 116.3, 126.9, 127.4, 129.8, 131.9, 141.3, 142.6, 151.9; IR (KBr) 1614, 1580, 1551, 1476 cm^{-1} ; EIMS m/z (rel int) 315 (100, M^+). Anal. Calcd for $C_{11}H_{10}INO_2$: C, 41.93; H, 3.20; N, 4.45. Found: C, 42.05; H, 3.60; N, 4.36.

1-Iodo-7-methoxyisooquinoline (10D): yield 84% (2.39 g, 8.4 mmol); white needle; mp 86.5–87 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 4.00 (3H, s), 7.35–7.37 (2H, m), 7.51 (1H, d, $J = 5.4$ Hz), 7.65 (1H, d, $J = 8.8$ Hz), 8.16 (1H, d, $J = 5.4$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 55.6, 110.5, 121.1, 124.1, 125.8, 128.9, 131.7, 133.3, 141.3, 159.8; IR (KBr) 1622, 1580, 1547, 1499 cm^{-1} ; EIMS m/z (rel int) 158 (100), 285 (31, M^+); HREIMS m/z calcd for $C_{10}H_8INO$ 284.9651, found 284.9641 [M^+].

1-Iodo-6,7-(methylenedioxy)isooquinoline (10E): yield 75% (2.24 g, 7.5 mmol); white needle; mp 181–182 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 6.14 (2H, s), 6.99 (1H, s), 7.41 (1H, d, $J = 5.5$ Hz), 7.46 (1H, s), 8.10 (1H, d, $J = 5.5$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 102.1, 102.8, 109.1, 120.8, 124.7, 129.6, 134.5, 142.1, 149.7, 151.4; IR (KBr) 1576, 1458, 1404 cm^{-1} ; EIMS m/z (rel int) 172 (100), 299 (80, M^+). Anal. Calcd for $C_{10}H_6INO_2$: C, 40.16; H, 2.02; N, 4.68. Found: C, 40.28; H, 2.14; N, 4.62.

General Procedure for the Preparation of (2-Bromophenyl)-isooquinolin-1-yl-acetonitrile 12. Sodium amide (43.8 mmol) was added to a mixture of the 1-iodoisooquinolines **10** (7.0 mmol) and (2-bromophenyl)acetonitriles **11** (8.8 mmol) in tetrahydrofuran (100 mL) while being stirred and cooled in an ice bath. The reaction mixture was then refluxed overnight under N_2 . After cooling, the resulting solution was quenched with water (50 mL) and extracted with EtOAc (3×100 mL). The combined extracts were washed with water, dried with anhydrous $MgSO_4$, filtered, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane–EtOAc (5:1) to give the pure (2-bromophenyl)isooquinolin-1-yl-acetonitriles **12**. The full spectral data of **12a–t** are described as follows.

(2-Bromophenyl)isooquinolin-1-ylacetonitrile (12a): Yield 97% (2.19 g, 6.8 mmol); white needle; mp 160–161 °C (hexane–EtOAc) (lit.⁴⁴ mp 152.5 °C); ¹H NMR (500 MHz, $CDCl_3$) δ 6.52 (1H, s), 7.21 (1H, td, $J = 7.8, 1.7$ Hz), 7.27 (1H, td, $J = 7.8, 1.7$ Hz),

7.41 (1H, dd, $J = 7.8, 1.7$ Hz), 7.60–7.63 (1H, m), 7.65 (1H, dd, $J = 7.8, 1.7$ Hz), 7.69–7.72 (2H, m), 7.89 (1H, d, $J = 8.2$ Hz), 7.97 (1H, d, $J = 8.2$ Hz), 8.61 (1H, d, $J = 5.7$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 42.2, 177.7, 121.5, 123.4, 123.7, 125.5, 127.7, 128.2, 128.3, 130.2, 130.3, 130.4, 133.1, 134.2, 136.6, 141.9, 153.0; IR (KBr) 2249 cm^{-1} ; ESIMS m/z (rel int) 323 (100, $[M + 1]^+$), 323 (100, $[M + 3]^+$). Anal. Calcd for $C_{17}H_{11}BrN_2$: C, 63.18; H, 3.43; N, 8.67. Found: C, 63.17; H, 3.20; N, 8.51.

(2-Bromo-4,5-dimethoxyphenyl)isoquinolin-1-yl-acetonitrile (12b): yield 72% (1.93 g, 5.0 mmol); white granule; mp 146–147 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 3.69 (3H, s), 3.86 (3H, s), 6.45 (1H, s), 6.91 (1H, s), 7.07 (1H, s), 7.62 (1H, t, $J = 8.3$ Hz), 7.69–7.72 (2H, m), 7.88 (1H, d, $J = 8.3$ Hz), 8.02 (1H, d, $J = 8.3$ Hz), 8.63 (1H, d, $J = 8.3$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 41.8, 55.9, 56.1, 112.4, 113.5, 115.3, 118.1, 121.5, 123.9, 125.6, 126.0, 127.7, 128.3, 130.5, 136.7, 141.9, 149.1, 149.9, 153.5; IR (KBr) 2243 cm^{-1} ; ESIMS m/z (rel int) 383 (100, $[M + 1]^+$), 385 (97, $[M + 3]^+$). Anal. Calcd for $C_{19}H_{12}BrN_2O_2$: C, 59.55; H, 3.95; N, 7.31. Found: C, 59.64; H, 4.07; N, 7.07.

(2-Bromo-4,5-methylenedioxypyhenyl)isoquinolin-1-ylacetone (12c): yield 76% (1.95 g, 5.3 mmol); white granule; mp 168–169 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 5.89 (1H, s), 5.95 (1H, s), 6.43 (1H, s), 6.81 (1H, s), 7.06 (1H, s), 7.60 (1H, t, $J = 8.2$ Hz), 7.68–7.71 (2H, m), 7.87 (1H, d, $J = 8.2$ Hz), 7.93 (1H, d, $J = 8.2$ Hz), 8.62 (1H, d, $J = 5.7$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 42.2, 102.2, 109.8, 112.8, 114.2, 117.9, 121.6, 123.8, 125.6, 127.3, 127.8, 128.4, 130.5, 136.8, 142.0, 148.1, 148.8, 153.3; IR (KBr) 2251 cm^{-1} ; ESIMS m/z (rel int) 367 (100, $[M + 1]^+$), 369 (98, $[M + 3]^+$); HRESIMS m/z calcd for $C_{18}H_{12}BrN_2O_2$ 367.0077, found 367.0079 [$M + 1]^+$.

(2-Bromo-5-trifluoromethylphenyl)isoquinolin-1-ylacetone (12d): yield 89% (2.44 g, 6.2 mmol); pale yellow needle; mp 147–148 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 6.54 (1H, s), 7.48 (1H, dd, $J = 8.3, 1.9$ Hz), 7.67 (1H, t, $J = 8.3$ Hz), 7.71–7.81 (4H, m), 7.92 (1H, d, $J = 8.3$ Hz), 8.03 (1H, d, $J = 8.3$ Hz), 8.58 (1H, d, $J = 5.7$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 41.7, 117.3, 121.9, 123.2 (q, $J = 271$ Hz), 123.4, 125.7, 126.9 (q, $J = 4$ Hz), 127.4, 127.5 (q, $J = 4$ Hz), 128.0, 128.6, 130.7, 130.8 (q, $J = 33$ Hz), 133.8, 135.4, 136.8, 142.1, 152.0; IR (KBr) 2245 cm^{-1} ; ESIMS m/z (rel int) 391 (100, $[M + 1]^+$), 393 (97, $[M + 3]^+$). Anal. Calcd for $C_{18}H_{10}BrF_3N_2$: C, 55.27; H, 2.58; N, 7.16. Found: C, 55.21; H, 2.66; N, 7.19.

(2-Bromophenyl)(5-methoxyisooquinolin-1-yl)acetonitrile (12e): yield 76% (1.88 g, 5.3 mmol); white granule; mp 146–147 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 4.00 (3H, s), 6.46 (1H, s), 6.99 (1H, t, $J = 7.8$ Hz), 7.19 (1H, t, $J = 7.8$ Hz), 7.25 (1H, t, $J = 7.8$ Hz), 7.37 (1H, d, $J = 7.8$ Hz), 7.49–7.50 (2H, m), 7.63 (1H, d, $J = 7.8$ Hz), 8.09 (1H, d, $J = 5.8$ Hz), 8.59 (1H, d, $J = 5.8$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 42.5, 55.7, 107.7, 115.4, 115.9, 117.8, 123.6, 126.5, 128.2, 128.5, 129.5, 130.2, 130.4, 133.2, 134.4, 141.6, 152.4, 155.2; IR (KBr) 2245 cm^{-1} ; ESIMS m/z (rel int) 353 (100, $[M + 1]^+$), 355 (93, $[M + 3]^+$). Anal. Calcd for $C_{18}H_{12}BrN_2O$: C, 61.21; H, 3.71; N, 7.93. Found: C, 61.40; H, 3.98; N, 8.04.

(2-Bromo-4,5-dimethoxyphenyl)(5-methoxyisooquinolin-1-yl)acetonitrile (12f): yield 85% (2.46 g, 6.0 mmol); white needle; mp 144–145 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 3.67 (3H, s), 3.85 (3H, s), 4.01 (3H, s), 6.40 (1H, s), 6.88 (1H, s), 7.00 (1H, d, $J = 7.4$ Hz), 7.06 (1H, s), 7.49–7.55 (2H, m), 8.09 (1H, d, $J = 5.8$ Hz), 8.62 (1H, d, $J = 5.8$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 42.1, 55.7, 56.0, 56.2, 107.7, 112.4, 113.6, 115.4, 115.5, 115.9, 118.2, 126.2, 126.4, 128.5, 129.5, 141.5, 149.1, 149.9, 152.9, 155.2; IR (KBr) 2245 cm^{-1} ; ESIMS m/z (rel int) 413 (100, $[M + 1]^+$), 415 (92, $[M + 3]^+$). Anal. Calcd for $C_{20}H_{12}BrN_2O_3$: C, 58.12; H, 4.16; N, 6.90.

(2-Bromo-4,5-methylenedioxypyhenyl)(5-methoxyisooquinolin-1-yl)acetonitrile (12g): yield 95% (2.64 g, 6.7 mmol); white crystal; mp 170–171 °C (hexane–EtOAc); ¹H NMR (500 MHz, $CDCl_3$) δ 4.01 (3H, s), 5.90 (1H, d, $J = 1.3$ Hz), 5.95 (1H, d, $J = 1.3$ Hz), 6.39 (1H, s), 6.80 (1H, s), 7.00 (1H, d, $J = 6.7$ Hz), 7.06 (1H, s), 7.46–7.52 (2H, m), 8.09 (1H, d, $J = 5.8$ Hz), 8.61 (1H, d, $J = 5.8$ Hz); ¹³C NMR (125 MHz, $CDCl_3$) δ 42.5, 55.8, 102.2, 107.7, 109.8, 112.8, 114.3, 115.4,

m/z (rel int) 411 (100, [M + 1]⁺), 413 (93, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₉H₁₂BrN₂O₄ 410.9975, found 410.9977 [M + 1]⁺.

(2-Bromo-5-trifluoromethylphenyl)(6,7-methylenedioxyisoquinolin-1-yl)acetonitrile (**12t**): yield 91% (2.77 g, 6.4 mmol); white granule; mp 226–228 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 6.10 (1H, s), 6.13 (1H, s), 6.29 (1H, s), 7.13 (1H, s), 7.22 (1H, s), 7.47 (1H, dd, *J* = 8.4, 1.5 Hz), 7.52 (1H, d, *J* = 5.6 Hz), 7.76–7.78 (2H, m), 8.41 (1H, d, *J* = 5.6 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 42.0, 99.5, 102.1, 103.7, 117.3, 121.4, 123.1, 123.3 (q, *J* = 271 Hz), 126.9 (q, *J* = 4 Hz), 127.4, 127.5 (q, *J* = 4 Hz), 130.9 (q, *J* = 33 Hz), 133.8, 135.5, 135.6, 141.6, 149.6, 150.0, 151.1; IR (KBr) 2243 cm⁻¹; ESIMS *m/z* (rel int) 435 (100, [M + 1]⁺), 437 (96, [M + 3]⁺). Anal. Calcd for C₁₉H₁₀BrF₃N₂O₂: C, 52.44; H, 2.32; N, 6.44. Found: C, 52.23; H, 2.27; N, 6.66.

General Procedure for the Preparation of (2-Bromophenyl)-isoquinolin-1-ylmethanone 13. A solution of nitriles **12** (2.00 mmol) in tetrahydrofuran (30 mL) was added to sodium hydride (60% dispersion in oil, 120 mg, 3.00 mmol), which had been washed with hexane, and the mixture was stirred at room temperature for 1 h. Then, a slow stream of oxygen gas was passed through the mixture for 3 h. The resulting solution was quenched with water (30 mL) and extracted with EtOAc (3 × 60 mL). The combined extracts were washed with water, dried with anhydrous MgSO₄, filtered, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel eluting with hexane–EtOAc (5:1) to give the pure (2-bromophenyl)isoquinolin-1-ylmethanones **13**. The full spectral data of **13a–t** are described as follows.

(2-Bromophenyl)isoquinolin-1-ylmethanone (**13a**): yield 95% (593 mg, 1.90 mmol); pale yellow crystal; mp 103–104 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 7.37 (1H, t, *J* = 7.8 Hz), 7.46 (1H, t, *J* = 7.8 Hz), 7.59 (1H, d, *J* = 7.8 Hz), 7.66 (1H, d, *J* = 7.8 Hz), 7.72–7.78 (2H, m), 7.81 (1H, d, *J* = 5.5 Hz), 7.91 (1H, d, *J* = 8.2 Hz), 8.53 (1H, d, *J* = 5.5 Hz), 8.79 (1H, d, *J* = 8.2 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 120.6, 124.0, 126.5, 126.8, 127.1, 127.3, 129.0, 130.6, 131.0, 132.1, 133.2, 137.0, 141.3, 141.4, 154.0, 196.7; IR (KBr) 1667 cm⁻¹; ESIMS *m/z* (rel int) 312 (100, [M + 1]⁺), 314 (94, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₆H₁₁BrNO 312.00185, found 312.00183 [M + 1]⁺.

(2-Bromo-4,5-dimethoxyphenyl)isoquinolin-1-ylmethanone (**13b**): yield 96% (714 mg, 1.92 mmol); pale yellow crystal; mp 109.5–110 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 3.92 (3H, s), 3.93 (3H, s), 7.02 (1H, s), 7.34 (1H, s), 7.71 (1H, t, *J* = 8.2 Hz), 7.76 (1H, t, *J* = 8.2 Hz), 7.80 (1H, d, *J* = 5.6 Hz), 7.91 (1H, d, *J* = 8.2 Hz), 8.54 (1H, d, *J* = 5.6 Hz), 8.55 (1H, d, *J* = 8.2 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.1, 56.3, 113.6, 114.0, 115.9, 123.4, 126.4, 126.7, 127.0, 128.7, 130.5, 132.4, 136.8, 141.3, 148.4, 152.2, 155.4, 195.5; IR (KBr) 1680 cm⁻¹; ESIMS *m/z* (rel int) 372 (100, [M + 1]⁺), 374 (95, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₈H₁₅BrNO₃ 372.0230, found 372.0240 [M + 1]⁺.

(2-Bromo-4,5-methylenedioxyphenyl)isoquinolin-1-ylmethanone (**13c**): yield 81% (577 mg, 1.62 mmol); white granule; mp 131–132 °C (hexane–EtOAc) (lit.⁴⁵ mp 150 °C); ¹H NMR (500 MHz, CDCl₃) δ 6.02 (2H, s), 7.00 (1H, s), 7.17 (1H, s), 7.68 (1H, t, *J* = 8.1 Hz), 7.72 (1H, t, *J* = 8.1 Hz), 7.60 (1H, d, *J* = 5.6 Hz), 7.88 (1H, d, *J* = 8.1 Hz), 8.51 (1H, d, *J* = 5.6 Hz), 8.61 (1H, d, *J* = 8.1 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 102.4, 111.1, 113.4, 113.8, 123.5, 126.3, 126.6, 127.0, 128.7, 130.5, 133.9, 136.8, 141.2, 147.4, 150.8, 154.8, 195.2; IR (KBr) 1665 cm⁻¹; ESIMS *m/z* (rel int) 356 (100, [M + 1]⁺), 358 (92, [M + 3]⁺). Anal. Calcd for C₁₇H₁₀BrNO₃: C, 57.33; H, 2.83; N, 3.93. Found: C, 57.03; H, 2.90; N, 3.61.

(2-Bromo-5-trifluoromethylphenyl)isoquinolin-1-ylmethanone (**13d**): yield 59% (448 mg, 1.18 mmol); white needle; mp 125–126 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 7.61 (1H, d, *J* = 8.3 Hz), 7.73 (1H, d, *J* = 8.3 Hz), 7.77–7.91 (2H, m), 7.85–7.87 (2H, m), 7.94–7.95 (1H, m), 8.53 (1H, d, *J* = 5.5 Hz), 8.90–8.92 (1H, m); ¹³C NMR (125 MHz, CDCl₃) δ 123.6 (q, *J* = 271 Hz), 124.3, 124.7, 126.3, 126.8, 127.2, 127.5 (q, *J* = 3 Hz), 128.1 (q, *J* = 3 Hz), 129.5, 130.0 (q, *J* = 33 Hz), 130.7, 133.7, 137.0, 141.3, 142.4, 152.4, 195.5; IR (KBr) 1682 cm⁻¹; ESIMS *m/z* (rel int) 380 (100, [M + 1]⁺), 382 (99, [M +

3]⁺); HREIMS *m/z* calcd for C₁₇H₁₀BrF₃NO 379.9892, found 379.9883 [M + 1]⁺.

(2-Bromophenyl)(5-methoxyisoquinolin-1-yl)methanone (**13e**): yield 94% (643 mg, 1.88 mmol); pale yellow granule; mp 141–142 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.04 (3H, s), 7.06 (1H, d, *J* = 8.0 Hz), 7.36 (1H, td, *J* = 8.0, 1.7 Hz), 7.45 (1H, td, *J* = 8.0, 1.0 Hz), 7.58 (1H, dd, *J* = 8.0, 1.0 Hz), 7.63 (1H, t, *J* = 8.0 Hz), 7.65 (1H, dd, *J* = 8.0, 1.7 Hz), 8.22 (1H, d, *J* = 5.7 Hz), 8.33 (1H, d, *J* = 8.0 Hz), 8.52 (1H, d, *J* = 5.7 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.6, 107.6, 117.9, 118.4, 120.5, 127.2, 127.4, 129.1, 129.7, 130.9, 131.9, 133.0, 140.8, 141.3, 153.2, 154.5, 196.6; IR (KBr) 1682 cm⁻¹; ESIMS *m/z* (rel int) 342 (100, [M + 1]⁺), 344 (90, [M + 3]⁺). Anal. Calcd for C₁₇H₁₂BrNO₂: C, 59.67; H, 3.53; N, 4.09. Found: C, 59.35; H, 3.28; N, 3.75.

(2-Bromo-4,5-dimethoxyphenyl)(5-methoxyisoquinolin-1-yl)methanone (**13f**): yield 84% (675 mg, 1.68 mmol); pale yellow crystal; mp 153–154 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 3.92 (6H, s), 4.04 (3H, s), 7.01 (1H, s), 7.05 (1H, d, *J* = 8.2 Hz), 7.33 (1H, s), 7.59 (1H, t, *J* = 8.2 Hz), 8.09 (1H, d, *J* = 8.2 Hz), 8.20 (1H, d, *J* = 5.7 Hz), 8.53 (1H, d, *J* = 5.7 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.7, 56.1, 56.3, 107.7, 113.6, 114.1, 116.0, 117.8, 118.0, 127.4, 128.8, 129.7, 132.5, 141.0, 148.4, 152.2, 154.6, 154.8, 195.6; IR (KBr) 1678 cm⁻¹; ESIMS *m/z* (rel int) 424 (100, [M + 1]⁺), 426 (99, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₉H₁₆BrNO₄Na 424.0160, found 424.0163 [M + Na]⁺.

(2-Bromo-4,5-methylenedioxypyrenyl)(5-methoxyisoquinolin-1-yl)methanone (**13g**): yield 67% (517 mg, 1.34 mmol); pale yellow crystal, mp 153–154 °C (hexane–EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 4.03 (3H, s), 6.05 (2H, s), 7.01 (1H, s), 7.04 (1H, d, *J* = 8.4 Hz), 7.17 (1H, s), 7.59 (1H, t, *J* = 8.4 Hz), 8.15 (1H, d, *J* = 8.4 Hz), 8.20 (1H, d, *J* = 5.7 Hz), 8.52 (1H, d, *J* = 5.7 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.7, 102.4, 107.7, 111.3, 113.5, 113.9, 118.0, 118.1, 127.5, 128.9, 129.8, 134.1, 141.0, 147.5, 150.9, 154.3, 154.6, 195.4; IR (KBr) 1668 cm⁻¹; ESIMS *m/z* (rel int) 386 (100, [M + 1]⁺), 388 (97, [M + 3]⁺). Anal. Calcd for C₁₈H₁₂BrNO₄: C, 55.98; H, 3.13; N, 3.63. Found: C, 56.28; H, 3.52; N, 3.76.

(2-Bromo-5-trifluoromethylphenyl)(5-methoxyisoquinolin-1-yl)methanone (**13h**): yield 55% (410 mg, 1.10 mmol); pale yellow crystal; mp 137–138 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.05 (3H, s), 7.08 (1H, d, *J* = 8.3 Hz), 7.59 (1H, dd, *J* = 8.3, 1.7 Hz), 7.76 (1H, t, *J* = 8.3 Hz), 7.71 (1H, d, *J* = 8.3 Hz), 7.87 (1H, d, *J* = 1.7 Hz), 8.27 (1H, d, *J* = 5.6 Hz), 8.44 (1H, d, *J* = 8.3 Hz), 8.52 (1H, d, *J* = 5.6 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.8, 107.8, 117.8, 119.2, 123.7 (q, *J* = 271 Hz), 124.3, 127.5 (q, *J* = 4 Hz), 127.6, 128.1 (q, *J* = 3 Hz), 129.7, 129.9 (q, *J* = 33 Hz), 130.0, 133.7, 141.0, 142.6, 151.8, 154.7, 195.6; IR (KBr) 1682 cm⁻¹; ESIMS *m/z* (rel int) 410 (100, [M + 1]⁺), 412 (96, [M + 3]⁺). Anal. Calcd for C₁₈H₁₁BrF₃NO₂: C, 52.71; H, 2.70; N, 3.41. Found: C, 52.70; H, 2.90; N, 3.40.

(2-Bromophenyl)(5,6-dimethoxyisoquinolin-1-yl)methanone (**13i**): yield 89% (662 mg, 1.78 mmol); pale yellow crystal; mp 125–126 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.03 (3H, s), 4.07 (3H, s), 7.37 (1H, td, *J* = 7.9, 1.6 Hz), 7.46 (1H, td, *J* = 7.9, 1.0 Hz), 7.51 (1H, d, *J* = 9.4 Hz), 7.59 (1H, dd, *J* = 7.9, 1.0 Hz), 7.64 (1H, dd, *J* = 7.9, 1.6 Hz), 8.08 (1H, d, *J* = 5.9 Hz), 8.46 (1H, d, *J* = 5.9 Hz), 8.58 (1H, d, *J* = 9.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.5, 61.3, 117.2, 117.7, 120.6, 122.5, 123.5, 127.3, 131.0, 132.0, 133.1, 133.2, 141.2, 141.3, 141.4, 151.3, 153.6, 196.9; IR (KBr) 1682 cm⁻¹; ESIMS *m/z* (rel int) 372 (100, [M + 1]⁺), 374 (100, [M + 3]⁺). Anal. Calcd for C₁₈H₁₄BrNO₃: C, 58.08; H, 3.79; N, 3.76. Found: C, 58.20; H, 4.06; N, 3.77.

(2-Bromo-4,5-dimethoxyphenyl)(5,6-dimethoxyisoquinolin-1-yl)methanone (**13j**): yield 91% (786 mg, 1.82 mmol); pale yellow granule; mp 161–162 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 3.92 (3H, s), 3.93 (3H, s), 4.03 (3H, s), 4.06 (3H, s) 7.02 (1H, s), 7.32 (1H, s), 7.48 (1H, d, *J* = 9.4 Hz), 8.06 (1H, d, *J* = 5.8 Hz), 8.35 (1H, d, *J* = 9.4 Hz), 8.47 (1H, d, *J* = 5.8 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.1, 56.3, 56.5, 61.3, 113.5, 114.0, 116.0, 116.9, 117.1, 122.4, 123.4, 132.4, 133.1, 141.4, 141.5, 148.4, 151.4, 152.2, 155.0, 195.7; IR (KBr) 1667 cm⁻¹; ESIMS *m/z* (rel int) 432 (100, [M +

+ 1]⁺), 434 (95, [M + 3]⁺); HRESIMS *m/z* calcd for C₂₀H₁₉BrNO₅ 432.0441, found 432.0442 [M + 1]⁺.

(2-Bromo-4,5-methylenedioxyphenyl)(5,6-dimethoxyisoquinolin-1-yl)methanone (13k): yield 94% (782 mg, 1.88 mmol); pale yellow granule; mp 190–191 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.02 (3H, s), 4.05 (3H, s), 6.05 (2H, s), 7.02 (1H, s), 7.16 (1H, s), 7.47 (1H, d, *J* = 9.4 Hz), 8.06 (1H, d, *J* = 5.8 Hz), 8.41 (1H, d, *J* = 9.4 Hz), 8.46 (1H, d, *J* = 5.8 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.5, 61.3, 102.4, 111.2, 113.4, 113.8, 117.0, 117.3, 122.4, 123.4, 133.1, 134.0, 141.3, 141.4, 147.4, 150.8, 151.3, 154.5, 195.5; IR (KBr) 1682 cm⁻¹; ESIMS *m/z* (rel int) 416 (100, [M + 1]⁺), 418 (100, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₉H₁₅BrNO₅ 416.0128, found 416.0130 [M + 1]⁺.

(2-Bromo-5-trifluoromethylphenyl)(5,6-dimethoxyisoquinolin-1-yl)methanone (13l): yield 65% (572 mg, 1.30 mmol); pale yellow crystal; mp 163–164 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.03 (3H, s), 4.08 (3H, s), 7.54 (1H, d, *J* = 9.4 Hz), 7.60 (1H, dd, *J* = 8.4, 1.8 Hz), 7.72 (1H, d, *J* = 8.4 Hz), 7.86 (1H, d, *J* = 1.8 Hz), 8.12 (1H, d, *J* = 5.8 Hz), 8.46 (1H, d, *J* = 5.8 Hz), 8.70 (1H, d, *J* = 9.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.5, 61.3, 117.6, 118.4, 122.5, 123.3, 123.5 (*q*, *J* = 271 Hz), 124.3, 127.4 (*q*, *J* = 4 Hz), 128.1 (*q*, *J* = 3 Hz), 129.9 (*q*, *J* = 33 Hz), 133.3, 133.6, 141.3, 141.5, 142.4, 151.4, 152.1, 195.7; IR (KBr) 1690 cm⁻¹; ESIMS *m/z* (rel int) 440 (100, [M + 1]⁺), 442 (98, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₉H₁₄BrF₃NO₃ 440.0104, found 440.0098 [M + 1]⁺.

(2-Bromophenyl)(7-methoxyisoquinolin-1-yl)methanone (13m): yield 88% (602 mg, 1.76 mmol); pale yellow granule; mp 104–105 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.00 (3H, s), 7.36 (1H, t, *J* = 7.9 Hz), 7.42 (1H, d, *J* = 9.0 Hz), 7.45 (1H, t, *J* = 7.9 Hz), 7.59–7.61 (2H, m), 7.75 (1H, d, *J* = 5.4 Hz), 7.81 (1H, d, *J* = 9.0 Hz), 8.26 (1H, s), 8.44 (1H, d, *J* = 5.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.6, 103.5, 120.5, 124.1, 124.2, 127.2, 128.5, 128.6, 130.6, 131.7, 132.9, 133.1, 139.7, 141.8, 151.3, 160.1, 197.4; IR (KBr) 1668 cm⁻¹; ESIMS *m/z* (rel int) 342 (100, [M + 1]⁺), 344 (97, [M + 3]⁺). Anal. Calcd for C₁₇H₁₂BrNO₂: C, 59.67; H, 3.53; N, 4.09. Found: C, 59.66; H, 3.85; N, 3.85.

(2-Bromo-4,5-dimethoxyphenyl)(7-methoxyisoquinolin-1-yl)methanone (13n): yield 96% (772 mg, 1.92 mmol); white granule; mp 158–159 °C (hexane–EtOAc). ¹H NMR (500 MHz, CDCl₃) δ 3.92 (3H, s), 3.93 (3H, s), 3.97 (3H, s), 7.03 (1H, s), 7.28 (1H, s), 7.41 (1H, d, *J* = 9.0 Hz), 7.73 (1H, d, *J* = 5.3 Hz), 7.81 (1H, d, *J* = 9.0 Hz), 7.99 (1H, s), 8.45 (1H, d, *J* = 5.3 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.5, 56.1, 56.3, 103.5, 113.3, 113.8, 115.9, 123.5, 124.1, 128.3, 128.5, 132.8, 132.9, 139.6, 148.3, 151.9, 152.8, 159.7, 196.2; IR (KBr) 1667 cm⁻¹; ESIMS *m/z* (rel int) 402 (100, [M + 1]⁺), 404 (95, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₉H₁₇BrNO₄ 402.03355, found 402.03348 [M + 1]⁺.

(2-Bromo-4,5-methylenedioxyphenyl)(7-methoxyisoquinolin-1-yl)methanone (13o): yield 88% (679 mg, 1.76 mmol); pale yellow crystal; mp 146–147 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 3.97 (3H, s), 6.05 (2H, s), 7.03 (1H, s), 7.12 (1H, s), 7.40 (1H, d, *J* = 9.0, 2.4 Hz), 7.73 (1H, d, *J* = 5.4 Hz), 7.80 (1H, d, *J* = 9.0 Hz), 8.07 (1H, d, *J* = 2.4 Hz), 8.44 (1H, d, *J* = 5.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.6, 102.4, 103.5, 111.0, 113.4, 113.6, 123.7, 124.1, 128.4, 128.6, 132.8, 134.5, 139.6, 147.4, 150.6, 152.2, 159.9, 196.0; IR (KBr) 1668 cm⁻¹; ESIMS *m/z* (rel int) 386 (100, [M + 1]⁺), 388 (95, [M + 3]⁺). Anal. Calcd for C₁₈H₁₂BrNO₄: C, 55.98; H, 3.13; N, 3.63. Found: C, 56.07; H, 3.22; N, 3.44.

(2-Bromo-5-trifluoromethylphenyl)(7-methoxyisoquinolin-1-yl)methanone (13p): yield 74% (607 mg, 1.48 mmol); pale yellow granule; mp 121–122 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 4.02 (3H, s), 7.44 (1H, dd, *J* = 9.1, 2.0 Hz), 7.59 (1H, dd, *J* = 8.4, 2.0 Hz), 7.73 (1H, d, *J* = 8.4 Hz), 7.78 (1H, d, *J* = 5.3 Hz), 7.81 (1H, d, *J* = 2.0 Hz), 7.83 (1H, d, *J* = 9.1 Hz), 8.39 (1H, d, *J* = 2.0 Hz), 8.44 (1H, d, *J* = 5.3 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 55.6, 103.4, 123.6 (*q*, *J* = 271 Hz), 124.2, 124.3, 124.8, 127.1 (*q*, *J* = 4 Hz), 127.8 (*q*, *J* = 3 Hz), 128.7, 128.8, 129.8 (*q*, *J* = 33 Hz), 133.0, 133.6, 139.7, 143.0, 149.7, 160.6, 196.2; IR (KBr) 1680 cm⁻¹; ESIMS *m/z* (rel int) 410 (100, [M + 1]⁺), 412 (93, [M + 3]⁺). Anal. Calcd for

C₁₈H₁₁BrF₃NO₂: C, 52.71; H, 2.70; N, 3.41. Found: C, 52.92; H, 2.79; N, 3.67.

(2-Bromophenyl)(6,7-methylenedioxyisoquinolin-1-yl)methanone (13q): yield 92% (655 mg, 1.84 mmol); pale yellow crystal; mp 182–183 °C (hexane–EtOAc) (lit.⁴⁶ mp 173–175 °C); ¹H NMR (500 MHz, CDCl₃) δ 6.14 (2H, s), 7.14 (1H, s), 7.35 (1H, td, *J* = 7.7, 1.7 Hz), 7.44 (1H, t, *J* = 7.7 Hz), 7.58 (1H, d, *J* = 7.7 Hz), 7.60 (1H, dd, *J* = 7.7, 1.7 Hz), 7.62 (1H, d, *J* = 5.4 Hz), 8.21 (1H, s), 8.37 (1H, d, *J* = 5.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 101.9, 102.5, 102.7, 120.4, 123.5, 124.7, 127.2, 130.6, 131.7, 133.0, 135.9, 140.7, 141.7, 150.3, 150.9, 151.5, 197.3; IR (KBr) 1680 cm⁻¹; ESIMS *m/z* (rel int) 356 (100, [M + 1]⁺), 358 (96, [M + 3]⁺). Anal. Calcd for C₁₇H₁₀BrNO₃: C, 57.33; H, 2.83; N, 3.93. Found: C, 57.55; H, 3.09; N, 3.67.

(2-Bromo-4,5-dimethoxyphenyl)(6,7-methylenedioxyisoquinolin-1-yl)methanone (13r): yield 79% (657 mg, 1.58 mmol); white granule; mp 169–170 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 3.92 (6H, s), 6.14 (2H, s), 7.01 (1H, s), 7.14 (1H, s), 7.28 (1H, s), 7.61 (1H, d, *J* = 5.5 Hz), 7.94 (1H, s), 8.38 (1H, d, *J* = 5.5 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.1, 56.3, 101.8, 102.4, 102.7, 113.2, 113.9, 115.9, 123.0, 124.5, 132.8, 135.7, 140.7, 148.4, 149.9, 151.0, 151.9, 153.0, 196.1; IR (KBr) 1674 cm⁻¹; ESIMS *m/z* (rel int) 416 (100, [M + 1]⁺), 418 (93, [M + 3]⁺); HRESIMS *m/z* calcd for C₁₉H₁₅BrNO₅ 416.0128, found 416.0124 [M + 1]⁺.

(2-Bromo-4,5-methylenedioxyphenyl)(6,7-methylenedioxyisoquinolin-1-yl)methanone (13s): yield 94% (752 mg, 1.88 mmol); pale yellow needle; mp 220–222 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 6.05 (2H, s), 6.13 (2H, s), 7.01 (1H, s), 7.12 (1H, s), 7.13 (1H, s), 7.61 (1H, d, *J* = 5.5 Hz), 8.02 (1H, s), 8.37 (1H, d, *J* = 5.5 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 101.9, 102.4, 102.5, 102.7, 111.0, 113.3, 113.5, 123.2, 124.6, 134.4, 135.8, 140.7, 147.4, 150.1, 150.6, 151.0, 152.4, 195.9; IR (KBr) 1667 cm⁻¹; ESIMS *m/z* (rel int) 400 (100, [M + 1]⁺), 402 (94, [M + 3]⁺). Anal. Calcd for C₁₈H₁₀BrNO₅: C, 54.02; H, 2.52; N, 3.50. Found: C, 53.90; H, 2.70; N, 3.41.

(2-Bromo-5-trifluoromethylphenyl)(6,7-methylenedioxyisoquinolin-1-yl)methanone (13t): yield 53% (449 mg, 1.06 mmol); pale yellow granule; mp 175–176 °C (hexane–EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 6.16 (2H, s), 7.15 (1H, s), 7.58 (1H, dd, *J* = 8.4, 2.1 Hz), 7.65 (1H, d, *J* = 5.4 Hz), 7.71 (1H, d, *J* = 8.4 Hz), 7.82 (1H, d, *J* = 2.1 Hz), 8.33 (1H, s), 8.36 (1H, d, *J* = 5.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 102.0, 102.3, 102.8, 123.6, 124.1, 124.2, 124.9, 127.1 (*q*, *J* = 4 Hz), 127.9 (*q*, *J* = 3 Hz), 129.9 (*q*, *J* = 33 Hz), 133.5, 136.1, 140.7, 142.9, 150.0, 150.7, 151.1, 196.0; IR (KBr) 1682 cm⁻¹; ESIMS *m/z* (rel int) 424 (89, [M + 1]⁺), 426 (89, [M + 3]⁺). Anal. Calcd for C₁₈H₉BrF₃NO₃: C, 50.97; H, 2.14; N, 3.30. Found: C, 51.17; H, 2.07; N, 3.22.

General Procedure for the Reductive Photocyclization of 1-(2-Bromobenzoyl)isoquinolines 13. NaBH₄ (0.750 mmol) was carefully added to a stirred solution of the bromo ketones 13 (0.500 mmol) in hot methanol (200 mL). After cooling, the resulting solution was irradiated in a quartz reactor equipped with a medium pressure mercury lamp (450 W) at room temperature. The reactions were monitored using TLC. Water (5 mL) was added until TLC showed the reaction to be complete. The solution was then evaporated, and the residue was dissolved in CHCl₃ (400 mL). The solution was washed with water (3 × 50 mL). The organic phase was dried with anhydrous MgSO₄, filtered, and evaporated under reduced pressure. The residue was purified by column chromatography over silica gel and eluted with CHCl₃–MeOH (20:1) to yield the pure dibenzo-[de,g]quinolin-7-ones 8. The full spectral data of dibenzo-[de,g]-quinolin-7-ones 8a–t are described as follows.

Dibenzo[de,g]quinolin-7-one (8a): yield 55% (63.5 mg, 0.275 mmol); yellow solid; mp 222–224 °C (lit.⁴⁷ mp 222–222.5 °C); ¹H NMR (600 MHz, CDCl₃) δ 7.60 (1H, t, *J* = 7.8 Hz), 7.78 (1H, td, *J* = 7.8, 1.2 Hz), 7.87 (1H, t, *J* = 7.8 Hz), 7.98–8.00 (2H, m), 8.30 (1H, d, *J* = 7.8 Hz), 8.45 (1H, d, *J* = 7.8 Hz), 8.56 (1H, dd, *J* = 7.8, 1.2 Hz), 9.06 (1H, d, *J* = 5.4 Hz); ¹³C NMR (150 MHz, CDCl₃) δ 123.1, 124.5, 124.6, 125.1, 128.6, 128.7, 129.1, 129.2, 131.0, 131.6, 134.1, 134.9, 136.9, 145.5, 146.7, 182.7; IR (KBr) 1665 cm⁻¹; EIMS *m/z* (rel int)

¹³C NMR (125 MHz, CDCl₃) δ 56.6, 111.4, 119.4, 123.8 (q, *J* = 271 Hz), 125.1, 125.6 (q, *J* = 4 Hz), 126.3, 129.4 (q, *J* = 33 Hz), 129.5, 129.6 (q, *J* = 3 Hz), 131.5, 131.9, 132.2, 137.4, 143.8, 144.9, 159.5, 181.3; IR (KBr) 1659 cm⁻¹; EIMS *m/z* (rel int) 324 (100, M⁺); HREIMS *m/z* calcd for C₁₈H₁₀F₃NO₂ 329.0664, found 329.0672 [M]⁺.

1,2-(Methylenedioxy)dibenzo[*d,e,g*]quinolin-7-one (8q): yield 45% (61.9 mg, 0.225 mmol); yellow solid; mp 279–281 °C (lit.⁴⁸ mp 280–282 °C); ¹H NMR (500 MHz, CDCl₃) δ 6.34 (2H, s), 7.14 (1H, s), 7.54 (1H, t, *J* = 7.2 Hz), 7.70 (1H, t, *J* = 7.2 Hz), 7.73 (1H, d, *J* = 5.4 Hz), 8.55 (1H, d, *J* = 7.2 Hz), 8.59 (1H, d, *J* = 7.2 Hz), 8.85 (1H, d, *J* = 5.4 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 102.4, 103.2, 108.1, 123.2, 124.2, 127.3, 128.6, 128.8, 131.3, 132.8, 133.9, 135.7, 144.9, 145.4, 147.9, 151.7, 182.4; IR (KBr) 1651 cm⁻¹; EIMS *m/z* (rel int) 275 (100, M⁺); HREIMS *m/z* calcd for C₁₇H₉NO₃ 275.0582, found 275.0579 [M]⁺.

Dicentrinone (8r): yield 81% (135.7 mg, 0.405 mmol); yellow solid; mp 300 °C dec (lit.⁴⁹ mp 300 °C dec); ¹H NMR (500 MHz, CDCl₃) δ 3.95 (3H, s), 4.06 (3H, s), 6.34 (2H, s), 7.06 (1H, s), 7.70 (1H, d, *J* = 5.2 Hz), 7.86 (1H, s), 7.92 (1H, s), 8.86 (1H, d, *J* = 5.2 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 56.1, 56.2, 102.4, 102.7, 108.2, 108.8, 109.5, 122.6, 124.0, 125.9, 127.7, 135.5, 144.7, 145.4, 147.0, 149.5, 151.5, 153.8, 181.2; IR (KBr) 1643 cm⁻¹; EIMS *m/z* (rel int) 335 (100, M⁺); HREIMS *m/z* calcd for C₁₉H₁₃NO₅ 335.0794, found 335.0790 [M]⁺.

Cassameridine (8s): yield 53% (84.5 mg, 0.265 mmol); brown solid; mp 320 °C dec (lit.⁵⁰ mp 300 °C); ¹H NMR (500 MHz, CDCl₃ + CF₃CO₂D + D₂O) δ 6.22 (2H, s), 6.60 (2H, s), 7.41 (1H, s), 7.77 (1H, s), 8.06 (1H, s), 8.30 (1H, d, *J* = 5.8 Hz), 8.94 (1H, d, *J* = 5.8 Hz); ¹³C NMR (125 MHz, CDCl₃ + CF₃CO₂D) δ 103.4, 103.5, 105.0, 107.5, 107.6, 108.6, 121.6, 124.7, 126.6, 130.1, 134.1, 135.1, 142.9, 150.0, 151.6, 155.9, 158.3, 174.4; IR (KBr) 1682 cm⁻¹; EIMS *m/z* (rel int) 319 (100, M⁺); HREIMS *m/z* calcd for C₁₈H₉NO₅ 319.0481, found 319.0488 [M]⁺.

9-Trifluoromethyl-1,2-(methylenedioxy)dibenzo[*d,e,g*]quinolin-7-one (8t): yield 72% (123.5 mg, 0.360 mmol); yellow solid; mp 350 °C dec; ¹H NMR (600 MHz, CDCl₃) δ 6.44 (2H, s), 7.25 (1H, s), 7.81 (1H, d, *J* = 5.4 Hz), 7.93 (1H, d, *J* = 8.4 Hz), 8.75 (1H, d, *J* = 8.4 Hz), 8.92 (1H, d, *J* = 5.4 Hz); ¹³C NMR (150 MHz, CDCl₃) δ 102.9, 104.3, 106.7, 123.3, 123.6 (q, *J* = 270 Hz), 124.7, 125.9, 128.0, 129.9, 130.3 (q, *J* = 33 Hz), 131.2, 135.6, 135.7, 144.8, 145.3, 149.0, 151.8, 181.4; IR (KBr) 1667 cm⁻¹; EIMS *m/z* (rel int) 343 (100, M⁺); HREIMS *m/z* calcd for C₁₈H₈F₃NO₃ 343.0456, found 343.0459 [M]⁺.

ASSOCIATED CONTENT

Supporting Information

Copies of the ¹H and ¹³C NMR spectra of compounds 8, 10, 12, and 13. This material is available free of charge via the Internet at <http://pubs.acs.org>.

AUTHOR INFORMATION

Corresponding Author

*E-mail: thchuang@mail.cmu.edu.tw.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

Financial support from the National Science Council of the Republic of China (NSC 101-2113-M-039-004-MY2) and China Medical University (CMU101-S-08) is gratefully acknowledged.

REFERENCES

- Bick, I. R. C.; Clezy, P. S.; Crow, W. D. *Aust. J. Chem.* **1956**, 9, 111–119.
- Costa, E. V.; Pinheiro, M. L. B.; Xavier, C. M.; Silva, J. R. A.; Amaral, A. C. F.; Souza, A. D. L.; Barison, A.; Campos, F. R.; Ferreira, A. G.; Machado, G. M. C.; Leon, L. L. P. *J. Nat. Prod.* **2006**, 69, 292–294.
- Zhang, Z.; ElSohly, H. N.; Jacob, M. R.; Pasco, D. S.; Walker, L. A.; Clark, A. M. *J. Nat. Prod.* **2002**, 65, 856–859.
- Rahman, M. M.; Lopa, S. S.; Sadik, G.; Rashid, H.; Islam, R.; Khondkar, P.; Alam, A. H. K.; Rashid, M. A. *Fitoterapia* **2005**, 76, 758–761.
- Montanha, J. A.; Amoros, M.; Boustie, J.; Girre, L. *Planta Med.* **1995**, 61, 419–424.
- Boustie, J.; Stigliani, J. L.; Montanha, J.; Amoros, M.; Payard, M.; Girre, L. *J. Nat. Prod.* **1998**, 61, 480–484.
- Zhang, C. F.; Nakamura, N.; Tewtrakul, S.; Hattori, M.; Sun, Q. S.; Wang, Z. T.; Fujiwara, T. *Chem. Pharm. Bull.* **2002**, 50, 1195–1200.
- Chen, J. J.; Chang, Y. L.; Teng, C. M.; Chen, I. S. *Planta Med.* **2001**, 67, 593–598.
- Chen, K. S.; Ko, F. N.; Teng, C. M.; Wu, Y. C. *Planta Med.* **1996**, 62, 133–136.
- Chang, F. R.; Wei, J. L.; Teng, C. M.; Wu, Y. C. *J. Nat. Prod.* **1998**, 61, 1457–1461.
- Chen, C. Y.; Chang, F. R.; Teng, C. M.; Wu, Y. C. *J. Chin. Chem. Soc.* **1999**, 46, 77–86.
- Chen, J. J.; Chang, Y. L.; Teng, C. M.; Chen, I. S. *Planta Med.* **2000**, 66, 251–256.
- Chang, F. R.; Hwang, T. L.; Yang, Y. L.; Li, C. E.; Wu, C. C.; Issa, H. H.; Hsieh, W. B.; Wu, Y. C. *Planta Med.* **2006**, 72, 1344–1347.
- Sonnet, P. E.; Jacobson, M. *J. Pharm. Sci.* **1971**, 60, 1254–1256.
- Chen, J. J.; Ishikawa, T.; Duh, C. Y.; Tsai, I. L.; Chen, I. S. *Planta Med.* **1996**, 62, 528–533.
- Woo, S. H.; Reynolds, M. C.; Sun, N. J.; Cassady, J. M.; Snapka, R. M. *Biochem. Pharmacol.* **1997**, 54, 467–473.
- Chen, K. S.; Chang, F. R.; Chia, Y. C.; Wu, T. S.; Wu, Y. C. *J. Chin. Chem. Soc.* **1998**, 45, 103–110.
- Chen, S. B.; Gao, G. Y.; Li, Y. S.; Yu, S. C.; Xiao, P. G. *Planta Med.* **2002**, 68, 554–556.
- Wirasathien, L.; Boonarkart, C.; Pengsuparp, T.; Suttsiri, R. *Pharm. Biol.* **2006**, 44, 274–278.
- Wu, Y. C.; Liou, Y. F.; Lu, S. T.; Chen, C. H.; Chang, J. J.; Lee, K. H. *Planta Med.* **1989**, 55, 163–165.
- Zhao, Q. Z.; Zhao, Y. M. *Nat. Prod. Res. Dev.* **2006**, 18, 316–324.
- Shamma, M. *The Isoquinoline Alkaloids*; Academic Press: New York, 1972.
- Shamma, M.; Castenson, R. L. In *The Alkaloids: Chemistry and Physiology*; Manske, R. H. F., Holmes, H. L., Eds.; Academic Press: New York, 1973; Vol. 14, pp 225–264.
- Shamma, M.; Moniot, J. M. *Isoquinoline Alkaloid Research 1972–1977*; Plenum Press: New York, 1978.
- Grundon, M. F.; Shamma, M. In *The Alkaloids*; Grundon, M. F., Eds.; London, 1981; Vol. 10, pp 126–134.
- Liu, Y. C.; Chen, Z. F.; Peng, Y.; Liang, H. *Linchan Huaxue Yu Gongye* **2011**, 31, 109–116.
- Shamma, M.; Guinaudeau, H. *Tetrahedron* **1984**, 40, 4795–4822.
- Singh, O. V.; Huang, W. J.; Chen, C. H.; Lee, S. S. *Tetrahedron Lett.* **2007**, 48, 8166–8169.
- Taylor, W. I. *Tetrahedron* **1961**, 14, 42–45.
- Castedo, L.; Saá, J. M.; Suau, R.; Villaverde, C. *Heterocycles* **1980**, 14, 1131–1134.
- Orito, K.; Uchiito, S.; Satoh, Y.; Tatsuzawa, T.; Harada, R.; Tokuda, M. *Org. Lett.* **2000**, 2, 307–310.
- Gerecke, M.; Brossi, A. *Helv. Chim. Acta* **1979**, 62, 1549–1558.
- Seá, C.; Guitián, E.; Castedo, L.; Seá, J. M. *Tetrahedron Lett.* **1985**, 26, 4559–4560.
- Nicolaides, D. N.; Papageorgiou, G. K.; Stephanidou-Stephanatou, J. *Tetrahedron Lett.* **1989**, 45, 4585–4592.
- Skiles, J. W.; Saa, J. M.; Cava, M. P. *Can. J. Chem.* **1979**, 57, 1642–1646.
- Kessar, S. V.; Gupta, Y. P.; Gupta, V. S.; Narula, M.; Mohammad, T. *Tetrahedron Lett.* **1980**, 21, 3307–3308.

- (37) Chuang, T. H.; Wu, P. L. *J. Chin. Chem. Soc.* **2006**, *53*, 413–420.
- (38) Krasovskiy, A.; Malakhov, V.; Gavryushin, A.; Knochel, R. *Angew. Chem., Int. Ed.* **2006**, *45*, 6040–6044.
- (39) Ikehara, M. *Pharm. Bull.* **1954**, *2*, 111–114.
- (40) Kharasch, M. S.; Sosnovsky, G. *Tetrahedron* **1958**, *3*, 97–104.
- (41) Tiner-Harding, T.; Mariano, P. S. *J. Org. Chem.* **1982**, *47*, 482–485.
- (42) Costanza, C.; Lenz, G. R.; Lessor, R. A. *Heterocycles* **1992**, *34*, 465–478.
- (43) Bissember, A. C.; Banwell, M. G. *J. Org. Chem.* **2009**, *74*, 4893–4895.
- (44) Ochiai, E.; Kuniyoshi, I. *Pharm. Bull.* **1957**, *5*, 292–297.
- (45) Hung, T. V.; Mooney, B. A.; Prager, R. H.; Tippett, J. M. *Aust. J. Chem.* **1981**, *34* (2), 383–395.
- (46) Gupta, Y. P.; Yadav, V. S.; Mohammad, T. *Indian J. Chem. Sect. B: Org. Chem. Incl. Med. Chem.* **1983**, *22*, 429–431.
- (47) Cannon, J. G.; Kim, J. C.; Aleem, M. A. *J. Heterocycl. Chem.* **1972**, *9*, 731–733.
- (48) Wang, H. M.; Yang, W. L.; Yang, S. C.; Chen, C. Y. *Chem. Nat. Compd.* **2011**, *47*, 316–318.
- (49) Castedo, L.; Saá, C.; Saá, J. M.; Suau, R. *J. Org. Chem.* **1982**, *47*, 513–517.
- (50) Cava, M. P.; Rao, K. V. *J. Org. Chem.* **1968**, *33*, 2443–2446.