Synthesis and Structure of New Substituted Pyrimidinone with Unsaturated Side Chain

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Abstract—Melting of a mixture of 5-substituted 2,4-dimethyl-1,6-dihydropyrimidin-6-ones with cinnamic aldehyde, 1-methylindoline-2,3-dione and 6-methoxy-2-chloroquinoline-3-carbaldehyde in the presence of ZnCl₂ led to the formation of substituted pyrimidines with conjugated bonds in the position 2. The structure of synthesized compounds as 2-isomers was confirmed by 2D ¹H NMR NOESY data.

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The interest to the synthesis of pyrimidines with a system of unsaturated double bonds in the side chain is due both to the high biological activity of styryl derivatives of pyrimidines [1, 2] and to the optical absorption and emission properties of heteroconjugated pyrimidines [3]. Here we report on a series of new substituted pyrimidines containing in the position 2 of the pyrimidine ring 1-methyl-2-oxo-3-indolylidenemethyl, 2-(3-quinolyl)ethylene, and 1-phenyl-1,3-butadiene groups. The initial 2,4-dimethyl-1,6-dihydropyrimidin-6-ones **1a–1f** were prepared by the reaction of 2-substituted ethyl 3-oxobutanoates [4] with acetamide in anhydrous methanol in the presence of sodium methylate. Pyrimidines **1a**, **1c**, and **1e–1g** were synthesized in [5], **1b**, in [6].

The structure of 5-butyl-2,4-dimethyl-1,6-dihydropyrimidin-6-one **1c** was proved by X-ray diffraction (XRD) analysis (Figs. 1, 2).

The heterocyclic fragment $N^{l}C^{2}N^{3}C^{4}C^{5}C^{6}$ is practically planar with a maximum deviation from the mean square plane 0.0091(3) Å. The molecules of compound **1c** form in the crystal centrosymmetric cyclic dimers (Fig. 2) owing to intermolecular hydrogen bonds $N^{l}-H^{l}\cdots O^{7}$ [the length of the donoracceptor bond is 2.804(2) Å].

The reaction of 5-substituted 2,4-dimethyl-1,6-dihydropyrimidin-6-ones 1a-1g with 1-methyl-indoline-2,3dione 2, 6-methoxy-2-chloro-3-quinolinecarbaldehyde 3, and cinnamic aldehyde 4 in the presence of $ZnCl_2$



Fig. 1. General arrangement of the molecule of 5-butyl-2,4dimethyl-1,6-dihydropyrimidin-6-one **1c** by XRD data.



Fig. 2. Fragment of crystal packing of 5-butyl-2,4-dimethyl-1,6-dihydropyrimidin-6-one **1c** showing the formation of centrosymmetric dimer.





1, R = (CH₂)₂Me (**a**), CH₂CH=CH₂ (**b**), (CH₂)₃Me (**c**), CH(CH₂)₄ (**d**), CH₂CH₂CHMe₂ (**e**), (CH₂)₅Me (**f**), Bn (**g**); **5**, R = (CH₂)₃Me (**a**), CH(CH₂)₄ (**b**), Bn (**c**); 7, R = CH₂CH=CH₂ (**a**), (CH₂)₃Me (**b**), CH₂CH₂CHMe₂ (**c**).

afforded 3-substituted 1-methylindolin-2-ones 5a-5c, substituted 2-(3-quinolyl)-pyrimidone 6, and 5-substituted buta-1,3-dienyl-1,6-dihydropyrimidin-6-ones 7a-7c (Scheme 1).

The condensation of 6-methoxy-2-chloro-3quinolinecarbaldehyde with pyrimidinone 1a is accompanied with a hydrolytic substitution of the chlorine atom in the position 2 of the quinolone ring with the formation of quinolinol derivative 6. The consideration of computer molecular models of compounds 5a-5c (program product Cambridge Soft Corporation Chem. 3D 5.0) did not give unambiguous indication of the prevailing formation of one among regioisomeros (2- or 4-styryl-substituted pyrimidine), yet such preliminary indications were obtained for geometric isomers. In *trans*-isomers of compounds **5a**–**5c** unfavorable steric interactions were revealed between the atoms H^4 of isatin benzene ring and HN^1 of pyrimidine in case of coplanar location of the pyrimidine and isatin rings, whereas in the *cis*-isomers such interactions are less significant (Fig. 3).

The comparison of the reactivity of groups 2- and 4 $-CH_3$ of 1,6-dihydropyrimidin-6-one in the reactions with carbonyl compounds where the methyl group



Fig. 3. Molecular models of compounds 5a-5c.

plays the role of a nucleophilic site with a partial negative sign shows the greater chemical activity of the 2-CH₃ group due to more efficient delocalization of the negative sign. The discussed assumptions were proved by the study of NMR spectra of compounds 5a, 5c, and 7b.

As show the 2D NOESY NMR spectra of compounds **5a**, **5c**, and **7b** the reaction proceeds at the 2-methyl group of the pyrimidine ring. It is confirmed by the NOE between the 6-methyl and the group α -CH₂ of substituents in the position 5 of the ring. Moreover, the presence of intensive NOE between the proton of the exocyclic double bond and H⁴ of the aromatic ring of 1- methylisatin shows that compounds **5a** and **5c** have Z-configuration.

EXPERIMENTAL

IR spectra were recorded on a spectrophotometer Nicolet Avatar 330 in mineral oil. ¹H NMR spectra were registered on a spectrometer Varian Mercury-300 at operating frequency 300 MHz in a mixture DMSO d_6 -CCl₄, 1 : 3, internal reference TMS. Elemental analysis was carried out on an automatic analyzer EA 3000 Eurovector (Italy). X-ray diffraction analysis was performed on a single crystal of compound 1c at room temperature using automatic diffractometer Enraf-Nonius CAD-4 (graphite monochromator, MoK_{q} radiation, $\theta/2\theta$ -scanning). The structure was solved by the direct method, hydrogen atoms coordinates were partially found from the difference Fourier syntheses. The coordinates of hydrogen atoms of methyl groups were geometrically calculated and refined in the *rider* model with the following restrictions: length of the bond C–H 0.96 Å, $U_{iso}(H) = 1.5 U_{eq}(C)$. The structure was refined by full-matrix least-squares method in anisotropic approximation for nonhydrogen atoms and in isotropic approximation for hydrogen atoms. All structural computations were carried out using SHELXTL software [6]. TLC was performed on a Silufol UV-254 plates, eluent ethanol-chloroform, 1:10, development in iodine vapor.

Crystallographic data of compound **1c** are deposited in the Cambridge Crystallographic Data Centre (CCDC 1536234).

5-Substituted 2,4-dimethyl-1,6-dihydropyrimidin-6-ones. General procedure. To a solution of alkali metal ethylate prepared from 0.46 g (0.02 mol) of metal sodium or potassium and 40 mL of anhydrous

Table 1. Main crystallographic characteristics andexperimental data for 5-butyl-2,4-dimethyl-1,6-dihydro-6-pyrimidinone 1c

Parameter	Value
Empirical formula	$C_{10}H_{16}N_2O$
Molecular mass	180.25
Crystal system	Triclinic
Space group	<i>P</i> -1
a, b, c, Å	4.8866(10), 10.088(2), 11.577(2)
α, β, γ, deg	114.58(3), 96.59(3), 90.66(3)
<i>V</i> , Å ³	514.4(2)
Ζ	2
$D_{\rm calc}, {\rm g/cm}^3$	1.164
$\mu(MoK_{\alpha}), mm^{-1}$	0.076
F(000)	196
Crystal size, mm	0.18×0.23×0.35
Temperature, K	293
Radiation, Å	0.71073
$\Theta_{\min}, \theta_{\max}, deg$	2.0, 30.0
Scan range	<i>−6≤h=6, −14=k=14, −16=l=16</i>
Number of measured reflections	5966
Observed reflections with $I > 3.0\sigma(I)$	1344
N _{ref} , N _{par}	2983, 160
R, wR_2, S	0.0593, 0.1609, 0.96

ethanol was added 0.95 g (0.01 mol) of preliminary dried acetamidine hydrochloride and 0.01 mol of 2substituted ethyl 3-oxobutanoate. The mixture was boiled for 6 h, alcohol was distilled off to dryness, the residue was dissolved in 10 mL of water and acidified with AcOH to pH 6. On cooling the separated crystals were filtered off, dried, and recrystallized from aqueous ethanol.

5-Allyl-2,4-dimethyl-1,6-dihydropyrimidin-6-one (1c), mp 151–152°C [7]. ¹H NMR spectrum, δ, ppm: 2.14 s (3H, CH₃), 2.22 (3H, CH₃), 3.13 d.t (2H, CH₂, *J* 6.0, 1.6 Hz), 4.93 d.d.t (1H, CH₂, *J* 10.1, 1.8, 1.6 Hz), 4.96 d.d.t (1H, CH₂, *J* 17.1, 1.8, 1.6 Hz), 5.76 d.d.t (1H, CH, *J* 17.1, 10.1, 6.0 Hz), 12.14 br.s [1H, NH (OH)].

2,4-Dimethyl-5-cyclopentyl-1,6-dihydropyrimidin-6-one (1d) was obtained from ethyl 2-cyclopentyl3-oxobutanoate. Yield 39%, mp 180–182°C, $R_{\rm f}$ 0.51 (2-propanol–dichloroethane, 1 : 9), IR spectrum, v, cm⁻¹: 3290, 3130 (NH), 1850, 1639 (CO), 1610 (C=C– C=N). ¹H NMR spectrum, δ , ppm: 1.51–1.66 m (4H), 1.78–2.02 m (4H) and 2.90–3.02 m (1H, C₅H₉), 2.18 s (3H) and 2.19 s (3H, 2CH₃), 11.95 br.s (1H, NH). ¹³C NMR spectrum, δ , ppm: 20.2 (CH₃), 21.1 (CH₃), 26.0 (2CH₃), 28.8 (2CH₃), 37.7 (CH), 122.8, 153.9, 158.0, 161.4. Found, %: C 68.89; H 8.51; N 14.65. C₁₁H₁₆N₂O. Calculated, %: C 68.72; H 8.39; N 14.57.

2-Substituted pyrimidin-6-ones (5a–5c, 6, 7a–7c) (general procedure). A mixture of 0.01 mol of 5-substituted 2,4-dimethyl-1,6-dihydro-6-pyrimidinone, 0.01 mol of *N*-methylisatin 2, substituted 3-quinolinealdehyde 3 or cinnamic aldehyde 4, 100 mg of anhydrous ZnCl₂ was heated at 170–180°C for 1 h. On cooling the product was triturated with ethanol, filtered off, and dried. The compounds obtained were purified by recrystallization from alcohol or by hot filtration of a dispersion of the compound in alcohol. Compounds 5a–5c are red-orange powders, 6, and 7a–7c are yellow or light yellow powders.

3-[(Z)-1-(5-Butyl-4-methyl-6-oxo-1,6-dihydropyrimidin-2-yl)methylidene]-1-methylindolin-2-one (5a) was prepared from 1-methylindoline-2.3-dione and pyrimidine 1a. Yield 60%, mp 243–245°C, $R_{\rm f}$ 0.75 (ethanol-dichloroethane, 1 : 1). IR spectrum, v, cm^{-1} : 1701, 1674, 1641 (CO), 1602 (C=C-C=N). ¹H NMR spectrum, δ, ppm: 0.97 t (3H, CH₃, J 6.9 Hz), 1.35-1.52 m (4H, <u>CH₂CH₂CH₃</u>), 2.31 s (3H, 4-CH₃), 2.44-2.50 m (2H, 5-CH₂CH₂), 3.36 s (3H, NCH₃), 7.02 br.d (1H, C₆H₄, H⁷_{indolinone}, J 7.7 Hz), 7.11 t.d (1H, C₆H₄, $H_{indolinone}^{5}$, ¹*J* 7.6, ²*J* 1.0 Hz), 7.26 s (1H, 2-CH=C), 7.38 t.d (1H, C₆H₄, H⁶_{indolinone}, ¹J 7.7, ²J 1.0 Hz), 7.73 br.d (1H, C₆H₄, H⁴_{indolinone}, J 7.6 Hz), 14.15 br.s (1H, NH). ¹³C NMR spectrum, δ, ppm: 13.5 (CH₃), 20.5 (CH₃), 22.2 (CH₂), 25.2 (CH₂), 26.0 (NCH₃), 29.7 (CH₂), 108.8 (CH), 120.5 (CH), 121.9, 122.6 (CH), 126.4, 129.1 (CH), 130.8 (CH), 131.8, 142.4, 149.9, 158.2, 160.3, 166.4. Found, %: C 70.68; H 6.31; N 12.85. C₁₉H₂₁N₃O₂. Calculated, %: C 70.57; H 6.54; N 12.99.

3-[(*Z*)-1-(4-Methyl-6-oxo-5-cyclopentyl-1,6-dihydropyrimidin-2-yl)methylidene]-1-methylindolin-2-one (5b) was prepared from 1-methylindoline-2,3dione and pyrimidine 1b. Yield 51%, mp 263–265°C, $R_{\rm f}$ 0.78 (ethanol–dichloroethane, 1 : 1). IR spectrum, v, cm⁻¹: 1671, 1641 (CO), 1605 (C=C–C=N). ¹H NMR spectrum, δ , ppm: 1.56–1.74 m (4H) and 1.84–2.11 m [4H, (CH₂)₄], 2.35 s (3H, 4-CH₃), 3.00–3.13 m (1H, CH_{cyclopentyl}), 3.36 s (3H, NCH₃), 7.01 d (1H, C₆H₄, J 7.8 Hz), 7.10 t (1H, C₆H₄, J 7.6 Hz), 7.26 s (1H, 2-CH=C), 7.38 d.d (1H, C₆H₄, ¹J 7.8, ²J 7.6 Hz), 7.73 d (1H, C₆H₄, J 7.6 Hz), 14.01 br.s (1H, NH). Found, %: C 71.77; H 6.18; N 12.40. C₂₀H₂₁N₃O₂. Calculated, %: C 71.62; H 6.31; N 12.53.

3-[(Z)-1-(5-Benzyl-4-methyl-6-oxo-1,6-dihydropyrimidin-2-yl)methylidene]-1-methylindolin-2-one (5c) was prepared from 1-methylindoline-2,3-dione and pyrimidine 1g. Yield 62%, mp 228–230°C, $R_{\rm f}$ 0.82 (ethanol-dichloroethane, 1 : 1). IR spectrum, v, cm^{-1} : 1703, 1675 (CO), 1642, 1601 (C=C-C=N). ¹H NMR spectrum, δ, ppm: 2.31 s (3H, CH₃), 3.36 s (3H, NCH₃), 3.87 s (2H, CH₂), 7.02 br.d (1H, H_{isatin}^7 , J 7.8 Hz), 7.11 t.d (1H, C₆H₄, J 7.8, 0.9 Hz), 7.09-7.16 m (1H, C₆H₅), 7.20-7.23 m (4H, C₆H₅), 7.29 s (1H, =CH), 7.38 t.d (1H, C₆H₄, J 7.8, 1.0 Hz), 7.74 br.d (1H, H⁴_{isatin}, J 7.6 Hz), 14.32 br.s (1H, NH). ¹³C NMR spectrum, δ , ppm: 21.0 (CH₃), 25.6 (CH₃), 30.6 (NCH₃), 108.9 (CH), 120.6 (CH), 121.8, 122.6 (CH), 124.8, 125.4 (CH), 127.7 (2CH, Ph), 127.8 (2CH, Ph), 129.6 (CH), 130.9 (CH), 132.2, 149.9, 138.8, 142.5, 150.5, 159.7, 160.7. Found, %: C 74.05; H 5.53; N 11.90. C₂₂H₁₉N₃O₂. Calculated, %: C 73.93; H 5.36; N 11.76.

2-[(E)-2-(2-Hydroxy-6-methoxyquinol-3-yl)-1ethenyl]-4-methyl-5-propyl-1,6-dihydropyrimidin-6one (6) was prepared from 6-methoxy-2-chloro-3quinolinecarbaldehyde and pyrimidinone 1a. Yield 38%, mp 320°C, $R_{\rm f}$ 0.67 (isobutanol-H₂O-AcOH, 2 : 2 : 1). IR spectrum, v, cm^{-1} : 1640 (CO), 1584 (C=C-C=N). ¹H NMR spectrum, δ , ppm: 0.92 t (3H, CH₃CH₂CH₂, J 7.3 Hz), 1.39–1.50 m (2H, CH₂CH₃), 2.28 s (3H, CH₃), 2.37-2.43 m (2H, CH₂C₂H₅), 3.87 s (3H, OCH₃), 7.21 d.d (1H, H⁷, C₆H₃, J 8.9, 2.8 Hz), 7.27 d (1H, H^5 , C_6H_3 , J 2.8 Hz), 7.28 d (1H, H^8 , C_6H_3 , J 8.9 Hz), 7.57 d [1H, (E) =CH, J 15.9 Hz], 7.83 d [1H, (E) = CH, J 15.9 Hz], 8.17 s (1H, H⁴), 11.96 s(1H, OH), 12.34 br.s (1H, NH). Found, %: C 68.57; H 5.85; N 11.67. C₂₀H₂₁N₃O₃. Calculated, %: C 68.36; H 6.02; N 11.96.

5-Allyl-4-methyl-2-[(*1E*,*3E*)-4-phenylbuta-1,3dienyl]-1,6-dihydropyrimidin-6-one (7a) was prepared from (2*E*,4*E*)-5-phenylpenta-2,4-dienal and pyrimidine 1b. Yield 48%, mp 200–202°C, $R_{\rm f}$ 0.57 (acetone–hexane, 2 : 3). IR spectrum, v, cm⁻¹: 1649, 1625 (CO), 1585 (C=C–C=N). ¹H NMR spectrum, δ, ppm: 2.23 s (3H, CH₃), 3.18 d.t (2H, CH₂, *J* 6.2, 1.5 Hz), 4.95 d.q (1H, =CH<u>H</u>, *J* 10.0, 1.5 Hz), 5.00 d.q (1H, =C<u>H</u>H, *J* 17.1, 1.5 Hz), 5.79 d.d.t (1H, =CH, *J* 17.1, 10.0, 6.2 Hz), 6.32 d [1H, (*E*)=CH, *J* 15.2 Hz], 6.86 d [1H, (*E*)=CH, *J* 15.5 Hz], 6.98 d.d [1H, (*E*)=CH, *J* 15.5, 10.5 Hz], 7.20–7.34 m (3H, $H_{Ph}^{3,4,5}$), 7.45–7.50 m (2H, $H_{Ph}^{2,6}$), 7.62 d.d [1H, (*E*)=CH, *J* 15.2, 10.5 Hz], 12.20 br.s (1H, NH). Found, %: C 77.83; H 6.47; N 10.30. C₁₈H₁₈N₂O. Calculated, %: C 77.67; H 6.52; N 10.06.

5-Butyl-4-methyl-2-[(1E,3E)-4-phenylbuta-1,3dienyl]-1,6-dihydropyrimidin-6-one (7b) was prepared from (2E, 4E)-5-phenylpenta-2,4-dienal and pyrimidine 1c. Yield 59%, mp 184-186°C, R_f 0.61 (acetone-hexane, 2 : 3). IR spectrum, v, cm^{-1} : 1646, 1626 (CO), 1570 (C=C-C=N). ¹H NMR spectrum, δ, ppm: 0.96 t (3H, CH₃, J 7.0 Hz), 1.32-1.48 m (4H, CH₂CH₂Me), 2.23 s (3H, CH₃), 2.35–2.44 m (2H, $CH_2C_3H_7$), 6.30 d [1H, H¹ (E)=CH, J 15.3 Hz], 6.85 d [1H, H⁴(E)=CH, J 15.6 Hz], 6.97 d.d [1H, H³(E)=CH, J 15.6 Hz], 6.97 Hz],J 15.6, 10.6 Hz], 7.20–7.35 m (3H, H^{3,4,5}), 7.45–7.50 m (2H, H^{2, 6}_{Ph}), 7.59 d.d [1H, H² (E)=CH, J 15.3, 10.6 Hz], 12.09 br.s (1H, NH). Found, %: C 77.37; H 7.68; N 9.40. C₁₉H₂₂N₂O. Calculated, %: C 77.52; H 7.53; N 9.52.

5-Isopentyl-4-methyl-2-[(*1E*,*3E*)-4-phenylbuta-**1,3-dienyl]-1,6-dihydropyrimidin-6-one** (7c) was prepared from (2*E*,4*E*)-5-phenylpenta-2,4-dienal and pyrimidine **1e**. Yield 51%, mp 192–194°C, $R_{\rm f}$ 0.60 (acetone–hexane, 2 : 3). IR spectrum, v, cm⁻¹: 1644, 1630 (CO), 1582 (C=C–C=N). ¹H NMR spectrum, δ , ppm: 0.97 d (6H, 2Me, *J* 6.6 Hz), 1.26–1.35 m (2H, <u>CH</u>₂CH), 1.62 nonet (1H, CH, *J* 6.6 Hz), 2.23 s (3H, CH₃), 2.36–2.43 m (2H, <u>CH</u>₂CH₂CH), 6.30 d [1H, (*E*) =CH, *J* 15.3 Hz], 6.85 d [1H, (*E*)=CH, *J* 15.5 Hz], 6.97 d.d [1H, (*E*)=CH, *J* 15.5, 10.5 Hz], 7.20–7.34 m (3H, $H_{P'h}^{3,4,5}$), 7.45–7.50 m (2H, $H_{Ph}^{2,6}$), 7.59 d.d [1H, (*E*)=CH, *J* 15.3, 10.5 Hz], 12.08 br.s (1H, NH). Found, %: C 78.05; H 7.53; N 9.28. C₂₀H₂₄N₂O. Calculated, %: C 77.89; H 7.84; N 9.08.

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