

View Article Online View Journal

ChemComm

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: M. Conejero Muriel, J. A. Gavira, E. Pineda-Molina, A. Belsom, M. Bradley, M. Moral, J. D. Duran, A. Luque-Gónzalez,



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

ChemComm

RSCPublishing

Journal Name

Influence of the Chirality of Short Peptide

Cite this: DOI: 10.1039/xoxxooooox

Received ooth January 2012, Accepted ooth January 2012

DOI: 10.1039/x0xx00000x

www.rsc.org/

Supramolecular Hydrogels in Protein Crystallogenesis

Mayte Conejero-Muriel,^a José A. Gavira,*^a Estela Pineda-Molina,^a Adam Belsom,^b Mark Bradley,^b Mónica Moral,^c Juan de Dios García-López Durán,^c Angélica Luque González,^d Juan J. Díaz-Mochón,^{*d} Rafael Contreras-Montoya,^e Ángela Martínez-Peragón,^e Juan M. Cuerva,^e and Luis Álvarez de Cienfuegos^{*e}

For the first time the influence of the chirality of the gel fibers in protein crystallogenesis has been studied. Enantiomeric hydrogels 1 and 2 were tested with model proteins lysozyme and glucose isomerase and a formamidase from B. cereus. Crystallization behaviour and crystal quality of these proteins in both hydrogels are presented and compared.

Chirality is ubiquitous in nature and has major implications in cell interactions and biological processes. Within this topic, chiral discrimination on chiral surfaces has significant effects in protein adsorption and cellular adhesion and proliferation.¹ It would be expected that similar chiral discrimination were relevant in 3D-enviroments, such as in protein crystallogenesis. Remarkably, the expected influence of the chirality in protein crystallogenesis remains unexplored probably owing to the difficulty to obtain a pair of enantiomeric environments suitable for crystallization. In this sense, the use of stereochemically well-defined fiber hydrogels can be an ideal media to study the influence of the chirality in the process of crystal growth. Recently, peptide hydrogels have found useful applications in biology due to their biocompatible and biodegradable nature.² It has been shown that these supramolecular fibrous aggregates interact selectively with proteins³ and also reduce denaturation.⁴ Considering this, we wondered if supramolecular hydrogels based on the self-assembly of short peptides could be compatible with the process of protein crystallization since gels are used as an excellent media for crystallization.⁵ It is known that peptides hydrogels are able to form stereochemically well defined 3D ordered structures⁶ and, therefore, the inherent chirality of peptides can interact diastereoisomerically with proteins, thus expecting different behaviors. It is worth noting that the structural simplicity of short peptides is essential for this purpose, allowing an easy access of their corresponding enantiomers. This property is not feasible with other common macromolecular hydrogels, which either have no stereocenters

(polyacrylamide) or are constituted by the natural chiral product (agarose).

To test this hypothesis we selected two cysteine-based peptides, compound 1 (N,N)-di(benzovl)-L-cysteine diamide) and its corresponding D enantiomer 2 (N,N)-di(benzoyl)-D-cysteine diamide) (Figure 1) due to their capacity to self-assembly in neat water to give the corresponding hydrogels.⁷ They were tested with two model proteins, chicken HEWL lysozyme and glucose isomerase, and one target protein, a formamidase from B. cereus being the first time that supramolecular peptide based hydrogels are used for this purpose. Additionally, the results were compared with crystal growth in agarose gel to investigate their potential in protein crystallization.

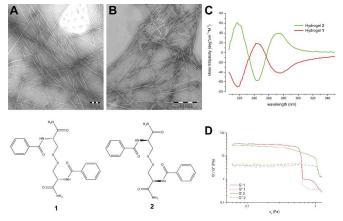


Figure 1. Characterization of hydrogels 1 and 2 by TEM microscopy, circular dichroism and rheology. TEM images of negatively stained dried hydrogels at 3 mM: A) 1; B) 2. Scale bar at 0.2 µm. C) CD spectra and D) rheology of hydrogels 1 and 2.

Compounds 1 and 2 were selected from a family of 22 cysteine derivatives⁷ based on its capacity to form a stable and completely transparent hydrogel. TEM images of hydrogel 1 showed self-assembled nanofibers of diameters ranging from Published on 29 January 2015. Downloaded by Memorial University of Newfoundland on 31/01/2015 05:34:37.

10 to 20 nm and lengths that go up to μ m in size (Figure 1).⁸ Fibrous aggregates co-exist with regions where fibers are more dispersed. Hydrogel 2 presented fibers of the same aspect. To gain insight into the molecular arrangement of these hydrogels circular dichroism (CD) spectra were measured (Figure 1). Hydrogel 1 exhibited a negative band near 218 nm ($n\pi^*$ transition), a positive band near 250 nm ($n\sigma^*$ disulfide band and/or $\pi\pi^*$ of the aryl groups) and a negative band near 280 nm $(\pi\pi^*$ of the aryl groups) in good agreement with the CD spectrum from other oligopeptides hydrogels9 identified as the β -sheet conformation. The CD spectrum of hydrogel 2 is the specular image of hydrogel 1 showing that the chirality of the peptides is conserved and transferred supramolecularly to the fibers. Hydrogels 1 and 2 were also analyzed using a regime of simple oscillatory shear stress through frequency (see Supporting Information) and amplitude sweep tests. Both hydrogels showed values of G' (elastic modulus) higher than G'' (viscous modulus) indicating the predominant elastic behavior of the gels (Figure 1). Based on the magnitude of the values of G', G'' and σ_c (G' \approx 30 Pa; G" \approx 4 Pa; $\sigma_c < 1$ Pa) we can conclude that these hydrogels are extremely weak. The similar values of G' and G'' for hydrogels 1 and 2 confirm that both enantiomeric hydrogels have the same rheological properties. VT-NMR studies for both hydrogels showed that the signals of the aromatic region were sifted 0.5 ppm to downfield as the temperature increased from 25 °C to 70 °C. The small values of T_2 (1.84 s for 7.74 ppm signal) and the negative values of nOe indicated that the ¹H-NMR signals corresponded to aggregates of the di-peptide which are not higher enough to be NMR-silent.¹⁰ The stability of these aggregates indicated that T_{gel} should be higher than 70 °C. In fact at 75/25 H₂O/DMSO mixture, the reported value of T_{gel} for 1 was above 90 °C.8 DSC scans in both directions of hydrogels^{9c} 1 and 2 did not show any endothermic or exothermic peaks. These results indicated continuous gel to sol and sol to gel transitions (see SI).

The two enantiomeric hydrogels **1** and **2** were then tested as the supramolecular medium for protein crystallization. Following the procedure described in SI, protein was allowed to diffuse in the gel prior to the addition of the precipitant. Figure 2A shows a typical counter-diffusion pattern of lysozyme crystals grown in hydrogel **1**. The effect of hydrogels **1** and **2** on the quality of lysozyme crystals was evaluated from a set of lysozyme crystals measured at beam-lines Xaloc (ALBA) following the standard quality criterion.¹¹

View Article Online DOI: 10.1039**/04009/024A10**

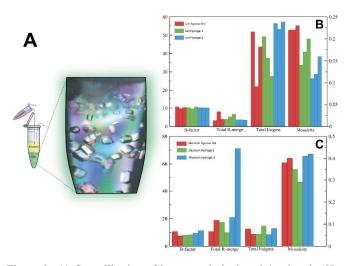


Figure 2. A) Crystallization of lysozyme in hydrogel **1** using the 2L configuration of the counterdiffusion technique set-up in Eppendorf tubes schematically represented. The evolution of the supersaturation is easily followed from top to bottom as an increase of crystal size and the reduction of the nucleation density (Right: optical microscopic picture under polarized light). B) and C) show the average values of standard quality indicators of lysozyme and glucose isomerase crystals, respectively, grown in hydrogels **1**, **2** and agarose from cryo-protected at 100 K keeping constant all data acquisition configuration set-up and analysis parameters.

Lysozyme crystals grown in hydrogel 2 were better than those obtained in hydrogel 1, reaching the resolution limit of the detector (0.95 Å) and having the best quality indicators in terms of $I/\sigma(I)$, Rmerge and mosaicity (Figure 2B). It is also worth mentioning the excellent reproducibility of crystals grown in hydrogel 2 (Figure S1 and Table S2). Moreover crystal quality indicators of crystals grown in agarose (able to produce protein crystals of high quality^{5a, d}) were similar than those obtained in hydrogel 1 but still inferior to hydrogel 2. Agarose gel has been proven to facilitate the soaking of different compounds^{5a} including cryo-protectants¹². In order to evaluate the possible cryo-protection capacity of these new hydrogels we compared the crystal quality of cryo-protected and "naked" (flash frozen without cryo-protectant) lysozyme crystals grown in hydrogel 1 and agarose from two data sets collected at beam-lines ID23 and BM30 (ESRF). It is important to notice that the polymeric nature of both agarose and hydrogel 1 allowed us to collect full data sets without needing to use any additive (Figure S2, S3 and Table S3).

A similar analysis was performed with glucose isomerase crystals grown in hydrogels **1**, **2** and agarose. In this case crystals obtained in hydrogel **1** and agarose were more homogeneous and of better quality than those grown in hydrogel **2** (Figure 2C, Table S4 and Figure S4). However, a second crystal form ($P2_12_12$) was identified only in the experiments run with hydrogel **2** that co-exist with the other polymorph (*I222*), the form normally found under similar crystallization conditions. Dauter and co-workers have described a new crystal form ($P2_12_12$) of glucose isomerase and have solved its structure (PDB ID. 1OAD).¹³ Remarkably, the polymorph found in hydrogel **2** presents different unit cell parameters (Å): 86.00, 93.68, 99.22 versus 98.45, 129.59, 78.33 for a, b and c respectively (Figure S4). To date there is only

one PDB entry (PDB ID. 101H) with similar unit cell and space group but obtained under different crystallization conditions.

The quality of this primitive orthorhombic crystal was the highest of all glucose isomerase crystals tested, with the structure determined at 1.20 Å and refined to R/Rfree values of 10.73/13.22 (%) (Table S5). When compared with the model used for molecular replacement (PDB. ID 10AD) the deviation of the main chain is minimal (rmsd of 0.3). It is worth mentioning that this new polymorph has only been obtained in free solution using other precipitant, as in the case of the PDB ID. 101H that crystallized using MPD, or in the presence of high agarose concentration.^{5a} In our case, under identical crystallization condition, the $P2_12_12$ polymorph was obtained only with hydrogel 2 and not in agarose at 0.5% (w/v), our control, suggesting a novel interaction between the chiral gel fibers and the protein at the nucleation stage. It has been previously demonstrated that the nature of the gel fibers, such as agarose¹⁵ and silica,¹⁶ affect protein nucleation density (promoted and inhibited respectively) with no effect on the final crystal packing i.e. formation of new polymorph. Up to date, the correlation between the protein-fiber interaction and polymorphism is unknown. Taking into account that both, chemical composition and the physical properties of the media are identical (hydrogels 1 and 2), we hypothesized that the formation of a different polymorph has to be an effect of the chirality at the nucleation state in terms of stabilization or induction of the second polymorph.

It is also known that the incorporation of hydrogel fibers in the protein crystal lattice occurs with agarose, silica and PEG hydrogels.^{5a,15a,16a,c,17} To prove that the same effect also happens with peptide hydrogels, dissolution experiments of lysozyme crystals grown in hydrogel **1** were carried out (Figure 3). This result showed that effectively the peptide fibers are incorporated within the protein crystals. The un-correlated behavior of incorporation and nucleation effects opens a question on the different roles that gel nature can have in the whole process of protein crystallogenesis.

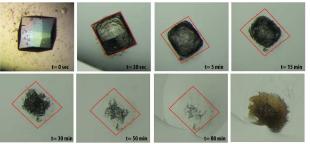


Figure 3. Dissolution sequence of a lysozyme crystal grown in hydrogel 1. Crystal extracted from the gel was placed in a water drop and allowed to dissolve (t = 0 sec). A red square is drawn around the remaining hydrogel. The last picture was taken after partially drying of the remaining gel framework.

Finally, formamidase from *B. cereus*, was selected as a target protein to test its crystallizability using hydrogel 1, 2, and agarose as a control. Good diffracting crystal was already obtained by capillary counter diffusion method diffracting to a maximum resolution of 1.78 Å (Table S6). From previous

experiments we knew that agarose used in combination with capillary counterdiffusion did not improve or even fully impeded crystallization. Crystals grown in three different crystallization conditions (Table S7) were evaluated. Only cryo-protected crystals obtained in hydrogel **2** using the condition C18 of the GSK crystallization screening kit (Triana S&T, Granada, Spain) produced the best crystals to date (1.4 Å resolution limit, Table S6). On the other hand hydrogel **1** inhibited nucleation while crystal grown in agarose were of low quality proving the different behavior of these hydrogels.

To sum up, we present the novel use of supramolecular hydrogels based on short-peptides in neat water for protein crystallization. These hydrogels, made of symmetrical L and D di-cysteine derivatives, were used to grow protein crystals of high quality. Differences in crystal quality and packing for the three proteins have been observed using enantiomeric hydrogels 1 and 2, suggesting a significant influence of the fibers' chirality in the crystallization process. The formation of the orthorhombic polymorph of glucose isomerase and the growth of the highest quality crystals of formamidase exclusively in hydrogel 2, support this hypothesis. This fact highlights the relevance that chirality may have in protein crystallogenesis for X-ray structural determination. Moreover these supramolecular peptide based hydrogels are excellent mediators for protein crystallization producing lysozyme and glucose isomerase and formamidase crystals with excellent quality indicators. Peptide based hydrogels also acted as a cryoprotectant avoiding the use of additives. Hence, these novel chiral gels would expand the field of protein crystallogenesis.

Notes and references

This research was funded by the MICINN (Spain) projects BIO2010-16800 (JAG), CTQ-2011.22455 (LAC & JMC), CTQ2012-34778 (JJDM & ALG), "Factoría Española de Cristalización" Consolider-Ingenio 2010 (JAG & MCM) and EDRF Funds (JAG, LAC & JMC), P12-FQM-2721 (LAC) Junta de Andalucía. JJDM thanks MICINN for a R&C Fellowship and MCM thanks CSIC for her JAE Fellowship. We would like to thank Dr. S. Martinez-Rodriguez for providing the plasmids of formamidase and B. Morel and E. Sánchez-Cobos for their help with the CD studies. We are very grateful to the staff at beam-line BM30, ID29 (ESRF) and Xaloc (ALBA) for support during X-ray data collection.

^a Laboratorio de Estudios Cristalográficos, Instituto Andaluz de Ciencias de la Tierra (CSIC-UGR), Av. de las Palmeras 4, 18100 Armilla, Granada, Spain. E-mail: jgavira@iact.ugr-csic.es, ^b School of Chemistry, University of Edinburgh, West Mains Road, EH9 3JJ Edinburgh, United Kingdom. ^c Dpt. de Física Aplicada, Facultad de Ciencias (UGR), 18071-Granada, Spain.^d Facultad de Farmacia, Dpt. de Química Farmacéutica y Orgánica (UGR). Centre for Genomics and Oncological Research: Pfizer/UGR/Andalusian Regional Government, PTS Granada, Av. de la Ilustración 114, 18016 Granada, Spain. Email: juandiaz@ugr.es, e Dpt. de Química Orgánica, Facultad de Ciencias (UGR). E-mail: lac@ugr.es, Supplementary Information (ESI) available: Electronic General experimental details. ¹H NMR and ¹³C NMR of 1 and 2. VT-NMR and DSC spectra of hydrogel 1 and 2. Crystallization and crystallography data. See DOI: 10.1039/b000000x/

 W. Wei, C. Xu, N. Gao, J. Ren, X. Qu, *Chem. Sci.* 2014, **5**, 4367-4374; G. Qing, S. Zhao, Y. Xiong, Z. Lv, F. Jiang, Y. Liu, H. Chen, M. Zhang, T. Sun, *J. Am. Chem. Soc.* 2014, **136**, 10736-10742; K. Benson, H.-J. Galla, N. S. Kehr, *Macromol. Biosci.* 2014, **14**, 793-798; Z. Li, A. Köwitsch, G. Zhou, T. Groth, B. Fuhrmann, M. Niepel, E. Amado, J. Kressler, *Adv. Healthcare Mater.* 2013, **2**, 1377-1387; J. El-Gindi, K. Benson, L. De Cola, H.-J. Galla, N. S. Kehr, *Angew.* Chem. Int. Ed. 2012, **51**, 3716-3720; M. Zhang, G. Qing, T. Sun, Chem. Soc. Rev. 2012, **41**, 1972-1984.

- J. Shi, X. Du, D. Yuan, J. Zhou, N. Zhou, Y. Huang, B. Xu, Biomacromol. 2014, 15, 3559–3568; A. Dasgupta, J. H. Mondal, D. Das, RSC Adv. 2013, 3, 9117-9149; N. Javid, S. Roy, M. Zelzer, Z. Yang, J. Sefeik, R. V. Ulijn, Biomacromol. 2013, 14, 4368–4376; R. Orbach, L. Adler-Abramovich, S. Zigerson, I. Mironi-Harpaz, D. Seliktar, E. Gazit, Biomacromol. 2009, 10, 2646–2651; F. Zhao, ML. Ma, B. Xu, Chem. Soc. Rev. 2009, 38, 883-891.
- 3 Y. Kuang, D. Yuan, Y. Zhang, A. Kao, X. Du, B. Xu, *RSC Adv.* 2013, **3**, 7704-7707; Y. Gao, M. J. C. Long, J. Shi, L. Hedstrom, B. Xu, *Chem. Commun.* 2012, **48**, 8404-8406.
- 4 S. Kiyonaka, K. Sada, I. Yoshimura, S. Shinkai, N. Kato, I. Hamachi, *Nat. Mater.* 2004, 3, 58-64.
- 5 S. Sugiyama, M. Maruyama, G. Sazaki, M. Hirose, H. Adachi, K. Takano, S. Murakami, T. Inoue, Y. Mori, H. Matsumura, J. Am. Chem. Soc. 2012, 134, 5786-5789; Y. Diao, K. E. Whaley, M. E. Helgeson, M. A. Woldeyes, P. S. Doyle, A. S. Myerson, T. A. Hatton, B. L. Trout, J. Am. Chem. Soc. 2012, 134, 673-684; J. A. Foster, M.-O. M. Piepenbrock, G. O. Lloyd, N. Clarke, J. A. K. Howard, J. W. Steed, Nat. Chem. 2010, 2, 1037-1043; B. Lorber, C. Sauter, A. Theobald-Dietrich, A. Moreno, P. Schellenberger, M.-C. Robert, B. Capelle, S. Sanglier, N. Potier, R. Giege, R. Prog. Biophys. Mol. Biol. 2009, 101, 13-25; R. I. Petrova, J. A. Swift, J. Am. Chem. Soc. 2004, 126, 1168-1173; C. Daiguebonne, A. Deluzet, M. Camara, K. Boubekeur, N. Audebrand, Y. Gérault, C. Baux, O. Guillou, Cryst. Growth Des. 2003, 3, 1015-1020.
- G. Qing, X. Shan, W. Chen, Z. Lv, P. Xiong, T. Sun, Angew. Chem. Int. Ed. 2014, 53, 2124-2129; S. Marchesan, C. D. Easton, K. E. Styan, L. J. Waddington, F. Kushkaki, L. Goodall, K. M. McLean, J. S. Forsythe, P. G. Hartle, Nanoscale, 2014, 6, 5172–5180; Silvia Marchesan, Y. Qu, L. J. Waddington, C. D. Easton, V. Glattauer, T. J. Lithgow, K. M. McLean, J. S. Forsythe, P. G. Hartley, Biomaterials 2013, 34, 3678-3687; J. Li, Y. Kuang, Y. Gao, X. Du, J. Shi, B. Xu, J. Am. Chem. Soc. 2013, 135, 542–545; S. Marchesan, C. D. Easton, F. Kushkaki, L. Waddington, P. G. Hartley, Chem. Commun., 2012, 48, 2195–2197; Y. Huang, Z. Qiu, Y. Xu, J. Shi, H. Lin, Y. Zhang, Org. Biomol. Chem., 2011, 9, 2149–2155; X. Yan, P. Zhu, J. Li, Chem. Soc. Rev. 2010, 39, 1877-1890.
- 7 A. Belsom, Doctor of Philosophy, PhD, School of Chemistry, University of Edinburgh 2010.
- 8 F. M. Menger, K. L. Caran, J. Am. Chem. Soc. 2000, 122, 11679-11691.
- 9 S. Marchesan, L. Waddington, C. D. Easton, F. Kushkaki, K. M. McLean, J. S. Forsythe, P. G. Hartley, *BioNanoSci.* 2013, **3**, 21-29; S. Marchesan, L. Waddington, C. D. Easton, D. A. Winkler, L. Goodall, J. Forsythe, P. G. Hartley, *Nanoscale*, 2012, **4**, 6752–6760; S. Maity, P. Kumar, D. Haldar, *Soft Matter*, 2011, **7**, 5239–5245.
- D. C. Duncan, D. G. Whitten, *Langmuir* 2000, 16, 6445-6452; B. Escuder, M. LLusar, J. F. Miravet. *J. Org. Chem.* 2006, 71, 7747-7752.
- 11 D. Maes, C. Evrard, J. A. Gavira, M. Sleutel, C. Van De Weerdt, F. Otalora, J. M. Garcia-Ruiz, G. Nicolis, J. Martial, K. Decanniere, *Cryst. Growth Des.* 2008, **8**, 4284-4290.
- 12 C. Biertumpfel, J. Basquin, D. Suck, C. Sauter, *Acta Crystallogr.*, Sect. D 2002, 58, 1657-1659.

- 13 U. A. Ramagopal, M. Dauter, Z. Dauter, Acta Crystallogr., Sect. D 2003, 59, 868-875.
- 14 C. M. Gillespie, D. Asthagiri, A. M. Lenhoff, *Cryst. Growth Des.* 2013, 14, 46-57.
- 15 J. A. Gavira, J. M. Garcia-Ruiz, *Acta Crystallogr., Sect. D* 2002, 58, 1653-1656; O. Vidal, M. C. Robert, F. Boué, *J. Cryst. Growth* 1998, 192, 257-270.
- 16 J. A. Gavira, A. E. S. V. Driessche, J. M. Garcia-Ruiz, *Cryst. Growth Des.* 2013, **13**, 2522-2529; O. Vidal, M. C. Robert, F. Boué, *J. Cryst. Growth* 1998, **192**, 271-281; J. M. García-Ruiz, J. A. Gavira, F. Otálora, A. Guasch, M. Coll, *Mater. Res. Bull.* 1998, **33**, 1593-1598.
- 17 J. A. Gavira, A. Cera-Manjarres, K. Ortiz, J. Mendez, J. A. Jimenez-Torres, L. D. Patiño-Lopez, M. Torres-Lugo, *Cryst. Growth Des.* 2014, 14, 3239–3248.

4 | J. Name., 2012, **00**, 1-3