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Synthesis of novel chromene derivatives of expected antitumor activity

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ABSTRACT

Inhibition of tubulin polymerization is one of the important tactics in cancer therapy. Since 4-aryl-4*H*-chromene derivatives are found to be microtubule-binding agents via interfering with tubulin polymerization so we decide to concentrate our exploration efforts on the combination of this nucleus with 5-, 6-, and/or 7-membered heterocyclic moieties in a novel series of compounds to explore the effect that might result from this combination. Ten novel compounds were selected for anticancer screening assay against MCF-7 breast cancer cell line in comparison to colchicine as positive control and most of them showed excellent activity.

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1. Introduction

Microtubule-binding agents (MBAs) are widely used in cancer chemotherapy [1]. There are two classes of MBAs; those that stabilize microtubules and promote polymerization, and those that destabilize the microtubules and promote depolymerization. Both types interfere with the mitotic spindle assembly during cell division, resulting in cell death. Colchicine (I), podophyllotoxin (II), and combretastatin A-4 (III) (Fig. 1) are representatives of compounds that inhibit microtubule assembly [2,3] and share a common structural feature; two aromatic rings directly bonded or separated by 1–4 carbon atoms, in such a way that they are close in space but not coplanar [4].

A major turning point in the development of MBAs as antitumor agents is the identification of the natural compound namely combretastatin A-4 (III) [4]. Compound III strongly inhibits the polymerization of tubulin by binding to the colchicine binding site [2,3]. Besides, it can inhibit angiogenesis, a process essential for tumor growth [5–7]. The synthesis of different 4-arylcoumarin analogs of combretastatin A-4 (III) lead to the discovery of promising series of potent apoptosis inducing agents possessing vascular targeting activity [8–14]. These compounds were found to be tubulin

* Corresponding author. Tel.: +20 1280007280 (mobile); fax: +20 2 23628426. *E-mail addresses:* K_aliaa2511@hotmail.com, K_aliaa2@hotmail.com (A.M. Kamal). destabilizers, binding at or nearly at the binding site of colchicine. They were also active in drug-resistant cancer cell lines and highly active as single agents or in combination with other anticancer agents in several tumor models, so they could be developed into new therapeutic anticancer regimes [15–19]. On the other hand, numerous triazolo, triazino and triazepino pyrimidine derivatives have been reported as cell cycle arrest agents that act via interfering with the polymerization [20,21] or depolymerization [22] of microtubules assembly leading to cell death. Prompted by these findings, we intended to synthesize novel series of compounds containing both 4-aryl-4H-chromenes and triazolo, triazino and triazepino pyrimidine moieties to explore the synergistic effect that might result from this combination (Schemes 1–5).

2. Results and discussion

2.1. Chemistry

2-Amino-4-(4-chlorophenyl)-7-hydroxy-4*H*-chromene-3carbonitrile (**1**) was synthesized according to the reported method [23,24] and alkylated using methyl or ethyl iodide to afford compounds **2a** and **2b**. The IR spectra of **2a** and **2b** assured the presence of both amino functionality and cyano group at 3432 (**2a**), 3332 (**2b**) cm⁻¹ and 2196 (**2a**), 2185 (**2b**) cm⁻¹ respectively. Meanwhile, ¹H NMR spectra of compounds **2a** and **2b** revealed the presence of one exchangeable singlet signal at δ 6.98 and 6.54 ppm respectively corresponding to NH₂ protons. The ¹³C NMR for



Original article



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Fig. 1. Colchicine I, podophyllotoxin II, and combretastatin A-4 III.

compound **2a** disclosed three signals at δ 39.9 ppm, δ 55.5 ppm and δ 114.2 ppm corresponding to the stereogenic carbon of the chromene nucleus, CH₃ and CN, sequentially.

Compounds **2a** and **2b** were reacted with triethyl orthoformate to give **3a** and **3b**. IR spectra of compounds **3a** and **3b** revealed the disappearance of NH₂ group while, their ¹H NMR spectra showed a triplet at δ 1.39, 1.36 ppm (OCH₂CH₃), a quartet at δ 4.43, 3.49 ppm (OCH₂CH₃) and a singlet at δ 8.30, 8.38 ppm indicated the presence of imino group (N=CH). Interaction of **3a** and **3b** with equimolar amount of either methylamine or hydrazine hydrate in absolute ethanol at ambient temperature gave the key intermediates **4a**– **d** (Scheme 1).

The structures of compounds **4a**–**d** were established on the basis of elemental analyses and spectral data. IR spectra of compounds **4a**–**d** showed no absorption band for CN group meanwhile revealed characteristic absorption bands at 3336–3156 cm⁻¹ indicating the presence of NH, NH₂ functionality. Furthermore, ¹H NMR spectra of compounds **4a**–**d** revealed the appearance of singlet signals at δ 7.79–8.17 ppm corresponding to C2H of the pyrimidine ring. ¹H NMR spectrum of compound **4a** revealed two singlets at δ 3.40 and 3.77 ppm assigned to N–CH₃ and O–CH₃ protons, respectively, ¹H NMR spectrum of compound

4b showed appearance of D_2O exchangeable broad signal at δ 4.74 ppm assigned to two protons of NH₂ group. It is worth to mention that, upon hydrazinolysis of **3a** and **3b** using excess hydrazine hydrate 99% instead of equimolar amounts, compounds **2a** and **2b** were recovered. This may be attributed to, after the reaction of 1 mol of hydrazine with the iminoether **3a** and **3b** a preferential elimination of ethyl formate hydrazone rather than cyclization to give the pyrimidine ring took place (Fig. 2).

Schemes 2–5 summarizes the synthesis of the triazolochromenopyrimidines with different substitution at C2 using different one-carbon donor cyclizing agents to react with the key intermediates **4b** and **4d**. Cyclization of **4b** and **4d** with either triethyl orthoformate, acetic anhydride or diethyl oxalate gave the corresponding 1,2, 4-triazolo[1,5-c]pyrimidines **5a** and **5b**; **5c** and **5d** and **5e** and **5f** respectively. Neither IR nor ¹H NMR spectra supported the presence of the NH or NH₂ protons that existed in the precursor **4b** and **4d**. IR spectra of compounds **5e** and **5f** showed a band at 1741, 1745 cm⁻¹ confirming the presence of (*C*=O) of the ester moiety. Moreover, the ¹H NMR spectra of **5a**–**f** showed two singlet signals at δ 5.63, 5.81 ppm and δ 9.00, 9.75 ppm indicating C12H and C5H, respectively. The increased chemical shift for these signals-compared to their values in tricyclic system **4b** and **4d**



Scheme 1. Synthesis of compounds 2a and 2b, 3a and 3b and 4a-d.



Scheme 2. Synthesis of compounds 5a and 5b.

 δ 4.88–5.22 (C5H) and δ 8.09–8.11(C2H) ppm- can be attributed to the deshielding effect of the diamagnetic current of the aryl π -electrons.

The ¹H NMR spectra of compounds **5a** and **5b** revealed a singlet signal at δ 8.28 ppm which was assigned to the C2H. While, ¹H NMR spectra of **5c** and **5d**, the methyl protons (C2–CH₃) were found as a singlet signal at δ 2.55, 2.57 ppm. Also, the ¹H NMR spectra of **5e** and **5f** showed triplet and quartet signals corresponding to ester protons.

In the course of this study, attempts of synthesis of chromenopyrimidinotriazepines **A** via reacting compounds **4b** and **4d** with ethyl ethoxymethylene cyanoacetate were not successful. Instead, this reaction gave a product which was identical in all respects (m.p., mixed m.p., IR and ¹H NMR) with compounds **5a** and **5b** (Scheme 2).

The suggested mechanism for the reaction of **4b** and **4d** with ethyl ethoxymethylene cyanoacetate might be illustrated in Fig. 3.

Reacting compounds **4b** and **4d** with acetyl acetone for obtaining compounds **B** was unsuccessful and compounds **5c** and **5d** were resulted instead (Scheme 3).

The suggested mechanism for the reaction of **4b** and **4d** with acetyl acetone might be illustrated in Fig. 4.

An attempt to prepare hydroxypyrimidotriazines **C** via heating compounds **4b** and **4d** with ethyl cyanoacetate in refluxing ethanol took place but unfortunately different products were obtained. These compounds were suggested to be the triazolopyrimidines **5g** and **5h** according to all available spectroscopic data and elemental analysis.

The IR spectra showed the presence of cyano group at 2251, 2259 cm⁻¹. Furthermore, ¹H NMR spectra revealed a singlet signal at δ 4.00 corresponding to two protons (*CH*₂CN) and no signals corresponding to OH either in IR or ¹H NMR spectra.

A suggested mechanism for the formation of **5g** and **5h** is shown (Fig. 5).

Preparation of compounds **D** was intended to be via the reaction of the key intermediates **4b** and **4d** with different chloro acid chlorides but interestingly, treating compounds **4b** and **4d** with equimolar amount of chloro acid chloride in dioxane containing catalytic amount of triethylamine did afford neither the starting materials nor compounds **D** but compounds **6a**–**d** were the products and this might be ascribed to the stability and aromaticity of the triazole ring system over the triazepine ring system (Scheme 5).

The structure of **6a–d** was suggested by different spectroscopic data. IR spectra displayed the absence of absorption bands corresponding to (C=O and NH) groups existing in the compounds **D**. 1 H NMR spectra of compounds 6a and 6c showed a singlet signal at δ 4.73 assigned to two protons of (CH₂Cl). Inspection of the ¹H NMR spectra of compounds **6b** and **6d** revealed the presence of pentet and two triplet signals at δ 2.22–2.38, 3.06–3.07 and 3.66 ppm due to $(CH_2-CH_2-CH_2)$ protons respectively (Scheme 5). Compounds 7a-f were prepared from the interaction of the key intermediates 4b and 4d with different arylidene malononitriles in the presence of piperidine under reflux in absolute ethanol. The structure of the given compounds was consistent with the IR, ¹H NMR and mass spectra. The absence of NH group as well as NH₂ group from the IR and ¹H NMR spectra of **7a**–**f** showed the increase in aromatic protons at δ 6.71–8.28 ppm confirmed their structures. Once more C12H and C5H of compounds 7a-f showed increase in chemical shift at δ 5.72–5.76 (C12H) and δ 9.05–9.09 (C5H) ppm – compared to their values in tricyclic system **4b** and **4d** δ 4.88–5.22 (C5H) and δ 8.09–8.11 (C2H) ppm – and this can be attributed to the deshielding effect of the diamagnetic current of the aryl π electrons.



Scheme 3. Synthesis of compounds 5c and 5d.

2.2. Antitumor activity

In this study, ten of the newly synthesized compounds were subjected to cytotoxic evaluation against human breast tumor cell line (MCF-7) using Colchicine as the reference drug. The response parameter calculated was IC_{50} value (Table 1), which corresponds to the compound concentration causing 50% mortality in net cells.

Compounds **4a**, **4b**, **5f**, **7c** and **7f** turned to be particularly promising for the development of new pharmacological inhibitors for breast tumor cells. The most promising results were obtained with compounds **4a** and **5f** ($IC_{50} = 0.007 \mu$ M) that showed almost double the activity of colchicine ($IC_{50} = 0.013 \mu$ M) (Fig. 6).

Three of the tested compounds **4b**, **7c** and **7f** ($IC_{50} = 0.008 \mu M$) (Fig. 7) showed nearly one and half the activity of colchicine.

Compounds **5a** and **5h** (IC₅₀ = 0.011 μ M) (Fig. 8) showed almost equal inhibitory effect on breast tumor cells as than the positive control (IC₅₀ = 0.013 μ M).

Moreover, compound **6d** (IC₅₀ = 0.029 μ M) (Fig. 9) revealed nearly half the activity of the positive control. Finally compounds **5c** and **5g** (IC₅₀ = 0.033 μ M, 0.039 μ M) (Fig. 10) are the least reactive ones.

3. Conclusion

In summary, the objective of the present study was to design, synthesize a library of novel hybrid compounds comprising chromenes and different triazolopyrimidinyl moieties and investigate the anticancer activity of these new compounds. It is worth to mention, many of unexpected compounds were resulted during the reaction of the key intermediates **4b** and **4d** with different organic reagents and this may be attributed to the electronic nature of this aminocyanochromene ring system where the amino group is a very weak basic center and carbon number 2 is carrying a highly positive charge. The preliminary biological evaluations revealed that the newly synthesized chromene derivatives **4a**, **4b**, **5f**, **7c** and **7f** showed superior *in vitro* antitumor activity when compared to colchicine as a reference drug. We might conclude that, this superior *in vitro* anticancer activity and good potential of causing cancer cells death might be attributed to their structure similarity to colchicine (I) and podophyllotoxin (II) (Fig. 1).

4. Experimental protocols

4.1. Chemistry

Progress of the reactions was monitored using thin layer chromatography (TLC) sheets that precoated with UV fluorescent silica gel MERCK 60 F 254 that was visualized by UV lamp and I₂ vapor. Solvent system was chloroform: methanol (in different ratios). ¹H NMR and ¹³C-NMR spectra were carried out on Varian Gemini 300 MHz Spectrometer, at the Microanalytical Center, Cairo University, Egypt. Using TMS as internal standard and chemical





shifts were recorded in ppm on δ scales. IR spectra were recorded on a Shimadzu 435 Spectrometer, using KBr discs and values were represented in cm⁻¹. GC Mass spectra were run on Shimadzu QP-2010 spectrometer and Mass spectra were run on Hewlett Packard 5988 spectrometer at the Microanalytical Center, Cairo University, Egypt and National Research Center, Giza, Egypt. Melting points were determined on a Griffin instrument and are uncorrected. All products reported showed ¹H NMR spectra in agreement with the assigned structures. Elemental Analyses were performed by the Micro-analytical Laboratory, Cairo University, Egypt. Compound **1** was prepared according to reported method [23,24].

4.1.1. General procedure for the preparation of compounds **2a** and **2b**

A mixture of compound **1** (2.98 g, 0.01 mol) and potassium hydroxide (0.56 g, 0.01 mol) in ethanol (20 mL) was stirred at room temperature for 1 h, then the appropriate methyl or ethyl iodide (0.015 mol) was added and the mixture was heated under reflux for 24 h. The reaction mixture was cooled, filtered and the formed precipitate was washed with ethanol and crystallized from absolute ethanol to yield **2a** and **2b**.

4.1.1.1. (*RS*) 2-*Amino-4-(4-chlorophenyl)-7-methoxy-4H-chromene-*3-*carbonitrile* (**2a**). Yield 84%. mp 167–168 °C. IR (KBr): 3432, 3347 (forked, NH₂), 3055 (CH arom.), 2960 (CH aliph.), 2185 (CN) cm⁻¹; ¹H NMR (DMSO-*d*₆): δ 3.72 (s, 3H, OCH₃); 4.72 (s, 1H, C4H); 6.55 (s, 1H, ArH); 6.66 (d, J = 8.7 Hz, 1H, ArH); 6.90 (d, J = 8.7 Hz, 1H, ArH); 6.98 (s, 2H, NH₂, D₂O exchangeable); 7.18 (d, J = 8.1 Hz, 2H, ArH); 7.35 (d, J = 8.1 Hz, 2H, ArH) ppm; ¹³C NMR (DMSO- d_6): δ 39.9, 55.5, 76.5, 101.4, 111.7, 114.2, 119.5, 128.9, 129.2, 130.1, 133.0, 143.2, 149.1, 159.0, 159.5; EIMS: m/z (%) = 314 (M + 2-]⁺, 2.8), 312 (M-]⁺, 8.4), 201 (M-C₆H₄Cl-]⁺, 100), 75 (C₆H₃-]⁺, 24.3); Anal. Calcd. for C₁₇H₁₃ClN₂O₂ (312.76): C 65.29, H 4.19, N 8.96; Found: C 65.50, H 4.31, N 9.16%.

4.1.1.2. (*RS*)-2-*Amino*-4-(4-*chlorophenyl*)-7-*ethoxy*-4*H*-*chromene*-3*carbonitrile* (**2b**). Yield 85%. mp 181–182 °C. IR (KBr): 3424, 3332 (forked, NH₂), 3070 (CH arom.), 2976 (CH aliph.), 2196(CN) cm⁻¹; ¹H NMR (DMSO-*d*₆): δ 1.29 (t, 3H, CH₃); 3.98 (q, 2H, CH₂); 4.72 (s, 1H, C4H); 6.54 (s, 1H, ArH); 6.65 (d, 1H, ArH + 2H, NH₂, D₂O exchangeable); 6.90 (d, 1H, ArH); 7.18 (d, *J* = 8.4 Hz, 2H, ArH); 7.36 (d, *J* = 8.4 Hz, 2H, ArH) ppm; Anal. Calcd. for C₁₈H₁₅ClN₂O₂ (326.79): C 66.16, H 4.63, N 8.57; Found: C 66.42, H 4.42, N 8.43%.

4.1.2. General procedure for the preparation of compounds (**3a** and **3b**)

A mixture of compound **2a** and **2b** (0.01 mol) and triethyl orthoformate (20 mL) was heated under reflux for 4 h. The excess of orthoformate was removed under reduced pressure, and then the residue was washed with ethanol and crystallized from the appropriate solvent to give **3a** and **3b**.

4.1.2.1. Ethyl (RS) (ZE) N-4-(4-chlorophenyl)-3-cyano-7-methoxy-4H-chromen-2-ylformimidate (**3a**). Crystallized from absolute



Scheme 5. Synthesis of compounds 6a-d and 7a-f.

ethanol. Yield 88%. mp 131–132 °C. IR (KBr): 3068 (CH arom.), 2965 (CH aliph.), 2200 (CN), 1644 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.39 (t, 3H, CH₂CH₃); 3.80 (s, 3H, OCH₃); 4.43 (q, 2H, CH₂); 4.80 (s, 1H, C4H); 6.60 (s, 1H, ArH); 6.65 (d, *J* = 8.7 Hz, 1H, ArH); 6.84 (d, *J* = 8.7 Hz, 1H, ArH); 7.14 (d, *J* = 8.7 Hz, 2H, ArH); 7.29 (d, *J* = 8.7 Hz, 2H, ArH); 8.38 (s, 1H, N=CH) ppm; Anal. Calcd. for C₂₀H₁₇ClN₂O₃ (368.82): C 65.13, H 4.65, N 7.60; Found: C 65.28, H 4.66, N 7.80%.

4.1.2.2. (*ZE*) (*RS*)-*Ethyl* N-4-(4-*chlorophenyl*)-3-*cyano*-7-*ethoxy*-4*Hchromen*-2-*ylformimidate* (**3b**). Crystallized from methanol. Yield 61%. mp 135–136 °C. IR (KBr): 3067 (CH arom.), 2981 (CH aliph.), 2208 (CN), 1619 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.28–1.36 (m, 6H, 2CH₂CH₃); 3.94 (q, 2H, OCH₂CH₃); 4.35 (q, 2H, N=CHOCH₂CH₃); 4.71 (s, 1H, C4H); 6.49 (s, 1H, ArH); 6.54 (d, *J* = 8.4 Hz, 1H, ArH); 6.74 (d, *J* = 8.4 Hz, 1H, ArH); 7.06 (d, *J* = 8.4 Hz, 2H, ArH); 7.20 (d, *J* = 8.4 Hz, 2H, ArH); 8.30 (s, 1H, N=CH) ppm; ¹³C NMR (CDCl₃): δ 14.3, 14.4, 32.2, 62.9, 65.2, 62.8, 95.0, 103.3, 108.9, 123.3, 129.4, 129.7, 129.8, 131.5, 141.3, 156.5, 157.0, 157.3, 163.6; EIMS: *m/z* (%) = 384 (M + 2 ⁺, 7.31), 382 (M ⁺, 20.40), 271 (M – C₆H₄Cl ⁺, 100); Anal. Calcd. for C₂₁H₁₉ClN₂O₃ (382.85): C 65.88, H 5.00, N 7.32; Found: C 66.09, H 4.80, N 7.12%.

4.1.3. General procedure for the preparation of compounds 4a-d

A solution of **3a** and **3b** (0.01 mol) and either methylamine or hydrazine hydrate (99%, 0.01 mol) in absolute ethanol (30 mL) was stirred at room temperature for 1 h. The solid formed was filtered and crystallized from benzene to give **4a**–**d**.

4.1.3.1. (*ZE*) (*RS*)-5-(4-*Chlorophenyl*) 8-*methoxy*-3-*methyl*-3*H*-*chromeno*[2,3-*d*]*pyrimidin*-4(5*H*)-*imine* (**4a**). Yield 95%. mp 128–129 °C. IR (KBr): 3221 (NH), 2966 (CH aliph.), 1643 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.40 (s, 3H, N–CH₃); 3.77 (s, 3H, OCH₃); 4.86 (s, 1H, C5H); 6.59–7.28 (m, 7H, ArH + 1H, NH, D₂O exchangeable); 7.79 (s, 1H, C2H) ppm; EIMS: *m*/*z* (%) = 355 (M + 2⁻⁺, 10.41), 353 (M⁻⁺, 28.27), 338 (M–CH₃⁻⁺; 29.34), 105 (M–C₁₂H₁₁ClN₃O⁻⁺; 100); Anal. Calcd. for C₁₉H₁₆ClN₃O₂ (353.81): C 64.50, H 4.56, N 11.88; Found: C 64.63, H 4.49, N 11.58%.

4.1.3.2. (*ZE*) (*RS*)-5-(4-Chlorophenyl)-4,5-dihydro-4-imino -8methoxy-3H-chromeno[2,3-d]pyrimidin-3-amine (**4b**). Yield 93%. mp 187–189 °C. IR (KBr): 3336, 3156 (NH, NH₂), 2928 (CH aliph.), 1639 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.78 (s, 3H, OCH₃); 4.74 (br s, 2H, NH₂, D₂O exchangeable); 4.88 (s, 1H, C5H); 6.60–7.36 (m, 7H, ArH + 1H, NH, D₂O exchangeable); 8.09 (s, 1H, C2H) ppm Anal. Calcd. for C₁₈H₁₅ClN₄O₂ (354.80): C 60.94, H 4.26, N 15.79; Found: C 60.76, H 4.20, N 15.49%.

4.1.3.3. (*RS*) (*ZE*) -5-(4-Chlorophenyl)-8-ethoxy-3-methyl-3H-chromeno[2,3-d]pyrimidin-4(5H)-imine (**4c**). Yield 84%. mp 169–170 °C. IR (KBr): 3324 (NH), 3028 (CH arom.), 2976 (CH aliph.), 1640 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.40 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 3.59 (s, 3H, N–CH₃); 3.99 (q, *J* = 7.2 Hz, 2H, CH₂CH₃); 5.64 (s, 1H, C5H); 6.32–7.37 (m, 7H, ArH); 7.87 (s, 1H, NH, D₂O exchangeable); 8.17 (s, 1H, C2H) ppm; Anal. Calcd. for C₂₀H₁₈ClN₃O₂ (367.84): C 65.31, H 4.93, N 11.42; Found: C 65.54, H 4.81, N 11.12%.

4.1.3.4. (*RS*) (*ZE*) -5-(4-*Chlorophenyl*)-4-*imino*-4,5-*dihydro*-8ethoxy-3H-chromeno[2,3-d]pyrimidin-3-amine (**4d**). Yield 89%. mp 168–169 °C. IR (KBr): 3333, 3180 (NH, NH₂), 3087 (CH arom.), 2981 (CH aliph.), 1629 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.29 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 4.00 (q, *J* = 7.2 Hz, 2H, CH₂CH₃); 5.22 (s, 1H, C5H); 6.65–7.36 (m, 7H, ArH + 3H, NH, NH₂, D₂O exchangeable); 8.11 (s, 1H, C2H) ppm;.¹³C NMR (CDCl₃): δ 14.3, 25.7, 65.1, 95.0, 103.2, 108.9, 123.1, 129.4, 129.7, 129.8, 131.8, 141.3, 157.0, 157.2, 163.0, 164.0; EIMS: *m/z* (%) = 370 (M + 2 — ⁺, 11.63), 368 (M — ⁺, 33.67), 352 (M – NH₂ — ⁺, 100); Anal. Calcd. for C₁₉H₁₇ClN₄O₂ (368.83): C 61.88, H 4.65, N 15.19; Found: C 61.90, H 4.70, N 14.91%.

4.1.4. General procedure for the preparation of compounds **5a** and **5b**

A solution of **4b** and **4d** (0.001 mol) and triethyl orthoformate (10 mL) was heated under reflux for 6 h. The precipitated solid formed after cooling were filtered, washed with ethanol, dried and recrystallized from absolute ethanol to afford compounds **5a** and **5b**.

4.1.4.1. (RS)-12-(4-Chlorophenyl)-9-methoxy-12H-chromeno[3,2-e] [1,2,4]triazolo[1,5-c]pyrimidine (**5a**). Yield 86%. mp 215–216 °C. IR (KBr): 3073 (CH arom.), 2932 (CH aliph.), 1626 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.84 (s, 3H, OCH₃); 5.67 (s, 1H, C12H); 6.73 (d, J = 8.4 Hz, 1H, ArH); 6.85 (s, 1H, ArH); 7.06 (d, J = 8.4 Hz, 1H, ArH); 7.23 (d, J = 8.7 Hz, 2H, ArH); 7.27 (d, J = 8.7 Hz, 2H, ArH); 8.28 (s, 1H, C2H); 9.14 (s, 1H, C5H) ppm; Anal. Calcd. for C₁₉H₁₃ClN₄O₂ (364.79): C 62.56, H 3.59, N 15.36; Found: C 62.70, H 3.40, N 15.56%.

4.1.4.2. (*RS*)-12-(4-Chlorophenyl)-9-ethoxy-12H-chromeno[3,2-e] [1,2,4]triazolo[1,5-c]pyrimidine (**5b**). Yield 81%. mp 177–178 °C. IR (KBr): 3077 (CH arom.), 2979 (CH aliph.), 1626 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.44 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 4.06 (q, *J* = 7.2 Hz, 2H, CH₂CH₃); 5.67 (s, 1H, C12H); 6.72 (d, *J* = 8.4 Hz, 1H, ArH); 6.84 (s, 1H, ArH); 7.05 (d, *J* = 8.4 Hz, 1H, ArH); 7.24 (d, *J* = 8.1 Hz, 2H, ArH); 7.29 (d, *J* = 8.1 Hz, 2H, ArH); 8.28 (s, 1H, C2H); 9.14 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃): δ 29.8, 56.0, 107.2, 111.6, 118.5, 127.6, 129.1, 129.6, 129.9, 130.8, 132.0, 144.1, 147.2, 147.9, 155.5, 160.8; EIMS: *m/z* (%) = 380 (M + 2⁻¹; 8.27), 378 (M⁻¹; 22.04), 267 (M– C₆H₄Cl⁻¹; 100); Anal. Calcd. for C₂₀H₁₅ClN₄O₂ (378.82): C 63.41, H 3.99, N 14.79; Found: C 63.30, H 3.80, N 14.49%.

4.1.5. General procedure for the preparation of compounds **5c** and **5d**

A mixture of **4b** and **4d** (0.001 mol) and acetic anhydride (10 mL) was heated under reflux for 6 h. The precipitated solid formed after cooling were filtered, washed with ethanol, dried and recrystallized to afford compounds **5c** and **5d**.



Fig. 2. The suggested mechanism for the reaction of compounds 3a and 3b with excess hydrazine hydrate.

4.1.5.1. (RS)-12-(4-Chlorophenyl)-2-methyl-9-methoxy-12H-chromeno[3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**5**c). Crystallized from methanol. Yield 85%. mp 231–232 °C. IR (KBr): 3061 (CH arom.), 2960 (CH aliph.), 1628 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 2.57 (s, 3H, C2CH₃); 3.84 (s, 3H, OCH₃); 5.65 (s, 1H, C12H); 6.73 (d, *J* = 8.7 Hz, 1H, ArH); 6.85 (s, 1H, ArH); 7.07 (d, *J* = 8.7 Hz, 1H, ArH); 7.24 (d, *J* = 8.4 Hz, 2H, ArH); 7.31 (d, *J* = 8.4 Hz, 2H, ArH); 9.01 (s, 1H, C5H) ppm; EIMS: *m*/*z* (%) = 380 (M + 2⁻⁺; 7.06), 378 (M⁻⁺⁻⁺;

19.72), 267 (M–C₆H₄Cl––)⁺, 100); Anal. Calcd. for $C_{20}H_{15}ClN_4O_2$ (378.82): C 63.41, H 3.99, N 14.79; Found: C 63.50, H 4.10, N 14.70%.

4.1.5.2. (RS)-12-(4-Chlorophenyl)-9-ethoxy-2-methyl-12H-chromeno [3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**5d**). Crystallized from methanol: chloroform (1:1). Yield 89%. mp 214–215 °C. IR (KBr): 3060 (CH arom.), 2981 (CH aliph.), 1626 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.43 (t, J = 7.2 Hz, 3H, CH₂CH₃); 2.55 (s, 3H, C2CH₃); 4.05 (q,



Fig. 3. The suggested mechanism for the reaction of 4b and 4d with ethyl ethoxymethylene cyanoacetate.



Fig. 4. The suggested mechanism for the reaction of 4b and 4d with acetyl acetone.

J = 7.2 Hz, 2H, CH₂–CH₃); 5.63 (s, 1H, C12H); 6.71 (d, *J* = 8.4 Hz, 1H, ArH); 6.83 (s, 1H, ArH); 7.05 (d, *J* = 8.4 Hz, 1H, ArH); 7.21 (d, *J* = 8.4 Hz, 2H, ArH); 7.29 (d, *J* = 8.4 Hz, 2H, ArH); 9.00 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃): δ 14.2, 29.8, 65.3, 107.3, 110.6, 118.2, 127.2, 129.4, 129.8, 130.0, 131.3, 142.1, 147.8, 153.5, 154.8, 157.3, 160.4, 171.6; EIMS: *m*/*z* (%) = 394 (M + 2 $^{-1}$ ⁺, 8.34), 392 (M $^{-1}$ ⁺, 22.62),

281 (M–C₆H₄Cl⁻⁻⁺, 100); Anal. Calcd. for C₂₁H₁₇ClN₄O₂ (392.85): C 64.21, H 4.36, N 14.26; Found: C 64.10, H 4.50, N 13.96%.

4.1.6. General procedure for the preparation of compounds **5***e*–**h** A mixture of **4b** and **4d** (0.001 mol) and either diethyl oxalate or ethyl cyanoacetate (0.001 mol) in absolute ethanol (15 mL) was



Fig. 5. The suggested mechanism for the reaction of 4b and 4d with ethyl cyanoacetate.

Results of *in vitro* cytotoxic activity of reference drug and the tested compounds on human breast tumor cell line (MCF-7).

| Cpd no | IC ₅₀ μM |
|------------|---------------------|
| Colchicine | 0.013 |
| 4a | 0.007 |
| 4b | 0.008 |
| 5a | 0.011 |
| 5c | 0.033 |
| 5f | 0.007 |
| 5g | 0.039 |
| 5h | 0.011 |
| 6d | 0.029 |
| 7c | 0.008 |
| 7f | 0.008 |

 IC_{50} : concentration of a drug that is required for 50% inhibition.

heated under reflux for 4 h. After cooling, the solid formed was collected and crystallized from the appropriate solvent to afford compounds **5e**–**h**.

4.1.6.1. (*RS*)-*Ethyl* 12-(4-*chlorophenyl*)-9-*methoxy*-12H-*chromeno*[3,2-*e*][1,2,4]*triazolo*[1,5-*c*]*pyrimidine*-2-*carboxylate* (*5e*). Crystallized from a mixture of methanol: benzene (3:1). Yield 75%. mp 229–230 °C. IR (KBr): 3067 (CH arom.), 2956 (CH aliph.), 1741 (C=O), 1626 (C=N) cm⁻¹; ¹H NMR (DMSO-*d*₆): δ 1.32 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 3.78 (s, 3H, OCH₃); 4.38 (q, *J* = 7.2 Hz, 2H, CH₂CH₃); 5.81 (s, 1H, C12H); 6.77–7.34 (m, 7H, ArH); 9.75 (s, 1H, C5H) ppm; EIMS: *m/z* (%) = 438 (M + 2 ⁺, 11.99), 436 (M ⁺, 31.30), 325 (M - C₆H₄Cl ⁺, 100); Anal. Calcd. for C₂₂H₁₇ClN₄O₄ (436.85): C 60.49, H 3.92, N 12.83; Found: C 60.69, H 4.10, N 12.86%.

4.1.6.2. (RS)-Ethyl 12-(4-chlorophenyl)-9-ethoxy-12H-chromeno[3,2e][1,2,4]triazolo[1,5-c]pyrimidine-2-carboxylate (**5f**).

Crystallized from a mixture of methanol: benzene (3:1). Yield 60%. mp 279–280 °C. IR (KBr): 3063 (CH arom.), 2983 (CH aliph.), 1745 (C=O), 1627 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.42–1.50 (m, 6H, 2 CH₃); 4.06 (q, *J* = 7.2 Hz, 2H, OCH₂–CH₃); 4.55 (q, *J* = 7.2 Hz, 2H, COOCH₂CH₃); 5.78 (s, 1H, C12H); 6.73–7.36 (m, 7H, ArH); 9.18 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃): δ 13.6, 14.3, 29.8, 59.1, 56.1, 107.3, 110.6, 118.7, 127.2, 129.4, 129.8, 130.0, 131.3, 141.1, 150.5, 151.9, 153.5, 154.8, 157.3, 167.0, 171.6; Anal. Calcd. for C₂₃H₁₉ClN₄O₄ (450.87): C 61.27, H 4.25, N 12.43; Found: C 61.13, H 4.28, N 12.61%.

4.1.6.3. (*RS*)-2-(12-(4-Chlorophenyl)-9-methoxy-12H-chromeno[3,2-e] [1,2,4]triazolo[1,5-c]pyrimidin-2-yl)acetonitrile (**5g**). Crystallized from benzene. Yield 80%. mp 207–208 °C. IR (KBr): 3072 (CH arom.), 2966 (CH aliph.), 2259 (CN), 1630 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.85 (s, 3H, OCH₃); 4.00 (s, 2H, CH₂CN); 5.64 (s, 1H, C12H); 6.75 (d, *J* = 8.7 Hz, 1H, ArH); 6.86 (s, 1H, ArH); 7.06 (d, *J* = 8.7 Hz, 1H, ArH); 7.21 (d,



Fig. 6. Structures of compounds 4a and 5f.



Fig. 7. Structures of compounds 4b, 7c and 7f.

J = 8.4 Hz, 2H, ArH); 7.26 (d, J = 8.4 Hz, 2H, ArH); 9.08 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃): δ 18.4, 29.5, 56.0, 107.4, 110.5, 114.9, 118.2, 127.9, 129.0, 129.4, 129.8, 130.4, 131.3, 141.1, 147.8, 153.5, 155.2, 160.5, 171.6; EIMS: m/z (%) = 405 (M + 2 $-^+$; 9.51), 403 (M $-^+$; 25.87), 292 (M $-C_6H_4Cl_{--}^+$; 100); Anal. Calcd. for C₂₁H₁₄ClN₅O₂ (403.82); C 62.46, H 3.49, N 17.34; Found: C 62.51, H 3.59, N 17.06%.

4.1.6.4. (RS)-2-(12-(4-Chlorophenyl)-9-ethoxy-12H-chromeno[3,2-e] [1,2,4]triazolo[1,5-c]pyrimidin-2-yl)acetonitrile (**5h**).

Crystallized from benzene Yield 78%. mp 218–219 °C. IR (KBr): 3060 (CH arom.), 2982 (CH aliph.), 2251 (CN), 1628 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.44 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 4.00 (s, 2H, CH₂CN); 4.08 (q, *J* = 7.2 Hz, 2H, CH₂–CH₃); 5.63 (s, 1H, C12H); 6.73 (d, *J* = 8.4 Hz, 1H, ArH); 6.84 (s, 1H, ArH); 7.05 (d, *J* = 8.4 Hz, 1H, ArH); 7.21 (d, *J* = 8.7 Hz, 2H, ArH); 7.29 (d, *J* = 8.7 Hz, 2H, ArH); 9.08 (s, 1H, C5H) ppm; Anal. Calcd. for C₂₂H₁₆ClN₅O₂ (417.85): C 63.24, H 3.86, N 16.76; Found: C 63.23, H 4.09, N 16.53%.

4.1.7. General procedure for the preparation of compounds 6a-d

To a solution of **4b** and **4d** (0.004 mol) in dioxane (15 mL) was added the appropriate acid chloride (0.004 mol), followed by the addition of triethylamine (0.40 g, 0.004 mol). The reaction mixture was stirred at room temperature for 24 h then poured into ice-cold water. The solid product that formed was filtered, washed with water, dried and crystallized from a mixture of ethanol: chloroform (3:1) to give **6a**–**d**.

4.1.7.1. (*RS*)-2-(*Chloromethyl*)-12-(4-*chlorophenyl*)-9-*methoxy*-12*Hchromeno*[3,2-*e*][1,2,4]*triazolo*[1,5-*c*]*pyrimidine* (**6***a*). Yield 80%. mp 200–201 °C. IR (KBr): 3056 (CH arom.), 2959 (CH aliph.), 1628 (C= N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.85 (s, 3H, OCH₃); 4.73 (s, 2H, CH₂); 5.68 (s, 1H, C12H); 6.75 (d, *J* = 8.7 Hz, 1H, ArH); 6.86 (s, 1H, ArH); 7.07 (d, *J* = 8.7 Hz, 1H, ArH); 7.23 (d, *J* = 8.4 Hz, 2H, ArH); 7.27 (d,



Fig. 8. Structure of compound 5a and 5h.

Table 1



Fig. 9. Structure of compound 6d.

J = 8.4 Hz, 2H, ArH); 9.07 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃): δ 29.8, 45.1, 56.2, 107.2, 110.6, 118.5, 127.7, 129.4, 129.6, 129.8, 130.9, 132.0, 141.1, 147.8, 153.5, 155.4, 160.4, 171.6; Anal. Calcd. for C₂₀H₁₄Cl₂N₄O₂ (413.26): C 58.13, H 3.41, N 13.56; Found: C 58.05, H 3.71, N 13.91%.

4.1.7.2. (*RS*)-12-(4-Chlorophenyl)-2-(chloropropyl)-9-methoxy-12*H*-chromeno[3,2-*e*][1,2,4]triazolo[1,5-*c*]pyrimidine (**6***b*). Yield 81%. mp 175–176 °C. IR (KBr): 3057 (CH arom.), 2951 (CH aliph.), 1628 (C= N) cm⁻¹; ¹H NMR (CDCl₃): δ 2.22 (pentet, *J* = 7.5 Hz, 2H, *CH*₂); 3.07 (t, *J* = 7.5 Hz, 2H, *CH*₂); 3.66 (t, *J* = 7.5 Hz, 2H, *CH*₂); 3.84 (s, 3H, OCH₃); 5.67 (s, 1H, C12H); 6.75 (d, *J* = 8.4 Hz, 1H, ArH); 6.85 (s, 1H, ArH); 7.07 (d, *J* = 8.4 Hz, 1H, ArH); 7.19 (d, *J* = 8.7 Hz, 2H, ArH); 7.29 (d, *J* = 8.7 Hz, 2H, ArH); 9.01 (s, 1H, C5H) ppm; Anal. Calcd. for C₂₂H₁₈Cl₂N₄O₂ (441.31): C 59.88, H 4.11, N 12.70; Found: C 60.09, H 4.25, N 12.50%.

4.1.7.3. (*RS*)-2-(*Chloromethyl*)-12-(4-*chlorophenyl*)-9-*ethoxy*-12*Hchromeno*[3,2-*e*][1,2,4]*triazolo*[1,5-*c*]*pyrimidine* (**6***c*). Yield 87%. mp 197–198 °C. IR (KBr): 3059 (CH arom.), 2980 (CH aliph.), 1628 (C= N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.44 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 4.06 (q, *J* = 7.2 Hz, 2H, CH₂-CH₃); 4.73 (s, 2H, CH₂Cl); 5.67 (s, 1H, C12H); 6.73 (d, *J* = 8.4 Hz, 1H, ArH); 6.84 (s, 1H, ArH); 7.05 (d, *J* = 8.4 Hz, 1H, ArH); 7.20 (d, *J* = 8.7 Hz, 2H, ArH); 7.29 (d, *J* = 8.7 Hz, 2H, ArH); 9.06 (s, 1H, C5H) ppm; EIMS: *m/z* (%) = 431 (M + 4 $^{-1}$; 9.96), 429 (M + 2 $^{-1}$; 25.76), 427 (M $^{-1}$; 39.64), 315 (M-C₆H₄Cl $^{-1}$; 100); Anal. Calcd. for C₂₁H₁₆Cl₂N₄O₂ (427.28): C 59.03, H 3.77, N 13.11; Found: C 59.09, H 3.89, N 13.02%.

4.1.7.4. (*RS*)-12-(4-Chlorophenyl)-2-(chloropropyl)-9-ethoxy-12*H*-chromeno[3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**6d**). Yield 88%. mp 171–172 °C. IR (KBr): 3071 (CH arom.), 2974 (CH aliph.), 1628 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.44 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 2.25 (pentet, *J* = 7.2 Hz, 2H, CH₂); 3.06 (t, *J* = 7.2 Hz, 2H, CH₂); 3.66 (t, *J* = 7.2 Hz, 2H, CH₂); 4.06 (q, *J* = 7.2 Hz, 2H, CH₂–CH₃); 5.65 (s, 1H, C12H); 6.72 (d, *J* = 8.4 Hz, 1H, ArH); 6.84 (s, 1H, ArH); 7.05 (d, *J* = 8.4 Hz, 1H, ArH); 7.30 (d, *J* = 8.7 Hz, 2H, ArH); 9.01 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃): δ 14.3, 29.6, 56.1, 107.1, 111.6, 118.2, 127.0, 127.5, 128.4, 129.0, 129.4129.8, 130.0, 131.3, 136.9, 141.1, 148.3, 153.9, 154.8, 157.7, 171.6; Anal. Calcd. for



Fig. 10. Structure of compounds 5c and 5g.

C₂₃H₂₀Cl₂N₄O₂(455.34): C 60.67, H 4.43, N 12.30; Found: C 60.51, H 4.51, N 12.51%.

4.1.8. General procedure for the preparation of compounds 7a-f

To a mixture of **4b** and **4d** (0.004 mol) and the respective arylidene malononitrile (0.004 mol) in absolute ethanol (25 mL), piperidine (0.2 mL) was added. The reaction mixture was heated under reflux for 2 h. The precipitate that formed was collected by filtration while hot and crystallized from a mixture of methanol: chloroform (3:1) to afford **7a**–**f**.

4.1.8.1. (RS)-12-(4-Chlorophenyl)-9-methoxy-2-phenyl-12H-chromeno[3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**7a**). Yield 75%. mp 264–265 °C. IR (KBr): 3058 (CH arom.), 2964 (CH aliph.), 1627 (C= N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.85 (s, 3H, OCH₃); 5.75 (s, 1H, C12H); 6.73–8.26 (m, 12H, ArH); 9.09 (s, 1H, C5H) ppm; Anal. Calcd. for C₂₅H₁₇ClN₄O₂ (440.88): C 68.11, H 3.89, N 12.71; Found: C 68.07, H 3.88, N 13.04%.

4.1.8.2. (*RS*)-2,12-bis-(4-Chlorophenyl)-9-methoxy-12H-chromeno [3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**7b**). Yield 82%. mp 280–281 °C. IR (KBr): 3078 (CH arom.), 2970 (CH aliph.), 1625 (C=N) cm⁻¹; ¹H NMR (CDCl₃) δ : 3.85 (s, 3H, OCH₃); 5.72 (s, 1H, C12H); 6.74 (d, *J* = 8.7 Hz, 1H, ArH); 6.86 (s, 1H, ArH); 7.08 (d, *J* = 8.7 Hz, 1H, ArH); 7.21 (d, *J* = 8.4 Hz, 2H, ArH); 7.36 (d, *J* = 8.4 Hz, 2H, ArH); 7.48 (d, *J* = 8.4 Hz, 2H, ArH); 8.20 (d, *J* = 8.4 Hz, 2H, ArH); 9.08 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃) δ : 29.8, 56.0, 107.3, 110.5, 118.2, 127.9, 128.4, 129.4, 129.8, 130.4, 131.2, 133.8, 134.7, 141.1, 148.0, 153.5, 155.2, 160.5, 171.6; Anal. Calcd. for C₂₅H₁₆Cl₂N₄O₂ (475.33): C 63.17, H 3.39, N 11.79; Found: C 63.44, H 3.59, N 11.94%.

4.1.8.3. (*RS*)-12-(4-Chlorophenyl)-9-methoxy-2-(4-methoxyphenyl)-12H-chromeno[3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**7c**). Yield 69%. mp 256–257 °C. IR (KBr): 3060 (CH arom.), 2933 (CH aliph.), 1619 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 3.85 (s, 3H, OCH₃); 3.89 (s, 3H, OCH₃); 5.77 (s, 1H, C12H); 6.76 (d, *J* = 9 Hz, 1H, ArH); 6.87 (s, 1H, ArH); 7.02 (d, *J* = 9 Hz, 1H, ArH); 7.07 (d, *J* = 8.4 Hz, 2H, ArH); 7.21 (d, *J* = 8.4 Hz, 2H, ArH); 7.37 (d, *J* = 8.7 Hz, 2H, ArH); 8.21 (d, *J* = 8.7 Hz, 2H, ArH); 9.06 (s, 1H, C5H) ppm; Anal. Calcd. for C₂₆H₁₉ClN₄O₃ (470.91); C 66.31, H 4.07, N 11.90; Found: C 66.20, H 4.20, N 12.20%.

4.1.8.4. (*RS*)-12-(4-Chlorophenyl)-9-ethoxy-2-phenyl-12H-chromeno [3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**7d**). Yield 87%. mp 219–220 °C. IR (KBr): 3058 (CH arom.), 2982 (CH aliph.), 1625 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.45 (t, J = 7.2 Hz, 3H, CH₂CH₃); 4.06 (q, J = 7.2 Hz, 2H, CH₂-CH₃); 5.74 (s, 1H, C12H); 6.71–8.28 (m, 12H, ArH); 9.08 (s, 1H, C5H) ppm; ¹³C NMR (CDCl₃) δ : 14.3, 29.8, 56.0, 107.3, 110.8, 118.2, 127.0, 127.9, 128.4, 129.4, 129.8, 130.4, 131.2, 136.7, 141.1, 153.5, 154.8, 155.2, 157.3, 160.6, 171.6; Anal. Calcd. for C₂₆H₁₉ClN₄O₂ (454.91): C 68.65, H 4.21, N 12.32; Found: C 68.61, H 4.30, N 12.60%.

4.1.8.5. (*RS*)-2,12-*bis*-(4-*Chlorophenyl*)-9-*ethoxy*-12*H*-*chromeno* [3,2-*e*][1,2,4]*triazolo*[1,5-*c*]*pyrimidine* (**7e**). Yield 88%. mp 264–265 °C. IR (KBr): 3067 (CH arom.), 2983 (CH aliph.), 1627 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.45 (t, J = 7.2 Hz, 3H, CH₂*CH*₃); 4.06 (q, J = 7.2 Hz, 2H, CH₂–CH₃); 5.76 (s, 1H, C12H); 6.74 (d, J = 8.4 Hz, 1H, ArH); 6.85 (s, 1H, ArH); 7.08 (d, J = 8.4 Hz, 2H, ArH); 7.22 (d, J = 8.4 Hz, 2H, ArH); 7.35 (d, J = 8.4 Hz, 2H, ArH); 7.48 (d, J = 8.4 Hz, 2H, ArH); 8.21 (d, J = 8.4 Hz, 2H, ArH); 9.09 (s, 1H, C5H) ppm; EIMS: m/z (%) = 493 (M + 4⁻¹, 1.05), 491(M + 2⁻¹, 6.86), 489 (M⁻¹, 1.056), 488 (M – 1⁻¹, 1768), 459 (M–C₂H₅⁻¹, 100); Anal. Calcd. for C₂₆H₁₈Cl₂N₄O₂ (489.35): C 63.81, H 3.71, N 11.45; Found: C 64.02, H 3.92, N 11.66%. 4.1.8.6. (*RS*)-12-(4-Chlorophenyl)-9-ethoxy-2-(4-methoxyphenyl)-12H-chromeno[3,2-e][1,2,4]triazolo[1,5-c]pyrimidine (**7f**). Yield 73%. mp 174–175 °C. IR (KBr): 2969 (CH aliph.), 1608 (C=N) cm⁻¹; ¹H NMR (CDCl₃): δ 1.45 (t, *J* = 7.2 Hz, 3H, CH₂CH₃); 3.89 (s, 3H, OCH₃); 4.07 (q, *J* = 7.2 Hz, 2H, CH₂–CH₃); 5.72 (s, 1H, C12H); 6.74–7.27 (m, 7H, ArH); 7.36 (d, *J* = 8.7 Hz, 2H, ArH); 8.20 (d, *J* = 8.7 Hz, 2H, ArH); 9.05 (s, 1H, C5H) ppm; Anal. Calcd. for C₂₇H₂₁ClN₄O₃ (484.93): C 66.87, H 4.36, N 11.55; Found: C 66.92, H 4.72, N 11.19%.

4.2. Antitumor activity

The breast tumor cell line was obtained frozen in liquid nitrogen $(-180 \,^{\circ}\text{C})$ from the American Type Culture Collection (ATCC) and was maintained at the National Cancer Institute, Cairo, Egypt, by serial sub culturing. Colchicine was used in this experiment as a positive control. The tested compounds were dissolved in 20% DMSO in concentration 1 mg/mL.

Serial dilutions were made reaching final concentration of the compounds to 0, 5, 12.5, 25, 50 μ g/mL. All chemicals used in this study are of high analytical grade. They either obtained from (Sigma–Alderich or Biorad).

4.2.1. Measurement of potential cytotoxic activity

The cytotoxic activity was measured *in vitro* on human breast tumor cell line (MCF-7) using Sulforhodamine-B stain (SRB) assay applying the method of Skehan et al. [25]. The results of *in vitro* cytotoxic activity experiments are presented in (Table 1).

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