Chemical transformations of 3-aminopyrrolidin-2-ones of the norbornane and cyclopropane series*

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3-Aminopyrrolidin-2-ones containing a fused norbornane or spirocyclopropane fragment react with benzaldehyde to give azomethines, which can be transformed into N-substituted 3-aminopyrrolidin-2-ones by reduction with sodium borohydride. Diazotization of amino-pyrrolidinones with NaNO₂ in acetic acid results in the elimination of molecular nitrogen and in the formation of acetoxy or unsaturated derivatives (through stabilization of intermediate carbocations). 6-Methylidene-4-azaspiro[2.4]heptan-5-one obtained by diazotization of 6-amino-6-methyl-4-azaspiro[2.4]heptan-5-one easily reacts with diazomethane, diazocyclo-propane, and benzonitrile oxide to yield heterocyclic spiranes by means of 1,3-dipolar cyclo-addition.

Key words: 3-aminopyrrolidin-2-ones, spirocyclopropane-containing azaheterocycles, azomethines, diazotization, 1,3-dipolar addition.

In the last few years, active interest has been shown in substituted 3-aminopyrrolidin-2-ones as promising biologically active compounds or synthons for target modification of this heterocyclic fragment.^{1–5} For instance, 5-amino-*exo*-3-azatricyclo[5.2.1.0^{2,6}]decan-4-one (1) containing a 3-aminopyrrolidin-2-one fragment (obtained by catalytic hydrogenation of a methyl diazoacetate— norbornene adduct with Raney nickel²) has high anti-arrhythmic and antifibrillatory effects on simulated arrhythmias of different types. Along with aminopyrrolidin-2-ones, their 5-spirocyclopropane-containing analogs have also been tested for biological activity.^{6,7}

Recently, we developed a method for the synthesis of pyrrolidin-2-ones containing the amino group and the spirocyclopropane fragment in positions 3 and 5.⁸ A general route to these compounds, in particular, 2-oxospiro-[pyrrolidine-5,1'-cyclopropanes] **2** and **3** (Scheme 1), involves catalytic hydrogenation of spiro[1-pyrazoline-5,1'-cyclopropane]-3-carboxylates or the corresponding amides (Raney Ni, H₂, 80–90 bar, 60–70 °C). As with the formation of compound **1**, the reaction occurs by

i

Scheme 1



Н

 NH_2



Reagents and conditions: *i*. H₂, Ni–Ra, 110 bar, 25–40 °C; *ii*. H₂, Ni–Ra, 80–90 bar, 60–70 °C.

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complete reduction of the pyrazolines to diamines followed by intramolecular cyclocondensation of the ester or amide fragment at the distant amino group.

In the present work, we studied some chemical transformations of 3-aminopyrrolidin-2-ones 1-3 mainly occurring at the amino group and obtained new spiro structures with the retained cyclopropane fragment.

Results and Discussion

First of all, we tried to obtain azomethines under mild conditions and reduce them to substituted amines. A similar approach has been employed earlier¹ for the synthesis of substituted 3-aminopyrrolidin-2-ones acting as farnesyltransferase inhibitors. We found that condensation of 3-aminopyrrolidin-2-one **2** or **3** with benzaldehyde in boiling CH₂Cl₂ in the presence of anhydrous MgSO₄ easily leads to azomethine **4** or **5**, respectively, in 78–85% yields (Scheme 2). These azomethines can be smoothly reduced with NaBH₄ in methanol to substituted 3-aminopyrrolidin-2-ones **6** and **7** in high yields, the other functional groups being intact.

Scheme 2



Reagents: i. MgSO₄, CH₂Cl₂; ii. NaBH₄, MeOH

2, **4**, **6**: $\mathbb{R}^1 = Me$, $\mathbb{R}^2 = H$ **3**, **5**, **7**: $\mathbb{R}^1 = H$, $\mathbb{R}^2 = CONHC_6H_4F$

The ¹H and ¹³C NMR spectra of azomethines **4** and **5** show signals at $\delta_{\rm H}$ 8.3–8.4 and $\delta_{\rm C}$ 158–164 (N=CH). Instead, the ¹H NMR spectra of reduction products **6** and **7** contain two doublets at $\delta_{\rm H}$ 3.8–4.0 (geminal coupling constant J = 12-13 Hz) and a broadened singlet at $\delta_{\rm H}$ 2.0–2.5 and their ¹³C NMR spectra exhibit a signal at $\delta_{\rm C}$ 48–51 (NH–CH₂). The NOESY experiment for compound **5** revealed a sufficiently strong coupling between the H(7) proton and either a cyclopropane proton or the amide proton of the aliphatic fragment, a coupling between the H(6) proton and the N=CH proton, and a very weak coupling between the vicinal H(6) and H(7) protons, which suggests the transoid arrangement of the substituents in pyrrolidinone 5 and, consequently, in the starting and substituted amines 3 and 7.

Polycyclic amine **1**, in which the *anti*-isomer is dominant (3.2 : 1), reacts with benzaldehyde or 2,4-dimethoxybenzaldehyde in a similar way. Although both isomers are involved in the reaction, double recrystallization from benzene—ethyl acetate (1 : 2) gives the individual *anti*isomers of azomethines **8a,b** in 62—68% yields (Scheme 3). In the ¹H NMR spectra of compounds **8a,b**, the signal for the methine H(5) proton is shifted downfield by ~0.5 ppm compared to the analogous signal for *anti*-1 but the coupling constant ($J_{5,6} \approx 3.5$ Hz) has the same order of magnitude,² which suggests the identical arrangement of the substituents in these compounds.

Scheme 3



Reagents and conditions: ArCHO, MgSO₄, CH₂Cl₂, 40 °C, 3 h.

Ar = Ph (**a**), 2,4-C₆H₃(OMe)₂ (**b**)

Acylation of pyrrolidinones 1 and 3 with acetyl chloride in triethylamine involves only the amino group, giving the corresponding *N*-acetyl derivatives 9 and 10in good yields (Scheme 4). As expected, the ratio of the

Scheme 4



Reagents and conditions: AcCl, Et₃N, CH₂Cl₂, 10 °C.

anti- and *syn*-isomers of product **9** is the same as in the starting pyrrolidinone **1** (~3.2 : 1). The major, *anti*-isomer was isolated in the individual state by preparative TLC; in its ¹H NMR spectrum, the coupling constant for the vicinal H(5) and H(6) protons has the same order of magnitude as in the starting *anti*-**1** (J = 4.4 Hz).

We studied transformations of compounds 1-3 under nitrosation conditions. It is known⁹ that 3-aminopyrrolin-2-ones, which contain a double bond in the heterocycle, can be diazotized with NaNO₂ in acetic acid to give 3-diazo-4-pyrrolin-2-ones as stable crystalline compounds. However, our attempted synthesis of diazo compounds from aminopyrrolidinones 1-3 failed under analogous conditions. It turned out that diazonium ions generated in the diazotization of compounds 1-3 with NaNO₂ in acetic acid at 10 °C tend to lose a nitrogen molecule, giving the corresponding carbocations rather than cyclic diazo amides. Further stabilization of these carbocations depends on the structure of the starting aminopyrrolidinone, mainly proceeding through their deprotonation or addition of an acetate anion. For instance, diazotization of amine 1 gives a mixture of products, isomeric anti- and syn-3-acetoxypyrrolidinones 11 being dominant. Their total yield is 50-55%. According to the ¹H NMR data, the ratio of the *anti*- and *syn*isomers of compound 11 is $\sim 1.6:1$ (for the starting aminopyrrolidinone 1, 3.2:1), which agrees with the carbocationic reaction mechanism (Scheme 5). Using preparative TLC, we isolated a fraction containing antiand syn-11 in a ratio of ~6: 1 and a fraction with a somewhat lower $R_{\rm f}$ value. The latter fraction contains syn-11 and probably a number of unsaturated compounds resulting from carbocationic rearrangements (the presence of four signals with different intensities at δ 5.6–6.2 in the

Scheme 5



Reagents and conditions: AcOH, NaNO₂, 10 °C.

¹H NMR spectrum); their structures were not determined. The ratio of the total integral intensity of the signals at δ 5.6–6.2 to the intensity of the signal at δ 5.30 for the single H(5) proton in the isomer *syn*-**11** is ~2.6 : 1.

Diazotization of aminopyrrolidinone **3** under similar conditions gives acetoxy derivative **12** and dihydropyrrolone **13** as major products in a total yield of 70–73% (Scheme 5). Compounds **12** and **13** were identified from the ¹H and ¹³C NMR spectra of the reaction mixture. The only isomer obtained of compound **12** seems to be a *trans*-isomer since the coupling constants for the protons of the heterocycle ($J_{6,7} = 7.3$ Hz) are close to those for compound **10** ($J_{6,7} = 8.4$ Hz).

In contrast to amines 1 and 3, nitrosation of amine 2 with NaNO₂ in acetic acid was more efficient, being accompanied only by deprotonation of the intermediate carbocation leading to a difficult-to-separate mixture of unsaturated compounds 14 and 15 (Scheme 6). The ratio of these isomers varies widely with the solvent and the acidity of the medium; however, compound 14 containing the endocyclic double bond is always dominant (Table 1).



Methylidenepyrrolidinone **15** is of interest for the synthesis of heterocyclic spiranes (*e.g.*, by analogy with 1,3-dipolar cycloaddition to 1-substituted 5,5-dimethyl-3-methylidenepyrrolidin-2-one^{10,11}). Its highest yield (up to 20%, ¹H NMR data) was achieved in a nonpolar solvent (CCl₄) at 30 °C (see Table 1). Because isomers **14** and **15** are difficult to separate, their further transformations were studied without separation of their reaction mixture.

It turned out that the action of diazomethane or diazocyclopropane (generated by decomposition of *N*-cyclo-

 Table 1. Yields and ratios of compounds 14 and 15 for different solvents and reaction conditions

Reagents and solvents	<i>T</i> /°C	Total yield (%)	Ratio of 14 and 15
NaNO ₂ /AcOH	5	80	8:1
$NaNO_2/(CO_2H)_2/$	5	75	10:1
Me ₂ CO-H ₂ O			
NaNO ₂ /AcOH/CCl ₄	30	88	3.5:1
BuONO/CH ₂ Cl ₂ /	5	73	6:1
Me ₃ COOH			
N ₂ O ₃ /CHCl ₃	15	48	>50:1

propyl-*N*-nitrosourea in the presence of Cs_2CO_3) on a mixture of compounds **14** and **15** (3.5 : 1) results in 1,3-dipolar cycloaddition involving only isomer **15**. In both cases, the reaction proceeds regioselectively to give di- and trispyro adducts **16** and **17** in high yields, the conversion of 3-methylidenepyrrolidin-2-one (**15**) being complete (Scheme 7). Products **16** and **17** can be easily separated from the unreacted dihydropyrrolone **14** by preparative TLC.

Scheme 7



Reagents and conditions: *i*. CH_2N_2 , Et_2O/CH_2Cl_2 ; *ii*. PhCCl=NOH, NEt₃, CH_2Cl_2 ; *iii*. \longrightarrow N(NO)CONH₂, Cs_2CO_3 , CH_2Cl_2 , 5 °C.

In the ¹³C NMR spectra, the signals at δ 37.4 and 97.3 are due to the C(3) and C(5) spiro atoms in compound **16** and the signals at δ 37.3, 95.9, and 70.3 are due to the C(3), C(5), and C(7) spiro atoms in compound **17**, respectively.

Nitrile oxide generated from benzohydroximoyl chloride and triethylamine in a mixture of compounds **14** and **15** also reacts only with the latter to give cycloadduct **18** in high yield (Scheme 7). As expected, the only one isomer was obtained, which agrees with the literature data on 1,3-dipolar cycloaddition of benzonitrile oxide to structurally similar 5,5-dimethyl-3-methylidenepyrrolidin-2-one.¹⁰ The resulting dispirane **18** can easily be isolated by precipitation from methanol at -15 °C. In the ¹³C NMR spectrum of this compound, the signals at δ 35.3 and 87.9 are due to the C(3) and C(5) spiro atoms.

Dihydropyrrol-2-one 14, which was isolated in the individual state from its mixtures with pyrazolines 16 and 17 by preparative TLC also reacts with benzohydroximoyl chloride in the presence of triethylamine in boiling benzene for 12 h. However, the reaction occurs by acylation of the N—H bond of dihydropyrrolone 14 rather than by

1,3-dipolar cycloaddition of benzonitrile oxide to the endocyclic double bond. With an equimolar ratio of the reagents, the conversion of compound 14 is $\sim 50\%$ and the yield of product 19 is $\sim 40\%$ (Scheme 8). According to the NOESY data, the hydroxy and phenyl groups in oxime 19 are on the same side of the double bond.

Scheme 8



19 (40%)

Reagents and conditions: PhCCl=NOH/NEt₃, C₆H₆, 80 °C.

Thus, we demonstrated that 3-aminopyrrolidin-2-ones containing a polycyclic or spirocyclopropane fragment can be used as new structural units for the targeted synthesis of their *N*-substituted derivatives and heterocyclic di- and trispyrans containing a cyclopropane fragment, which are of interest as potential biologically active compounds and as structural fragments of more complicated polycyclic systems.

Experimental

¹H and ¹³C NMR spectra were recorded on a Bruker AVANCE II 300 spectrometer (300 and 75.5 MHz, respectively) for solutions in CDCl₃, (CD₃)₂CO, or (CD₃)₂SO with 0.05% Me₄Si as the internal standard. Mass spectra were measured on a Finnigan MAT INCOS-50 instrument (EI, 70 eV, direct inlet probe). Elemental analysis was carried out on a Perkin-Elmer Series II, 2400 C,H,N-analyzer. N-Cyclopropyl-N-nitrosourea, ¹² N-(4-fluorophenyl)maleimide, ¹³ 5-amino-*exo*-3-aza-tricyclo[5.2.1.0^{2,6}]decan-4-one (1),² 6-amino-6-methyl-4-azaspiro[2.4]heptan-5-one (2),⁸ and benzohydroximoyl chloride¹⁴ were prepared as described earlier. Compound 1 was additionally recrystallized from ethyl acetate; the ratio of its anti- and syn-isomers was 3.2 : 1. Thin-layer chromatography was carried out on Silica gel 60 plates (Merck); spots were visualized with the iodine vapor. For preparative separation, Silica gel 60 plates (0.040-0.063 mm, Merck) with a layer thickness of 1.5-1.8 mm were employed. Reagent-grade solvents (>99.6%) were used without additional purification.

N-(4-Fluorophenyl)-6-amino-5-oxo-4-azaspiro[2.4]heptane-7-carboxamide (3). A 100-cm³ steel autoclave was charged with 7-(4-fluorophenyl)spiro{cyclopropane-1,4'-2,3,7-triazabicyclo-[3.3.0]oct-2-ene-6,8-dione} (0.52 g, 2 mmol) (prepared from diazocyclopropane and *N*-(4-fluorophenyl)maleimide by analogy with the synthesis of the corresponding 4-bromophenyl derivative),⁸ ethanol (50 mL), and Raney nickel (0.05 g). Hydrogenation was carried out at 60 °C and a hydrogen pressure of 80 bar for 4 h. Then the reaction mixture was filtered and concentrated, the residue was treated with ethyl acetate, and the precipitate that formed was filtered off and dried *in vacuo*. The yield of compound **3** was 0.21 g (40%), colorless crystals, m.p. 256–258 °C. Found (%): C, 59.04; H, 5.55; N, 15.80. C₁₃H₁₄FN₃O₂. Calculated (%): C, 59.31; H 5.36; N, 15.96. Partial MS, *m/z* (I_{rel} (%)): 263 (3) [M]⁺, 246 (2), 207 (5), 153 (12), 126 (42), 125 (67), 111 (100), 109 (97). ¹H NMR ((CD₃)₂SO), & 0.52, 0.81 (both m, 2 H each, CH₂CH₂); 2.00 (br.s, 2 H, NH₂); 3.11 (d, 1 H, H(7)), J = 8.8 Hz); 3.83 (d, 1 H, H(6)), J = 8.8 Hz); 7.12, 7.60 (both m, 2 H each, C₆H₄); 7.81, 10.3 (both br.s, 1 H each, 2 NH). ¹³C NMR ((CD₃)₂SO), & 7.9, 8.7 (CH₂CH₂); 37.2 (C(3)); 54.8, 55.4 (C(6), C(7)); 114.8 (d, C_m, $J_{C,F} = 22.0$ Hz); 128.6 (d, C_o, $J_{C,F} = 7.8$ Hz); 134.6 (d, C_{ipso}, $J_{C,F} = 2.3$ Hz); 157.7 (d, C_p, $J_{C,F} = 239$ Hz); 167.8, 175.2 (2 CO).

Synthesis of azomethines (general procedure). A suspension of 3-aminopyrrolidin-2-ones 1–3 (0.38 mmol), benzaldehyde (0.4 mmol), and anhydrous MgSO₄ (0.7 mmol) in CH₂Cl₂ (5 mL) was stirred at 40 °C for 3–4 h. The reaction mixture was filtered and concentrated. The product was extracted from the residue with ethyl acetate and recrystallized from benzene–AcOEt (1:2) to give azomethines 4, 5, and *trans*-8a,b.

6-Benzylideneamino-6-methyl-4-azaspiro[2.4]heptan-5-one (**4**), 85% yield, m.p. 174–176 °C. Found (%): C, 73.31; H, 7.35; N, 12.08. $C_{14}H_{16}N_2O$. Calculated (%): C, 73.66; H 7.06; N, 12.27. MS, m/z (I_{rel} (%)): 228 (2) [M]⁺, 199 (5), 158 (17), 130 (26), 125 (100). ¹H NMR ((CD₃)₂SO), δ : 0.56–0.77 (m, 4 H, CH₂CH₂); 1.42 (s, 3 H, Me); 2.19, 2.49 (both d, 1 H each, H₂C(7), ²J = 12.9 Hz); 7.43, 7.72 (both m, 3+2 H, Ph); 7.98 (br.s, 1 H, NH); 8.41 (s, HC=N). ¹³C NMR ((CD₃)₂SO), δ : 9.0, 9.3 (CH₂CH₂); 22.3 (Me); 35.1 (C(3)); 44.7 (C(7)); 67.1 (C(6)); 127.4 (C_m); 128.2 (C_o); 130.3 (C_p); 135.8 (C_{ipso}); 158.6 (C=N), 175.1 (CO).

N-(4-Fluorophenyl)-*trans*-6-benzylideneamino-5-oxo-4-azaspiro[2.4]heptane-7-carboxamide (5), 78% yield, m.p. 222–224 °C. Partial MS, m/z (I_{rel} (%)): 351 (2) [M]⁺, 213 (17), 137 (10), 106 (100). ¹H NMR ((CD₃)₂SO), δ : 0.56, 0.70, 0.90 (all m, 1+1+2 H, CH₂CH₂); 3.68 (d, 1 H, H(7), ³J = 7.4 Hz); 4.52 (d, 1 H, H(6), ³J = 7.4 Hz); 7.11, 7.56 (both m, 2 H each, C₆H₄); 7.43, 7.74 (both m, 3+2 H, Ph); 8.18, 10.2 (both br.s, 1 H each, 2 NH); 8.40 (s, HC=N). ¹³C NMR ((CD₃)₂SO), δ : 8.4, 8.7 (CH₂CH₂); 33.4 (C(3)); 52.9 (C(7)); 71.9 (C(6)); 114.8 (d, C_m, J_{C,F} = 21.7 Hz); 120.8 (d, C_o, J_{C,F} = 8.0 Hz); 127.7, 128.3 (C_o, C_m); 130.7 (C_p); 134.5 (d, C_{ipso}, J_{C,F} = 2.2 Hz); 135.0 (C_{ipso}); 157.5 (d, C_p, J_{C,F} = 238 Hz); 164.1 (C=N); 167.6, 171.0 (2 CO).

anti-5-Benzylideneamino-*exo*-3-azatricyclo[5.2.1.0^{2,6}]decan-4-one (8a), 62% yield, m.p. 168–170 °C. Partial MS, m/z (I_{rel} (%)): 254 (1) [M]⁺, 151 (100), 122 (47), 110 (23), 104 (12). ¹H NMR (CDCl₃), δ : 1.19 (m, 3 H, H_{endo} (8), H_{endo} (9), H_{anti} (10)); 1.51 (m, 3 H, H_{exo} (8 and 9), H_{syn} (10)); 2.21 (m, 2 H, H(1), H(7)); 2.48 (ddd, 1 H, H(6), $J_{2,6}$ = 7.2 Hz, $J_{5,6}$ = 3.5 Hz, J = 1.5 Hz); 3.65 (br.d, 1 H, H(5), $J_{5,6}$ = 3.5 Hz); 3.67 (br.d, 1 H, H(2), $J_{2,6}$ = 7.2 Hz); 7.08 (br.s, 1 H, NH); 7.40, 7.79 (both m, 3+2 H, Ph); 8.37 (s, HC=N). ¹³C NMR (CDCl₃), δ : 25.3, 28.2 (C(8), C(9)); 32.2 (C(10)); 40.8, 41.2 (C(1), C(7)); 49.9 (C(6)); 60.8 (C(2)); 75.3 (C(5)); 128.5 (C_o, C_m); 131.0 (C_p); 135.8 (C_{ipso}); 163.4 (C=N), 176.3 (CO). *anti-5-*(2,4-Dimethoxybenzylidene)amino-*exo-*3-azatricyclo-

anti-5-(2,4-Dimethoxybenzylidene)amino-*exo*-3-azatricyclo-[5.2.1.0^{2,6}]decan-4-one (8b), 68% yield, m.p. 188–190 °C. Partial MS, m/z (I_{rel} (%)): 314 (9) [M]⁺, 218 (5), 164 (28), 151 (100), 122 (89), 110 (43). ¹H NMR (CDCl₃), δ : 1.18 (m, 3 H, $\begin{array}{l} H_{endo}(8), \ H_{endo}(9), \ H_{anti}(10)); \ 1.52 \ (m, \ 3 \ H, \ H_{exo}(8 \ and \ 9), \\ H_{syn}(10)); \ 2.20 \ (m, \ 2 \ H, \ H(6), \ H(7)); \ 2.46 \ (m, \ 1 \ H, \ H(1)); \\ 3.62 \ (br.d, \ 1 \ H, \ H(5), \ J_{5.6} = \ 3.4 \ Hz); \ 3.66 \ (br.d, \ 1 \ H, \ H(2), \\ J_{2.6} = \ 7.3 \ Hz); \ 3.83, \ 3.84 \ (both \ s, \ 3 \ H \ each, \ 2 \ OMe); \ 6.20 \ (br.s, \\ 1 \ H, \ NH); \ 6.41 \ (dd, \ 1 \ H, \ H(3'), \ ^4J = \ 2.3 \ Hz, \ ^5J = \ 1.0 \ Hz); \ 6.49 \ (dd, \ 1 \ H, \ H(5'), \ ^3J = \ 8.6 \ Hz, \ ^4J = \ 2.3 \ Hz); \ 7.90 \ (dd, \ 1 \ H, \ H(6'), \ ^3J = \ 8.6 \ Hz, \ ^5J = \ 1.0 \ Hz); \ 8.65 \ (s, \ HC=N). \ ^{13}C \ NMR \ (CDCl_3), \\ 8.25.4, \ 28.2 \ (C(8), \ C(9)); \ 32.2 \ (C(10)); \ 40.8, \ 41.4 \ (C(1), \ C(7)); \ 50.2 \ (C(6)); \ 55.5, \ 55.6 \ (2 \ OMe); \ 60.5 \ (C(2)); \ 75.4 \ (C(5)); \ 98.0 \ (C(3')); \ 105.4 \ (C(5')); \ 117.7 \ (C(1')); \ 128.9 \ (C(6')); \ 159.0 \ (C=N), \ 160.4, \ 163.4 \ (C(2'), \ C(4')); \ 176.5 \ (CO). \end{array}$

6-Benzylamino-6-methyl-4-azaspiro[2.4]heptan-5-one (6). Sodium borohydride (19 mg, 0.5 mmol) was added to a solution of azomethine 4 (34 mg, 0.15 mmol) in anhydrous methanol (3 mL). The reaction mixture was stirred at 20 °C for 12 h. The solvent was removed in vacuo, the residue was treated with water, and the product was extracted with CH₂Cl₂. The organic layer was separated, dried with anhydrous Na_2SO_4 , and concentrated in vacuo. The yield of amine 6 was 29 mg (85%), m.p. 109–111 °C. Partial MS, m/z (I_{rel} (%)): 230 (1) [M]⁺, 201 (4), 173 (5), 146 (15), 125 (22), 106 (84), 91 (100). ¹H NMR (CDCl₂), δ: 0.70, 0.87 (both m, 2 H each, CH₂CH₂); 1.46 (s, 3 H, Me); 1.88, 2.50 (both d, 1 H each, $H_2C(7)$, ${}^2J = 13.0$ Hz); 1.95 (br.s, 1 H, NH); 3.73, 3.83 (both d, 1 H each, H₂CN, ${}^{2}J$ = 12.1 Hz); 6.95 (br.s, 1 H, NH); 7.20–7.40 (m, 5 H, Ph). ¹³C NMR (CDCl₂), δ: 9.8, 12.2 (CH₂CH₂); 24.6 (Me); 35.9 (C(3)); 41.4 (C(7)); 48.2 (NCH₂); 62.7 (C(6)); 127.1 (C_n); 128.4,128.5 (C_o, C_m); 140.3 (C_{ipso}); 179.8 (CO).

N-(4-Fluorophenyl)-*trans*-6-benzylamino-5-oxo-4-azaspiro-[2.4]heptane-7-carboxamide (7) was obtained analogously by reduction of azomethine 5. The yield was 88%, m.p. 146–148 °C. Found (%): C, 67.54; H, 5.44; N, 12.12. $C_{20}H_{20}FN_3O_2$. Calculated (%): C, 67.97; H, 5.70; N, 11.89. Partial MS, *m/z* (I_{rel} (%)): 353 (2) [M]⁺, 262 (6), 215 (10), 125 (19), 106 (34), 91 (100). ¹H NMR ((CD₃)₂SO), & 0.52, 0.82, 0.96 (all m, 1+2+1 H, CH₂CH₂); 2.48 (br.s, 1 H, NH); 3.38 (d, 1 H, H(7), ³J = 10.4 Hz); 3.93, 4.07 (both d, 1 H each, NCH₂, ²J = 12.7 Hz); 4.04 (d, 1 H, H(6), ³J = 10.4 Hz); 6.98, 7.41 (both m, 2 H each, C₆H₄); 7.30 (m, 5 H, Ph); 7.10, 9.3 (both br.s, 1 H each, 2 NH). ¹³C NMR ((CD₃)₂SO), & 8.6, 9.7 (CH₂CH₂); 38.4 (C(3)); 50.7 (C(7)); 51.2 (NCH₂); 61.5 (C(6)); 115.6 (d, C_m, J_{C,F} = 22.2 Hz); 121.7 (d, C_o, J_{C,F} = 8.0 Hz); 127.8 (C_p); 128.5, 128.8 (C_o, C_m); 133.8 (d, C_{ipso}, J_{C,F} = 2.5 Hz); 138.7 (C_{ipso}); 159.4 (d, C_p, J_{C,F} = 241 Hz); 167.4, 174.6 (2 CO).

anti- and syn-5-Acetamido-exo-3-azatricyclo[5.2.1.0^{2,6}]decan-4-one (9). Acetyl chloride (47 mg, 0.6 mmol) in CH₂Cl₂ (0.5 mL) was added dropwise to a stirred solution of amine 1 (100 mg, 0.6 mmol) and triethylamine (67 mg, 0.66 mmol) in CH₂Cl₂ (5 mL). The reaction mixture was stirred at 20 °C for 10 h. Then water (3 mL) was added and the product was extracted with CH₂Cl₂ (3×10 mL). The combined organic extracts were dried with anhydrous MgSO4 and concentrated in vacuo. The yield of compound 9 as a 3.2:1 mixture of transand cis-isomers was 116 mg (93%). The anti-isomer was isolated using preparative TLC on SiO₂ with benzene-AcOEt (1:3) as an eluent, yellowish crystals, R_f 0.30, m.p. 64-67 °C. anti-Isomer of compound 9. Found (%): C, 63.06; H, 7.89; N, 13.16. C₁₁H₁₆N₂O₂. Calculated (%): C, 63.44; H, 7.74; N, 13.45. Partial MS, m/z (I_{rel} (%)): 208 (16) [M]⁺, 190 (5), 166 (24), 165 (70) $[M - Ac]^+$, 150 (10) $[M - NHAc]^+$, 43 (100). ¹H NMR (CDCl₃), δ : 1.19 (m, 3 H, H_{endo}(8), H_{endo}(9), H_{anti}(10)); 1.41 (dq, 1 H, H_{syn}(10), ${}^{2}J = 10.2$ Hz, $J \approx 1.8$ Hz); 1.55 (m, 2 H, H_{exo}(8 and 9)); 2.00 (s, 3 H, Me); 2.18 (m, 2 H, H(6), H(7)); 2.44 (m, 1 H, H(1)); 3.54 (br.d, 1 H, H(2), $J_{2,6} = 7.5$ Hz); 3.81 (dd, 1 H, H(5), $J_{5,6} = 4.4$ Hz, ${}^{3}J = 6.6$ Hz); 6.66 (br.d, 1 H, NHAc, ${}^{3}J = 6.6$ Hz); 6.70 (br.s, 1 H, NH). 13 C NMR (CDCl₃), δ : 23.0 (Me); 25.4, 27.8 (C(8), C(9)); 32.0 (C(10)); 41.1, 41.3 (C(1), C(7)); 49.9 (C(6)); 56.3 (C(2)); 60.1 (C(5)); 170.8 (CO); 176.5 (C(4)). syn-Isomer of compound 9 (in the mixture with the *trans*-isomer). ¹H NMR (CDCl₃), non-overlapping signals, δ : 2.07 (s, 3 H, Me); 2.59 (br.d, 1 H, H(6), $J_{2,6} = 6.9$ Hz, $J_{5,6} = 10.1$ Hz); 3.59 (br.d, 1 H, H(2), $J_{2,6} = 6.9$ Hz); 4.58 (dd, 1 H, H(5), $J_{5,6} = 10.1$ Hz, ${}^{3}J = 6.0$ Hz); 6.52 (br.d, 1 H, NHAc, ${}^{3}J = 6.0$ Hz); 6.60 (br.s, 1 H, NH). 13 C NMR (CDCl₃), δ : 22.8 (Me); 24.7, 28.5 (C(8), C(9)); 29.7 (C(10)); 36.4 (C(7)); 41.7 (C(1); 44.4 (C(6)); 52.3 (C(2)); 60.0 (C(5)); 171.0 (CO); 176.3 (C(4)).

N-(4-Fluorophenyl)-trans-6-acetamido-5-oxo-4-azaspiro-[2.4]heptane-7-carboxamide (10). Acetyl chloride (32 mg, 0.4 mmol) in CH₂Cl₂ (0.5 mL) was added dropwise to a stirred mixture of amine 3 (105 mg, 0.4 mmol) and triethylamine (42 mg, 0.42 mmol) in CH₂Cl₂ (6 mL). The reaction mixture was stirred at 20 °C for 12 h. Then 0.3 M NaOH (1.4 mL) was added and the mixture was evaporated in vacuo almost to dryness. The product was extracted from the residue with ethyl acetate (3×10 mL). The combined extracts were concentrated in vacuo. The yield of compound 10 was 110 mg (90%), colorless small crystals, m.p. 288-290 °C. Found (%): C, 58.66; H, 5.49; N, 13.37. C₁₅H₁₆FN₃O₃. Calculated (%): C, 59.01; H, 5.28; N, 13.76. ¹H NMR ((CD₂)₂SO), δ: 0.49, 0.59, 0.85 (all m, 1+1+2 H, CH₂CH₂); 1.82 (s, 3 H, Me); 3.52 (d, 1 H, H(7), $J_{6,7} = 8.4 \text{ Hz}$; $\overline{4.60}$ (dd, 1 H, H(6), $J_{6,7} = 8.4 \text{ Hz}$, ${}^{3}J = 8.1 \text{ Hz}$); 7.14, 7.59 (both m, 2 H each, $C_6H_4^{(3)}$; 7.99, 10.2 (both br.s, 1 H each, 2 NH); 8.44 (br.d, 1 H, NHAc, ${}^{3}J = 8.1$ Hz). ¹³C NMR ((CD₂)₂SO), δ: 8.7, 8.8 (CH₂CH₂); 22.1 (Me); 37.6 (C(3)); 51.2 (C(7)); 53.4 (C(6)); 114.9 (d, $C_{m,}$, $J_{C,F} = 22.0$ Hz); 120.6 (d, C_{o} , $J_{C,F} = 8.0$ Hz); 134.5 (d, C_{ipso} , $J_{C,F} = 2.8$ Hz); 157.6 (d, C_{p} , $J_{C,F} = 240$ Hz); 167.8, 169.0, 171.2 (3 CO).

anti- and syn-5-Acetoxy-exo-3-azatricyclo[5.2.1.0^{2,6}]decan-4-one (11). Sodium nitrite (207 mg, 3 mmol) was added at 10 °C to a solution of aminopyrrolidinone 1 (100 mg, 0.6 mmol) in acetic acid (5 mL) containing CH₂Cl₂ (0.5 mL). The reaction mixture was stirred for 2 h. Then the acetic acid was removed in vacuo, water (1 mL) was added, and the product was extracted with CH₂Cl₂. The extract was concentrated in vacuo. According to the integral intensities of the non-overlapping signals at $\delta 4.82$ and 5.30 in the ¹H NMR spectrum of the reaction mixture, the ratio of anti- and syn-11 was ~1.6:1. Acetoxy derivatives 11 were isolated by preparative TLC on SiO₂ (benzene-AcOEt, 1:3) as a ~5:1 mixture of the *anti*- and *syn*-isomers ($R_f 0.32$). Another fraction with $R_f 0.24$ contained syn-11 (35-40%). The total yield of compounds 11 was 50–55%. Partial MS, m/z $(I_{\rm rel} (\%)): 209 (17) [M]^+, 167 (40), 166 (21) [M - Ac]^+, 149 (64)$ $[M - AcOH]^+$, 106 (84), 67 (90), 43 (100).

anti-Isomer of compound 11. ¹H NMR (CDCl₃), &: 1.15 (m, 2 H, H_{endo}(8), H_{endo}(9)); 1.22 (dq, 1 H, H_{anti}(10), ²J = 10.8 Hz, $J \approx 1.8$ Hz); 1.36 (dq, 1 H, H_{syn}(10), ²J = 10.8 Hz, $J \approx 1.8$ Hz); 1.55 (m, 2 H, H_{exo}(8), H_{exo}(9)); 2.07 (m, 1 H, H(6)); 2.14 (s, 3 H, Me); 2.20 (m, 1 H, H(7)); 2.48 (m, 1 H, H(1)); 3.57 (br.d, 1 H, H(2), $J_{2,6} = 7.0$ Hz); 4.82 (d, 1 H, H(5), $J_{5,6} = 3.6$ Hz); 6.98 (br.s, 1 H, NH). ¹³C NMR (CDCl₃), &: 20.9 (Me); 25.2 (C(9)); 27.9 (C(8)); 32.1 (C(10)); 40.1, 40.9

(C(1), C(7)); 49.0 (C(6)); 60.5 (C(2)); 75.4 (C(5)); 170.5 (CO); 174.3 (C(4)).

syn-Isomer of compound 11. ¹H NMR (CDCl₃), non-overlapping signals, δ : 2.02, 2.09 (both m, 1 H each, H(1), H(7)); 2.20 (s, 3 H, Me); 2.55 (m, 1 H, H(6)); 3.57 (br.d, 1 H, H(2), $J_{2.6} = 7.0$ Hz); 5.30 (d, 1 H, H(5), $J_{5.6} = 10.0$ Hz); 6.85 (br.s, 1 H, NH). ¹³C NMR (CDCl₃), δ : 20.6 (Me); 24.7 (C(9)); 28.3 (C(8)); 32.3 (C(10)); 36.2 (C(7)); 41.4 (C(1)); 44.0 (C(6)); 60.0 (C(2)); 71.1 (C(5)); 170.5 (CO); 174.1 (C(4)).

N-(4-Fluorophenyl)-*trans*-6-acetoxy-5-oxo-4-azaspiro-[2.4]heptane-7-carboxamide (12) and *N*-(4-fluorophenyl)-5-oxo-4-azaspiro[2.4]hept-6-ene-7-carboxamide (13). Sodium nitrite (0.19 g, 2.8 mmol) was added at 10 °C to a solution of aminopyrrolidinone **3** (0.13 g, 0.5 mmol) in acetic acid (5 mL) containing CH_2Cl_2 (0.5 mL). The reaction mixture was stirred for 2 h. Then the acetic acid was removed *in vacuo*, water (1 mL) was added, and the product was extracted with ethyl acetate. The extract was concentrated *in vacuo*. A fine crystalline substance (0.10 g, 70–73%) consisting of compounds **12** and **13** (molar ratio 1.4 : 1) was isolated by preparative TLC on SiO₂ with AcOEt as an eluent, $R_c 0.26-0.35$.

Compound 12. ¹H NMR ((CD₃)₂CO), & 0.80, 0.96 (both m, CH₂CH₂); 1.96 (s, 3 H, Me); 3.53 (d, 1 H, H(7), $J_{6,7} = 7.3$ Hz); 5.82 (d, 1 H, H(6), $J_{6,7} = 7.3$ Hz); 7.08 and 7.63 (both m, 2 H each, C₆H₄); 7.35, 9.45 (both br.s, 1 H each, 2 NH). ¹³C NMR ((CD₃)₂CO), & 9.0, 9.5 (CH₂CH₂); 19.8 (Me); 38.4 (C(3)); 52.6 (C(7)); 73.3 (C(6)); 115.2 (d, C_m, $J_{C,F} = 22.4$ Hz); 121.4 (d, C_o, $J_{C,F} = 8.2$ Hz); 135.2 (d, C_{ipso}, $J_{C,F} = 2.8$ Hz); 157.4 (d, C_p, $J_{C,F} = 240$ Hz); 167.2, 169.6, 175.8 (3 CO). **Compound 13.** ¹H NMR ((CD₃)₂CO), & 1.46, 1.91 (both m,

Compound 13. ¹H NMR ((CD₃)₂CO), δ : 1.46, 1.91 (both m, 2 H each, CH₂CH₂); 6.71 (s, 1 H, H(6)); 7.06, 7.72 (both m, 2 H each, C₆H₄); 7.85, 9.65 (both br.s, 1 H each, 2 NH). ¹³C NMR ((CD₃)₂CO), δ : 13.4 (CH₂CH₂); 37.4 (C(3)); 115.1 (d, C_m, J_{C,F} = 22.5 Hz); 122.0 (d, C_o, J_{C,F} = 8.0 Hz); 126.8 (C(6)); 134.9 (d, C_{ipso}, J_{C,F} = 2.8 Hz); 158.5 (d, C_p, J_{C,F} = 238 Hz); 164.2, 168.1, 170.0 (C(7), 2 CO).

Nitrosation of 6-amino-6-methyl-4-azaspiro[2.4]heptan-5-one (2). Sodium nitrite (1.04 g, 15 mmol) was added at 30 °C for 30 min to an actively stirred solution of aminopyrrolidinone 2 (1.12 g, 8 mmol) in CCl_4 (14 mL) and acetic acid (6 mL). The reaction mixture was stirred for 2 h, poured into cold water, and neutralized with K₂CO₃. The product was extracted with CH₂Cl₂. The organic layer was dried with anhydrous Na₂SO₄ and concentrated in vacuo. The residue was washed with a small amount of ether and dried in vacuo. According to ¹H and 13 C NMR data, the resulting colorless crystals (0.87 g, ~88%) consisted of 6-methyl-4-azaspiro[2.4]hept-6-en-5-one (14) and 6-methylidene-4-azaspiro[2.4]heptan-5-one (15) in a ratio of \sim 3.5 : 1. Compound 14 was isolated in the individual state upon treatment of the mixture of the isomers obtained with diazomethane or diazocyclopropane, because it was inert under those conditions (see below).

Compound 14. ¹H NMR (CDCl₃), δ : 1.23, 1.44 (both m, 2 H each, CH₂CH₂); 1.92 (d, 3 H, Me, J = 1.6 Hz); 6.40 (dq, 1 H, =CH, J = 1.5 Hz, J = 1.6 Hz); 8.10 (br.s, 1 H, NH). ¹³C NMR (CDCl₃), δ : 10.8 (Me); 11.5 (CH₂CH₂); 43.3 (C(3)); 133.0 (C(6)); 144.8 (C(7)); 175.3 (C=O).

Compound 15. ¹H NMR (CDCl₃), δ : 0.71, 0.95 (both m, 2 H each, CH₂CH₂); 2.87 (t, 2 H, H₂C(7), J = 2.5 Hz); 5.36, 6.02 (both m, 1 H each, =CH₂); 8.30 (br.s, 1 H, NH). ¹³C NMR (CDCl₃), δ : 11.8 (CH₂CH₂); 35.1 (C(7)); 36.8 (C(3)); 115.6 (=CH₂); 140.9 (C(6)); 170.7 (C=O). The yields and ratios of compounds 14 and 15 for different solvents and reaction conditions are given in Table 1.

6,7,11-Triazadispiro[**2.1.4**⁵.**2**³]**undec-6-en-10-one (16).** A 0.5 *M* solution of diazomethane (0.8 mL, 0.4 mmol) in ether was added to a solution of a ~3.5 : 1 mixture (55.4 mg) of isomers **14** and **15** (this corresponds to ~0.1 mmol of compound **15**) in CH₂Cl₂ (1 mL). The reaction mixture was kept at 18 °C for two days. The solvents were removed *in vacuo* and the residue was separated by preparative TLC (SiO₂, AcOEt). The yield of dispirane **16** was 14.4 mg (87% with respect to compound **15**), colorless crystals, R_f 0.46, m.p. 156–157 °C The unreacted compound **14** (39 mg) was recovered, R_f 0.32, m.p. 123–124 °C.

Compound 16. Found (%): C, 58.46; H, 6.39; N, 25.20. $C_8H_{11}N_3O$. Calculated (%): C, 58.17; H, 6.71; N, 25.44. Partial MS (EI), *m/z* (I_{rel} (%)): 149 (4), 136 (25), 122 (100), 110 (12), 109 (24). ¹H NMR (CDCl₃), & 0.77, 0.96 (both m, 1+3 H, CH₂CH₂); 1.46 (ddd, 1 H, H_a(9), ²J = 12.6 Hz, ³J = 8.0 Hz, ³J = 9.0 Hz); 2.28 (ddd, 1 H, H_b(9), ²J = 12.6 Hz, ³J = 6.4 Hz, ³J = 8.1 Hz); 2.37, 2.53 (both d, 1 H each, H₂C(4), ²J = 13.3 Hz); 4.66 (m, 2 H, H₂C(8)); 5.85 (br.s, 1 H, NH). ¹³C NMR (CDCl₃), &: 10.2, 11.7 (CH₂CH₂); 26.4 (C(9)); 37.4 (C(3)); 40.6 (C(4)); 77.6 (C(8)); 97.3 (C(5)); 173.6 (CO).

Compound 14. Partial MS (EI), m/z ($I_{rel}(\%)$): 123 (100) $[M]^+$, 122 (38) $[M - H]^+$, 108 (15), 94 (27). For the ¹H and ¹³C NMR spectra, see the section "Nitrosation of compound **2**").

10,11,13-Triazatrispiro[2.1.1.2⁷.2⁵.2³]tridec-10-en-12-one (17). N-Cyclopropyl-N-nitrosourea (20.5 mg, 0.15 mmol) and Cs_2CO_2 (65 mg, 0.2 mmol) were added at 5–7 °C to a solution of a ~3.5:1 mixture (55.3 mg) of isomers 14 and 15 (this corresponds to ~0.1 mmol of compound 15) in CH_2Cl_2 (3 mL). The reaction mixture was vigorously stirred for 30 min, filtered, and concentrated in vacuo. The residue was separated by preparative TLC (SiO₂, AcOEt). The unreacted compound 14 (38 mg) was recovered, $R_{\rm f}$ 0.32 (its characteristics are given above). The yield of trispyran 17 was 15.8 mg (83% with respect to compound 15), colorless crystals, R_f 0.55, m.p. 176-177 °C. Found (%): C, 62.53; H, 6.70; N, 22.03. $C_{10}H_{13}N_3O$. Calculated (%): C, 62.81; H, 6.85; N, 21.97. Partial MS, *m/z* (*I*_{rel} (%)): 162 (14), 148 (27), 135 (100), 120 (29), 95 (38), 79 (90). ¹H NMR (CDCl₃), δ: 0.72, 0.93, 1.11, 1.23 (all m, 1+3+1+1 H, protons of the cyclopropane rings); 1.66, 2.44 (both d, 1 H each, $H_2C(4)$, $^{2}J = 12.7$ Hz); 1.81 (m, 2 H, CHCH of the cyclopropane ring next to the N=N fragment); 2.37, 2.53 (both d, 1 H each, $H_2C(6)$, $^2J = 13.3 Hz$; 6.32 (br.s, 1 H, NH). ^{13}C NMR (CDCl₂), δ: 10.2, 12.2, 14.6, 15.1 (2 CH₂CH₂); 33.6 (C(6)); 37.3 (C(3)); 41.9 (C(4)); 70.3 (C(7)); 95.9 (C(5)); 173.4 (CO).

8-Phenyl-6-oxa-7,11-diazadispiro[2.1.4⁵.2³]undec-7-en-**10-one (18).** Benzohydroximoyl chloride (0.16 g, 1 mmol) and triethylamine (0.11 g, 1.1 mmol) were added to a solution of a ~3.5 : 1 mixture (0.22 g) of isomers **14** and **15** (this corresponds to ~0.4 mmol of compound **15**) in CH₂Cl₂ (5 mL). The reaction mixture was kept at 18 °C for 20 h and treated with water (10 mL). The organic layer was separated and concentrated *in vacuo*. The residue was dissolved in a minimum amount of methanol at 30 °C and left at -15 °C for 12 h. The precipitate that formed was filtered off, washed with a small amount of chloroform, and dried *in vacuo*. The yield of compound **18** was 0.091 g (94%), colorless crystals, m.p. 235–236 °C. Found (%): C, 69.15; H, 5.88; N, 11.23. C₁₄H₁₄N₂O₂. Calculated (%): C, 69.41; H, 5.82; N, 11.56. Partial MS, *m/z* (I_{rel} (%)): 242 (3) $[M]^{+}, 214 (14), 144 (36), 117 (18), 82 (40), 77 (100). {}^{1}H NMR ((CD_3)_2SO), \delta: 0.62-0.82 (m, 4 H, CH_2CH_2); 2.30, 2.46 (both d, 1 H each, H_2C(4), {}^{2}J = 13.9 Hz); 3.52, 3.69 (both d, 1 H each, H_2C(6), {}^{2}J = 17.2 Hz); 7.47, 7.69 (both m, 3+2 H, Ph); 8.33 (br.s, 1 H, NH). {}^{13}C NMR ((CD_3)_2SO), \delta: 8.4, 10.2 (CH_2CH_2); 35.3 (C(3)); 41.2, 41.5 (C(4), C(9)); 87.9 (C(5)); 126.2 (C_0); 128.4 (C_m); 128.5 (C_{ipso}); 129.8 (C_p); 155.5 (C(8)); 171.8 (CO).$

4-[Hydroxyimino(phenyl)methyl]-6-methyl-4-azaspiro-[2.4]hept-6-en-5-one (19). A mixture of compound 14 (49 mg, 0.4 mmol), benzohydroximoyl chloride (62 mg, 0.4 mmol), and triethylamine (46 mg, 0.45 mmol) was refluxed in benzene (5 mL) for 12 h. On cooling, the precipitate that formed was filtered off, the solvent was removed in vacuo, and the residue was separated by preparative TLC (SiO₂, AcOEt). The yield of compound **19** was 39 mg (40%), yellow crystals, $R_{\rm f}$ 0.55, m.p. 113–115 °C. MS, m/z (I_{rel} (%)): 242 (52) [M]⁺, 178 (45), 123 (100), 105 (43). ¹H NMR (CDCl₃), δ: 1.26 (m, 2 H, $CH_{2}CH_{2}$); 1.97 (d, 3 H, Me, ${}^{4}J = 1.5$ Hz); 2.48 (m, 2 H, $CH_{2}CH_{2})$; 6.54 (q, 1 H, =CH, ${}^{4}J$ = 1.5 Hz); 7.08 (tt, 1 H, H_{p} , ${}^{3}J = 7.7 \text{ Hz}$, ${}^{4}J = 1.3 \text{ Hz}$; 7.31 (dd, 2 H, H_{m} , ${}^{3}J = 8.4 \text{ Hz}$, ${}^{3}J = 7.7 \text{ Hz}$); 7.50 (dd, 2 H, H_{o} , ${}^{3}J = 8.4 \text{ Hz}$, ${}^{4}J = 1.3 \text{ Hz}$); 11.0 (br.s, 1 H, NOH). ${}^{13}\text{C}$ NMR (CDCl₃), δ : 10.6 (Me); 11.6 (CH₂CH₂); 48.0 (C(3)); 120.4 (C₂); 124.0 (C₂); 129.0 (C_m) ; 129.8 $(\overline{C_{inso}})$; 137.5 (C(6)), 149.6 (C=N); 149.7 (C(7)); 173.7 (CO).

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References

- I. M. Bell, S. N. Gallicchio, M. Abrams, D. C. Beshore, C. A. Buser, J. C. Culberson, J. Davide, M. Ellis-Hutchings, C. Fernandes, J. B. Gibbs, S. L. Graham, G. D. Hartman, D. C. Heimbrook, C. F. Homnick, J. R. Huff, K. Kassahun, K. S. Koblan, N. E. Kohl, R. B. Lobell, J. J. Lynch, Jr, P. A. Miller, C. A. Omer, A. D. Rodriguez, E. S. Walsh, T. M. Williams, J. Med. Chem., 2001, 44, 2933.
- V. A. Gorpinchenko, E. A. Yatsynich, D. V. Petrov, L. T. Karachurina, R. Yu. Khisamutdinova, N. Zh. Baschenko, V. A. Dokichev, Yu. V. Tomilov, M. S. Yunusov, O. M. Nefedov, *Khim.-Farm. Zh.*, 2005, **39**, 89 [*Pharm. Chem. J. (Engl. Transl.*), 2005, **39**].
- S. J. F. Macdonald, G. D. E. Clarke, M. D. Dowle, L. A. Harrison, S. T. Hodgson, G. G. A. Inglis, M. R. Johnson, P. Shah, R. J. Upton, S. B. Walls, *J. Org. Chem.*, 1999, 64, 5166.
- A. Avenoza, C. Cativiela, J. Peregrina, M. Zurbano, *Tetrahedron: Asymmetry*, 1997, 8, 863.

- 5. P. Camps, D. Munoz-Torrero, J. Rull, M. Font-Bardia, X. Solans, *Tetrahedron: Asymmetry*, 2007, **18**, 2947.
- 6. M. Kordes, H. Winsel, A. de Meijere, Eur. J. Org. Chem., 2000, 18, 3235.
- 7. C. Laroche, D. Harakat, Ph. Bertus, J. Szymoniak, Org. Biomol. Chem., 2005, 3, 3482.
- I. V. Kostyuchenko, E. V. Shulishov, V. A. Korolev, V. A. Dokichev, Yu. V. Tomilov, *Izv. Akad. Nauk, Ser. Khim.*, 2005, 2482 [*Russ. Chem. Bull., Int Ed.*, 2005, 54, 2562].
- 9. H. Dobenek, A. Uhl, Liebigs Ann. Chem., 1974, 1550.
- 10. P. Oravec, L. Fišera, P. Ertl, D. Végh, *Monatsh. Chemie*, 1991, **122**, 821.

- 11. E. Jedlovska, L. Fišera, J. Heterocycl. Chem., 2004, 41, 677.
- 12. V. P. Gol'mov, Zh. Obshch. Khim., 1935, 5, 1562 [Chem. Zentralbl., 1936, 107 (II), 1905].
- 13. J. M. Barrales-Rienda, J. G. Ramos, M. S. Chavez, J. Fluorine Chem., 1977, 9, 293.
- 14. A. Werner, H. Buss, Chem. Ber., 1894, 27, 2193.

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