A natural carbohydrate substrate for *Mycobacterium tuberculosis* methionine sulfoxide reductase A[†]

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Enzymatic reduction of the methylsulfinylxylofuranosyl (MSX) groups in lipoarabinomannan provides proof of the absolute configuration of MSX and a possible biochemical mechanism for oxidative protection in *Mycobacterium tuberculosis*.

Tuberculosis (TB) remains a major threat to world health,¹ with 9.2 million new cases and 1.7 million deaths from TB in 2006.² The disease is caused by *Mycobacterium tuberculosis* (Mtb) which invades and colonises macrophage cells. The ability of Mtb to survive within this inhospitable environment has been attributed to its robust cell wall,³ that comprises complex glycolipids including mycolyl-arabinogalactan-peptidoglycan (mAGP)⁴ and lipoarabinomannan (LAM).⁵ LAM facilitates the entry of the bacterium into macrophages,⁵ prevents macrophage activation, and protects Mtb from damage by superoxide and hydroxyl radicals.⁶ Although macrophages also produce H₂O₂ during a respiratory burst,⁷ the effect of H₂O₂ on LAM was not reported alongside the studies on superoxide and hydroxyl radicals.⁶

Recent structural studies have revealed two novel sugar substituents on the terminal mannose caps of LAM,^{8,9} which were identified as an α -5-methylthioxylofuranosyl (MTX) group and its corresponding sulfoxide derivative (MSX) (Scheme 1b).^{10,11} The cell surface location of MTX⁸ will make it particularly exposed to damage by reactive oxygen species (ROS) produced by macrophages. Oxidation of the sulfide group would almost certainly modulate any functional interactions between the sugar and other biomolecules. For example, an MTX- α (1,4)-mannosyl disaccharide has been shown to inhibit the production of cytokine TNF- α in the human monocyte cell line THP-1,¹¹ but the effect is reduced upon oxidation of the sugar.

Conversely, sacrificial oxidation of MTX would sequester ROS from solution and thus prevent other sensitive molecules from oxidative damage. Levine has proposed¹² that

methionine residues in proteins could provide anti-oxidative protection by sequestering peroxides through a redox cycle (Scheme 1a) comprising oxidation by H_2O_2 and subsequent enzymatic reduction by methionine sulfoxide reductase (Msr) enzymes that are found in all organisms.^{13,14} Indeed, overexpression of the methionine sulfoxide reductase A enzyme (MsrA) in human fibroblasts can reduce the level of protein carbonyl formation when challenged with H₂O₂.¹⁵ The MsrA enzyme from Mycobacterium smegmatis has also been shown to be important for survival of mycobacteria inside macrophage cells.¹⁶ MsrA is known to accommodate alternative substrates bearing a methylthio group.¹⁷ Therefore, the structural similarities between methionine and MTX led us to consider if a similar redox cycle could be possible in which methionine is replaced by MTX (Scheme 1b). We also use this oxidation/reduction mechanism to provide a stereochemical proof for the absolute configuration of MTX.

LAM was purified from the Mtb clinical isolate CSU20 as described previously.⁸ ¹H NMR spectroscopy of CSU20-LAM (Fig. 1a) indicated that there was approximately one MTX or MSX residue for every 45 monosaccharide units. This ratio corresponds to one xylofuranosyl residue per LAM molecule, or one MTX/MSX for every 5-6 mannose caps.^{5,8} When one considers the relative concentrations of proteins and LAM in mycobacterial cell walls,¹⁸ MTX could be as prevalent as methionine residues in this part of the cell. The presence of two overlapping doublets (corresponding to H-1) at ca. 5.45 ppm indicated that the sample was already partially oxidised to a 1 : 1 mixture of MSX diastereoisomers. The occurrence of MSX in previous studies has been attributed to oxidation during purification of the LAM.⁸ Upon addition of excess H2O2 to the sample, the MTX anomeric signal decreased in size to be replaced by the overlapping MSX



Scheme 1 Oxidation and reduction of (a) methionine and (b) MTX.

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Fig. 1 Anomeric (H-1) region of the ¹H NMR spectra of LAM (a) before oxidation, and after successive treatment with (b) H_2O_2 and (c) MsrA–DTT.

anomeric signals (Fig. 1b). It should be noted that no other anomeric signals in the spectrum changed when the sample was treated with H₂O₂. Therefore, it can be concluded that MTX is oxidised selectively to MSX by H₂O₂, and no other monosaccharide component of LAM is capable of sequestering H₂O₂. This observation was corroborated by experiments on methyl α -mannopyranoside and methyl α/β -arabinofuranoside which showed no oxidation by H₂O₂ under conditions that allowed complete conversion of MTX to MSX (data not shown).

Mycobacterium tuberculosis has two genes encoding Msr enzymes: MsrA and MsrB which reduce *S*- and *R*-configured sulfoxides, respectively. Cell wall/membrane and soluble protein fractions were prepared according to Mueller-Ortiz *et al.*¹⁹ Immunoblotting experiments showed that while MsrB is confined to the cytosol of the bacterium, MsrA is present in both the soluble and membrane/cell wall fractions (Fig. 2). As deletion of the gene encoding MsrB is generally less important for anti-oxidative protection than MsrA,¹⁴ we chose to focus first on reduction of MSX by MsrA.

MSX-LAM was treated with MsrA in the presence of excess dithiothreitol (DTT) as a co-reductant to regenerate the



Fig. 2 Immunoblot analysis of MsrA and MsrB in different fractions of Mtb cells: (a) whole Mtb cell, (b) the cytosolic fraction and (c) the cell wall/membrane fraction.

catalytically active form of the enzyme. The ¹H NMR spectrum of the reaction mixture showed that the low field MSX anomeric signal disappeared to be replaced by the anomeric signal for MTX (Fig. 1c). The MTX anomeric signal is slightly larger than that for the remaining MSX as a consequence of incomplete oxidation of the original MTX residues (Fig. 1b). These results demonstrate that MSX is reduced stereoselectively by MsrA. A molecular model of methionine sulfoxide docked into the crystal structure of Mtb MsrA indicates that only the *S*-configured sulfoxide can be accommodated in the enzyme active site.²⁰ Therefore, we would assign the remaining MSX signal to correspond to the *R*-sulfoxide. Treatment of MSX-LAM with DTT alone had no effect on the NMR spectrum, and hence did not modify the polysaccharide structure.

The stereoselective reduction of MSX presented us with an opportunity to determine the absolute stereochemistry of MTX while still attached to the polysaccharide. Inspection of a model of MSX docked into the MsrA crystal structure indicated that little of the MSX structure would be bound within the enzyme active site (Supporting information Fig. S1[†]). Therefore, either D- or L-configured MSX should be able to act as a substrate for the enzyme, but in each case only the *S*-configured sulfoxide should be reduced. The remaining *R*-sulfoxides would be a pair of diastereoisomers and would thus have different NMR spectra. We anticipated that only one of these isomers would match the spectrum of the natural polysaccharide.

The methyl α -furanosides of D- and L-MTX were prepared by chemical synthesis (see supporting information†) and oxidised to MSX by exposure to H₂O₂. No evidence for over-oxidation to the sulfone was observed. The ¹H NMR



S-sulfinyl-D-MSX

spectra for D- and L-MSX were identical (Fig. 3a,d), as expected for pairs of enantiomers. The difference in chemical shift for the anomeric proton of methyl α-MSX from that of MSX-LAM (Fig. 1b) can be attributed to the different anomeric substituents.¹⁰ Upon treatment with MsrA and DTT, only the S-sulfoxide isomers were reduced to give mixtures of MTX and either the *R*-methylsulfinyl-D-xylofuranoside (Fig. 3b) or *R*-methylsulfinyl-L-xylofuranoside (Fig. 3c).²¹ As the downfield MSX anomeric signal is lost for both MSX-LAM (Fig. 1c) and methyl α -D-MSX (Fig. 3b), we can thus conclude that MTX is p-configured. This result provides independent corroboration of the findings of Lowary and co-workers who compared the NMR spectra of six synthetic MTX-mannosyl disaccharides with the published NMR spectra of LAM.¹¹

R-sulfinyl-D-MSX

11 O

D-MTX

L-MTX

MeO/

S-sulfinyI-L-MSX

5.05

НÒ

NOMe

Me

L-MSX

`S // 0

(ppm)

юн

D-MSX

In conclusion, we have demonstrated that exposure of LAM to the biological oxidant H2O2 results in oxidation of only the MTX substituent. Stereoselective reduction of the S-sulfoxide isomer of MSX-LAM using MsrA provides proof of the absolute configuration of this novel sugar without the need to remove the sugar from the polysaccharide. While oxidation of MTX would almost certainly affect its biological function in vivo, this damage could be repaired, in part, by MsrA which is present in the mycobacterial cell wall/membrane fraction. Alternatively, there exists the possibility that MSX on the surface of M. tuberculosis may be reduced by host MsrA and MsrB when the bacteria reside inside macrophages. If so, then MSX would be the first natural non-protein substrate for these enzymes. Furthermore, this redox cycle of chemical oxidation

and enzymatic reduction could also provide a mechanism for more general anti-oxidative protection, as has been established previously for methionine oxidation.^{12,15} Indeed, as H_2O_2 is a direct precursor of hydroxyl radicals in vivo, sequestration of H₂O₂ by MTX–MsrA could also reduce the production of OH• in the mycobacterial cell wall. Future studies will focus on testing this hypothesis in vivo.

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